

Regulation of Phosphoenolpyruvate Carboxylase from the Green Alga *Selenastrum minutum*¹

Properties Associated with Replenishment of Tricarboxylic Acid Cycle Intermediates during Ammonium Assimilation

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ABSTRACT

Two isoforms of phosphoenolpyruvate carboxylase (PEPC) with very different regulatory properties were partially purified from the green alga *Selenastrum minutum*. They were designated PEPC₁ and PEPC₂. PEPC₁ showed sigmoidal kinetics with respect to phosphoenolpyruvate (PEP) whereas PEPC₂ exhibited a typical Michaelis-Menten response. The $S_{0.5}(\text{PEP})$ of PEPC₁ was 2.23 millimolar. This was fourfold greater than the $S_{0.5}(\text{PEP})$ of PEPC₂, which was 0.57 millimolar. PEPC₁ was activated more than fourfold by 2.0 millimolar glutamine and sixfold by 2.0 millimolar dihydroxyacetone phosphate (DHAP) at a subsaturating PEP concentration of 0.625 millimolar. In contrast, PEPC₂ showed only 8% and 52% activation by glutamine and DHAP, respectively. The effects of glutamine and DHAP were additive. PEPC₁ was more sensitive to inhibition by glutamate, 2-oxoglutarate, and aspartate than PEPC₂. Both isoforms were equally inhibited by malate. All of these metabolites affected only the $S_{0.5}(\text{PEP})$ not the V_{max} . The regulatory properties of *S. minutum* PEPC *in vitro* are discussed in terms of (a) increased rates of dark carbon fixation (shown to be catalyzed predominantly by PEPC) and (b) changes in metabolite levels *in vivo* during enhanced NH_4^+ assimilation. Finally, a model is proposed for the regulation of PEPC *in vivo* in relation to its role in replenishing tricarboxylic acid cycle intermediates consumed in NH_4^+ assimilation.

It has been proposed the PEPC² in C_3 plants functions to replenish TCA cycle intermediates consumed in amino acid biosynthesis (5, 12, 16). During NH_4^+ assimilation, the net synthesis of glutamine and glutamate utilizes 2-oxoglutarate in the GS/GOGAT reactions (17). Increased rates of NH_4^+ assimilation have been correlated with elevated rates of dark carbon fixation in cyanobacteria, green algae, and higher

plants (4–6, 10, 11, 21, 22, 24) and these elevated rates of dark carbon fixation have been attributed to increased activity of PEPC. However, the factors responsible for increased PEPC activity have not been identified.

In contrast to the situation in C_3 plants, the regulation of PEPC in C_4 and CAM plants is well understood (23). The enzyme purified from leaf tissue of C_4 and CAM plants is inhibited by malate and activated by glucose-6-P (1). In the CAM plant *Crassula argentea*, PEPC undergoes interconversion between a more active tetrameric 'night form' (less sensitive to malate inhibition, more sensitive to glucose-6-P activation) and a less active dimeric 'day form' (more sensitive to malate inhibition, less sensitive to glucose-6-P activation) (35, 36). Both CAM and C_4 PEPC are regulated in a light dependent manner by protein phosphorylation (8, 20). The phosphorylated enzyme, which is more abundant in the dark in CAM plants and in the light in C_4 plants, is less sensitive to malate inhibition than the dephosphorylated enzyme. In the C_4 plant maize, where the day form of PEPC is more active than the night form, phosphorylation status is the most important factor, and changes in oligomeric state apparently are not involved in regulation by light (15). The properties of C_4 and CAM PEPC are consistent with its role in the C_4 acid cycle of photosynthesis. PEPC from C_3 plants and nonphotosynthetic forms of the enzyme from C_4 and CAM plants might, therefore, be expected to have very different properties from the photosynthetic forms of PEPC (1, 12). PEPC from C_3 plants has a lower K_m (PEP) and is less sensitive to malate inhibition and glucose-6-P activation than the C_4 enzyme (12). Beyond this, very little is known of the regulation of PEPC in C_3 plants, especially as pertains to its potential anaplerotic functioning during NH_4^+ assimilation.

In the present study we confirm the stimulation of dark carbon fixation associated with an increase in the rate of NH_4^+ assimilation in a C_3 plant, the green alga *Selenastrum minutum*, and present evidence that PEPC is the predominant enzyme catalyzing this process. We also report the partial purification of two isoforms of PEPC from *S. minutum* and examine their regulatory properties. This allows us to propose an integrative model for the regulation of PEPC in a C_3 plant during increased demand for the replenishment of TCA cycle intermediates consumed in the assimilation of NH_4^+ into amino acids.

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² Abbreviations: PEPC, phosphoenolpyruvate carboxylase; DHAP, dihydroxyacetone phosphate; DIC, dissolved inorganic carbon; FPLC, fast protein liquid chromatography; GOGAT, glutamine oxoglutarate aminotransferase; GS, glutamine synthetase; n_H , Hill coefficient; PEP, phosphoenolpyruvate; PK, pyruvate kinase; PK_c, cytosolic pyruvate kinase; Rubisco, ribulose biphosphate carboxylase/oxygenase.

MATERIALS AND METHODS

Materials

The green alga *Selenastrum minutum* (Naeg.) Collins (UTEX 2459) was cultured in NO_3^- -limited chemostats as previously described (4). The medium was buffered to pH 8.0 with 25 mM Hepes-KOH. The steady state growth rate of cells for dark carbon fixation and metabolite experiments was 0.3 d^{-1} and for enzyme extraction and purification it was 1.2 d^{-1} . All biochemicals and malate dehydrogenase were from Sigma Chemical Co. Other coupling enzymes were from Boehringer-Mannheim. $\text{Na}_2^{14}\text{CO}_3$ was from Nordion International Inc. All other reagents were analytical grade.

Spectrophotometric PEPC Assay

The PEPC reaction was coupled with the malate dehydrogenase reaction and assayed at 25°C by monitoring NADH utilization at 340 nm using a Varian DMS 200 spectrophotometer. The standard assay contained 25 mM Bis Tris Propane (pH 8), 15% (v/v) glycerol, 5 mM MgCl_2 , 5 mM KHCO_3 , 5 mM DTT, 0.2 mM NADH, 2.5 mM PEP, and 10 units malate dehydrogenase in a final volume of 1 mL. Assays were initiated by addition of PEPC. One unit is defined as the amount of enzyme catalyzing the utilization of $1 \mu\text{mol NADH} \cdot \text{min}^{-1}$.

Dark Carbon Fixation and Metabolite Analyses

Dark carbon fixation was assayed by following the incorporation of $\text{H}^{14}\text{CO}_3^-$ into acid stable products in the dark as previously described (34), except that unfixed counts were removed by drying the sample in a stream of air and redissolving in 1 mL of kill solution, three times. The dried sample was taken up in 1 mL H_2O and 2 mL Aquasol II (Dupont) liquid scintillation cocktail. Experimental protocols for the extraction in CHCl_3 -methanol and the analysis of metabolites were as previously described (30, 32).

Radiometric PEPC and PEP Carboxykinase Assay

S. minutum cells growing in a chemostate at a rate of 0.3 d^{-1} or in batch culture (13) were extracted in the medium of Jiao and Chollet (8) with 30 mM NaF added and only 10% (v/v) glycerol. The homogenate obtained after two passes through a French press at 18,000 p.s.i. was used in the assay. The assay buffer contained, in a final volume of 1 mL, 50 mM Hepes-KOH (pH 8), 10 mM MgCl_2 , 5 mM DTT, 1 mM NADH, 5 units malate dehydrogenase, 5 mM PEP, 0.5 mM ADP (PEP carboxykinase only), and 2 mM NaHCO_3 (specific radioactivity, $4.5 \mu\text{Ci}/\mu\text{mol DIC}$). The assay was initiated with the addition of extract and run in the dark. Incorporation of ^{14}C into acid stable products was determined as above. PEPC activity was taken as the rate in the absence of ADP, and PEP carboxykinase activity was calculated as the difference between this rate and that obtained in the presence of ADP.

Pyruvate Carboxylase and NAD-Malic Enzyme Assays

Crude cell extracts were prepared as described for the radiometric PEPC and PEP carboxykinase assay (see above) except that the extraction buffer contained 100 mM Hepes-KOH (pH 7.8), 2 mM EDTA, and 10 mM DTT. The assay buffer contained, in a final volume of 1 mL, 100 mM Hepes-KOH (pH 7.8), 10 mM MgSO_4 , 5 mM DTT, 1 mM NADH, 10 units malate dehydrogenase, 4 mM ATP, 0.15 mM acetyl CoA, 10 mM pyruvate, and 10 mM KHCO_3 (specific radioactivity, $8.56 \mu\text{Ci}/\mu\text{mol DIC}$). The assay was initiated with the addition of extract and run in the dark. Incorporation of label into acid stable products was determined as above. For malic enzyme, the assay procedure was the same as for pyruvate carboxylase, except that the pH was 7.0 and ATP was omitted.

Buffers Used in PEPC Purification

Buffer A: 25 mM K-Pi (pH 7.0), 20% (v/v) glycerol, 5 mM malate, 2 mM DTT, 1 mM EDTA, 10 mM MgCl_2 , 0.1% (v/v) Triton X-100, 4% (w/v) PEG-8000, and 2 mM PMSF. Buffer B: 10 mM K-Pi (pH 7.5), 20% (v/v) glycerol, 5 mM malate, 2 mM DTT, 1 mM EDTA, and 5 mM MgCl_2 . Buffer C: As for buffer B but with 1 M KCl added. Buffer D: 25 mM KPi (pH 7.0), 20% (v/v) glycerol, 5 mM malate, 5 mM thiourea, 2 mM DTT, 1 mM EGTA, 1 mM EDTA, 10 mM MgCl_2 , 1 mM NaF, 1 mM PMSF, 50 mM KCl, and 0.02% (w/v) NaN_3 .

Native Molecular Mass Determination

Native molecular mass estimations were made using a 100 mL FPLC Superose 6 HR 16/50 column equilibrated with buffer D. The sample, 200 to 300 mg protein in a volume of 2 mL, was applied to the column at a flow rate of 0.3 mL/min. Molecular masses were determined from a plot of K_{av} (partition coefficient) versus log molecular mass for standard proteins: thyroglobulin (669 kD), ferritin (440 kD), and catalase (232 kD). Blue dextran was used to determine the void volume.

Protein Assay

Protein was determined according to the method of Bradford (2) using bovine γ -globulin as standard.

Enzyme Kinetic Analyses

All kinetic studies were performed using the spectrophotometric assay. Apparent $S_{0.5}$ values and Hill coefficients were calculated from Hill plots using Enzfitter (Sigma Chemical Co.), a nonlinear regression program. Activation constants (K_a) were determined using double-reciprocal plots of $1/(v - v_0)$ versus $1/[\text{activator}]$. I_{50} values (inhibitor concentration producing 50% inhibition of enzyme activity) were determined by the method of Job *et al.* (9). All data are the means of duplicate determinations performed on one representative enzyme preparation.

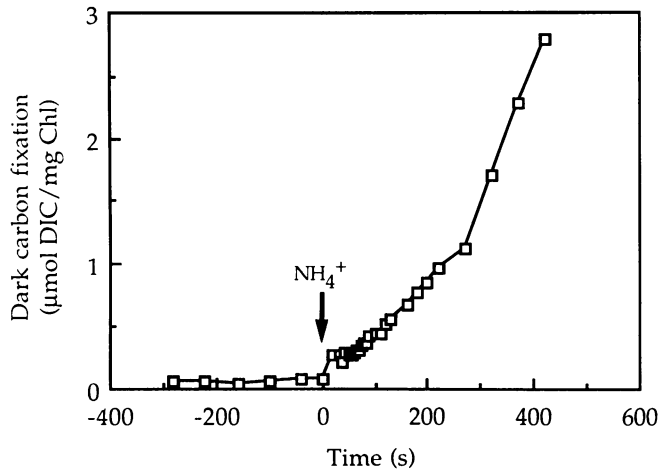


Figure 1. Effect of NH₄⁺ assimilation on dark carbon fixation by N-limited *S. minutum*. Cells, at a density of 2 μg Chl/mL, were supplied with Na₂¹⁴CO₃ (specific radioactivity = 6.1 μCi/μmol DIC). Total DIC was 4.3 mM. NH₄Cl (2 mM) was added at time zero. Data are the means of three replicate experiments with different batches of cells.

RESULTS

In Vivo Effects of NH₄⁺ Assimilation

Effects of NH₄⁺ Assimilation on Dark Carbon Fixation

Figure 1 shows the effects of resupplying NH₄⁺ to N-limited *Selenastrum minutum* cells on incorporation of H¹⁴CO₃⁻ into acid stable products in the dark. Prior to NH₄⁺ resupply the rate of dark carbon fixation was low, less than 1 μmol·mg⁻¹ Chl·h⁻¹. Immediately following NH₄⁺ addition it increased to 12 μmol·mg⁻¹ Chl·h⁻¹ (0–180 s). Subsequently it increased still further to 40 μmol·mg⁻¹ Chl·h⁻¹ (270–420 s). PEPC activity in crude cell extracts of *S. minutum* was 30 μmol·mg⁻¹ Chl·h⁻¹ (radiometric assay) which was almost sufficient to account for the maximum rate of dark carbon fixation. Three other enzymes (PEP carboxykinase, pyruvate carboxylase, and NAD-malic enzyme) which could potentially catalyze dark carbon fixation showed no activity in the carboxylating direction in *S. minutum* extracts (data not shown).

Effects of NH₄⁺ Assimilation on PEP and Malate Levels

Resupply of NH₄⁺ to N-limited *S. minutum* cells caused a rapid decline in the PEP level (Fig. 2A). Within 5 s of NH₄⁺ addition, PEP declined to 41% of its control value. Malate remained relatively constant for the first 2 min of NH₄⁺ assimilation (Fig. 2A) but had declined slightly by 5 min. These changes resulted in a large increase in the malate/PEP ratio (Fig. 2B).

Partial Purification of Two Isoforms of PEPC

Two isoforms of PEPC were partially purified from *S. minutum* using the following procedure. All steps were per-

formed at 4°C and PEPC activity was determined using the spectrophotometric assay.

Preparation of the Crude Extract

Cells (50 g) stored at -80°C were thawed into 100 mL of buffer A and passed twice through a French press at 18,000 p.s.i. The homogenate was centrifuged at 24,000g for 20 min. The supernatant fluid was retained and designated the crude extract. Inclusion of PMSF, a serine protease inhibitor, in the extraction buffer was very important for the stability of PEPC. Activity losses in the range 40 to 60% occurred between preparation of the crude extract and resuspension of the PEG pellets if PMSF was omitted (data not shown).

PEG Fractionation

The crude extract was brought to 20% (w/v) PEG with a 50% (w/v) solution of PEG-8000 containing 50 mM KH₂PO₄ (pH 7) and 2 mM EDTA. After 20 min stirring, the extract was centrifuged as above and the pellets retained. No PEPC activity was detected in the supernatant fluid.

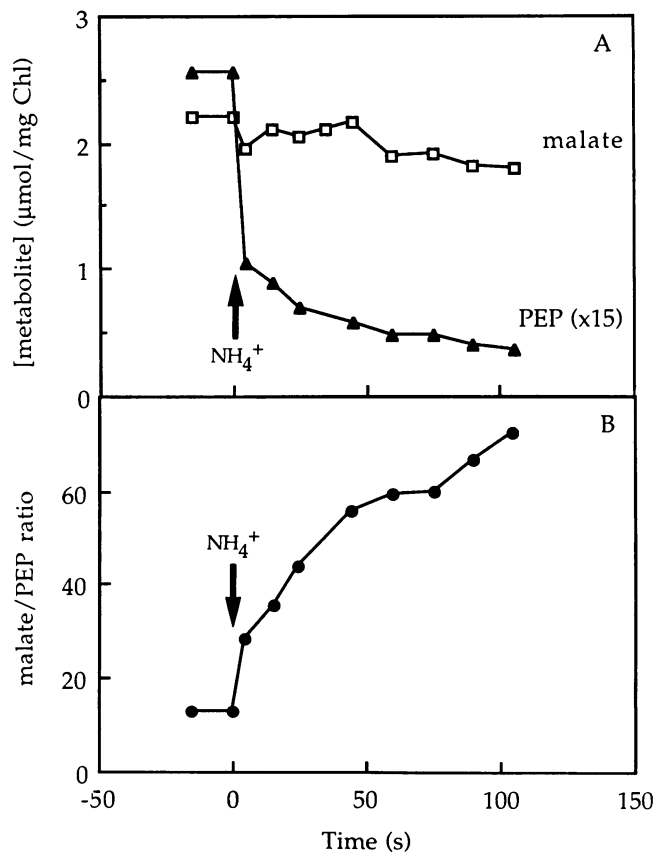


Figure 2. Effect of NH₄⁺ assimilation on total cellular malate and PEP levels in N-limited *S. minutum*. NH₄Cl (5 mM) was added at time zero. Data are the means of triplicate analyses of duplicate experiments.

Table I. Partial Purification of *S. minutum* PEPC₁ and PEPC₂

The standard spectrophotometric assay was used for PEPC activity.

Fraction	Volume mL	Total Activity units	Total Protein mg	Specific Activity units/mg	Yield %	Purification -fold
I. Crude extract	83	63	2573	0.025	100	1.0
II. PEG-8000 fractionation (4–20% w/v)	200	63	1840	0.035	100	1.4
III. Q-Sepharose						
PEPC ₁	2.3	7.6	209	0.036	12	1.4
PEPC ₂	5.1	22.4	163	0.138	36	5.5

Q-Sepharose Chromatography

The PEG pellets were resuspended in buffer B to give a protein concentration of $9.2 \text{ mg} \cdot \text{mL}^{-1}$. Complete recovery of the total PEPC activity assayed in the crude extract was obtained (Table I). The resuspended pellets were centrifuged as above to remove insoluble material and the supernatant liquid was adsorbed batchwise with gentle stirring for 2 h onto 70 mL of Q-Sepharose resin preequilibrated with buffer B. The slurry was then packed into a 2.5-cm diameter column and washed, at 2 mL/min, until the A_{280} fell below 0.05. PEPC activity eluted as two distinct peaks when a 350 mL linear gradient of 0 to 400 mM KCl in buffer B was applied to the column, using an FPLC system (Fig. 3). The early and later eluting peaks were designated PEPC₁ and PEPC₂, respectively. PEPC₁ began to elute at a conductivity of 0.7 mmho which was only marginally above the conductivity of the start buffer (buffer B = 0.4 mmho). PEPC₂ was well separated from PEPC₁, beginning to elute at 2.0 mmho. The recovery of the total activity-units loaded onto the Q-Sepharose column was 48%, with 25% and 75% of this accounted for in PEPC₁ and PEPC₂, respectively (Table I). Pooled peak fractions containing PEPC₁ or PEPC₂ were concentrated approximately 25-fold using an Amicon YM-30 ultrafilter, divided into 200 μL aliquots, frozen with liquid N₂, and stored at -80°C . The purified enzymes were stable for at least 1 month when stored frozen. All subsequent kinetic analyses were performed on the two isoforms of PEPC purified through the Q-Sepharose step.

Kinetic Characterization of the Two Isoforms of PEPC

Effect of pH

The effect of pH on the two isoforms of PEPC was determined in the presence of the substrate and cofactor concentrations used in the standard spectrophotometric assay. As will be shown later (see Table II), the PEP concentration (2.5 mM) was only about 10% above the $S_{0.5}$ for PEPC₁ but saturating for PEPC₂. Under these conditions, the pH profiles of the two isoforms were very similar with broad optima centered around pH 9.0 (data not shown). Near maximal activity was observed for both between pH 8.0 and 9.5, with a marked drop-off below pH 7.5.

PEP Kinetics

PEP saturation curves suggested a sigmoidal response for PEPC₁ (Fig. 4) but hyperbolic for PEPC₂ (Fig. 5). Hill plot analyses confirmed the existence of cooperativity in the binding of PEP to PEPC₁ ($n_H = 1.55$) but not to PEPC₂ ($n_H = 0.81$) (Table II). The $S_{0.5}(\text{PEP})$ of PEPC₁ was 2.23 mM (Table II). This was fourfold greater than the $S_{0.5}(\text{PEP})$ of PEPC₂ which was 0.57 mM. It should be noted that all PEPC activities reported here were determined in the presence of 15% (v/v) glycerol (see "Materials and Methods"). This was because preliminary studies had shown that glycerol substantially reduced the $S_{0.5}(\text{PEP})$ of both isoforms (data not shown). Other authors have cited stabilization of the quaternary structure of

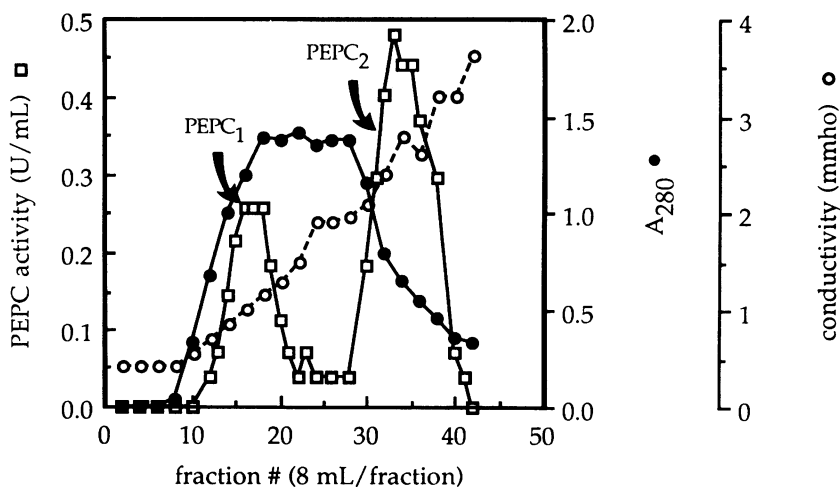


Figure 3. Q-Sepharose chromatography of the 4–20% (w/v) PEG-8000 fraction prepared from a crude extract of *S. minutum*. Details of the procedure are described in the text.

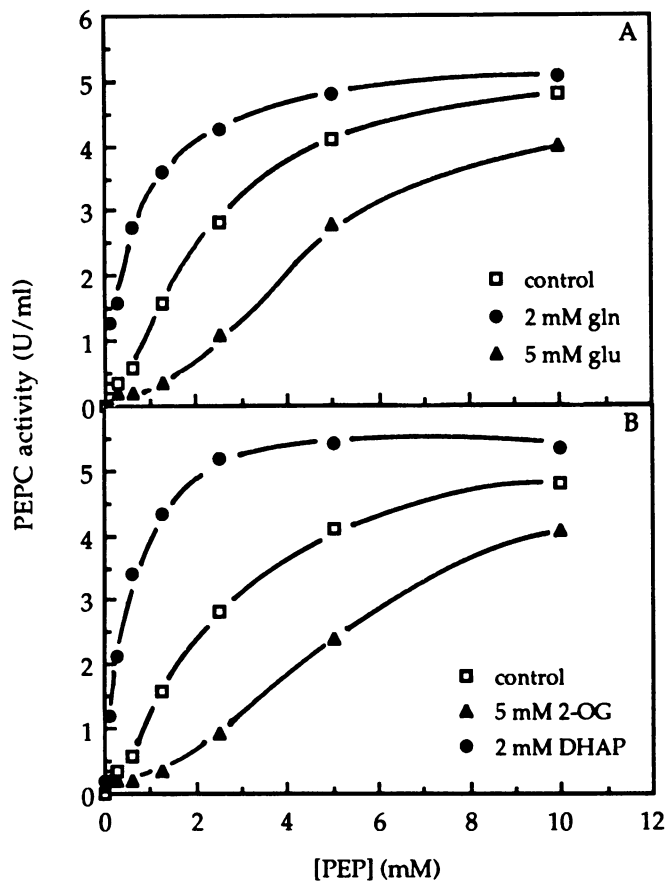


Figure 4. Metabolite effects on PEP saturation kinetics of PEPC₁. The standard spectrophotometric assay was used except that the PEP concentration was varied and the effectors were added at the concentrations shown in the figure.

PEPC, due to exclusion of solvent molecules, as the explanation for the favorable effect of glycerol on the affinity of PEPC for its substrate (28, 29, 31).

Metabolite Effects

A variety of metabolites were tested for effects on PEPC₁ and PEPC₂ at substrating concentrations of PEP (2.5 mM for PEPC₁ and 0.32 mM for PEPC₂). The following had no effect ($\pm 10\%$ of control velocity) on either isoform: glucose-6-P (1 mM), fructose-6-P (1 mM), fructose-1,6-bisP (1 mM), fructose-2,6-bisP (1 mM), acetyl CoA (0.1 mM), Pi (1 mM), and pyruvate (0.4 mM).

Activators

Glutamine markedly activated PEPC₁ but hardly affected PEPC₂ (Table II; Figs. 4A, 5A). At a concentration of 2 mM, glutamine caused a threefold improvement in the affinity of PEPC₁ for PEP (Table II). The K_a (glutamine) of PEPC₁ was 0.2 mM (Table III). DHAP activated both forms of PEPC, but PEPC₁ more than PEPC₂ (Table II; Figs. 4B, 5B). At a concentration of 2 mM, DHAP caused an almost fivefold improvement in the affinity of PEPC₁ for PEP but the change was only threefold for PEPC₂ (Table II). The K_a (DHAP) of PEPC₁ was 1.6 mM. Both activators had their effect on PEPC₁ by reducing the sigmoidicity of the PEP saturation curve (Fig. 4, A and B). Neither activator affected V_{max} (Table II).

Inhibitors

Both isoforms of PEPC were inhibited by glutamate, aspartate, 2-oxoglutarate, and malate (Tables II, III; Figs. 4, 5). PEPC₁ was more sensitive than PEPC₂ to inhibition by glutamate, aspartate, and 2-oxoglutarate, but malate inhibited both forms to a similar extent (Table III). The I_{50} (glutamate) of PEPC₂ was threefold greater than that of PEPC₁, the I_{50} (aspartate) was almost sevenfold greater, and the I_{50} (2-oxoglutarate) was twofold greater (Table III). The two isoforms had similar I_{50} values for malate. The sigmoidicity of the PEP saturation curve for PEPC₁ was greatly enhanced by either glutamate (Fig. 4A) or 2-oxoglutarate (Fig. 4B). All four inhibitors caused marked increases in n_H and/or $S_{0.5}$ (PEP) for

Table II. Metabolite Effects on n_H , $S_{0.5}$ (PEP) and V_{max} of PEPC₁ and PEPC₂

The standard spectrophotometric assay was used except that the PEP concentration was varied and effectors were added at the concentrations shown in the table.

Effector	PEPC ₁			PEPC ₂		
	n_H	$S_{0.5}$ (PEP) mM	V_{max} units/mL	n_H	$S_{0.5}$ (PEP) mM	V_{max} units/mL
Control	1.55	2.23	5.29	0.81	0.57	5.71
2 mM glutamine	1.31	0.79	5.84	1.10	0.32	5.15
5 mM glutamate	2.10	4.26	4.62	0.83	0.75	4.78
2 mM malate	1.56	3.98	5.55	0.85	1.09	5.25
2 mM aspartate	1.70	4.03	5.73	0.96	0.65	5.32
5 mM 2-OG	1.76	6.11	5.80	0.76	1.68	6.08
2 mM DHAP	1.42	0.47	5.56	1.10	0.21 ^a	6.10 ^a

^a Only 1 determination.

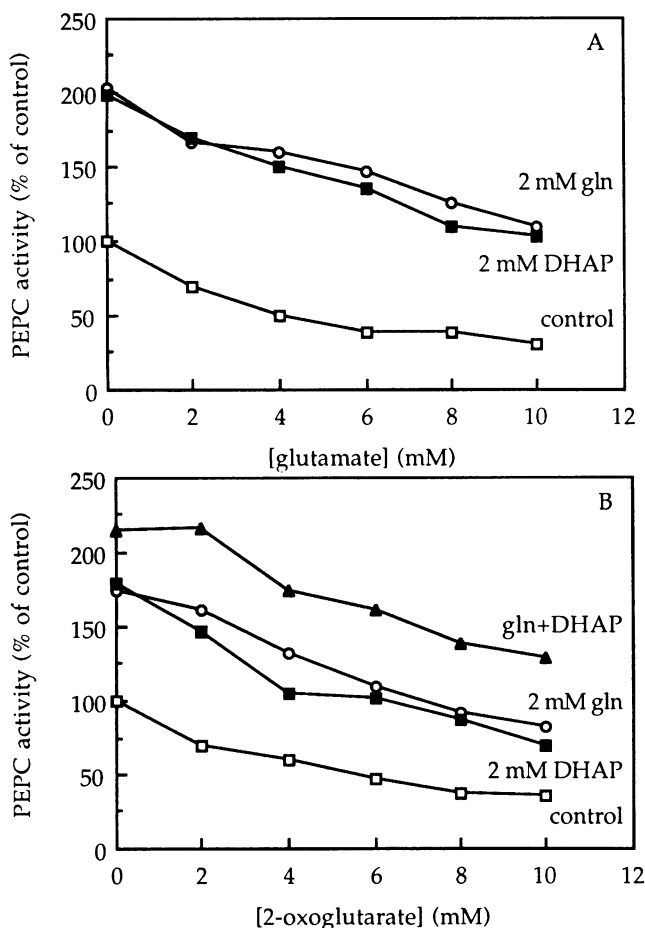


Figure 5. Metabolite effects on PEP saturation kinetics of PEPC₂. The standard spectrophotometric assay was used except that the PEP concentration was varied and the effectors were added at the concentrations shown in the figure.

PEPC₁, but the effects on these parameters were not very pronounced for PEPC₂ (Table II). None of the inhibitors affected V_{max} of either isoform (Table II).

Interacting Effects of Glutamine, DHAP, Glutamate, and 2-Oxoglutarate on PEPC₁

In view of the more dramatic responses of PEPC₁, the effects of combinations of activators and inhibitors on the isoform were studied in some detail at a fixed subsaturating concentration of PEP. Figure 6A shows that either glutamine or DHAP could relieve the inhibition by glutamate. Equivalent results were obtained when 2-oxoglutarate replaced glutamate as the inhibitor (Fig. 6B). Furthermore, the activating effects of glutamine and DHAP were additive, suggesting separate and independent binding sites for these two effectors.

Native Molecular Mass Estimation

All active fractions (PEPC₁ plus PEPC₂) from the Q-Sepharose purification step were combined and concentrated to 2 mL by ultrafiltration through a YM-30 membrane. When the

concentrate was applied to a Superose 6 gel filtration column, PEPC activity eluted in two peaks. The first peak began to elute at the void volume and was very broad making it impossible to obtain a molecular mass estimate. The second peak corresponded to a molecular mass of 279 ± 21 kD (\pm SD, $n = 3$). Experiments where PEPC₁ and PEPC₂ from the Q-Sepharose step were applied separately to the Superose 6 column showed the PEPC₁ corresponded to the low molecular mass form of the enzyme and PEPC₂ corresponded to the very high molecular mass form.

DISCUSSION

Effects of NH₄⁺ Assimilation on *in Vivo* Activity of PEPC

Resupply of NH₄⁺ to N-limited *Selenastrum minutum* cells resulted in an immediate and dramatic (40-fold) increase in the rate of dark carbon fixation (Fig. 1). This was accompanied by a rapid decline in the PEP level and a rapid increase in the malate/PEP ratio (Fig. 2). PEPC activity in crude cell extracts was $30 \mu\text{mol}\cdot\text{mg}^{-1}\text{Chl}\cdot\text{h}^{-1}$ (see "Results"), which was almost sufficient to account for the maximum rate of dark carbon fixation, $40 \mu\text{mol}\cdot\text{mg}^{-1}\text{Chl}\cdot\text{h}^{-1}$ (Fig. 1). This result, coupled with the absence of alternative carboxylating activities, PEP carboxykinase, pyruvate carboxylase, and NAD-malic enzyme (data not shown) indicates that PEPC is the predominant enzyme catalyzing dark carbon fixation in *S. minutum*. This is consistent with ¹³C isotopic discrimination values observed by Guy *et al.* (5). The decline in PEP, the substrate of PEPC, relative to its produce oxaloacetate (assumed to be in equilibrium with malate) (Fig. 2.) implies that the increased activity of PEPC (dark carbon fixation) during NH₄⁺ assimilation is due to activation and not a change in the mass action ratio of this enzyme.

Regulatory Properties of PEPC *in Vitro*

Isolation of Two Isoforms of PEPC

In an attempt to explain the *in vivo* activation of PEPC, the response of this enzyme to various metabolites was studied *in vitro*. This approach was complicated by the fact that two isoforms of PEPC were obtained from *S. minutum* (Table I;

Table III. Kinetic constants for several effectors of *S. minutum* PEPC₁ and PEPC₂

The standard spectrophotometric assay was used except that the PEP concentration was subsaturating (2.5 mM for PEPC₁ and 0.32 mM for PEPC₂).

Effector	PEPC ₁		PEPC ₂	
	I_{50}	K_a	I_{50}	K_a
	<i>mM</i>			
Glutamine		0.2		ND ^a
DHAP		1.6		ND
Glutamate	3.6		11.6	
2-Oxoglutarate	2.1		4.4	
Aspartate	1.4		9.2	
Malate	3.5		2.9	

^a Not determined.

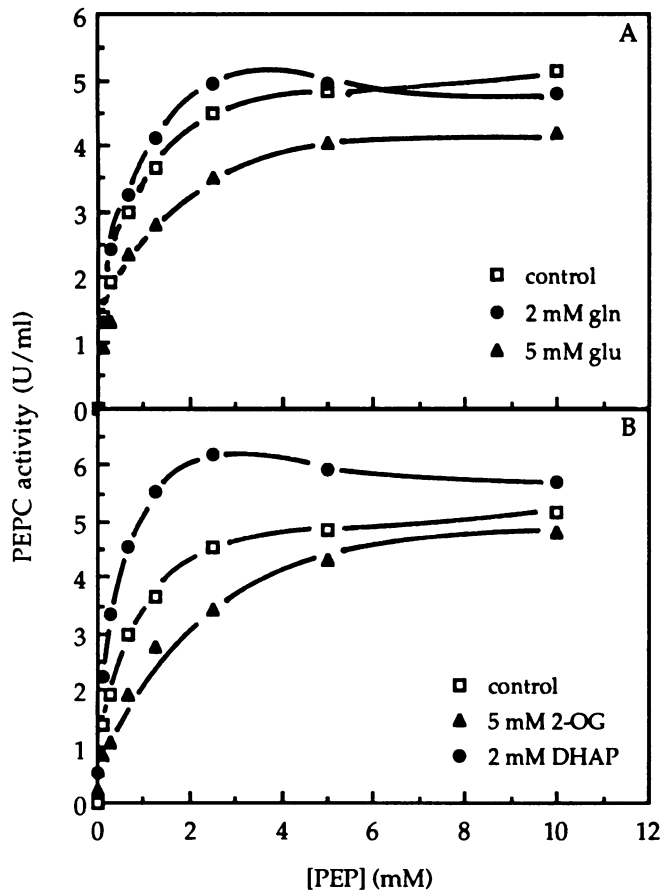


Figure 6. Relationship between PEPC₁ activity and the concentration of glutamate (A) and 2-oxoglutarate (B) as determined in the presence and absence of 2 mM DHAP and/or 2 mM glutamine. The standard spectrophotometric assay was used.

Fig. 3). PEPC₁ was much more responsive than PEPC₂ to most of the regulatory metabolites (glutamine, DHAP, glutamate, 2-oxoglutarate, and aspartate) (Tables II, III; Figs. 4, 5). This could largely be explained by the pronounced sigmoidal PEP saturation kinetics of PEPC₁, which contrasted with the typical Michaelis-Menten response of PEPC₂ (Figs. 4, 5). PEPC₁ had a molecular mass of 279 ± 21 kD as determined by gel filtration chromatography, whereas PEPC₂ was a much larger protein eluting as a broad peak beginning at the void volume. PEPC₁ and PEPC₂ may be distinct isozymes or they may simply be interconvertible forms of the same enzyme. Cytosolic and plastidic isozymes of PEPC have been demonstrated immunologically in two C₃ plants, *Phaseolus* and spinach (26) and interconvertible forms of PEPC are well documented in C₄ and CAM plants (1). However, at present we have no evidence for either of these possibilities in *S. minutum*.

Effect of pH on PEPC₁ and PEPC₂

The pH profiles of the two forms of *S. minutum* PEPC both showed quite high optima, near pH 9. This might be considered consistent with a plastidic location. However, it is more likely that this high pH optimum is related to the pH

stat function of PEPC (12). In other words, as the cytosol becomes more alkaline, PEPC is activated, ultimately resulting in an increase in the malic acid concentration and a return to neutral pH.

Metabolite Effects on PEPC₁ and PEPC₂

PEPC₁ was activated more than 4-fold by 2 mM glutamine and 6-fold by 2 mM DHAP, at a subsaturating PEP concentration of 0.625 mM (Fig. 4). In contrast, PEPC₂ showed only 8% activation by glutamine and 52% by DHAP (Fig. 5). Glutamine has been estimated to increase from 0.13 to 25 mM and DHAP from 57 to 330 μ M following resupply of NH₄⁺ to N-limited *S. minutum* cells (32). Therefore, the glutamine concentration attained *in vivo* is sufficient but the DHAP concentration is somewhat low for maximal activation of PEPC *in vitro* (Table III). The effects of glutamine and DHAP on PEPC₁ are additive (Fig. 6B). Consequently, under the conditions given above (2.0 mM glutamine, 2.0 mM DHAP, 0.625 mM PEP), a 10-fold activation of PEPC₁ could be expected. This is short of the 40-fold activation seen *in vivo* (Fig. 1) but if declining pool sizes of PEPC inhibitors (32) are also taken into account then we may go further toward explaining the *in vivo* activation of this enzyme.

PEPC₁ was more sensitive than PEPC₂ to inhibition by glutamate, aspartate, and 2-oxoglutarate but both forms were similarly inhibited by malate (Table III). Preliminary observations indicate that the effects of these inhibitors are not additive (KA Schuller, unpublished data). Glutamate has been estimated to decline from 5.6 to 2.2 mM and 2-oxoglutarate from 0.73 to 0.13 mM following NH₄⁺ resupply, but aspartate and malate do not change appreciably over the short term (32). Based on these *in vivo* levels, and the I₅₀ values for the various inhibitors (Table III), glutamate and 2-oxoglutarate are likely to be important regulators of *S. minutum* PEPC during enhanced NH₄⁺ assimilation. Aspartate and malate may be less important depending on their distribution between and concentration in the various subcellular compartments.

The properties of PEPC₁ discussed above and the changes in the pool sizes of the various metabolites (32) are consistent with the rapid increase in the rate of dark carbon fixation following NH₄⁺ resupply. However, PEPC₁ constituted only 25% of the total activity recovered in the two isoforms of PEPC from *S. minutum* (Table I). Therefore, more detailed studies of the combined effects of the various activators and inhibitors, at PEP and effector concentrations prevailing *in vivo*, are required to determine whether metabolite regulation is sufficient to explain the activation of PEPC during enhanced NH₄⁺ assimilation or whether other factors are also involved.

Comparison of *S. minutum* PEPC with the Enzyme from Other Algae and Higher Plants

Two isoforms of PEPC isolated from the photosynthetic alga *Euglena* were inhibited by malate, citrate, succinate, and 3-phosphoglycerate (25). PEPC purified from leaves of the C₄ plant *Amaranthus viridis* was inhibited by malate, aspartate, 2-oxoglutarate, and glutamate (7). Neither glutamine nor

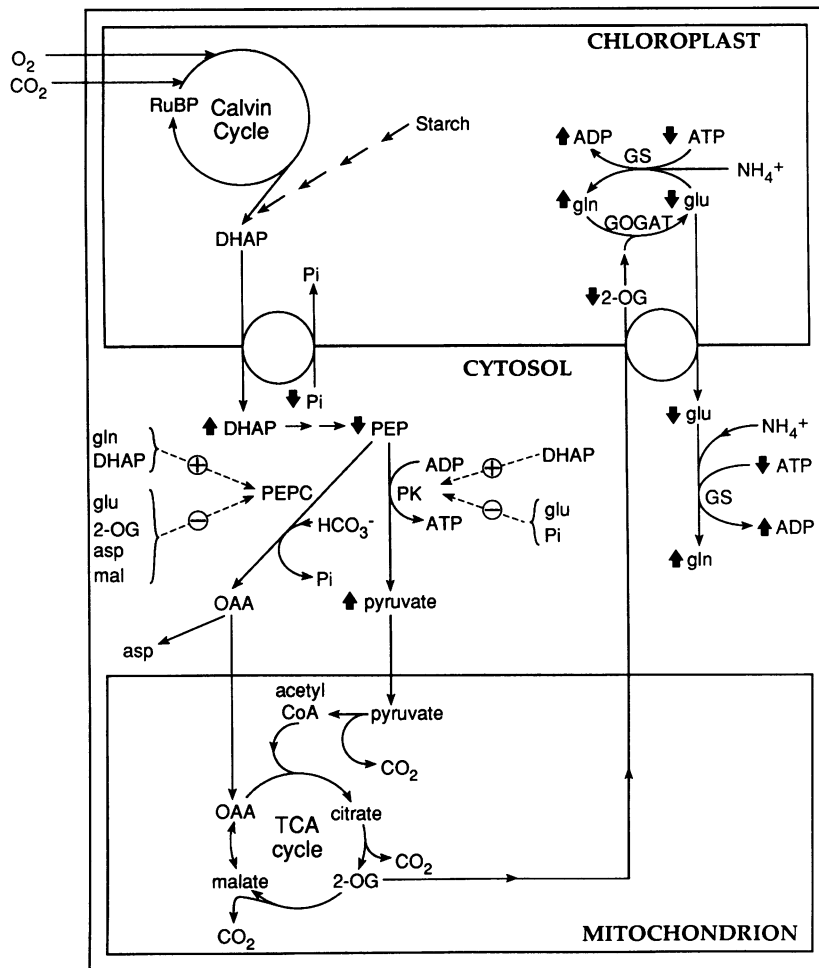


Figure 7. A model for the coordinate regulation of PEPC and PK_c in *S. minutum*. Solid lines represent the flow of metabolites whereas dashed lines indicate regulation of PEPC and PK_c by effectors; (+), activation; (–), inhibition. The broad solid arrows indicate the changes in pool size of the various metabolites following resupply of NH₄⁺ to N-limited *S. minutum* cells (30, 32).

DHAP was tested as an effector of either *Euglena* or *Amaranthus* PEPC. PEPC purified to homogeneity from spinach leaves had a $K_m(\text{PEP})$, extrapolated to infinite Mg^{2+} , of 0.1 mM (19). This is the same order of magnitude as the $S_{0.5}(\text{PEP})$ of PEPC₂ but is very much lower than the $S_{0.5}(\text{PEP})$ of PEPC₁ (Table II). Also like PEPC₂, the PEP saturation curve obtained for the spinach enzyme was hyperbolic (Fig. 5) (19). The native molecular mass of spinach PEPC was 560 kD, which is twice that of PEPC₁. None of the metabolites used in the present study were tested as potential effectors of purified spinach PEPC. PEPC from the nodules of a range of legumes is immunologically related to C₄ leaf PEPC (18). This is surprising since it is thought that an important function of PEPC in nodules is anaplerotic carbon fixation (3) as it is in C₃ plant leaves (12). None of the metabolites tested in the present study had any effect on either alfalfa or soybean nodule PEPC where the same compounds were used (27, 33). In contrast, two isoforms of PEPC from lupin nodules were inhibited by aspartate, malate, 2-oxoglutarate, and glutamate (14). Glutamine had no effect.

Coordinate Regulation of PEPC and PK_c during NH₄⁺ Assimilation

We have presented evidence that *S. minutum* PEPC is activated in response to an increased requirement for the

replenishment of TCA cycle intermediates (particularly 2-oxoglutarate) consumed in the assimilation of NH₄⁺ into amino acids. However, if the TCA cycle is to continue to function under the increased demands imposed by NH₄⁺ assimilation, then there must be an increased input of acetyl CoA as well as oxaloacetate. In this regard, the cytosolic isozyme of pyruvate kinase (PK_c) from *S. minutum* has been shown to be activated by DHAP and inhibited by glutamate (13), properties shared by PEPC. Since DHAP levels rise and glutamate levels fall following NH₄⁺ resupply (30, 32), this would serve to coordinately activate PEPC and PK_c. Activation by DHAP suggests a feed-forward effect due to the activation of glycolysis which has previously been shown to occur following NH₄⁺ resupply to N-limited *S. minutum* cells (32). The decline in glutamate can be explained by its rapid conversion to glutamine (catalyzed by glutamine synthetase) in the presence of excess NH₄⁺.

The unique feature of PEPC is its activation by glutamine. As glutamate and 2-oxoglutarate levels fall following NH₄⁺ resupply, glutamine levels rise (30, 32). The rise in glutamine can be explained by a limitation on GOGAT activity imposed by the falling 2-oxoglutarate level. Rising glutamine and falling 2-oxoglutarate would serve to activate PEPC, resulting in replenishment of TCA cycle intermediates and consequent alleviation of any 2-oxoglutarate limitation on GOGAT activity. Conversely, when NH₄⁺ assimilation ceases, due to NH₄⁺

depletion, glutamine levels would be expected to fall and glutamate levels to rise, thereby inhibiting PEPC. In this way, PEPC activity may be linked to carbon skeleton (2-oxoglutarate) availability for NH₄⁺ assimilation. Figure 7 summarizes the regulatory properties of PK_c and PEPC and integrates them into a model of the anaplerotic function of PEPC during NH₄⁺ assimilation.

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