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# **The non-ribosomal peptide synthetase-independent siderophore (NIS) rhizobactin produced by** *Caballeronia mineralivorans* **PML1(12) confers the ability to weather minerals**

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**ABSTRACT** To mobilize nutrients entrapped into minerals and rocks, heterotrophic bacteria living in nutrient-poor environments have developed different mechanisms based mainly on acidolysis and chelation. However, the genetic bases of these mechanisms remain unidentified. To fill this gap, we considered the model strain *Caballeronia mineralivorans* PML1(12) known to be effective at weathering. Based on its transcriptomics and proteomics responses in Fe-depleted conditions, we pointed a cluster of genes differentially expressed and putatively involved in the production of siderophores. In this study, we report the characterization of this gene region coding for the production of a non-ribosomal peptide synthetase-independent siderophore (NIS). Targeted mutagenesis associated with functional assays and liquid chromatography coupled to high-resolution tandem mass spectrometry demonstrated the production of a single siderophore, identified as rhizobactin. This siderophore represents the first NIS containing malic acid in its structure. The evidence for the implication of rhizobactin in mineral weathering was demonstrated during a hematite dissolution assay. This study provides the first demonstration of the synthesis of a NIS in the genus *Caballeronia* and its involvement in mineral weathering. Our conclusions reinforce the idea that strain PML1(12) is particularly well adapted to nutrient-poor environments.

**IMPORTANCE** This work deciphers the molecular and genetic bases used by strain PML1(12) of *Caballeronia mineralivorans* to mobilize iron and weather minerals. Through the combination of bioinformatics, chemical, and phylogenetic analyses, we characterized the siderophore produced by strain PML1(12) and the related genes. This siderophore was identified as rhizobactin and classified as a non-ribosomal peptide synthetase-independent siderophore (NIS). Contrary to the previously identified NIS synthetases that form siderophores containing citric acid, α-ketoglutarate, or succinic acid, our analyses revealed that rhizobactin contains malic acid in its structure, representing, therefore, the first identified NIS with such an acid and probably a new NIS category. Last, this work demonstrates for the first time the effectiveness at weathering minerals of a siderophore of the NIS family. Our findings offer relevant information for different fields of research, such as environmental genomics, microbiology, chemistry, and soil sciences.

**KEYWORDS** bacteria, mineral weathering, NRPS-independent siderophore, rhizobactin, malic acid

I ron represents an essential micronutrient required for living organisms. This element is used as a cofactor by many enzymes and is involved in a wide range of physiological ron represents an essential micronutrient required for living organisms. This element is processes such as respiration, DNA synthesis, or photosynthesis [\(1\)](#page-17-0). While iron is the fourth most abundant element in the Earth crust, it is mainly found under insoluble forms [\(2,](#page-17-0) 3) in primary (*e.g.*, biotite, granite;  $Fe^{3+}/Fe^{2+}$ ) and secondary minerals (*e.g.*,

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hematite, goethite, and magnetite;  $Fe^{3+}$ ) [\(4\)](#page-17-0). In addition, iron is usually poorly bioavailable at neutral to basic pH and under oxygenic conditions. To deal with these limiting conditions and their physiological requirements, bacteria have developed a diversity of mechanisms to mobilize iron and other nutrients poorly available or entrapped into minerals and rocks, a global process termed mineral weathering (MWe).

In heterotrophic bacteria, two main MWe mechanisms have been evidenced: acidolysis and chelation [\(5\)](#page-17-0). The first mechanism is related to the action of the byproducts of the primary and secondary metabolisms of bacteria. Indeed, by consuming carbon substrates, bacteria acidify their local environment due to the production and accumulation of organic acids (*e.g*., citrate, gluconate) and protons. Such acidification allows the dissolution of the minerals and the release of their nutritive content. The effectiveness of this mechanism is clearly associated with the carbon substrates consumed, with glucose usually giving stronger acidification [\(6,](#page-17-0) 7). The second mechanism is based on the production of chelating compounds such as organic acids (e*.g.*, oxalate) and siderophores. Bacteria can weather minerals using one or both mechanisms (*i.e.*, acidolysis and chelation) depending on the bacterial strain, the nutrient availability, the type of mineral, and the ionic strength of the local environment (*i.e.,*  buffering capacity).

Many microorganisms, such as bacteria and fungi, can produce one or more type(s) of siderophore. These particular metabolites are low-molecular-weight compounds, characterized by a strong affinity for iron [\(8\)](#page-17-0), although they can also chelate other divalent and trivalent cations such as Al, Mg, or Cr [\(9,](#page-17-0) 10). Siderophores are produced when microorganisms experience nutrient limitation and especially iron deficiency [\(11\)](#page-17-0). In this sense, concentrations up to 2 and 12 nM of siderophores have been measured in the solution of nutrient-poor soils [\(12\)](#page-17-0). To date, two classes of siderophores have been identified: (i) the non-ribosomal peptide synthetase (NRPS) siderophores (*e.g*., pyoverdin) and (ii) the NRPS-independent siderophores (NIS) (*e.g*., aerobactin) [\(13\)](#page-17-0). The contribution of siderophores to MWe has been known for a long time for different types of molecules (*i.e.*, desferrioxamine) in abiotic experiments [\(14\)](#page-17-0). However, under such experimental conditions, the use of relatively high concentrations of siderophore did not allow to appreciate how microorganisms produced and regulated this production according to their local environment (*i.e*., pH, ionic strength, nutrient availability) [\(9,](#page-17-0) 15). Despite a constant increase in the number of new siderophores identified, their role in MWe has rarely been assessed in a cellular context [\(6,](#page-17-0) 16, 17). Indeed, few studies have considered how the MWe ability of a mutant impaired in siderophore production compares to that of the wild-type (WT) strain and how the nutrient limitations impact the action of the siderophore-mediated weathering.

In this study, we have focused on the model strain PML1(12) from the genus *Caballeronia mineralivorans*, which has been isolated from a nutrient-poor soil and is known to be effective at MWe and at promoting plant growth [\(7,](#page-17-0) 18). However, the molecular basis of its MWe ability is not fully elucidated. The dual transcriptomic and proteomic analyses recently applied on mineral/bacteria interactions highlighted the specific increased expression of a cluster of genes related to iron mobilization and the production of a potential siderophore under Fe-depleted conditions [\(7\)](#page-17-0). Among these genes, the most highly expressed presented high homology with a putative diaminopimelate decarboxylase, cysteine synthase, and IucA/IucC synthetase in conditions devoid of iron [\(7\)](#page-17-0). In this context, the objectives of this study were to (i) characterize the cluster of genes conferring the ability to mobilize iron of strain PML1(12); (ii) purify and characterize the siderophore produced by strain PML1(12); (iii) quantify the contribution of siderophore production on mineral weathering; and (iv) study the conservation of the genes involved in siderophore production in the *Burkholderia, Caballeronia, Paraburkholderia* (BCP) group. To do it, different bioinformatic tools were combined to characterize the cluster of genes related to the production of the NIS produced by strain PML1(12) and to determine the relative conservation of this cluster of genes in the genome of different strains from the BCP group. The use of a mutant impaired in its ability to

produce the siderophore allowed for the determination of its role in hematite weathering. Ultra-high-performance liquid chromatography coupled with high-resolution mass spectrometry allowed the identification of the chelating molecule.

## **RESULTS**

#### **A single genomic region is potentially involved in siderophore production**

The Chrome Azurol S (CAS) assay performed on strain PML1(12) demonstrated its ability to mobilize iron as stated by the formation of a yellow halo around the colony, suggesting the production of one or several siderophores. An antiSMASH analysis of the PML1(12) genome identified a single siderophore production locus. This region is characterized by 10 genes, among which a single gene encoding a protein of 635 amino acids, we named RhiE, presenting the characteristic signature of the *iucA/iucC* domain. Such domain is associated with the synthesis of enzymes involved in the production of NIS. The other genes present in the region upstream and downstream of the *rhiE*  gene could be involved in siderophore production (*rhiBCD*, intermediate production of the siderophore), transport (*rhiA*, TonB transporter; *rhiF*, MFS transporter; *rhiGHI*, ABC systems) and in the regulation of siderophore production (*rhiJ*). The detail of the functions encoded by each gene of the *rhiABCDEFGHIJ* cluster is presented in Table 1.

# **The** *rhiABCDEFGHIJ* **region confers the ability to mobilize iron to strain PML1(12)**

To investigate the implication of this region in siderophore biosynthesis, a *rhiE*::Gm knock-out mutant was constructed and tested on solid CAS assay to compare its siderophore production with that of the WT strain (Fig. S1). After 7 days of incubation, the WT strain presented a yellow halo of *ca*. 1.7 cm, while no halo was visible for the *rhiE*::Gm mutant strain. The lack of production of chelating molecules by the *rhiE*::Gm mutant strain was confirmed with the liquid CAS assay, which showed a very slight optical density (OD) measured at 655 nm (OD<sub>655nm</sub>) decrease from 0.48  $\pm$  0.01 to 0.43  $\pm$ 



**TABLE 1** Description of the NRPS-independent gene region involved in siderophore production by *Caballeronia mineralivorans* strain PML1(12)

*<sup>a</sup>*Each gene product of the cluster is presented by its NCBI accession number. The annotation is based on BLASTp homology. NIS synthetases showing the highest homology with RhiE of strain PML1(12) are shown.

0.01 between the non-inoculated treatment and the *rhiE*::Gm mutant, when the decrease reached a final value of 0.11  $\pm$  0.01 for the WT strain indicative of a trapping of iron from the CAS-Fe(III) complex.

# **The** *rhiABCDEFGHIJ* **cluster permits the growth of strain PML1(12) in iron-deficient condition**

The growth capabilities of the WT and *rhiE*::Gm strains were measured in a medium devoid of iron (ABm-Fe) amended or not with various concentrations of iron (0, 0.5, 1, 5, and 10 mg L<sup>-1</sup> of Fe) (Fig. 1). In absence of iron, the growth of the WT strain was delayed and reached a maximum OD<sub>595nm</sub> value of 0.58  $\pm$  0.01. The growth rate in absence of iron was significantly lower ( $P < 0.05$ ) than that observed in the presence of iron at 5 or 10 mg L−1 concentration (OD595nm = 0.88 ± 0.01). The *rhiE*::Gm mutant strain reached the same  $OD_{595nm}$  as the WT strain when growing at the highest iron concentrations (*i.e.*, 5 and 10 mg L<sup>-1</sup>), but with an important growth delay. The maximal OD<sub>595nm</sub> was observed at 50 h vs 26 h for the *rhiE*::Gm and WT strains, respectively. The largest growth differences (*i.e.*, maximal OD<sub>595nm</sub> and delay) between WT and *rhiE*::Gm strains were observed when iron concentrations were below 1 mg L−1 in the solution.

# **Inhibition of the siderophore production according to Fe concentration**

As the production of chelating molecules is usually dependent on Fe concentration in the medium, we assayed the production of siderophores in a range of concentrations of Fe from 0 to 5 mg L−1. After a 3-day incubation period in ABm-Fe amended or not with iron (*i.e.*, FeCl<sub>3</sub>), the analyses done on the culture supernatant revealed the lack of chelating activity in the non-inoculated controls and in the WT cultures in presence of iron concentrations at 0.8 mg L<sup>-1</sup> (Fig. 2), while this activity was detectable at 0.7 mg L<sup>-1</sup> and below. Chelating activity reached its maximum in absence of iron in the medium. The same experiment performed with the *rhiE*::Gm mutant did not reveal any chelating activity whatever the iron concentration tested.

# **Hematite weathering**

The ability of the chelating molecule(s) produced by strain PML1(12) to weather minerals was evaluated considering hematite, an iron-rich mineral, as the sole source of iron in our experimental conditions. To limit the impact of acidification on the mineral



**FIG 1** Growth assay. The growth of the WT strain (A) and *rhiE*::Gm mutant (B) was monitored in presence of different iron concentrations (*i.e.*, 0, 0.5, 1, 5, and 10 mg L<sup>-l</sup> of Fe) in AB medium depleted of iron. The growth was studied under orbital shaking at 25°C for 90 h. The absorbance was measured at 600 nm every 3 h. Each dot is the mean of independent triplicates.



**FIG 2** Impact of increasing concentrations of iron on the chelating activity of the WT and mutant (*rhiE*::Gm) strains. To determine the concentration of iron inhibiting siderophore production, liquid cultures of WT (red) and *rhiE*::Gm (blue) strains were performed with different concentrations of iron (*i.e.,* 0, 0.1, 0.2, 0.3, 0.4, 0.5, 0.6, 0.7, 0.8, 0.9, 1, 1.5, 2, 2.5, and 5 mg L<sup>-1</sup>). Pure water (gray) was used as a negative control. After 3 days, the absorbance at 655 nm was measured. The decrease of the measured absorbance means the presence of chelating molecules since the media coloration changes from blue to yellow and the absorbance at 655 nm measures blue color intensity.

weathering process, experiments were done in buffered ABm-Fe medium over 7 days of incubation. The pH of the medium was controlled before and after the experiments in presence/absence of bacterial inoculum and hematite and remained stable (pH 6.1). In absence of hematite, the WT strain grew in the buffered ABm-Fe medium to an OD<sub>595nm</sub> of 0.47  $\pm$  0.01. The presence of hematite in the medium significantly increased growth of the WT strain (OD<sub>595nm</sub> = 0.82  $\pm$  0.01), while hematite addition had no impact on growth of the *rhiE*::Gm mutant (OD<sub>595nm</sub> = 0.45  $\pm$  0.01 in presence vs OD<sub>595nm</sub> = 0.44  $\pm$  0.02 in absence of the mineral). The CAS assay confirmed that only the WT strain was presenting an active chelating activity in both conditions (*e.g.*, in the presence or in the absence of hematite). Quantification of the iron released in solution from hematite highlighted significantly higher concentrations of iron for the WT strain (Fe = 0.76  $\pm$  0.02 mg L<sup>-1</sup>; *P* < 0.05) than for the *rhiE*::Gm mutant (0.04 ± 0.02 mg L−1) and the non-inoculated control  $(0.06 \pm 0.02 \text{ mg } L^{-1})$  (Fig. 3).

# **Rhizobactin is the siderophore produced by strain PML1(12)**

Pre-purified supernatants from 3-day cultures of *rhiE*::Gm mutant and WT strains were analyzed by high-performance liquid chromatography (HPLC) after purification. For this experiment, the growth was performed in ABs-Fe with succinate as sole carbon source to limit the production of exopolysaccharides, which are not compatible with the purification steps. The comparison of the HPLC chromatograms from the WT and *rhiE*::Gm strains as well as from the medium alone highlighted one major difference, a peak (peak 1) with a retention time of 0.84 min in the WT supernatant. This peak was



**FIG 3** Hematite weathering potential of strain PML1(12) and its *rhiE*::Gm mutant. The weathering ability was evaluated by the measure of iron released from hematite in ABm medium devoid of iron, after 7 days of incubation at 25°C under agitation (200 rpm). (A) Concentration of iron released in solution. The iron released from hematite was measured by ferrospectral determination (optical density measured at 595<sub>nm</sub>). Abbreviations were used as follows: WT + H, wild-type strain plus hematite; WT, wild-type strain; *rhiE*::Gm + H, mutant strain plus hematite; *rhiE*::Gm, mutant strain; abiotic + h, non-inoculated condition plus hematite; abiotic, non-inoculated condition. Non-inoculated conditions (abiotic control with and without hematite) were used as controls. Samples with the same letter indicate no significant difference (*P* < 0.05). (B) Siderophore activity. The siderophore activity of the different samples was determined using the CAS method. The yellow color indicates siderophore activity. For each measure, the results are the mean of independent triplicates.

2.8 and 3.8 times more intense for the WT strain than for the *rhiE*::Gm mutant strain and the non-inoculated medium, respectively (Fig. 4A). To correlate the HPLC chromatograms with the presence of chelating activities, the HPLC fractions were collected and assayed using the CAS assay. The CAS assay was positive for two fractions for the WT samples. The first positive fraction with the CAS assay (OD $_{655nm}$  = 0.06) was collected between 0.77 and 0.97 min corresponding to peak 1 region. The second positive fraction (OD $_{655nm}$  = 0.14) was identified between 3.74 and 3.95 min. Detailed analysis of the HPLC profiles within this retention time highlighted the presence of a small peak (peak 2), only for WT strain (Fig. 4B).

To identify the siderophore produced by strain PML1(12), we performed an HPLCphotodiode array detector (PDA)-high-resolution tandem mass spectrometry (HRMS/MS) analysis of the two fractions mentioned above. High-resolution MS analysis revealed the presence of a single molecule with the same exact mass of 377.1868 Da (monoprotonated ion seen at  $m/z = 378.1868$ ) in both active fractions and with the theoretical formula  $C_1$ <sub>5</sub>H<sub>25</sub>O<sub>8</sub>N<sub>3</sub> (Fig. 5A). Such compound was found only in the WT fractions (and not in the *rhiE*::Gm strain, nor in the negative control). An additional MS signal with  $m/z = 431.0982$  was also specifically observed in the two active WT fractions and was unambiguously attributed to the ferric complex  $[(C_{15}H_{25}O_8N_3-2H) + {^{56}Fe(HI)} + (m/z_{theo} =$ 377.1788-2  $\times$  1.008 + 55.935 = 431.0978, with <sup>56</sup>Fe = 55.935 u and <sup>1</sup>H = 1.008 u). Indeed, the experimental isotopic pattern was in perfect agreement with the simulated one (Fig. 6), thanks to the clear observation of the contribution of  $^{54}$ Fe isotope at m/z = 429.1029 { $[(C_{15}H_{25}O_8N_3-2H) + {}^{54}Fe(HI)]^+$ }. The previous mass spectrometric observations allow to conclude with certainty that compound of exact mass  $M = 377.1788$  Da can bind iron (III) efficiently and corresponds to the siderophore produced by strain PML1(12).

Such exact mass of 377.1868 Da has been previously associated with the siderophore rhizobactin by Smith et al. [\(19\)](#page-18-0) in a strain of *Ensifer meliloti* and Kügler et al. [\(20\)](#page-18-0) in the strain *Pseudomonas* sp. FEN. In order to elucidate the siderophore highlighted



**FIG 4** HPLC analysis and UV detection. The pre-purified supernatant of the WT strain (red), *rhiE*::Gm mutant (blue) and non-inoculated medium (gray) were monitored at 210 nm after HPLC separation using a Milli-Q water/acetonitrile + 0.2% trifluoroacetic acid gradient for 25 min at 1 mL min<sup>-1</sup>. To determine the presence of siderophore activity, fractions were collected during the HPLC run and tested using the liquid CAS assay. Positive fractions for siderophore presence are highlighted with black arrows. (A) A major peak (peak 1) at retention time of 0.84 min was identified. (B) A second peak (peak 2) of lower intensity, with a retention time of 3.88 min, was identified after CAS test.

in PML1(12), an MS/MS experiment was performed (Fig. 5B). Noticeably, our analyses identified the same four major product ions characterizing the rhizobactin described by Kügler et al. [\(20\)](#page-18-0) (Fig. 5C) as follows: (i)  $m/z = 289.1351$  (formed as a result of loss of alanine), (ii)  $m/z = 200.0882$  (formed as a result of loss of ethylenediamine and alanine and an additional carboxylic group), (iii)  $m/z = 116.0681$  (identified as malic acid ion), and (iv)  $m/z = 84.0790$  (identified as decarboxylated lysine ion). Overall, these analyses demonstrate that PML1(12) produces rhizobactin, a siderophore composed of a lysine, an alanine, an ethylenediamine group, and malic acid. This work represents the first demonstration that rhizobactin is an NRPS-independent siderophore. Furthermore, it



**FIG 5** Identification, fragmentation profile, and chemical structure of rhizobactin. (A) Mass spectrum of HPLC positive fractions showing the parent ion of the putative chelating molecule at m/z = 378.1868 (raw formula: C<sub>15</sub>H<sub>25</sub>O<sub>8</sub>N<sub>3</sub>). (B) Fragmentation pattern of the chelating molecule seen at m/z = 378.1868, revealing a perfect agreement with previous experimental MS<sup>2</sup> results done on the rhizobactin identified in Sinorhizobium meliloti DM4 by Smith et al. [\(19\)](#page-18-0) and in *Pseudomonas* sp. FEN by Kügler et al. [\(20\)](#page-18-0). The major product ions are as follows: 289.1351 (in blue, consecutive to alanine loss), 200.0882 (in yellow, formed as a result of loss of ethylenediamine and alanine and an additional carboxylic group), 116.0681 (in red, corresponding to malic acid ion-H2O), and 84.0790 (in green, corresponding to decarboxylated lysine ion). (C) Chemical structure [according to Smith et al. [\(19\)](#page-18-0)] and fragmentation pattern of the rhizobactin. Same colors as in panel B are used to highlight the correspondence of the chemical moieties of rhizobactin and product ions. Dotted boxes are used to highlight lost ions, while undotted boxes mark the remaining structure of the molecule.



**FIG 6** Isotopic pattern of iron complexes. Comparison between (A) the experimentally observed isotopic pattern of the putative iron(III) siderophore complex ion (m/z = 431.0963) and (B) the theoretical isotopic profile of the complex  $[(C_{15}H_{25}O_8N_3-2H)+Fe^{in}]^+$ ,  $C_{15}H_{25}O_8N_3$  being the raw formula associated with the putative siderophore of m/z = 378.1868.

confirms that rhizobactin is based on malic acid and not on other known NIS substrates (citric acid, α-ketoglutarate, or succinic acid).

#### **Comparative genomics of rhizobactin synthesis**

A BlastP search of the nr database using the RhiE protein as template allowed the identification of several homologous proteins presenting a percentage of identity ranging from 64% to 85% and good sequence coverage (95% to 100%). The best hits appeared assigned to the genera *Paraburkholderia*, *Trinickia*, *Vogesella*, *Polaromonas*, or *Pseudomonas*. Noticeably, a homolog was found in strain FEN of *Pseudomonas* sp. (63.5% homolog to RhiE; NCBI accession number WP\_191487624.1), a strain described previously for its ability to produce rhizobactin, but in which the synthesis genes had not been identified. Focusing on the *Burkholderiaceae*, we identified 38 proteins presenting a percentage of identity ranging from 34% to 85% and belonging to the genus *Paraburkholderia*, *Trinickia*, *Cupriavidus*, and *Ralstonia*. Considering the BCP group, besides the homologs in *Caballeronia*, we found RhiE homologs only in five strains assigned to *Paraburkholderia* (*P.* sp. BL23I1N*, P. haematera, P.* sp. T12-10*, P.* sp. NMBU\_R16, and *P.* sp. Ac-20340), organized almost identically as in PML1(12) (*i.e*., cluster *rhiABCDEFJ*) (Fig. 7), but with a variable number of genes. Noticeably, the ABC transport systems were not always present.

The phylogenetic analysis of NIS synthetases usually clusters them according to their substrate specificity [\(13\)](#page-17-0). To position the RhiE synthetase of strain PML1(12) in the NIS classification and to determine its relatedness to the known NIS synthetases, we constructed a N-J phylogenetic tree of a representative set of known NIS synthetases (Fig. 8). A total of 24 protein sequences of NIS synthetases which function and substrate are known were considered, as well as RhiE protein sequence of PML1(12) and the two homologous RhiE protein sequences identified in *Pseudomonas* FEN and *Ensifer (Sinorhizobium) meliloti* AK83 which are also potentially involved in rhizobactin production. Overall, our results confirm that NIS synthetases are distributed according

to their substrate specificity [\(13,](#page-17-0) 21[–23\)](#page-18-0). The newly characterized RhiE NIS synthetase of strain PML1(12), as well as the two homologs from *Pseudomonas* and *Ensifer*, cluster in group A', forming a well-supported sub-branch within this group. Cluster A' is the only in the phylogeny which does not seem to follow substrate specificity. Indeed, as demonstrated in the current work, RhiE and homologs synthesize rhizobactin, which is based on the use of malic acid as substrate (see above), while other NIS from this cluster use citric acid to synthesize siderophores such as rhizoferrin. The low support of the tree at the base of this cluster (bootstrap values of 37 and 42) may indicate a problem of phylogenetic resolution using the current set of data.

The comparison of the complete genomic region of the cluster of genes involved in rhizobactin transport and production (*rhiABCDEFGHIJ*) (Table 1) found in strain PML1(12) with the genomes of the strains already known to produce rhizobactin (*Ensifer meliloti* and *Pseudomonas* sp. FEN) highlighted a partial conservation of this cluster (*i.e.*, *rhiBCDEF*). The clusters differed, however, in several instances: (i) the presence of a TonB system in the upstream region in strain PML1(12) and at the downstream end in strain AK83 (AEG58212.1) and (ii) the absence of two ABC-transport systems (*rhiHI*) and the transcriptional regulator (*rhiJ*) in strains AK83 and FEN (Fig. 7).

# **DISCUSSION**

# **Rhizobactin identification**

Rhizobactin was first described in the plant-associated bacterial strain *Ensifer meliloti*  DM4 (former *Sinorhizobium meliloti*) [\(19\)](#page-18-0). This first study provided a detailed chemical composition based on mass spectrometry and NMR analyses, revealing that this carboxylate siderophore was formed by an ethylenediamine group coupled with alanine, lysine, and malic acid. The same siderophore was reported more recently for a bacterial strain isolated from a nutrient-poor environment, the strain *Pseudomonas* sp. FEN [\(20\)](#page-18-0). Both studies provided the chemical characterization of this molecule, but neither the genetic characterization nor the classification of rhizobactin. Interestingly, *Pseudomonas* sp. FEN can produce rhizobactin and rhizobactin B, rhizobactin B being the main siderophore produced. Structurally, these two siderophores differ only by the presence of a methyl group at the C2 position of malic acid, giving a citramalic acid in rhizobactin



**FIG 7** Organization of the cluster of genes involved in rhizobactin biosynthesis in different strains. Each arrow represents one gene, and its size was scaled according to gene length. Gene annotation is shown in the right, and the complete description is provided in Table 1. Light gray, dark gray, and green genes are involved in siderophore transport. In blue, red, and yellow, genes are involved in siderophore biosynthesis. In black, genes involved with siderophore production regulation. In bold, strains for which rhizobactin synthesis was demonstrated.



**FIG 8** Neighbor-joining phylogenetic analysis of NIS synthetases. A total of 24 experimentally validated and two proposed NIS-synthetases were aligned using MUSCLE algorithm, and the phylogenetic tree was constructed using SeaView. The code between brackets corresponds with the NCBI accession number for each protein sequence. NIS synthetases are colored depending on their specificity of substrate and thus their group: A (pink), A' (red), B (green), C (blue), and C' (yellow). The enzymes presented in this study (*i.e.,* RhiE and putative rhizobactin synthetases) are shown in black.

B instead of a malic acid in rhizobactin. A third rhizobactin type was identified in strain 1021 of *Ensifer meliloti* (*i.e.*, rhizobactin 1021), but completely different in terms of chemical structure from the one described above. This hydroxamate siderophore of 530 Da is based on a citrate derivative with the carboxyl groups linked to two different side chains [\(23\)](#page-18-0). In our study, we identified the siderophore produced by strain PML1(12) using a combination of CAS assay, HPLC purification, and MS/MS analyses. This molecule not only binds the different iron isotopes but also presented an exact mass (377.1868 Da) and a fragmentation pattern like that of the rhizobactin identified in *Ensifer meliloti* strain DM4 and *Pseudomonas* sp. strain FEN [\(19,](#page-18-0) 20). Our *rhiE*::Gm mutant of strain PML1(12), which is unable to chelate iron, proves the involvement of the *rhiE* gene cluster in the production of the unique siderophore molecule produced by strain PML1(12). Noticeably, the rhizobactin gene cluster appeared relatively conserved in the two strains already described for the production of rhizobactin, although the strain *Pseudomonas* sp. FEN presented an additional gene identified as a citrate lyase (WP\_191487621.1), which is the first gene of the cluster in this strain, upstream of the *rhiE* homolog. Citrate lyase may be involved in the process of transformation of malic acid into citramalic acid, explaining why rhizobactin B is the main siderophore produced by strain *Pseudomonas* sp. FEN, while rhizobactin is produced by *Ensifer meliloti* AK83 and strain PML1(12).

#### **Rhizobactin as NRPS-independent siderophore**

Two groups of bacterial siderophores are currently described according to their biosynthetic pathway. The first pathway uses complex, modular, multidomain enzymes known as NRPS [\(6,](#page-17-0) 13, 24). The second relies on NRPS-independent synthetases [\(13,](#page-17-0) 25). In comparison to the wealth of information available on the genetics and chemistry of the NRPS siderophores, NIS and their related biosynthetic pathways are still poorly

documented [\(25\)](#page-18-0). This is an important point because siderophores cannot be categorized as NIS or NRPS without a proper genetic characterization. In this sense, rhizobactin was discovered and chemically characterized by Smith et al. [\(19\)](#page-18-0), as early as 1984, in strain *Ensifer meliloti* DM4, but could only be classified as an NIS thanks to the current study and its homology at chemical  $(C_15H_25Q_8N_3)$  and genomic level (gene cluster) to that of strain PML1(12). The strain PML1(12) presents a gene cluster of 10 genes (Fig. 7) with the characteristic signature of the NIS synthesis pathway as identified for the first described NIS (*i.e.*, aerobactin) [\(26,](#page-18-0) 27). The aerobactin synthetases, IucA and IucC, became, since that, the hallmarks of the presence of NIS in genomes [\(22\)](#page-18-0), either in a single copy such as for the synthesis of rhizoferrin (*e.g*., FslA) in *Francisella tularensis* [\(28\)](#page-18-0) or rhizobactin in strain PML1(12) (*e.g*., RhiE, this study) or either in several copies such as for the synthesis of aerobactin (*e.g*., IucA and IucC) [\(27\)](#page-18-0). Our findings will enrich the NIS database and improve the siderophore determination using AntiSMASH.

#### **Rhizobactin a new class of NIS?**

NIS synthetases traditionally separate in five phylogenetic groups (A, A', B, C, and C'). This classification which was obtained on experimentally demonstrated NIS synthetases is congruent with their substrate specificity with: (i) type A and A' specific for citric acid, (ii) type B for α-ketoglutarate intermediates, and (iii) type C and C' specific for citryl or succinyl-based intermediates [\(13\)](#page-17-0). The number of experimentally demonstrated NIS synthetases is limited, and consequently, their position in phylogenetic trees may change with the evolution of NIS studies. As an example, the SfaB and SfaD enzymes responsible for the synthesis of staphyloferrin A which presents a citric acid in its structure [\(29\)](#page-18-0) were initially included in group B, a group of NIS synthetases specific for α-ketoglutarate. With the discovery of other novel citric acid-based and chiral siderophores (*e.g*.., rhizoferrin, legiobactin), SfaB and SfaD now cluster within a new group, named A'. Our phylogenetic tree agrees with those previously published [\(13,](#page-17-0) 21, 22), confirming that the NIS synthetases mostly cluster according to their substrate specificity. In this respect, the rhizobactin synthetase clusters with group A', which corresponds to citrate-based and chiral siderophores. However, rhizobactin and rhizobactin B synthetases form a wellsupported branch within group A'. Malic acid (rhizobactin) and citramalic acid (rhizobactin B) have never been described previously as possible substrates for NIS synthetases. Therefore, the difference of substrate specificity between rhizobactin synthetases (malic for rhizobactin and citramalic acid for rhizobactin B) and synthetases from group A' (citric acid) and the absence of other already characterized synthetases with malic acid as substrate allow us to propose a new group of NIS called D. The future characterization of new NIS synthetases that use malic/citramalic acid as substrate should make more robust the separation from A' group, as it happened previously for groups B and A' [\(29\)](#page-18-0).

#### **Role of rhizobactin in mineral weathering**

Iron is considered as an essential element, and its deficiency alters the growth of bacteria [\(6,](#page-17-0) 30, 31). Strain PML1(12) follows this rule, with a strong delay and decrease in growth in iron deprived conditions compared to non-limiting conditions. To adapt to low-iron conditions, bacteria can produce siderophores, whose production is regulated by the quantity of iron available and stopped at a maximum of 5 mg  $L^{-1}$  of Fe [\(32\)](#page-18-0). Here, we showed that the production of rhizobactin by strain PML1(12) was detectable until an iron concentration of 0.7 mg  $L^{-1}$ . Such iron regulation fits with the observations done by the dual transcriptomic and proteomic analyses on strain PML1(12) in the presence/absence of a mineral carrying iron (*i.e.*, biotite), which evidenced that the genes involved in siderophore production were only up-regulated in Fe-depleted conditions and in the absence of mineral [\(7\)](#page-17-0). Although minerals represent an important source of iron, the production of siderophores remains strongly determined by the iron bioavailability and the local conditions (pH, ionic strength). Several studies have demonstrated that, in oligotrophic conditions, minerals may be seen as a source of iron if siderophore-producing bacteria are present in the environment [\(33\)](#page-18-0). The effectiveness

of siderophores at solubilizing iron from different minerals is known and demonstrated for a long time in abiotic conditions. As an example, desferrioxamine B was proved to weather Fe from several iron oxides [\(15\)](#page-17-0). Even more, rhizoferrin and desferrioxamine A and B are not only effective at mobilizing Fe from hydroxide minerals, but also other metals such as chromium [\(34,](#page-18-0) 35). However, such abiotic experiments were usually based on high concentrations of siderophores and did not permit to consider all the regulation occurring in bacteria according to nutrient availability, ionic strength of the medium, and mineral type. In contrast, few studies investigated the ability of siderophore-producing bacteria to liberate nutrients trapped in minerals in a cellular context, considering mutants impaired in their ability to produce siderophores. Our study represents the first demonstration of the role of an NIS (*i.e*., rhizobactin) produced in biotic conditions in the weathering of hematite. Such demonstration completes the list of siderophores already characterized for their role in mineral weathering [\(6,](#page-17-0) 16, 17, 33).

#### **Occurrence of the rhizobactin siderophore in the BCP group**

Among members of the same genus or species, the type of siderophore produced can strongly differ (*i.e*., rhizoferrin biosynthesis) [\(36,](#page-18-0) 37) or on the contrary be very conserved (*i.e.*, pyoverdine biosynthesis) [\(38\)](#page-18-0). Members of the *Burkholderia sensu lato* are well known for the different strategies they use to mobilize iron based on siderophore-independent (Ftr) [\(39\)](#page-18-0) and siderophore-dependent (*e.g.*, ornibactin) [\(40,](#page-18-0) 41) systems, but not yet based on an NIS. Our study represents the first evidence of the production of rhizobactin in this group. Based on this evidence, we evaluated the distribution of a homologue of the *Caballeronia mineralivorans* strain PML1(12) RhiE NIS synthetase and the conservation of the *rhi* cluster of genes. Our analysis revealed that homologs of RhiE can be found in many genera among beta-proteobacteria, but the most represented remains the *Pseudomonas* genus. In this context, *Pseudomonas* sp. FEN is the only strain characterized for its ability to produce rhizobactin, while many *Pseudomonas* strains have been described for their ability to produce NRPS siderophores (*e.g.*, pyoverdin) [\(38\)](#page-18-0). Focusing on the BCP group (including *Burkholderia*, *Paraburkholderia*, and *Caballeronia*), only five *Paraburkholderia* strains (*P*. sp. BL23I1N*, P. haematera,* sp. T12-10*, P*. sp. NMBU\_R16, and *P*. sp. Ac-20340) presented the rhizobactin cluster, questioning its origin. In terms of ecology, many of the strains we identified to possess a rhizobactin synthesis locus homolog originate from the rhizosphere (*e.g*.*, Paraburkholderia* sp. T12-10, R16, and Ac-20340 and most of the members of *Trinickia, Vogesella, Polaromonas*, or *Pseudomonas*) and nutrient-poor ecosystems suggesting a role in plant nutrition and adaptation to nutrient-poor conditions. The role of siderophore-producing bacteria in plant nutrition and growth promotion has already been demonstrated as in the case of the strains *Burkholderia pyrrocinia* JK-SH007 [\(40\)](#page-18-0) or *Caballeronia mineralivorans* PML1(12) [\(42,](#page-18-0) 43).

# **Conclusion**

It is clearer and clearer that the ability of bacteria to weather minerals is finely regulated by local conditions and physiological processes. Nutrient availability, ionic strength of the medium, and the type of mineral and of the carbon substrate metabolized strongly determine the mechanism used by bacteria and the effectiveness at weathering. While the global mechanisms are identified (*i.e.*, acidification, chelation, oxido-reduction), the genetic and geochemical bases remain poorly documented. Demonstrations in a cellular context (*i.e.*, effect of a gene mutation) are essential to cover without *a priori* the diversity of mechanisms used by phylogenetically different bacteria and to decipher the regulations of these mechanisms occurring according to the mineral type and the local conditions. The combination of molecular and (geo-)chemical techniques used here allowed not only the identification of the siderophore produced by strain *Caballeronia mineralivorans* PML1(12) as rhizobactin but also its classification as a NIS and its contribution to mineral weathering. Our study provides new information regarding the complexity of mechanisms used by this model strain according to the local conditions and its metabolism. Our results highlight that siderophores play an essential role in

mineral weathering under buffered and non-acidifying conditions, while organic acid and proton-based mechanisms represent the main mineral weathering mechanism used by this strain under non-buffered and acidifying conditions. The ability of rhizobactin to weather more complex and less weatherable minerals, with a lower iron content (*e.g*., biotite) remains to be determined. Together, our results clearly highlight how the strain PML1(12) is well adapted to live in nutrient-poor environments and to develop complex and finely regulated mechanisms to mobilize nutrients from minerals.

#### **MATERIALS AND METHODS**

## **Bacterial strains and growth media**

Bacterial strains and plasmid are listed in Table 2. The model *Caballeronia mineralivorans*  strain PML1(12) was isolated from the ectomycorrhizosphere of *Scleroderma citrinum*  associated with oak [for sampling details see Calvaruso et al. [\(44\)](#page-18-0)].

Strains, with the exception of *Escherichia coli*, were grown in AB medium [\(45\)](#page-18-0) supplemented with mannitol (ABm) or succinic acid (ABs) (2 g L<sup>-1</sup> final concentration) at 25°C. A version of the AB medium devoid of iron was also used (AB-Fe). *E. coli* strains were grown in Luria-Bertani at 37°C. When required, gentamycin, kanamycin and ampicillin were added to the media at a final concentration of 20, 50, and 100 μg mL<sup>-1</sup>, respectively.

# *Preparation of bacterial inoculum of Caballeronia mineralivorans strain PML1(12) and its mutant*

For each of the assays presented in this work, the WT strain and the *rhiE*::Gm mutant strain were recovered from glycerol stock (−80°C) and grown on solid ABm medium (25°C) for 2 days. One isolated colony of each strain was inoculated in 10 mL of liquid ABm medium and incubated 3 days at 25°C under 150 rpm agitation. Liquid cultures were centrifugated at 9,000  $\times$  *g* for 15 min at 4°C, and the pellet was washed three times with sterile Milli-Q water. The pellet was resuspended in sterile Milli-Q water, and the optical density was adjusted to OD<sub>595nm</sub>= 0.9  $\pm$  0.03 (corresponding to 2.10 $^{\circ}$  cell/mL).

## *rhiE***::Gm mutant construction**

Total DNA was extracted from the WT strain PML1(12) using the protocol of Pospiech and Neumann [\(46\)](#page-18-0). To construct a targeted mutant of the gene homologous to *iucA/iucC*  (*rhiE*), the total gene was amplified using a combination of primers including For-*rhiE* and Rev-*rhiE* for a complete amplification and Rev-*rhiE*\_*Sma*I and For-*rhiE*\_*Sma*I to insert a *Sma*I site (Table 2; Fig. S2 for details). The insertion of a *Sma*I site in the middle of the *rhiE*  gene was done following three PCR rounds: (i) For-*rhiE* with Rev-*rhiE*\_*Sma*I (967 bp), (ii) For-*rhiE*\_*Sma*I with Rev-*rhiE* (966 bp), and (iii) For-*rhiE* and Rev-*rhiE* (1,908 bp). The PCR products of PCR1 and 2 were purified and used as matrix for PCR 3. After amplification and purification, PCR3 product was ligated into the pGEM-T Easy plasmid (Promega) resulting in the plasmid pGEM-*rhiE*\_*Sma*I (Table 2). The plasmid pGEM-*rhiE*\_*Sma*I was then digested with *Sma*I to introduce the *Sma*I-digested gentamycin (Gm) resistance cassette from plasmid pUC1318, resulting in the plasmid pGEM-*rhiE*::Gm. A fragment containing the *rhiE*::Gm cassette was then obtained by *Eco*RI restriction and cloned in the plasmid pK19mob, resulting in the pK19mob-*rhiE*::Gm plasmid. This construction was transferred to *E. coli* S17.1λpir and then to strain PML1(12) by bi-parental conjugation. Δ*rhiE* mutants were recovered after 5 days of incubation at 25°C on gentamycin-containing ABm plates. The transconjugants obtained were verified by PCR with For-*rhiE*  and Rev-*rhiE* primers, to differentiate the single and double cross-over events. Several single and double cross-over mutants were obtained and conserved. The experiments described in our study were done using a double cross-over mutant named *rhiE*::Gm.

**TABLE 2** List of bacterial strains, constructions, and primers used in this study*<sup>a</sup>*



ªGm<sup>8</sup>, Km<sup>R</sup>, Sm<sup>R</sup>, and Amp<sup>R</sup> indicate gentamicin, kanamycin, spectinomycin, and ampicillin resistance cassettes, respectively.

## **Siderophore activity**

Siderophore activity was detected using the liquid and solid versions of the CAS assay according to Schwyn and Neilands [\(47\)](#page-18-0). For both assays, the color change, sign of the chelating activity of the siderophore for the iron (III) of the CAS-Fe complex, can be quantified by the presence of a halo on the solid CAS assay or through the measurement of the optical density at 655 nm for the liquid assay.

For the solid CAS assay, 5 µL of the bacterial inoculum [both WT and mutant strains; corresponding to 5  $\times$  10<sup>6</sup> colony-forming units (CFU)] were inoculated in triplicate in agar plates and incubated for 3 days at 25°C. The diameter of colony and the halo surrounding it were measured to determine growth and siderophore(s) production, respectively.

For the liquid CAS assay, 0.5 mL of the bacterial inoculum (both WT and mutant strains; corresponding to  $5 \times 10^8$  CFU) were inoculated in triplicate in 4.5-mL ABm-Fe. After 3 days of incubation at 25°C under agitation (150 rpm), the cultures were centrifugated at 9,000  $\times$  *g* during 15 min at 4°C to recover culture supernatants. A volume of 100 µL of these supernatants was mixed with 100 µL of CAS-Fe(III) and incubated during 1 h at room temperature in the dark. Then, the OD was measured at 655 nm using a microplate reader (Bio-Rad, model iMark). Decrease of absorbance at this specific wavelength is synonym of siderophore presence [change of media coloration from blue to yellow [\(6,](#page-17-0) 47)]. Non-inoculated medium was used as negative control for the liquid assay.

### **Impact of iron concentration on the growth the WT and** *rhiE***::Gm strains**

To determine if the concentration of iron and the mutation of the *rhiE* gene (*rhiE*::Gm mutant) affected the growth of strain PML1(12), growth was determined at different iron concentrations (*i.e.*, 0, 0.5, 1, 5, and 10 mg  $L^{-1}$  of Fe). A bacterial inoculum of 10  $\mu$ L (corresponding to  $6 \times 10^6$  CFU) was added to 190 µL of ABm-Fe and incubated in 96-well microplates for 5 days. This experience was performed in triplicate and conducted at 25°C on orbital shaking in a microplate reader (Tecan infinite M200 pro). The absorbance was measured at 595 nm every 3 h. To evaluate the growth of the strains, the slope of the exponential phase (*i.e*., growth rate) was calculated for each condition and replicate.

# **Siderophore production inhibition**

Siderophore production is known to be regulated by the availability of nutritive elements in the culture medium, especially iron. To evaluate the concentration of Fe necessary to inhibit the siderophore production, the WT and mutant strains were incubated at different iron concentrations. Non-inoculated conditions were used as negative controls for each concentration. A volume of 10 µL of inoculum was added to 180 µL of ABm-Fe supplemented with 10 µL of different FeCl<sub>3</sub> stock solutions ranging from 0 to 5 mg L<sup>-1</sup> of Fe to obtain the following final concentrations in the medium: 0, 0.1, 0.2, 0.3, 0.4, 0.5, 0.6, 0.7, 0.8, 0.9, 1, 1.5, 2, 2.5, and 5 mg L<sup>-1</sup>. The 96-well microplates were incubated 3 days at 25°C. Then, microplates were centrifugated at 3,000 × *g* during 15 min at 4°C, and 100 µL of supernatant was mixed with 100 µL of CAS-Fe(III). After 1-h incubation in the dark, the plates were read at 655 nm using a microplate reader (Bio-Rad, model iMark) as described above.

# **Hematite weathering assay**

# *Hematite characteristics*

Hematite (Fe<sub>2</sub>O<sub>3</sub>) is an iron-oxide mineral widely distributed in all soils [\(48\)](#page-18-0) and characterized by a high iron content. The hematite used in this study comes from a batch extracted in Brazil. The chemical composition was determined on crushed powder by inductively coupled plasma-atomic emission spectrometry (700 Series ICP-AES, Agilent Technologies) as described in Picard et al. [\(49\)](#page-18-0). This analysis revealed that the hematite used is mainly composed of Fe $_2O_3$  (99.3%) and also contains in small amounts: 0.40% TiO2, 0.18% SiO2, 0.16% Al2O3, 0.10% MgO, 0.037% MnO.

# *Mineral weathering assay*

To avoid contamination by residues associated with glass tubes, all glassware was rinsed once with 3.6% HCl and then three times with Milli-Q water. The tubes were filled with 200 mg of hematite and 9 mL of ABm-Fe medium and autoclaved at 121°C. To evaluate the differences in the weathering ability of the *rhiE*::Gm mutant and the WT strains, 1 mL of each inoculum was added to the tubes (procedure performed in triplicate). Non-inoculated media with and without hematite were used as abiotic and sterility controls for weathering. Samples were incubated for 7 days under agitation at 150 rpm and 25°C.

The cultures were used to determine the following: (i) bacterial growth, (ii) solution pH, (iii) siderophore production, and (iv) iron concentrations. Bacterial growth was estimated directly on the liquid culture by measuring the OD<sub>595nm</sub> on 200 µL. The other measurements were done after centrifugation at  $14,000 \times q$  for 15 min to remove bacterial cells and hematite particles. A volume of 100 µL of supernatant was used to evaluate the level of siderophore(s) production using the liquid CAS assay according to the protocol described above. A volume of 180 µL of supernatant was used to measure the pH by mixing with 20 µL of bromocresol green  $(1 \, \text{q L}^{-1})$  according to Uroz et al. [\(18\)](#page-18-0). For the measure of the iron concentration, the supernatants were filtered at 0.22  $\mu$ M (GHP Acrodisc 25 mm syringe filter; PALL) before mixing 180 µL of supernatant with 20 µL of ferrospectral. Both pH and iron determinations are based on optical density measurements at 595 nm using a microplate reader (Bio-Rad, model iMark) and after conversion with the calibration curves done on the bromocresol green and ferrospectral dyes, according to Uroz et al. [\(18\)](#page-18-0).

# **Chemical characterization of the siderophore**

# *Siderophore production and purification*

Volumes of 50 mL of ABs-Fe were inoculated with each strain (WT and *rhiE*::Gm) and incubated at 25°C under 150 rpm shaking for 3 days. Succinic acid was used here as sole carbon source to limit the production of exopolysaccharides. A non-inoculated

condition was also included as negative control. After the incubation time, cultures were centrifugated at 9,000  $\times$  *g* during 15 min at 4°C. The supernatant was filtered using 0.45 µM filters (Millipore) and treated with a Sep pack Vac 6cc tC18 Cartridge (Waters) to prepurify and concentrate the siderophore(s). The columns were first washed with 10 mL of 100% ethanol and then equilibrated with 10 mL of 0.01% trifluoroacetic acid (TFA). After this equilibration step, a volume of 25 mL of supernatant was added progressively to the column, and finally, the different fractions of the supernatant were eluted by the addition of 5 mL of 10% and then 20% methanol in water. At each step, the eluted volume (*ca*. 5 mL) was recovered and concentrated in the SpeedVac system. The dried samples were resuspended in 100 µL of Milli-Q water. These pre-purified fractions were used for further characterization using the liquid CAS assay to verify the presence of siderophore.

#### *High-performance liquid chromatography analyses and fractionation*

The different pre-purified fractions recovered above (from WT, mutant, and negative control samples) were analyzed by HPLC. A volume of 50 µL of each extract was injected in a Gemini C18 column (150  $\times$  3.0 mm, 5 µM particle size, Phenomenex). Solvent A consisted of Milli-Q water, and solvent B consisted of acetonitrile plus 0.2% of TFA. A linear gradient elution from 2% to 40% of solvent B over 25 min at a flow rate of 1 mL min−1 was used to separate compounds. Absorbance was monitored at 210 nm.

Peaks of potential interest were identified by comparing the chromatogram of the WT sample with that of the *rhiE*::Gm mutant sample and the negative control. To determine the presence of the siderophore(s) activity, a fraction collection was performed in 96-well microplates using a fraction collector system (FRC-10A fraction collector system, Shimadzu). Fractions of 200 µL were recovered each 0.2 s using the same HPLC conditions as described before. Fraction collection was performed three times independently, and fractions with the same retention times were pooled. The fractions collected were then concentrated using the SpeedVac and resuspended in 100 µL of Milli-Q water. To determine the presence of siderophore(s) activity, a volume of 25 µL of the collected fractions was mixed with 25 µL of liquid CAS, and the absorbance at 655 nm was measured after overnight incubation.

# *Qualitative HPLC-DAD-MS-MS/MS analysis of siderophores and ferrosiderophores*

A Thermo Scientific UHPLC-HRMS/MS system composed of a Vanquish liquid chromatography unit coupled to a PDA and an Orbitrap ID-X Tribrid high-resolution mass spectrometer operating in electrospray ionization mode (ESI) was used to detect and identify the putative siderophores. Ten microliters of each bacterial extract were injected onto a Hichrom Alltima C18 column (150 × 2.1 mm − 5 µM) maintained at 25°C. The flow rate was set at 0.2 mL min<sup>-1</sup> and mobile phases consisted of water modified with formic acid (0.1%) for A and acetonitrile modified with formic acid (0.1%) for B. Compounds of interest were eluted using a linear gradient from 5% to 20% of B for 10 min, and then, a 5-min isocratic step was applied at 98% of B to wash the column, before returning to the initial composition of 5% B for 6 min to reach the equilibrium. Mass analysis was carried out in ESI positive ion mode (ESI<sup>+</sup>), and mass spectrometry conditions were as follows: spray voltage was set at 3.5 kV; source gases were set (in arbitrary units min<sup>-1</sup>) for sheath gas, auxiliary gas, and sweep gas at 35, 7 ,and 10, respectively; vaporizer and ion transfer tube temperatures were both set at 300°C. Survey scans of precursors were performed from 150 to 2,000 m/z at 60 K resolution (FWHM at 200 m/z) with MS parameters as follows: RF-lens, 35%; maximum injection time, 50 ms; data type, profile; internal mass calibration EASY-IC TM activated; normalized AGC target, 25%. A top speed data-dependent MS<sup>2</sup> (0.6 s for the whole cycle time) was carried out using a wide quadrupole isolation (1.5 Th), an HCD fragmentation with a stepped collision energy (20, 35, and 50%) and an Orbitrap measure at 15 K resolution (high-resolution MS/MS analysis). Precursors with an intensity upper than  $2.10<sup>4</sup>$  were automatically sampled for MS<sup>2</sup>. Dynamic exclusion was

<span id="page-16-0"></span>used, and the time of exclusion was set at 2.5 s, with a 10-ppm tolerance around the selected precursor (isotopes excluded). Other MS<sup>2</sup> parameters were as follows: data type, profile; normalized AGC target, 20%; AGC target, 10,000.

Mass spectrometer calibration was performed by using the Thermo Scientific Pierce FlexMix calibration solution. MS data acquisition was carried out by using the Xcalibur v. 3.0 software (Thermo Scientific).

#### **Bioinformatic analyses**

#### *Genome analysis*

Genome analysis was performed using MicroScope MaGe (Magnifying Genomes) [\(50\)](#page-18-0) and NCBI. The presence of *iucA/iucC* gene homolog was identified by antiSMASH [\(51\)](#page-18-0) and confirmed by BLASTP analyses based on sequence homology [\(52\)](#page-19-0).

# *Phylogenetic analyses*

A total of 27 protein sequences (Table S1) homologous to IucA/IucC*,* including the homolog of PML1(12) (RhiE), were recovered from NCBI to build a phylogenetic neighbor-joining (N-J) tree using the Seaview platform (version 5.0.4) [\(53\)](#page-19-0). For this analysis, the protein sequences considered were as follows: (i) sequences which role in siderophore biosynthesis were experimentally demonstrated; (ii) sequences based on their homology with the RhiE protein identified in strain PML1(12). Protein sequences were aligned using MUSCLE algorithm, and then, an N-J tree with 100.000 iterations was constructed.

# **Statistical analyses**

Statistical analyses were performed in R software (Tidyverse packages) [\(54\)](#page-19-0). Triplicate samples were used in all experiments. Differences between samples means (pH, growth, CAS assay, and released iron) or growth slopes were analyzed by analysis of variance and TukeyHSD tests.

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#### **ADDITIONAL FILES**

The following material is available [online.](https://doi.org/10.1128/aem.00453-23)

#### Supplemental Material

**Supplemental file 1 (AEM00453-23-S0001.pdf).** Fig. S1 and S2 and Table S1

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