Hindawi Advances in Pharmacological and Pharmaceutical Sciences Volume 2023, Article ID 5593350, 21 pages https://doi.org/10.1155/2023/5593350

# Research Article

# Combination of Chemically-Characterized Essential Oils from Eucalyptus polybractea, Ormenis mixta, and Lavandula burnatii: Optimization of a New Complete Antibacterial Formulation Using Simplex-Centroid Mixture Design

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Received 20 June 2023; Revised 3 August 2023; Accepted 9 August 2023; Published 21 August 2023

Academic Editor: Kuldeep Singh

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This study aims to identify the volatile profile of three essential oils obtained from Eucalyptus polybractea cryptonifera (EPEO), Ormenis mixta (OMEO), and Lavandula burnatii briquet (LBEO) and to examine their combined antibacterial activity that affords the optimal inhibitory ability against S. aureus and E. coli using simplex-centroid mixture design and checkerboard assay. Essential oils (EOs) were isolated by hydrodistillation and characterized using gas chromatography-mass spectrometry (GC-MS) and gas chromatography coupled with flame-ionization detector (GC-FID). The antibacterial activity was performed using disc diffusion and microdilution assays. The chemical analysis revealed that 1,8-cineole (23.75%), p-cymene (22.47%), and  $\alpha$ -pinene (11.20%) and p-menthane-1,8-diol (18.19%), α-pinene (10.81%), and D-germacrene (9.17%) were the main components detected in E. polybractea and O. mixta EOs, respectively. However, L. burnatii EO was mainly represented by linalool (24.40%) and linalyl acetate (18.68%). The EPEO, LBEO, and OMEO had a strong antibacterial effect on S. aureus with minimal inhibitory concentrations (MICs) values ranging from 0.25 to 0.5% (v/v). Furthermore, the combination of 1/2048 MIC<sub>EPEO</sub> + 1/4 MIC<sub>LBEO</sub> showed a synergistic antibacterial effect on S. aureus with a FIC index of 0.25, while the formulation of  $1/4 \text{ MIC}_{\text{EPEO}} + 1/4$ MICOMEO demonstrated an antibacterial synergistic activity on E. coli with a FIC index of 0.5. Moreover, the simplex-centroid mixture design reported that the most effective combinations on E. coli and S. aureus correspond to 32%/28%/40% and 35%/30%/ 35% of E. polybractea, O. mixta, and L. burnatii, respectively. Presented information highlights the action of antibacterial formulations of these EOs and suggests their potential applications as alternatives to commercialized drugs to contract the development of bacteria causing serious infections and food deterioration.

## 1. Introduction

Infectious diseases triggered by antimicrobial resistance (AMR) are among the key issues impacting morbidity and fatality in patients suffering from such problems. The AMR

has become a main public health concern that threatens the effective treatment of a broad range of infections caused by bacteria, fungi, viruses, and parasites no longer vulnerable to the common antibiotics used to prevent them. For multiple decades, bacteria causing common or severe infections have

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developed resistance to each novel antibiotic coming to market. The impact of antimicrobial resistance concerning the public health charge is quite difficult to predict. The Center for Disease Control and Prevention (CDC) estimated that more than two million people per year are affected with antibiotic-resistant infections, with at least 23,000 deaths as a consequence of infectious diseases. In light of this fact, it is crucial to take action to prevent a developing healthcare crisis [1]. In fact, natural products represent a source of safe and effective agents which can be used as alternative to antimicrobial medications since their continuous use can increase the resistance of microorganisms, thus decreasing the efficiency of these drugs [2]. Essential oils (EOs) have been broadly employed for treating various ailments due to their famous antimicrobial activities [3]. Nowadays, EOs have been suggested and proved as antimicrobial products in complementary medicine. These properties are associated with the complex bioactive compounds of EOs, especially terpenes, aldehydes, phenylpropanoids, alcohols, esters, and ketones, which possess varied antimicrobial actions [4]. These bioactive molecules may act through different modes of action to apply their antimicrobial effect. Generally, EOs can induce membrane disruption and metabolic damages leading to cell death [5]. The application of EOs has been restricted due to their effects on sensory characteristics, specifically at elevated doses. Therefore, it is imperative to detect minimum inhibitory concentration for EOs and to evolve EO formulations in order to obtain synergistic effects, thus diminishing the amounts of EOs impacting the organoleptic properties without altering their antimicrobial effects [6]. Numerous works have established synergism between EOs using the checkerboard method without highlighting optimal EO formulations. However, only a few studies have recently used mixture design [7-10]. The mixture design approach can diminish the number of tests and predicted results can be modeled graphically and statistically aiming to minimize the total error. Through this approach, the accurate amounts of different EOs can be optimally associated to attain better mixtures [7]. In this context, we selected three medicinal plants to test the single and combined antibacterial effect of their EOs on pathogenic strains. These plants were chosen based on our previous ethnobotanical surveys which have demonstrated their therapeutic and culinary virtues [11-14]. In addition, some in vitro studies have revealed their significant antimicrobial abilities [8, 15, 16].

Eucalyptus polybractea, also named blue-leaved or blue mallee, is a multistemmed mallee eucalypt. This species belongs to the Myrtaceae family [17]. Eucalyptus has been used in traditional medicine to heal a variety of disorders including flu, fever, colds, sores, muscular pains, internal aches, and inflammation [18]. This plant oil is chiefly characterized by 1,8-cineole. The volatile oils derived from this genus are widely applied for pharmaceuticals, cosmetics, and food crops. Indeed, Eucalyptus oils have been reported to exhibit significant antibacterial, antifungal, anti-inflammatory, and antioxidant properties [15, 19].

Ormenis mixta, known as wild chamomile, is an endemic species of western and central Morocco. This plant belongs to the Asteraceae family and is usually used as general tonic, neurotonic, and aphrodisiac. It is endowed with anti-infectious, parasiticidal, antimutagenic, anticholesterol, and wound-healing properties [12]. It was previously reported that the plant oil is mainly characterized by D-germacrene and 1,8-cineole and possess promising anti-oxidant, anti-inflammatory, antidiabetic, and antimicrobial properties [20].

Lavandula burnatii, commonly known as Burnat's lavender, is a perennial herb native to the Mediterranean region and attributed to the Lamiaceae family. It has slender, graygreen leaves and bluish-purple flowers [21]. Previous phytochemical investigations showed that lavender essential oils are rich in linalool, linalyl acetate, and camphor. Prior research indicated that the antimicrobial effects of lavender oils are mainly related to their bioactive compounds [16, 22].

Previous investigations have reported the antibacterial effects of the three studied EOs. Indeed, it has been reported that the OMEO and its major constituents, such as p-menthane, germacrene D, and  $\alpha$ -pinene, exhibit powerful antibacterial properties against various strains, especially *Escherichia coli* and *Staphylococcus aureus* [20, 23–25]. Numerous studies reported the antimicrobial properties of the species belonging to the genus *Lavandula* and *Eucalyptus*, mainly *Lavandula stoechas*, *Lavandula intermedia*, and *Eucalyptus globulus* against a plethora of microbial strains [26–28].

The present investigation aims to analyze the phytochemistry of the EOs extracted from Eucalyptus polybractea cryptonifera (EPEO), Ormenis mixta (OMEO), and Lavandula burnatii briquette (LBEO) and evaluate their single antimicrobial activity as well as to determine the combination that affords the optimal inhibitory ability against the tested bacterial strains. Consequently, the simplex-centroid mixture design was employed to create polynomial models elucidating the relationship between the antibacterial effect and the amount of each volatile oil. Eucalyptus polybractea cryptonifera, Ormenis mixta, and Lavandula burnatii briquette were selected based on their medicinal applications. To the best of our knowledge, the essential oils derived from these plants have not been the subject of previous investigations. Hence, this is the first research aiming to combine these plants' essential oils and elucidate their antimicrobial and phytochemical characteristics.

The following points represent the hypothesis of this study:

- (i) The phytochemical analysis of the studied EOs will reveal the presence of diverse chemical compounds that could be responsible for the antimicrobial properties of these EOs
- (ii) Every single EO will exhibit important antimicrobial activity against the tested bacterial strains, demonstrating their potential as natural antibacterial agents

- (iii) The optimal mixtures of EPEO, OMEO, and LBEO resulting from the simplex-centroid mixture design will exhibit enhanced antibacterial effects compared to the single EOs
- (iv) The polynomial models will elucidate the quantitative relationship between the amount of each oil in the mixture and its corresponding antibacterial effect, providing valuable insights into the synergistic interactions of the studied EOs

#### 2. Materials and Methods

2.1. Plant Material and EOs Extraction. The aerial parts (stems, leaves, and flowers) of Eucalyptus polybractea cryptonifera, Ormenis mixta, and Lavandula burnatii briquet were harvested during the period between March and April at the flowering stage, since this is the best time to use plant essences more effectively [29].

The three plants were collected in the region of Taounate, Morocco (34° 32′ 09″ N, 4° 38′ 24″ W). Plant authenticity was carried out by the botanists of the Sidi Mohamed Ben Abdellah University, Fez, Morocco, and deposited under voucher codes of BLMUP 386-388.

The samples were dried to a constant weight for 10 days at 25°C in a dark place, crushed employing an electric blender, and sieved to attain a fine powder. Extraction of EOs from the plant aerial parts was executed by the hydrodistillation technique using Clevenger-type device. This technique features a recycling system and operates at atmospheric pressure. The system enables the preservation of mass plant/water proportion at its initial level. Through each extraction, 500 g of the dried aerial part of each plant was placed in a 1 L flask, and then 800 ml of distilled water was added. The solution was heated to boiling temperature (100°C) during 3 h; the released steams traversed the column and passed out of the condenser in a liquid form (extraction was performed in three separate replicates (n = 3)). At the end of the distillation process, two phases were noticed, an organic phase (EOs) and an aqueous phase (aromatic water). The attained oils were desiccated by anhydrous sodium sulfate and kept at 4°C pending upcoming tests.

- 2.2. Characterization of EOs Components. The analysis of E. polybractea (EPEO), O. mixta (OMEO), and L. burnatii (LBEO) was executed by gas chromatography coupled with mass spectrometry (GC/MS) and gas chromatography coupled with flame-ionization detector (GC-FID).
- 2.2.1. GC-FID Analysis. Analytical GC was performed using a Hewlett-Packard (HP/Agilent 6890) device equipped with a FID apparatus. The separation was accomplished using an HP-5 MS no-polar capillary column (length 30 m, diameter 0.25 mm, film thickness  $0.25 \,\mu\text{m}$ ).

The column temperature was set from 50°C to 200°C at 4°C/min. The chromatography carrier gas (nitrogen) was fixed to 1.4 ml/min, and split injection mode was used with a 1/50 split ratio; the temperature of injector and detectors (FID) was set at 250°C. The volume of oils (diluted 1/20 v/v

in methanol) injected was approximately 1  $\mu$ l. The device was controlled by an "HP Chem station" computer system, which managed its operation and allowed the monitoring of chromatographic analyses.

2.2.2. GC-MS Analysis. The GC-MS analysis was executed with a Hewlett-Packard gas chromatography (HP6890) equipped with a mass spectrometry system (HP 5973). Chromatographic separations were carried out using an HP-5 MS capillary column ( $30 \,\mathrm{m} \times 0.25 \,\mathrm{mm}$  i.d.,  $0.25 \,\mu\mathrm{m}$  film thickness). The carrier gas was helium whose flow rate was fixed at  $1.4 \,\mathrm{ml/min}$ .

The column temperature was managed at 50 and 200°C with a rate of 4°C/min. The injection of 1  $\mu$ L of EOs (diluted 1/20 in methanol solution) was performed, split ratio 1:30. The mass spectra (MS) were programmed over a scan range of 30 to 1000 amu; 0.5 s/scan. The ionization energy was 70 eV. The temperature at the ionization source and the input was 280°C.

The volatile compounds were identified based on their retention indices (RIs) and MS, which were compared with those obtained in the literature [30, 31]. Moreover, the mass spectra (MS) of various compounds were verified using standardized data from chemical libraries, including the NIST 2022 and the Wiley 275. Finally, commercialized standards (terpenes with purities ranged between 80 and 98%) were also used for external standardization.

- 2.3. Antibacterial Assays. Prior to formulation, the antibacterial activities of *E. polybractea*, *O. mixta*, and *L. burnatii* EOs were investigated separately. In order to assess the EOs antibacterial activity, the first step was the use of the disc diffusion method which allowed identifying the concentration giving a response classified as sensitive. Second, the minimum inhibitory concentration (MIC) and minimum bactericidal concentration (MBC) were determined for each EO.
- 2.3.1. Microorganisms. The EOs were examined against two reference bacterial strains: Escherichia coli ATCC 25922 and Staphylococcus aureus ATCC 29213. Both strains were obtained from the laboratory of Microbial Biotechnology and Bioactive Molecules, Science, and Technology Faculty, Fez. Before being used, the strains were reactivated by subculturing in Luria-Bertani (LB) plates at 37°C for 18–24 h.
- 2.3.2. Disc Diffusion Method. The antimicrobial activity of the studied EOs was determined using the agar disc diffusion method with minor reforms [32]. This method has been used as a first step to assess the inhibition diameters generated by the EOs around the disk.

A fresh culture suspension was prepared in sterile saline solution and adjusted to 0.5 McFarland ( $10^8$  CFU/mL), then inoculated into Mueller Hinton Agar (MHA) plates and incubated briefly for 20 min, and the culture's excess was eliminated. The sterile paper discs (6 mm diameter) were soaked with 5  $\mu$ l of pure EO from each plant before being put

on an inoculated agar plate. In addition, gentamicin ( $50 \,\mu\text{g}/\text{disc}$ ) and kanamycin ( $50 \,\mu\text{g}/\text{disc}$ ) were employed as standards to identify the sensitivity of the tested strains, while DMSO ( $5 \,\mu\text{L}$ ; 5%) was used as growth control. These plates were maintained at 4°C for 3 h and then incubated at 37°C for 24 h. After incubation, the inhibition zone diameters were measured in millimeters using a digital Vernier caliper (Mitutoyo). The results were represented as the mean± standard deviation for three separate tests (n=3).

2.3.3. Determination of MIC and MBC. MICs of EPEO, OMEO, and LBEO were assessed using the broth microdilution assay in 96-well microplates as previously explained [32, 33] with minor adjustments. Bacteriological agar was employed as an emulsifier of EOs in a culture medium at 0.15% (v/v), and p-iodo-nitrotetrazolium chloride (INT) 95% (Sigma-Aldrich) was used as a bacterial growth indicator. First, a microtiter plate was filled from the second to the twelfth well, with  $50 \,\mu$ l of Mueller-Hinton broth (MHB) supplemented with agar (0.15%). Then, 100 µL of EOs dilution prepared in MHB with agar (0.15%) was added to the first test well of each microtiter row. Next, 50 µl of scalar dilution was moved from the second to the eleventh. The 12th well was considered as growth control. Then, 50 µL of the bacterial suspension prepared and adjusted to 0.5 McFarland (10<sup>8</sup> CFU/mL) was deposited in each well.

The plates were incubated at 37°C for one day (24 h). Then,  $5\,\mu$ l of INT was added to each well. After 2 hours of incubation, the MIC was defined as the maximum EOs dilution where the white-to-red color shift was unnoticeable. Tests were conducted in triplicate (n=3). MBC was evaluated by subculturing  $15\,\mu$ L from each negative well on LB agar plates and incubating for 18-22 h at  $37^{\circ}$ C. MBC was the lowest concentration at which no growth was detected. Moreover, the MBC/MIC ratios were also determined to identify the possible mechanism of the studied EOs [34].

2.4. Synergism Testing. The checkerboard technique was performed to verify synergistic interactions between the tested EOs against bacterial strains [35].

The tested sample concentrations were prepared in MHB containing 0.15% bacteriological agar. On the microplate x-axis, 25  $\mu$ L of the weakest active EO concentration, which corresponds to the highest MIC value of EO (determined by microdilution assay) was added to the well from the 1<sup>th</sup> to the 11<sup>th</sup> one. Regarding the y-axis, 25  $\mu$ L of each higher active EO concentration, which consists of the lowest MIC value of EO (determined by microdilution assay) was added to each well. The 12<sup>th</sup> well served as a negative control. Then, 50  $\mu$ L of the bacterial suspensions, at a concentration of  $2 \times 10^6$  UFC/mL, was added to each well.

Thereafter, the microplate was incubated at 37°C for 18 to 20 h. After that,  $10 \,\mu\text{L}$  of resazurin was added to each well as a bacterial growth indicator. After incubation at 37°C for 90 min, bacterial growth was detected by reduction of blue dye (resazurin) to pink (resorufin). Experiments were perfomed in triplicate (n = 3).

The synergy was evaluated based on the instructions described by the American Society for Microbiology [36].

The fractional inhibitory concentration (FIC) index values were determined as follows:

$$\sum \text{FICI} = \text{FIC}(E_A) + \text{FIC}(E_B), \tag{1}$$

with

$$FIC(X_A) = \frac{MIC(E_{Acombination})}{MIC(E_{Aalone})},$$

$$FIC(X_B) = \frac{MIC(E_{Bcombination})}{MIC(E_{Balone})}.$$
(2)

The  $\sum$  FICI index was interpreted as follows: synergistic = FIC  $\leq$  0.5; partial synergy = 0.5 < FIC  $\leq$  0.75; additive interaction = 0.76  $\leq$  FIC  $\leq$  1.0; indifferent (noninteractive) = 1.0 < FIC  $\leq$  4.0; and antagonistic interaction = FIC >4.0.

## 2.5. Experimental Design

2.5.1. Mixture Design. The optimization and evaluation of antibacterial activity were generated using an augmented simplex-centroid design for three compounds [37]. For this, ten trials were conducted, using three pure EOs at the triangle's vertices  $(E_1 - E_2 - E_3)$ , binary mixtures at the three triangle's sides  $(E_4 - E_5 - E_6)$ , equiproportional mixture of the EOs at the triangle's centroids (central point)  $(E_7)$ , and control points  $(E_8 - E_9 - E_{10})$ .

The experiment of the equal proportionate mixtures has been tripled to determine the lack of the model fit. Ultimately, 12<sup>th</sup> experiments were used in the current exploratory design (Figure 1).

Factors ( $E_1$ ,  $E_2$ , and  $E_3$ ) are used to explain response variation in a mixture, and they represent a portion of each experimental component in the mixture, which has a value range between 0 and 1 without constraints [38], with

 $E_1$ : the proportion of EOEO

 $E_2$ : the proportion of OMEO

 $E_3$ : the proportion of LBEO

The responses assessed in this investigation were the antibacterial action against two bacteria, including *S. aureus* and *E. coli*. Next, the data were fitted to a special cubic model employing least-squares regression to reveal the unidentified coefficients in the following equation [39]:

$$Y = \delta_1 E_1 + \delta_2 E_2 + \delta_3 E_3 + \delta_{12} E_1 E_2 + \delta_{13} E_1 E_3 + \delta_{23} E_2 E_3 + \delta_{123} E_1 E_2 E_3 + \varepsilon,$$
(3)

where Y is the MIC response in % (v/v).  $\delta_1$ ,  $\delta_2$ , and  $\delta_3$  are the coefficients of linear terms.  $\delta_{12}$ ,  $\delta_{23}$ , and  $\delta_{23}$  are the coefficients of binary terms.  $\delta_{123}$  is the coefficient of ternary term.  $\varepsilon$  is the error term.

2.5.2. Statistical Analysis. The  $F_{\rm LOF/PE}$  ratio between the mean square lack of fit (MS<sub>LOF</sub>) and the mean square pure error (MS<sub>PE</sub>) was used to ensure the accuracy of the model

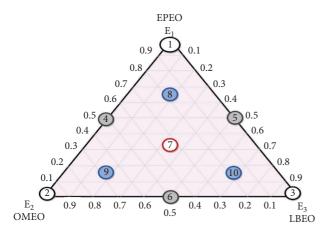


FIGURE 1: An overview of the simplex-centroid design for three-compound mixtures.

with observations. High  $F_{\rm LOF/PE}$  values signify a mismatch to the model [40, 41]. Moreover, the validation of the fitted models was verified using the ANOVA F-test. To determine the statistical significance of the model, we used the ratio between the mean square regression (MS $_R$ ) and the mean square residual (MS $_r$ ) [42]. Then, the quality of the assumed model was further assessed using the coefficient of determination  $R^2$ . In fact, it is frequently presented as a percentage (%) and used to assess the correlation among observed and expected responses [43], whereas Student's t-test was employed to determine the importance of the model's coefficients [44]. This analysis was executed applying JMP software version 14 and Design Expert version 12.

2.5.3. Optimization Tools. The contour plot and 3D surface, based on iso-response curves, were employed to identify the ideal EOs formulation, resulting in the required responses. These curves were used to look for the factors modification intervals to get the desired response [43, 45].

Besides, the desirability test was applied to identify the desired response values based on the optimal conditions. Thanks to this tool, we can provide the precise optimum setting with a percentage between 0 and 1. A value of 1 is given when factors result in the maximum desired response, whereas a value of 0 signifies an inadmissible response [46].

#### 3. Results and Discussion

3.1. Chemical Composition. The EOs yields (v/w) were 2.13%, 0.9%, and 1.38% for *E. polybractea*, *O. mixta*, and *L. burnatii*, respectively.

EOs generally include compounds derived from two major groups: monoterpenes and sesquiterpenes (hydrocarbons and their derivatives) [47]. Monoterpenes are a chemical family of terpenes that possess two isoprene units per component with a structural formula of  $C_{10}H_{16}$  (monoterpenes hydrocarbons). Monoterpenes may be linear

(acyclic) or comprise a single (monocyclic) or double rings (bicyclic) [48]. Modified terpenes, such as oxygen-containing components are known as oxygenated monoterpenes. Furthermore, sesquiterpenes are secondary metabolites consisting of 15-carbon components containing three isoprenoid units and representing a multifaceted and heterogeneous subclass of bioactive molecules [47, 49].

The chemical analyses of EPEO, OMEO, and LBEO, including the percentage of each constituent, elution order, molecular formula, and retention index, are summarized in Table 1. A total of 70 volatile components have been detected in the three studied EOs. Indeed, twenty-five, twenty-nine, and thirty-three components were identified in EPEO, OMEO, and LBEO, representing 98%, 94.82%, and 97.47% of these oils, respectively.

EPEO was characterized by a high amount of monoterpene hydrocarbons (57.18%) and a low amount of sesquiterpene hydrocarbons (0.97%). Besides, LBEO has revealed an important percentage of oxygenated monoterpenes (62.49%) and a moderate proportion of oxygenated sesquiterpenes (0.49%). As regards, OMEO, sesquiterpene hydrocarbons, and oxygenated monoterpenes were the most dominant constituents with 36.27% and 27.67%, respectively.

The chemical analysis of EOs attained from the three studied plants revealed that 1,8-cineole (23.75%), p-cymene (22.47%), and  $\alpha$ -pinene (11.20%) were the main bioactive compounds detected in the EPEO. These findings are in agreement with those cited by prior investigations mostly for 1,8-cineole dominance [50–53]. However, many studies that focused on the chemical composition of the genus *Eucalyptus* (*E. dives*) EO revealed other chemotypes: piperitone (40.50%),  $\alpha$ -phellandrene (17.40%), and p-cymene (8.50%) with a low rate of 1,8-cineole (0.70%) [54].

This difference may be ascribed to several factors, including environmental conditions (soil type, precipitation, and climate), plant origin, harvesting time, extraction and processing methods, and phenological stage of plant concerned, and it could be genetically determined [41, 55, 56].

As regards the analyzed OMEO, twenty-nine compounds were identified, with p-menthane-p1,8-diol (18.19%),  $\alpha$ -pinene (10.81%), and D-germacrene (9.17%) as major compounds. Prior investigations have determined the chemical composition of OMEO from other Moroccan areas and have revealed varying volatile constituents that depend on the plants' origin. For instance, the oil of O. mixta collected in the Taounate region mainly contains santolina alcohol, farnesene, and epi- $\alpha$ -macrogol [8]. Furthermore, Elouaddari et al. [57] found that the chemical composition of OMEO from Morocco varies qualitatively and quantitatively based on geographical location and growth conditions. Indeed, the OMEO collected from Benguerir, Kenitra, Settat, Meknes, and Tamesna regions contained camphor and  $\beta$ -myrcene, whereas the  $\beta$ -myrcene and  $\beta$ -farnesene chemotypes were present in the Chefchaouane's sample. However, the Bouznika sample contained methacrylate and 2-methyl-2-trans-buteny [58].

Concerning the phytochemical profile of LBEO, thirtythree constituents were identified, of which linalool was the major component with a percentage of 24.40%, followed by

TABLE 1: Chemical composition of EPEO, OMEO, and LBEO.

NI a	0 1 h	) ( 1	DIC	DT 1:4d	% of	% of relative peak area			
No. <sup>a</sup>	Compounds <sup>b</sup>	Molecular formula	RI <sup>c</sup>	RI lit <sup>d</sup>	EPEO	OMEO	LBEO	Identification	
1	α-Thujene	C <sub>10</sub> H <sub>16</sub>	902	905	0.94	_	_	MS, $I_R$	
2	Camphene	$C_{10}H_{16}$	943	943	0.26	_	_	MS, $I_R$	
3	α-Pinene	$C_{10}H_{16}$	948	948	11.20	10.81	0.33	MS, $I_R$	
4	3-Octanone	$C_8H_{16}O$	952	948	_	_	0.40	MS, $I_R$	
5	$\beta$ -Myrcene	$C_{10}H_{16}$	958	958	0.72	_		MS, $I_R$	
6	<i>p</i> -Menthane-1,8-diol	$C_{10}H_{20}O_2$	970	973	_	18.19		MS, $I_R$	
7	$\beta$ -Pinene	$C_{10}H_{16}$	972	972	8.08	3.17	3.73	MS, $I_R$	
8	Sabinene	$C_{10}H_{16}$	975	976	_	2.29	_	MS, $I_R$	
9	Yomogi alcohol	$C_{10}H_{18}O$	998	999	_	1.96	_	MS, $I_R$	
10	lpha-Phellandrene	$C_{10}H_{16}$	1005	1002	1.76	_	_	MS, $I_R$	
11	D-Limonene	$C_{10}H_{16}$	1018	1020	_	_	1.10	MS, $I_R$	
12	Sabinene hydrate	$C_{10}H_{18}O$	1041	1040	_	_	1.92	MS, $I_R$	
13	p-Cymene	$C_{10}H_{14}$	1042	1041	22.74	_	0.33	MS, $I_R$	
14	$\beta$ -Ocimene	$C_{10}H_{16}$	1051	1050	_	_	2.51	MS, $I_R$	
15	Terpinolene	$C_{10}H_{16}$	1052	1053	0.87	_	_		
16	1,8-cineole	$C_{10}H_{18}O$	1059	1059	23.75	_	6.57	MS, $I_R$	
17	γ-Terpinene	$C_{10}H_{16}$	1062	1060	7.29	0.97	_	MS, $I_R$	
18	Linalool oxide	$C_{10}H_{18}O_2$	1080	1079	_	_	1.21	MS, $I_R$	
19	Linalool	$C_{10}H_{18}O$	1082	1082	0.62	_	24.40	MS, $I_R$	
20	Artemisia alcohol	$C_{10}H_{18}O$	1084	1083		1.95		MS, $I_R$	
21	1-Octenyl-3-acetate	$C_{10}H_{18}O_2$	1109	1115	_	_	0.49	MS, $I_R$	
22	Camphor	$C_{10}H_{16}O$	1121	1122	_		8.01	MS, $I_R$	
23	(E)-Pinocarveol	$C_{10}H_{16}O$	1131	1132	0.59	1.17	<del>-</del>	MS, $I_R$	
24	Borneol	$C_{10}H_{18}O$	1138	1134	0.56	_	5.26	MS, $I_R$	
25	Fenchyl alcohol	$C_{10}H_{18}O$	1138	1137	0.48	_	_	MS, $I_R$	
26	α-Terpineol	$C_{10}H_{18}O$	1143	1144	3.11	_	1.71	MS, $I_R$	
27	Lavandulol	$C_{10}H_{18}O$	1146	1148	_	_	1.16	MS, $I_R$	
28	Allo-Ocimene	$C_{10}H_{16}$	1147	1147	0.66	_	0.61	MS, $I_R$	
29	Artemisyl acetate	$C1_2H_{20}O_2$	1153	1152	_	1.75		MS, $I_R$	
30	Terpinen-4-ol	$C_{10}H_{18}O$	1177	1177	2.85	1.26	3.79	MS, $I_R$	
31	Hexyl butanoate	$C_{10}H_{20}O_2$	1183	1182	_	_	0.84	MS, $I_R$	
32	Cryptone	$C_9H_{14}O$	1188	1188	2.69	_	_	MS, $I_R$	
33	[E]-3-Caren-2-ol	$C_{10}H_{16}O$	1188	1188	2.66	_	_	MS, $I_R$	
34	Cumin aldehyde	$C_{10}H_{12}O$	1230	1232	1.57	_	_	MS, $I_R$	
35	Carvacrol	$C_{10}H_{14}O$	1262	1269	0.70	_	2.00	MS, $I_R$	
36	Lavandulyl acetate	$C_{12}H_{20}O_2$	1270	1271	_	_	2.98	MS, $I_R$	
37 38	Linalyl acetate	$C_{12}H_{20}O_2$	1272 1277	1270 1273	_	1.39	18.68	$MS, I_R$	
39	Bornyl acetate Cuminol	$C_{12}H_{20}O_2$	1277	12/3	0.42	1.39 —	_	MS, $I_R$ MS, $I_R$	
40		$C_{10}H_{14}O$	1319	1318	0.42	_	0.55	MS, $I_R$	
41	Linalyl propionate Hexyl tiglate	$ C_{13}H_{22}O_2 $ $ C_{11}H_{20}O_2 $	1319	1318	_	_	0.33	MS, $I_R$	
42	α-Terpinyl acetate	$C_{11}H_{20}O_2$ $C_{12}H_{20}O_2$	1333	1333	1.35	_	—	MS, $I_R$	
43	Geranyl acetate	$C_{12}H_{20}O_2$ $C_{12}H_{20}O_2$	1352	1352	_	_	3.75	MS, $I_R$	
44	$\delta$ -Elemene	$C_{12}H_{20}C_{2}$ $C_{15}H_{24}$	1365	1462	_	7.18	_	MS, $I_R$	
45	$\beta$ -Elemene	$C_{15}H_{24}$ $C_{15}H_{24}$	1398	1393	<u>_</u>	2.33	_	MS, $I_R$	
46	Bornyl isobutyrate	$C_{14}H_{24}O_2$	1412	1402	_	1.03	_	MS, $I_R$	
47	α-Bergamotene	$C_{14}H_{24}C_{2}$ $C_{15}H_{24}$	1430	1430	_	_	0.43	MS, $I_R$	
48	γ-Muurolene	$C_{15}H_{24}$ $C_{15}H_{24}$	1435	1435		_	0.63	$MS, I_R$	
49	β-cis-Farnesene	$C_{15}H_{24}$	1440	1446		_	0.92	$MS, I_R$	
50	$(Z)$ - $\beta$ -Farnesene	$C_{15}H_{24}$ $C_{15}H_{24}$	1440	1443	_	7.58	_	MS, $I_R$	
51	$\alpha$ -Farnesene	$C_{15}H_{24}$ $C_{15}H_{24}$	1458	1456	_	2.46	_	MS, $I_R$	
52	Alloaromadendrene	$C_{15}H_{24}$ $C_{15}H_{24}$	1461	1461	0.97	_	_	MS, $I_R$	
53	Caryophyllene	$C_{15}H_{24}$ $C_{15}H_{24}$	1494	1493	_	3.59	2.21	MS, $I_R$	
54	α-Muurolene	$C_{15}H_{24}$	1499	1500	_	2.97	_	MS, $I_R$	
55	$\beta$ -Bisabolene	$C_{15}H_{24}$	1500	1500	_	_	0.59	MS, $I_R$	
56	Lavandulyl	$C_{15}H_{26}O_2$	1504	1502	_	_	0.84	$MS, I_R$	
57	γ-Cadinene	$C_{15}H_{24}$	1513	1513	_	_	0.28	MS, $I_R$	
58	$\delta$ -Cadinene	$C_{15}H_{24}$	1524	1524	_	1.99	_	$MS, I_R$	
								, 10	

TABLE 1: Continued.

NT - a	C 1-b	Molecular formula	RI°	RI lit <sup>d</sup>	% of	relative peak	area	Identification	
No. <sup>a</sup>	Compounds <sup>b</sup>	Molecular formula	KI	KI III	EPEO	OMEO	LBEO	Identification	
60 Spathulenol		C <sub>15</sub> H <sub>24</sub> O	1575	1572	3.14	1.40	_	MS, $I_R$	
61	Caryophylene oxide	$C_{15}H_{24}O$	1581	1583	_	0.90	0.90	MS, $I_R$	
62	Germacrene D	$C_{15}H_{24}$	1588	1587	_	9.17	_	MS, $I_R$	
63	Carotol	$C_{15}H_{26}O$	1593	1594	_	1.60	_	MS, $I_R$	
64	Cedrenol C <sub>15</sub> H <sub>24</sub>		1604	1604	_	0.91	_	MS, $I_R$	
65	Caryophyllene oxide $C_{15}H_{24}O$		1632	1630	_	1.63	_	MS, $I_R$	
66	τ-Muurolol	$\tau$ -Muurolol $C_{15}H_{26}O$		1640	_	1.04	_	MS, $I_R$	
67	au-Cadinol	$C_{15}H_{26}O$	1644	1640	_	_	0.59	MS, $I_R$	
68	$\alpha$ -Bisabolol	$C_{15}H_{26}O$	1682	1683	_	_	1.28	MS, $I_R$	
69	cis-Lanceol	$C_{15}H_{24}O$	1743	1746	_	0.84	_	MS, $I_R$	
70	Butanoic acid	$C_{15}H_{22}O_2$	1773	1773	_	1.48	_	MS, $I_R$	
Total ic	dentified (%)				99.98	95.82	97.47		
Monote	erpene hydrocarbons				57.18	17.24	8.61		
Oxyger	nated monoterpenes				38.73	27.67	62.49		
Sesquiterpene hydrocarbons						36.27	5.65		
Oxygenated sesquiterpenes						11.13	0.9		
Others						1.03	22.28		
Yield (	%, v/w)				2.13	0.7	1.48		

<sup>&</sup>lt;sup>a</sup>In order of elution on HP-5MS, <sup>b</sup>components identified by RI and MS. <sup>c</sup>Retention index determined from alkanes series (C9–C31). <sup>d</sup>Retention index from data libraries (NIST) [30, 31]. Bold values represent the proportions of the major components for each oil.

linally acetate (18.68%) and camphor (8.01%). According to Lesage-Meessen et al. [59], these compounds are primarily in charge of lavandin distinctive flavor as well as its biological and therapeutic characteristics. This outcome is in concordance with those described in the literature. Indeed, for the lavandin EO from Iran, 1,8-cineole was the major component (47.90%) followed by borneol (26.40%) and camphor (14.40%) [60]. Moreover, L. angustifolia EO from Italian origin (Salerno) was characterized by a high content of linalool (33.10%) and linally acetate (10.40%) [61]. Furthermore, the main constituents in lavandin (Lavandula × intermedia) grown in Lazio region (Italy) were also linalool (41.60%) and linalyl acetate (23%), with smaller amounts of 1,8-cineole (5.20%) [62]. In contrast, Romanian lavandin EO was richer in camphor and eucalyptol [26]. The fluctuations in the percentages of detected compounds may be related to either the altitude of the cultivation area and/or its specific microclimate conditions [27, 63-65].

3.2. Single Antibacterial Activity. Results of the antibacterial activity of EPEO, OMEO, and LBEO examined by disc diffusion method against two bacterial strains are summarized in Figure 2.

The inhibition zone diameter values are depending on the tested EOs' nature and the tested species' susceptibility. LBEO, EPEO, and OMEO showed an important antibacterial effect against *S. aureus* with respective inhibition zone diameters of  $24.66\pm1.45$ ,  $18.4\pm0.655$ , and  $14.9\pm0.75$  mm. Moreover, this noticeable antibacterial effect has been also revealed against *E. coli* with  $17.13\pm2.85$  mm,  $11.03\pm1.05$  mm, and  $8.52\pm0.82$  mm for EPEO, LBEO, and OMEO, respectively. These effects are less effective compared with the standard antibiotics

(gentamicin and kanamycin), which may be explained by the fact that antibiotics are particular antimicrobial drugs with specific site of action on bacteria from the one hand [66]. On the other hand, EOs consist of a large number of bioactive compounds, which may lead to antagonistic interactions, thus limiting the antimicrobial activities of EOs.

The quantitative antibacterial values (MIC and MBC) of the three tested EOs against *S. aureus* and *E. coli* are presented in Table 2.

All studied EOs have presented significant antibacterial activity. In fact, *S. aureus* appears to be the most sensitive strain to the three EPEO, LBEO, and OMEO, with MIC values ranging from 0.25 to 0.5% (v/v). Indeed, LBEO and EPEO have the strongest antibacterial activity with MIC values of 0.25% (v/v). Moreover, noticeable antibacterial effect has also been revealed for OMEO with MIC equal to 0.5% (v/v).

However, a lower antibacterial effect has been shown against *E. coli* with MIC values ranging from 1 to 6% (v/v). In fact, LBEO exhibited the highest antibacterial activity (1% v/v) followed by EPEO (2% v/v), while OMEO showed the weakest antibacterial effect (6% v/v). This is to note that the standard antibiotics gentamicin and kanamycin have shown MIC values ranging from 8 to 32% (v/v) against both strains. The MBC values of the three tested EOs are quite similar to the MIC values obtained against *E. coli* and *S. aureus*. However, for EPEO, the MBC value was twice as high as the MIC. Concerning this bactericidal effect, it has been found that the Grampositive (Gram+) bacterial strain is more sensitive to EPEO, LBEO, and OMEO than the Gram-negative (Gram-) bacteria.

The antibacterial activity of EOs could be elucidated by the molecular interaction of the functional groups of their components with the bacterial cell wall, causing wholecell lysis.

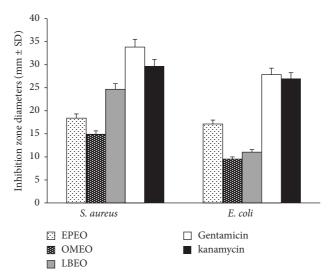


FIGURE 2: Screening of antibacterial action by disc diffusion method of EPEO, LBEO, and OMEO and standard antibiotics (gentamicin and kanamycin) against *S. aureus* and *E. coli*.

TABLE 2: The MIC and MBC values of EPEO, LBEO, and OMEO against bacteria strains using microdilution assay.

Bacteria <sup>a</sup>	EPEO (%v/v)				LBEO (%v/v)			OMEO (%v/v)			Gentamicin <sup>b</sup> (μg/mL)		Kanamycin <sup>b</sup> (μg/mL)	
	MIC	MBC	Effect	MIC	MBC	Effect	MIC	MBC	Effect	MIC	MBC	MIC	MBC	
E. coli	2	4	Bacteriostatic	1	1	Bactericidal	6	6	Bactericidal	16	32	32	32	
S. aureus	0.25	0.25	Bactericidal	0.25	0.25	Bactericidal	0.5	0.5	Bactericidal	8	32	16	32	

<sup>&</sup>lt;sup>a</sup>Final bacterial concentration was 10<sup>6</sup> CFU/mL. <sup>b</sup>Gentamicin and kanamycin were used as references.

Liberation of internal components from the cell is a good indicator of membrane integrity, with small ions such as phosphate and potassium that have a tendency to diffuse before large molecules such as RNA and DNA and other substances [28, 67]. Moreover, the resistance of Gramnegative (Gram<sup>-</sup>) bacteria is strongly linked to the composition of their cell wall which limits the diffusion of hydrophobic components such as EOs and their bioactive components through the lipopolysaccharide (LPS) layer [68, 69]. However, LPS are absent in Gram-positive bacteria, which possess a cell wall mainly constituted by a peptidoglycan layer facilitating the diffusion of EOs through cell membrane and thereby distributing cell permeability and binding with vital macromolecules, such as proteins, DNA, and RNA and thus causing cell death [66].

Previous investigations have reported the antibacterial effects of the three studied EOs. Our findings are consistent with those reported by Ouedrhiri et al. [20] who showed that OMEO causes a significant inhibition on Gram<sup>+</sup> bacteria (*S. aureus* and *B. subtilis*), while a weak antibacterial effect was determined against Gram<sup>-</sup> bacteria, including *Pseudomonas aeruginosa* and *E. coli* [8, 20].

As it can be observed, the Gram strains are more resilient than the Gram ones. Indeed, the Gram have an envelope that consists of three layers. The first layer is the outer membrane, a protective and a unique feature that distinguishes Gram from Gram bacteria [3]. The outer membrane of Gram strains is the principal reason for resistance to a broad variety of antimicrobial agents

including essential oils due to their hydrophobic characteristics [10]. Besides, the Gram<sup>-</sup> strains can change their hydrophobic properties via mutations, creating resistance to EOs, while the Gram<sup>+</sup> ones do not have this strong layer, which makes Gram<sup>-</sup> bacteria more resistant than Gram<sup>+</sup> ones [70].

Nevertheless, it has been previously shown that the major constituents of OMEO (p-menthane, germacrene D, and  $\alpha$ -pinene) exhibit antibacterial effect against E. coli and S. aureus [20, 23–25]. Therefore, the lower antibacterial action found against Gram bacteria might account for an antagonistic interaction among its volatile compounds. Indeed, numerous researches have shown that the antibacterial effect of EOs is controlled by the intricate interplay between their minor and major constituents. In some cases, these components are active against bacterial cells when evaluated separately [20, 71].

To the best of our knowledge, there is no literature data demonstrating the antibacterial activity of LBEO. However, numerous studies reported the antibacterial activity of the genus *Lavandula*. Indeed, Bouyahya et al. [72] showed that *Lavandula stoechas* exhibits antibacterial activity with some variability depending on the tested bacteria, experimental methods used, and/or chemical composition of the EOs. In addition, similar findings have been reported by Dadaliogÿlu and Evrendilek and Cherrat et al. [73, 74].

Furthermore, Garzoli et al. [62] showed that lavandin (*Lavandula intermedia*) EO exhibits bactericidal activity against a wide range of pathogenic bacteria. This effect can

be attributed mainly to the dominant presence of linalool (41.60%) in this oil. Indeed, Silva et al. [75] also attributed *Ocimum basilicum* EO activity to linalool. Furthermore, Hussain et al. [76] have demonstrated that *Ocimum basilicum* EO and linalool compound display antibacterial and antifungal actions against *E. coli*, *S. aureus*, *B. subtilis*, and *Aspergillus niger*. The research suggests that the antimicrobial activity of the LBEO could be associated with its high content of oxygenated monoterpenes which are highly effective against microbial cells [77].

In addition, EPEO has not gained much attention regarding its antibacterial properties. In this context, we have investigated this study to report the antibacterial activity of this plant. Djenane et al. and Oyedeji et al. [78, 79] obtained similar results with some variability against a panel of bacteria. Our EO exhibited significant bacteriostatic and bactericidal effects against *E. coli* and *S. aureus* compared to the results obtained by Fahad et al. [50], while these findings are corroborated with those reported by Assaggaf et al. [80], which indicated a strong antibacterial effect at low MIC and MBC values. These differences may be attributable to the variations in the EOs' chemical composition and variations in the experimental conditions as well as to bacterial strains tested.

The mechanism of EPEO, LBEO, and OMEO remains unresolved. Nevertheless, it has been found that other EOs have many antibacterial mechanisms [68, 81, 82]. Indeed, the mechanisms of action include the capacity to cross the cell membrane, the disturbance in the electron respiratory chain, and the leakage of electrolytes [81]. Other mechanisms have shown that EO components inhibit the quorum sensing signaling pathways, thus decreasing the bacterial resistance [83–85].

Besides, the antimicrobial properties of EOs can be attributed to the composition, functional groups of the bioactive compounds, and their synergistic interactions. Generally, terpenes and terpenoids are the main groups of EOs. They are characterized by a small molecular weight. The terpenoids group can be partitioned into alcohols, ketones, esters, phenols, aldehydes, epoxides, and ethers [81]. Many terpenoid compounds have demonstrated significant antimicrobial effects against various Gram<sup>+</sup> and Gram<sup>-</sup> strains, especially thymol, geraniol, carvacrol, linalyl acetate, menthol, piperitone, and linalool which are the major compounds of terpenoids. These compounds are able to interact with membrane proteins and disrupt the outer and inner membrane of bacteria, resulting in bacterial death [70, 80].

#### 3.3. Optimization of Antibacterial Action by Mixture Design

3.3.1. Antibacterial Formulations Design. Table 3 displays the mixtures design, which comprises different mixtures of OMEO, LBEO, and EPEO, along with the observed responses for each experiment on *S. aureus* and *E. coli*. The experiments were conducted randomly and each response is the mean of three repetitions. The results demonstrated that the equiproportional mixture of three EOs and the ternary mixture (0.167: OPEO/0.167: OMEO/0.667: LBEO) were the

most performant formulations, presenting the highest antibacterial activity against the two studied strains.

3.3.2. Statistical Validation of the Postulated Model. The experimental response data were statistically analyzed to confirm the chosen model for each bacterial strain, which represents the relationship between responses and factors.

The variance analysis reveals that the regression main impact is statistically significant for the two examined responses since their risk significance (p value) is less than 0.05 (0.0001\* and 0.001\* for E. coli and S. aureus, respectively). In addition, the  $F_{\mathrm{ratio}(R/r)}$  calculated for both studied responses are higher than the theoretical value at the 95% confidence level. As shown in Table 4, the  $F_{\mathrm{ratio}(R/r)}$  for E. coli (117.694) and S. aureus (29.41) exceeded the tabular value of F at a 95% confidence level. Besides, the ANOVA F-test indicated that the both postulated models had no lack of adjustment, because their p values were higher than 0.05 (0.07 and 0.051). The computed  $F_{\mathrm{Ratio}(\mathrm{LOF/PE})}$  of the investigated responses was also observed to be lower than the theoretical value  $F_{(0.05;3.2)} = 19.16$  at the 95% confidence level.

The coefficients of determination  $R^2$  for *S. aureus* and *E. coli* are 97% and 99%, respectively. These values indicate a sufficient agreement between the experimental values and those predicted by the fitted model. These findings were supported by the graph in Figure 3, which shows that the curves of observed values as a function of predicted values look exactly like a straight line.

3.3.3. Factors Effects and the Fitted Model of Both Responses. The impact of all investigated factors, their corresponding t-student statistical values, and the p values, are summarized in Table 5. The mathematical model coefficients were statistically significant when their p values were lower than 0.05, while those with a p value higher than 0.05 were excluded from the presumed model.

The statistically significant coefficients for the MIC response of *E. coli* are linear terms ( $\delta_1$ ,  $\delta_2$ , and  $\delta_3$ ), binary interaction terms between EPEO and OMEO ( $\delta_{12}$ ) as well as OMEO and LBEO ( $\alpha_{23}$ ) and finally ternary term ( $\delta_{123}$ ). These outcomes showed that the antibacterial action on *E. coli* is dependent on all terms in the modulated mathematical model, except for the coefficient associated with the binary term ( $\alpha_{13}$ ) between EPEO and LBEO.

The mathematical model was estimated according to the following formula:

$$Y = 2.02E_1 + 5.92E_2 + 0.96E_3 - 8.12E_1E_2 - 6.21E_2E_3$$
$$-32.04E_1E_2E_3 + \varepsilon.$$

(4)

Regarding the response MIC<sub>S.aureus</sub>, all the terms in the adapted mathematical model ( $\delta_1$ ,  $\delta_2$ ,  $\delta_3$ ,  $\delta_{12}$ ,  $\delta_{13}$ , and  $\delta_{123}$ ) are statistically significant, except for the coefficient representing the binary mixture of EPEO and LBEO ( $\alpha_{13}$ ). These findings reflect that the antibacterial effect against this bacterial strain depends on all interactions except those obtained by EPEO\* LBEO.

98%

85%

N	EDEO	OMEO	LDEO	MIC % (v/v) <sup>b</sup>		
Number of experiments <sup>a</sup>	EPEO	OMEO	LBEO	E. coli	S. aureus	
1	1	0	0	2	0.25	
2	0	1	0	6 1 2	0.5 0.25 0.25	
3	0	0	1 0			
4	0.5	0.5				
5	0.5	0	0.5	2	0.25	
6	0	0.5	0.5	2	0.25	
7	0.333	0.333	0.333	0.5	0.125	
8	0.333	0.333	0.333	0.5	0.125	
9	0.333	0.333	0.333	0.5	0.125	
10	0.667	0.167	0.167	1	0.125	
11	0.167	0.667	0.167	2	0.25	
12	0.167	0.167	0.667	0.5	0.125	

Table 3: Various combinations generated by the chosen mixture design and response data for each trial.

Models			E. coli					S. aureus		
	DF	SS	MS	F	p value	DF	SS	MS	F	p value
R	6	25.486212	4.24770	117.6945	< 0.0001*	6	0.125	0.021	29.41	0.001*
r	5	0.180455	0.03609			5	0.004	0.001		
LOF	3	0.190	0.06015	14.32	0.07	3	0.0036	0.00122	12.2	0.051
PE	2	0.00012	0.0042			2	0.0001	0.0001		
Total	11	25.666667				11	0.129			
$R^2$	99%					97%				

TABLE 4: Variance analysis for the three fitted models.

R: regression; r: residual; LOF: lack of fit; PE: pure error;  $R^2$ : coefficient of determination; adj: adjusted; pred: predicted; DF: degrees of freedom; SS: sum of squares; MS: mean square; \*statistically significant.

93%

42%

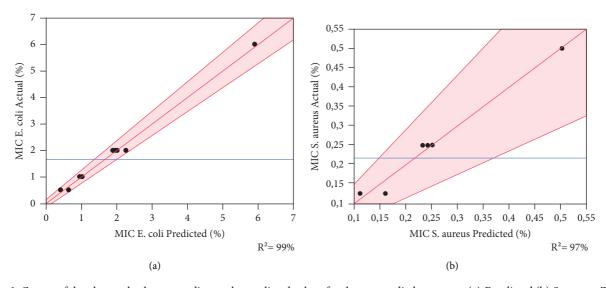


FIGURE 3: Curves of the observed values according to the predicted values for the two studied responses (a) *E. coli* and (b) *S. aureus*. The red lines show the curve of actual values of MIC as a function of those predicted for both tested strains. The blue horizontal lines indicate the mean of the observed values.

<sup>&</sup>lt;sup>a</sup>Experiments were carried out after randomization. <sup>b</sup>Each response is the average of three replicates.

E. coli S. aureus p value Term Coefficient Estimation t-ratio p value Estimation t-ratio **EPEO**  $\delta_1$ 2.02 10.99 0.0001\*0.24 9.40 0.0002\***OMEO**  $\delta_2$ 5.92 32.29 < 0.0001\* 0.50 19.56 < 0.0001\*  $\delta_3$ LBEO 0.97 5.29 0.0032\*0.24 9.40 0.0002\*EPEO\* OMEO -8.790.0111\* $\delta_{12}$ -8.120.0003\*-0.51-3.93EPEO \* LBEO 1.97 0.8158  $\delta_{13}$ 2.13 0.086 -0.03-0.25OMEO \* LBEO -6.210.0011\*0.0111\* $\delta_{23}$ -6.72-0.51-3.93EPEO \* OMEO \* LBEO  $\delta_{123}$ -32.04-6.380.0014\*-2.70-3.830.0123\*

TABLE 5: Estimated regression coefficients for the uncompleted cube regression model.

The predictive mathematical mode was determined as follows:

$$Y = 0.24E_1 + 0.5E_2 + 0.24E_3 - 0.51E_1E_2 - 0.51E_2E_3 - 2.7E_1E_2E_3 + \varepsilon.$$
 (5)

3.3.4. Formulation Optimization and Desirability Study. The optimization process adopting the mixture design approach enables us to determine the optimal formulation of the three EOs demonstrating the best MIC value, that is to say, the smallest concentration which exhibits the highest sensitivity. The smallest MIC values noticed during the experiments were 0.5% and 0.125% for *E. coli* and *S. aureus*, respectively. Consequently, a formulation of the three EOs allowing to obtain responses smaller than or equal to these values will be elucidated.

In the present research, it is important to emphasize that the ternary mixture indicates stronger antibacterial activity against both bacteria than single oils and binary mixtures. This positive combination is depicted in Figure 4, where the optimum mixture zone is situated in the center of the triangle. The contour plot and 3D surface (Figure 4) demonstrate the interaction among each component of the mixtures. This graph illustrates changes in response, with the dark blue zone representing weak MIC values and greater bacterial potential. However, the green to orange hue indicates elevated MIC values. Hence, the mixture design approach optimized the amounts of individual active constituents to generate an optimal formulation established by its potent antibacterial activity.

## (i) Effect of the EOs formulation against E. coli

The MIC<sub>E. coli</sub> response achieved in the various tests ranged from 0.5 to 6% (Table 3). Figure 4(a) indicates the contour and surface plots of the MIC<sub>E. coli</sub> response found with diverse mixtures of the EPEO, OMEO, and LBEO. A MIC value equal to 0.40% was determined as a compromise against *E. coli*. From the 2D and 3D mixing graph, we can conclude that a mixture of EPEO, OMEO, and LBEO, is necessary to achieve this MIC value. In addition, the desirability function (Figure 5(a)) exhibited the precise amounts of EPEO, OMEO, and LBEO leading to the desired MIC value of 0.37%. Thus, the desirability test revealed that there is a 99.9% chance of reaching

this value using the following mixture: 32%, 28%, and 40% of EPEO, OMEO, and LBEO, respectively.

## (ii) Effect of the EOs formulation against S. aureus

The results of the microdilution assay indicated that the MIC<sub>S.aureus</sub> response varied between 0.125 and 0.5% (Table 3). Moreover, 2D and 3D mixture plots (Figure 4(b)) illustrated the optimal compromise zone, revealing that the mixture comprising EPEO, OMEO, and LBEO is required to attain the desired MIC of about 0.12%.

In fact, the desirability function (Figure 5(b)) confirms this finding, indicating that the ternary mixture with the following proportion (35:30:35 v/v/v) EPEO, OMEO, and LBEO leads to the best achievable MIC value (0.11%), with a compromise percentage of 99.9%. These outcomes demonstrate the synergistic interaction between these components.

Numerous investigators are currently using a mixture design methodology to assess the potential interactions between various mixture constituents in order to estimate the optimal formulation [38, 43, 86, 87].

Zieniuk and Bętkowska [86] employed the mixture design approach to evaluate and optimize the synergistic antibacterial effect among tea tree, rosewood, and lavender EOs against *E. coli, Listeria monocytogenes*, and *Rhodotorula mucilaginosa*. Within this line, the synergistic effects of *Eucalyptus camaldulensis*, *Mentha pulegium*, and *Rosmarinus officinalis* EOs against bacteria responsible for nosocomial infections were also investigated using a simplexcentroid design [38]. In addition, Fadil et al. [43] optimized the proportions of *Thymus. vulgaris* L., *R. officinalis* L., and *Myrtus communis* EOs by applying a centroid mixture design. The optimal formulation corresponded to 45% of myrtle and 55% of thyme EOs, which showed synergistic activity against *Salmonella typhimurium* strain.

3.3.5. Interaction between EOs. The mixture contour plot of E. coli and S. aureus responses, generated by the three EOs, EPEO, OMEO, and LBEO, elucidates the impact of the simultaneous optimization. The optimal compromise area for the two strains indicated that the desired MIC requires a ternary mixture of the abovementioned EOs (Figure 6).

The mixture of these components allowed particularly oxygenated monoterpenes (1,8-cineole, p-menthane, and

<sup>\*</sup>Statistically significant at p < 0.05.

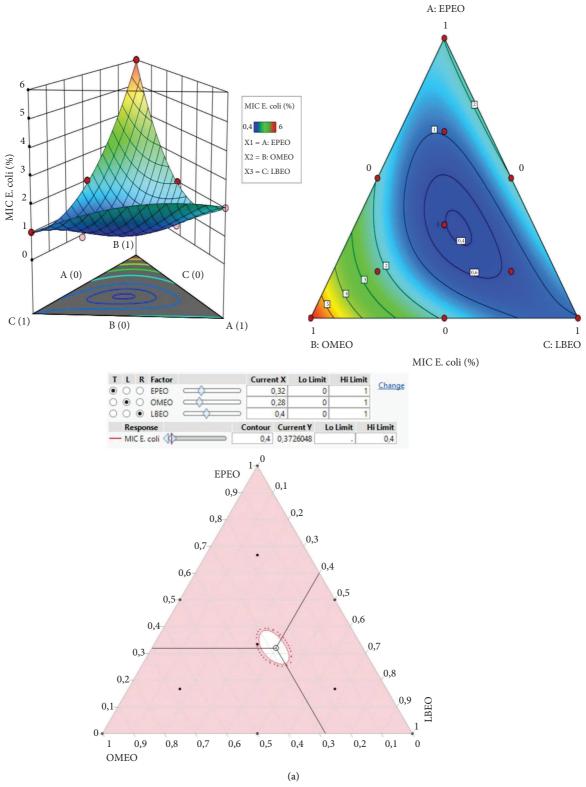


FIGURE 4: Continued.

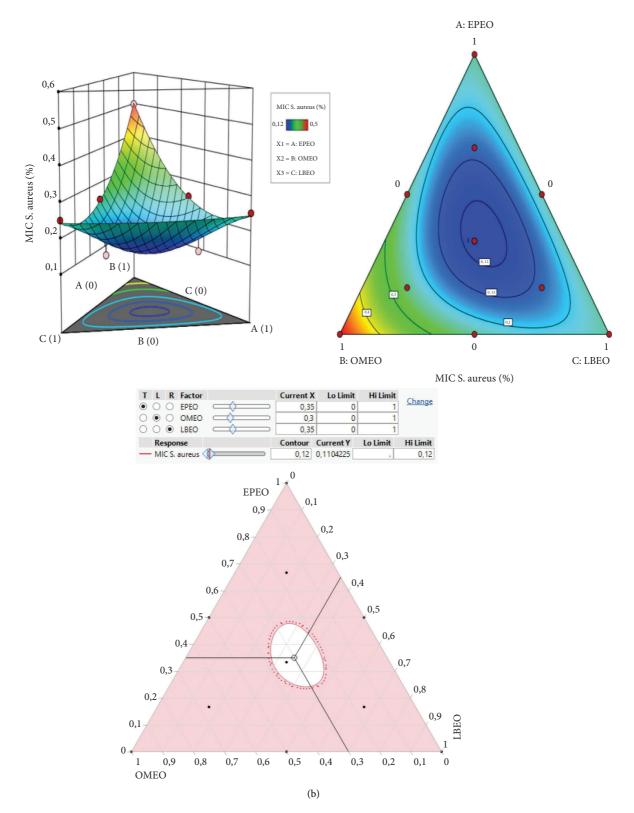


Figure 4: 2D and 3D mixture plot displaying the optimal compromise area leading to the desired MIC values against *E. coli* (a) and *S. aureus* (b).

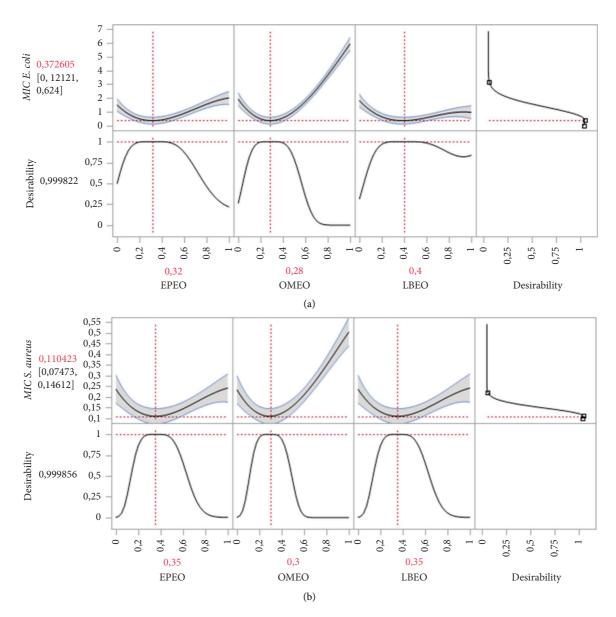


FIGURE 5: Desirability plot revealing the precise proportions of LBEO, OMEO, and EPEO leading to the best antibacterial action against *E. coli* (a) and *S. aureus* (b).

linalool) and monoterpene hydrocarbons (p-cymene and  $\alpha$ -pinene) to assemble. Each volatile compound has different sites within the bacterial cell where it can act [88]. The oxygenated terpenoids, including 1,8-cineole, p-menthane, and linalool, are the major antibacterial constituents as compared with the terpene hydrocarbons, which lack hydroxyl groups (-OH) [39, 89]. In addition, the combination of minor and/or major constituents may be responsible for the synergistic effect on bacteria [90]. p-Cymene and  $\alpha$ -pinene are not effective antibacterials when acting alone, but their combination with oxygenated terpenoids such as linalool and/or 1,8-cineole has shown promising antibacterial activities. Therefore, p-cymene can swell bacterial cell membranes, facilitating the diffusion of linalool and 1,8-cineole into the cell membrane, leading to bacterial death.

Various synergistic antibacterial effects have been described for compounds or fractions of EOs when studied in

binary/ternary mixtures [71, 91–95]. Ultee et al. [91] reported that cymene, combined with an oxygenated monoterpene (carvacrol), possess a synergistic activity against Gram<sup>+</sup> bacteria. Moreover, the combination eugenol/linal-ool has shown a synergistic activity toward Gram<sup>-</sup> bacteria [92]. Additionally, combinations of 1,8-cineole/limonene, 1,8-cineole/thymol, 1,8-cineole/p-cymene, and  $\alpha$ -pinene/linalool have been found to exhibit synergistic antimicrobial activity [93, 94, 96].

In this respect, to our knowledge, the current study reports for the first time the synergistic activity of 1,8-cineole, p-Menthane-1,8-diol, and linalool. In addition, this combination could increase the levels of components with antimicrobial properties, such as D-limonene,  $\gamma$ -cadinene, and  $\alpha$ -pinene [97, 98].

Taken together, these findings provide scientific evidence for potential applications of studied oils in combination to

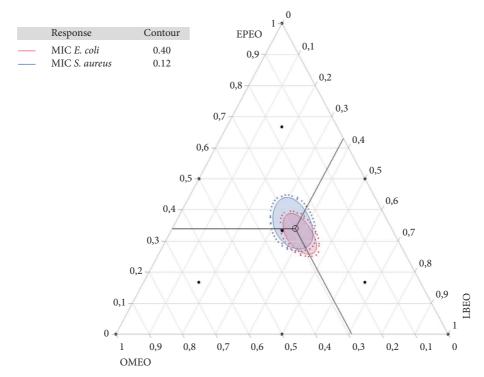


FIGURE 6: Mixture plot showing the optimal combination zone among O. mixta, E. polybractea, and Lavandula briquit against bacterial strain.

develop novel and effective antimicrobial agents, which may be useful in food packaging, food preservation, and elaboration of biopharmaceuticals. However, further investigations on the antibacterial action of these oils alone and/or in mixture are strongly required to clearly describe in detail how they could interact with each type of bacteria. Furthermore, in the literature, the results showed that the three EOs have effectively inhibited the growth of foodborne pathogens in vitro, whereas their action in food system model (*in vivo*) has not been reported. Indeed, more studies are needed in this subject to validate the possible applications of the three studied oils as natural additive in foods to persevere their microbiological security.

3.3.6. Synergistic Activity Using Checkerboard Assay. FIC index findings of the binary combination between tested EOs on *E. coli* and *S. aureus* are summarized in Table 6. Three dual combinations were assessed, including EPEO/OMEO, EPEO/LBEO, and LBEO/OMEO.

FIC index of the binary combination between tested EOs ranged from 0.156 to 0.75. As regards S.~aureus, it has been noticed that the combinations of 1/4  $\rm MIC_{EPEO}+1/32$   $\rm MIC_{OMEO}$ , 1/8  $\rm MIC_{EPEO}+1/16$   $\rm MIC_{OMEO}$ , 1/32  $\rm MIC_{EPEO}+1/8$   $\rm MIC_{OMEO}$ , and 1/64  $\rm MIC_{EPEO}+1/4$   $\rm MIC_{OMEO}$  displayed synergistic effects with respective FIC indexes of 0.281, 0.1874, 0.156, and 0.265, whereas the combinations 1/2  $\rm MIC_{EPEO}+1/4$   $\rm MIC_{OMEO}$  and 1/256  $\rm MIC_{EPEO}+1/2$   $\rm MIC_{OMEO}$  exhibited a partial synergy with respective FIC indexes of 0.656 and 0.503.

Furthermore, 1/2048 MIC<sub>EPEO</sub> + 1/4 MIC<sub>LBEO</sub> showed a synergistic antibacterial activity on *S. aureus* with a FIC index of 0.25, while the combination of 1/4096 MIC<sub>EPEO</sub> + 1/2 MIC<sub>LBEO</sub> has shown a partial synergistic effect with a FIC index of 0.5002. Moreover, the combination of 1/4 MIC<sub>LBEO</sub>/1/2<sub>OMEO</sub> also presented a partial synergistic activity, with a FIC index value of 0.75.

Concerning the effect of the tested EOs combination treatment against  $E.\ coli$ , it has been found that the combinations of 1/4 MIC<sub>EPEO</sub> + 1/4 MIC<sub>OMEO</sub> had an antibacterial synergistic activity with a FIC index of 0.5, whereas the combination 1/2 MIC<sub>EPEO</sub> + 1/8 MIC<sub>OMEO</sub> exhibited partial synergy with a FIC index of 0.625. The combinations of EPEO/LBEO and LBEO/OMEO demonstrated a synergistic and partial synergistic effect against  $E.\ coli.$ 

The mixture of 1/4 MIC<sub>EPEO</sub> + 1/512 MIC<sub>LBEO</sub> and 1/2 MIC<sub>LBEO</sub> + 1/4 MIC<sub>OMEO</sub> displayed synergy with respective FIC index values of 0.251 and 0.5.

In contrast, four combinations showed partial synergy with a FIC index ranging from 0.50024 to 0.625. Each checkerboard assay produces diverse combinations. However, the FIC values of the most efficient combination are employed to assess the FIC index [71].

Taken together, the binary combinations between the three tested EOs had a greater antibacterial effect (synergistic) than the application of EOs alone, when tested against *E. coli* and *S. aureus*. It has been demonstrated that the minor compounds are involved in antibacterial effect and may exhibit synergistic interactions with other constituents [68].

Pactoria	Sam	ples	MI	C % (v/v)	FIC %	FIC index	Effect	
Bacteria	Sall	ipies	Alone	Combination	(v/v)	FIC index	Епест	
	1/2	EPEO	0.25	0.125	0.5	0.656	Dantial over anon	
	1/4	OMEO	0.5	0.0078	0.0156	0.030	Partial synergy	
	1/4	EPEO	0.25	0.0625	0.25	0.281	Crrnanar	
	1/32	OMEO	0.5	0.0156	0.0312	0.281	Synergy	
	1/8	EPEO	0.25	0.3125	0.125	0.1874	Synergy	
	1/16	OMEO	0.5	0.0312	0.0624	0.16/4	Syllergy	
	1/32	EPEO	0.25	0.007813	0.03125	0.156	Synergy	
	1/8	OMEO	0.5	0.0625	0.125	0.130	, .,	
S. aureus ATCC 29213	1/64	EPEO	0.25	0.0039	0.0156	0.2656	Synergy	
3. uureus ATCC 29213	1/4	OMEO	0.5	0.125	0.25	0.2030	Syneigy	
	1/256	EPEO	0.25	0.00976	0.003907	0.5039	Partial synergy	
	1/2	OMEO	0.5	0.25	0.25	0.3039	raitiai syneigy	
	1/2048	EPEO	0.25	0.0001220	0.00048828	0.25	Synergy	
	1/4	LBEO	0.25	0.0625	0.25	0.23	Syneigy	
	1/4096	EPEO	0.25	$6.103.10^{-5}$	0.00024412	0.5002	Partial synerg	
	1/2	LBEO	0.25	0.125	0.5	0.3002	raitiai syneigy	
	1/4	LBEO	0.25	0.0625	0.25	0.75	Partial synerg	
	1/2	OMEO	0.5	0.25	0.5	0.73	r artiar synergy	
	1/2	EPEO	2	1	0.5	0.625	D	
	1/8	OMEO	6	0.75	0.125	0.625	Partial synergy	
	1/4	EPEO	2	0.5	0.25	0.5	C	
	1/4	OMEO	6	1.5	0.25	0.5	Synergy	
	1/4096	EPEO	2	0.00048828	0.00024414	0.50024414	D	
	1/2	OMEO	6	3	0.5	0.50024414	Partial synergy	
	1/8	EPEO	2	0.25	0.125	0.625	D	
	1/2	LBEO	1	0.5	0.5	0.625	Partial synergy	
E. coli ATCC 25922	1/4	EPEO	2	0.5	0.25	0.25105212	C**** ****	
E. con ATCC 25922	1/512	L.BEO	1	0.00195313	0.00195313	0.25195313	Synergy	
	1/2	EPEO	2	1	0.5	0.50024414	Partial synergy	
		LBEO	1	0.00024414	0.00024414	0.50024414 Par		

0.5

0.75

0.25

1.5

0.0625

3

TABLE 6: FIC indices of the combined antibacterials against the tested bacterial strains.

The synergistic effect observed in this research between the studied EOs could be explained by the molecular interaction of the functional groups (monoterpene hydrocarbons, oxygenated monoterpenes, and sesquiterpene hydrocarbons). They integrate and disrupt the cell membrane and thereby facilitate the uptake of the active compounds, leading to cell lysis [99].

1/2

1/8

1/2

1/4

1/16

1/2

**LBEO** 

**OMEO** 

**LBEO** 

**OMEO** 

**LBEO** 

**OMEO** 

1

6

1

6

1

6

Other investigations also reported a synergic effect among O. mixta, Eucalyptus, and Lavandula EOs with other EOs [8, 20, 70, 100, 101]. In fact, Ouedrhiri et al. [20] highlighted the effect of the combination of O. Mixta and Pelargonium asperum EOs. The results showed that O. mixta MIC decreased from 2 to 0.007813% (v/v) against S. aureus after combination with Pelargonium asperum EOs. On the other hand, a synergistic activity was attained between Eucalyptus and Dracocephalum EOs against S. aureus and E. coli [70].

Furthermore, Moussii et al. [101] showed that the combination of lavender, wormwood, and rosemary EOs display a synergistic effect against Gram<sup>+</sup> and Gram<sup>-</sup>

bacteria. A number of reports have proposed certain specific mode of action of antibacterial interaction that produce synergism outcomes, including modulating certain common biochemical pathways, inhibiting the protective enzymes, and using the active agents on the cell wall to increase the absorption of other antimicrobials [102]. In addition, volatile compounds derived from various medicinal plants possess hydroxyl functions (-OH) in their structure, which potentiate the antibacterial properties of terpenes [103].

0.625

0.5

0.5625

Partial synergy

Synergy

Partial synergy

0.5

0.125

0.25

0.25

0.0625

0.5

Furthermore, the presence and position of functional groups in EO compounds may effectively modulate its antibacterial effect [66, 104]. In fact, in Gram bacteria, the presence of phenolic groups in carvacrol and thymol have been shown to interact with the outer membrane constituents, causing its breakdown and thereby leading to the liberation of LPS and increasing the membrane permeability with significant loss of ATP [105]. Moreover, some components such as carvone, which have a hydroxyl group (in position 3), appear to be responsible for its interaction with the bacterial wall, causing significant injury, especially in

Gram+ bacteria [106]. Furthermore, it has been demonstrated that the antibacterial activity of terpene aldehydes is related to the electronegative characteristics of aldehyde group [107]. In fact, aldehydes can act on bacterial cell wall, restricting its biological functionality, especially electron transfer.

#### 4. Conclusion

In this exploratory investigation, we reported the antibacterial formulation of three EOs derived from Eucalyptus polybractea cryptonifera, Ormenis mixta, and Lavandula burnatii briquet using checkerboard and mixture design approaches. The antibacterial action of these EOs depends on the proportion of each constituent and the target bacteria. As a result, it has been shown that the MIC values were considerably reduced using the combination of E. polybractea, O. mixta, and L. burnatii. These effects are mainly ascribed to the synergistic action of the major and/or minor molecules identified in the combined EOs. The most effective combinations on E. coli and S. aureus correspond to 32%, 28%, and 40% and 35%, 30%, and 35% of E. polybractea, O. mixta, and L. briquet, respectively. These antibacterial formulations may be suitable as alternative to commercialized antibacterial and preservative agents, which are increasingly becoming nonactive against a panel of bacterial strains causing serious infections and undesirable deteriorations of some food-based products.

# **Data Availability**

All the data supporting the findings of this study are included in this article.

#### **Conflicts of Interest**

The authors declare that they have no conflicts of interest.

#### **Authors' Contributions**

Mohamed Jeddi conceptualized the study, wrote the original draft, curated the data, developed methodology, formally analyzed the study, and investigated the study. Naoufal El Hachlafi reviewed and edited the manuscript, developed methodology, formally analyzed the study, and wrote the original draft. Mouhcine Fadil developed methodology, investigated the study, and was responsible for software. Nesrine Benkhaira investigated the study, wrote the original draft, developed methodology, and formally analyzed the study. Samir Jeddi investigated the study and curated the data. Fikri-Benbrahim conceptualized the study, supervised the study, and reviewed and edited the manuscript.

## **Supplementary Materials**

Supplement file 1: Chromatogram of gas chromatography (GC) analysis of *Ormenis mixta* EO. Supplement file 2: Chromatogram of gas chromatography (GC) analysis of *Eucalyptus polybractea* EO. Supplement file 3: Chromatogram of gas chromatography (GC) analysis of *Lavandula burnatii* EO. (Supplementary Materials)

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