

Characterization of Dicarboxylate Stimulation of Ammonia, Glutamine, and 2-Oxoglutarate-Dependent O₂ Evolution in Isolated Pea Chloroplasts

Received for publication September 28, 1982 and in revised form January 31, 1983

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ABSTRACT

Intact isolated chloroplasts from pea (*Pisum sativum*) leaves carried out light-dependent (NH₃, 2-oxoglutarate) and (glutamine, 2-oxoglutarate)-dependent O₂ evolution at rates of 3.3 ± 0.7 ($n = 7$) and 6.0 ± 0.4 ($n = 5$) micromoles per milligram chlorophyll per hour, respectively. Malate stimulated the rate of (NH₃, 2-oxoglutarate)-dependent O₂ evolution 2.1 ± 0.5 ($n = 7$)-fold in the absence of glutamine, and 3.3 ± 0.4 ($n = 11$)-fold in the presence of glutamine. Malate also stimulated (glutamine, 2-oxoglutarate)-dependent O₂ evolution in the presence of high concentrations of glutamine. The affinity ($K_{1/2}$) of (NH₃, glutamine, 2-oxoglutarate)-dependent O₂ evolution for 2-oxoglutarate was estimated at 200 to 250 micromolar in the absence of malate and 50 to 80 micromolar when malate (0.5 millimolar) was present. In contrast to malate and various other dicarboxylates, aspartate, glutarate, and glutamate did not stimulate (NH₃, glutamine, 2-oxoglutarate)-dependent O₂ evolution in isolated pea chloroplasts. Using both *in vitro* assays and reconstituted chloroplast systems, malate was shown to have no effect on the activities of either glutamine synthetase or glutamate synthase.

The concentration of malate required for maximal stimulation of O₂ evolution was dependent on the concentration of 2-oxoglutarate present. However, the small extent of the competition between malate and 2-oxoglutarate for uptake was not consistent with that predicted by the current 'single carrier' model proposed for the uptake of dicarboxylates into chloroplasts.

It is now widely accepted that refixation of photorespiratory NH₃ takes place via the GS/GOGAT² pathway (12). However, while both enzymes are located within the chloroplast, the actual site of refixation of NH₃ by GS is still the subject of some conjecture, because up to 50% of the total cellular GS activity is cytoplasmic in some plant species (22). For refixation by cytoplasmic GS to be a feasible proposition, potential uptake of glutamine into the chloroplast must be sufficiently rapid to maintain the proposed rate of photorespiratory NH₃ recycling estimated at approximately $30 \mu\text{mol mg}^{-1} \text{Chl h}^{-1}$ (24). Evidence to date, however, on glutamine uptake (3) shows this to be unlikely, thereby providing support for the hypothesis that all NH₃ refixa-

tion takes place within the chloroplast.

Similarly, 2-OG must also be transported into the chloroplast at rates approaching those cited above. On the basis of ¹⁴C uptake experiments with isolated chloroplasts from both spinach (8) and peas (16), it seems that potential rates of 2-OG uptake are sufficient to account for the predicted rate of operation of the photorespiratory cycle. However, despite these findings, attempts to demonstrate similar rates of (NH₃, 2-OG)-dependent O₂ evolution by isolated chloroplasts have not been successful (1). In a recent publication, Woo and Osmond (25) re-examined this problem and showed that the rate of (NH₃, 2-OG)-dependent O₂ evolution by isolated chloroplasts could be significantly enhanced by the addition of certain dicarboxylates, e.g. malate, succinate, or fumarate to the assay system.

Given our present understanding of the uptake of dicarboxylate molecules into chloroplasts, *i.e.* it is mediated by a broadly specific dicarboxylate carrier (8), one might predict that malate should in fact inhibit (NH₃, 2-OG)-dependent O₂ evolution, because both malate and 2-OG would compete for entry into the chloroplast on the same carrier. We therefore set out to examine in more detail the phenomenon of dicarboxylate stimulation of (NH₃, 2-OG)-dependent O₂ evolution in isolated pea chloroplasts and have found evidence that malate and 2-OG do indeed compete for uptake into the chloroplast. However, the competition is rather weak and our results are best explained by the operation of two separate carriers in the chloroplast membrane and not that of a single broadly specific dicarboxylate carrier.

MATERIALS AND METHODS

Pea seedlings (*Pisum sativum* vars. Massey Gem, Greenfeast) were grown in vermiculite for 11 to 13 d in a glasshouse. All biochemicals were obtained from Sigma Chemical Co. Spinach ferredoxin was prepared by the method of Rao *et al.* (17).

Isolation of Chloroplasts. Chloroplasts were obtained from pea shoots according to the method of Robinson and Wiskich (18).

Assay Procedures. O₂ evolution was measured in a Rank O₂ electrode. The vessel was maintained at 25°C and illuminated with a 150-w tungsten halogen projector lamp giving a light intensity of $550 \mu\text{E m}^{-2} \text{s}^{-1}$ at the center of the vessel. Chloroplasts (50 μg Chl) were illuminated in a 2-ml standard reaction medium containing 330 mM sorbitol, 2 mM EDTA, 1 mM MgCl₂, 1 mM MnCl₂, and 50 mM Hepes, pH 7.6. For CO₂-dependent O₂ evolution, 4.5 mM Na₂P₂O₇, 0.9 mM ATP, 0.18 mM NaH₂PO₄/Na₂HPO₄ were also added. Glyceraldehyde (10 mM) was present during (NH₃, 2-OG) (NH₃, glutamine, 2-OG)- and (glutamine, 2-OG)-dependent O₂ evolution to inhibit endogenous CO₂-dependent O₂ evolution (20). Chloroplast intactness was measured with ferricyanide according to the method of Lilley *et al.* (10). Chl was determined from 80% acetone extracts using the method of Arnon

¹ Recipient of a George Fraser Scholarship, University of Adelaide. Financial support from the Australian Research Grants Committee is gratefully acknowledged.

² Abbreviations: GS, glutamine synthetase; 2-OG, 2-oxoglutarate; Asp, aspartate; OH-Pyr, hydroxypyruvate; Aza, azaserine; MSO, methionine sulfoximine; GOGAT, glutamate synthase.

(2). All dicarboxylate compounds were newly prepared and adjusted to pH 7.0 with NaOH prior to use.

GS was assayed spectrophotometrically according to the method of O'Neal and Joy (14). GOGAT was assayed polarographically (4) using a reconstituted chloroplast system prepared according to the method of Lilley *et al.* (9). The reaction mixture contained envelope-free chloroplasts (250 μg Chl), 100 μl chloroplast extract, 330 mM sorbitol, 2 mM EDTA, 10 mM MgCl₂, 1 mM MnCl₂, 50 mM Hepes, 200 units of catalase, 1 mM NH₄Cl, and 180 μg spinach ferredoxin in a final volume of 2 ml at pH 7.6 and 25°C.

RESULTS

Isolated pea chloroplasts carried out light-dependent O₂ evolution in the presence of NH₄Cl and 2-OG (trace A, Fig. 1) and in the absence of PPi, ADP, and large concentrations of MgCl₂, used previously by other researchers for (NH₃, 2-OG)-dependent O₂ evolution (1, 25). Glutamine addition caused only a small increase in the rate of O₂ evolution (trace A, Fig. 1), whereas in the

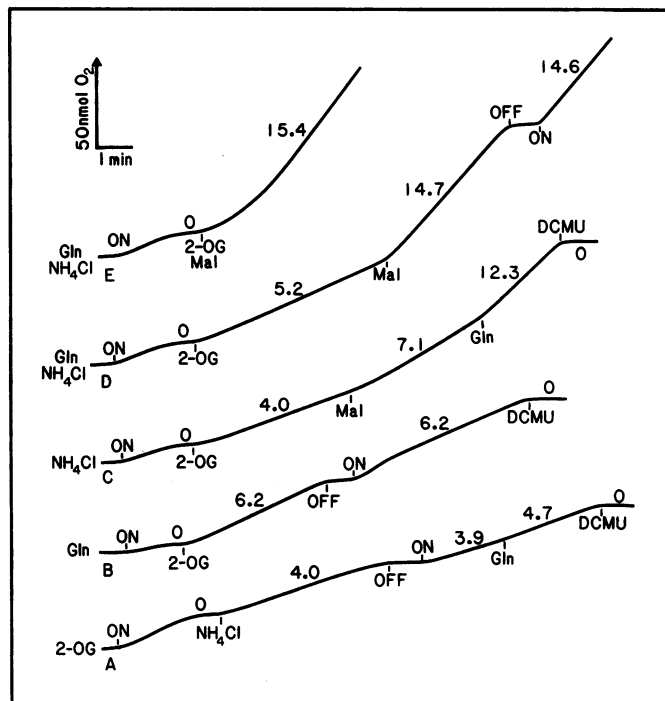


FIG. 1. Effect of glutamine, DCMU and malate, on (NH₃, 2-OG)- and (glutamine, 2-OG)-dependent O₂ evolution by isolated pea chloroplasts. Additions were made to the following concentrations: NH₄Cl, 1 mM; 2-OG, 1 mM; glutamine, 1 mM; malate (Mal), 1 mM; DCMU, 5 μM , Chl concentration: 47 μg ml⁻¹. CO₂-dependent O₂ evolution: 92 μmol mg⁻¹ Chl h⁻¹. Intactness: 69%. Values beside the curves represent μmol O₂ evolved mg⁻¹ Chl h⁻¹.

Table 1. Effect of Malate on *In Vitro* Glutamate Synthase Activity Measured with a Reconstituted Chloroplast System

The reaction mixture containing 5 mM glutamine was illuminated for 2 min prior to the first addition.

Sequential Additions	Glutamate Synthase Activity
	<i>nmol</i> O ₂ evolved min ⁻¹
2-Oxoglutarate (1 mM)	34.6
Malate (1 mM)	34.6
Malate (4 mM)	34.6
Chloroplast extract (150 μl)	59.3
Azaserine (0.5 mM)	0

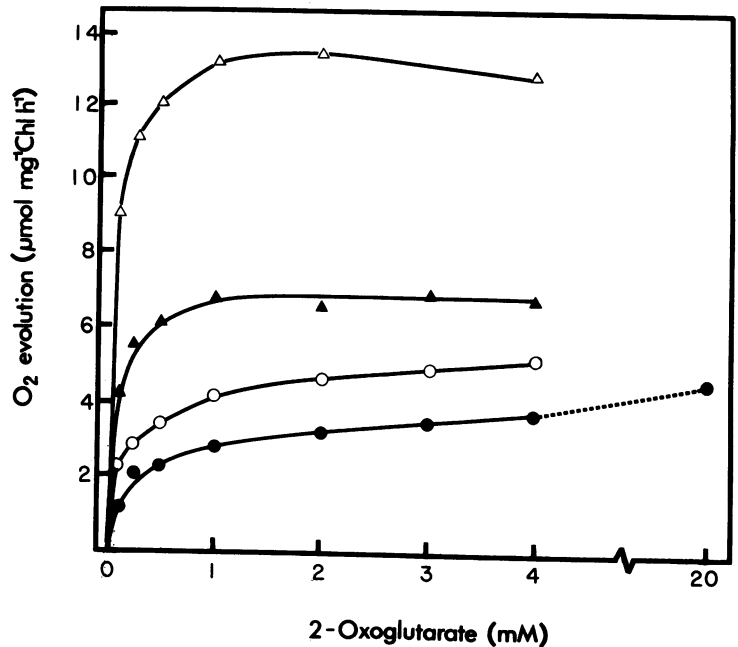


FIG. 2. Effect of 2-OG concentration on (NH₃, 2-OG)-dependent O₂ evolution in the presence and absence of glutamine and malate. Where added, NH₄Cl, glutamine, and malate were present at a concentration of 1 mM. O₂ evolution was initiated by the addition of 2-OG (\pm malate) and the recorded rates represent the steady state rates of O₂ evolution after 5 min. (NH₃, 2-OG), (●); (NH₃, glutamine, 2-OG), (○); (NH₃, 2-OG, malate), (▲); (NH₃, glutamine, 2-OG, malate), (Δ). Chl concentration: 45 μg ml⁻¹. CO₂-dependent O₂ evolution: 70 μmol mg⁻¹ Chl h⁻¹. Intactness: 75%.

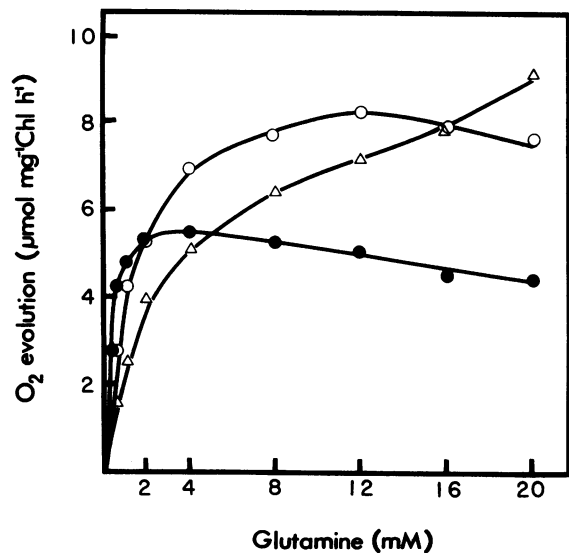


FIG. 3. Effect of malate on (glutamine, 2-OG)-dependent O₂ evolution by isolated pea chloroplasts in the presence of varying glutamine concentrations. 2-OG was present at 1 mM. Malate was added at a concentration of either 0.1 (○) or 1.0 mM (Δ) following the establishment of a steady state rate of O₂ evolution. Chl concentration: 50 μg ml⁻¹. CO₂-dependent O₂ evolution: 64 μmol mg⁻¹ Chl h⁻¹. Intactness: 66%.

presence of malate it produced a marked stimulation (trace C, Fig. 1). The addition of malate stimulated O₂ evolution, the extent of which was found to be dependent on the presence or absence of added glutamine, *i.e.* while malate was found to stimulate (NH₃, 2-OG)-dependent O₂ evolution in the absence of glutamine, the full extent of malate stimulation was only observed when gluta-

Table II. Summary of Mean Values for Various Forms of O₂ Evolution by Isolated Chloroplasts over n Preparations

Rates corrected for chloroplast breakage during preparation (CFB) are also given (mean intactness: 68%; n = 13). Assay of A, B, D, and E were always carried out with a 1 mM concentration of all substrates involved. Assay of C involved glutamine, 2 mM, 2-oxoglutarate, 1 mM. Values represent $\mu\text{mol O}_2$ evolved mg^{-1} Chl h^{-1} .

Type of O ₂ Evolution	n	Mean	SD	CFB
A. NH ₃ , 2-OG	7	3.3	0.7	4.8
B. NH ₃ , glutamine, 2-OG	12	4.4	0.9	6.4
C. Glutamine, 2-OG	5	6.0	0.4	8.8
D. NH ₃ , 2-OG, Malate	7	6.9	1.6	10.0
E. NH ₃ , Glutamine, 2-OG, Malate	11	14.6	2.2	21.5
D/A	7	2.1	0.5	
D/E	7	0.5	0.1	
E/B	11	3.3	0.4	

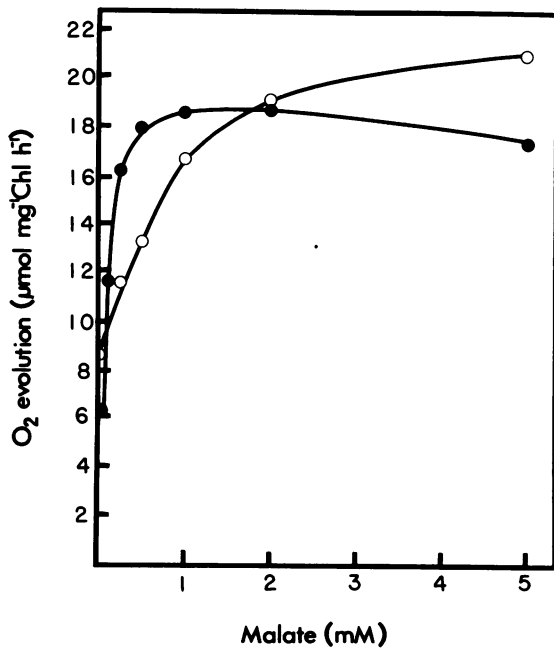


FIG. 4. Effect of malate concentration on the stimulation of (NH₃, glutamine, 2-OG)-dependent O₂ evolution by isolated pea chloroplasts in the presence of two different 2-OG concentrations. Standard assay contained NH₄Cl, 1 mM; glutamine, 1 mM, and 2-OG at a concentration of either 1 (●) or 10 mM (○). Malate, at varying concentrations, was added at the same time as 2-OG which was used to initiate O₂ evolution. Chl concentration: 40 $\mu\text{g ml}^{-1}$. CO₂-dependent O₂ evolution: 85 $\mu\text{mol mg}^{-1}$ Chl h^{-1} . Intactness: 70%.

mine was also present (traces C, D, and E, Fig. 1). Isolated pea chloroplasts also supported light-dependent O₂ evolution in the presence of glutamine and 2-OG, and, like (NH₃, 2-OG)-dependent O₂ evolution, this activity was inhibited by DCMU (traces A and B, Fig. 1). MSO or Aza inhibited (NH₃, 2-OG)-dependent O₂ evolution, both in the presence and absence of malate, at concentrations of 2.5 and 0.25 mM, respectively (data not shown), indicating that O₂ evolution resulted from the operation of the GS/GOGAT pathway (19, 21) and not from the metabolism of malate within the chloroplast.

The possibility that malate stimulated O₂ evolution by activating either GS or GOGAT was investigated using broken chloroplasts (GS) or a reconstituted assay system (GOGAT). Malate had no

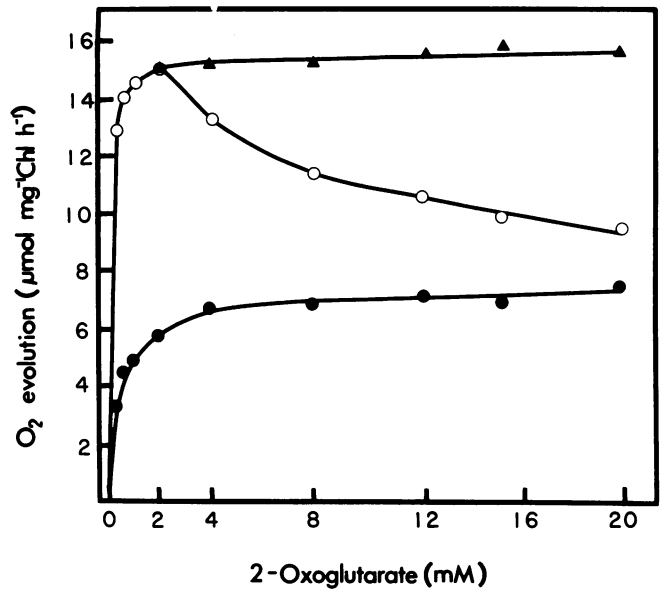


FIG. 5. Effect of 2-OG concentration on the stimulatory effect of malate on (NH₃, glutamine, 2-OG)-dependent O₂ evolution by isolated pea chloroplasts. Standard assay contained NH₄Cl, 1 mM, glutamine, 1 mM (●), and malate, 0.5 mM (○), or at a concentration equaling the 2-OG concentration used in the assay (▲). Where present, malate addition was made with 2-OG, which was used to initiate O₂ evolution. Chl concentration: 38 $\mu\text{g ml}^{-1}$. CO₂-dependent O₂ evolution: 98 $\mu\text{mol mg}^{-1}$ Chl h^{-1} . Intactness: 72%.

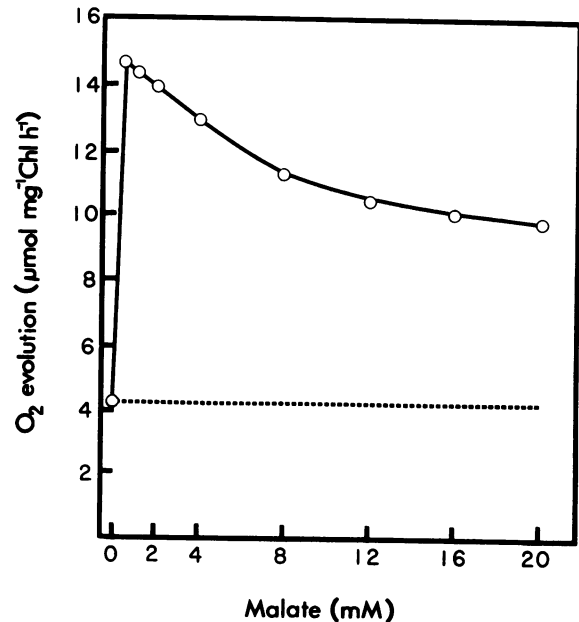


FIG. 6. Effect of increasing malate concentration on its stimulatory effect on (NH₃, glutamine, 2-OG)-dependent O₂ evolution by isolated pea chloroplasts. Standard assay contained NH₄Cl, 1 mM, glutamine, 1 mM, 2-OG, 0.5 mM. Malate addition was made with 2-OG, which was used to initiate O₂ evolution. The dotted line represents the rate of O₂ evolution in the absence of added malate. Chl concentration: 43 $\mu\text{g ml}^{-1}$. CO₂-dependent O₂ evolution: 100 $\mu\text{mol mg}^{-1}$ Chl h^{-1} . Intactness: 71%.

effect on the *in vitro* activities of either GS (data not shown) or GOGAT (Table I) over the wide range of substrate concentration tested, indicating that the stimulation of (NH₃, glutamine, 2-OG)-dependent O₂ evolution by malate required the presence of an intact chloroplast membrane.

Table III. *Specificity of Stimulation by Various Decarboxylates of (NH₃, Glutamine, 2-OG)-Dependent O₂ Evolution by Isolated Chloroplasts*

Standard assay contained NH₄Cl, 1 mM; glutamine, 1 mM; 2-OG, 1 mM; and dicarboxylate, 2 or 5 mM. Dicarboxylate addition was made with 2-OG which was used to initiate O₂ evolution. Aminoxyacetic acid (5 mM) was used to inhibit (Asp, 2-OG)-dependent O₂ evolution in assays involving Asp. There was no effect of aminoxyacetic acid on the rate of (NH₃, glutamine, 2-OG)-dependent O₂ evolution. The control rate represents the rate of O₂ evolution in the absence of added dicarboxylate.

Dicarboxylate	Experiment 1		Experiment 2	
	O ₂ Evolution	Stimulation by Malate	O ₂ Evolution	Stimulation by Malate
	$\mu\text{mol mg}^{-1}$ Chl h^{-1}	%	$\mu\text{mol mg}^{-1}$ Chl^{-1}	%
Control	4.6		5.2	
Malate (2 mM)	16.6	100	15.0	100
Succinate (2 mM)	16.3	98		
Fumarate (2 mM)	16.6	100		
L-Tartrate (2 mM)	12.8	68	11.1	60
L-Tartrate (5 mM)	13.4	73	12.3	72
Aspartate				
2 mM	5.4	6	6.0	8
5 mM	6.1	12	6.5	13
Glutarate				
2 mM	6.4	15		
5 mM	7.1	21	6.5	13
Glutamate				
2 mM	2.0			
5 mM	1.0			
Thiomalate				
2 mM			8.9	38
5 mM			11.4	63
Malonate				
2 mM	4.6	0		
5 mM	4.1			

The dependence of (NH₃, 2-OG)-dependent O₂ evolution by isolated pea chloroplasts on 2-OG concentration, both in the presence and absence of glutamine and malate, is shown in Figure 2. Again it can be seen that malate stimulates (NH₃, 2-OG)-dependent O₂ evolution both with and without added glutamine, over the range of 2-OG concentrations tested. The $K_{1/2}$ for 2-OG in the absence of malate, was estimated at 200 to 250 μM . Glutamine addition had no effect on this value, while the presence of malate reduced the $K_{1/2}$ estimate to 50 to 80 μM . Malate was also found to stimulate (glutamine, 2-OG)-dependent O₂ evolution (*i.e.* no added NH₄Cl) when isolated pea chloroplasts were incubated with high concentrations of glutamine (Fig. 3).

Table II gives the mean values of the various types of O₂ evolution shown in Figure 2 as well as (glutamine, 2-OG)-dependent O₂ evolution, measured with a number of separate chloroplast preparations. It also includes the estimated rates of O₂ evolution when corrected for chloroplast breakage during isolation. It should be noted that the most variable parameter listed in Table II was found to be the ratio D/E, *i.e.* the rate of malate-stimulated (NH₃, 2-OG)-dependent O₂ evolution in the absence of glutamine, compared to the maximal rate of malate-stimulated O₂ evolution in the presence of glutamine. This was not only true for chloroplast preparations within the variety Massey Gem, but also between pea varieties, with the variety Greenfeast consistently having a much smaller value for this ratio (*e.g.* 0.20 to 0.25) than Massey Gem. This most probably reflects differences in the levels of endogenous intermediates of the GS/GOGAT pathway, within chloroplasts isolated from these different pea varieties. The optimum pH for (NH₃, glutamine, 2-OG)-dependent O₂ evolution, both with and without added malate, was 7.4.

The relationship between malate concentration and the degree

of stimulation of (NH₃, glutamine, 2-OG)-dependent O₂ evolution is illustrated in Figure 4. In the presence of 1 mM 2-OG, half-maximal stimulation of O₂ evolution was achieved on addition of malate to a concentration of 150 μM and maximal stimulation was reached when the value of the ratio of concentrations of malate to 2-OG was in the range of 1 to 2. However, when malate was added in the presence of an increased concentration of 2-OG (10 mM), stimulation at each malate concentration was found to be reduced and the $K_{1/2}$ for malate increased to approximately 700 μM .

These results indicated that while malate could significantly stimulate (NH₃, glutamine, 2-OG)-dependent O₂ evolution, the degree of stimulation produced by a particular malate concentration was dependent on the concentration of 2-OG also present. This is further illustrated in Figure 5. Here, the malate concentration was kept constant (0.5 mM) while the 2-OG concentration was varied between 0 and 20 mM. The resultant stimulation of (NH₃, glutamine, 2-OG)-dependent O₂ evolution, was compared to that produced by malate, when present at the same concentration as 2-OG, *i.e.* a malate/2-OG concentration ratio of 1. Again, the results show that the degree of stimulation of (NH₃, glutamine, 2-OG)-dependent O₂ evolution, produced by a particular malate concentration, was dependent on the concentration of 2-OG present. Increasing the amount of malate added in the presence of a constant 2-OG concentration was also found to lead to a decrease in the rate of malate-stimulated (NH₃, glutamine, 2-OG)-dependent O₂ evolution (Fig. 6).

The effects of a range of dicarboxylates on (NH₃, glutamine, 2-OG)-dependent O₂ evolution are shown in Table III. Malate, succinate, fumarate, L-tartrate and thiomalate were all found to stimulate O₂ evolution markedly. In contrast, Asp (in the presence

of aminoxyacetic acid to inhibit GOT) and glutamate produced only very minor increases, while glutamate addition resulted in an inhibition of (NH₃, glutamine, 2-oxoglutarate)-dependent O₂ evolution. Malonate which is not thought to be transported into the chloroplast (8) had no effect.

DISCUSSION

Our results demonstrate that isolated pea chloroplasts are capable of supporting both (NH₃, 2-OG)- and (glutamine, 2-OG)-dependent O₂ evolution in the absence of PP_i, ADP, or excessively high concentrations of MgCl₂ (10 mM) as used previously (1, 25). This is especially significant with respect to MgCl₂, inasmuch as recent reports (6, 11) have shown that the presence of MgCl₂ at concentrations as low as 4 mM can cause significant acidification of the chloroplast stroma and an inhibition of CO₂-dependent O₂ evolution.

In contrast to the observations of Woo and Osmond (25), malate was found to stimulate (NH₃, 2-OG)-dependent O₂ evolution in isolated pea chloroplasts in the absence of added glutamine. However, the full stimulatory effect of malate was only seen when glutamine was also present, with NH₄Cl. This requirement for glutamine most probably results from a deficiency of glutamate for GS, due to the unusually high *K_m* of this enzyme for this substrate, e.g. 3.5 to 13 mM (15), and the presence, in isolated pea chloroplasts, of endogenous glutamate at concentrations of only 5 to 6 mM (13). This is in contrast to spinach chloroplasts, which did not require glutamine for maximal stimulation of (NH₃, 2-OG)-dependent O₂ evolution by malate (25), and which have been found to contain glutamate at concentrations of up to 15 mM (8).

The demonstration of malate-stimulated (glutamine, 2-OG)-dependent O₂ evolution (Fig. 3) indicates that the effect of malate is on GOGAT-dependent O₂ evolution. It is also clear that the extent of the stimulation is dependent on the relative availability of glutamine to the GOGAT enzyme. When the only source of glutamine is via uptake (e.g. Fig. 3), malate may be seen to have little or no effect, due to the slow transport of this amino acid across chloroplast membranes (3). When isolated pea chloroplasts are incubated with both glutamine and NH₄Cl, however, the availability of glutamine within the chloroplast is markedly increased through the operation of GS, thereby allowing the maximal stimulatory effect of malate on GOGAT-dependent O₂ evolution to be observed (Figs. 1 and 2).

No evidence was found to suggest that the observed stimulation of (NH₃, glutamine, 2-OG)-dependent O₂ evolution by malate resulted from the metabolism of malate within the chloroplast, or that the addition of malate led to any activation of the pathway enzymes. Because malate can be shown to stimulate both (NH₃, 2-OG)- and (glutamine, 2-OG)-dependent O₂ evolution, it is unlikely that the malate effect is mediated through direct effects on glutamine or NH₃ uptake. More probably, malate increases the availability to GOGAT of its other substrate (2-OG) either by stimulating 2-OG uptake or increasing the affinity of the GOGAT enzyme for this substrate. Woo and Osmond (25) observed a dramatic decrease in the *K*_{1/2} for 2-OG in (NH₃, 2-OG)-dependent O₂ evolution from 6 mM to 73 μM on the addition of malate to spinach chloroplasts. Our measurements with pea chloroplasts, however, showed the *K*_{1/2} estimates for 2-OG, both in the presence and absence of malate (50 to 250 μM), to be in the range expected from previous studies on chloroplast transporters (8) and the GOGAT enzyme (21). Thus, the stimulatory effect of malate on (NH₃, glutamine, 2-OG)-dependent O₂ evolution probably results from an increase in the rate of 2-OG uptake, via exchange reactions with malate across the chloroplast membrane.

It is important to note that glutamate, a dicarboxylate known to exchange across the chloroplast membrane and produced during (NH₃, glutamine, 2-OG)-dependent O₂ evolution, could be expected to exchange for 2-OG *in vivo*, in order to maintain the

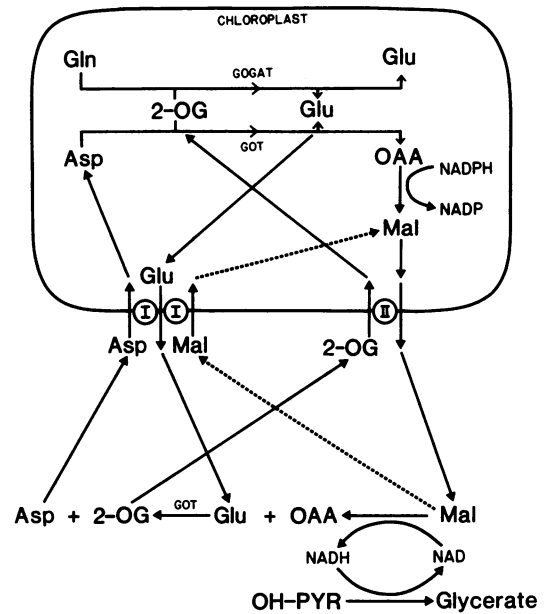


FIG. 7. Movement of dicarboxylate molecules during (a) the operation of a malate/Asp shuttle (solid lines) and (b) malate-stimulated uptake of 2-OG for GOGAT-catalyzed O₂ evolution (broken lines). See text for explanation of symbols I and II. GOT, glutamate oxaloacetate transaminase; Mal, malate.

stoichiometric balance of the photorespiratory cycle. However, it seems from the limited rate of (NH₃, glutamine, 2-OG)-dependent O₂ evolution in the absence of malate that glutamate (in contrast to malate) may not be the optimal substrate for exchange reactions involving 2-OG.

Isolated chloroplasts would not contain sufficient endogenous dicarboxylates (8), such as malate, to maintain this type of exchange for any length of time following the addition of 2-OG. This may explain the apparent anomaly between the estimated rates of 2-OG uptake by isolated chloroplasts as measured in ¹⁴C uptake experiments over 10 to 20 s, i.e. 26 to 32 μmol mg⁻¹ Chl h⁻¹ (8, 16), and those rates approximated from polarographic measurements, as done here, of between 8 and 12 μmol mg⁻¹ Chl h⁻¹ (Table I), which represent steady state rates calculated over 5–10 min.

Asp is rapidly taken up in isolated pea chloroplast as indicated by the high rates of (Asp, 2-OG)-dependent O₂ evolution which can be observed (20–25 μmol mg⁻¹ Chl h⁻¹). However, this amino acid has little effect on (NH₃, glutamine, 2-OG)-dependent O₂ evolution (Table III). This suggests that under the experimental conditions used internal Asp does not exchange for external 2-OG, as readily as do the C₄-type dicarboxylates such as malate, succinate, or L-tartrate.

Another major point regarding malate-stimulated (NH₃, glutamine, 2-OG)-dependent O₂ evolution is that it is not consistent with a single dicarboxylate carrier model. If the chloroplast membrane contained only one type of dicarboxylate carrier, the addition of malate during (NH₃, glutamine, 2-OG)-dependent O₂ evolution should result in strong competition between malate and 2-OG for uptake and consequently inhibit, not stimulate, O₂ evolution. Figure 6 illustrates that while some competition is observed at very high malate 2-OG ratios, the rate is always well above the rate achieved in the absence of malate, i.e. full inhibition of 2-OG uptake is never approached, as one might expect at malate/2-OG ratios of 40:1. Conversely, high concentrations of 2-OG should inhibit the uptake of malate into the chloroplast, thereby reducing its stimulatory effect on O₂ evolution. The results of Figures 4 and 5 confirm this; however, again the extent of the competition or inhibition at high 2-OG/malate ratios is much

lower than would be predicted from the similarity in affinity of the dicarboxylate carrier for these two substrates (23). For example, Lehner and Heldt (8) showed that the uptake of malate into spinach chloroplasts could be inhibited 60% by the presence of a 3-fold higher concentration of succinate or fumarate. Our results show that the presence of a 3-fold higher concentration of 2-OG had virtually no effect on the stimulation of (NH₃, glutamine, 2-OG)-dependent O₂ evolution by malate.

We feel that results of this type strongly suggest the possible existence of separate carriers in the chloroplast membrane as depicted in Figure 7. Using this model, it can be seen that the explanation for malate-stimulated 2-OG uptake lies in the ability of malate to 'short circuit' the reaction and transport sequence which can support rapid rates of 2-OG uptake during (Asp, 2-OG)-dependent O₂ evolution, via the operation of a malate-Asp shuttle proposed originally by Heber (5) as a method for transferring reducing equivalents out of the chloroplast. *In vivo* uptake of 2-OG may be coupled to an OAA-malate shuttle (25) which could supply reducing equivalents for the hydroxypyruvate reduction step in peroxisomes (Fig. 7). However, the feasibility of such a shuttle is under question from suggestions that it is the mitochondrion and not the chloroplast which supplies the reducing equivalents for the hydroxypyruvate reaction (7).

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