

Communication

# Simultaneous Transport of CO<sub>2</sub> and HCO<sub>3</sub><sup>-</sup> by the Cyanobacterium *Synechococcus* UTEX 625<sup>1</sup>

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## ABSTRACT

A mass spectrometer was used to simultaneously follow the time course of photosynthetic O<sub>2</sub> evolution and CO<sub>2</sub> depletion of the medium by cells of the cyanobacterium *Synechococcus leopoliensis* UTEX 625. Analysis of the data indicated that both CO<sub>2</sub> and HCO<sub>3</sub><sup>-</sup> were simultaneously and continuously transported by the cells as a source of substrate for photosynthesis. Initiation of HCO<sub>3</sub><sup>-</sup> transport by Na<sup>+</sup> addition had no effect on ongoing CO<sub>2</sub> transport. This result is interpreted to indicate that the CO<sub>2</sub> and HCO<sub>3</sub><sup>-</sup> transport systems are separate and distinctly different transport systems. Measurement of CO<sub>2</sub>-dependent photosynthesis indicated that CO<sub>2</sub> uptake involved active transport and that diffusion played only a minor role in CO<sub>2</sub> acquisition in cyanobacteria.

Cyanobacteria such as *Anabaena variabilis* (1, 10, 20), *Coccolithis penicystis*, (7, 12, 13), *Synechococcus* sp. (4), and *Synechococcus leopoliensis* UTEX 625 (7, 8, 14, 16) possess mechanisms for the active transport of both CO<sub>2</sub> and HCO<sub>3</sub><sup>-</sup>. The ability to experimentally distinguish between the two transport processes, however, is complicated by the fact that CO<sub>2</sub> and HCO<sub>3</sub><sup>-</sup> are readily interconvertible. To circumvent this problem many of the assay procedures for CO<sub>2</sub> and HCO<sub>3</sub><sup>-</sup> transport (1, 4, 7, 12, 13, 19) involve transient measurement of activity and are conducted under experimental conditions which favor transport of the DIC<sup>2</sup> species of interest. These protocols, however, generally preclude the simultaneous measurement of the rate of transport of the other species of DIC. Consequently, there is uncertainty as to the relative contribution of each species of DIC to total DIC uptake and the extent to which each transport system functions *in situ*. In the present communication, we demonstrate that CO<sub>2</sub> and HCO<sub>3</sub><sup>-</sup> are simultaneously and continuously taken up during steady state photosynthesis by the cyanobacterium *Synechococcus* UTEX 625. Implications regarding the mechanism of transport are briefly explored.

## MATERIALS AND METHODS

*Synechococcus* UTEX 625 (*Synechococcus leopoliensis* University of Texas Culture Collection), also known as *Anacystis*

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<sup>2</sup> Abbreviations: DIC, dissolved inorganic carbon (CO<sub>2</sub> + HCO<sub>3</sub><sup>-</sup> + CO<sub>3</sub><sup>2-</sup>); BTP, 1,3-bis[tris(hydroxymethyl)methylamino]propane; CA, carbonic anhydrase; Rubisco: D-ribulose 1,5-bisphosphate carboxylase/oxygenase (EC 4.1.1.39).

*nidulans* (17, 18), was grown on air (0.05% v/v CO<sub>2</sub>) in unbuffered Allen's medium (2) as described previously (8). Cells were harvested for experiments when the [Chl] reached 6 to 9 μg/ml. At harvest, the pH of the culture medium was between 9.5 and 10.2. Prior to experiments, the cells were washed three times by centrifugation (1.0 min at 10,000g, Beckman Microfuge B) in 25.0 mM BTP/23.5 mM HCl buffer (pH 8.0), which contained only contaminant levels of DIC (15 μM) and Na<sup>+</sup> (5 μM) (17). Subsequently, 5 ml of cell suspension was placed in a thermostated (30°C) glass reaction vessel and allowed to equilibrate for 3 to 5 min in the dark. The vessel was closed to the atmosphere during experiments. Dissolved CO<sub>2</sub> and O<sub>2</sub> in the cell suspension were alternately measured at 3 s intervals using a magnetic sector mass spectrometer (VG Gas Analysis, Middlewich, England, model MM14-80 SC) equipped with a membrane inlet system (9). Data were stored on magnetic discs for later retrieval and analysis. Alternatively, when a single dissolved gas was measured, the output from the mass spectrometer was directed to a chart recorder. The mass spectrometer was calibrated for CO<sub>2</sub> by adding a known volume of 100 mM K<sub>2</sub>CO<sub>3</sub> to a darkened cell suspension. The CO<sub>2</sub> concentration generated (pH 8.0, 30°C) was calculated from the equations of Buch (5) and corrected for ionic strength ( $I = 0.025$  M) (15). The added K<sub>2</sub>CO<sub>3</sub> also served as the DIC source for photosynthesis. Calibration for dissolved O<sub>2</sub> was achieved using the catalase-H<sub>2</sub>O<sub>2</sub> procedure (6).

**Calculations.** Theoretical rates of photosynthesis based on the passive uptake of CO<sub>2</sub> alone were calculated using the equations described by Miller (11). For passive uptake it was assumed that the CO<sub>2</sub> concentration required to attain one-half the maximum rate of photosynthesis was equal to the  $K_m$ (CO<sub>2</sub>) of isolated cyanobacterial Rubisco (250 μM) (3). A measured value of 300 μmol O<sub>2</sub>·mg<sup>-1</sup> Chl·h<sup>-1</sup> was taken as the maximum rate of photosynthesis ( $V_{max}$ ). The apparent first order rate constant for photosynthesis,  $kp'$ , was calculated from the equation

$$kp' = V_{max}/K_m(\text{CO}_2) \quad (1)$$

For the analysis we varied  $K_m(\text{CO}_2)$  between 250 μM CO<sub>2</sub> and 1 nM CO<sub>2</sub>.

The maximum rate of CO<sub>2</sub> supply, occurring from HCO<sub>3</sub><sup>-</sup> dehydration, to cells in a closed system was calculated in two ways: method A from the equations of Miller and Colman (13) or method B from the equation

$$\frac{d[\text{CO}_2]}{dt} = k_d [\text{HCO}_3^-] \quad (2)$$

Use of Eq. 2 was justified since the actual measured level of CO<sub>2</sub> approached zero during experiments; hence the back reaction from CO<sub>2</sub> to HCO<sub>3</sub><sup>-</sup> was not a major factor in determining the CO<sub>2</sub> concentration. The value for  $k_d$  ( $0.87 \times 10^{-3}$  s<sup>-1</sup>) was determined empirically by using the mass spectrometer to follow

the time course of  $\text{CO}_2$  appearance in BTP/HCl reaction buffer (pH 8.0,  $30^\circ\text{C}$ ) of the appropriate ionic strength, following the addition of  $1\text{ mM K}_2\text{CO}_3$ . Eq. 2 yielded approximately a two-fold lower rate of  $\text{CO}_2$  supply than the equations of Miller and Colman (Fig. 1). The difference lies in the fact that ionic strength was taken into account in Eq. 2 but not in the other equations (13).

## RESULTS AND DISCUSSION

**Simultaneous Transport of  $\text{CO}_2$  and  $\text{HCO}_3^-$ .** Figure 1 illustrates a typical time course of  $\text{CO}_2$  removal from the medium (pH 8) by cells of *Synechococcus* following illumination in the presence of  $55\ \mu\text{M DIC}$ . The first phase of the time course lasted for about 40 s and involved rapid  $\text{CO}_2$  depletion of the medium, to a level near zero. In the second phase of the time course, this low level of  $\text{CO}_2$  was maintained constant until the cells were darkened. If CA, a catalyst for  $\text{CO}_2$ - $\text{HCO}_3^-$  interconversion, was added to the illuminated cell suspension shortly after attaining steady state  $\text{CO}_2$  uptake (phase 2), the extracellular  $[\text{CO}_2]$  rapidly returned to a level near that found prior to illumination (Fig. 2). These results show that, in the absence of CA, the rapid removal of  $\text{CO}_2$  from the medium by the cells caused the  $\text{CO}_2$ - $\text{HCO}_3^-$  system to be drawn far from chemical equilibrium. Second, these results also show that  $\text{CO}_2$ , rather than  $\text{CO}_2 + \text{HCO}_3^-$ , was selectively removed from the medium during the initial phase of the time course, since a return to a value close to the pre-illumination level of  $\text{CO}_2$  (Fig. 2) would not be expected if all the DIC ( $\text{CO}_2 + \text{HCO}_3^-$ ) was sequestered within the intracellular pool. Photosynthetic fixation may have accounted for some of the initial consumption of  $\text{CO}_2$  (Fig. 1). However, the brevity of the initial phase of uptake and the low rate of photosynthesis, in the absence of  $\text{Na}^+$  (Fig. 1), indicated that  $\text{CO}_2$  fixation would only

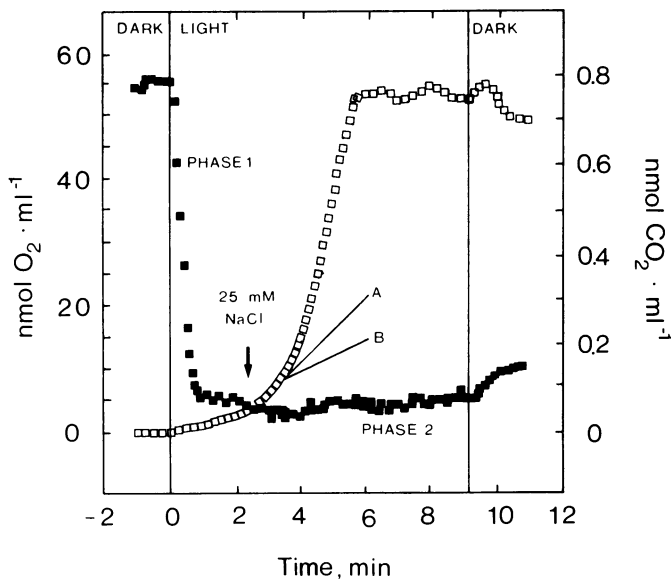


FIG. 1. Simultaneous measurement of photosynthetic  $\text{O}_2$  evolution ( $\square$ ) and  $\text{CO}_2$  depletion ( $\blacksquare$ ) of the medium by *Synechococcus* following illumination in the presence of  $55\ \mu\text{M DIC}$ . Cells ( $12.9\ \mu\text{g Chl}\cdot\text{ml}^{-1}$ ) were suspended in BTP/HCl buffer (pH 8.0), at  $30^\circ\text{C}$ , and light was provided at  $600\ \mu\text{E}\cdot\text{m}^{-2}\cdot\text{s}^{-1}$  PAR. Active transport of  $\text{HCO}_3^-$  was initiated by the addition of  $25\ \text{mM NaCl}$ . The rates of photosynthesis ( $\mu\text{mol O}_2\cdot\text{mg}^{-1}\cdot\text{Chl}\cdot\text{h}^{-1}$ ) prior to and after  $\text{Na}^+$  addition were 16 and 128, respectively. Parallel experiments (not shown) demonstrated that the increase in photosynthesis was, in fact, dependent upon  $\text{Na}^+$ . Solid lines (A, B) represent photosynthesis ( $\text{O}_2$  evolution) which could be supported by  $\text{CO}_2$  uptake alone, as calculated by method A or B.

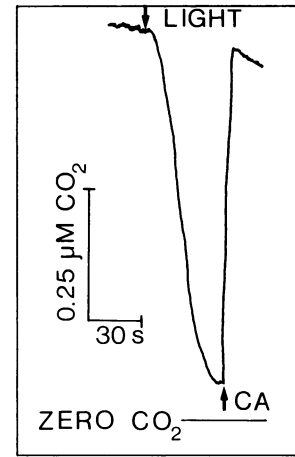


FIG. 2. Time course of  $\text{CO}_2$  depletion of the medium by *Synechococcus* following illumination in the presence of  $65\ \mu\text{M DIC}$ . Shortly after attaining steady state  $\text{CO}_2$  uptake (phase 2),  $25\ \mu\text{g}\cdot\text{ml}^{-1}$  CA ( $62.5$  Wilbur-Anderson units/ml) was added to the cell suspension. Cells ( $7.7\ \mu\text{g Chl}\cdot\text{ml}^{-1}$ ) were suspended in BTP/HCl buffer (pH 8.0), containing  $100\ \mu\text{M NaCl}$  at  $30^\circ\text{C}$ , and light was provided at  $600\ \mu\text{E}\cdot\text{m}^{-2}\cdot\text{s}^{-1}$  PAR.

be a minor factor in the disappearance of  $\text{CO}_2$  from the medium. In fact, we have obtained nearly identical time courses when the cells have been poisoned with iodoacetamide (15), an inhibitor of the Calvin cycle. Thus, the initial removal of  $\text{CO}_2$  from the medium by *Synechococcus* represents uptake of  $\text{CO}_2$  into the intracellular pool of inorganic carbon.

The near-zero, nonequilibrium, level of  $\text{CO}_2$  found during the second phase of the time course (Fig. 1) was maintained in spite of the fact that extracellular  $\text{HCO}_3^-$  dehydration served as a means for the regeneration of  $\text{CO}_2$  within the system. In light of this consideration, maintenance of a low nonequilibrium  $\text{CO}_2$  concentration must be interpreted as indicating continuous uptake of  $\text{CO}_2$  by the cells at a rate equal to the  $\text{HCO}_3^-$  dehydration rate. If  $\text{CO}_2$  uptake ceased after attaining phase 2, it would be expected that the  $\text{CO}_2$  concentration would return to its equilibrium level. This was clearly not the case (Fig. 1).

Figure 1 also shows the simultaneously recorded time course for  $\text{O}_2$  evolution following illumination of the cells. In the absence of  $\text{Na}^+$ , a rate of  $16\ \mu\text{mol O}_2\cdot\text{mg}^{-1}\cdot\text{Chl}\cdot\text{h}^{-1}$  was observed and this closely corresponded to the theoretical rate ( $13\ \mu\text{mol O}_2\cdot\text{mg}^{-1}\cdot\text{Chl}\cdot\text{h}^{-1}$ ) of  $\text{CO}_2$  supply to the system from  $\text{HCO}_3^-$  dehydration (method B). Addition of  $25\ \text{mM NaCl}$  to the cell suspension enhanced photosynthetic  $\text{O}_2$  evolution eightfold, but had no apparent effect on the rate (phase 2) of  $\text{CO}_2$  uptake. Control experiments demonstrated that the increase in photosynthesis was, in fact, a consequence of  $\text{Na}^+$  addition (not shown; see 8, 14, 17).

Previous studies (8, 14, 17) have shown that millimolar levels of  $\text{Na}^+$  stimulate the active transport of  $\text{HCO}_3^-$  in *Synechococcus*. The criterion that we used to establish  $\text{HCO}_3^-$  uptake was to compare the rate of photosynthesis to the maximum rate which could be supported by  $\text{CO}_2$  uptake alone (Fig. 1). In this experiment photosynthesis exceeded the  $\text{CO}_2$  supply rate calculated from method A, by 5.2-fold or by 9.8-fold as calculated from method B. Thus, the additional DIC required to support the increased rate of photosynthesis must be obtained from the medium through  $\text{HCO}_3^-$  uptake. The necessity for  $\text{HCO}_3^-$  uptake in support of photosynthesis is further emphasized by the fact that the  $\text{CO}_2$  concentration was near zero just prior to the addition of  $\text{Na}^+$  (Fig. 1) and thus cannot be the source of additional C for fixation. Collectively, the data in Figure 1 indicate that both  $\text{CO}_2$  and  $\text{HCO}_3^-$  were simultaneously and continuously taken up

by the cells during steady state photosynthesis at pH 8.

**Interaction between CO<sub>2</sub> and HCO<sub>3</sub><sup>-</sup> Uptake.** It is of interest to note that the initiation of HCO<sub>3</sub><sup>-</sup> transport by Na<sup>+</sup> addition had little effect on the concentration of CO<sub>2</sub> maintained in the medium during phase 2. It might be expected that any competitive effect of HCO<sub>3</sub><sup>-</sup> transport on CO<sub>2</sub> transport (20) would be manifested as an increase in the steady state (phase 2) level of CO<sub>2</sub> in solution. The fact that the initiation of HCO<sub>3</sub><sup>-</sup> transport did not alter the extracellular CO<sub>2</sub> concentration suggests that CO<sub>2</sub> and HCO<sub>3</sub><sup>-</sup> were taken up by distinctly different transport systems. This finding is in contrast to data obtained for *Anabaena variabilis* (20), where it was proposed that CO<sub>2</sub> and HCO<sub>3</sub><sup>-</sup> interacted at a common site and that the carrier systems for both species of DIC were linked (20). It is possible CO<sub>2</sub> transport was inhibited by HCO<sub>3</sub><sup>-</sup> uptake but that the cells possessed sufficient CO<sub>2</sub> transport capacity to maintain the extracellular CO<sub>2</sub> concentration near zero in spite of the inhibition. This explanation, however, was not sufficient since even under conditions where the rate of CO<sub>2</sub> transport was near maximum (high [DIC]) Na<sup>+</sup> addition (HCO<sub>3</sub><sup>-</sup> transport) had little effect on the steady state CO<sub>2</sub> concentration in the medium (not shown).

Experiments similar to that shown in Figure 1 have been conducted by Badger and Andrews (4) using a marine species of *Synechococcus*. In both cyanobacterial species, an initial rapid depletion of CO<sub>2</sub> from the medium occurred upon illumination. With the marine species, however, this was followed by a massive efflux of CO<sub>2</sub> from the cells rather than by the establishment of a low and constant extracellular CO<sub>2</sub> concentration (Fig. 1). Consequently, for the marine species, it is difficult to ascertain whether or not inwardly directed active CO<sub>2</sub> transport continued during the CO<sub>2</sub> efflux stage, although it was clear that HCO<sub>3</sub><sup>-</sup> transport was occurring (4).

**Active Transport of CO<sub>2</sub>.** The uptake of CO<sub>2</sub> alone, before the addition of 25 mM NaCl, was sufficient to support a low rate of photosynthesis (Fig. 1). However, the observed rate of photosynthesis was some 22-fold higher than expected based on the *in vitro* kinetic characteristics of cyanobacterial Rubisco and the ambient CO<sub>2</sub> concentration passively distributed between cells and the medium (Fig. 3). Although the rate of conversion of HCO<sub>3</sub><sup>-</sup> to CO<sub>2</sub> was just adequate to support the observed rate of photosynthesis (Fig. 1), this photosynthesis could only occur if the CO<sub>2</sub> taken up was accumulated within the cells to a level which far exceeded the extracellular concentration. This would require that CO<sub>2</sub> be actively transported across the plasma-membrane. In other work (15), we have used the mass spectrometer to measure uptake of CO<sub>2</sub> by *Synechococcus* against concentration gradients in excess of 17,000-fold. The present analysis provides independent evidence for the active transport of CO<sub>2</sub> and indicates that maintenance of even low rates of photosynthesis requires active accumulation of CO<sub>2</sub>. Consequently, the diffusive uptake of CO<sub>2</sub> plays a very minor role in the supply of DIC for photosynthesis in cyanobacteria.

**Relative Contribution of CO<sub>2</sub> and HCO<sub>3</sub><sup>-</sup> Uptake to Photosynthesis.** Maintenance of a near-zero concentration of CO<sub>2</sub> (Fig. 1) showed that the rate of CO<sub>2</sub> resupply to the system, from HCO<sub>3</sub><sup>-</sup> dehydration, occurred at or near its maximum theoretical rate. This observation in conjunction with the fact that the rate of photosynthesis (O<sub>2</sub>-evolution) closely corresponded to the theoretical rate of CO<sub>2</sub> supply, in the absence of Na<sup>+</sup>, indicated that little, if any, HCO<sub>3</sub><sup>-</sup> was taken up by the air-grown cells through an Na<sup>+</sup>-independent (8) mechanism. *Synechococcus* was also capable of reducing the CO<sub>2</sub> concentration to near zero at extracellular DIC concentrations up to 200 μM, both in the absence and presence of Na<sup>+</sup> (HCO<sub>3</sub><sup>-</sup> transport) (not shown). Although the actual extent of CO<sub>2</sub> depletion depended upon cell density, these results showed that under certain circumstances the cellular capacity for active CO<sub>2</sub> transport was sufficient to

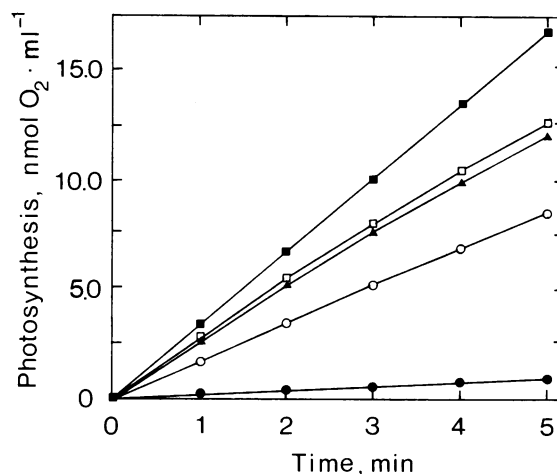


FIG. 3. Comparison between the observed, CO<sub>2</sub>-dependent, rate of photosynthesis (■) and that expected (●, ○, ▲, □) if Rubisco uses CO<sub>2</sub> from the medium without transport and subsequent intracellular accumulation. For the analysis, the  $K_m$  (CO<sub>2</sub>) of Rubisco was taken as 250 μM (●), its measured value (3), and it was assumed to be equal to the  $K_{1/2}$  (CO<sub>2</sub>) of photosynthesis. Alternatively, a value of 10 μM (○), 1 μM (▲), or 1 nM (□) was used. The  $V_{max}$  was held constant at 300 μmol O<sub>2</sub> · mg<sup>-1</sup> Chl · h<sup>-1</sup>. The CO<sub>2</sub>-dependent rate of photosynthesis, prior to Na<sup>+</sup> addition, was 3.4 nmol O<sub>2</sub> · ml<sup>-1</sup> · min<sup>-1</sup> (Fig. 1). Other conditions as in Figure 1.

support photosynthesis at rates close to the maximum rate of CO<sub>2</sub> supply (method B) at DIC concentrations up to saturation. Thus, the net contribution made by CO<sub>2</sub> uptake to the supply of DIC for photosynthesis could be defined.

Rates of photosynthesis in excess of the CO<sub>2</sub> supply rate require a direct contribution from HCO<sub>3</sub><sup>-</sup> uptake to provide the additional substrate for photosynthesis (Fig. 1). In this case (Fig. 1), after the addition of Na<sup>+</sup>, 12.5% of the DIC for photosynthetic fixation would have been supplied through CO<sub>2</sub> uptake while the remainder was supplied through HCO<sub>3</sub><sup>-</sup> uptake. The relative contribution of CO<sub>2</sub> or HCO<sub>3</sub><sup>-</sup> to photosynthesis would, of course, depend upon experimental conditions. At low levels of extracellular DIC (<100 μM) and/or at alkaline pH (>8.5), Na<sup>+</sup>-dependent HCO<sub>3</sub><sup>-</sup> uptake was the major means of DIC acquisition for photosynthesis in air-grown cells (8, 16). As the extracellular DIC concentration was increased, CO<sub>2</sub> uptake provided an increasing proportion of the total DIC for photosynthesis as governed by the rate of CO<sub>2</sub> supply and the kinetic characteristics of the transport system. At high DIC concentration (>500 μM, pH 8), where the rate of CO<sub>2</sub> supply is not limiting, it is theoretically possible that all of the CO<sub>2</sub> required for fixation would be acquired through CO<sub>2</sub> transport. Whether HCO<sub>3</sub><sup>-</sup> transport continues under these conditions is not known. A putative cessation of HCO<sub>3</sub><sup>-</sup> transport at high DIC concentration would require some means of regulating the transport system.

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