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Latest advances in dual inhibitors of acetylcholinesterase and monoamine oxidase B against Alzheimer's disease

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ABSTRACT

Alzheimer's disease (AD) is a progressive brain disease characterised by progressive memory loss and cognition impairment, ultimately leading to death. There are three FDA-approved acetylcholinesterase inhibitors (donepezil, rivastigmine, and galantamine, AChEls) for the symptomatic treatment of AD. Monoamine oxidase B (MAO-B) has been considered to contribute to pathologies of AD. Therefore, we reviewed the dual inhibitors of acetylcholinesterase (AChE) and MAO-B developed in the last five years. In this review, these dual-target inhibitors were classified into six groups according to the basic parent structure, including chalcone, coumarin, chromone, benzo-fused five-membered ring, imine and hydrazine, and other scaffolds. Their design strategies, structure-activity relationships (SARs), and molecular docking studies with AChE and MAO-B were analysed and discussed, giving valuable insights for the subsequent development of AChE and MAO-B dual inhibitors. Challenges in the development of balanced and potent AChE and MAO-B dual inhibitors were noted, and corresponding solutions were provided.

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Introduction

AD is a brain disorder hallmarked with intracellular neurofibrillary tangles (NFTs) and extracellular β -amyloid (A β) deposits, which induces progressive memory loss and cognitive dysfunction, afflicting over 50 million people worldwide and posing a tremendous threat on the health of the elderly¹⁻³. Although many hypotheses have been proposed and studied in the process of investigating AD⁴, and some milestone discoveries have been made⁵, the pathological mechanism of AD is still unclear. The cholinergic system, as a primary neurotransmitter system, has been successfully applied in the development of anti-AD drugs. Several AChEls successively approved by U.S. Food and Drug Administration (FDA) were the primary drugs for AD therapy (Figure 1)^{6,7}. Furthermore, increasing evidence indicated that the peripheral anion site (PAS) of AChE was believed to induce $A\beta$ aggregation neurotoxicity, providing a prospect for the development of AChE-based multifunctional anti-AD agents⁸.

The amine homeostasis in the brain was affected by the oxidative deamination of monoamine oxidase (MAO), which was a flavin-containing enzyme present in the outer membrane of mitochondria. MAO could be classified into two types, MAO-A and MAO-B^{9,10}. Recently, the MAO-B has been found to be overexpressed in the hippocampus and cerebral cortex of AD patients¹¹. Moreover, various neurotoxic byproducts produced in MAO-B catalysed biochemical reactions would further induce mitochondrial dysfunction and neuronal death^{12,13}, implying that MAO-B inhibitors can provide beneficial neuroprotective effects for AD therapy by increasing monoamine neurotransmitters and reducing the production of the reactive oxygen species (ROS)¹⁴.

Considering the complexity of AD pathology, the "one drug, one target" paradigm could not achieve the desired therapeutic effects¹⁵. However, the multi-target-directed ligands (MTDLs) strategy, a single multifunctional molecule that could modulate multiple interconnected pathologic pathways, could be an effective way to alter the course of the disease^{16,17}. This approach was thought to be more advantageous than combination therapy because it not only maintained the multiple pharmacodynamics of combination therapy, but also added some additional benefits, such as avoiding potential risks caused by drug-drug interactions, lower risk of adverse effects, and a simplified dosing regimen, which improved compliance¹⁸.

Multifunctional molecules were obtained through three rational design approaches: linkage, fusion, and incorporation (Figure 2). The linkage approach connecting two prototype molecules through a linker was simple to implement and could be applied in structurally incompatible molecular scaffolds. But with the increase in molecular size, this approach might lead to poor drug-like properties including solubility, blood-brain barrier (BBB) permeability, and bioavailability. The fusion approach directly combined two scaffolds without a linker, and the incorporation approach was applied by integrating or overlapping scaffolds^{18–21}. Despite there might be disadvantages and challenges, the MTDLs strategy was still most favourable in terms of multifactorial diseases.

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Figure 2. Three rational design strategies to generate multifunctional molecules.

As the first AChE inhibitor approved for AD therapy, tacrine had been widely applied in MTDLs strategy to obtain potent AChEls with additional pharmacological benefits²². Previous study revealed that Rhein possessed hepatoprotective, anti-A β aggregation, and metal-chelating activities. The potent candidate tacrine-rhein-1 was synthesised by Liu et al.²³ using the linkage approach, and exhibited greater activity against AChE compared to tacrine (IC₅₀ = 27.3 and 135.0 nM, respectively). Tacrine-rhein-1 could bind to Cu²⁺ and inhibit the A β aggregation induced by AChE (70.2% at 100 µM). Notably, compared to tacrine, lower levels of aspartate aminotransferase and alanine aminotransferase were observed in mice serum after administration of tacrine-rhein-1, indicating lower hepatotoxicity.

Given that A β aggregation and ROS production induced by metal ions were two significant neurotoxic events in AD²⁴, and metal-induced A β aggregation was reversible²⁵, thus, chelation therapy by administration of metal-specific chelating ligands seemed to be a reasonable choice for the treatment of AD²⁶. Thioflavin-T, a fluorescent dye binding to A β , and metal chelator di(2-picolyl)amine were chosen to synthesise FC-1 through the fusion approach²⁵. As expected, FC-1 was not only capable of inhibiting the A β_{40} aggregation but also resulting in the dissociation of Cu²⁺ and Zn²⁺-triggered A β_{40} aggregates. Rodríguez-Rodríguez et al.²⁷ chose the incorporation approach of integrating the features of thioflavin-T and clioquinol to synthesise a multifunctional candidate HBXI that could inhibit Cu²⁺-promoted A β_{40} aggregation and efficiently chelate Cu²⁺ (pCu = 11.06, pClioquinol = 10.77) and Zn²⁺ (pZn = 10.98, pClioquinol = 11.22) ions.

Therefore, in recent years, the concept of AChE/MAO-B dual inhibitors has gained significant interest for the treatment of AD²⁸, and numerous AChE and MAO-B dual inhibitors designed using MTDLs strategy have been well-documented. In particular, several novel techniques played an important role in the discovery of new hit inhibitors.

In recent years, nanotechnology, especially magnetic nanoparticles (MNPs), has been widely applied in immobilisation due to their unique properties. Immobilising important enzymes onto nanoparticles can provide comprehensive benefits²⁹. In this context, the Zhang group³⁰ employed AChE-immobilized MNPs as a tool to screen AChEIs from natural products and found an amentoflavone, which displayed more potent AChE inhibition (IC₅₀ = $0.73 \,\mu$ M) than tacrine. Li and colleagues³¹ created a non-conjugated polymer dots (NCPDs)-MnO₂ biosensing system to screen AChEIs, which exhibited a broader detection range (12.3–3675 U L⁻¹ for AChE), and an impressive low limit of detection (LOD) at 4.9 U L⁻¹, showing a promising prospect in the discovery of new AD therapy drugs.

Computer-aided drug design (CADD) has become an indispensable tool for discovering lead compounds and providing useful structural information of receptor-ligand complexes for subsequent research³². Zhou group³³ discovered a well-known natural product kaempferol as an AChE inhibitor, while another group³⁴ identified this compound as a selective human MAO-A inhibitor. Gogineni et al.³⁵ adopted this strategy to enhance the selectivity of flavonoid acacetin 7-O-methyl ether, resulting in a thousandfold increase of the selectivity index for MAO-B. In addition, Wang group³⁶ adopted the Three-Dimensional Biologically Relevant Spectrum (BRS-3D), a representative compound database used in CADD, to evaluate the subtype selectivity of MAO inhibitors, observing a thousand-fold selectivity index improvement for MAO-B.

Light irradiation has extensive application in the field of disease treatment such as the Photodynamic therapy (PDT)³⁷, a therapeutic method for various cancers. Interestingly, Paolino et al.³⁸ designed a small library of AChE and MAO-B dual inhibitors that could modify their activity through light irradiation, representing a novel exploration in the design of dual inhibitors. In summary, these novel techniques expedite lead compound discovery and expand the pool of potential candidates for AD treatment.

This review provided a summary and discussion of AChE and MAO-B dual inhibitors published from 2018 to 2022, including the design strategies employed, the SARs, molecular docking studies, the challenges faced and corresponding solutions.

The AChE and MAO-B binding sites

As a serine hydrolase, AChE was extensively distributed in the peripheral and central nervous system and was responsible for hydrolysing acetylcholine (ACh)³⁹. AChE was an ellipsoidal protein with dimensions of approximately 45 Å × 60 Å × 65 Å. It possessed a deep and narrow gorge (~20 Å) formed by 14 aromatic residues (e.g. Ser200, Phe289, Tyr121, and Gly199) consisting of two binding sites, known as the PAS and the catalytic anionic site (CAS)⁴⁰. The PAS guided ACh into the CAS site through the "cation- π " interaction, and ACh was subsequently hydrolysed at the CAS site⁴¹.

MAO-B was responsible for catalysing the metabolism of monoamine neurotransmitters, and one of its key binding sites was the flavin adenine dinucleotide (FAD)¹⁴. The binding pocket of MAO-B (~700 Å³) was divided into the entrance cavity and substrate cavity by the lle199 and Tyr326 residues⁴². The flexible lle199 residue was pivotal in controlling the entry of ligands into the enzyme cavity⁴³⁻⁴⁵. In addition, a smaller hydrophobic entrance cavity, lined with specific residues Tyr398 and Tyr435, was adjacent to the substrate cavity, could generate an "aromatic cage", interacting with the amino group of the substrate through π -cation interactions^{44,46}.

AChE and MAO-B dual inhibitors

The fundamental principle for designing AChE and MAO-B dual inhibitors was to incorporate the pharmacophores of anti-AD molecules or drugs (Figure 3). For example, the cholinesterase (ChE) inhibitory moiety of donepezil was amalgamated into the potent MAO-B inhibitor PF9601N to create the lead ASS234, which was further modified to obtain a highly stable candidate Contilisant in human liver microsomes. The carbamate group of rivastigmine was merged into potent MAO-B inhibitors M30 and rasagiline, synthesising the M30D and Ladostigil, respectively^{47–50}. In particular, Ladostigil displayed a brain-selective inhibitory profile on MAO in an *in vivo* study. When administered to mice (26 mg·kg⁻¹) for two

weeks, it resulted in approximately 70% inhibition of brain MAO activity, but there was almost no MAO inhibition in the liver and small intestine⁵¹. This unique advantage enabled it to avoid side effects caused by cheese reactions during the treatment of AD⁵². Recently, a Phase III clinical trial of Ladostigil to treat mild cognitive impairment (MCI) suggested that the combination of AChE and MAO-B inhibitors is a promising treatment for AD⁵³.

Chalcone-based inhibitors

Chalcone, a compound found in many plants, is composed of two benzene rings separated by three carbon atoms⁵⁴. In this paper, the aromatic ring connected to the carbonyl group is named the A ring, while the other benzene ring is defined as the B ring. Chalcone and its analogs have attracted great interest for decades on account of their diverse pharmacological properties such as antitumor, anti-inflammatory, anti-diabetic, antioxidant, and antimicrobial agents⁵⁵. In particular, certain natural and synthesised chalcone derivatives (Figure 4) exhibited both AChE and MAO-B inhibitory activities.

Oh et al.⁵⁶ reported that compouds 1, 2, 4, 5, and 6 exhibited moderate AChE inhibitory activities ($IC_{50} = 2.79$, 1.25, 6.07, 6.02, and 2.46 µM, respectively) and potent MAO-B inhibitory effects $(IC_{50} = 0.082, 0.066, 0.029, 0.061 \text{ and } 0.075 \,\mu\text{M}, \text{ respectively})$. But compound **3** showed weak activity against both enzymes ($IC_{50} =$ 31.30 and 10.06 µM, respectively). This difference was attributed to the formation of hydrogen bonds between Cys172 of MAO-B (PDB: 4A79) with 1, 2, 4, 5, and 6, but not with 3, probably due to the steric hindrance and possible charge effect caused by the carboxymethyl group. Interestingly, 3 displayed weaker AChE inhibition despite forming a hydrogen bond with AChE (PDB: 1GQS) Phe288, which was not observed in 1, 2, 4, 5, and 6. In addition, the inhibitory activities of both enzymes decreased or disappeared when the 2' position of the A ring was substituted with methoxy $(7, 8)^{46}$, ethoxy $(9)^{57}$, morpholine $(10)^{58}$ or piperazinyl (11)⁵⁹ groups due to the blocked interaction between the A ring and two enzymes.

Furthermore, both inhibitory activities decreased sharply (**12**)⁶⁰ when the A ring underwent cyclisation using a benzothiazine ring, while AChE inhibitory activity significantly increased when the A ring was cyclized by 1, 3-dioxolane and the B ring was replaced by thiophene ring (**13**)⁶¹ (Figure 5). These observations could be explained through docking studies of **13**. The 1,3-benzodioxole ring established π - π interactions with Tyr72 and Trp286 of AChE (PDB: 4EY7), and Tyr326 of MAO-B (PDB: 2V5Z), a hydrogen bond with Tyr188 of MAO-B, respectively. Additional π - π interactions were observed between the thienyl ring and Tyr341 of both AChE and MAO-B.

Chalcone and donepezil-based dual inhibitors

Donepezil has been extensively investigated in the development of multifunctional anti-AD candidates for its good bioavailability, high selectivity, and low toxicity (Figure 6). In light of this, several moderate and balanced dual inhibitors of AChE and MAO-B were synthesised by Sang et al. ⁶² via a linkage approach. Compound **14** showed potent AChE inhibitory activity ($IC_{50} = 0.41 \mu$ M) with moderate MAO-B inhibitory activity ($IC_{50} = 8.8 \mu$ M). Moreover, it revealed antioxidant and metal chelate abilities. A docking study revealed that **14** interacted with AChE (PDB: 1EVE) through hydrogen bonds and π - π interactions established between its hydroxyl, methoxy groups with Arg289, Tyr334. Besides, an important intermolecular hydrogen bond was found between the *N*-(2-



Figure 3. Structures of promising MTDLs.

		С	IC ₅₀ ·(µM)↔	
	1 $R^1 = C1, R^2 = R^3 = R^4 = R^5 = H$	Compounds	AChE	MAO-B _e
R^3 B R^2	2 $R^1 = NO_2, R^2 = R^3 = R^4 = R^5 = H$	1.0	2.79 ± 0.11+	0.082 ± 0.0016
	3 R^1 =CH ₂ COOH, R^2 = R^3 = R^4 = R^5 =H	2.0	1.25 ± 0.42	0.066 ± 0.0025
	4 $R^1 = N(CH_3)_2$, $R^2 = R^3 = R^4 = R^5 = H$	3.0	31.30±·1.65+	10.06 ± 3.44
	5 $R^1 = N(CH_3)_2$, $R^2 = R^3 = R^4 = H$, $R^5 = Cl$	4.0	6.07±0.17+	0.029 ± 0.0028
	6 R ¹ =N(CH ₃) ₂ , R ² =R ⁴ =H, R ³ =R ⁵ =Cl	50	6.02 ± 0.04	0.061 ± 0.0021
	7 $R^1 = R^5 = OH, R^2 = R^3 = H, R^4 = OMe$	6.0	$2.46 \pm 0.078_{\circ}$	0.075 ± 0.0028
	8 $R^1 = R^2 = R^5 = OH, R^3 = H, R^4 = OMe$	7.0	> 40.0	0.38 ± 0.023+
	9 $R^1 = R^2 = R^3 = R^4 = H, R^5 = OC_2H_5$	8.0	25.01 ± ·1.03+	1.01 ± 0.12*
	10 $R^1 = R^2 = R^3 = R^4 = H, R^5 = morpholine$	9₽	>. 50.	0.57±0.030+
	11 $R^1 = R^2 = R^3 = R^4 = H$, $R^5 = piperazinyl$	10 <i>e</i>	$12.07 \pm 1.18_{\circ}$	0.32 ± 0.16
		11 +	No inhibition.	1.37 ± 0.0027

Figure 4. Structures of natural and synthesised chalcone.



Figure 5. Structures of compound 1-based derivatives.

methoxybenzyl) ethylamine fragment and Ser122. In addition, several intermolecular hydrogen bonds (e.g. between the methoxyl group and Ala325) were also observed between the **14** and MAO-B (PDB: 2V60).

To develop improved anti-AD candidates, **15** and **16** were synthesised by the same research group using a linkage approach^{63,64}. As expected, these compounds displayed more balanced AChE (IC₅₀ = 0.13 and 1.3 μ M, respectively) and MAO-B



Figure 6. Structures of chalcone and donepezil-based dual inhibitors.

inhibitory activity (IC_{50}~=~1.0 and 0.57 $\mu\text{M},$ respectively), along with excellent anti-self-induced $A\beta$ aggregation and potent antioxidant abilities. What's more, 15 revealed no neurotoxicity (1000 mg·kg⁻¹) in mice, and demonstrated improvement in memory impairment induced by scopolamine. The 2-(dimethylamino)-N-ethyl-benzenemethanamine moiety of 15 could enter into the CAS and interact with AChE (PDB: 4EY7) Trp286. A hydrogen bond was formed between the hydroxyl group on the A ring and Tyr314, and π - π interactions were established between the A ring and Trp286, as well as Tyr341. In addition, hydrophobic interactions were also observed with several AChE residues (e.g. Tyr337, Trp286, and Trp86). The 2-hydroxy-acetophenone moiety of 15 generated two hydrogen bonds with Phe168, and Cys172 of MAO-B (PDB: 2V60). Additionally, the aromatic rings (A and B rings of chalcone) established π - π interactions with Leu171 and Tyr435 of MAO-B. These interactions were likely resposible for the MAO-B inhibitory activity exhibited by 15. Compound 16 fully occupied the pocket of AChE (PDB: 1EVE), with the A ring forming a π - π interaction with Phe330. The oxygen atom of the B ring formed an intermolecular hydrogen bond with Phe288. In addition, the hydroxyl group generated two hydrogen bonds with Ile198, and Gln206 of MAO-B (PDB: 2V60), respectively. The carbonyl moiety generated hydrogen bonds with Ile199 and Cys172, and the B ring also established π - π interactions with MAO-B residues, Tyr398 and Tyr435.

Other chalcone-based dual inhibitors

Compound **17**, synthesised by Sang et al.⁶⁵ using the fusion approach, is a carbamate derivative of chalcone. Unfortunately, It showed no significant AChE inhibition $(26.2 \pm 0.89\% \text{ at } 25 \,\mu\text{M})$, but moderate MAO-B inhibition $(IC_{50} = 1.3 \pm 0.06 \,\mu\text{M})$ in *in vitro* biological test. In the molecular docking study of **17**, limited interactions between **17** and AChE (PDB: 1EVE) were observed, which was consistent with its extremely weak AChE inhibitory activity. However, it was observed that the hydroxyl and carbonyl groups formed hydrogen bonds with two key residues lle199 and Tyr326 of MAO-B (PDB: 2V60), respectively. Moreover, the oxygen atom and carbonyl group of the carbamate fragment established hydrogen bonds with Tyr398.

Phenolic Mannich bases have demonstrated various neuroprotective effects including antioxidant⁶⁶, AChE inhibition⁶⁷, and metal chelation⁶⁸. Most importantly, the conversion of these bases into ammonium salts resulted in better solubility⁶⁹. In light of this, several research groups created a series of chalcone-polyphenol Mannich bases using the fusion approach (Figure 7). Compound 18 exhibited balanced inhibitory activity against the target enzymes, with an IC₅₀ value of 7.15 μ M for AChE and 0.43 μ M for MAO-B70. In the Torpedo californica AChE (TcAChE)-18-complex (PDB: 1EVE), the phenylacetate group occupied the PAS by forming a key hydrogen bond with Ser286, and the protonated piperidine fragment bound to the CAS through a hydrogen bond and salt bridge with Asn85 and Asp72, respectively. In addition, the chalcone skeleton could bind to the mid-gorge through π - π interaction with residues such as Tvr334. On the other hand, in the docking study of MAO-B (PDB: 2V5Z), the 3'-OH on the B ring of 18 established a hydrogen bond with Gly101 and the piperidine moiety formed a salt bridge with Glu84. Moreover, the A ring occupied the entrance cavity and interacted with the key residue Ile199, while a π - π interaction was also found in the substrate cavity between the phenyl at 4-position and Tyr398.

Another structurally simplified chalcone-Mannich base derivative **19** demonstrated similar AChE and MAO-B inhibitory potency $(IC_{50} = 0.44 \text{ and } 1.21 \,\mu\text{M}$, respectively)⁷¹. In addition, it exhibited antioxidant activity, the ability to inhibit A β aggregation and metal-chelating properties. The molecular modelling study revealed that **19** could interact simultaneously with the two binding sites (CAS and PAS) of AChE (PDB: 1EVE). The carbonyl moiety and 3'-OH group generated hydrogen bonds with Tyr435 and Ile199 of MAO-B (PDB: 2V60), respectively. Furthermore, A ring established a π - π interaction with Tyr398.

Drugs containing oxime ether fragments are widely administrated for various diseases, including antibiotic, antifungal, and anti-inflammatory. Oh et al.^{72,73} synthesised a series of chalcone oxime ethers using the fusion approach and evaluated their inhibitory potency on AChE and MAO-B (Figure 7). Among them, compound **20** exhibited the most balanced inhibitory against both enzymes ($IC_{50} = 4.39$ and 0.028 μ M, respectively), and displayed superior MAO-B inhibitory potency compared to the positive drug lazabemide ($IC_{50} = 0.042 \,\mu$ M). Simplification of the oxime ether moiety in compound **21** led to **a** significant decrease in AChE inhibitory potency ($IC_{50} = 0.012 \,\mu$ M). A similar result was observed when the A ring of chalcone was decorated with the dioxime ether (**22**, 11.2% at 10 μ M for AChE; $IC_{50} = 0.018 \,\mu$ M



Figure 7. Structures of other chalcone-based dual inhibitors.

for MAO-B). In the molecular docking studies, these chalcone oxime ethers could interact with the aromatic cage composed of FAD, Tyr435, and Tyr398, while the B ring established π - π interaction with the MAO-B (PDB: 2V5Z) selective residue Tyr326.

In an earlier study by Kumar's team⁷⁴, they developed novel pyrimidine derivatives as multi-target anti-AD agents by cyclizing the three-carbon fragment of chalcone using the incorporation approach. Most of them exhibited potent selective AChE inhibition, but weak inhibitory potency on MAO-B (e.g. **23**, IC₅₀ = 0.020 μ M and 7.31 μ M, respectively). To enhance the inhibitory potency on MAO-B of these compounds, the morpholine moiety was replaced with propargyl group⁷⁵. However, *in vitro* assays showed that the optimised compound exhibited only a slightly increase in inhibitory potency against MAO-B, but a dramatic decrease in AChE inhibitory activity (**24**, IC₅₀ = 1.49 and 1.74 μ M, respectively), suggesting that the morpholine moiety was indispensable for the potent AChE inhibitory activity.

3.1.3. Summary of SARs

Based on the SARs of chalcone derivatives (Figure 8), the AChE inhibitory activities decrease with the substitution of hydroxyl, alkoxy groups at 2-position, morpholine, piperazine groups or oxime ether moiety at the 4-position of A ring. Meanwhile, the cyclisation of A ring into a 10*H*-phenothiazine scaffold with greater rigidity also leads to decreased AChE inhibitory activity. Additionally, the substitution with a hydroxyl group at the 4'-

position, or carbamate or oxime ether moieties at the 3'-position of B ring results in the decrease of AChE inhibitory activities. Conversely, the AChE inhibitory activities increase when the substitution with chlorine, N,N-dimethylamino or aminoalkyl groups at the 4'-position of B ring. The MAO-B inhibitory potency increases when substituted with oxime ether moiety or propargyl group at the 4-position of A ring, or chlorine, nitro groups at the 4'-position of B ring. Conversely, the MAO-B inhibitory activity decreases when substituted with morpholine, or piperazine groups at 4-position at A ring. Substitution with the oxime ether moieties in the B ring does not significantly affect MAO-B inhibitory activities. The substitution of the carboxymethyl group at the 4'-position of ring B was disadvantageous for both AChE and MAO-B inhibition. In addition, the cyclisation of the carbonyl group of chalcone yields potent and balanced AChE and MAO-B dual inhibitors. The AChE inhibitory activities is determined by crucial residues such as Arg289, Ser122, Tyr314, Phe330, and Trp286 of AChE, which facilitate the A ring of chalcones to bind to the middle gorge and CAS through hydrogen bonds or $\pi - \pi$ stacking interactions. Other important residues, including Trp286, Phe288, and Tyr341 in the PAS, can interact with the B ring via the π - π interactions. In addition, the α,β -unsaturated carbonyl group can form hydrogen bonds with specific residues. On the other hand, MAO-B inhibitory activities are influenced by the residues Ile199, Cys172, Tyr 435, Tyr326, and Tyr398 of MAO-B, which enable the chalcone derivatives to enter the active pockets by forming hydrogen bonds and $\pi - \pi$ interactions.

Substitution at the **2-position** of **A ring** Hydroxyl and alkoxy groups decrease AChE inhibition. Substitution at the **4-position** of **A ring** Morpholine, piperazine, oxime ether moieties, or cyclization into 10H-phenothiazine scaffold reduce AChE inhibition; Morpholine and piperazine groups decrease MAO-B inhibition, while oxime ether and propargyl moieties increase it. **Arg289, Ser122, Tyr314,** and **Trp286** of **AChE**, along with **Tyr398** and **Tyr 435** of **MAO-B** are crucial for inhibition through hydrogen bonds and π -

Figure 8. SAR of chalcone-based dual inhibitors targeting AChE and MAO-B.

Coumarin-based inhibitors

 π interactions with A ring.

Coumarins, characterised by 1,2-benzopyranone or benzopyran-2-one, are widely found in various plants. Studies have demonstrated that natural coumarins and synthesised analogs exhibit anti-AChE and MAO-B activities^{76–81}. Coumarins have been reported to bind to PAS of AChE, suggesting their potential in designing and synthesising new analogs with excellent and balanced AChE and MAO-B inhibitory activities. As a result, numerous dual inhibitors targeting both enzymes have been developed by incorporating coumarin core with some catalytic site interacting moieties.

Modifications at position 3 or 4 of coumarin

Structural decoration at position 3 or 4 of coumarin has been a significant strategy to develop effective AChE and MAO-B inhibitors (Figure 9). In previous studies, representative compounds **25** and **26**, modified on the 3 position of coumarin bearing an amide linker by using the linkage approach, showed weak AChE inhibition ($IC_{50} = 591.44$ and 69.47 μ M, respectively), but remarkable MAO-B inhibition at the nanomolar level ($IC_{50} = 4.6$ and 760 nM, respectively)^{82,83}. However, **27** and **28** displayed potent and balanced inhibitory activities against both AChE and MAO-B ($IC_{50} = 0.163$ and 0.068 μ M for AChE; 8.13 and 6.31 μ M for MAO-B, respectively) when alkylamine served as the linker⁸⁴, suggesting that the alkylamine linker is more advantageous for developing coumarin-based dual inhibitors targeting these enzymes.

Rehuman et al.⁸⁵ synthesised a series of coumarin-chalcone compounds using the fusion approach to introduce α,β -unsaturated ketone unit of chalcone at the 3 position of coumarin and evaluated their AChE and MAO-B inhibitory potencies. Among them, **29** displayed weak AChE inhibiton (IC₅₀ = 40.1 µM), but potent MAO-B inhibitory activity with an IC₅₀ value of 0.51 µM. The weak AChE inhibitory activity might be due to the inability of **29** to reach the CAS centre of AChE (PDB: 4EY7) and interact with Trp82.

In addition, Zhang *et al.* also modified the 4 position of coumarin using the linkage approach to obtain the representative compound **30**, which showed effective AChE inhibition with an IC₅₀ value of 0.84 μ M, but displayed weak MAO-B inhibitory potency (IC₅₀ = 12.31 μ M)⁸⁴.

Modification at position 7 of coumarin

In most studies, the focus has been on designing dual inhibitors targeting AChE and MAO-B by using the linkage approach to introduce moieties at position 7 of coumarin, which bound to the CAS site (Figure 9). However, when an amide linker was used as the connecting group, the inhibitory activity of the compounds against AChE and MAO-B was not balanced, as observed with **31** and **32**⁸⁶. When an aromatic amide was substituted at position 7 of coumarin, **31** Substitution at the **3'-position** of **B ring** Carbamate and oxime ether moieties decrease AChE inhibition. Substitution at the **4'-position** of **B ring** Hydroxyl group decrease AChE inhibition, while chlorine, *N*,*N*dimethylamino, and aminoalkyl groups enhance it. Chlorine and nitro groups elevate MAO-B inhibition. Carboxymethyl group decrease both AChE and MAO-B inhibition. **Trp286, Phe288** of **AChE**, coupled with **Cys172, Ile199**, **Tyr326** of **MAO-B**, are pivotal for inhibition, establishing

showed no significant AChE inhibition (45–50% at 100 μ M), while the MAO-B inhibitory activity of compound **32** was almost disappeared (IC₅₀ > 100 μ M) when fatty amide was decorated at the same position. This could be attributed to the inability of **31** and **32** to simultaneously bind to the active sites of both AChE and MAO-B.

hydrogen bonds and π - π interactions with **B** ring.

On the other hand, 33⁸⁷, 34⁸⁸, 35 and 36⁸⁹, displayed potent and balanced activity against both enzymes ($IC_{50} = 0.55$, 0.114, 1.3, and 1.4 μ M for AChE, respectively; IC₅₀ = 0.0082, 0.101, 0.051 and 0.076 µM for MAO-B, respectively) when the CAS binding groups were introduced at the position 7 of coumarin through a flexible alkyl ether linker. Molecular docking studies have shown that these molecules occupied the active sites of the target enzyme and interacted with key residues of AChE (PDB: 4EY7) such as Trp286, Tyr341, Trp86, Tyr337, Gly448, His447, and Gly121. Meanwhile, the coumarin core was accommodated within the substrate cavity of MAO-B (PDB: 2V61) and established interactions with residues such as Tyr398, Phe343, Tyr60, and Tyr435. The carbonyl mojety of coumarin formed hydrogen bonds with residues such as Tyr188. Other fragments of these compounds were observed to occupy the entrance cavity of MAO-B and interacted with residues including Pro102, Pro104, and Ile316.

Summary of SARs

Coumarin is an extremely valuable skeleton for the discovery and development of highly potent AChE and MAO-B dual inhibitors. SARs of coumarin derivatives, as shown in Figure 10, reveal that substitution with benzylamine derivatives at the 3- or 4-position of the coumarin scaffold can significantly improve the AChE inhibitory activities. AChE inhibitory activities decrease sharply if the benzylamine groups at the 3-position of the coumarin core are replaced with benzamide moieties, but MAO-B inhibitory activities can be remarkably increased. The introduction of α,β -unsaturated ketone moiety at the 3-position of the coumarin core reduces the AChE inhibitory activities of coumarin derivatives. Substitution with benzamide moieties at the 7-position of the coumarin core makes it difficult to obtain balanced AChE and MAO-B dual inhibitors. On the contrary, the introduction of flexible AChE inhibitory pharmacophores at the same 7-position of coumarin is advantageous for obtaining potent and balanced AChE and MAO-B dual inhibitors. Furthermore, the presence of the N-benzylpiperidine or N-alkyloxy moiety at the 7-position of the coumarin has no significant effect on the AChE inhibition, but exhibited a remarkable impact on the MAO-B inhibition, suggesting that decoration of the protonatable portion has a greater influence on MAO-B affinity compared to AChE.

The Trp286 in AChE PAS plays an important role in interacting with the coumarin core through a π - π interaction. While, residues Trp86, His447, and Gly121 in AChE CAS interact with the protonatable basic moieties through cation- π interactions. The Tyr188 at





Figure 9. Structures of coumarin-based inhibitors.

Substitution at the **7-position** of coumarin core Incorporating *N*-benzylpiperidine or *N*-alkyloxy moieties leads to enhanced and balanced AChE and MAO-B inhibition. Decorating the protonatable moieties has a greater influence on MAO-B. Benzamide moieties disturb AChE and MAO-B inhibition balance. **Trp286**, **Gly121**, **Trp86**, and **His447** play a key role in AChE inhibition through π - π and cation- π interactions with coumarin core and protonatable moieties. **Tyr188**, **Pro102**, **Pro104**, and **Ile316** facilitate coumarin entry into MAO-B's acityc cavities.



the bottom of the substrate cavity of MAO-B can form a hydrogen bond with the carbonyl oxygen of the coumarin, and certain residues (e.g. Tyr398, Tyr435) stabilise coumarins by establishing hydrophobic interactions. Additionaly, residues Pro102, Pro104, and Ile 316 enable coumarins to enter into the entrance cavity of MAO-B.

Chromone-based inhibitors

3.3.1. Modification of chromones

Chromone (4*H*-chromen-4-one), an isomer of coumarin (2*H*-chromen-2-one), is a kind of significantly important scaffold due to its potential for modification, generating chromone derivatives with diverse biological activities. For instance, compounds **37** and **38** (Figure 11) displayed potent MAO-B inhibitory activity, and were synthesised by the introducing carboxyl group onto the chromone $core^{90}$. The screening results showed that the position of the



carboxyl group is critical for the selectivity of MAO inhibitors. Compound **37**, with the carboxyl group substituted at position 3 of the chromone core, displayed selective MAO-B inhibition with an IC₅₀ value of 48 nM (SI >2083) due to a crucial hydrogen bond established between the carboxyl group and Tyr326. However, isomer **38**, with the same carboxyl substitution at position 2, showed weaker MAO-B inhibitory potency (IC₅₀ > 100 nM). Therefore, previous studies^{91,92} have used the linkage approach to develop selective MAO-B inhibitors (e.g. **39–41**) featured a carboxylamide linker substitution at the 3-position of the chromone core. However, recent studies have focused on the synthesis of numerous chromone derivatives to obtain multifunctional anti-AD lead compounds (Figure 11).

By using the fusion approach to introduce AChE inhibitory pharmacophores, two compounds (**42** and **43**) bearing carboxamide substitution at position 3 of chromone were synthesised, and their inhibitory activities against the target enzymes were



Figure 11. Structures of chromone-based inhibitors.

evaluated^{93,94}. Both compounds were AChE and MAO-B dual inhibitors, but compared to 42 (IC₅₀ = $3.69 \,\mu$ M), 43 displayed more potent AChE inhibitory activity (IC₅₀ = $0.37 \,\mu\text{M}$), suggesting that the N-benzylpiperidine moiety, serving as the pharmacophore of AChEls, might be more noteworthy in the design of AChE and MAO-B dual inhibitors. Furthermore, 43 was proved to cross the BBB and show safety on rat pheochromocytoma (PC12) cells. Docking studies of **43** revealed that the π -cation and π - π interactions were generated between the N-benzylpiperidine moiety and Trp84, as well as Phe330 in CAS of AChE (PDB: 1EVE). Additionally, the chromone fragment established π - π interaction with Trp279 in PAS. In the case of MAO-B (PDB: 2V60), the benzyloxy moiety established $\pi - \pi$ interactions with Tyr435 and Tyr398, while the amide carbonyl group formed a hydrogen bond with Tyr326. In contrast, the N-benzylpiperidine fragment was located in the entrance cavity.

Interestingly, compared to **42**, using the fusion approach, dual inhibitor **44** showed more potent AChE inhibitory activity ($IC_{50} = 0.21 \,\mu$ M), but weaker MAO-B inhibitory potency ($IC_{50} = 3.81 \,\mu$ M) when the carboxamide linker was substituted at position 2 of chromone core⁹³. In addition, compound **45** obtained using the fusion approach with carboxamide group substituted on the same position of chromone core, displayed balanced inhibition towards both enzymes with IC_{50} values of 1.0 and 8.1 μ M, respectively⁹⁵. This indicated that the introduction of AChE inhibitory moieties, specifically when the carboxamide group was substituted at position 2 of the γ -pyrone nucleus, may lead to a dual inhibitor with stronger AChE inhibitory activity.

Łażewska et al.⁹⁶ synthesised potent dual inhibitors **46** (IC_{50} = 180, and 775 nM, respectively) and **47** (IC_{50} = 136, and 897 nM, respectively) using the incorporation approach to cyclize the chromone skeleton. The docking studies indicated that the azepane rings interacted with His440, Trp84 and Phe330 of AChE (PDB: 1EVE). Moreover, the *N* atoms established hydrogen bonds with a

crucial water molecule 1159, and the alkyl moiety generated hydrophobic interactions with several residues such as Phe330. and Phe290. On the other hand, the xanthone core was located in PAS, and engaged π - π and hydrophobic interactions with several residues (e.g. Tyr334, Tyr70, Tyr121, and Trp279). In addition to the cyclisation of the chromone skeleton, the balanced AChE and MAO-B inhibitor 48 (IC_{50} = 9.2 and 0.0077\,\mu\text{M}, respectively) with anti-A eta_{40} aggregation (IC_{50} = 3.2 μ M), was obtained by using the incorporation approach to substitute the O atom of pyranone with the carbonyl group⁹⁷. Docking studies demonstrated that the naphthoquinone core interacted with Trp286 of AChE (PDB: 4EY7) via π - π interaction, while the carbonyl group bound to Tyr72 and Tyr124 by hydrogen bonds. On the other hand, the carbonyl groups established three hydrogen bonds with Tyr326, Tyr435, and GIn206 of MAO-B (PDB: 6FWC). In addition, an intriguing halogen bond was observed between the chlorine atom of 48 and Cys172 of MAO-B.

Summary of SARs

Chromone, as a significant scaffold, is widely found in numerous natural products and synthetic compounds, exhibiting a wide range of biological activities. SARs of chromone-based inhibitors are shown in Figure 12. By introducing AChE inhibitory pharmacophores onto the chromone scaffold bearing formamide group as linker at 3-position of chromone core, balanced dual inhibitors of AChE and MAO-B are created. Substitution with formamide moiety at 2-position of chromone core, and introduction of AChE inhibitory pharmacophores onto the 6-position of chromone can increase the AChE inhibitory activities, but decrease MAO-B inhibitory potencies. Introduction of donepezil-like pharmacophores at 2-position of chromone core using formamide moiety as linker leads to a decrease in both AChE and MAO-B inhibitory activities. Furthermore, if donepezil-like pharmacophores are introduced at



Figure 12. SAR of chromone-based dual inhibitors targeting AChE and MAO-B.

the 3-position of the chromone core, both AChE and MAO-B inhibitory activities increase. Additionally, the chromone nucleus is easily modifiable, allowing for the acquisition of novel scaffolds (xanthone and naphthoquinone) by cyclisation or core-hopping strategies. The introduction of flexible alkoxy amine at the 2-position of xanthone results in AChE and MAO-B inhibitory activities increased. Similarly, the benzylamino group at the 2-position of naphthoquinone significantly enhances the MAO-B inhibitory activity.

Docking studies indicated that residues Trp84, and Phe330 in CAS, Trp279 in PAS were important factors in determining the AChE inhibitory activities of chromone derivavtives by forming π -cation and π - π interactions. While, residues Tyr435, Tyr398, and Tyr326 of MAO-B played an important role in the MAO-B inhibitory activities by establishing π - π interactions or hydrogen bound.

Benzo five-membered ring-based inhibitors

Heteroaromatic compounds, including benzofuran, benzothiophene, indole, and others, are essential organic compounds found in nature. These compounds serve as common structural units in many bioactive molecules, and have wide applications in medicine, dyes, agriculture, new materials and other fields⁹⁸. As a result, studies of these compounds have gained increasing attention, particularly in the field of pharmaceutical chemistry (Figure 13).

Compound 49, an AChE and MAO-B dual inhibitor, was synthesised based on benzo[d]isothiazol-3(2H)-one⁹⁹ using the linkage approach. It showed potent AChE inhibitory activity ($IC_{50} =$ $0.29 \,\mu$ M), but weak MAO-B inhibitory potency with an IC₅₀ value of 20.1 µM. This potent AChE inhibition could be attributed to the compound's ability to fully occupy the entire active pocket of AChE (PDB: 1EVE), while the benzo[d]isothiazol-3-one scaffold and the benzylamine moiety generated π - π interactions with Phe331, Phe288 and Tyr130, respectively. In addition, the S atom established a hydrogen bond with Tyr121 in the PAS. Compared with 49, using the fusion approach, the inhibitor 50 based on isobenzofuran-1(3H)-one exhibited more potent and balanced inhibitory activity against AChE and MAO-B (IC_{50} = 0.041 and 0.30 $\mu\text{M},$ respectively)¹⁰⁰. It was also characterised for Cu²⁺ chelation and favourable drug-like properties. Molecular docking studies indicated that the phthalide group interacted with Tyr334 through π - π interaction, while the butyl moiety established alkyl interactions with Trp279 and Phe290 of AChE (PDB: 1EVE), respectively. Furthermore, **50** exhibited various π - π interactions and π -cation bonds with Phe330, Trp84, and Glu199 of AChE. The benzyl ether group was located in the MAO-B (PDB: 2V60) entrance cavity and formed several interactions (hydrogen bonds, π -sigma and π - π interactions) with key amino acids such as Ile199, Cys172, and Tyr326.



While most potent MAO-B inhibitors are characterised by small molecular size, an indole-based molecule of large size synthesised using the linkage approach, compound 51, displayed potent and balanced AChE and MAO-B inhibitory activities with IC₅₀ values of 0.042 and 0.33 μ M, respectively¹⁰¹. These effects were attributed to its interactions with Trp86, Tyr124, Trp286, Tyr341, and Phe338 of AChE (PDB: 4EY7) through $\pi - \pi$ interactions, as well as interactions with Gly120 and Tyr133 through hydrogen bonds. Additionally, two balanced dual inhibitors 52 and 53 ($IC_{50} = 3.70$ and 2.41 µM for AChE, respectively; 2.62 and 1.84 µM for MAO-B, respectively) were synthesised using the fusion approach to fuse the carbamate moiety into the indole core¹⁰². As expected, the carbamate moieties established π -H interactions with CAS Trp84 of AChE (PDB: 1EVE), which is an important interaction for AChE inhibition. The indole cores interacted with Trp334 and Phe330 through $\pi - \pi$ interactions. On the other hand, **52** and **53** entered the substrate cavity of MAO-B (PDB: 2V5Z) and generated the π -H interactions with the FAD. These interactions played a crucial role in supporting the moderate MAO-B inhibitory activities observed for these two compounds.

To improve the brain permeability, the fusion approach was used to synthesise compound ${\bf 55}^{103}$ by replacing the imidazole ring of the lead 54 (IC_{50} = 0.324 and 1.427 μM , respectively) with the benzimidazole ring. Although the MAO-B inhibitory potency slightly decreased (IC₅₀ = $2.117 \,\mu$ M), the AChE inhibitory activity significantly increased with an IC_{50} value of 0.032 $\mu\text{M},$ and the BBB permeability improved. Furthermore, 55 demonstrated the ability to reverse cognitive impairment induced by scopolamine in mice, and showed good oral bioavailability (F = 45.55%), suggesting its potential for AD therapy. Using the fusion approach, benzimidazole-based molecule 56 was obtained, exhibiting more potent and balanced dual inhibition against the two enzymes ($IC_{50} = 0.024$ and 0.041 μ M, respectively)¹⁰⁴. A β_{42} ligand investigation revealed that dual inhibitor **56** showed higher anti-A β_{42} aggregation than curcumin and similar activity to that of donepezil. Docking studies showed that the benzimidazole and the phenyl moieties generated $\pi - \pi$ interactions with AChE (PDB: 4EY7) Trp286 and Tyr341, respectively. Besides, the guaternary amine established a hydrogen bond with Glu202. Regarding MAO-B (PDB: 2V5Z), the benzimidazole core generated π - π interaction with Trp119, and the carbonyl group formed a hydrogen bond with Tyr345.

Two indazole scaffold-based candidates **57** and **58** were created using the linkage approach, but they exhibited unbalanced dual inhibitory activities. Although they showed no AChE inhibition (21.6%, and 17.9% at 100 μ M), exhibited notably MAO-B inhibitory activity (IC₅₀ = 0.586, 0.386 nM respectively)¹⁰⁵. This result suggested that the presence of AChE inhibitory moieties was indispensable for the AChE and MAO-B dual inhibitors. The docking study of potent and balanced dual inhibitors **59** and **60** (IC₅₀ = 23.4, and 27.8 nM for AChE, respectively; 40.3 and 56.7 nM



Figure 13. Structures of benzo five-membered ring-based inhibitors.

for MAO-B, respectively)¹⁰⁶ obtained by using the fusion approach, further supported this observation. Both compounds were derived from a benzothiazole scaffold. In the AChE (PDB: 4EY7) docking study, the benzothiazole moiety of dual inhibitor 59 established π - π interaction with Trp286, and its carbonyl moiety formed a hydrogen bond with Phe295 in PAS. The N atoms of piperazine bound to Trp86 and Try337 by cation- π interactions, and a salt bridge was established with Asp74. Additionally, the N atom of dimethylaminoethyl moiety generated a hydrogen bond with Ser125. In the MAO-B (PDB: 2V5Z) docking study, the benzothiazole ring formed an aromatic-hydrogen bond with Pro102. The terminal N atom of the dimethylaminoethyl side chain and the N atom in the 1st position of the piperazine established cation- π interactions with Try435 and Tyr398, respectively. In addition, the piperazine ring formed a hydrogen bond with the carbonyl adjacent to the N5 of FAD.

Imine and hydrazone-based inhibitors

The C=N group has been widely reported as effective regulator of the interaction between inhibitors and enzyme, thus making imine and hydrazone derivatives potent MAO-B inhibitors in the past¹⁰⁷. Moreover, the imine nitrogen of the hydrazine could form a hydrogen bond with Tyr407¹⁰⁸. Therefore, the C=N group has

been widely investigated as an effective linker in seeking for potent AChE and MAO-B dual inhibitors (Figure 14).

Several imine-based anti-AD compounds have been synthesised using the linkage approach, and evaluated by researchers^{105,109,110}. However, these compounds such as compounds **61** and **62**, showed unbalanced dual-target inhibitory potency, with no AChE inhibition, but potent MAO-B inhibitory abilities (IC₅₀ values of 0.612 and 0.049 μ M, respectively). Compared to **61** and **62**, compound **63** displayed potent AChE inhibitory activity (IC₅₀ = 0.057 μ M), but remained an unbalanced MAO-B inhibitor (52.28% at 1 mM). Interestingly, a series of potent and balanced dual inhibitors (**64–67**)^{111–113} were obtained using the linkage approach when the hydrazone was designed as a linker to the substitute imine group.

The piperonylic acid derived hydrazones **64** ($IC_{50} = 0.052 \,\mu$ M for AChE, and 0.89 μ M for MAO-B) and **65** ($IC_{50} = 0.85 \,\mu$ M for AChE and 0.034 μ M for MAO-B) displayed balanced inhibitory activities against target enzymes. But **65** demonstrated more potent MAO-B inhibitory potency, suggesting that the propargyl substitution could improve the MAO-B inhibitory potency. In the AChE (PDB: 4EY7) binding mode, **64** formed two hydrogen bonds with Asp74 and Phe295, and **65** formed four hydrogen bonds with Phe295, Gly122, Gly121, and Ala204. Additionally, hydrophobic interactions established between both compounds and Val294, Phe338, Tyr337, and Trp28 were observed. Regarding MAO-B



Figure 14. Structures of imine and hydrazone-based inhibitors.

(PDB: 2V5Z), the isatin moieties of both compounds were oriented towards FAD, and a hydrogen bond was established between **64** and Tyr435, while four hydrogen bonds were observed between **65** and Ile199, Tyr326, Gln206, and Ile198.

The aromatic aldehyde hydrazone derivative **66**¹¹² was created by using the linkage approach. It exhibited weak AChE inhibition $(IC_{50} = 16.5 \,\mu\text{M})$, but showed excellent MAO-B inhibitory potency with an IC_{50} value of $0.025 \,\mu$ M, possibly due to the absence of AChE inhibitory pharmacophore. In the light of this, the potent and balanced guinoxaline scaffold-based dual inhibitor 67 was developed by using the linkage approach to fuse the AChE CAS inhibitory pharmacophore¹⁰³. This lead compound showed potent and balanced AChE and MAO-B inhibitory activity ($IC_{50} = 0.028$ and 0.046 µM, respectively). In the AChE (PDB: 4EY7) docking study, $\pi - \pi$ interactions were established between the guinoxaline and Trp286, as well as the phenyl ring and Phe338, and Tyr341. Two crucial hydrogen bonds were established between the hydrazone moiety with Arg296 and Phe295, respectively. In addition, the N atom of the dimethylaminoalkyl fragment formed a salt bridge with Glu202. In the docking study of MAO-B (PDB: 2V5Z), the guinoxaline core and phenyl ring bound to Tyr435 and Tyr326 through $\pi - \pi$ interactions, respectively. The hydrazine and dimethylaminoalkyl moieties formed hydrogen bonds with Glu206 and Pro102, respectively.

Other scaffold-based inhibitors

In addition to the AChE and MAO-B candidates mentioned above, extensive research has been conducted on other scaffold-based dual inhibitors (Figure 15). Compound **68**, 3,4-dihydropyrimidin-2(1*H*)-one-based dual inhibitor¹¹⁴, showed moderate AChE inhibitory potency and potent MAO-B inhibitory activity with IC₅₀ values of 2.86 and 0.34 μ M, respectively. Another potent and balanced dual inhibitor, pyridoxine scaffold-based compound **69** (IC₅₀ = 0.081 and 0.039 μ M, respectively) was synthesised using the linkage approach¹¹⁵. A docking study showed that it bound to Tyr435 and Tyr398 of AChE (PDB: 4EY7) through π - π interactions, and embedded in the entrance cavity of MAO-B (PDB: 2BYB) which was formed by Tyr69, Ile199, Leu167, Gln206, Ile316, and Leu171. In addition, notable π - π interactions were observed with Tyr435 and Tyr398 in the substrate cavity of MAO-B.

Youdim⁵⁰ designed and synthesised the inhibitor **70** using the fusion approach, which served as the prodrug for compound **71**. Compound **70** showed potent AChE inhibitory activity with an

 IC_{50} value of $0.52\,\mu\text{M},$ but exhibited limited MAO-B inhibitory potency (IC₅₀ = 7.90 μ M) and metal affinity (Fe, Cu, and Zn). Interestingly, upon metabolism by AChE, it released 71 to restore the abilities of metal ions chelation and MAO-B inhibitory activity $(IC_{50} = 0.057 \,\mu\text{M})$, suggesting that **70** hold promise as a multi-target lead for AD therapy. Compound 72, synthesised using the linkage approach, was a dual inhibitor based on 1(2H)-phthalazinone scaffold¹¹⁶. Although various interactions such as hydrogen bond and $\pi - \pi$ interactions have been established (AChE, PDB: 1EVE), it displayed only moderate AChE inhibition (IC₅₀ = $8.2 \,\mu$ M). However, it exhibited potent MAO-B inhibitory activity with an IC_{50} value of 0.7 μ M. In the MAO-B (PDB: 2V60) docking study, the phthalazinone core of 72 formed hydrogen bonds with crucial residues Cys172 and Tyr407. Similarly, compounds 73, 74, and 75 were obtained using the linkage or fusion approaches. They displayed moderate AChE inhibition (IC₅₀ = 2.9, 0.442, and 7.31 μ M, respectively), but weak MAO-B inhibition ($IC_{50} = 16.6$, 6.43, and 26.1 μ M, respectively)^{117–119}, suggesting that a rigid scaffold is necessary for potent MAO-B inhibitors.

Fluoxetine and sertraline, used as antidepressant drugs in clinical practice, have recently been explored for the design of anti-AD drugs. Nadeem et al.¹²⁰ developed a series of potent and balanced AChE and MAO-B dual inhibitors by using the linkage approach to fuse the structural features from fluoxetine and sertraline into the tacrine, donepezil, and certain antidiabetic drugs (pioglitazone, ciglitazone, rosiglitazone, and troglitazone). Among them, **76–80** exhibited the best dual inhibitory activities ($IC_{50} =$ 0.024, 0.010, 0.035, 0.015, and 0.008 μ M for AChE, respectively; IC₅₀ = 0.011, 0.015, 0.180, 0.157, and 0.031 µM for MAO-B, respectively), and showed good safety profiles (2000 mg·kg⁻¹). Molecular docking studies revealed that these compounds could interact with residues of AChE (PDB: 2CKM) including Trp84, Phe330, Trp279, and Tyr70 through π - π interactions. Besides, hydrogen bonds were observed with Tyr121 and Tyr334. In the case of MAO-B (PDB: 2V5Z), these compounds established hydrogen bonds with Tyr188, Tyr326, and Trp388, and π - π interactions with Tvr398 and Tvr435.

The diclofenac derivative **81** displayed moderate AChE inhibition (IC₅₀ = 2.58 μ M) but weak MAO-B inhibition (IC₅₀ = 13.07 μ M). In order to discover potent and balanced AChE and MAO-B dual inhibitors, Javed's group¹²¹ synthesised several diclofenac derivatives based on the lead **81** by using the linkage or fusion approaches. Among them, **82**, **83**, and **84** demonstrated potent and balanced AChE and MAO-B inhibitory activities. In





69





72

MAO-B IC₅₀= $0.7 \pm 0.1 \ \mu M$





75

MAO-B IC_{50}= $7.90 \pm 1.34 \ \mu M$

AChE IC₅₀= $0.52 \pm 0.07 \ \mu M$ MAO-B IC₅₀= $0.057 \pm 0.01 \ \mu M$ AChE IC₅₀= $8.2 \pm 0.2 \ \mu M$

73 AChE IC₅₀ = $2.9 \pm 0.03 \mu$ M MAO-B IC₅₀ = $16.6 \pm 0.43 \mu$ M

74 AChE IC₅₀ = $0.442 \pm 0.03 \mu M$ MAO-B IC₅₀ = $6.43 \pm 0.62 \mu M$

AChE IC₅₀= 7.31 ± 1.4 μ M MAO-B IC₅₀= 26.1 ± 3.9 μ M





Figure 15. Structures of other scaffold-based inhibitors.

particular, **84** exhibited the ability to cross the BBB and demonstrated non-neurotoxicity at the dose of 2000 mg kg⁻¹ *in vivo*. Docking studies demonstrated the formation of π - π , hydrogen bond, and π -S interactions between **84** and Trp84, Phe330, Phe331, Tyr121, Trp279, and Tyr334 of AChE (PDB: 1EVE). On the other hand, six hydrogen bonds were observed between **84** and Arg42, Cys197, Ile199, Gln206, Tyr326, and Trp388 of MAO-B (PDB: 2V5Z). These interactions provided an explanation for its functional profiles.

Conclusion and prospect

In spite of AChE has been the most successful target in development of anti-AD drugs, AChEIs approved by FDA have been only employed for symptomatic treatment of AD. MAO-B has emerged as another promising target against AD due to the discovery of its overexpression in AD patients' brain and the establishment of an oxidative stress state. Therefore, we have conducted a comprehensive review of the dual inhibitors targeting both AChE and MAO-B published in recent years.

The methods for implementing MTDLs strategy and the primary characteristics of AChE and MAO-B active pockets were summarised in this review. Based on this, dual inhibitors targeting both AChE and MAO-B, including chalcone, couumarin, chromone, benzo five-membered ring, imine, and hydrazone scaffolds, were systematically classified according to their structural characteristics. Additionally, an in-depth analysis of their design strategies, docking studies, and SARs was conducted. Phe295, Phe288, and Tyr121 of AChE could form hydrogen bonds with ligands, and Trp279, Trp286, Phe330, Phe334, Phe338, and Tyr341 generated π - π interactions. Moreover, Tyr72 and Tyr334 established hydrogen bonds and π - π interactions. Similarly, Cys172, Ile198, Ile199, Gln206, Tyr188, Arg42, and Pro102 of MAO-B could form hydrogen bonds with ligands, Tyr326, Tyr398, and Tyr435 established hydrogen bonds and π - π interactions. These above-mentioned residues played a crucial role in determining the AChE and MAO-B inhibitory activities of ligands, providing the useful clues for designing reasonable dual inhibitors of AChE and MAO-B by predicting whether ligands could generate interactions with these crucial residues.

However, great challenges still exist to develop balanced and potent dual inhibitors targeting both AChE and MAO-B due to the inherent structural limitations of both enzymes themselves. The CAS site for AChE is located at the bottom of a deep and narrow gorge (about 20 Å long and as narrow as 4.5 Å). While the active sites of MAO-B is composed of the "entrance cavity" (about 290 Å³) and a larger hydrophobic substrate cavity (420 Å³). This meant that both enzymes often lack the ability to share the same pharmacophores, making the discovery of potent and balanced dual inhibitors of AChE and MAO-B an extremely difficult process. With the advancement of artificial intelligence, CADD strategies have played an important role in new drug discovery, offering an effective approach to solve this challenge. The high-throughput screening, energy calculation, affinity prediction based on this technique can significantly improve the efficiency in designing dual inhibitors of AChE and MAO-B, avoiding the arduous process

of blind phenotype-based screening, and creating the well-balanced and potent candidates that cross the BBB with high bioavailability and low toxic side effects.

Encouragingly, lecanemab, the second $A\beta$ -targeting drug after aducanumab, has been fully approved by the FDA for the treatment of AD this year. This approval not only provides solid evidence for the A β hypothesis of AD, but also represents a significant milestone in AD therapy, bringing new hope for AD patients. In addition, due to the complex nature of AD, various therapies are undergoing clinical trials, such as anti-tau, synaptic plasticity/neuroprotection, and anti-inflammation therapies. Furthermore, recent studies have identified new opportunities and possible targets for AD therapy such as β 2-microglobulin, microRNA-132, and Bacillus Calmette-Guérin. With the continuous advancement of research, the complex nature of AD will gradually be understood, leading to the discovery of more effective drugs. The treatment of AD will present diverse prospects and the combination of multiple approaches may bring more clinical benefits to AD patients.

Authors contributions

Writing-original draft preparation, Dajiang Zou; writing-review and editing, Dajiang Zou, Renzheng Liu, Yangjing Lv, Jianan Guo, Changjun Zhang; supervision, Yuanyuan Xie. All authors have read and agreed to the published version of the manuscript.

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