

Aldehyde-Associated Mutagenesis—Current State of Knowledge

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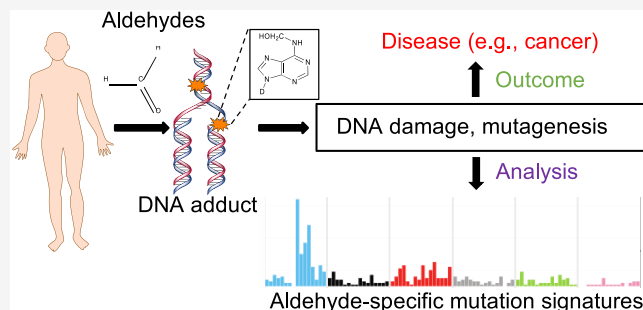
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ABSTRACT: Aldehydes are widespread in the environment, with multiple sources such as food and beverages, industrial effluents, cigarette smoke, and additives. The toxic effects of exposure to several aldehydes have been observed in numerous studies. At the molecular level, aldehydes damage DNA, cross-link DNA and proteins, lead to lipid peroxidation, and are associated with increased disease risk including cancer. People genetically predisposed to aldehyde sensitivity exhibit severe health outcomes. In various diseases such as Fanconi's anemia and Cockayne syndrome, loss of aldehyde-metabolizing pathways in conjunction with defects in DNA repair leads to widespread DNA damage. Importantly, aldehyde-associated mutagenicity is being explored in



a growing number of studies, which could offer key insights into how they potentially contribute to tumorigenesis. Here, we review the genotoxic effects of various aldehydes, focusing particularly on the DNA adducts underlying the mutagenicity of environmentally derived aldehydes. We summarize the chemical structures of the aldehydes and their predominant DNA adducts, discuss various methodologies, *in vitro* and *in vivo*, commonly used in measuring aldehyde-associated mutagenesis, and highlight some recent studies looking at aldehyde-associated mutation signatures and spectra. We conclude the Review with a discussion on the challenges and future perspectives of investigating aldehyde-associated mutagenesis.

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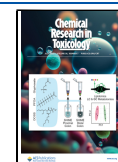
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INTRODUCTION

Aldehydes are a ubiquitous class of chemicals that are widely present in our diets, the immediate environment, as well as the intracellular milieu. Because of their high reactivity, these molecules can chemically modify all major biomolecules and impede their function. As a result, aldehyde exposure, especially from miscellaneous environmental sources, poses a

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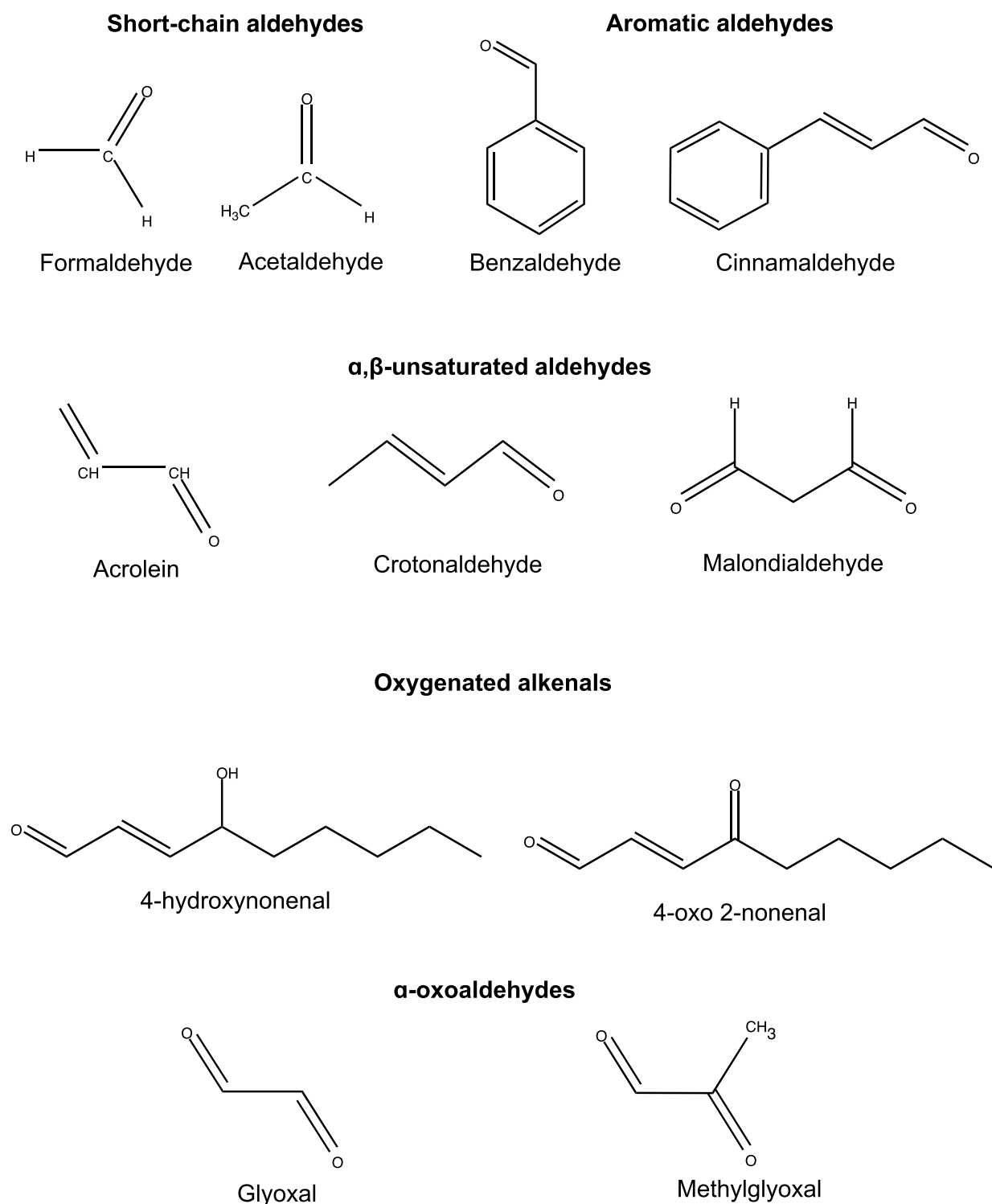


Figure 1. Chemical classification and structures of common environmental and endogenous aldehydes.

high risk to human health. Unsurprisingly, aldehyde-mediated toxicity underlies several human diseases, as noted in the upcoming sections.

Aldehydes belong to a group of chemicals referred to as reactive carbonyls. These molecules typically have a polarized carbon–oxygen (C=O) double bond, which imparts a substantial dipole moment to aldehydes. This difference in electronegativity makes the carbonyl carbon a strong electrophile and therefore readily reactive toward nucleophilic

molecules like amino groups of proteins and nucleobases of DNA. Depending on their chemical complexity, aldehydes can be classified into various chemical categories (Figure 1). These range from unbranched simple aldehydes, such as formaldehyde and acetaldehyde, to α,β -unsaturated aldehydes, such as acrolein and crotonaldehyde, and aromatic aldehydes, such as benzaldehyde, cinnamaldehyde, and vanillin.¹ Due to their short chain lengths, shorter unbranched simple aldehydes like

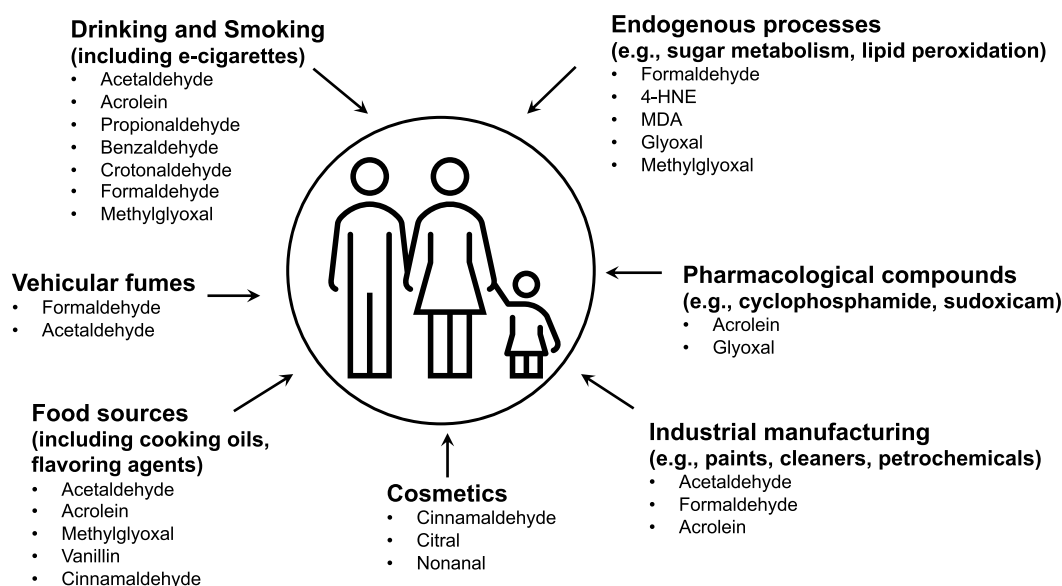


Figure 2. Common environmental and endogenous sources of aldehydes.

formaldehyde are strongly hydrophilic, which greatly enhances their toxicity.

Given the variety in chemical structures and properties of diverse aldehydes, the molecular mechanism of how these molecules impact the genome, the underlying chemical modifications, adducts, and downstream repair pathways for exposure to most aldehydes remain incompletely understood. The purpose of this Review is to highlight recent advances in understanding aldehyde-associated mutagenesis. Readers will be provided a short primer on the extant knowledge on the various classes of aldehydes, their sources, mechanisms of toxicity, and the corresponding detoxification machinery. The primary focus of the Review is to highlight studies that explored aldehyde mutagenicity and to discuss future avenues to investigate aldehyde-associated mutagenesis.

A Brief Overview of the Major Aldehydes and Their Sources. Aldehydes can occur in the environment from a multitude of natural sources (Figure 2). A variety of day-to-day human activities either use or release aldehydes into the ambient environment, including air and soil. Several excellent reviews provide an in-depth analysis of aldehyde sources, mechanism of cytotoxicity, and aldehyde clearance systems.^{1–3} A summary of the main environmental sources for well-known aldehydes is listed below.

Smoking and alcohol consumption carry the highest risk of individual exposure to aldehydes. Cigarette smoke is particularly enriched for several different types of aldehydes, chiefly acetaldehyde, acrolein, and crotonaldehyde.⁴ Cumulatively, the concentration of aldehydes in tobacco smoke is reportedly ~1000 times higher than polycyclic aromatic compounds and tobacco-specific nitrosamines.⁵ E-cigarettes also result in exposure of users to a variety of reactive carbonyl compounds including acetaldehyde, formaldehyde and acrolein, and methylglyoxal.^{6–9} These aldehydes can lead to the formation of DNA adducts; for example, tobacco smoke exposure induced increased production of the aldehyde adduct γ -hydroxypropanodeoxyguanosine (γ -OH-PrdG) in mice and humans.¹⁰

Alcoholic beverages are consumed worldwide and represent one of the largest and most ubiquitous sources of acetaldehyde

exposure. Acetaldehyde toxicity is thought to be the primary contributor to alcoholic liver disease and likely the key underlying factor in the genotoxicity of ethanol and the associated cancers of the esophagus, upper respiratory tract, gastrointestinal tract, and liver.^{11–15}

Fuel combustion is an additional major contributor to environmental aldehydes, chiefly via the photooxidative conversion of emitted hydrocarbons to aldehydes. A variety of aldehydes are released to the environment as volatile organic compounds (VOCs) in this manner, including formaldehyde, acetaldehyde, acrolein, as well as aromatic aldehydes such as benzaldehyde and tolualdehyde.^{2,16} Newer modifications to fuel sources such as the addition of ethanol have been shown to further contribute to overall environmental acetaldehyde levels.^{17,18} Aldehydes are additionally generated as intermediate chemicals during many chemical manufacturing processes.¹⁹

Diet and cooking processes release multiple complex aldehydes into the environment. Aldehyde emissions are associated with cafes and coffee-roasting facilities.²⁰ The artificial sweetener aspartame is metabolized to formaldehyde in the gastrointestinal tract,²¹ and multiple dairy products, fruits, vegetables, and meats contain formaldehyde and acetaldehyde. Food processing, especially deep-frying, can greatly increase generation of aldehydes through prolonged cooking times and high-temperature exposure of polyunsaturated fats in edible oils.^{22–24}

Many commonly available pharmacological drugs are metabolized to produce aldehydes. For example, the antineoplastic agent cyclophosphamide is metabolized by cytochrome P450 enzymes to generate acrolein, which is associated with an elevated risk of renal toxicity.²⁵ Similarly, the enol-carboxamide drug sudoxicam—the first nonsteroidal anti-inflammatory drug shown to have anti-inflammatory effects in animals—is metabolized to glyoxal and is associated with acute hepatotoxicity.²⁶

Hair treatment reagents often contain formaldehyde, resulting in high indoor concentrations in locations like hair salons.²⁷ Cinnamaldehyde is found in components of deodorants and is associated with allergic contact dermatitis.²⁸ Other

cosmetic products such as skin care treatments, colognes, nail polish removers, and fragrant hand sanitizers can be a source of toxic aldehydes like benzaldehyde and acetaldehyde, further increasing the risk of topical allergies and other skin-related diseases with prolonged contact and usage.²⁹

Lastly, several endogenous cellular pathways produce toxic aldehydes that can damage DNA. DNA metabolism can yield furfural.³⁰ Oxidative demethylation reactions involving RNA and DNA can generate formaldehyde as a byproduct, as can neutrophilic myeloperoxidase enzymes.^{31,32} The oxo-aldehyde methylglyoxal is commonly formed from triose phosphate intermediates during respiration,³³ while glyoxal is the common byproduct of sorbitol and ascorbate metabolism.² α,β -unsaturated aldehydes including acrolein and 4-hydroxynonenal (4-HNE) are commonly generated during lipid peroxidation.³⁴

Aldehyde Exposure and Human Health. Environmental aldehydes put millions of individuals at risk of aldehyde-associated health problems. Both formaldehyde and acetaldehyde are listed as Class I agents by the IARC^{35–38} and are associated with multiple cancer types including nasopharyngeal cancer, esophageal carcinoma, hepatocellular carcinoma, head-and-neck cancers, and blood cancers.^{39–45} Defects in the aldehyde dehydrogenase gene *ALDH2* are responsible for “flushed face” syndrome.⁴⁶ This phenomenon occurs in roughly 36% of all individuals of East Asian descent, is marked by facial redness, palpitations, and muscle weakness upon alcohol consumption,⁴⁷ and additionally puts such individuals at increased risk for esophageal cancer.⁴⁸

Formaldehyde is a driver of Cockayne syndrome, which is a rare autosomal recessive genetic disorder marked by cachexia, renal failure, and neurodevelopmental defects such as microencephaly.^{49,50} The transcription-coupled nucleotide excision repair factor *ERCC6* and aldehyde dehydrogenase factor *ADHS* synergize to prevent DNA damage induced by endogenous formaldehyde, and defects in both genes phenotype symptoms of Cockayne syndrome.⁵⁰ Digenic defects in *ALDH2* and *ADHS* underlie the recently described AmeD (“aplastic anemia, mental retardation, dwarfism”) syndrome and are likely linked to inefficient clearance of endogenous formaldehyde.⁵¹ Fanconi anemia, an aplastic anemia that impairs bone marrow function, is prevalent among individuals of Ashkenazi Jewish ancestry.^{52–54} Defects in formaldehyde-induced DNA damage repair,⁵⁵ as well as defects in the aldehyde dehydrogenase gene *ALDH2*,⁵⁶ predispose individuals to Fanconi anemia. Failure to efficiently clear formaldehyde in *ALDH2*-deficient mothers, combined with *FANC* deficiencies in the embryo, likely exposes fetal genomes to toxic aldehydes and produce early embryonic defects in neurodevelopment and hematopoiesis, thus providing a molecular basis for the disease.⁵⁷

Other aldehydes have been implicated in human diseases. Lipid peroxidation-associated aldehydes such as malondialdehyde (MDA) and 4-HNE can be detected as protein–aldehyde adducts in atherosclerotic lesions from human aortas,⁵⁸ and serum levels of isopentanaldehyde were increased in samples from patients of cardiovascular disease (CVD).⁵⁹ Furthermore, high plasma concentrations of 4-HNE were detected in patients of chronic kidney disease prior to hemodialysis.⁶⁰ In addition, aldehydes have been proposed as key mediators in several neuropathologies; aldehydes such as 3-aminopropanal are derived from the metabolism of polyamines (e.g., spermine), and polyamine levels are greatly elevated in

oligodendrocytes from multiple sclerosis patients.⁶¹ Additionally, aldehyde–protein adducts are elevated in amyotrophic lateral sclerosis (ALS) and Alzheimer’s disease.⁶⁰ In neurons from Parkinson’s disease patients, the metabolism of neuroendocrine factors such as dopamine and epinephrine as well as increased lipid peroxidation produces elevated levels of biogenic aldehydes such as MDA, 4-HNE, and 3,4-dihydroxyphenylacetaldehyde (DOPAL).⁶² Oxo-aldehydes like glyoxal and methylglyoxal can react with amino acid residues on proteins and precipitate the formation of advanced glycation end products (AGEs) and additionally make nucleobase adducts with guanines. Complications associated with diabetes, including vascular damage, are often attributed to decreased protein functions resulting from AGEs, likely formed due to elevated blood glucose levels that lead to a net increase in methylglyoxal levels.²

Air concentrations of formaldehyde and acetaldehyde are significantly elevated in occupational settings such as indoor nail salons, manufacturing plants for carpets, paints, furniture, and fabrics, healthcare locations, chemical laboratories, and funeral parlors.^{29,63} As such, workers in these settings are at an increased risk for respiratory and dermal problems. In combination with the prevalence of aldehydes in day-to-day emissions such as cigarette smoke, motor vehicle exhaust, and cooking oil fumes, elevated environmental aldehyde levels pose a major and constant threat to human health.

Mechanisms of Detoxification. The toxic effects of aldehydes are mitigated in large part owing to robust detoxification machinery. These consist of alcohol dehydrogenases, aldehyde dehydrogenases, aldehyde oxidases, and the cytochrome P450 reductase family of enzymes, among other proteins. Their mechanisms are briefly discussed below.

Aldehyde dehydrogenases (*ALDH*) are a class of enzymes that rely on NAD(P)-dependent oxidation of aldehydes, with varying substrate specificities. In humans, aldehyde dehydrogenases are roughly divided into three classes, *ALDH1–3*, with members typically functioning as homodimers or tetramers. Conserved residues line the catalytic pocket and participate in NAD binding, substrate alignment and deprotonation, and subsequent oxidation steps.⁶⁴ *ALDH2* deficiencies are associated with multiple disease states. In particular, the *ALDH2**2 variant, which loses the ability to bind the NAD cofactor, results in a loss of catalytic activity and accumulation of toxic levels of intracellular acetaldehyde. This results in dire health consequences including alcohol toxicity, cardiovascular complications, and cancers.^{42,43,56} *ALDH7A1* is involved in the metabolism of the amino acid lysine, and dysfunction is correlated with developmental defects and seizures, such as pyridoxine-associated epilepsy (PDE).^{65,66}

Cytochrome P450 (*CYP450*) enzymes are a class of membrane-bound enzymes present in the mitochondria and endoplasmic reticulum and are ubiquitous in the animal kingdom. These enzymes are monooxygenases and have a heme molecule as a prosthetic group that is sulfenylated via a cysteine–thiolate bond in the protein. In addition to performing multiple detoxifying reactions, including metabolism of drugs and xenobiotics and fat-soluble vitamins, *CYPs* play a key role in the metabolic activation of many compounds, steroid hormone synthesis, and breakdown of unsaturated fatty acids (reviewed in ref 67). *CYPs* catalyze the oxidation of aldehydes to their corresponding carboxylic acids using oxygen and NADPH. In addition to simple aldehydes such as acetaldehyde, *CYPs* act on a range of aldehydes including

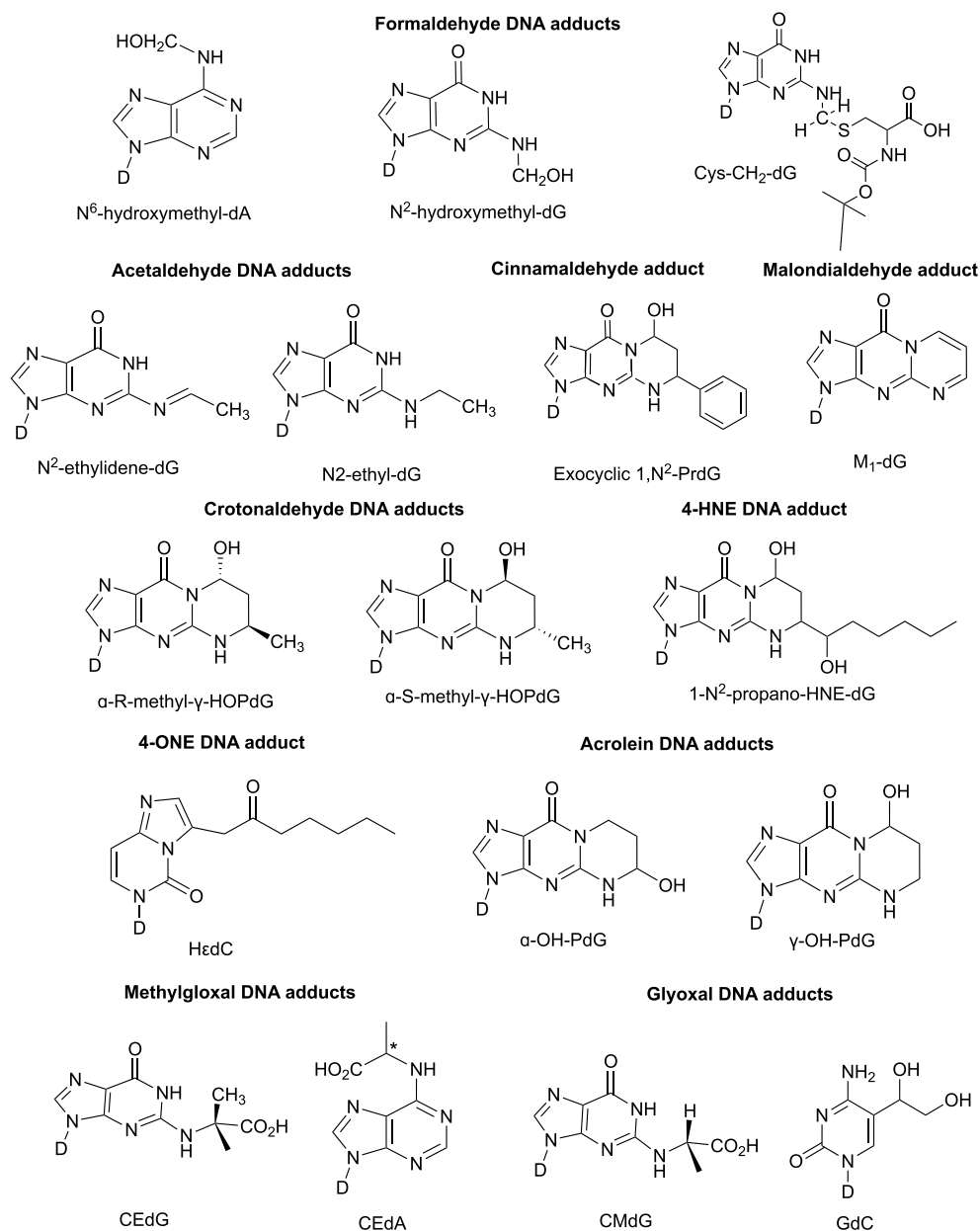


Figure 3. Chemical structures of the major DNA adducts associated with common environmental and endogenous aldehydes. All adducts are shown on deoxyribonucleosides (nucleobase-linked sugar moiety labeled D in all the above figures). dA, dG, and dC in adduct names indicate whether the nucleoside is in deoxyriboseadenosine, deoxyriboseguanosine, or deoxyribosecytidine, respectively. Refer to the section [Aldehyde Adducts and the Genome](#) for details.

acrolein, monoterpene aldehydes such as citronellal, benzaldehydes, and α,β -unsaturated aldehydes such as 4-HNE, which is derived from lipid peroxidation and is associated with chronic inflammation and neurodegeneration.^{68,69}

α -Oxoaldehydes are detoxified by the glyoxylase system that consists of glutathione-*S*-transferase (GSH)-dependent enzymes. The most-well studied members of this system are the Glo1 isomerase and Glo2 thioesterase, which are highly conserved glyoxylases that are critical for metabolizing methylglyoxal (MG) and glyoxal within tissues.²⁷⁰ Glo1 acts on glutathione conjugates of oxoaldehydes such as the MG-derived hemithioacetal and the glyoxal-derived glycolate, while Glo2 acts in the later steps of the pathway, for example, hydrolysis of the hemiacetal-derived thioester to D-lactate.^{71,72}

Moreover, monomeric aldo-keto reductases (AKR) act via general acid–base catalysis reactions that reduce aldehydes and ketones to primary and secondary alcohols. Currently three AKR families comprising 14 members are described for humans, which utilize NADP(H) as a cofactor. Endogenous aldehydes resulting from lipid peroxidation and sugar metabolism, as well as xenobiotic-derived aldehydes (e.g., aflatoxin), are among the substrates for detoxification by AKRs.⁷³

Aldehyde oxidases (AOX) are a small group of enzymes that require molybdenum, iron–sulfur clusters, and FAD as cofactors. Humans have a single AOX1 enzyme, which follows general base catalysis to oxidize aldehydes, first via the generation of activated molybdenum (mo-O^-), nucleophilic attack on the substrate by activated molybdenum (mo-O^-),

and generation of a stable intermediate, followed by release of product, reoxidation of molybdenum, and transfer of reducing equivalents via Fe–S clusters.⁷⁴ AOX enzymes are largely confined to the hepatic tissues, where they help in the metabolism of several antitumor, immunosuppressive drugs such as methotrexate, as well as antiviral compounds. Crotonaldehyde, benzaldehyde, the aromatic aldehyde vanillin, and retinal are among the common aldehyde substrates of AOX.⁷⁵

In humans, there are five classes of alcohol dehydrogenases (ADH1–V). This group of enzymes typically acts as homo- or heterodimers, has two zinc atoms in the active site, and uses NADP(H) as a cofactor. Class I ADHs are primarily expressed in the stomach and are involved in the metabolism of alcohol-derived aldehydes like acetaldehyde, whereas class II, III, and IV metabolize benzaldehyde, formaldehyde, and lipid peroxidation-derived aldehydes and are expressed in the liver and other tissues including brain.⁷⁶ ADHs can reversibly reduce aldehydes to primary alcohols, when physiological concentrations of aldehydes and reduced cofactors (NADP(H)) remain higher than alcohol and NADP⁺.⁷⁷

In addition to the mechanisms listed above, there are various nonenzymatic routes for the detoxification of aldehydes. For example, 4-HNE can be conjugated to glutathione, which links with aldehyde molecules through its sulfhydryl moiety and aids in their detoxification by aldol reductases.⁷⁸ Methylglyoxal can be efficiently scavenged by hydrazone-forming compounds such as hydrazine and its derivatives and by amino-guanidine.^{79,80} The histidine dipeptide carnosine is endogenously present across different organs including skeletal tissues, the brain, and the gastrointestinal system; it can nonenzymatically scavenge reactive oxygen species (ROS), prevent glycation, and react with several aldehyde and carbonyl compounds including methylglyoxal, lipid-peroxidation-derived malondialdehyde, and acetaldehyde.^{81–83}

■ ALDEHYDE ADDUCTS AND THE GENOME

Covalent adduct formation is a hallmark of many different aldehydes due to the presence of a highly electrophilic carbonyl which tends to readily react with strong nucleophiles like the N² nitrogen on deoxyguanosine in DNA molecules and the ϵ -amino groups on lysine residues in proteins. Conversely, unsaturated α,β -unsaturated aldehydes are less electrophilic and tend to target sulfhydryl thiolate sites that are often present on cysteine residues within proteins.⁸⁴ This becomes even more critical given that the catalytic centers of many key enzymes harbor cysteine residues, implying that adduct formation on such residues is highly detrimental to many key cellular processes. The following sections summarize specific types of DNA adducts associated with aldehyde exposure. The best-described adducts for each of the aldehydes listed below are illustrated in Figure 3.

1. Acetaldehyde Adducts. Acetaldehyde reacts with the N² group on deoxyguanosine bases to make an unstable adduct N²-ethylidene-deoxyguanosine.⁸⁵ Detection of this adduct *in vitro* requires a subsequent reduction step to convert it to a more stable N²-ethyl-deoxyguanosine. Recent studies have shown that the reduced form is also detected in human tissues in oral mucosa, urine, and leukocytes from drinkers' samples.^{86,87} Additionally, N²-ethyl-deoxyguanosine has been biochemically shown to stall elongation by the replicative polymerase Pol α but not the translesion polymerase Pol η .^{88,89} Furthermore, *aldh2*^{-/-} mice that were exposed to ethanol had

elevated levels of N²-ethylidene-dG compared to wildtype mice.⁹⁰

Interaction of acetaldehyde with an intermediate adduct in the presence of positively charged amino groups, such as those present on the basic amino acid lysine, can make DNA–DNA or DNA–protein cross-links, both of which are highly detrimental to genome stability.^{91–93} Hodkinson *et al.* elegantly showed using *Xenopus* egg extracts that when DNA is treated with acetaldehyde to produce hydroxy-PrdG adducts (see Figure 3 for example structures). The resulting interstrand cross-links are efficiently repaired by two separate repair pathways, including the Fanconi anemia pathway.⁹⁴ High-throughput sequencing showed that such adducts lead to mutagenic repair across the damaged guanine base, leading to C \rightarrow A and C \rightarrow T substitutions in a Rev1-dependent manner.⁹⁴ Besides acetaldehyde, other types of aldehydes such as crotonaldehyde and malondialdehyde can form PrdG adducts upon reacting with DNA.^{94–97} To a lesser extent acetaldehyde can also react with deoxyadenosine and deoxycytosine residues, as measured by reverse-phase HPLC analysis, to produce lower yield adducts Aa-dAdo and Aa-dCyd, respectively.⁸⁵

2. Formaldehyde Adducts. Formaldehyde is a potent cross-linking agent through its ability to form methylene bridges between the exocyclic amino groups of nucleoside bases like deoxyguanosine and the N/S groups in the side chains of amino acids like lysine, histidine, and cysteine.^{98,99} DNA–protein cross-links (DPCs) formed in this manner are relatively stable and major impediments to genome stability.^{92,100} Specialized proteases have evolved to digest DPCs and protect the genome in lower and higher eukaryotes, and their absence has been shown to make cells more susceptible to DNA damage by formaldehyde.^{101–103} In addition to cross-links, formaldehyde can make hydroxymethyl DNA adducts with N² of deoxyguanosine residues (N²-HOMEg) and with N⁶ of deoxyadenine residues (N⁶-HOMEa).^{104,105} Because hydroxymethyl adducts are unstable, their *in vitro* analysis relies upon a reduction step with sodium cyanoborohydride (NaBH₃CN) to generate stable methyl-deoxynucleosides.^{106,107} Consistent with tobacco smoke being a primary source of exogenous formaldehyde, methyl deoxyadenosine adducts are greatly enriched in DNA samples from smokers compared to nonsmokers.¹⁰⁸ Notably, mice lacking *ALDH2* and *ADHS* dehydrogenase aldehyde genes had 10-fold higher levels of blood formaldehyde and >20-fold increase in N²-MeG in tissue samples including brain, kidney, and liver compared to wildtype mice.³²

3. Methylglyoxal and Glyoxal Adducts. Methylglyoxal (MG) and glyoxal belong to a class of α -oxoaldehydes that are produced as endogenous byproducts of sugar metabolism and from other dietary sources such as meats and high-sugar snacks.^{109,110} Both molecules react with and irreversibly modify various biomolecules. In particular, the sequential modification of arginine residues on proteins by these compounds (referred to as glycation) generates advanced glycation end products (AGEs), which reduce the cellular half-life of proteins and greatly destabilize the proteome (reviewed in refs 111 and 112). As such, AGEs contribute to several pathologies including neurodegenerative diseases like Parkinson's, diabetes, aging, cardiovascular disease, and cancer. Both methylglyoxal and glyoxal can react with DNA, and the resulting reactions have been found genotoxic and/or mutagenic in human cells, reporter systems, as well as

Salmonella typhimurium.^{113,114} Yeast cells deleted for DNA damage repair genes *RAD23* and *RAD50* are sensitive to methylglyoxal.¹¹⁵ Carbonyl stress induced by methylglyoxal and glyoxal treatments was shown to cause DNA breaks and protein–DNA cross-links in human skin cells.¹¹⁶ The primary DNA adducts of MG are N²-(1-carboxyethyl)-2'-deoxyguanosine (CEdG) and N⁶-(1-carboxyethyl)-2'-deoxyadenosine and have been shown to induce genotoxicity in the form of single-strand breaks (Figure 3).^{117,118} Similarly, glyoxal reacts with dG and dC residues in DNA to make N²-carboxymethyl-deoxyguanosine (CMdG) and 5-glycolyldeoxycytidine (gdC) adducts.¹¹⁹ Further, glyoxal treatment increases G → C and G → T substitutions in reporter systems in mammalian cells.¹²⁰ Interestingly, wildtype *Escherichia coli* strains treated with methylglyoxal exhibited a higher mutation frequency and a different spectrum from *uvrC* strains that were deficient in nucleotide-excision repair; this phenomenon may be explained by the pro-mutagenic nature of NER under conditions of oxidative stress due to erroneous gap filling by DNA Pol I.¹²¹ Finally, methylglyoxal treatment inflicted greater DNA damage in cells deficient in folic acid¹²² and is able to induce chromosomal instability even at low concentrations,¹²³ which suggests that the genotoxicity of oxoaldehydes is further elevated from errors in metabolism.

4. Malondialdehyde Adducts. Malondialdehyde (MDA) has been shown to make adducts with deoxyguanosine, deoxyadenosine, and deoxycytidine molecules. Pyrimido[1,2 α]purin-10(3H)-one (M₁G) is the major adduct formed by the reaction between MDA and dG (Figure 3). Other adducts such as N⁶-(3-oxo-propenyl) deoxyadenosine (M₁A) are also formed, although less frequently.¹²⁴ Additionally, the enol moiety on MDA uniquely allows it to make oligomeric adducts, although these are not typically observed under physiological conditions.¹²⁴ MDA adducts have been shown to be mutagenic in assays in *E. coli* using a *lacZ* α containing M13 shuttle vector, whereby elevated mutagenicity correlated with increasing concentrations of MDA.¹²⁵ The predominant mutation class observed was single-base substitution, with G → T transversions as the main mutation type, along with G → A transitions and frameshift mutations. Similar studies done with genome-incorporated adducts showed M₁G-associated mutagenicity in *E. coli*.¹²⁶ Human studies have shown that M₁G is present in transplant liver samples¹²⁷ and breast tissue.¹²⁸ Finally, dietary fat intake has been correlated with increased MDA-associated adducts in humans.⁸⁶

5. Crotonaldehyde Adducts. The principal DNA adducts of crotonaldehyde are enantiomers of α -methyl- γ -hydroxypropanodeoxyguanosine (α -R-methyl- γ -HOPdG and α -S-methyl- γ -HOPdG) (Figure 3).¹²⁹ These adducts are mutagenic due to their ability to inhibit topoisomerase-I mediated DNA cleavage,¹³⁰ and by generating interstrand cross-link formation in a CG context.¹³¹ Using a single-stranded shuttle vector system containing ligated oligonucleotides with crotonaldehyde-derived adducts, it was shown that in mammalian cells both stereoisomers of the adduct were equally mutagenic, with G → T transversions being the main mutation type, as evaluated via autoradiography.¹³² However, these adducts are predominantly reported to adopt an open-ring confirmation in duplex DNA, which favors correct base pairing and could account for their low mutagenicity compared to other aldehyde adducts.¹³¹

6. 4-HNE Adducts. Peroxidation of linoleic acid generates 4-hydroxy-2-nonenal (4-HNE), which is an oxygenated alkenal

and a potent biomarker of oxidative stress.¹³³ Much like the short-chain aldehydes, reactions involving alkenals and DNA bases result in exocyclic adducts, the most common being cyclic 1,N²-dG (Figure 3).¹³⁴ Two out of the four reported stereoisomers of 4-HNE-derived adducts are more mutagenic in human cells but not strongly miscoding, resulting in G → N base substitutions (N = T, C, or A) on tandem (GG) bases.^{135,136} Urine samples from hepatitis B virus (HBV)-infected patients of chronic liver diseases had >70 fold higher concentrations of secreted ethenobase adducts such as the 4-HNE associated adduct N⁶-etheno-2'-deoxyadenosine, suggesting that robust inflammation and lipid peroxidation may contribute to aldehyde-mediated DNA damage in the cells of such patients.¹³⁷ Mouse embryonic fibroblasts deficient in the Y-family error-prone polymerase iota (Pol ι) are sensitized to 4-HNE treatment, suggesting that Pol ι is required for the efficient bypass of 4-HNE-derived genomic lesions.¹³⁸

7. 4-ONE Adducts. Linoleic acid hydroxyperoxides are broken down to oxynonenals. These aldehydes are highly electrophilic; they can undergo Michael-type addition reactions with sulfhydryl groups, such as those present on GSH, to form Schiff bases or react with nucleobases, creating etheno adducts such as heptanone-etheno-2'-deoxycytidine (HedC) (Figure 3).^{139–142} The occurrence of such adducts could be as frequent as 100 adducts/10⁸ bases, as suggested by a liquid chromatography-based adductome analysis of >60 autopsy samples from cardiopulmonary, hepatic, and gastrointestinal tissues of deceased individuals.¹⁴²

8. Benzaldehyde and Cinnamaldehyde Adducts. Despite evidence that aldehydes are respiratory irritants, flavoring agents such as benzaldehyde and cinnamaldehyde are widespread in e-cigarettes, present in >70% and >50% of all e-cigarettes, respectively.¹⁴³ Toxicological assessments show that both agents are highly cytotoxic and genotoxic.^{144,145} Intriguingly, several studies show an apparent antimutagenic activity for cinnamaldehyde in the presence of other mutagens, although it is possible that such effects are dependent on various extraneous factors such as the metabolic state of the cell, genetic background, and varying repair mechanisms.^{146–148} Although benzaldehyde adducts are poorly understood in the literature, *in silico* models predict possible adduct formation by cinnamaldehyde on N²-dG residues, creating exocyclic substituted N²-PrdG adducts (Figure 3).^{149,150}

9. Acrolein Adducts. The major DNA adducts of acrolein are α - and γ -OH-PdG (α - and γ -hydroxy-1,N²-propano-2'-deoxyguanosine (Figure 3)). The adducts are mutagenic and capable of forming both DNA interstrand cross-links as well as DNA–protein cross-links.^{151–155} Moreover, regions of the genome enriched in guanine residues, such as CpG sites, preferentially act as mutational hotspots for acrolein-association especially where such sites are methylated.¹⁵⁶

MUTAGENICITY OF ALDEHYDES

The electrophilic properties of alkanals and alkenals make them particularly strong mutagens because of the ease with which they react with nucleobases. The resulting adducts are remarkably stable and can impede genome stability in a variety of ways, including replication fork stalling, sister chromatid exchanges, interstrand cross-links, and DNA single- and double-stranded breaks.^{3,157} While such phenotypes provide a good measure of bulk genome instability, genotoxic agents often exert more subtle effects in the form of point mutations

or small insertions and deletions (InDels). Genomes exposed to toxic agents often gradually accumulate such mutations and can result in distinct mutational signatures, which serve as molecular imprints of the past or prevailing insults faced by genomes (reviewed in ref 158). A summary of studies assessing aldehyde mutagenesis is presented below. In addition, these studies are listed in Table 1, along with the proposed mechanism(s) underlying mutagenesis by the listed aldehydes.

Acetaldehyde Mutagenesis. Early studies of acetaldehyde mutagenicity came from analyzing the replication of acetaldehyde-treated plasmids carrying the *supF* tRNA reporter gene in human fibroblasts. Using both single- and double-stranded plasmids, tandem-base substitutions were found to be the predominant mutation type, especially GG → TT transversions.¹⁵⁹ Based on these studies, in liver cancers, a dinucleotide base signature was identified and attributed to acetaldehyde exposure, comprised primarily of CC → AA changes and additionally lower levels of CC → AG and CC → AT changes.¹⁶⁰ Interestingly, CC → AA mutations are the predominant double-base substitutions present in the COSMIC signature DBS2, which is widely observed in tobacco-smoking associated lung cancers.¹⁶⁰ Whether this mutation subtype is linked to acetaldehyde exposure in this signature has not been formally explored but nevertheless remains a tantalizing possibility. In another study, T lymphocytes treated with acetaldehyde reported a higher mutation frequency in the *HPRT* gene.¹⁶¹ Like the prior study, G → A transitions as well as A → T transversions were identified, with a preference for a 5'-AAG-3' or 5'-AGG-3' motif. Both the above studies are consistent with N²-Eth-dG-based mutagenesis. Interestingly, mutations in the *TP53* gene isolated from esophageal carcinoma patients display a preponderance of G → A transitions, with more mutations associated with smokers and drinkers.¹⁶² Such studies provided an early indication that acetaldehyde-induced mutagenesis might lie at the core of carcinogenesis associated with alcohol and smoking. Several other studies using similar reporters have corroborated the predominant *in vivo* mutation spectrum of acetaldehyde to be comprised of mutations in guanines (Table 1).^{163–165}

An obvious caveat to the above studies is their reliance on a single gene or reporter to measure mutations. Several studies have shown that mutation frequencies and spectra can greatly vary with several factors including cell type, replication timing, local chromatin context, and the presence of nearby DNA-bound transcription factors (reviewed in ref 166). Moreover, the dynamic interplay between DNA damage and DNA repair processes can further determine mutagenicity.^{160,167} Therefore, whole-genome analysis provides an unbiased view of the full landscape of mutations associated with a given mutagen. In an elegant study, somatic mutation loads of engrafted hematopoietic stem cells from the bone marrow of wildtype and *aldh2*^{-/-}*fancd2*^{-/-} mice were compared.¹⁶⁸ *aldh2*^{-/-}*fancd2*^{-/-} mice displayed an increased frequency of base substitutions, rearrangements, and indels and displayed evidence of stochastic DNA damage. However, low mutation loads in this study precluded accurate mutation signature analysis. Follow-up mechanistic studies revealed that cells rely on a two-step response to mitigate aldehyde-associated damage—first via an excision repair-dependent pathway that removes and repairs interstrand DNA cross-links (ICLs) and second a mutagenic translesion synthesis pathway that requires DNA replication and involves cutting within the cross-link itself to unhook the ICL.⁹⁴ Using *Xenopus* egg extracts, researchers

monitored the replication of plasmids carrying acetaldehyde-derived cross-links and showed that both the above repair pathways were mutagenic. Unlike prior reports showing G → A transitions, the *Xenopus* experiment G → T transversions were the most common base substitutions. In agreement with the role of excision repair in mutagenesis, ablation of the FA pathway via p97 inhibition removed this mutagenic signature, while depletion of the TLS factor REV1 inhibited bypass of the ICL lesion in a strand-specific fashion.⁹⁴ On the other hand, acetaldehyde treatment elicited a DNA damage response without any associated mutagenesis *in vivo* with human-induced pluripotent stem cells (iPSCs) (Table 1).¹⁶⁹

Recent work from our group demonstrated that acetaldehyde is highly mutagenic; however, its primary genomic substrate is single-stranded DNA (ssDNA).¹⁷⁰ In line with earlier studies, mutagenesis was found to be TLS-dependent and gave rise to a preponderance of C → A (G → T) base substitutions. Further, we observed that the ssDNA-associated mutations were present in a gCn (nGc) motif, which revealed a ssDNA-specific gCn → A (nGc → T) mutation signature for acetaldehyde (Table 1).¹⁷⁰ Even more remarkably, mutation loads in this genomic context were enriched in whole-genome- and whole-exome-sequenced cancers associated with smoking/alcohol, especially liver cancers. Interestingly, the enriched mutations observed in cancers strongly associated with the nontranscribed strand of genes, which suggests that ssDNA formed during increased transcription in cancers likely acts as an ideal substrate for acetaldehyde-induced mutagenesis.¹⁷⁰ A related study demonstrated the mutagenesis of ssDNA in response to acetaldehyde, albeit with different base substitutions, i.e., more C → T and T → A changes; interestingly, acetaldehyde treatment led to an increase in the proportion of ssDNA-associated deletions of ≥5 bp but without any associated microhomology at break points.¹⁷¹

Formaldehyde Mutagenesis. In budding yeast, formaldehyde-mutated *CAN1* mutants and *lys2* frameshift revertants were sequenced and found to contain frameshifts consisting of NER-dependent large deletions as well as complex insertions in hotspots of the *LYS2* gene. Comparison of mutational spectra from strains lacking *REV3*, *RAD30*, or *RAD14* showed that these complex mutations are dependent on NER as well as mutagenic bypass via Pol ζ-mediated translesion synthesis.¹⁷² A recent study demonstrated that formaldehyde, like acetaldehyde, was mutagenic to ssDNA in yeast. Formaldehyde treatment generated C → T and T → A transversions and revealed a mutational signature akin to COSMIC SBS40, which is a remarkably common signature among most cancer types. However, unlike acetaldehyde, no appreciable increase in indels was observed with formaldehyde exposure (Table 1).¹⁷¹

Beside yeast, mice serve as excellent models for aldehyde-associated human diseases such as Fanconi anemia and Cockayne syndrome, owing to the conservation of hematopoietic cell lineages and physiological aldehyde response systems between rodents and humans.^{32,94,168,173,174} In mice lacking *CSB5* (*ERCC6*), an excision repair gene, and *ADHS*, which is involved in endogenous formaldehyde clearance, formaldehyde-associated adducts (N²-methyl-dG) are elevated in kidneys and brain, and cells display transcriptional stress.⁵⁰ In response to methanol treatment, *adh5*^{-/-} mice accumulated DNA damage, including increased frequency of SCE events.³² Further, γH2AX levels are elevated in hematopoietic stem cells from *adh5*^{-/-}*fancd2*^{-/-} mice, indicating DNA damage, with

Table 1. Studies Exploring Aldehyde-Associated Mutagenesis^a

agent	system	chemical agent	reporter	mutagenicity	sequencing	signature	proposed mechanism	ref
acetaldehyde (AA)	SV-40 transformed fibroblasts	direct AA treatment of reporter plasmid	<i>supF</i> tRNA gene	yes	Sanger	GG → TT (CC → AA)	ICL formation impaired NER	159
	peripheral T lymphocytes	AA	<i>HPRT</i>	yes	Sanger	G → A	transcription-associated NER	161
	HEK 293 cells	N ² ,O ⁶ -dG adduct-containing oligos	<i>supF</i> tRNA gene	yes	Sanger	G:C → A:T	error-prone TLS, postreplicative MMR	163
	human XPA cells	site-specific plasmid-borne adduct	<i>bsd</i> resistance (survival)	yes	Sanger	G → T	ICL formation, TLS	164
	human fibroblasts	AA	<i>TP53</i>	yes	Sanger	G → A, G → T	ICL formation, TLS	165
	mice	AA	HSCs	yes	WGS		<i>aldh2</i> ^{-/-} <i>jancd2</i> ^{-/-} cells had more base substitutions, microhomology (MH)-mediated deletions, rearrangements, likely stochastic damage	168
	human iPSC	AA	<i>CANI</i>	no	WGS	increased C → T with EHOH	DNA damage response induction without mutagenesis	169
	<i>Saccharomyces cerevisiae</i>	AA, EHOH	<i>CANI</i>	none for AA	Sanger	G → T	EHOH-induced replication stress, followed by error-prone DNA repair but no AA-specific mutations	197
	<i>Xenopus</i> egg extracts	AA-ICL adduct plasmid replication		yes	high-throughput sequencing		Rev1, pol ζ-mediated TLS, ICL repair via FA pathway	94
formaldehyde (FA)	<i>S. cerevisiae</i>	AA	<i>CANI ADE2</i>	yes	WGS	gCn → gAn (nGc → nTc), ssDNA-specific	TLS mediated by pol ζ, ssDNA-specific signature also observed in alcohol-associated cancers	170
	<i>S. cerevisiae</i>	AA	<i>CANI</i>	yes	WGS	C:G → T:A, T:A → C:G, small deletions	likely TLS	171
	<i>S. cerevisiae</i>	FA	<i>CANI</i> , <i>lys2ΔA746</i> , NR	yes	Sanger	large deletions in direct repeats, complex insertions	lesion bypass by NER, Pol ζ-mediated TLS	172
	mice	methanol	endogenous N ² -hydroxymethyl-dG	DNA damage		increased γH2AX, p53 induction	FA cross-link repair pathway and <i>ADH5</i> mediated protection	99
	human iPSC	FA		no	WGS		no effect on DDR pathway or mutagenesis	169
	mice	FA		yes	WGS (HPSCs)	C → T, T → A	SBS signatures 3, 5, 25, 40 observed. age-associated damage, FA pathway defect	32
	FA SCCs	likely elevated aldehyde levels from smoking, alcohol		yes	PacBio, WGS	multiple COSMIC SBS and ID signatures	CIN is high in FA cells, likely drives mutagenesis, complex SV formation	186
	<i>S. cerevisiae</i>	FA	<i>CANI</i>	yes	WGS	C:G → T:A, T:A → C:G, no indels	likely TLS	171
acrolein (Acr)	<i>S. typhimurium</i>	Acr	S9 fraction enzyme activation	yes				175
	COS-7 cells	pMS2 shuttle vector with adduct (γ-HO-PdG)		yes	Sanger	G → T, G → A, G → C	lesion bypass by error-prone polymerases, DNA:DNA, DNA: protein cross-links	176
	NHLF cells	Acr	<i>supF</i> , <i>p53</i>	yes	Sanger	G → T, G → A, G → C	Acr in cigarette smoke can mutate cancer drivers, NER-mediated bulky lesion repair	153
	mouse embryonic fibroblasts, human XPA fibroblasts	Acr	<i>dI</i> transgene, <i>supF</i>	no			dose-dependent, chromosome context-dependent mutagenicity of Acr, error-free lesion bypass	198
	human CCL-202 lung fibroblasts	Acr	<i>supF</i>	yes	Sanger	G → T, G → A, G → C	Acr mutations scale proportionately with Acr-DNA adducts, NER-dependent repair.	177
oxoaldehydes	human iPSC	Acr		no	WGS		no effect on DDR pathway or mutagenesis	169
	<i>E. coli</i>	methylglyoxal	<i>lacI</i>	yes	Sanger	G:C → C:G, G:C → T:A	NER-dependent repair	181
	COS-7	methylglyoxal	<i>supF</i>	yes	Sanger	G:C → C:G, G:C → T:A	NER-dependent repair	114
	human fibroblasts (XP+ and XP ⁻)	CEdG-adducted shuttle vector	<i>supF</i>	yes	Sanger	A:T → G:C	NER-dependent adduct removal and DNA repair	182
	COS-7	glyoxal	<i>supF</i>	yes	Sanger	G:C → C:G, G:C → T:A	NER-dependent repair	120

Table 1. continued

agent	system	chemical agent	reporter	mutagenicity	sequencing	signature	proposed mechanism	ref
crotonaldehyde	COS-7 cells	N ² ,O ⁶ -dG adduct-containing vector	GFP vector (transfection efficiency)	yes	probe hybridization	G → T	replication blockage, NER	132
malondialdehyde (MDA)	human XPA-cells	shuttle vector with site-specific ICL	blasticidin S	yes	probe hybridization	G → T	NER-independent repair at replication forks, TLS	199
	<i>E. coli</i>	ssDNA M13 vector replication	<i>lacZ</i> alpha forward mutation	yes	Sanger	G → T, A → G, C → T		125
	human fibroblasts	M ₁ G adduct in ssDNA vector	MDA-treated pSP189 shuttle vector	yes	probe hybridization	G → A, G → T	NER, adduct blocks replication fork passage	126
			<i>supF</i> reporter mutations	yes	Sanger	small indels, GC → AT, GC → TA, GC → CG	NER	200

^aRefer to main text for details of reporter systems and additional references.

animals displaying sensitivity to exogenous methanol. *aldh2*^{-/-}*adh5*^{-/-} mice display sensitivity to acetaldehyde and formaldehyde and confer a variety of deficiencies including lymphoid cancers. DNA from several organs of *aldh2*^{-/-}*adh5*^{-/-} mice also showed increased N²-meth-dG adduction, with bone marrow from these animals displaying a >two-fold increase in both single-base substitutions and indels. Whole-genome sequencing of hematopoietic stem and progenitor cells (HSPC) from bone marrow of *aldh2*^{-/-}*adh5*^{-/-} mice show increased formaldehyde-associated base substitutions, with a predominant T → A and T → C bias, and a preference for adenines on the transcribed strand.³² *fancd2*^{-/-}*aldh2*^{-/-} mice have similarly been utilized to model the pathogenesis of fetal alcohol syndrome, which is related to maternal alcohol consumption.¹⁷³

Acrolein Mutagenesis. Sublethal doses of acrolein were shown to be mutagenic in *S. typhimurium* assays.¹⁷⁵ Subsequent work with human cell lines similarly noted acrolein-associated mutagenesis in reporter genes. Plasmids incorporated with acrolein-derived γ HOPdG adducts permitted relatively efficient elongation by replicative polymerases in human COS-7 cells, albeit with a small percentage of predominantly G → T base substitutions.¹⁷⁶ Other studies examined the impact of acrolein-derived adducts on *TP53* mutations in lung fibroblasts and observed that the site of adduct formation within *TP53* correlated with mutational hotspots that are commonly seen with smoking-associated lung cancer.¹⁵³ *supF* reporter analyses show that G → T transversions are greatly enhanced in acrolein-treated samples (Table 1).^{153,177} However, acrolein was not found to be mutagenic in a whole-genome analysis of treated human iPSC cells nor did it elicit a DNA damage response.¹⁶⁹ Therefore, the molecular mechanism(s) underlying acrolein-induced mutagenicity and any associated mutation signatures are still relatively unclear and warrant further investigation.

Mutagenicity of Oxaldehydes. Endogenous metabolic processes such as lipid peroxidation and glycolysis can produce oxaldehydes such as methylglyoxal and glyoxal, which can rapidly cross-react with amino acid side chains and give rise to advanced glycation end products (AGEs). In addition to proteins, studies have shown oxaldehydes to react with nucleobases, chiefly guanosine and produce adducts.^{178–180} Even so, the biological relevance and mutagenicity of such adducts is an open question. Initial studies in *E. coli* and human COS-7 cells showed that methylglyoxal can induce G → C and G → T mutations and 20–300 bp deletions.^{114,181} Reporter gene studies with glyoxal yielded similar results.¹²⁰ In subsequent studies, the primary methylglyoxal adduct N²-(1-carboxyethyl)-2'-deoxyguanosine (CEDG) led to increased mutation frequencies, with G:C → T:A transversions being the major mutation type.¹⁸² Interestingly, although repair-deficient cells had a higher mutation frequency, they nevertheless had an unchanged mutation spectrum compared to repair-proficient cells. A role for the Y-family error-prone polymerase Pol iota (*pol i*) in the gap-filling repair step of NER has been suggested as a possible mechanism for the observed mutation spectrum.^{182,183} *In vitro* studies also suggest frequent purine incorporation opposite CEDG adducts.¹⁸⁴ Treatment of keratinocytes and fibroblasts suggests that distinct DNA damage can accumulate in response to glyoxal and methylglyoxal. While the former was seen to induce a higher frequency of double-strand breaks, the latter caused a higher incidence of DNA–protein cross-links.¹¹⁶ Glyoxal and

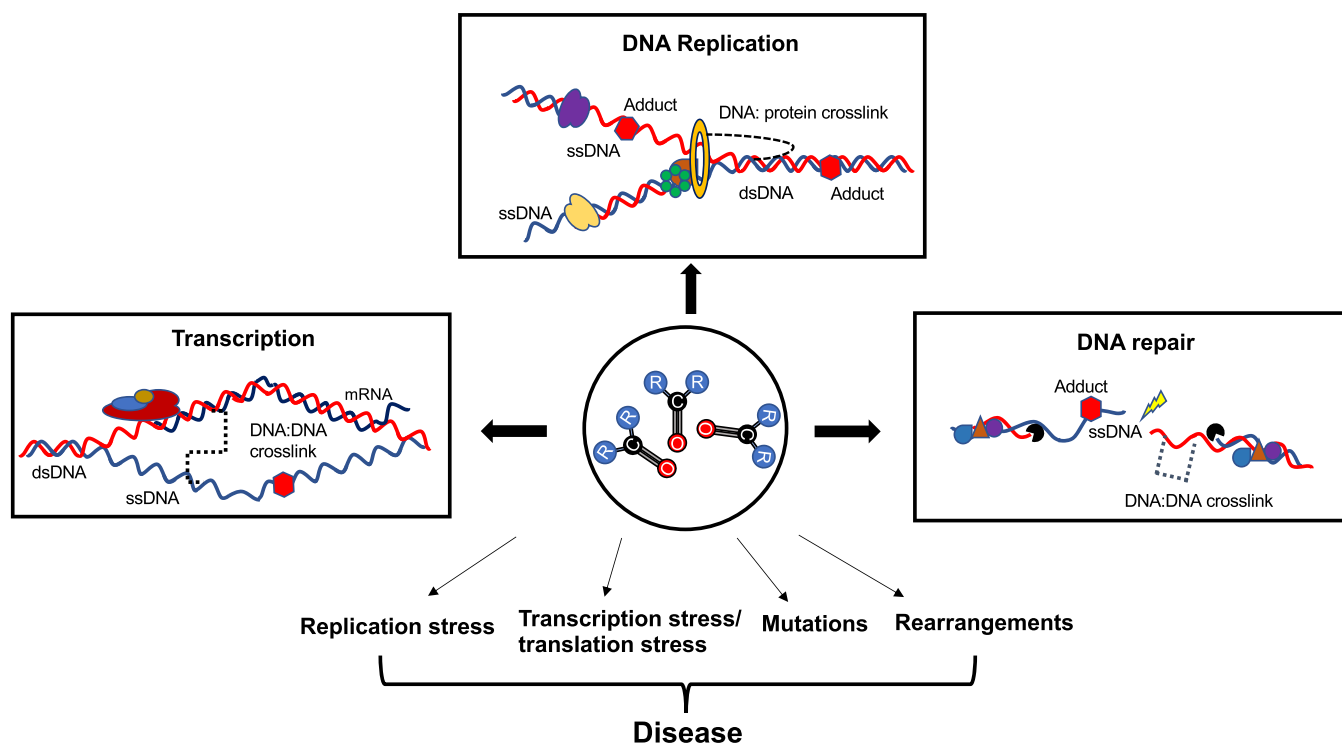


Figure 4. Mechanisms of aldehyde-associated genome instability. Major genome-associated pathways are illustrated in open boxes. All the listed processes involve unwinding of the double helix, leading to the generation of single-stranded DNA (ssDNA). In the presence of reactive carbonyls (ball and stick molecules), open circle, genomes can accumulate a variety of lesions on both ssDNA as well as double-stranded DNA (dsDNA), including DNA:protein and DNA:DNA cross-links (broken connectors) and adducts (red hexagons). Failure to repair such lesions or erroneous bypass can result in severe genome instability, which can contribute to aldehyde-related diseases.

methylglyoxal have also been shown to make imidazopurinone derivatives with deoxyguanosine bases, which have been shown to increase the frequency of double-strand breaks in HL60 cells treated with methylglyoxal¹⁸⁵ and might contribute to differential oxoaldehyde-associated mutagenesis.

Similar reporter-based mutagenesis studies have been performed for other aldehydes and are summarized in Table 1.

Other Mutational Events Associated with Aldehyde Exposure. Finally, recent work explored the mutational landscape of squamous cell carcinoma patients with defects in the Fanconi anemia pathway, which is characterized at the molecular level by increased DNA interstrand cross-links (ICLs), a common occurrence in response to toxic aldehydes. The major mutational event in these cells was chromosome instability in the form of large structural variants in the range of 1–100 kb and MMEJ-mediated rearrangements. Such events are thought to contribute to overall oncogenic activation and provide insights into how genomes exposed to toxic aldehydes through smoking, drinking, or other environmental pollutants undergo mutagenesis and accumulate genome instability.¹⁸⁶

A General Model for Aldehyde-Associated Genome Instability and Mutagenesis. Historically, the lesions most associated with aldehyde exposure are DNA:DNA cross-links (ICLs) as well as DNA:protein cross-links.^{92,94,102,154,159,187} Unless resolved, such lesions can destabilize the genome in multiple ways, including stalling replication forks, blocking transcription, and initiating double-strand breaks. We showed that in yeast single-stranded DNA, while acetaldehyde exposure resulted in an uptick in G → T mutations, there was no substantial enrichment for GG → TT double-base substitutions, which is the preferred genomic context for

acetaldehyde-induced intrastrand cross-links.¹⁵⁹ This implies that additional mechanisms could drive aldehyde-associated mutagenesis.

Surveys of aldehyde-associated mutagenesis listed so far seem to share certain commonalities. The overall mutation spectra look remarkably similar, with guanines being the preferred base for adduction, and NER appears to be the preferred pathway for minimizing aldehyde-associated DNA damage (Table 1). Although such observations are not exclusive to aldehydes, they nevertheless hint at a potential mechanistic link through which different aldehydes target the genome and stimulate mutagenesis.

One such link could be mutagenesis of single-stranded DNA. The genome is transiently single-stranded during several processes such as DNA replication, transcription, and DNA repair (Figure 4). ssDNA is highly mutagenizable due to the lack of any associated repair pathways. Work from our lab and others has shown that ssDNA is mutagenized by both formaldehyde and acetaldehyde in yeast, which raises the possibility of other aldehydes following a similar mechanism. If ssDNA is in fact the preferred genomic substrate for aldehydes, perhaps that would explain why most prior research has yielded mixed results on classifying aldehydes as mutagenic *in vivo*. Most studies using cell lines or mouse models would not have sufficient ssDNA formed as the substrate for aldehyde mutagenesis, leading them to conclude somewhat erroneously that the given aldehyde is not mutagenic. When genomes are repeatedly exposed to toxic aldehydes, the presence of vulnerable substrates (e.g., ssDNA) would enhance aldehyde genotoxicity, generating several types of lesions (such as cross-links) and bulky adducts. Erroneous processing of such lesions

(e.g., in repair-deficient genetic backgrounds) would result in marked genome instability in the form of replication/transcriptional stress, accumulation of mutations, and widespread DNA damage, which would ultimately contribute to diseases such as carcinogenesis. (Figure 4). In fact, most cancers are marked by replication stress, replication–transcription collisions and R-loop formation, and hypertranscription, all events that generate copious ssDNA.^{188,189} As such, mutations in ssDNA regions could represent a genomic record of aldehyde exposures, such as what we and others have demonstrated.^{170,171}

CONCLUSIONS AND FUTURE PERSPECTIVES

Decades of elegant research have led to the discovery of the main adducts of several environmental and endogenous aldehydes and shaped the field of aldehyde mutagenesis. However, several challenges remain. First, a direct cause-and-effect relationship between aldehyde-associated mutagenesis and the offending adduct is not straightforward. Mutagenesis could result from the processing of either DNA adducts or a combination of DNA, RNA, and protein adducts. Second, depending on the cell type, local genomic context, DNA accessibility, and variable DNA repair efficiencies, aldehydes could be processed to one or several adducts *in vivo*. Third, different adducts might have varying degrees of stability *in vivo*, as well as strand preferences. Consequently, it is possible to have varying degrees of mutagenesis even with the same aldehyde. Discrepancies between *in vivo* and *in vitro* adduct chemistries could partly explain why reports of mutagenesis for several aldehydes drastically vary between different studies (Table 1).

Several unresolved questions remain that would help broaden our understanding of aldehyde-associated mutagenesis. For instance, is single-stranded DNA the preferred *in vivo* genomic substrate for other aldehydes? What are the determinants of the genomic distribution of aldehyde-associated DNA adducts? Are there other specialized repair pathways that are dedicated to specific aldehydes, much like acetaldehyde-associated FANCDJ? Depending on the cell or tissue type, metabolic state, and genetic background, do aldehydes have multiple mutation signatures? Do aldehydes broadly damage the genome, or are there damage and mutational hotspots? On a related note, can localized aldehyde-associated damage generate clustered mutations (kategias) that are often a hallmark of many different types of cancers?

Exploring the above themes can help determine if certain genomic contexts are “at-risk” of aldehyde-associated damage and allow us to identify which genomic lesions are mutagenic.

A combination of cutting-edge adduct-mapping techniques and high-throughput sequencing now provides researchers means to extensively survey the genotoxic impacts of aldehyde exposure. Precision methods, such as LC-NSI-HRMS-based mass spectrometry, have been successfully used for *in vivo* mapping of acrolein-derived adducts.¹⁹⁰ Similar applications would enable the generation of high-resolution adductome maps, predict which chemicals acted as precursors to the adducts, and can eventually help trace back to the source environmental chemicals that are responsible for genotoxicity.^{187,191} Similarly, newer sequencing technologies can help define a much broader mutational landscape for a given aldehyde. Newer versions of single-molecule real-time (SMRT) sequencing improve significantly upon their prede-

cessors to provide single-base resolution of DNA damage with multiple types of modified bases such as alkylated purines (O⁶-methyl guanine, 6-methyladenine), hydroxylated bases (e.g., 5-hydroxycytosine), and UV-damaged cyclobutene dimers and can be potentially leveraged to study aldehyde-derived adducts.^{192,193} Similarly, PCR-free amplification methods like nanopore sequencing (NPS) are greatly suited to adduct mapping. NPS has been used for mapping a wide range of adducts associated with many common exogenous genotoxins.^{194–196} Finally, elegant computational pipelines, including NMF- and/or knowledge-based signature analyses, and machine learning/AI can refine the key genomic coordinates that drive mutagenesis and lead to development of prognostic genomic biomarkers for aldehyde exposure, thus greatly informing on therapeutics and leading to better health outcomes.

Aldehyde exposure is intricately tied to human health. From a clinical standpoint, it is imperative to explore research avenues that clearly outline the genomic risk factors for aldehyde-associated toxicity to formulate better strategies for health risk mitigation. Although significant advances have been made toward determining aldehyde genotoxicity, these only represent the tip of the iceberg. As such, several additional avenues should be explored to comprehend the full spectrum of aldehyde-associated genome instability. Such efforts are vital for providing well-defined, publicly available data sets that link distinct aldehydes to specific mutational patterns and eventually diseases.

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Notes

The authors declare no competing financial interest.

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Sriram Vijayraghavan: Dr. Vijayraghavan received his Ph.D. from the University of Pittsburgh, where his research focused on the interplay of DNA replication and repair using yeast as a model system. He did his postdoctoral research at the Duke University Medical Center, North Carolina, studying mitochondrial DNA variation and RNA viruses in natural yeast populations, using a combination of genomics and molecular biology. He is currently a staff scientist in Dr. Natalie

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Natalie Saini: Dr. Saini received her Ph.D. from the Georgia Institute of Technology, where she studied the impact of fragile DNA repeats on genome stability. She proceeded to do her postdoctoral work at the National Institute of Environmental Health Sciences, North Carolina, where her research focused on how the accumulation of somatic mutations impacts normal cells. She obtained a K99/R00 Pathway to Independence grant and transitioned to her current position as an Assistant Professor in the Department of Biochemistry at the Medical University of South Carolina, where she continues to study the impact of environmental and endogenous DNA damage on genome stability.

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ABBREVIATIONS

4-HNE 4-hydroxynonenal
 4-ONE 4-oxynonenal
 γ -HOPdG γ -hydroxy-1,N-2-propano-2'-deoxyguanosine
 β -HOPdG β -hydroxy-1,N-2-propano-2'-deoxyguanosine
 AA acetaldehyde
 AI artificial intelligence
 Acr acrolein
 AGE advanced glycation end product
 ALS amyotrophic lateral sclerosis
 CE dG N2-(1-carboxyethyl)-2'-deoxyguanosine
 CM dG N2-carboxymethyl-deoxyguanosine
 COSMIC Catalog of Somatic Mutations in Cancers
 CPD cyclobutane pyrimidine dimers
 CRC colorectal cancer
 CVD cardiovascular disease
 CYP450 cytochrome P450
 DNA deoxyribonucleic acid
 DOPAL 3,4-dihydroxyphenylacetaldehyde
 DPC DNA-protein cross-link
 EMS ethylmethanesulfonate
 FA Fanconi anemia
 FAD flavin adenine dinucleotide
 gdC 5-glycolyldeoxycytidine
 GSH glutathione S-transferase
 H2AX gamma H2AX
 HBV hepatitis B virus
 H ϵ dC heptanone-etheno-2'-deoxycytidine
 HPLC high-performance liquid chromatography
 HPRT hypoxanthine-guanine phosphoribosyltransferase
 LC-NSI-HRMS/MS liquid chromatography-nanoelectrospray ionization-high-resolution tandem mass spectrometry
 HSPC hematopoietic stem and progenitor cells
 IARC International Agency for Research on Cancer
 ICL inter/intrastrand cross-link
 InDels insertions and deletions
 ISC intestinal stem cells
 LC liquid chromatography
 M₁A N⁶-(3-oxo-propenyl) deoxyadenosine
 M₁G pyrimido[1,2 α]purin-10(3H)-one
 MG methylglyoxal
 MDA malondialdehyde

MMEJ microhomology-mediated end joining
 MMS methylmethanesulfonate
 MS mass spectrometry
 N²-Eth-dG N²-ethylidenedeoxyguanosine
 N²-MeG N²-hydroxymethyl-deoxyguanosine
 N⁶-MeA N⁶-hydroxymethyl-deoxyadenosine
 NAD nicotinamide adenine dinucleotide phosphate
 NER nucleotide excision repair
 NMF non-negative matrix factorization
 NMR nuclear magnetic resonance
 NPS nanopore sequencing
 PCR polymerase chain reaction
 PDE pyridoxine-associated epilepsy
 PrdG 1,N²-propanodeoxyguanosine
 iPSC induced pluripotent stem cells
 RNA ribonucleic acid
 ROS reactive oxygen species
 SBS single-base substitution
 SCE sister chromatid exchange
 SMRT single-molecule real time
 SNP single-nucleotide polymorphism
 SNV single-nucleotide variant
 ssDNA single-stranded DNA
 TLC thin layer chromatography
 TLS translesion synthesis
 VOC volatile organic compounds

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