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Pd(II)-Catalyzed Synthesis of Bicyclo[3.2.1] Lactones via Tandem Intramolecular β -C(sp³)-H Olefination and Lactonization of Free Carboxylic Acids

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Abstract

Bicyclo[3.2.1] lactones are chemical scaffolds found in numerous bioactive natural products. Herein, we detail the development of a novel palladium(II)-catalyzed tandem intramolecular β -C(sp³)-H olefination and lactonization reaction that rapidly transforms linear carboxylic acid possessing a tethered olefin into the bicyclo[3.2.1] lactone motif. This transformation features a broad substrate scope, shows excellent functional group compatibility, and can be extended to the preparation of the related seven-membered bicyclo[4.2.1] lactones. Additionally, we demonstrate the synthetic potential of this annulation by constructing the 6,6,5-tricyclic lactone core structure of the meroterpenoid cochlactone A. We anticipate that this compelling reaction may provide a novel synthetic disconnection that can be broadly applied towards the preparation of a variety of bioactive natural products.

Graphical Abstract

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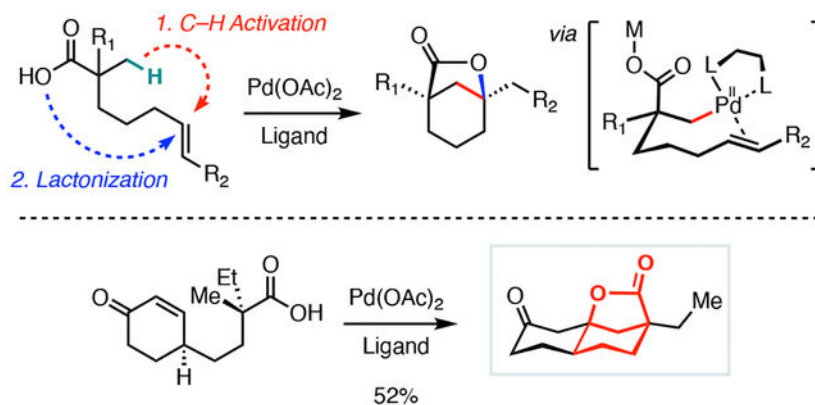
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Supporting Information

The Supporting Information is available free of charge on the ACS Publications website.

Full experimental details, product characterization, and X-ray crystallographic data (PDF)

The authors declare no competing financial interests.



Bicyclo[3.2.1] lactones represent a structural motif present in numerous bioactive natural products, such as picrotoxinin (**1**), dendrobine (**2**), and cochlactone A (**3**, Figure 1A).¹ Conventionally, this structure is accessed by synthesis of a functionalized six-membered ring with a pendant carboxylic acid (Figure 1B). From elaborated cyclohexenes, halolactonization or an acid/base-mediated opening of an epoxide ring can be used to form the lactone; however, both of these approaches introduce a potentially undesirable heteroatom that may need to be removed.² An alternative preparation of bicyclo[3.2.1] lactones from cyclohexanones employs a two-step protocol comprising of an initial carbonyl reduction followed by a hydroxyl-directed lactonization reaction.³ Successful implementations of this protocol can be limited by stereoselectivity issues in the initial reduction step. Given the recent developments in carboxylic-acid directed C–H functionalization reactions,⁴ we reasoned that it might be possible to synthesize this intriguing lactone scaffold *de novo* via a palladium(II)-catalyzed tandem intramolecular β -C(sp³)-H olefination and lactonization sequence of linear carboxylic acid precursors possessing a tethered olefin (Figure 1D). Such approach would significantly streamline synthetic access to these bridged scaffolds and would also provide a novel disconnection that was previously inaccessible via conventional methodologies.

To date, intramolecular C–H olefination strategies have only been successfully coupled with C(sp²)-H activation reactions. For instance, palladium mediated oxidative couplings between indoles and olefins have been reported as key steps in syntheses of several natural products.⁵ In 2004, Stoltz and co-workers disclosed the catalytic intramolecular Fujiwara–Moritani/oxidative Heck cyclization of arenes (Figure 1C).⁶ Although this impressive transformation provided access to functionalized benzofurans and dihydrobenzofurans, the non-directed electrophilic palladation mechanism was limited to highly electronrich arenes. Additionally, Wang and co-workers disclosed a directed macrocyclization protocol for the synthesis of stapled peptides via the intramolecular olefination of phenylalanines,⁷ arylsulfonamides,⁸ and arylacetamide.⁹ While intermolecular β -C(sp³)-H olefination reactions have been achieved with numerous directing groups, including perfluorinated *N*-arylamide,¹⁰ 8-aminoquinoline,¹¹ pyridine,¹² free carboxylic acid,^{4e} and native amides,¹³ the development of the corresponding intramolecular C(sp³)-H olefination reaction remains a significant challenge.

As part of our research program aimed at developing new C–H functionalization reactions that rapidly increase molecular complexity, we herein report a palladium(II)-catalyzed synthesis of bicyclo[3.2.1] lactones from linear carboxylic acids (Figure 1D). At the outset of our work, we hypothesized that this transformation may proceed by an initial carboxylic-acid directed β -C(sp³)-H activation followed by intramolecular olefin migratory insertion, β -H elimination of the resulting Pd(II)alkyl complex, and lactonization via a conjugate addition of the carboxylic acid. However, we recognized that several challenges might make this transformation difficult to realize. First, competitive chelation of the pendant double bond to the palladium center was recognized as a potential impediment to C–H activation. We surmised that the use of an electron-deficient alkene might solve this problem, but this raises the potential for a competitive intramolecular 7-*exo*-trig oxy-Michael addition prior to the C–H activation event. Second, the introduction of the olefin moiety into the backbone of the linear substrate **6** introduces a reactive allylic C–H bond that can compete with the desired β -C–H bond activation.¹⁴ Third, the putative carbon–carbon bond forming step would involve intramolecular migratory insertion of the alkylpalladium species **7** via the 6–4 bicyclic palladacycle **8**, which has been previously described only for a few examples of highly activated benzyl halide mediated C(sp³)-C(sp²) Heck reactions.¹⁵

We began our investigation by preparing the linear cyclization substrate **10a**, which was easily accessed in four steps from the commercial isobutyric acid (see the Supporting Information). Our group and others have showed that mono-*N*-protected amino acids (MPAAs) are excellent ligands for enabling a range of C(sp³)-H activation reactions of free carboxylic acids.¹⁶ Guided by these findings, our initial reaction conditions examined a series of commercially available MPAAs. We were pleased to find that the phenylalanine derivative (**L1**) provided the desired bicyclo[3.2.1] lactone product **11a** in a modest 43% yield without any detectable side products other than the unreacted **10a** (Scheme 1). Several additional ligands were examined, including the pyridine-based **L2** and **L3** and the monoprotected aminoethyl amine derivative **L4**, but these did not improve the reaction yield (39–42%). Additionally, we found that the use of the bidentate acyl-protected amino oxazoline **L5** or the monodentate pyridone **L6** resulted in a decreased yield (25% and 20%, respectively). Next, we turned our attention to thioether containing ligands that can act as soft σ -donors capable of stabilizing the palladium catalyst through their strong coordination.^{4e,17} When the *N*-acyl aminoethyl phenyl thioether (**L7**) was employed, we observed a significant increase in the yield of the product. Following additional modifications to the reaction conditions such as varying the base, bystandant oxidant, and solvent we were able to obtain the bicyclo[3.2.1] lactone product **11a** in 87% yield (see the Supporting Information).¹⁸

With the optimal reaction conditions in hand, we next investigated the substrate scope with respect to the olefin moieties (Scheme 2). Using olefin cross metathesis with the appropriate acrylate, it was possible to prepare a range of annulation precursors bearing a variety of different functional groups. We found that simple alkyl- (**11b–11e**) and the methoxyethyl ether acrylates **11f** were tolerated in excellent yields ranging from 73% to 82%. Other electron-withdrawing substituents such, as the methyl ketone (**11g**), cyanide (**11h**), and phosphonate (**11i**) also provided the bicyclo[3.2.1] lactones in high yields (82–88%).

While the tertiary amides (**11j**–**11l**) were compatible with the intramolecular β -C(sp³)-H olefination and lactonization reaction, products were formed in slightly lower yields (53–64%). We attribute this to the reduced electron withdrawing nature of the unsaturated amides compared to the corresponding unsaturated esters. Lastly, the sulfonyl phenyl group was also tolerated and provided the crystalline **11m** product in 78% yields. In the absence of a terminal electron withdrawing group we did not observe formation of the bicyclo[3.2.1] lactone but instead obtained product derived from allylic C–H functionalization (see the Supporting Information).¹⁹

A wide variety of substituents at the β -position of the carboxylic acid residue were found to be compatible with reactivity occurring exclusively at the methyl C–H bond (Scheme 3). For example, the alkyl bearing substrates (**13a**–**13d**) provided the desired lactone products in 61–74% isolated yields. The alkyl chloride **12e** was transformed to the lactone **13e** in 60% yield. This substituent provides a useful synthetic handle for further product elaboration. The α -hydrogen substrate **12f** that lacks the favorable Thorpe–Ingold effect coupled with the typically interfering acidic C–H bond displayed a significantly reduced reactivity. However, we found that increasing the palladium and ligand loadings by twofold resulted in the formation of the α -hydrogen bicyclo[3.2.1] lactone **13f** in 27% yield. Substrates containing the chelating oxygen heteroatoms such as the methoxy-**12g** and the benzyloxy-**12h** substituents showed good reactivity and provided the desired products in 73% and 65% yields.

Our reaction protocol was also compatible with the incorporation of electron-neutral (**13i**), electron-donating (**13j**), and electron-withdrawing (**13k**) aryl substituents and furnished the lactone products in overall good yields (56–67%, Scheme 3). This transformation also tolerated the 2-(trifluoromethyl)pyridine and tosyl protected indole heterocycles and provided the annulated lactone products **13l** (66%) and **13m** (64%). Lastly, we evaluated the influence of oxygen and nitrogen substituents in the linear backbone and prepared the corresponding precursors **12n** and **12o**. Both of these substrates were well tolerated and generated the tetrahydropyran and piperidine bicyclo[3.2.1] lactones in 88% and 70%, respectively.

We also sought to investigate what role, if any, the geometry of alkene plays in the tandem intramolecular β -C(sp³)-H olefination and lactonization reaction (Scheme 4). The preparation of all of the substrates shown in Schemes 2 and 3 employed olefin cross metathesis and consequently furnished the depicted *E*-alkene configuration for all substrates. To arrive at the opposite *Z*-alkene isomer, we utilized the *Z*-selective Still–Gennari²⁰ olefination reaction to generate the linear precursor (*Z*)-**10c**. Exposure of the (*Z*)-**10c** to our standard reaction conditions provided the bicyclo[3.2.1] lactone **11c** in 84% yield. This result highlighted that both olefin isomers are transformed to product in comparable reaction yield, and that this transformation can be used in a setting where it may not be possible to exclusively prepare a single olefin isomer.

It was of interest to determine if this methodology is amenable to preparation of bicyclic lactones with varying numbers of carbon atoms in the central ring system (Scheme 4B). When this strategy was applied towards the synthesis of bicyclo[2.2.1] lactone, we observed

formation of the product **14** in 13% yield. We attribute the reduction in product yield to the considerable increase in steric strain about the alkylpalladium species during the intramolecular migratory insertion step. Pleasingly, we found that the addition of one extra carbon into the backbone of the annulation precursor was tolerated and the reaction provided the seven-membered bicyclo[4.2.1] lactone **15** in 81% yield. A limitation of this methodology was met when we attempted to prepare the eight-membered bicyclo[5.2.1] lactone **16**. Exposure of the corresponding starting material to our reaction conditions resulted in a recovery of the starting material and no detectable levels of the lactone **16** were observed.

To showcase the potential synthetic utility of this reaction, we applied this method to the synthesis of the 6,6,5-tricyclic lactone **24** that constitutes the core of the meroterpenoid cochlactone A (**3**, Scheme 5). Cochlactone A (**3**) was isolated as a racemate from the fungus *Ganoderma cochlear* in 2018 and showed promising anti-inflammatory activity (nitric oxide production inhibitory assay: $IC_{50} = 5.9 \pm 0.1 \mu\text{M}$).^{1d} Our short approach began with a sequential alkylation of **17** (lithium diisopropylamide, methyl iodide followed by lithium diisopropylamide, ethyl iodide) to provide the α -quaternary ester **18** (88% over two steps). Subsequent oxidative cleavage (ozone, triphenyl phosphine) of the terminal alkene provided the aldehyde **19** (94%). Robinson annulation between methyl vinyl ketone and the aldehyde **19** was catalyzed by (diethylamino)trimethylsilane to first generate the Michael adduct **20**, which was subsequently treated with potassium hydroxide and tetrabutylammonium hydroxide to provide the cyclohexanone **21** (81% yield, 1:1 diastereomers at C5).²¹ Removal of the trimethylsilyl ethyl ester (refluxing trifluoroacetic acid) generated the cyclization precursor **23** (49%) along with the C5 diastereomer **22** (49%). The pivotal tandem intramolecular β -C(sp³)-H olefination and lactonization reaction of **23** using standard conditions with two equivalents of lithium carbonate provided the 6,6,5-tricyclic lactone core **24** in 52% yield. Our effort towards this interesting tricyclic scaffold underscores the ability of this newly developed transformation to provide access to synthetically useful building blocks and, importantly, suggests that it may find use in complex natural product syntheses.

In conclusion, we have developed a palladium(II)-catalyzed tandem intramolecular β -C(sp³)-H olefination and lactonization reaction that rapidly transforms a linear free acid with a tethered olefin into the bicyclo[3.2.1] lactone scaffold. This transformation is amenable to broad structural variability and is compatible with a range of common functional groups. Co-ordinating heteroatoms such as oxygen and nitrogen on the backbone of the annulation substrates are also tolerated and provide a concise synthetic entry to novel tetrahydropyran and piperidine bicyclo[3.2.1] lactones. This reaction can be extended to the preparation of the related seven-membered bicyclo[4.2.1] lactones and proceeds with comparable efficiency using (*Z*)- and (*E*)-alkene isomers. We demonstrate the synthetic utility of this reaction by constructing the 6,6,5-tricyclic lactone core structure of the meroterpenoid cochlactone A (**3**) in seven steps. We anticipate that this newly developed synthetic strategy may find use in the preparation of other natural products and can also be used in medicinal chemistry to generate a library of potentially bioactive compounds.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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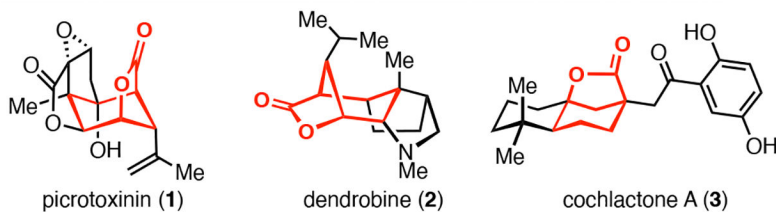
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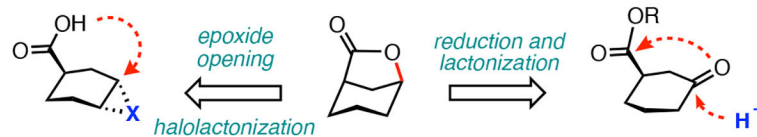
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18. We examined the enantioselectivity of the chiral thioether ligands **L9**, **L10**, and the *N*-Ac-Phe (**L1**) but in all instances the ee was <10% (see the Supporting Information).
19. We also explored the viability of a trisubstituted olefin substrate **S21** in our transformation. Exposure of **S21** to our standard reaction conditions did not provide any of the corresponding bicyclo[3.2.1] lactone product (see the Supporting Information).
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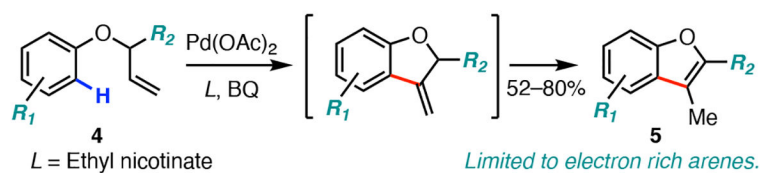
A. Examples of bioactive natural products containing bicyclo[3.2.1] lactones.



B. Conventional methodologies to synthesize bicyclo[3.2.1] lactones.



C. Intramolecular Fujiwara–Moritani olefination of arenes (Stoltz, 2004).



D. This work: Synthesis of bicyclo[3.2.1] lactones via a Pd(II)-catalyzed tandem intramolecular β -C(sp³)-H olefination and lactonization.

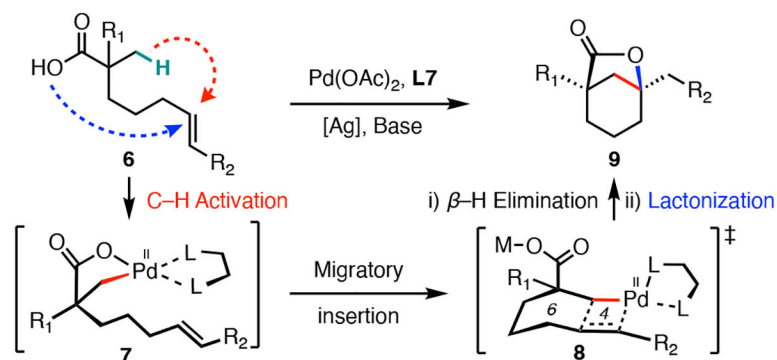
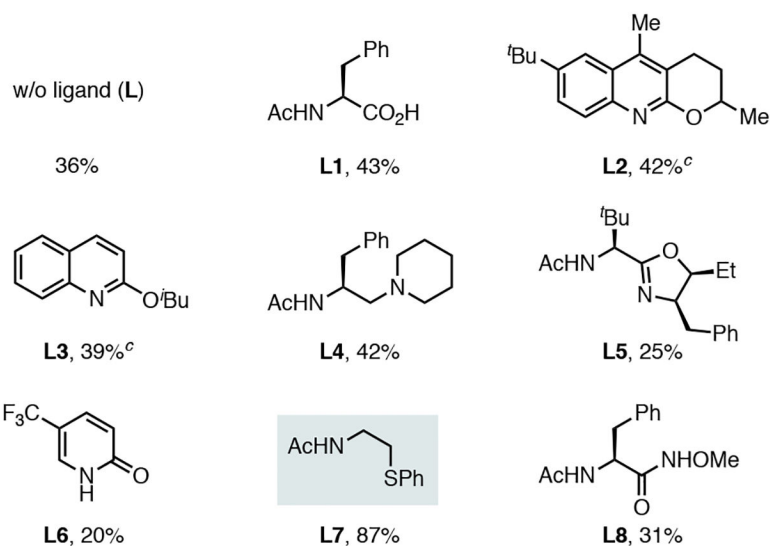
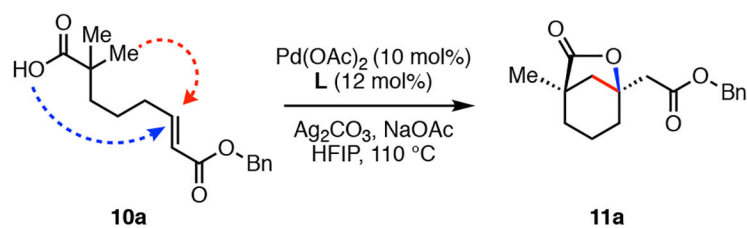
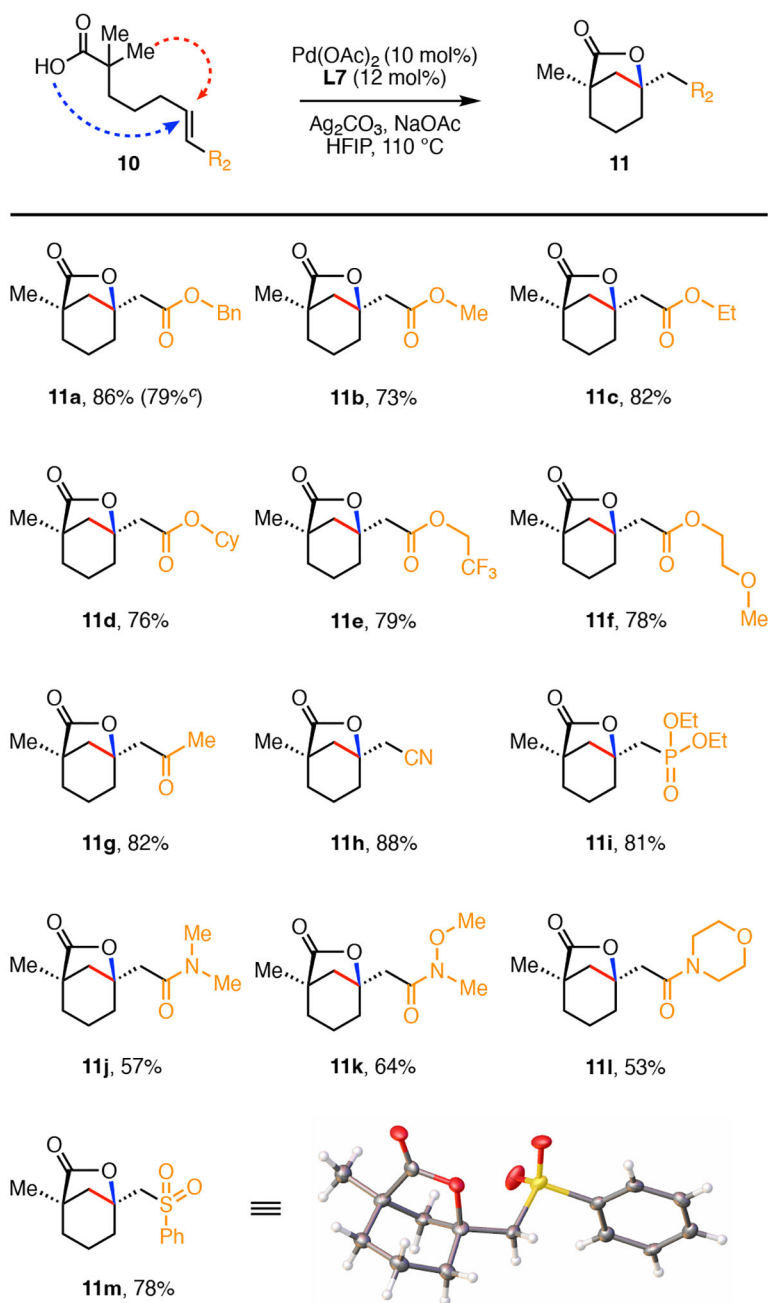


Figure 1. Pd(II)-catalyzed synthesis of bicyclo[3.2.1] lactones.

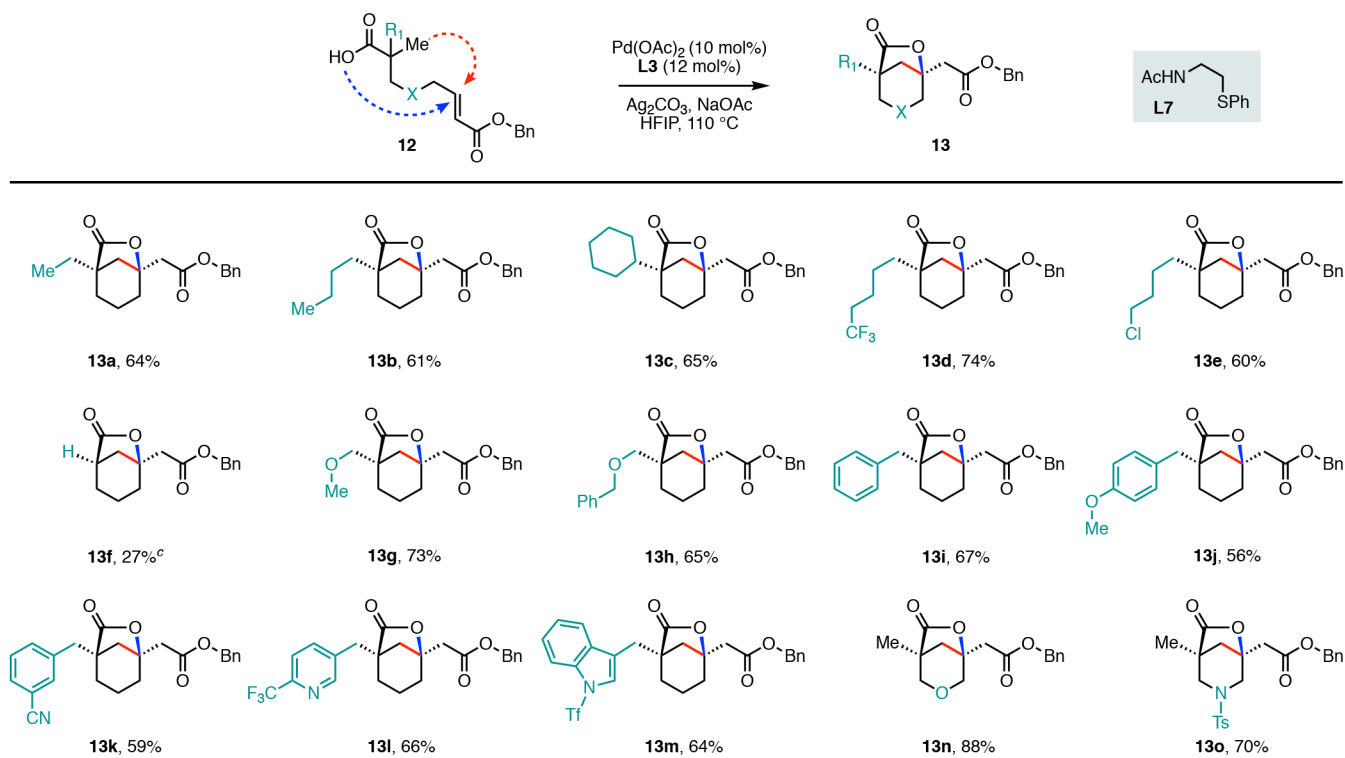
**Scheme 1.**

Ligand Investigation for the Tandem Intramolecular $\beta\text{-C}(\text{sp}^3)\text{-H}$ Olefination and Lactonization.^{a,b}

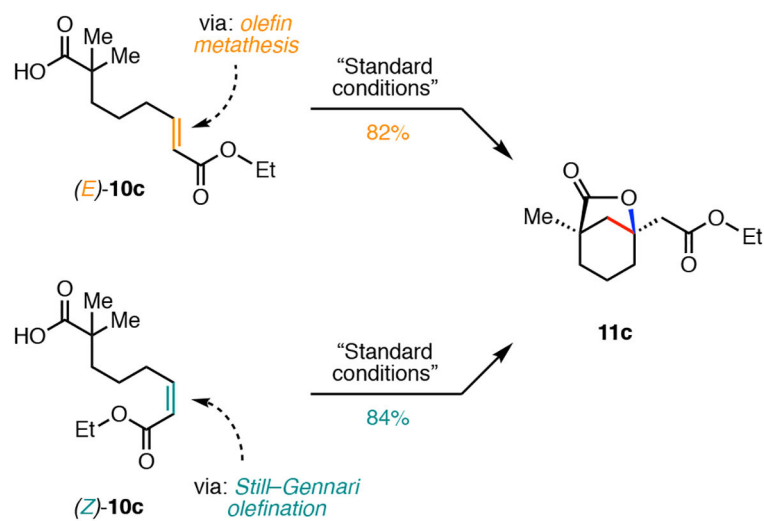
^aConditions: **10a** (0.1 mmol), Pd(OAc)₂ (10 mol%), Ligand (**L**) (12 mol%), Ag₂CO₃ (2.0 equiv), NaOAc (1.0 equiv), HFIP (1.0 mL), 110 °C, 15h. ^bYields were determined by ¹H NMR analysis using CH₂Br₂ as an internal standard. ^cLigand (**L**) (20 mol%).

**Scheme 2.**Substrate Scope of the Alkene Residue.^{a,b}

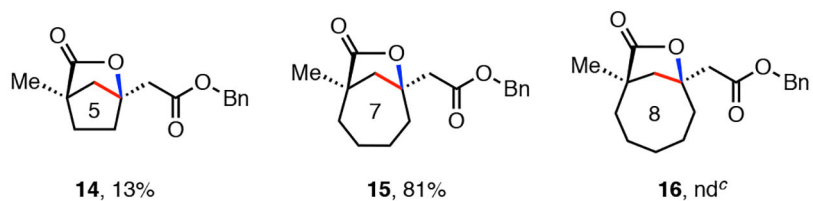
^aConditions: **10** (0.1 mmol), $\text{Pd}(\text{OAc})_2$ (10 mol%), Ligand (**L7**) (12 mol%), Ag_2CO_3 (2.0 equiv), NaOAc (1.0 equiv), HFIP (1.0 mL), 110 °C, 15h. ^bIsolated Yields. ^cIsolated yield on 1.0 mmol scale.



A. Investigation into the role of the alkene geometry.



B. Synthesis of bicyclo[2.2.1], [4.2.1], and [5.2.1] lactones.

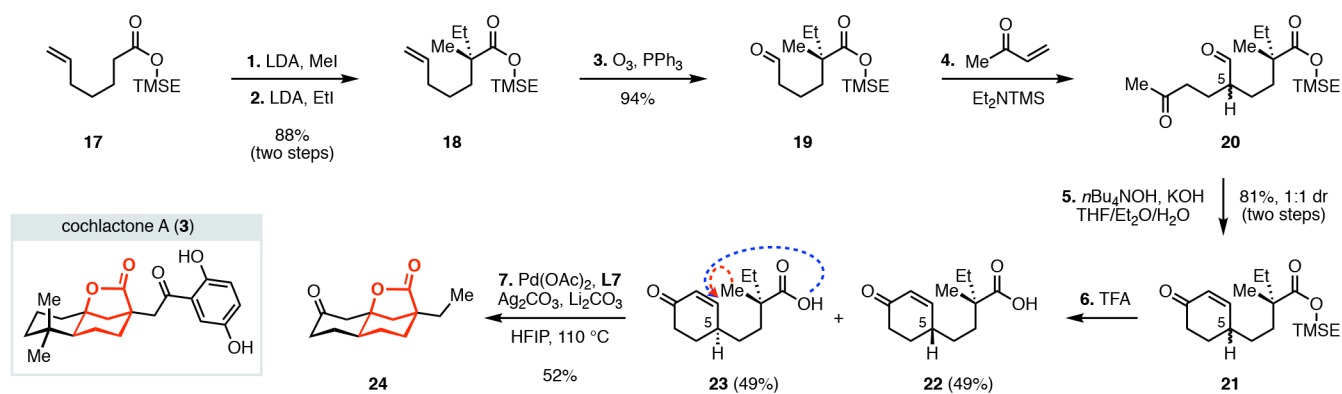


Scheme 4.

Alkene Geometry and Synthetic Application.^{a,b}

^aConditions: Carboxylic acid (0.1 mmol), Pd(OAc)₂ (10 mol%), Ligand (**L7**) (12 mol%), Ag₂CO₃ (2.0 equiv), NaOAc (1.0 equiv), HFIP (1.0 mL), 110 °C, 15h. ^bIsolated yield.

^cNone detected.

**Scheme 5.**

Synthesis of the 6,6,5-Tricyclic Lactone Core Structure of Coch lactone A (3).^{a,b}

^aReagents and conditions: (1) LDA (1.2 equiv), MeI (1.5 equiv), THF, -78→0 °C; (2) LDA (1.2 equiv), EtI (1.5 equiv), THF, -78→0 °C; (3) O₃, PPh₃ (1.5 equiv), DCM–MeOH (10:1 v/v) -78→0 °C; (4) MVK (1.5 equiv), Et₂NTMS (0.2 equiv), CH₃CN, reflux; (5) KOH (0.9 equiv), *n*Bu₄NOH (0.25 equiv, 40% aq), Et₂O–THF (3:1 v/v) 50 °C; (6) TFA (6.5 equiv), DCM, 40 °C; (7) Pd(OAc)₂ (20 mol%), Ligand (**L7**) (24 mol%), Ag₂CO₃ (2.0 equiv), Li₂CO₃ (2.0 equiv), HFIP, 110 °C, 15h. ^bTMSE = 2-(trimethylsilyl)ethyl