



Effect of green Fe₂O₃ nanoparticles in controlling *Fusarium* fruit rot disease of loquat in Pakistan

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Abstract

The subtropical fruit known as the loquat is prized for both its flavour and its health benefits. The perishable nature of loquat makes it vulnerable to several biotic and abiotic stressors. During the previous growing season (March–April 2021), loquat in Islamabad showed signs of fruit rot. Loquat fruits bearing fruit rot symptoms were collected, and the pathogen that was causing the disease isolated and identified using its morphology, microscopic visualisation, and rRNA sequence. The pathogen that was isolated was identified as *Fusarium oxysporum*. Green synthesized metallic iron oxide nanoparticles (Fe₂O₃ NPs) were employed to treat fruit rot disease. Iron oxide nanoparticles were synthesized using a leaf extract of the *Calotropis procera*. Characterization of NPs was performed by different modern techniques. Fourier transform infrared spectroscopy (FTIR) determined the existence of stabilizing and reducing compounds like phenol, carbonyl compounds, and nitro compounds, on the surface of Fe₂O₃ NPs. X-ray diffraction (XRD) explained the crystalline nature and average size (~49 nm) of Fe₂O₃ NPs. Energy dispersive X-ray (EDX) exhibited Fe and O peaks, and scanning electron microscopy (SEM) confirmed the smaller size and spherical shape of Fe₂O₃ NPs. Following both in vitro and in vivo approaches, the antifungal potential of Fe₂O₃ NPs was determined, at different concentrations. The results of both in vitro and in vivo analyses depicted that the maximum fungal growth inhibition was observed at concentration of 1.0 mg/mL of Fe₂O₃ NPs. Successful mycelial growth inhibition and significantly reduced disease incidence suggest the future application of Fe₂O₃ NPs as bio fungicides to control fruit rot disease of loquat.

Keywords *Fusarium oxysporum* · Fruit rot · Green synthesis · *Calotropis procera* · Fe₂O₃ NPs · NPs characterization

Introduction

Loquat (*Eriobotrya japonica* Lindl.) is a member of family Rosaceae, subfamily Pomoideae, and it is mostly cultivated in subtropical areas [1]. Loquat tree has medicinal and ornamental importance. It is used to treat respiratory disorders, infection, and cancer [2, 3]. Major loquat producing countries are China, Pakistan, Mauritius Island, Japan, Reunion Island, India, and Mediterranean

countries [4–6]. Approximate fruit production of loquat is 549,220 tonnes, worldwide. China is the major fruit producing country followed by Spain, Japan, India, and Pakistan [7, 8]. In Pakistan, loquat is grown in Murree, Hasan Abdal, Mardan, Tret, Kalar Kahar, Wah, Haripur, Sargodha, Choa Saidu Shah, Kasur, and Chhattar [9]. In Pakistan, Punjab and Khyber Pakhtunkhwa are the major loquat producing provinces [10].

Fungi are major biotic factors to causes 50% yield losses in fruit crops. Approximately one third of all food crops are destroyed by fungal pathogens [11]. About 248 species of fungi have been reported to cause diseases of *E. japonica* [12]. *Colletotrichum gloeosporioides* and *Colletotrichum truncatum* have been reported to cause pre-harvest and post-harvest diseases of loquat [13, 14]. Previous studies have described fungal diseases of loquat including anthracnose [15], fruit rot by *Colletotrichum* species [16], fruit rot by *Lasiodiplodia theobromae* [17, 18], circular brown leaf

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spot [19], *Fusarium* fruit rot [17], and scab by *Fusicladium eriobotryae* [20–22]. In Pakistan, few major fungi including *Diplodia seriata* [23], *Fusarium solani* [17], *C. gloeosporioides* [24], *Curvularia lunata* [25], and *Alternaria mali* [26] have been reported to cause diseases on different parts of loquat tree.

Use of chemical fungicide is very common in different parts of the world [27]. In Pakistan, approximately 108 types of insecticides are being used [28]. Fungicides have possible side effects, and they have been reported to affect human health [29]. Due to these health risks, scientists are working on environment-friendly disease control technologies for controlling post-harvest fruit disease [30, 31]. Nowadays, there is too much attention for alternatives of chemical fungicides and development of biological control measures against plant diseases [32]. Due to their biological origin, bio fungicides have minimal toxicity and increased resistance against diseases, and they are cheaper [33].

Nanotechnology is inspiring weapon against many factors that threaten plant health [34]. Green synthesis of nanoparticles involving the use of plants to synthesize metallic nanoparticles is safe and has environment friendly approach. To synthesize nanoparticles, different plant parts such as stem, seeds, leaves, roots, and fruits are being used. All these parts contain phytochemicals that have stabilizing and reducing abilities [35]. The aqueous leaf extract of *C. procera* has been used to synthesize nanoparticles. Antimicrobial potential of *C. procera* is well-documented [36]. To overcome drawbacks of classical hazardous disease-control methods, use of plants to synthesize nanoparticles is a novel approach, as the phytochemicals of plants act as capping and reducing agents [37]. Efficient application of Fe₂O₃ NPs in controlling fruit rot diseases has been reported earlier [38]. Iron oxide nanoparticles synthesized in aqueous leaf extract of *Calotropis procera* has been reported to control fruit rot of cherry [39]. Iron oxide NPs have also diminished the rot of apple [40].

The goal of this study was to synthesize iron oxide nanoparticles using the aqueous leaf extract of *C. procera* that could be used to treat post-harvest pathogen of loquat.

Materials and methods

Collection of diseased fruit

During the months of March–April 2021, field surveys were conducted, and diseased loquat samples were collected from the orchards of Quaid-i-Azam University Islamabad. Collected fruits were transferred to the lab in sterile polythene bags for further analyses [39].

Isolation of pathogen

Loquat fruit surface sterilization was done with 1% sodium hypochlorite solution (1 mL of sodium hypochloride was dissolved in 99 mL of distilled water) and then washed with distilled water for 2 min before isolation of causative agent. With sterilized blade, segments of diseased fruits were cut from edges and put on potato dextrose agar (PDA) media plates. Petri plates were sealed using parafilm and were kept in an incubator at 25 ± 2 °C. After 5 to 7 days, fungus morphology was observed by observing the colony growth, color, and its pattern from front and backside of the plate [40].

In vivo pathogenicity test

Pathogenicity was confirmed by following Koch's postulates. Mycelia of 7-day-old fungus were transferred to Czapek broth media and placed in shaking incubator at 25 ± 2 °C. After achieving desired concentration of conidia suspension (10⁶ conidia/ mL), broth culture was filtered to remove mycelia. To infect healthy loquat fruit, wounded sterile needle was used, and 5 µL conidial suspension was injected into three randomly selected fruit. Control fruits were injected with sterile distilled water. To protect all treated fruits, muslin cloth was used and kept in an incubator at 25 ± 2 °C. After 1 week, symptoms of disease with earlier field symptoms were compared. Pathogen was re-isolated by culturing on PDA media for 5–7 days at 25 ± 2 °C [39].

Microscopic identification

For identification, hyphae along with reproductive structures of isolated fungus were analyzed under light microscope. Slides were prepared using lactophenol blue [41]. Drop of lactophenol blue was placed in the center of slide, and growing mycelium was mounted on slides. By avoiding air bubbles, cover slip was placed, and slide was observed under microscope at ×100 magnification to examine spores, hyphae, and other pathogenic characteristics [42].

Molecular characterization and phylogenetic analysis

For genetic identification, rDNA of isolated fungus was amplified using specific small subunit ribosomal RNA primers. CTAB method was used for the extraction of fungal DNA [43]. The fungal mycelia (50 mg) were scraped from 10-day-old PDA cultures, manually ground in 1.5 mL of microfuge tubes with micro pestle adding 500 µL of pre-warmed (60 °C) TES lysis buffer (100 mM Tris pH 8.0; 10 mM EDTA pH 8.0; 2% SDS) of proteinase K

that was added to the ground material, incubated in 60 °C for 60 min. One hundred forty microliter of 5M NaCl and 64 µL of 10% (w/v) of CTAB were added to the suspension incubated at 65 °C for 10 min. DNAs were extracted by adding equal volume of chloroform: isoamylalcohol (24:1) centrifuged at 14000 × g/10 min. DNA was precipitated by adding 0.6 volume of cold isopropanol and 0.1 volume of 3M sodium acetate pH 5.2, centrifuged, and washed twice with 70% ethanol suspended in 100 µL of TE (10 mM Tris pH 8.0; 1 mM EDTA pH 8.0). RNA was digested by adding 10 mg/mL of RNase A and incubating at 37 °C for 45 min and stored at –20 °C for further use. For the amplification of rRNA, in polymerase chain reaction (PCR), ITS1 forward (TCCGTAGGTGAACCTGCGG) and ITS4 reverse (TCCTCCGCTTATTGATATGC) primers were used [44]. Reaction mixture contains 0.5 µL of dNTPs, 1.5 µL of *Taq* DNA polymerase, 5 µL of 10× polymerase buffer, 1 µL of genomic DNA, and 1 µL of each primer. PCR reaction temperature was 94 °C for 4 min, followed by 35 cycles of 94 °C for 40 s, 58 °C for 40 s, and 72 °C for 40 s. Sequencing of amplified PCR product was done and subjected to BLAST analysis on NCBI database. MEGA 7 with maximum likelihood method with 1000 bootstrap was used for phylogenetic analysis [45].

Preparation of leaf extract

To prepare extract, fresh leaves of *Calotropis procera* were washed and shade dried for 1 week. Leaves were ground into fine powder. *C. procera* extract was prepared by mixing of 30 g of powder into 300 mL of de-ionized water and boiled for 5 to 10 min. Solution was warmed up in water bath at temperature of 80 °C for 30 min. Extract was permissible to cool down and then filtered through muslin cloth. Further filtration was performed using Whatman filter paper and kept at 4 °C for storage [39].

Calotropis procera-mediated synthesis of iron nanoparticles

For the preparation of Fe₂O₃ NPs, 1 mM FeCl₃ solution was prepared and added into *C. procera* extract in 1:1 ratio. Heating temperature of hot plate for blend was 80 °C for 120 min, until the color was changed. Color change was the indication of reduction process of Fe₂O₃ NPs. Mixture was centrifuged for 10 min at 6000 rpm, and the pellet was washed with de-ionized water. Pellet was dried in an incubator for 3 to 4 h at 100 °C. Crystalline form of Fe₂O₃ NPs was obtained by calcination at 500 °C for 2 to 3 h. The synthesized nanoparticles were characterized before their antifungal activity analyses [39].

Characterization of nanoparticles

Fe₂O₃ NPs were characterized by the different laboratory techniques [46].

FTIR spectroscopy

Fourier transform infrared spectroscopy (FTIR) was used to determine the nature and types of functional groups associated with iron nanoparticles, in the range of 400 to 4000 cm⁻¹ with resolution power of 4cm⁻¹⁰. KBr pellet method was adopted to prepare sample by enveloping 10 mg of nanoparticles in 100 mg of KBr pellet.

XRD

To understand the nature and size of nanoparticles, X-ray diffraction (XRD) spectroscopy was used. Size of synthesized Fe₂O₃ NPs was assessed by the following Scherrer equation:

$$D = K\lambda/\beta \cos \theta$$

D is the crystalline size, shape factor is *K*, λ is the wavelength of X-ray, β is the full width at half-maximum of radians, and θ is angle of diffraction.

SEM-EDX characterization

Scanning electron microscopy (SEM) was used to see the morphology of NPs. There was use of energy dispersive X-ray spectroscopy (EDX) to reveal elemental composition of Fe₂O₃ NPs. Fe and O peaks were observed in X-ray spectra.

In vitro antifungal assay of Fe₂O₃ NPs

Fe₂O₃ NP fungicidal activity against *F. oxysporum* was tested by poisoned food technique [47]. Five concentrations of Fe₂O₃ NPs (0.1, 0.25, 0.50, 0.75, and 1 mg/ml) were mixed with PDA media. Fungal discs (5 mm) were positioned in the middle of each Petri plate. One Petri plate with no Fe₂O₃ NPs was used as a control. Plates were kept in an incubator at 25 °C. Percentage inhibitions and diameters of fungal mycelia were measured at regular intervals in 7 days. Following formula was used to calculate inhibition percentage of fungal growth:

$$\text{Growth inhibition percentage} = (C - T)/C \times 100$$

where C is the growth in the control and T is the sample mycelial growth [36].

In vivo antifungal activity analysis of Fe₂O₃ NPs

For the control of disease on loquat fruit, Fe₂O₃ NPs were used, following “detached fruit inoculation method” [48]. For this purpose, 18 fruits were surface sterilized, wounded with needle, and inoculated with fungus. After inoculation of fungus, Fe₂O₃ NPs were applied in five different concentrations (0.1, 0.25, 0.50, 0.75, and 1.0 mg/mL). Each concentration was applied on three inoculated fruit, and all samples were kept at 25 °C in an incubator. Lesions on fruit were observed, and area of lesion was measured in regular intervals of 72 h of treatment.

Statistical analysis

Experimental data was statistically analyzed by following one-way ANOVA. The significance of difference between treatments was tested by least significance difference (LSD) post hoc test ($\alpha = 0.05$). All data expressed as mean \pm standard error.

Results

Morphological and microscopic identification of pathogen

Diseased samples were collected (Fig. 1a), and after isolation from diseased fruit, the color of aerial mycelium was faded white, which later changed to purplish blue color

(Fig. 1b). Mycelial color, on the backside of petri plates, was light yellow (Fig. 1c). Macro conidial septation was observed under microscope (Fig. 1d).

Abundant chlamydozoospores were produced. These all-surface pattern and microscopic examinations demonstrated this pathogen as *F. oxysporum* [17].

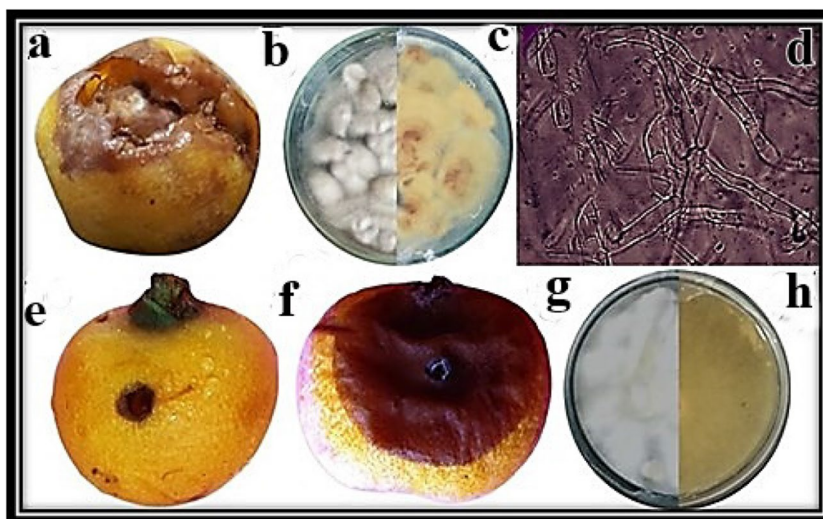
In vivo pathogenicity test

Koch’s postulates validated pathogenicity of isolated *F. oxysporum*. After 3 days of inoculation of healthy fruit, light brown disease circles were observed (Fig. 1e), which enlarged in size, later (Fig. 1f). These disease symptoms were similar to the symptoms of initially collected diseased fruit. Fungus was re-isolated and showed similar growth pattern to inoculated fungus (Fig. 1g, h). Based on these findings, *F. oxysporum* was demonstrated as causal agent of loquat fruit rot disease.

Molecular identification and phylogenetic analysis of isolated fungus

Isolated fungus sequence was 100% similar with *F. oxysporum* strain FO_99 (Accession no. MT649536.1). The sequence was submitted in NCBI database with accession number OL744437. Maximum likelihood method was successfully used to construct phylogenetic tree, which showed that the presence of obtained sequence is the same clade with *F. oxysporum* (Fig. 2). It also confirmed the evolutionary relationship of isolated fungus with *F. oxysporum*.

Fig. 1 Fruit rot symptoms were observed on loquat fruit (a). Disease-causing pathogen was isolated and observed from the front side (b) and back side of the Petri plates (c). Microscopic observation of fungus was performed at $\times 40$ magnification (d). Fungus was re-inoculated on healthy fruit, and disease symptoms were observed after 3 days (e) and 5 days post inoculation (f). Pathogen was re-isolated on PDA and observed from front side (g) and back side (h)



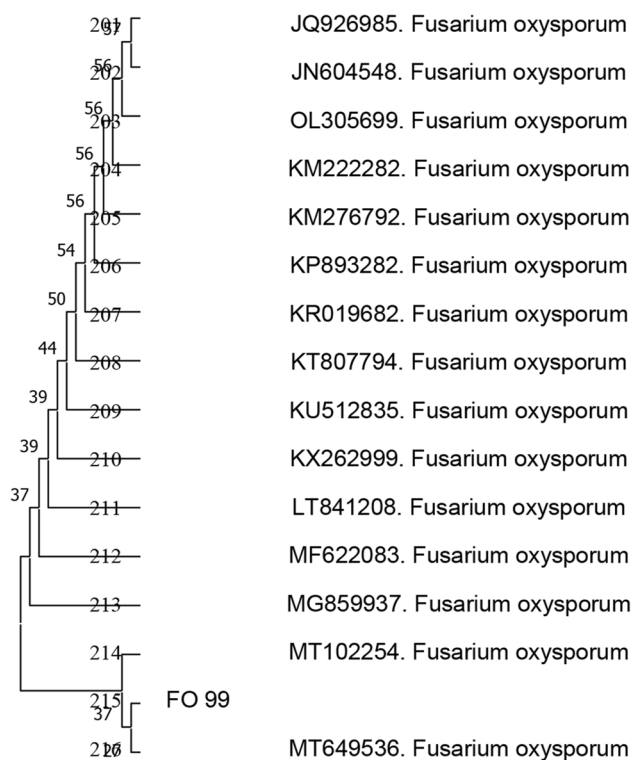


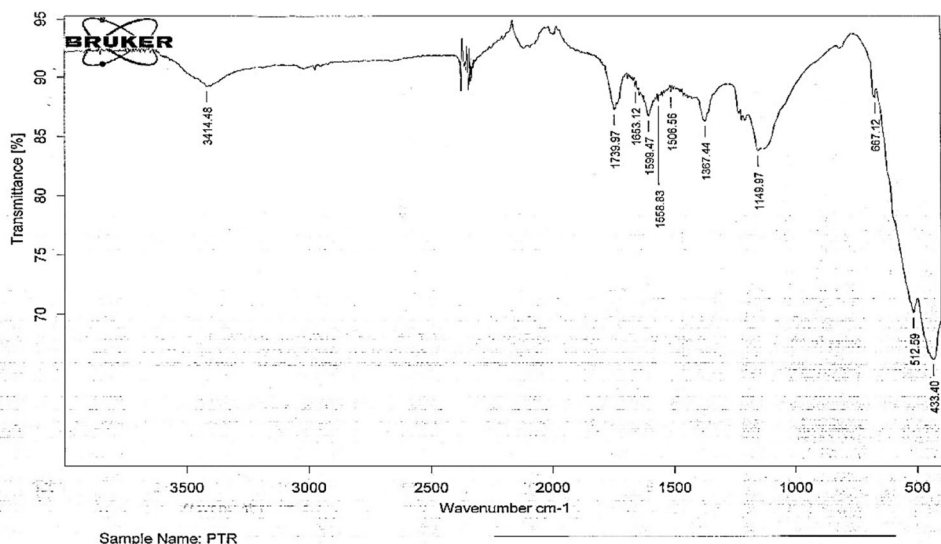
Fig. 2 Phylogenetic analysis of isolated pathogen with 15 related gene sequences from GenBank

Characterization of green synthesized Fe₂O₃ NPs

FTIR

FTIR spectra of Fe₂O₃ NPs showed different absorption band (Fig. 3). Absorption peak at 3414.48 cm⁻¹ represented O-H stretching. Absorption peak at 1739.97 cm⁻¹ showed C=O stretch of aldehyde group. Stretching

Fig. 3 FTIR spectrum of Fe₂O₃ NPs synthesized in the leaf extract of *C. procera*



vibrations at 1653.12 cm⁻¹ described alkene group while 1599.47 cm⁻¹ attributed polyphenols [49]. Stretching vibrations at 1558.83 cm⁻¹ and 1506.56 cm⁻¹ demonstrated nitro compounds (N-O). Peak at 1367.44 cm⁻¹ signified S=O stretching of sulfonamide group. Peak at 1149.97 cm⁻¹ indicated C-O stretching of aliphatic ether. Stretching vibrations of Fe-Ofe were observed at 667 cm⁻¹, 512 cm⁻¹, and 433 cm⁻¹ [14].

XRD

Crystalline structure of Fe₂O₃ NPs was described by XRD spectroscopy (Fig. 4). XRD pattern showed five notable peaks at 2θ angle between 10⁰ and 80⁰. Peaks were observed at 11⁰, 15.8⁰, 20.5⁰, 30.7⁰, and 43⁰. The average size of nanoparticle was observed to be 49 nm. Sharp peaks revealed crystalline nature of Fe₂O₃ NPs. Crystalline nature of NPs is inclined by plant extract [43]. Other than these peaks, there were no other detectable peaks, which indicated the purity of sample. XRD pattern of Fe₂O₃ NPs was in accordance with previous studies [7].

SEM and EDX analyses

SEM analysis explored the surface of synthesized NPs (Fig. 5). From the micrographs, spherical shape of Fe₂O₃ NPs was observed. These NPs were present in a size range of 40 to 50 nm and confirmed the size of NPs, estimated from XRD spectra. Fe₂O₃ NPs were spherical in shape. Peaks of Fe and O in EDX confirmed the formation of NPs. Weight percentage of chlorine, iron, and oxygen was 45.67%, 43.06%, and 20.18%, respectively (Fig. 6).

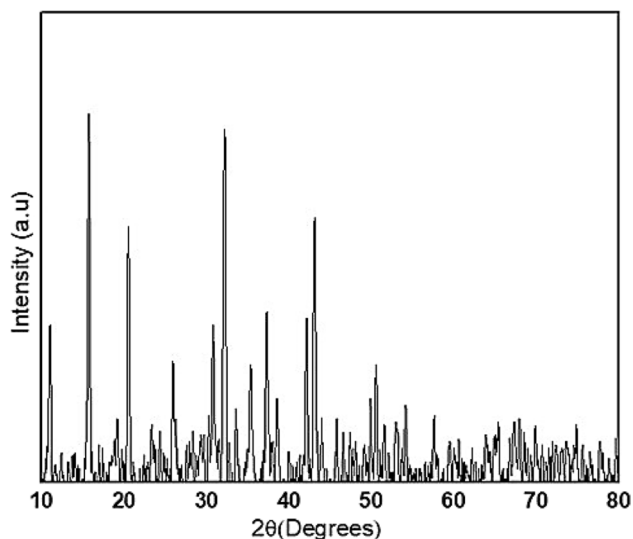


Fig. 4 XRD pattern of Fe_2O_3 NPs synthesized in the leaf extract of *C. procera*

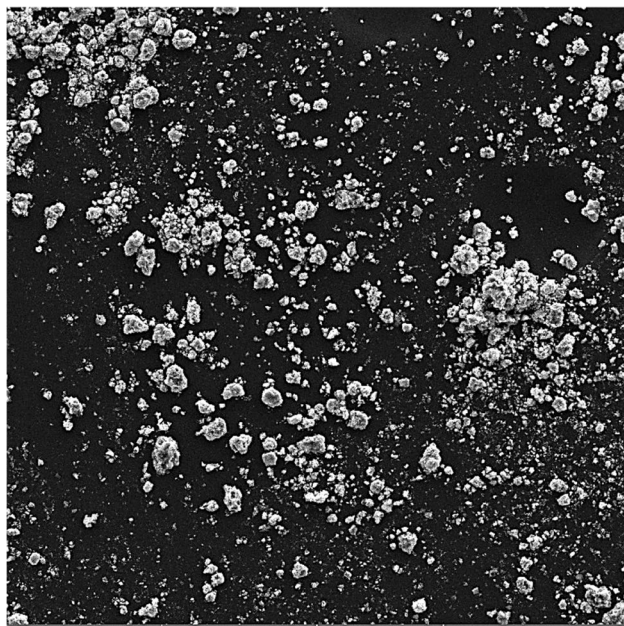


Fig. 5 SEM of Fe_2O_3 NPs synthesized in the leaf extract of *C. procera*

Antifungal activity of Fe_2O_3 NPs, in vitro

NPs of Fe_2O_3 showed maximum mycelial growth suppression (72%) at the concentration of 1 mg/mL. While growth inhibition at 0.75, 0.50, 0.25, and 0.1 mg/mL was 64.44%, 60%, 52%, and 37% (Table 1) (Fig. 7).

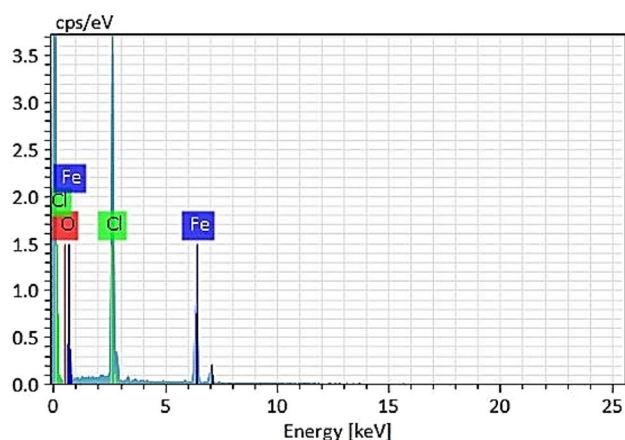


Fig. 6 Energy dispersive X-ray (EDX) spectroscopy of Fe_2O_3 NPs revealed the presence of Fe, Cl, and O

Table 1 Mycelial growth inhibition at different concentration of Fe_2O_3 NPs

Concentration mg/mL	Inhibition percentage
1.0	72.22 ± 1.0
0.75	64.44 ± 1.0
0.50	60.77 ± 0.5
0.25	52.22 ± 0.8
0.1	37.77 ± 1.2

Antifungal activity of Fe_2O_3 NPs, in vivo

In vivo antifungal activity analysis revealed the best disease control at concentration of 1.0 mg/mL of Fe_2O_3 NPs. All concentration of Fe_2O_3 NPs exhibited variable disease control (Fig. 8).

Discussion

Most frequent disease of loquat and other perishable fruits is fruit rot [15]. This work includes a thorough examination of the mycobiota linked to the fruit rot of loquat in Islamabad. To obtain pure colonies, the pathogen was isolated from rotten fruit segments and cultured on Petri plates. Utilizing morphological and molecular traits, fungal isolates were identified. Isolated fungus from loquat was identified as *F. oxysporum*. Microscopic observation revealed similar morphology, and macro conidial septation has been described earlier [50].

To control plant diseases, biocontrol efforts are being done for last few decades. Nanomaterials have gained popularity in the modern period as a result of their unique features and important applications in the food and agricultural industries [51]. The

Fig. 7 Inhibitory effect of Fe_2O_3 NPs on *F. oxysporum* using five different concentrations of NPs. Fungus growth was observed in control (a) and at five concentrations of NPs including 0.1 mg/mL (b), 0.25 mg/mL (c), 0.50 mg/mL (d), 0.75 mg/mL (e), and 1.0 mg/mL (f)

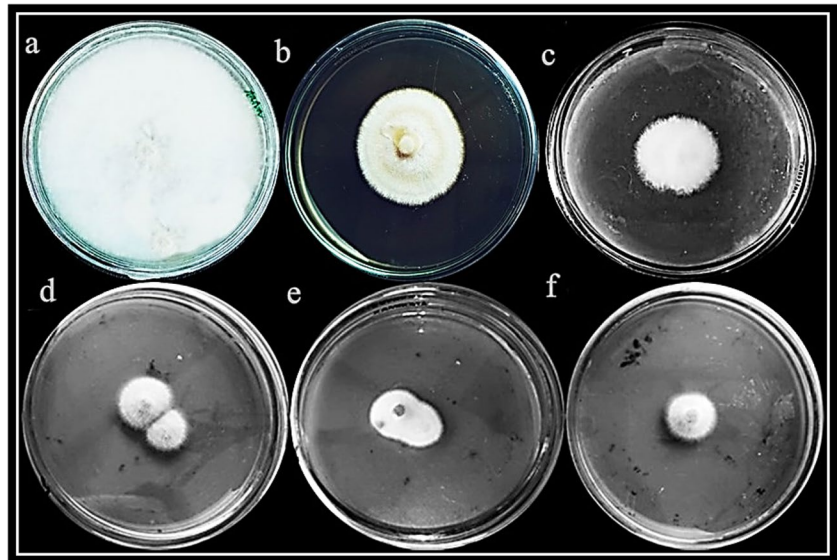
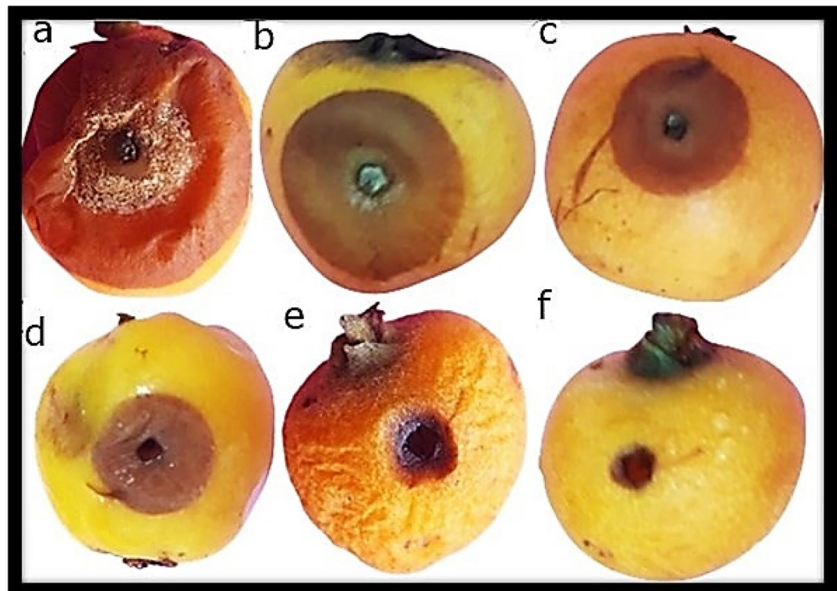


Fig. 8 Control of fruit rot disease using Fe_2O_3 NPs on loquat. Control fruit exhibited maximum disease (a). Variable fruit rot disease was observed at 0.1 mg/mL (b), 0.25 mg/mL (c), 0.50 mg/mL (d), 0.75 mg/mL (e), and 1.0 mg/mL (f) concentration of Fe_2O_3 NPs



manufacturing of nanoparticles has advanced greatly, and a wide range of biological agents, including bacteria and plants, are being exploited [52]. Use of plants to synthesize metallic nanoparticles is a more simple method, and many plant metabolites act as stabilizing and reducing agents [49].

In this study, *C. procera* was successfully used to synthesize iron oxide nanoparticles. Antifungal potential of *C. procera*-mediated NPs has been described earlier [39]. With the aid of FTIR, XRD, SEM, and EDX, iron oxide nanoparticle characterization was performed. Different functional groups were identified by FTIR spectrum. These functional groups consist of bioactive molecules. While reviewing FTIR spectra, amino groups were

present, and they show the presence of proteins all around iron oxide NPs. Results affirmed that protein molecules are present in the extract, and they are able to function as stabilizing and reducing agent by attachment to iron oxide NPs, with the aid of free primary amino group [53]. Carboxylic groups help to bind on Fe surface [54]. XRD affirmed crystal nature and smaller size of NPs. Previous research has demonstrated that small size and crystalline structure of NPs provide them strong antimicrobial capabilities [40, 46].

SEM images revealed spherical shape of iron oxide NPs. Additionally, high surface area to volume ratio was observed that depicted that nanoparticles have a propensity to aggregate in suspension [55]. An EDX study

verified the elemental composition of the iron oxide nanoparticles' elemental makeup. Iron oxide nanoparticles' EDX spectrum pattern revealed narrow peaks that confirmed the crystalline structure of the iron nanoparticles [56]. The EDX spectra revealed the existence of iron peaks in two distinct regions. The culture filtrate may be responsible for the existence of additional organic substances like O and C [57]. Iron oxide NPs were synthesized using a precursor solution that contained the element chlorine [58].

In this study, different concentrations of iron oxide nanoparticles synthesized using *C. procera* (0.1, 0.25, 0.5, 0.75, and 1 mg/mg) were utilized to both in vitro and in vivo test to suppress the fungus that causes fruit rot disease in loquat. It was found that the antifungal activity increased with increasing NP concentration. NP application results in membrane disability of fungus, by producing active oxygen radical suppression of transporter genes and genotoxicity [59]. NPs affect morphological properties of fungal membrane like depolarization of membrane and permeability. This breakdown causes leakage of different substances like enzymes, proteins, and DNA that result in cell death. Furthermore, NPs penetrate through the holes of cell wall of microorganisms [60]. NPs also cause lipid peroxidation and depletion fungal membrane ergosterol contents [61]. Higher ratios of surface-to-volume allow NPs to adhere the fungal cell surface, penetrate directly into cell, and cause damage of cell wall [62]. In the past, it has been reported that mycosynthesized iron oxide nanoparticles can prevent apple brown rot [46]. Green iron oxide nanoparticles prevented cherry fruit rot [40]. Citrus brown rot was prevented by bio-fabricated iron oxide nanoparticles [39].

This study has described a simple, predictable, and ecologically secure technique of preventing fruit rot disease of loquat by using green iron oxide nanoparticles. It has been concluded that the synthesized Fe₂O₃ NPs show effective antifungal properties. The outcomes showed that rot disease loquat could be effectively controlled by using the optimal concentration of Fe₂O₃ NPs (1.0 mg/ml).

Author contribution F.N. conducted all experiments, collected data, and wrote up the manuscript; M.A. compiled and organized the data; U.H. supervised the isolation of strains; F. and A.K. helped in the characterization of nanoparticles, T.R. and F.A. assisted in antifungal activity analyses, R.N. assisted in field experiments, H.J.C. supervised the write-up, and M.F.H.M. designed and supervised the whole study. All authors have read and agreed to the published version of the manuscript.

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Code availability Not applicable

Declarations

Ethics approval and consent to participate Not applicable

Consent for publication Not applicable

Conflict of interest The authors declare no competing interests.

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