

Purification and Separation of Plant Gibberellins from Their Precursors and Glucosyl Conjugates¹

Received for publication August 13, 1982 and in revised form May 2, 1983

MASAJI KOSHIOKA, KIYOTASHI TAKENO, FREDERICK D. BEALL, AND RICHARD P. PHARIS²
Department of Biology, University of Calgary, Calgary, Alberta T2N 1N4 Canada

ABSTRACT

A procedure using two small preparative columns (in sequence) of C₁₈ reverse phase Bondapak B material with methanolic extracts of plant tissue (*Pisum sativum* L., *Malus domestica* Borkh., *Pimpinella anisum* L.) yields two fractions: (i) gibberellin (GA) precursors, and (ii) free GA/GA methyl esters (GA-Me)/GA glucosyl conjugates. The discrete separation of (iii) free GA/GA-Me from (iv) GA glucosyl conjugates is then accomplished by a combination of differential solvent solubility and SiO₂ partition chromatography. All fractions are almost pigment free, and appreciable dry weight purification was accomplished for the GA precursor and free GA/GA-Me fractions. Solvent volumes can be kept low, no buffer salts are introduced, and each fraction (i, iii, iv) can be subjected directly to preparative or analytical reverse phase C₁₈ high performance liquid chromatography without recourse to solvent partitioning, and often without further purification.

Conventional methods for purification and isolation of GA³ and GA glucosyl conjugates from plant tissues involve partitioning of aqueous solutions against various organic solvents at one or more steps throughout the procedure. There exists in the literature of the past two decades numerous examples of various partitioning solvents used with a wide variety of plant tissues to extract or purify a structurally diverse range of acidic GA (2, 4, 13, 16, 17, 21, 31) and GA glucosyl conjugates (11, 27, 34, and numerous references cited in 26). However, partitioning techniques, as useful as they have been, do have a number of drawbacks. They are time consuming, especially where additional partitioning steps may be required through the use of PVPP (11) and/or charcoal (4, 36) purification procedures. They require a knowledge of the partition coefficient of each GA/GA precursor/GA glucosyl conjugate in question, for each of the various solvent systems used, in order to avoid variable and significant losses (5). Partitioning exposes the rather labile GA/GA conjugates to extremes of pH for varying periods of time, and partitioning may introduce salts into the organic phase containing the GA/GA conjugates. Finally, GA precursors, GA, and GA glucosyl conjugates have wide and overlapping ranges

of partition coefficients for any one solvent system (5, 11, 26, 27), making it impossible to design a simple partitioning sequence that will allow their resolution into discrete classes.

The above reasons have prompted us and others (1) to utilize a reverse phase C₁₈ material⁴ in an attempt to avoid the many problems of conventional solvent partitioning, the inconvenience of PVPP and charcoal purification columns, and still obtain extracts which are properly fractionated (*e.g.* GA separated from GA precursors and from GA glucosyl conjugates) and are pure enough for subsequent chromatography by preparative or analytical C₁₈ HPLC prior to more definitive characterization and quantitation. An earlier technique (using C₁₈ Sep-Paks[®]) will remove appreciable amounts of nonpolar substances, including pigments (1, 9), but as used by Barendse *et al.* (1) does not solve the problem of separating free GA from GA precursors or GA glucosyl conjugates. This inability to separate free GA (as a class) from their precursors and conjugates prior to reverse phase C₁₈ preparative or analytical HPLC columns is exacerbated by the fact that free GA and their glucosyl conjugates have very similar chromatographic behavior on reverse phase C₁₈ HPLC (14, 32).

In the present paper, we describe a method using two small preparative columns of C₁₈ reverse phase material on Bondapak B⁴ (C₁₈-PC) in combination with differential solvent solubility and SiO₂ partition column chromatography to yield three discrete fractions: (i) GA precursors such as kaurene, kaurenoid acid; (ii) acidic GA/GA-Me; and (iii) GA glucosyl conjugates.

MATERIALS AND METHODS

Plant Materials. Eleven-d-old seedlings of Alaska pea (*Pisum sativum* L.) were fed [³H]GA₂₀ (2.36 Ci/mmol) at 1.88 μCi/plant by spreading it on the first trifoliate leaf in 5 μl of 95% ethanol, and harvested 11 d later. One-year-old potted propagules (about 2 m in height) of apple (*Malus domestica* Borkh.) were fed [³H]GA₄ (1.3 Ci/mmol) at 8 μCi/propagule by injecting it in 400 μl of 0.6% ethanol into the stem at about 1 m height. Leaf tissue was harvested 7 d later. Suspension cultures of somatic cells of anise (*Pimpinella anisum* L.) were fed [³H]GA₄ (1.3 Ci/mmol) in aqueous solution in varying amounts and harvested 2, 204, and 348 h later (15).

C₁₈ HPLC Columns, HPLC Conditions, and HPLC-RC Conditions. A Waters Associates reverse phase C₁₈ μ-Bondapak column (3.9 mm × 30 cm) was used with gradient elution at 2 ml/min as described previously (14). The HPLC was a Waters Associates ALC/GPC R-401 liquid chromatograph with two model 6000 pumps, a model 660 solvent flow programmer, and a model U6K universal injector. The manually implemented linear gradient program was 0 to 10 min (pump A, 100%), 10 to 40 min (pump B, 0–70%), 40 to 50 min (pump B, 70%), 50 to 80 min (pump B, 100%); solvents, pump A: 10% MeOH in 1% acetic acid/H₂O, pump B: 100% MeOH (see also Figs. 2 and 3). The temperature was 22 to 25°C. A Berthold HPLC radioactivity

¹ Supported by Natural Sciences and Engineering Research Council Grants A-2585 to R. P. P., GO-633 to R. P. P. and J. N. Owens, and GO-154 to J. D. Bewley, D. M. Reid, and R. P. P.

² To whom all requests for reprints should be sent.

³ Abbreviations: C₁₈-PC, small preparative columns of 55- to 105-μm particle size C₁₈ monochlorooctadecylsilane material on Porasil B⁴; d.w., dry weight; EtOAc, ethyl acetate; GA, gibberellin(s); GA Me, GA methyl ester(s); MeOH, methanol; PVPP, polyvinylpyrrolidone; RC, radio-counting.

⁴ Waters Scientific Ltd., Mississauga, Ontario, Canada.

monitor (LB 503) with either a flow-through cell (300 μ l) or a pump delivering a Triton-X:xylylene:MeOH:PPO (330 ml:630 ml:150 ml:10 g) scintillant to a mixing cell (200 μ l, ratio 1:4 [effluent volume:scintillant volume]) was used for some $^3\text{H}/^{14}\text{C}$ standards and extracts, and was set at time constant 0.4 min, threshold factor 1.00, background 30 cpm, peak reject 20 counts, ratemeter range 300 cpm.

Bioassay. The Tan-ginbozu dwarf rice micro-drop assay (18) modified in that 0.5 μ l drops were applied, and measurement took place at hour 48, was used in serial dilution on each HPLC fraction (Fig. 4). For determination of GA_3 glucoside and GA_4 glucosyl ester, the Tan-ginbozu immersion assay (19) was utilized, an aliquot of the C_{18} -PC eluate being incorporated into 2 ml of 0.5% aqueous agar medium, on which 10 germinated seeds were planted. Length of the second leaf sheath was measured 6 d after planting.

Solvents. The H_2O for HPLC, hexane and EtOAc for SiO_2 partition columns, MeOH for extraction, and C_{18} -PC were purified by fractional distillation. The MeOH for HPLC was purchased as HPLC grade solvent. The HPLC solvents were filtered through 0.45- μm (HATF for H_2O) and 0.5- μm (FHUP for MeOH) pore size filters, respectively. The HPLC solvents were allowed to equilibrate, degassed, mixed, and during the elution of the column were maintained under vacuum over a magnetic stirrer to prevent gas accumulation. Extraction and C_{18} -PC solvents were reagent grade or distilled.

SiO_2 Partition Column. The procedure varied from that described earlier (3) in that the H_2O partition on the Woelm SiO_2^5 was made by gently rotating (15 rph) the container containing 500 g SiO_2 for 2 weeks with 20% H_2O by weight, the H_2O being added over 30 min. The SiO_2 was then allowed to equilibrate a further 7 d before use. The sample was dissolved in EtOAc:MeOH (1:1) or varying percentages of MeOH: H_2O (depending on its solubility) and loaded onto GFA paper discs (Whatman) or adsorbed onto a small amount of the SiO_2 and dried in a warm air stream, prior to placing onto the column. After completion of the gradient (HCOOH saturated, hexane:EtOAc), which elutes the free GA, the column was washed with at least 100 ml of 100% MeOH (or the discs and SiO_2 removed with air pressure, slurried in MeOH, and washed with excess MeOH on a Büchner funnel), to remove GA glucosyl conjugates. The MeOH wash of the SiO_2 was neutralized with 3 N NH_4OH , then taken to dryness *in vacuo* at 35°C prior to subsequent HPLC, either alone, or combined with the MeOH- and H_2O -soluble fractions from the C_{18} -PC (Fig. 1A and below).

Use of the C_{18} -PC Method for Authentic GA, GA Precursors, and GA Glucosyl Conjugates, and for Plant Extracts. C_{18} reverse phase Bondapak material⁴ was used to build small columns, either in glass tubing or syringe barrels (i.d. 2 cm, length 1.5–2.5 cm, weight of C_{18} material optimized at 4 times sample d.w.). Freeze-dried (or fresh) tissue (0.5–1.0 g d.w.) was homogenized with 40 ml of MeOH (MeOH: H_2O , 80:20 v/v) and (pH 6.7) filtered through Whatman No. 1 filter paper. The 80% MeOH filtrate (pH 6.7) (or 40 ml solution containing the standard compound; Table I) was passed under pressure at 2 to 4 ml/min through the first C_{18} -PC, the column then being washed with an additional 20 ml of 80% MeOH. An additional 36 ml of H_2O was added to the combined eluates of 80% MeOH and the resultant 50% MeOH (pH 6.5) solution (96 ml) was passed through a second C_{18} -PC (dimensions as noted above). This second column was then washed with an additional 20 ml of 50% MeOH.

Subsequent elution of the second C_{18} -PC with 60 ml 100% MeOH will remove the GA precursors (e.g. kaurene and kaurenolic acid) retained on the column. The combined 50% MeOH eluates (116 ml), which will contain all of the free GA and GA glucosyl conjugates (see "Results"), were taken to dryness in a rotary evaporator flask at 35°C by the addition of excess absolute

MeOH, or by removing most of the MeOH *in vacuo* at 35°C, and freezing and freeze-drying the residue. The dried residue of the 50% MeOH eluate was then extracted with 90 ml (30 ml \times 3) of EtOAc:MeOH (1:1) to dissolve all of the free GA. The residue remaining thereafter was further extracted with 90 ml (30 ml \times 3) of MeOH. The final residue was dissolved in 40 ml of H_2O , from which aliquots could be taken for radio- or bioassay.

Although the above technique of sequential solvent extraction of the dried residue of the 50% MeOH eluate from the second C_{18} -PC worked well (see "Results") for tissue amounts up to 1 g for most extracts, some extracts high in sugars may, when taken to dryness *in vacuo*, yield a residue that might occlude free GA during the above washing process. These residues can be dissolved in 98% MeOH (or dissolved first in a small amount of H_2O , MeOH then being added to make up a 98% solution). An equal volume of EtOAc is then added, yielding a flaky precipitate. This precipitate is then washed with two additional volumes of EtOAc:MeOH (1:1), the solution filtered and taken to dryness *in vacuo*. This residue is dissolved in minimal amounts of EtOAc:MeOH (1:1) followed by a few drops of 80% MeOH, and loaded onto the SiO_2 partition column as noted above. An alternative procedure, especially useful with extract residues from small tissue amounts (1.0 g or less), involves dissolving the entire residue from the second C_{18} -PC in a minimal amount of 80% MeOH, followed by EtOAc:MeOH, and applying it to the SiO_2 partition column as noted above. However, care must be taken not to overload the GFA paper discs or SiO_2 particles if occlusion effects with the hexane:EtOAc solvents of the SiO_2 partition column are to be avoided. Since all of the highly H_2O -soluble conjugates will be present on this SiO_2 partition column, an excess (about 200 ml) of absolute MeOH should be used when washing the column after completion of the hexane:EtOAc gradient. When extracts have been rich in [^3H]GA glucosyl conjugates of a highly polar nature (e.g. [^3H]GA₈-glucoside), we have used an 80% MeOH wash of the SiO_2 after the wash with absolute MeOH. This second wash yielded up to 10% additional GA conjugate radioactivity, and allowed for a quantitative recovery of radioactivity. However, significant amounts of SiO_2 are dissolved in 80% MeOH, and this may limit its further use.

Analytical or preparative C_{18} reverse phase HPLC was accomplished on the various fractions (after taking to dryness) as follows: (i) 100% MeOH eluate of the second C_{18} -PC (GA precursors fraction)—dissolve in 100% MeOH and inject in 10 to 100 μ l; (ii) EtOAc:MeOH-soluble fraction—dissolve in 50% MeOH and dilute if possible with H_2O to same MeOH percentage as initial solvent prior to injecting in 10 to 100 μ l; (iii) MeOH-soluble (EtOAc-MeOH insoluble)—dissolve in 10% MeOH and inject in 100 to 500 μ l; (iv) H_2O -soluble fraction (MeOH insoluble)—dissolve in 10% MeOH and inject in 100 to 500 μ l; (v) MeOH wash of the SiO_2 partition column—neutralize as noted above prior to taking to dryness, then dissolve in 10% MeOH and inject in 100 to 500 μ l.

Standard Samples. [^3H]GA₁⁶, [^3H]GA₄ (6), [^3H]GA₇ (synthesized by H. Aoki, N. Murofushi, K. Ogiyama, M. Noma, and T. Sassa), [^3H]GA₈ (20), [^3H]GA₉ (35), [^3H]GA₁₄ (8), [^3H]GA₂₀ (20), the methyl esters of [^3H]GA₄, [^3H]GA₉, and [^3H]GA₁₄, [^3H]kaurenolic acid (23), [^{14}C]GA₃⁷, [^{14}C]kaurene (synthesized by J. T. Lin and E. Heftmann), [^3H]GA₁₂ aldehyde, GA₃-O (3) glucoside (synthesized by G. Schneider and G. Sembdner), and GA₄ glucosyl ester (synthesized by N. Murofushi and N. Takahashi) were used, and detected by radioassay (aliquots dissolved in 1 ml MeOH or MeOH: H_2O were assayed by liquid scintillation spectrometry) or immersion bioassay (GA glucosyl conjugates).

Solvent Partitioning. The procedure used (Fig. 1A, right side) was a modification of a number of traditional procedures (2, 4,

⁵ Woelm SiO_2 , Cat. No. 02766, purchased from ICN Cleveland, OH.

⁶ Purchased from New England Nuclear, Lachine, Quebec, Canada.

⁷ Purchased from Amersham, Oakville, Ontario, Canada.

Table I. Elution of Authentic Standards of Gibberellins, Gibberellin Methyl Esters, Gibberellin Precursors, and Gibberellin Glucosyl Conjugates from C₁₈-PC

	Fraction (MeOH Concn.)								
	1 (10%)	2 (20%)	3 (30%)	4 (40%)	5 (50%)	6 (60%)	7 (70%)	8 (80%)	9 (100%)
	% eluted ^b								
Sample I ^a									
[³ H]GA ₁	92.2	6.3							
[¹⁴ C]GA ₃	98.3	1.2							
[³ H]GA ₄	97.5	2.5							
[³ H]GA ₇	93.0	4.2	0.3						
[³ H]GA ₈	94.5	2.9							
[³ H]GA ₉	96.9	2.1							
[³ H]GA ₁₄	97.0	2.1							
[³ H]GA ₂₀	94.5	3.7							
[³ H]GA ₄ -Me	78.4	19.8	4.7	2.0	3.7				
[³ H]GA ₉ -Me			4.9	39.0	45.0	5.2			
[³ H]GA ₁₄ -Me	42.7	19.4	14.1	6.5	10.5	9.7	5.0		
[³ H]GA ₁₂ aldehyde				0.4	53.5	39.8	4.4	1.4	0.5
[³ H]kaurenoic acid						1.6	5.5	86.7	3.3
[¹⁴ C]kaurene						2.7	8.3	57.8	27.2
Sample II ^a									
GA ₃ glucoside				97.7 ^d	0.5	0.5	0.4	0.4	0.5
GA ₄ glucosyl ester				92.0	3.1	0.9	1.1	1.1	1.8

^a 2×10^5 to 6×10^6 dpm of standard (in 1–100 μ l of methanolic solution) was diluted into 40 ml of 10% MeOH. Each solution was then passed through the column of C₁₈ reverse phase material^a (C₁₈-PC) (i.d., 1 cm; length, 1 cm; weight of C₁₈ material, 0.35 g). The column was then eluted stepwise with 40 ml of each of varying percentages of MeOH:H₂O. at pH 6.3 (10% MeOH) to 6.7 (80% MeOH).

^b Values are percentages of total radioactivity applied to column.

^c One or 2 μ g of GA₃ glucoside and GA₄ glucosyl ester, respectively, were dissolved in 40 ml of 40% MeOH.

Each solution was passed through the C₁₈-PC as noted above in footnote a.

^d Values based on assay (immersion) with dwarf rice var Tan-ginbozu.

13, 16, 17, 21, 31). The 80% MeOH extraction solvent was taken to the aqueous phase *in vacuo* at 35°C, and an equal volume of pH 9.0, 0.5 M KH₂PO₄:Na₂HPO₄ buffer added. This aqueous fraction was filtered through Whatman No. 1 paper on a Büchner funnel before adjusting the pH to 9.0 with 1 N KOH, and partitioning 3× against diethyl ether (3:5 ether:buffer). The aqueous phase was then adjusted to pH 3.0 with 1 N HCl and partitioned 5× against EtOAc (3:5 EtOAc:buffer). The acidic, EtOAc phase was then dried by freezing at –70°C and filtering out the ice on a Büchner funnel, prior to taking to dryness in a rotary evaporator flask at 35°C *in vacuo*. The dry residue was then placed in a freeze drier overnight to remove any traces of acetic acid. These acidic, EtOAc-soluble substances were then solubilized in as small an amount as possible (about 1–2 ml) of 0.5 M, pH 9.0 phosphate buffer (higher pH used if solubility was a problem), and loaded onto a PVPP column (10) eluted with 0.1 M pH 8.0 phosphate buffer. The 0.1 M buffer eluate was adjusted to pH 3.0 with 1 N HCl and partitioned as noted above with EtOAc. The acidic, EtOAc-soluble substances were dried as noted above, then dissolved in dry EtOAc:MeOH (1:1) for loading onto GFA (Whatman) discs for subsequent SiO₂ partition column chromatography.

RESULTS

Elution Characteristics from the C₁₈-PC of GA, GA Me, GA Precursors, and GA Glucosyl Conjugates. The elution/retention profiles of authentic GA with a wide variety of functional groups, and of representative GA-Me, GA precursors, and GA glucosyl conjugates were determined (Table I).

Eight GAs of widely varying polarity (5), three GA-Me, and three GA precursors were tested. After loading in 40 ml of 10% MeOH, all eight GAs were quantitatively eluted with the solvent (10% MeOH) followed by 40 ml 20% MeOH. Also, GA₉-Me,

GA₄-Me, and GA₁₄-Me (GA-Me should not be overlooked as being potential native compounds in plants; [see 22, 31]), were quantitatively eluted by the elution sequence of 10 to 50% MeOH (Table I). However, two GA precursors, kaurene and kaurenoic acid, did not begin to elute until a 60% MeOH concentration was reached. A third GA precursor, GA₁₂ aldehyde, was only half-eluted at a 50% MeOH concentration (Table I), the rest eluting at 60 and 70% MeOH concentration. As representatives of GA glucosyl conjugates, GA₃ glucoside and GA₄ glucosyl ester were dissolved in 40 ml of 40% MeOH, applied to the C₁₈-PC and eluted as above. The majority of each eluted with the first 40 ml of 40% MeOH (Table I; and would also be eluted by lower MeOH concentrations because of their high polarity).

These elution patterns from C₁₈-PC coincide with those noted on reverse phase C₁₈ HPLC, namely, a large number of GA, GA-Me, and GA glucosyl conjugates were eluted by a linear gradient (10–73% MeOH:1% AcOH:H₂O), but kaurene, kaurenoic acid, and GA₁₂ aldehyde are only eluted with MeOH concentrations greater than 73% (14). Thus, the data provided by Table I, combined with our knowledge of GA elution from reverse phase C₁₈ HPLC (14), indicated that elution of the C₁₈-PC with 50% MeOH would remove GA, GA and GA glucosyl conjugates Me, and about one-half the GA₁₂ aldehyde, leaving about 45% of GA₁₂ aldehyde and most of the kaurene and kaurenoic acid (Table I). A complete retention of GA₁₂ aldehyde could be obtained on the C₁₈-PC if 40% MeOH was used, but this is accomplished only by splitting the GA₉ Me fraction (Table I).

Effects of pH, in varying percentages of MeOH:H₂O, on elution of authentic GA from the C₁₈-PC. The influence of pH on the elution of [³H]GA₄, [³H]GA₁₄, and [³H]GA₁₂ aldehyde from the C₁₈-PC in various percentages of MeOH:H₂O was also examined (Table II). If the pH is decreased appreciably below that normally found in the MeOH:H₂O mixtures (e.g. pH 6.3–6.7

[Table I]), there was a greater retention of the GA on the C₁₈-PC. This retention effect was most pronounced at lower MeOH percentages (Table II). Therefore, the pH of methanolic plant extracts must be adjusted to approximately 6.5, if necessary, before performing the C₁₈-PC procedure to ensure proper elution of free GA while GA precursors are retained. Further, the pH of the eluate after the second C₁₈-PC should be checked and, if it has dropped below 6.0, then an additional volume of pH 6.5 MeOH:H₂O should be flowed through the second C₁₈-PC.

It is apparent from Table II that, while use of either 40% or 50% MeOH at pH 6.5 will elute 98+% of GA₄, the less polar GA₁₄ is retained in part. Hence, if the tissue of interest contains GA₁₄ (or GA₁₄ aldehyde), a higher percentage (60%) of MeOH must be used if elution of GA₁₄ or GAA₁₄ aldehyde from the second C₁₈-PC is desired. Gibberellin Ag may also be partially retained on the second C₁₈-PC at 40% MeOH (e.g. its elution characteristics lie between GA₄ and GA₁₄ [14]).

Although the present work does not attempt to explore the ramifications of the use of various pH and MeOH:H₂O percentages on separation of GA (other than those shown in Table II), it is obvious that such a use could become a powerful tool in isolating specific GA of interest in a relatively highly purified fraction.

Removal of Plant Pigments by C₁₈-PC Columns. Leaf tissue harvested from apple propagules was processed through the C₁₈-PC two column sequence in Figure 1A, and an estimate of the removal of neutral plant pigments was obtained. Absorption at 640 and 436 nm (9) was considered indicative of pigment (Chl and carotenoids) levels. Virtually all of the Chl were removed as was much of the 436 nm absorbing pigments. Thus, as noted earlier (1, 9), the C₁₈-PC method can be used in place of conventional ether partitioning at high pH for removal of pigments and other nonpolar substances with similar C₁₈ retention characteristics.

Ratio of C₁₈ Reverse Phase Bondapak B Material to Tissue Dry Weight and Recovery of [³H]GA and Their ³H Metabolites, Including [³H]GA Glucosyl Conjugate-like Compounds. Methanolic extracts from six samples of tissue (varying in d.w. from 0.5 to 1.46 g) containing variable amounts of the precursor [³H]GA₄, ³H-free GA metabolites, and ³H highly water-soluble metabolites (e.g. glucosyl conjugate-like compounds) were obtained from suspension cultures of somatic embryos of anise (15). The 80% MeOH extracts of six samples were passed through two C₁₈-

PC where the ratio of C₁₈ material to tissue dry weight varied from 2.05 to 4.00. The recovery of total radioactivity varied between 86% and 96% (mean = 90.3 ± 4.4). The 5 to 14% of radioactivity retained on the columns may represent incomplete elution of GA and GA glucosyl conjugates due to differential response of GA to pH (Table II) or may be accounted for by nonpolar conjugates (see 30) of the applied [³H]GA₄ or its metabolites.

Thus, the two column sequence of C₁₈-PC yields reproducibly high recoveries, and the amount of C₁₈-PC material, relative to the dry weight of tissue, does not significantly affect the recovery of GA or GA glucosyl conjugate-like substances.

Removal of Dry Weight by the C₁₈-PC Method. Extraction of four apple leaf tissue samples with 80% MeOH yielded about 19% of the tissue d.w. (18.8–21.5%). The C₁₈-PC two column sequence was quite effective in removing d.w. from the fraction containing all of the GA/GA-Me (Fig. 5; EtOAc:MeOH soluble [ii and iii]); only 0.7 to 2.3% of the MeOH-extractable d.w. remained. The GA precursor fraction (Fig. 5i) was also effectively purified; it contained about 6 to 9% of the MeOH-extractable d.w. Elution of the first C₁₈-PC with 100% MeOH (after first eluting with 80% MeOH) yielded the plant pigments in a fraction that contained 24% of the MeOH-extractable d.w. Based on Figure 1, where conjugates formed by metabolism of [³H]GA₄ were analyzed, only about one-third of the GA glucosyl conjugates will be present in the relatively low d.w. fraction (0.7–2.3% d.w.) that also contains all of the GA/GA-Me. The majority of conjugates will occur in the high d.w. residues that are dissolved only by MeOH (about one-half of conjugates and 34–41% of MeOH-extractable d.w.) → H₂O (about one-sixth of conjugates and 24–28% of the MeOH-extractable d.w.). Thus, the C₁₈-PC method does not significantly purify fractions containing the majority of GA glucosyl conjugates (and GA₃₂ if present). If use of high capacity C₁₈ reverse phase HPLC columns is not feasible, or use of aliquots is impractical, then some additional purification can be accomplished by partitioning into 1-butanol at pH 3.0 (25).

A Comparison of the C₁₈-PC Method with a Solvent Partitioning Procedure for [³H]GA Metabolites and Endogenous GA-Like Substances. The solvent partitioning and C₁₈-PC methods (Fig. 1) were compared in their efficacy by analyzing metabolites in an extract from pea seedlings fed [³H]GA₂₀. Both methods yielded 90% recovery of radioactivity after removal of nonpolar pigments (Fig. 1A). Highly buffer-soluble substances (presumably

Table II. Elution of Authentic Gibberellins from the C₁₈-PC at pH 3.0 to 6.5 in 40 to 80% Aqueous Methanol.

Eluent Solutions	GA ^a	Recovered Radioactivity at Following pH of Eluant Solution				
		3.0	4.0	5.0	6.9	6.5
80% MeOH	[³ H] GA ₄	65.2	90.4	96.8	99.3	100.0
	[³ H] GA ₁₄	84.0	99.8	100.0	100.0	100.0
	[³ H] GA ₁₂ aldehyde	62.4	97.8	98.7	99.2	99.6
60% MeOH	[³ H] GA ₄	30.0	91.9	98.0	99.5	99.9
	[³ H] GA ₁₄	19.1	86.2	96.9	99.3	99.7
	[³ H] GA ₁₂ aldehyde		1.1	7.6	26.4	74.2
50% MeOH	[³ H] GA ₄	2.4	36.8	69.0	95.8	99.7
	[³ H] GA ₁₄	2.9	8.0	16.3	47.8	98.3
	[³ H] GA ₁₂ aldehyde			1.4	11.8	33.7
40% MeOH	[³ H] GA ₄		1.0	45.2	95.2	97.6
	[³ H] GA ₁₄	1.0	2.3	8.6	26.8	52.7
	[³ H] GA ₁₂ aldehyde				1.4	5.2

^a Radioactive GA standard (5 × 10⁵ dpm) was diluted into 40 ml of pH 3.0 MeOH:H₂O solution. Each solution was then passed through the column of C₁₈ reverse phase material⁵ (C₁₈-PC) (i.d., 2.2 cm; length, 1.7 cm; weight of C₁₈ material, 2.5 g). The column was then eluted stepwise with 40-ml volumes of MeOH:H₂O at varying percentages and pH.

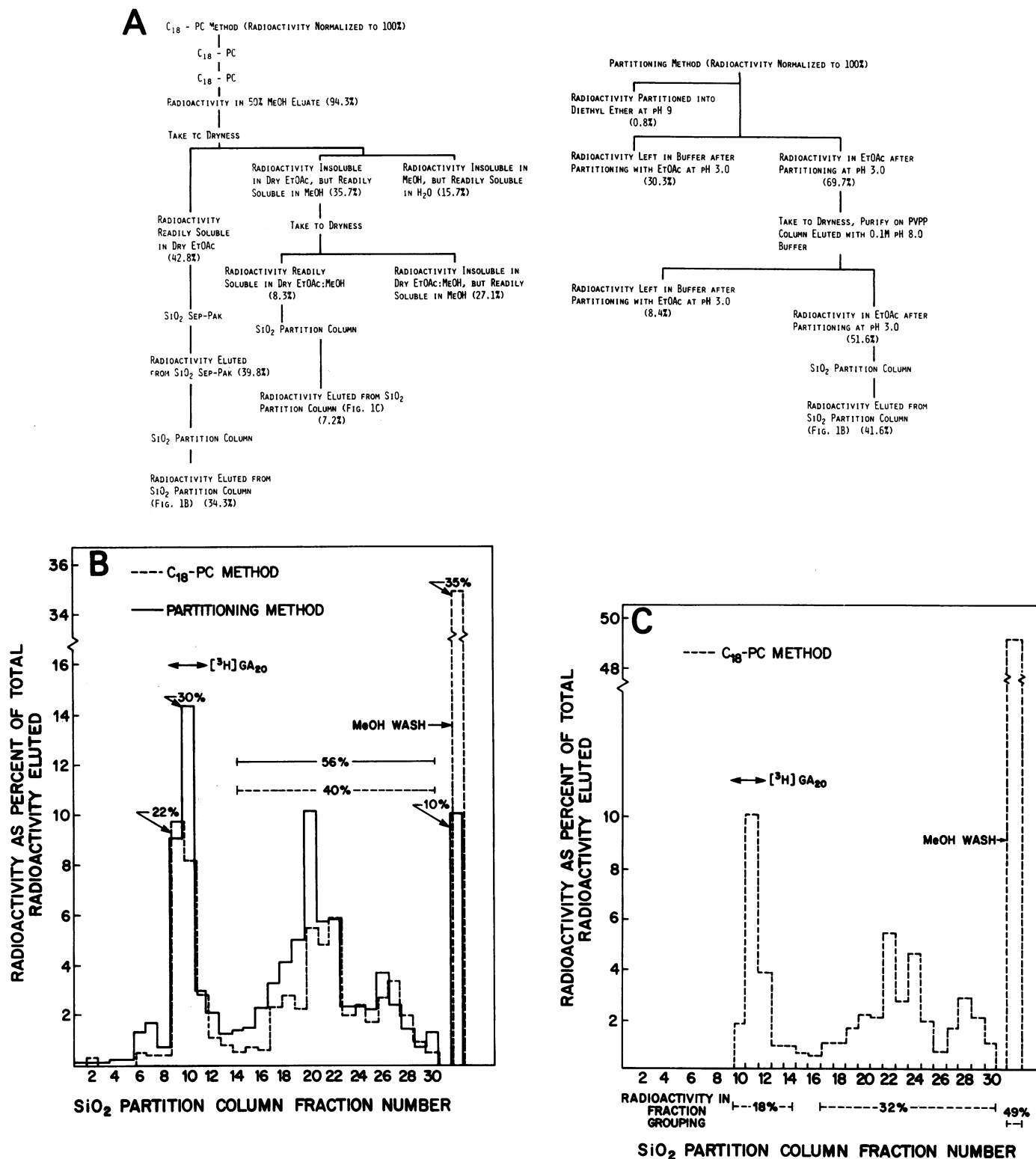


FIG. 1A. Purification and chromatography schemes for the C₁₈-PC method (left) and traditional solvent partitioning method (right) utilized for each of two halves of an 80% MeOH extract of 2.6 g d.w. of freeze-dried shoots of Alaska pea seedlings to which [³H]GA₂₀ (2.36 Ci/mmol) at the dosage of 1.58 μCi/plant had been applied (first trifoliolate leaf 11 d prior to harvest). The initial radioactivity has been normalized to 100% for each half of the MeOH extract. Specific details of both schemes are given in "Materials and Methods." Two SiO₂ partition columns were developed for the C₁₈-PC method in order to determine the relative proportion of 'GA conjugate-like' radioactivity in residues that were sequentially soluble in (i) dry EtOAc (Fig. 1B) and (ii) dry EtOAc:MeOH (50:50) (Fig. 1C).

FIG. 1B. SiO₂ partition column profiles of radioactivity (extracted from pea seedlings fed [³H]GA₂₀; see Fig. 1A) which is soluble in dry EtOAc (---) from the C₁₈-PC method, and which is acidic, EtOAc soluble (—) from the solvent partitioning method. The precursor, [³H]GA₂₀, elutes in fractions 9 to 11. A logical metabolite GA₂₉ (27) would elute in fractions 20 to 22. A MeOH wash of the column, following completion of the hexane:EtOAc gradient, was used to remove radioactivity which was not eluted by the gradient solvents. Glucosyl esters and glucosides of GA will be washed from the SiO₂ partition column by MeOH (Figs. 2 and 3). See Figure 1A and "Materials and Methods" for further details on the extract, tissue, or purification scheme.

FIG. 1C. SiO₂ partition column profile of the radioactivity that is insoluble in dry EtOAc, but is soluble in a 1:1 mixture of dry EtOAc:MeOH (---). The dry EtOAc-soluble radioactivity profile is shown in Figure 1B. Additional details are given in Figure 1, A and B, and in "Materials and Methods."

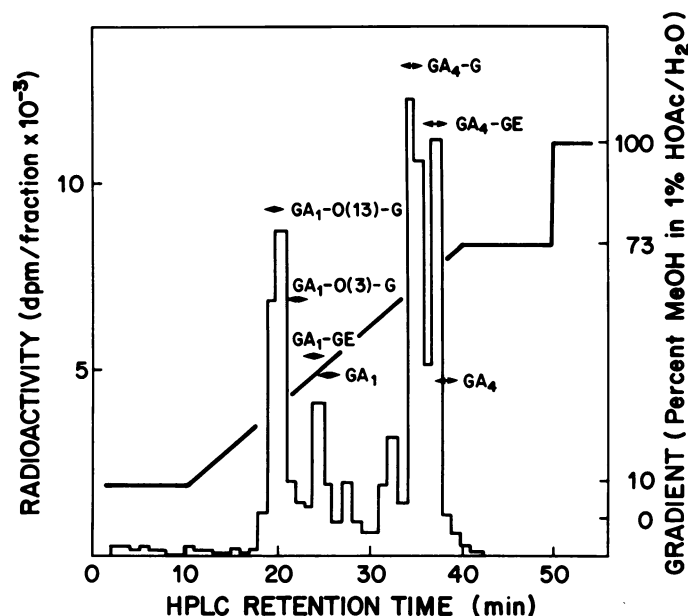


FIG. 2. Reverse phase analytical C_{18} HPLC profile of $[^3H]$ GA glucosyl conjugate-like substances from the MeOH wash fraction of a SiO_2 partition column which had been loaded with the dry EtOAc:MeOH (1:1)-soluble substances from leaves of apple propagules which had been fed $[^3H]GA_4$ 7 d prior to harvest (M Koshioka, G Edwards, J Taylor, and R Pharis, unpublished data). This profile is typical of similar fractions from extracts of other plants fed $[^3H]GA_4$, or other less polar $[^3H]GA$, in that GA glucosyl ester-like conjugates (especially) and GA glucoside-like conjugates (to a lesser degree) of the precursor tend to predominate, with conjugates of more polar metabolites being in the minority (due to differential solubility in the dry EtOAc:MeOH). Retention times of standard GA_1 and GA_4 glucosides (GA_1G , GA_4G) and GA_1 and GA_4 glucosyl esters (GA_1GE , GA_4GE) are shown by (\leftrightarrow).

GA glucosyl conjugates) constituted only 38.7% of the radioactivity in the solvent partitioning method. To this amount should be added the radioactivity eluted in the MeOH wash of the SiO_2 partition column (Fig. 1B, normalized to 5.2% of the initial radioactivity), bringing the total of GA glucosyl conjugate-like substances to 43.9% for the solvent partitioning method. Substances which were insoluble in dry EtOAc, or a mixture of dry EtOAc:MeOH (50:50), constituted 42.8% of the radioactivity for the C_{18} -PC method, to which should be added the 19.1% (calculated on basis of initial radioactivity) radioactivity eluted in the MeOH wash of the two SiO_2 partition columns (Fig. 1B, 35% normalized to 15.0%; Fig. 1C, 49% normalized to 4.1%; total = 19.1%), bringing the total of GA glucosyl conjugate-like substances to 61.9% for the C_{18} -PC method. The solvent partitioning procedure also yielded an increased amount of a compound which is probably 'ring C/D rearranged GA_{20} ' (Fig. 1B, fractions 6 and 7) and part or all of the increase in the fraction 6, 7 compound may originate from hydrolysis of a conjugate (7).

Thus, solvent partitioning, at least as used by us, may hydrolyze a significant portion of conjugate-like substances into free GA.

The qualitative spectrum of free $[^3H]GA$ metabolites is similar between the two methods (Fig. 1B). Procedures used to dissolve the dry residue after C_{18} -PC were examined through the use of SiO_2 partition column chromatography and analytical C_{18} HPLC. The radioactivity readily soluble in dry EtOAc (42.8% of initial radioactivity, Fig. 1A, left) is 65% free GA, and 35% GA glucosyl conjugate-like substances (MeOH wash; Fig. 1B; see also 11, 26). The radioactivity insoluble in dry EtOAc, but readily soluble in MeOH (35.7% of initial radioactivity; Fig. 1A, left) can be further categorized by the sequential use of EtOAc:MeOH

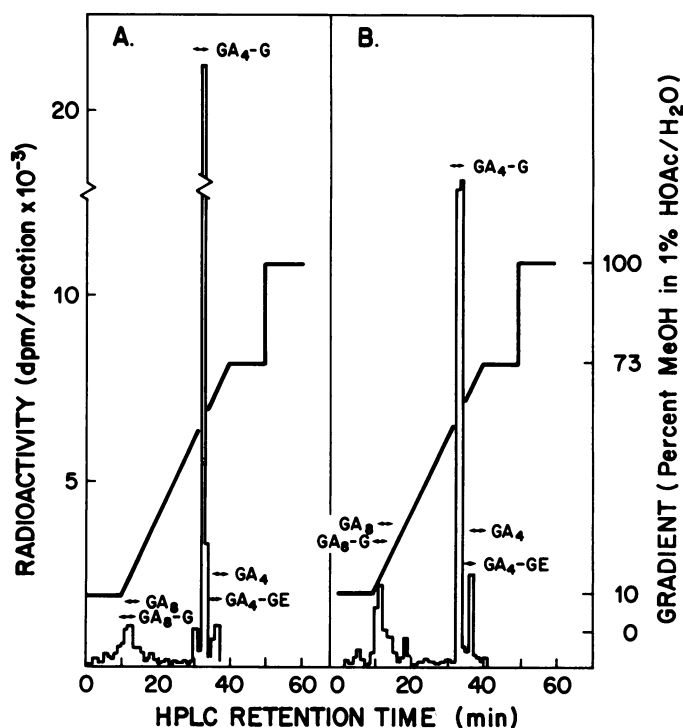


FIG. 3. Reverse phase analytical C_{18} HPLC profile of $[^3H]GA$ glucosyl conjugate-like substances from the MeOH-soluble (e.g. dry EtOAc:MeOH-insoluble) fraction (A) and the H_2O -soluble (e.g. MeOH-insoluble) fraction (B) of extracts of somatic embryos of anise which had been fed $[^3H]GA_4$ 348 h previously (15). The retention times of standards of GA_4 and GA_8 , GA_4 glucosyl ester (GA_4G), and GA_4 and GA_8 glucosides (GA_4G and GA_8G) are also given (\leftrightarrow).

(8.3%) \rightarrow MeOH (27.1%). The SiO_2 partition column of the EtOAc:MeOH-soluble fraction (8.3% of the initial radioactivity) yielded about half as free $[^3H]GA$, and half as $[^3H]$ glucosyl conjugate-like compounds (e.g. MeOH wash of the SiO_2 partition column, Fig. 1C). Thus, a good solvent mixture for dissolving all of the free GA in a small volume of liquid is EtOAc:MeOH (1:1). However, one must be aware that appreciable amounts of GA glucosyl conjugates are also dissolved by EtOAc:MeOH (e.g. see Fig. 1, B and C, 'MeOH wash' fractions).

An indication of the qualitative spectrum of $[^3H]GA$ glucosyl conjugate-like substances can be obtained by subjecting the purported conjugate fractions to analytical C_{18} HPLC (14). Thus, for apple tissue fed $[^3H]GA_4$, EtOAc:MeOH-soluble compounds that are eluted in the MeOH wash of a SiO_2 partition column (Fig. 1, B and C) include GA_4 glucoside- and GA_4 glucosyl ester-like conjugates and also conjugates of one of its metabolic products, $[^3H]GA_1$ (Fig. 2). Compounds that are sequentially soluble in MeOH \rightarrow H_2O (but insoluble in dry EtOAc:MeOH) as per the C_{18} -PC method (Fig. 1A) are shown in Figure 3, A and B. They include not only the more polar conjugates (e.g. $[^3H]GA_8$ glucoside-like substance), but also appreciable amounts of less polar conjugates (e.g. $[^3H]GA_4$ glucoside- and $[^3H]GA_4$ glucosyl ester-like substances). Although we term these 'conjugate-like substances,' all of the above-mentioned $[^3H]GA_4$ metabolites were subsequently subjected to isocratic HPLC-RC before and after hydrolysis, with identification of the GA moiety after hydrolysis being made by isocratic HPLC-RC and/or GLC-RC (15). Virtually no free GA were present upon isocratic HPLC analysis before hydrolysis, thereby confirming that occlusion of free GA in the EtOAc:MeOH-insoluble residue was not occurring and that complete solubilization of free GA by the EtOAc:MeOH solvent is effected (however, see "Materials and Methods" and Fig. 5 in reference to use of $H_2O \rightarrow$ EtOAc:MeOH to prevent

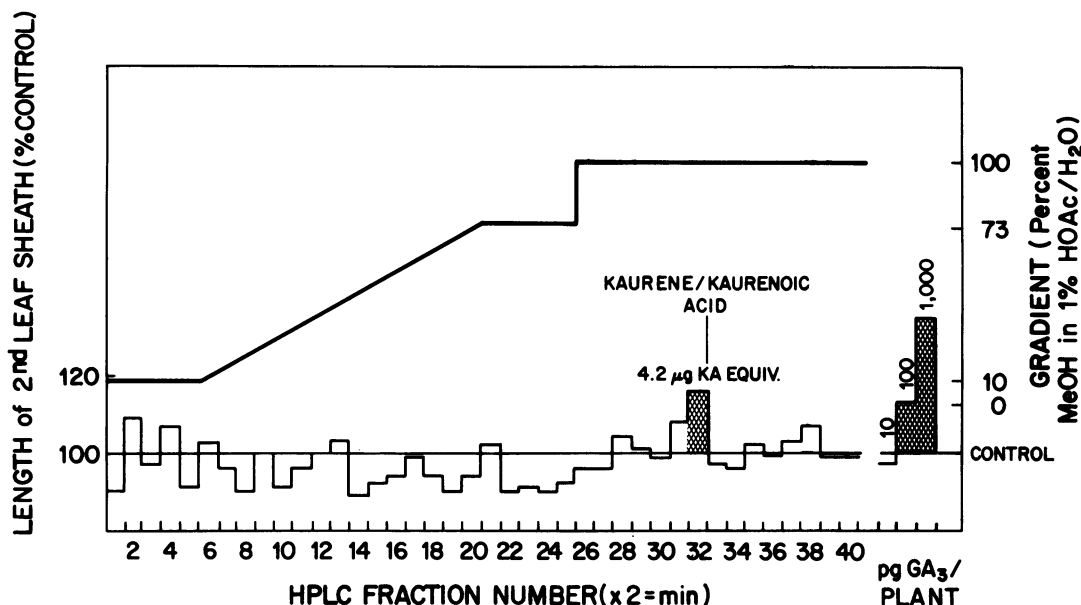


FIG. 4. Reverse phase C_{18} HPLC profile of the GA precursor fraction eluted with 100% MeOH from the C_{18} -PC (Fig. 5) of a MeOH (80%) extract of 0.88 g d.w. somatic grape embryos (29). Biological activity (4.2 μ g kaurenoic acid eq/0.88 g tissue d.w.) had initially been detected on the dwarf rice microdrop bioassay (in serial dilution) in the bulked GA precursor fraction, and was similarly detected in HPLC fractions 31 and 32. Statistically significant activity is shown by cross-hatching (Rt 60–64 min). Fractions 31 and 32 were subsequently subjected to packed column GLC-MS; kaurene and kaurenoic acid were identified based on complete MS which were identical to MS of authentic standards.

possible occlusion effects).

Thus, while there is a relationship between polarity of the conjugate and its solubilization in the sequence of dry EtOAc \rightarrow EtOAc:MeOH \rightarrow MeOH \rightarrow H₂O, appreciable amounts of the more polar GA conjugates are solubilized by dry EtOAc:MeOH (Fig. 2). However, even absolute MeOH will not completely solubilize all of the [³H]GA₄ glucosyl ester-like conjugate (Fig. 3, A and B). Therefore, in order to have all of the GA glucosyl conjugates in one fraction prior to gradient-eluted reverse phase C_{18} HPLC, we recommend that the MeOH wash fraction from the SiO₂ partition column (e.g. on which is loaded substances soluble in dry EtOAc:MeOH) be combined with those residual substances that are soluble in MeOH \rightarrow H₂O, or soluble in 10% MeOH (e.g. scheme shown in Fig. 5).

Use of the C_{18} -PC Method for Endogenous GA-Like Substances. Barendse *et al.* (1) have shown the usefulness of the C_{18} reverse phase Bondapak B material for obtaining biologically active GA-like substances, although their technique would have grouped free GA with at least a portion of GA precursors, and with GA glucosyl conjugates. We have used the C_{18} method successfully on a variety of plant materials, and describe one example below for endogenous GA precursors.

The second C_{18} -PC after elution with 50% MeOH will retain about 70% of the kaurene, almost all of the kaurenoic acid (Table I), about 46% of the GA₁₂-aldehyde, and many metabolites of kaurene. Significant biological activity was thus obtained in a 100% MeOH eluate of the second C_{18} -PC (after it had first been eluted with 50% MeOH; as in Fig. 1, A and C) for extracts of somatic embryos of grape (29). The 100% MeOH eluate (Fig. 5i) was subjected directly to a gradient-eluted analytical reverse phase C_{18} HPLC (14). Bioassay of the HPLC fractions showed a significant peak of activity in the kaurene/kaurenoic acid region (Fig. 4). This peak was subjected to GLC-MS and yielded MS identical to those of kaurene and kaurenoic acid (29). The C_{18} -PC \rightarrow HPLC sequence may thus be useful in obtaining these and other GA precursors from plant extracts. However, one should keep in mind the possibility that nonpolar GA conjugate-like compounds (28), if present, may also elute in the 100% MeOH eluate of the second C_{18} -PC.

DISCUSSION

The C_{18} -PC method (as outlined in Figs. 1A and 5) can safely replace solvent partitioning techniques as an extraction/purification method for GA precursors, free GA, GA-Me, and GA glucosyl conjugates from the same tissue sample. When used in combination with a SiO₂ partition column (isocratic, step or gradient-eluted), GA and/or GA-Me can be readily separated from GA glucosyl conjugates, thus yielding three discrete fractions: (i) many GA precursors, including kaurene and kaurenoic acid, (ii) free GA/GA-Me, and (iii) GA glucosyl conjugates. All of fractions (i–iii) can be subjected directly to preparative and/or analytical reverse phase C_{18} HPLC, with detection and quantitation by appropriate means (e.g. bioassay, radioassay, GLC-MS, GLC-MS-SIM). For use in bioassay, purity of free GA (ii) is as good or better than PVPP- and charcoal:celite-purified fractions from the solvent partitioning method. And, use of the bulked fraction of GA glucosyl conjugates (iii) in serial dilution on a combination of micro-drop (18) and immersion (19) bioassays on Tan-ginbozu dwarf rice has yielded significant activity without severe toxicity, indicating that these bioassays can also be used to assay fractions eluted from reverse phase C_{18} HPLC of (iii).

The C_{18} -PC method is of special interest for work on the metabolism of radioactive GA/GA precursor/GA glucosyl conjugates, and lends itself well to preliminary separation of these three categories of compounds prior to more definitive assay by analytical reverse phase C_{18} HPLC-RC and/or sequential GLC-RC.

Our recommended procedure using the C_{18} -PC method is detailed in the form of a flow chart (Fig. 5). Where isolation of only one or a few free GA/GA precursors/GA glucosyl conjugates is desired, the procedure can be simplified, purification enhanced, and indeed preliminary chromatography often can be accomplished by loading the extract onto the second (or a third) C_{18} -PC in a reduced volume of MeOH (0–50%), and monitoring the eluate fractions. The elution pattern from such C_{18} -PC will be the same as that noted earlier (12, 14), although it should be remembered that GA glucosyl conjugates, if present, will also be eluted with the GA if the SiO₂ partition column (or analogous

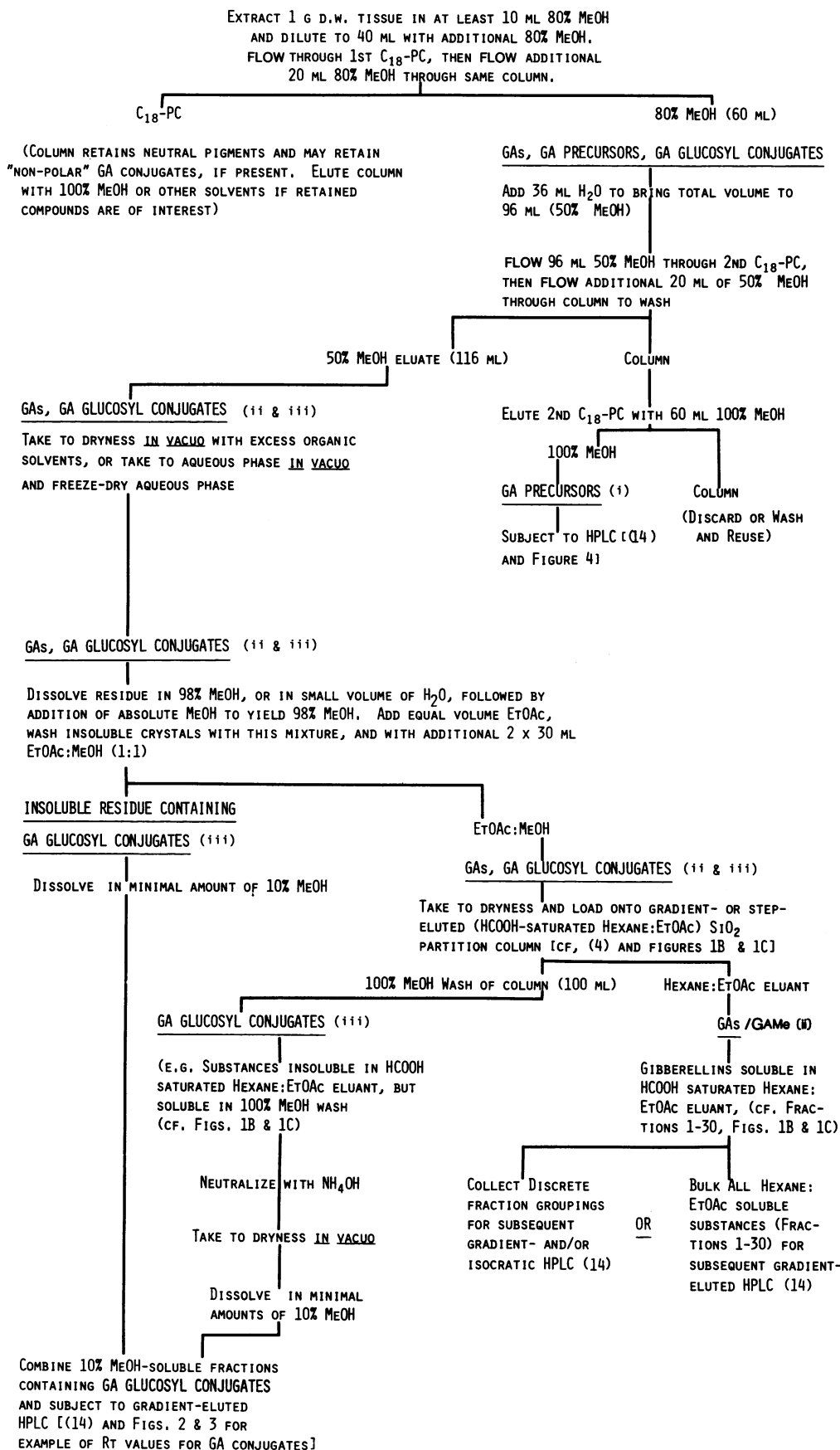


FIG. 5. A flow chart of the recommended procedure for use of the C₁₈-PC method for purification of (i) GA precursors, (ii) GA/GA-Me, and (iii) GA glucosyl conjugates from a 1-g d.w. sample of plant tissue. Increased amounts of tissue will require a proportional scaling up of amounts of C₁₈-reverse phase Bondapak B, and at least a partial scaling up of extraction and C₁₈-PC elution solvents. The pH of the MeOH:H₂O solutions should be maintained at 6.5 (see Table II). This procedure, however, splits GA₁₂ aldehyde into fractions (i) and (ii). Use of 40% MeOH (116 ml) to elute the second C₁₈-PC column will allow GA₁₂ aldehyde to be subsequently eluted quantitatively in fraction (i) by 60 ml of 100% MeOH. But, use of 40% MeOH then splits GA₉-Me between fractions (i) and (ii), so the user must decide which procedure is most appropriate. Gradient and/or isocratic C₁₈ HPLC (1, 12, 14, 32) is recommended for subsequent separation of components within (i, ii, and iii), with detection and quantitation by appropriate methods (e.g. radioassay [6-8, 14, 15], definitive bioassay [18, 19], GC-MS [4, 11], or GC-MS-SIM [12]).

procedure) step has not been accomplished.

A convenient procedure (see "Materials and Methods" for details and comments on avoiding occlusion effects) for tissue amounts of about 1 g (or less) involves loading the 50% MeOH eluate residue (about 130 mg for a 1 g d.w. tissue sample) from the second C₁₈-PC (e.g. GA and GA glucosyl conjugates) directly on to the SiO₂ partition column, free GA being eluted in the hexane:EtOAc gradient, the GA glucosyl conjugates being washed from the SiO₂ column with excess absolute MeOH.

Where chromatography of free GA at the SiO₂ partition column stage is not required (e.g. the free GA and GA conjugate fractions are to be chromatographed subsequently on C₁₈ HPLC *in toto*), we have found that short SiO₂ partition columns (5 g SiO₂; column diameter 15 mm i.d.) eluted isocratically with HCOOH-saturated hexane:EtOAc are a rapid and efficient way to effect separation of free GA from GA glucosyl conjugates. Elution of such a column with 70 ml of 5:95 (v/v) hexane:EtOAc yields >95% of the highly polar free [³H]GA₈, and <5% of most nonpolar of GA conjugates, [³H]GA₉ glucosyl ester (Pharis, Koshioka, Janzen, unpublished). Subsequent elution of the column with MeOH (as described in the "Materials and Methods" and "Results") yields almost all of the GA glucosyl conjugates. This procedure would be useful for nonradioactive extracts where one wanted to ensure that free GA would be quantitatively separated from GA glucosyl conjugates. When using radioactive GA, large amounts of residual precursor [³H]GA can be removed from free GA metabolites by step-eluting the SiO₂ partition column sequentially with two isocratic mixtures (e.g. 120 ml 45:55 [v/v] hexane:EtOAc to elute the precursor GA, [³H]GA₄, or [³H]GA₂₀, followed by 60 ml of 5:95 [v/v] hexane:EtOAc to elute the free [³H]GA metabolites), GA glucosyl conjugates being eluted subsequently with MeOH (25).

Reuse of the C₁₈-PC material is possible after extensive washing with methanol, chloroform, EtOAc, dimethyl sulfoxide, and H₂O. Washed C₁₈ material should of course be checked to ensure that proper elution characteristics are retained (e.g. to make certain the C₁₈ material is not coated with contaminants that might alter the normal elution pattern) and also for the presence of GA contaminants by an appropriate assay (e.g. ³H oxidation; bioassay of MeOH eluate) prior to reuse. Even so, we recommend that only new C₁₈ material be used for extracts known to be low in biological activity, or for radioactive extracts where very low levels of radioactivity are expected in metabolites.

Acknowledgments—We acknowledge with thanks the able technical assistance of Stania Horacek and Mayumi N. Koshioka, and gifts of C₁₈ Bondapak B material (Waters Scientific Ltd., Mississauga, Ontario) and standard compounds from A. Crozier (Department of Botany, Glasgow University, Scotland, U. K.), R. C. Durley (Department of Crop Science, University of Saskatchewan, Saskatchewan, Canada), E. Heftmann (U. S. Department of Agriculture, Berkeley, CA), N. Takahashi and N. Murofushi (Department of Agricultural Chemistry, University of Tokyo, Japan), G. Sembdner (Institut für Pflanzen Biochemie, Halle [Saale], DDR).

LITERATURE CITED

- BARENDSE GWM, PH VAN DE WERKEN, N TAKAHASHI 1980 High-performance liquid chromatography of gibberellins. *J Chromatogr* 198: 449-455
- COOMBE BG 1971 A polar gibberellin with high biological potency. *Science* 172: 856-857
- DURLEY RC, A CROZIER, RP PHARIS, GE McLAUGHLIN 1972 Chromatography of 33 gibberellins on a gradient eluted silica gel partition column. *Phytochemistry* 11: 3029-3033
- DURLEY RC, J MACMILLAN, RJ PRYCE 1971 Investigation of gibberellins and other growth substances in the seed of *Phaseolus multiflorus* and of *Phaseolus vulgaris* by gas chromatography-mass spectrometry. *Phytochemistry* 10: 1891-1908
- DURLEY RC, RP PHARIS 1972 Partition coefficient of 27 gibberellins. *Phytochemistry* 11: 317-326
- DURLEY RC, RP PHARIS 1973 Interconversion of gibberellin A₄ to gibberellins A₁ and A₂₄ by dwarf rice, cultivar Tan-ginbozu. *Planta* 109: 357-361
- DURLEY RC, RP PHARIS, JAD ZEEVAART 1975 Metabolism of [³H]gibberellin A₂₀ by plants of *Bryophyllum daigremontianum* under long- and short-day conditions. *Planta* 126: 139-149
- DURLEY RC, ID RAILTON, RP PHARIS 1974 Conversion of gibberellin A₁₄ to other gibberellins in seedlings of *Pisum sativum*. *Phytochemistry* 13: 547-551
- ESKINS K, HJ DUTTON 1979 Sample preparation for high-performance liquid chromatography of higher plant pigments. *Anal Chem* 51: 1885-1886
- GLENN JL, CC KUO, RC DURLEY, RP PHARIS 1972 Use of insoluble polyvinylpyrrolidone for purification of plant extracts and chromatography of plant hormones. *Phytochemistry* 11: 345-351
- HIRAGA K, T YOKOTA, N MUROFUSHI, N TAKAHASHI 1972 Isolation and characterization of a free gibberellin and glucosyl esters of gibberellins in mature seeds of *Phaseolus vulgaris*. *Agric Biol Chem* 36: 345-347
- JONES MG, JD METZGER, JAD ZEEVAART 1980 Fractionation of gibberellins in plant extracts of reverse phase high performance liquid chromatography. *Plant Physiol* 65: 218-221
- KAMIENSKA A, RC DURLEY, RP PHARIS 1976 Isolation of gibberellins A₃, A₄ and A₇ from *Pinus attenuata* pollen. *Phytochemistry* 15: 421-424
- KOSHIOKA M, M NOMA, T SASSA, K OGIYAMA, JS TAYLOR, SB ROOD, RL LEGGE, J HARADA, K TAKENA, RP PHARIS 1983 Reverse phase C₁₈ high pressure/performance liquid chromatography of acidic and conjugated gibberellins. *J Chromatogr* 256: 101-115
- KOSHIOKA M, TJ DOUGLAS, D ERNST, J HUBER, RP PHARIS 1983 Metabolism of [³H] gibberellin A₄ in somatic suspension cultures of anise. *Phytochemistry*. In press
- MACMILLAN J, PJ SUTER 1958 The occurrence of gibberellin A₁ in higher plants: isolation from the seed of runner bean (*Phaseolus multiflorus*). *Naturwissenschaften* 45(2): 1-2
- METZGER JD, JAD ZEEVAART 1980 Identification of six endogenous gibberellins in spinach shoots. *Plant Physiol* 65: 623-626
- MURAKAMI Y 1968 A new rice seedling bioassay for gibberellins, "Microdrop Method", and its use for testing of rice and morning glory. *Bot Mag (Tokyo)* 81: 33-43
- MURAKAMI Y 1973 A method for detecting bound gibberellin by rice seedling test. *Chem Regul Plants (Tokyo)* 8: 40-44
- MUROFUSHI N, RC DURLEY, RP PHARIS 1977 Preparation of radioactive GA₂₀, GA₂₅ and GA₂₉. *Agric Biol Chem* 41: 1075-1079
- MUROFUSHI N, S IRIUCHIJIMA, N TAKAHASHI, S TAMURA, J KATO, Y WADA, E WATANABE, T AOYAMA 1966 Isolation and structure of a novel C₂₀ gibberellin in bamboo shoots. *Agric Biol Chem* 30: 917-924
- NOMA M, J HUBER, RP PHARIS 1979 Occurrence of Δ¹⁰⁽¹⁰⁾gibberellin A₁ counterpart, GA₁, GA₄ and GA₇ in somatic cell embryo cultures of carrot and anise. *Agric Biol Chem* 43: 1793-1794
- RAILTON ID, RC DURLEY, RP PHARIS 1975 A lack of correlation between the biological activity and rate of metabolism of *ent*-[³H]-17-kaurenoic acid by seedlings of dwarf rice cv. Tan-ginbozu. *Plant Cell Physiol* 16: 943-951
- REEVE DR, A CROZIER 1978 The analysis of gibberellins by high performance liquid chromatography, isolation of plant growth substances. *Soc Exp Biol Semin Ser* 4: 41-77
- ROOD SB, RP PHARIS, M KOSHIOKA 1983 Reversible conjugation of gibberellins *in situ* in maize. *Plant Physiol*, 73: 0000-0000
- SCHNEIDER G 1982 Gibberellin conjugates. In A Crozier, ed, *The Biochemistry and Physiology of Gibberellins*. Praeger, New York. In Press
- SEMBDNER G, G SCHNEIDER, J WEILAND, K SCHREIBER 1964 Über ein gebundenes Gibberellin aus *Phaseolus coccineus* L. *Experientia* 20: 89-90
- SPONSEL VM, J MACMILLAN 1975 The metabolism of gibberellins A₉, A₂₀ and A₂₉ in immature seeds of *Pisum sativum* cv. Progress No. 9. *Planta* 125: 181-195
- TAKENO K, M KOSHIOKA, RP PHARIS, K RAJASEKARAN, MG MULLINS 1983 Endogenous gibberellin-like substances in somatic embryos of grape (*Vitis vinifera* × *V. rupestris*) in relation to embryogenesis and the chilling requirement for subsequent development of mature embryos. *Plant Physiol* 73: 0000-0000
- WAMPLE RL, RC DURLEY, RP PHARIS 1975 Metabolism of gibberellin A₄ by vegetative shoots of Douglas fir at three stages of ontogeny. *Physiol Plant* 31: 273-278
- YAMAGUCHI I, T YOKOTA, N MUROFUSHI, Y OGAWA, N TAKAHASHI 1970 Isolation and structure of a new gibberellin from immature seeds of *Prunus persica*. *Agric Biol Chem* 34: 1439-1441
- YAMAGUCHI I, T YOKOTA, S YOSHIDA, N TAKAHASHI 1979 High pressure liquid chromatography of conjugated gibberellins. *Phytochemistry* 18: 1099-1102
- YAMANE H, N TAKAHASHI, K TAKENO, M FURUYA 1979 Identification of gibberellin A₉ methyl ester as a natural substance regulating formation of reproductive organs in *Lygodium japonicum*. *Planta* 147: 251-256
- YOKOTA T, N MUROFUSHI, N TAKAHASHI, S TAMURA 1971 Gibberellins in immature seeds of *Pharbitis nil*. III. Isolation and structure of gibberellin glucosides. *Agric Biol Chem* 35: 583-595
- YOKOTA T, DR REEVE, A CROZIER 1976 The synthesis of [³H] gibberellin A₉ with high specific activity. *Agric Biol Chem* 40: 2091-2094
- ZEEVAART JAD 1971 Effects of photoperiod on growth rate and endogenous gibberellins in the long day rosette plant spinach. *Plant Physiol* 47: 821-827