

Scalable and Consolidated Microbial Platform for Rare Earth Element Leaching and Recovery from Waste Sources

Nathan M. Good, Christina S. Kang-Yun, Morgan Z. Su, Alexa M. Zytneck, Colin C. Barber, Huong N. Vu, Joseph M. Grace, Hoang H. Nguyen, Wenjun Zhang, Elizabeth Skovran, Maohong Fan, Dan M. Park, and Norma Cecilia Martinez-Gomez*



Cite This: *Environ. Sci. Technol.* 2024, 58, 570–579



Read Online

ACCESS |



Metrics & More



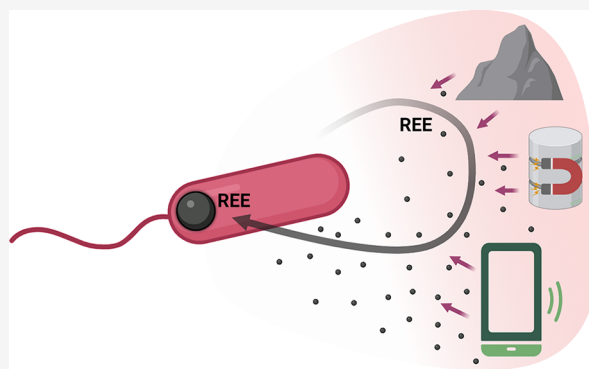
Article Recommendations



Supporting Information

ABSTRACT: Chemical methods for the extraction and refinement of technologically critical rare earth elements (REEs) are energy-intensive, hazardous, and environmentally destructive. Current biobased extraction systems rely on extremophilic organisms and generate many of the same detrimental effects as chemical methodologies. The mesophilic methylotrophic bacterium *Methylobacterium extorquens* AM1 was previously shown to grow using electronic waste by naturally acquiring REEs to power methanol metabolism. Here we show that growth using electronic waste as a sole REE source is scalable up to 10 L with consistent metal yields without the use of harsh acids or high temperatures. The addition of organic acids increases REE leaching in a nonspecific manner. REE-specific bioleaching can be engineered through the overproduction of REE-binding ligands (called lanthano-phores) and pyrroloquinoline quinone. REE bioaccumulation increases with the leachate concentration and is highly specific. REEs are stored intracellularly in polyphosphate granules, and genetic engineering to eliminate exopolyphosphatase activity increases metal accumulation, confirming the link between phosphate metabolism and biological REE use. Finally, we report the innate ability of *M. extorquens* to grow using other complex REE sources, including pulverized smartphones, demonstrating the flexibility and potential for use as a recovery platform for these critical metals.

KEYWORDS: bioaccumulation, bioconcentration, electronic waste, lanthanide, metal-binding protein, neodymium, bioleaching, acid-free leaching



INTRODUCTION

Global demand for rare earth elements (REEs) is at an all-time high and is steadily increasing, but their supply is susceptible to market and national interest perturbations.^{1–3} REEs, composed of the lanthanides (Lns), scandium, and yttrium, are critical metals for modern clean energy, communication, advanced transportation, consumer electronics, and defense technologies, underpinning a broader economy.^{4–6} China maintains its status as the world's foremost producer of REEs, generating heavy reliance on a single market that could jeopardize supply and compromise national security.^{7,8} Though domestic REE production is on the rise, environmental and health concerns loom over traditional mining operations.^{9–11} The development of technologies for REE reuse and recycling has garnered interest as a means of moving toward independence from foreign importation while simultaneously generating a robust, resilient supply of these critical metals.¹²

Several challenges continue to stymie the development of a safer, more environmentally sustainable REE supply chain. These include (1) the requirement of high temperatures and

pressures and the inclusion of harsh acids for the extraction of poorly soluble REE, which can produce radioactive waste products; (2) the preference for high-grade sources (>0.2% REE m/m) for cost-efficient recovery, leaving low-grade and waste sources as untapped reservoirs of valuable REE; and (3) the necessity for hundreds of costly, hazardous processing steps for successful separation of co-occurring REE in minerals.^{2,13}

Microbiological REE leaching and extraction methods offer promising alternatives to current state-of-the-art methods (hydrometallurgical, pyrometallurgical, and electrometallurgical approaches) that produce large quantities of sludge, acidic wastewater, atmospheric pollution, and radioactive tail-

Received: August 24, 2023

Revised: November 13, 2023

Accepted: November 20, 2023

Published: December 27, 2023



ings.^{9,14–16} Given that conventional rare earth ores (e.g., bastnaesite, monazite, and xenotime) are typically oxidized and nonsulfidic,¹⁷ conventional bioleaching using acidophilic sulfur and iron oxidizers, which has been widely practiced commercially for sulfidic copper and gold ores, is not directly applicable. Rather, REE bioleaching approaches typically involve the production of organic acids (e.g., citric, gluconic, and oxalic acids)^{18,19} from sugar-based carbon sources by heterotrophic bacteria or fungi to leach REEs into solution. While these approaches can be effective for overall leaching and even provide an economical benefit,^{19–21} they indiscriminately dissolve metal feedstocks and require additional process(es) to separate desired metals from the remaining leachate. Microbe-mediated REE adsorption using native and engineered cell surface functional groups (e.g., carboxylates, phosphates, and metal-binding outer membrane proteins)^{22–25} can be an effective and potentially economic strategy^{26,27} for selective sequestration of REE from bulk samples, but this method requires solution-based feedstocks derived from costly chemical leaching and pH-adjustment processes.

Microbial metal accumulation and biomineralization have been shown to be effective strategies for the recovery of Hg,²⁸ Au,^{29–31} and U,^{32,33} but has been underexplored as an REE recovery approach. Bioaccumulation and biomineralization of REEs in mesophilic bacteria were first shown in the model methylotroph *Methylobacterium* (also known as *Methylorubrum*) *extorquens* AM1.³⁴ Methylotrophic bacteria, organisms that thrive on inexpensive, readily available one-carbon compounds such as methane and methanol, therefore provide an attractive new approach for REE bioleaching and extraction due to their natural ability to acquire Lns from the surrounding environment.^{35,36} This includes soluble and insoluble REE sources, such as electronic waste (E-waste).^{37–39} *M. extorquens* AM1 was the first organism reported to grow using REE from electronic waste.³⁶ Mesophilic methylotrophs like *M. extorquens* AM1 have dedicated systems for acquisition, uptake, and intracellular storage of Lns as polyphosphate granules, making them effective agents of bioleaching and bioaccumulation without the need for high acidity or temperature.³⁴ REE use by mesophilic methylotrophs was thought to be restricted to the light Ln, but recently, a genetic variant of *M. extorquens* AM1 was isolated and characterized that can transport, store, and grow using the heavy Ln, gadolinium.⁴⁰ Detailed genetic and Ln uptake studies indicate the likely possibility of an additional system dedicated to the acquisition and transport of heavy Lns. Thus, methylotrophs, such as *M. extorquens* AM1, may already possess the biological means to separate light and heavy Lns, and have the potential to be engineered for uptake of specific Ln species from complex feedstocks. To date, the potential of methylotrophs as a platform for REE recovery has not been rigorously investigated, including the selectivity of the REE uptake.

Here, we show the scalability of this growth up to 10 L with Nd magnet swarf, waste powders, and filings generated during magnet production and an E-waste analogue without the requirements of high temperatures or harsh acids. We also show the potential for process improvement by leveraging the genetic tractability and suite of genetic tools available for *M. extorquens* AM1 to engineer enhanced, nonacidic REE bioleaching (~20-fold) and bioaccumulation (~50-fold) attributes. This study provides a proof-of-principle demonstration of *M. extorquens* AM1 as a scalable platform for biomining and bioaccumulation of REEs from complex

feedstocks, such as magnet swarf, without the need of harsh acids and high temperatures, greatly reducing potential hazards and obstacles to generating a circular economy (Figure 1).

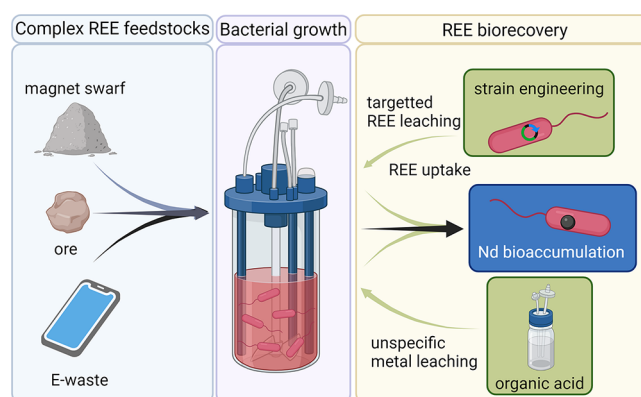


Figure 1. *M. extorquens* AM1 is a platform for Nd recovery through efficient, high-yield bioaccumulation from complex sources. REEs in complex feedstocks are accessible to *M. extorquens* AM1 due to its innate ability to selectively acquire, transport, and accumulate metals in polyphosphate granules. Metal leaching and REE uptake can be enhanced by the addition of organic acid to the culture medium and strain engineering, resulting in greater Nd recovery yields.

MATERIALS AND METHODS

Chemicals. All chemicals were purchased from Fisher Scientific (Pittsburgh, PA) or Sigma-Aldrich (St. Louis, MO) unless otherwise specified. Mineral ores containing lanthanides were purchased from vendors via [Ebay.com](https://www.ebay.com). Magnets were removed from their protective stainless-steel cases prior to processing. Lanthanide-containing ores and magnets were powdered for subsequent experiments. NdFeB magnet swarf was provided by Urban Mining Company (Austin, TX).

Bacterial Strains and Growth Conditions. All growth studies were performed using the $\Delta mxaF$ strain of *M. extorquens* AM1. The *mxoF* gene encodes the large subunit of the calcium-dependent methanol dehydrogenase, MxaFI. Deletion of this gene ensures REE dependency for growth with methanol via primary oxidation catalyzed by alternative REE-dependent alcohol dehydrogenases.⁴¹ For routine culturing of *M. extorquens* AM1, strains were grown in 16 × 125 mM borosilicate culture tubes shaking at 200–220 rpm, 29–30 °C in modified *Methylobacterium* PIPES (MP) or Hypho minimal medium containing 2–20 μM calcium chloride,^{41,42} with 1.75 g/L succinate as a sole carbon source. Except when indicated otherwise, a modified formulation of Hypho (Hypho^{MOD}) containing 1.27 g/L of K₂HPO₄ and 1.30 g/L of NaH₂PO₄·H₂O was used. For growth measurements with solid ores and minerals, cells were grown in 16 × 125 or 25 × 150 mm borosilicate culture tubes using an Excella E25 shaking incubator (New Brunswick Scientific, Edison, NJ) at 180 rpm.

Growth Analysis with Organic Acids. *M. extorquens* AM1 $\Delta mxaF$ was grown overnight in 3 mL of MP medium with 1.77 g/L succinate as a sole carbon source at 30 °C and 220 rpm in an I 24 shaking incubator (New Brunswick Scientific, Edison, NJ). Cells were harvested via centrifugation at 6,900g for 10 min (Centrifuge S420; Eppendorf, Hamburg, Germany), then washed with 1 volume of fresh, Hypho^{MOD} medium. The resulting cell pellet was resuspended in 0.1

volume of fresh medium, then used to inoculate 1 mL cultures in 24-well suspension culture plates (Greiner Bio-One, Monroe, NC) at a final dilution rate of 1:10. Control cultures were incubated at 30 °C and 567 rpm in a microplate reader (BioTek Synergy H1; Agilent, Santa Clara, CA), and growth was monitored continuously by measuring optical density at 600 nm (OD_{600}). Cultures with magnet swarf were incubated at 30 °C and 500 rpm in a Thermoshaker PHMP-4 (Grant Instruments, Royston, U.K.), and periodically removed to measure OD_{600} in the microplate reader. These cultures were passed over a permanent magnet to clear the well center of magnet swarf particles prior to measurements. To investigate the effect of organic acids on methylotrophic bioleaching and bioaccumulation with 1% (w/v) Nd magnet swarf and 1.6 g/L methanol as sole sources of REE and carbon, respectively, cultures were grown with and without 5 mM or 15 mM citrate, 55 mM gluconate, or 5 mM oxalate. These organic acids were provided as trisodium citrate dihydrate, sodium gluconate, and oxalic acid. The cells were harvested at the early stationary phase by transferring the culture into a 1.5 mL microcentrifuge tube to separate from the remaining magnet swarf. The cultures were then centrifuged, and cell pellet and spent medium fractions were separately stored at -20 °C. Biological triplicates were analyzed for every condition.

Bioreactor Cultivation. Precultures grown in Hypho^{MOD} medium with succinate were prepared as mentioned and then subcultured into 50 mL of Hypho^{MOD} medium with 1.75 g/L succinate and 1.6 g/L methanol in 250 mL shake flasks. Cultures were incubated with shaking at 200 rpm, 30 °C in an Innova S44i shaker (Eppendorf, Hamburg, Germany) for 16 h. 30 mL of culture was then transferred to a 2 L B1One 1250 bioreactor (Distek, North Brunswick, New Jersey) with 0.75 L of Hypho^{MOD} medium and 1% (w/v) NdFeB magnet swarf. For 10 L bioreactor cultivations, 500 mL of overnight cultures was transferred to 9.5 L of Hypho^{MOD} medium. Bioreactor parameters were as follows: agitation, 500 rpm; air flow, 200–2000 sccm; temperature, 29.5 °C; and pH, 6.9. The pH of the bioreactor was maintained by using 1 M NaOH.

Metal Quantification. Samples for the determination of Nd content were prepared as follows: 10 mL of culture was removed from the bioreactor, and cells were pelleted by centrifugation at 4121 × *g* in a Multifuge X Pro Series centrifuge (Thermo Fisher Scientific, Waltham, MA) for 12 min. The culture supernatant was decanted, and cell pellets were washed four times with double-distilled water before drying at 65 °C. After achieving complete dryness, dry weight was measured, and then cells were deconstructed in 20% metal-grade nitric acid at 90 °C and diluted to 2.3% acid before analysis by inductively coupled plasma mass spectrometry (ICP-MS) at the Laboratory for Environmental Analysis (Center of Applied Isotope Studies, University of Georgia). Values were determined by ICP-MS and normalized per unit dry weight as recommended for comparison of bioaccumulation across samples and methods.⁴³

Cell pellets obtained from growth in 24-well plates were acidified in 35% (v/v) HNO₃ for 2 h at 95 °C with mixing by inversion every 30 min. Prior to dilution, the REE concentrations in spent medium and acidified cell samples were quantified by the Arsenazo III assay, as described previously.⁴⁴ Briefly, 40 μL of sample was combined with 40 μL of 12.5% (w/v) trichloroacetic acid (TCA), then added with 120 μL of filtered 0.1% (w/v) Arsenazo in 6.25% (w/v) TCA. Absorbance at 652 nm was measured, and REE

concentration was calculated based on a calibration curve with Nd. For metal content analysis using ICP-MS, digested cell suspensions and spent medium samples were diluted with Milli-Q water or 4% (v/v) HNO₃, respectively, to achieve a final HNO₃ concentration of 2%. The amount of metal associated with cells was normalized to biomass using a predetermined factor of 0.422 ± 0.018 mg dry weight mL⁻¹ OD_{600}^{-1} (*n* = 5).

Statistical Analysis. Data were analyzed using one-way analysis of variance (ANOVA) followed by Tukey's honestly significant difference (HSD) test to determine significant differences between conditions. Pairwise comparisons were performed by using the Student's *t* test with Bonferroni correction. Maximum growth rates in 24-well cultures were determined via linear fitting of natural log transformed OD_{600} using the R package growth rates.⁴⁵

DNA Manipulation, Molecular Cloning, and Mutagenesis. To increase REE bioaccumulation in polyphosphate granules, the gene *ppx*, encoding the exopolyphosphatase enzyme that catalyzes polyphosphate degradation, was deleted. A genetic system for overproducing the lanthanophore was constructed and transformed into *M. extorquens* AM1 with the purpose of increasing nonacidic bioleaching of REEs. Detailed methods for the construction of the *ppx* deletion mutant and for pAZ1, for the overproduction of the lanthanophore, can be found in the Supporting Information (Supplemental Methods and Figure S9).

RESULTS AND DISCUSSION

Scalable REE Bioleaching and Bioaccumulation from Electronic Waste. *M. extorquens* growth performance with 1% (m/v) Nd magnet swarf was benchmarked in a 0.75 L benchtop bioreactor with 1.6 g/L methanol (Figure 2A). Next, we investigated the impact of inorganic phosphate on growth performance because REEs are poorly soluble as phosphate compounds,⁴⁶ the formation of which could hinder REE-dependent methanol growth in media with high phosphate concentrations. Comparative growth analysis showed that lowering inorganic phosphate concentrations in the growth medium resulted in higher growth rates (Figures 2B and S1) but limited growth yields if reduced too much (Figure S1). Therefore, we used Hypho^{MOD} medium containing half the standard phosphate concentration (see Methods) and 1% (m/v) Nd magnet swarf for all growth experiments going forward. *M. extorquens* AM1 cultures grown in Hypho^{MOD} in a 0.75 L benchtop bioreactor generated growth rates that were significantly increased compared to microplate (+2.3-fold) (Figure 2B,D) and shake flask cultures (+1.5-fold) (Figure 2C,D). In the bioreactor, cultures reached maximum densities in 25 h, representing a 2.2-fold decrease in growth period relative to culturing in microplates and a 0.5-fold reduction compared to growth in shake flasks (Table 1). At higher pulp densities, the leaching of REEs and/or other metals, such as iron, could negatively impact growth performance and bioaccumulation. However, when the magnet swarf was increased to 10% (m/v) in the same bioreactor setup, growth rates (Figure 2E) and cycle times (Table 1) were similar to what was observed at the lower pulp density (Figure 2D), indicating that growth was not inhibited.

M. extorquens AM1 stores REE as intracellular polyphosphate granules.³⁴ Baseline Nd bioaccumulation levels were measured from bioreactor cultures grown in standard Hypho minimal medium (Table 1). Relative to the baseline

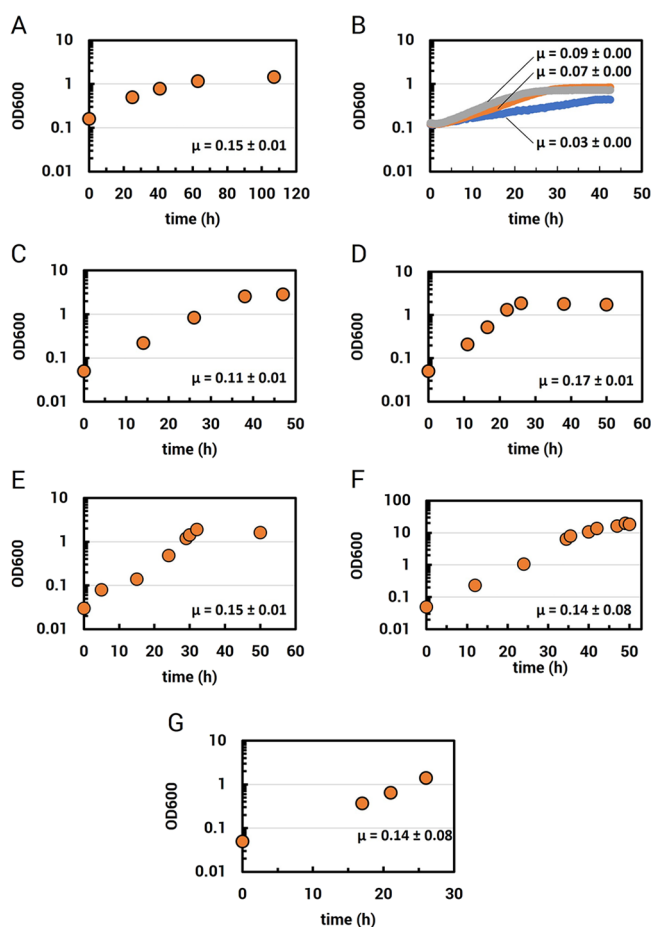


Figure 2. *M. extorquens* growth with Nd magnet swarf and methanol. (A) 0.75 L bioreactor growth in full inorganic phosphate Hypho minimal medium (see Methods) and 1% (w/v) Nd magnet swarf. (B) Microplate growth with full phosphate (blue), half phosphate (Hypho^{MOD}; orange), and quarter phosphate (gray) medium and 1% (w/v) magnet swarf. (C) 50 mL shake flask culture growth with Hypho^{MOD} and 1% (w/v) magnet swarf. (D) 0.75 L bioreactor growth with Hypho^{MOD} medium and 1% (w/v) magnet swarf. (E) 0.75 L bioreactor growth with Hypho^{MOD} medium and 10% magnet swarf. (F) 0.75 L bioreactor growth with 1% (w/v) magnet swarf. (G) 10 L bioreactor growth with Hypho^{MOD} medium and 1% magnet swarf. Growth curves are representative of each condition (bioreactors $n = 2$, microplates $n = 6$, flasks $n = 3$). Growth rates are the mean rates for each condition, with standard deviations.

bioaccumulation level, *M. extorquens* AM1 grown in Hypho^{MOD} accumulated 3.6-fold more Nd when grown with 1% pulp density (Table 1). Reducing inorganic phosphate, therefore, not only boosted growth performance in the bioreactor (Figure 2A,D), but also positively affected REE bioaccumulation, underscoring the link between phosphate metabolism and REE use in microbial systems. Removal of calcium from the growth medium did not significantly impact REE bioaccumulation ($p > 0.05$ by ANOVA and Tukey's HSD) (Figure S2). Even at 10% pulp density, growth rate REE bioaccumulation by *M. extorquens* AM1 was still high (Figure 2E and Table 1) showing that the process is not inhibited by a 10-fold increase in REE feedstock.

Next, we tested the performance of *M. extorquens* AM1 under methanol fed-batch conditions in a 0.75 L bioreactor. By repeated feeding of methanol over the run cycle, culture densities of 20 OD were achieved without a significant

Table 1. Growth Performance and Bioaccumulation of Nd by *M. extorquens* AM1 during Bioreactor Cultivation with a Nd Magnet Swarf^a

growth setup ^c	growth cycle (h)	bioaccumulation (mg Nd/g DW ^b)	leachate (mg/L)
bioreactor (baseline)	63	3.2 ± 0.1	n.m.
microplate	28–54	n.m.	n.m.
flask	38	n.m.	n.m.
bioreactor	25	11.5 ± 0.3	<0.001
bioreactor (10% pulp)	30	10.8 ± 0.2	<0.001
bioreactor (fed batch)	50	11.0 ± 0.4	<0.001
bioreactor (10 L)	26	13.2 ± 2.9	<0.001

^an.m. is not measured. ^bNd content of cell samples were measured by ICP-MS and normalized to dry weight. Means and standard deviations of 3 independent measurements. ^cBioreactor measurements are from two independent replicates with 1% Nd magnet swarf pulp density and 0.75 L volume, unless indicated otherwise. All bioreactors were run with Hypho^{MOD} medium except baseline. 10% pulp indicates increase of swarf pulp density to 10%. Fed batch indicates methanol growth substrate was refed until maximum optical density of the culture was reached. 10 L indicates a scale increase to a 10 L bioreactor system.

reduction in growth rate (Figure 2F). In total, ~17.3 g/L of methanol was fed over a cycle time of ~50 h. Again, there was no detectable residual Nd found in the leachate, and the REE to biomass ratio was the same as that in batch cultures (Table 1). Finally, we assessed process performance at a 10 L scale using optimized media and 1% Nd magnet swarf pulp density. The growth cycle and growth rate were like the 0.75 L scale, and Nd yields increased by nearly 15 ± 17%, but the differences in each of these parameters between the 0.75 and 10.0 L scales were not statistically significant (Student's *t* test with Bonferroni correction, $p_{\text{Bonf}} > 0.05$). Together, these results show the scalability of this process without loss of efficiency in REE recovery from E-waste.

Selective Bioaccumulation of REE from Nd Magnet Swarf. Even high-grade feedstocks like Nd magnet swarf are of mixed composition and therefore vary in the content of highly valuable REE and associated metals,⁴⁷ making selectivity crucial for the development of a successful recovery platform. We determined the precise metal content of the Nd magnet swarf used in our growth and recovery experiments using ICP-MS. Iron was the primary metal component of Nd magnet swarf (68.0% Fe), rendering nonselective leaching and uptake mechanisms insufficient for effective REE bioaccumulation. Nd was the second most abundant metal measured (26.7% Nd). In addition to Nd, the magnet swarf contained significant amounts of the light Ln praseodymium (Pr, 4.35%) and the heavy Ln dysprosium (Dy, 3.34%), both of which have high technological value.^{48,49}

To assess the selectivity of *M. extorquens* AM1 uptake for REE, we determined the metal content of the abiotic leachate and cells and supernatants of cultures grown with 1% (w/v) magnet swarf. Approximately 60% of the metal leached by uninoculated Hypho^{MOD} medium consisted of REE (Figure 3A, abiotic control). In cultures grown under the same conditions, most of the available REE was taken up by the cell, leaving the supernatant with 98% Fe (Figure 3A). Next, we assessed the specificity of REE bioaccumulation in batch bioreactor cultures with 1% magnet swarf (w/v). Cultures were

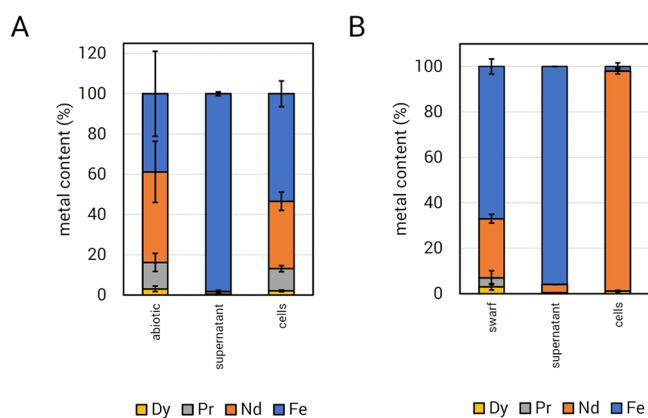


Figure 3. Selective bioconcentration of REE. *M. extorquens* AM1 was grown in Hypho^{MOD} methanol medium with a 1% Nd magnet swarf. Cell and supernatant samples were taken once the culture reached the early stationary growth phase. (A) Metal content from 1 mL cultures. Abiotic data represents leachates of inoculated growth medium incubated simultaneously as cultures. (B) Metal content from 0.75 L benchtop bioreactor cultures. Swarf data shows the composition of the REE source material used. Cell and supernatant samples were taken once the culture reached the early stationary growth phase. Plots reflect the mean metal contents of three independent growth experiments, with error bars showing standard deviations. Dy, dysprosium; Pr, praseodymium; Nd, neodymium; Fe, iron.

grown to maximum culture density of 1.8 OD (600 nm) with 1.6 g/L methanol, after which Fe, Nd, Pr, and Dy content in cells and supernatants were measured. Supernatant metal content consisted of mostly Fe (95.9%) with low amounts of REE (Nd, 3.6%; Pr, < 1%; Dy, < 1%) (Figure 4). In cell

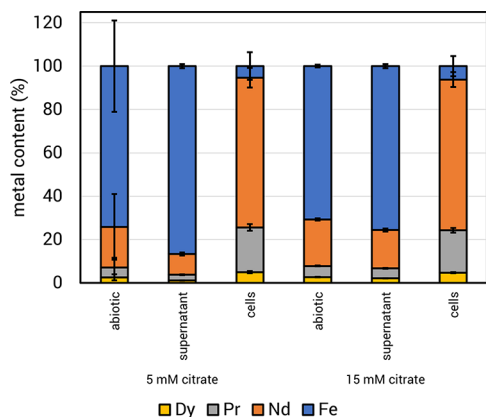


Figure 4. Selective bioconcentration of REE. *M. extorquens* AM1 was grown in 1 mL of Hypho^{MOD} methanol medium with 1% Nd magnet swarf and 0, 5, or 15 mM citrate. Cell and supernatant samples were taken once the culture reached the early stationary growth phase. Cell and supernatant samples were taken once the culture reached the early stationary growth phase. Plots reflect the mean metal contents of three independent growth experiments, with error bars showing standard deviations. Dy, dysprosium; Pr, praseodymium; Nd, neodymium; and Fe, iron.

samples, 98.0% of the accumulated metal was REE, 96.8% of which was Nd, while Pr and Dy made up a combined 1.2% (Figure 3B). This was ~1.6-fold higher than what was observed in the 1 mL microplate cultures as mentioned above and could possibly be accounted for by more efficient mixing of swarf with *M. extorquens* AM1 in a stirred-tank

system. Fe accounted for only 2.0% of the measured intracellular metal content (Figure 3B). Given that cells acquire Fe as a micronutrient, it is difficult to differentiate Fe from the growth medium from that obtained from the magnet swarf, but it can be concluded that the latter is less than 2.0%. Together, these data show preferential REE uptake and bioaccumulation in the bioreactor.

Chemical Strategies for Increased REE Leaching from Nd Magnet Swarf. Only trace amounts of REE were detected after process runs in bioreactor culture supernatants, indicating that nearly all leached REE was scavenged (Table 1). REE leaching, therefore, could be a limiting factor for bioaccumulation. Microbially produced organic acids are an environmentally and financially attractive alternative to traditional hydrometallurgical methods.^{50–52} Organic acids derived from heterotrophic microbial metabolism have been successfully employed as biolixiviants for metal dissolution, including REEs.^{18,19,53–62}

Adding citric acid to Hypho^{MOD} medium with a magnet swarf increased the level of chemical leaching of all metals (Table S1 and Figure S3). After 28 h of incubation, 8.2 ± 0.1 and 58.7 ± 0.6 ppm of REE were leached into the supernatant supplemented with 5 mM and 15 mM citrate, respectively, which were 15- and 108-fold greater than that found in the supernatant with no chelator (Figure S3). However, Fe was preferentially leached to concentrations of 23.7 ± 2.5 and 142.0 ± 3.2 ppm in the presence of citrate, i.e., there was a 67- and 404-fold increase in Fe concentrations under these conditions (Table S1). The increase in leached Dy with the addition of citrate was twice that of Nd or Pr (Table S1), which may be attributable to its higher complex formation constant with citrate ($\log K = 10.69$ vs 9.76 – 9.94).⁶³ Addition of 55 mM gluconate or 5 mM oxalate, the highest permissible concentration of each for comparable growth of *M. extorquens*, did not increase REE leaching as compared to the negative control condition, though Fe concentrations did increase by 12- and 4.2-fold, respectively (Table S1). Gluconic acid was previously shown to promote leaching of REEs from solid feedstocks, though the pH conditions were more acidic compared to the buffered pH of Hypho^{MOD} medium (<pH 3.3 vs pH 6.9).¹⁹

We investigated whether nonspecific leaching by the biolixiviant citrate would alter the specificity of REE uptake by *M. extorquens* AM1. When 5 or 15 mM citrate was added to the medium, there was a ~2-fold increase in cellular REE content, from 47 to >93% (Student's *t* test with Bonferroni correction; $p_{\text{Bonf}} = 0.001$ and 0.007 , respectively), consisting of 69% Nd, 20% Pr, and 5% Dy. The REE content in the culture supernatant also increased from 2% to 13% and 24%, respectively (Figure 4). *M. extorquens* AM1 ΔmxaF grew well with 1% (w/v) magnet swarf as the sole source of REE, with a significantly higher growth rate as compared to that grown with soluble La (Figures 5A and S4; ANOVA with Tukey HSD, $p_{\text{Tukey}} = 5.4 \times 10^{-9}$).

Addition of citrate to the growth medium with magnet swarf did not significantly affect growth ($p_{\text{Tukey}} \geq 0.98$). However, addition of gluconate reduced the growth rate as compared to that of citrate ($p_{\text{Tukey}} \leq 0.042$), and oxalate further negatively impacted growth ($p_{\text{Tukey}} \leq 1.6 \times 10^{-10}$). As expected from the abiotic leaching results (Figure S3), addition of citrate greatly improved REE leaching (Figures 5B and S5A; $p_{\text{Bonf}} \leq 0.001$), and led to greater REE uptake by *M. extorquens* AM1 ΔmxaF (Figures 5C and S5B; $p_{\text{Bonf}} \leq 0.001$). Interestingly, more REE

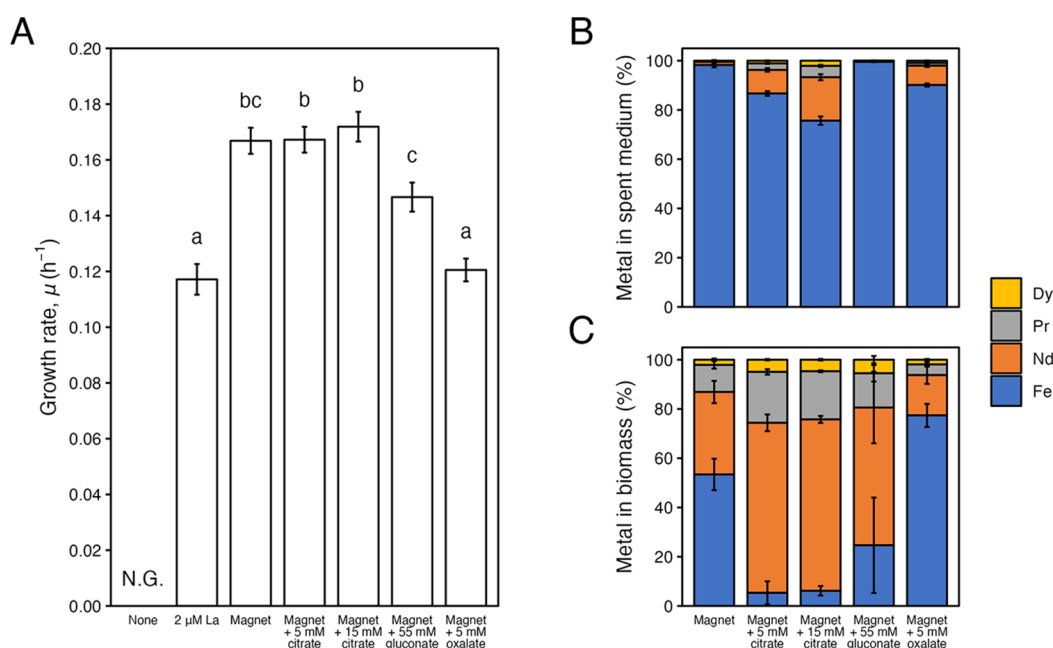


Figure 5. Growth of *M. extorquens* AM1 $\Delta mxaF$ in Hypho^{MOD} medium with 1.6 g/L MeOH and 2 μM La, 1% (w/v) magnet swarf, or 1% (w/v) magnet swarf supplemented with citrate (5 or 15 mM), gluconate (55 mM), or oxalate (5 mM). (A) Growth rate of *M. extorquens* AM1 $\Delta mxaF$, (B) metals leached into the spent medium, and (C) metals associated with biomass were determined based on biological triplicates. Bars and error bars represent model estimates and the standard error of growth rates, or the mean and standard deviation of metal content. The letters above bars represent statistical significance as determined by ANOVA followed by the Tukey HSD test ($p < 0.05$). N.G. represents no growth.

was associated with biomass when 5 mM, rather than 15 mM, citrate was supplemented, suggesting that there exists a balance between abiotic leaching and biological uptake at the scale tested (Figure S5B; $p_{\text{Tukey}} \leq 0.04$). In fact, upon scale-up of the culture volume to 100 mL, 15 mM citrate inhibited growth of *M. extorquens* AM1 $\Delta mxaF$ (Figure S6). Importantly, unlike the increased uptake of REE in the presence of citrate, the level of Fe associated with the cell remained constant across all conditions (Figure S5B; $p_{\text{Tukey}} \geq 0.34$).

Next, we assessed how adding 5 mM citrate impacts batch bioreactor cultures and found that total REE bioaccumulation increased to 98.8% of metal uptake with Nd accounting for 88.9%, Pr 9.2%, and Dy 0.7% (Figure S7). Supernatant REE concentrations were notably higher with the addition of citrate to the growth medium. Nd accounted for 24.5%, Pr 2.3%, and Dy 7.9%, with Fe accounting for only 65.2% (Figure S7). Nd and Fe concentrations in citrate-supplemented supernatants were similar to those of the original swarf composition, but Pr and Dy concentrations were notably different. Pr in the supernatant was ~ 2 -fold lower and Dy ~ 2 -fold higher in supernatant, indicating preferential uptake of the light REE and/or preferential leaching of the heavy REE. Overall, these data show highly selective, ~ 3.6 -fold bioconcentration of REE from magnet swarf by *M. extorquens* AM1. Based on these highly promising results, *M. extorquens* AM1 appears naturally poised as a competitive REE bioaccumulation microbe with scalable growth.

Current REE bioaccumulation yields do not account for 100% of the REE in the magnet swarf. We tested whether a single magnet swarf batch could be processed for further REE extraction in both small- and large-scale cultivations. In 1 mL cultures, the same batch of magnet swarf was able to sustain growth of *M. extorquens* AM1 $\Delta mxaF$ for five cycles, though the final OD and total leached REE concentrations decreased with continuing growth cycles (Figure S7). Based on the

bioaccumulation yields from the first bioreactor run (Figure 6), we estimated $8.25 \pm 0.04\%$ of the swarf Nd could be recovered

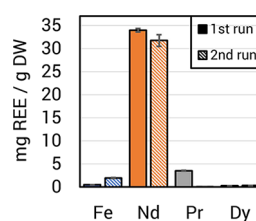


Figure 6. Total REE bioaccumulation is increased by reusing magnet swarf feedstock. Each experiment consisted of two complete process runs in which the $\Delta mxaF$ strain was grown with 1% magnet swarf (w/v) in a 0.75 L bioreactor with 1.6 g/L methanol and 5 mM citrate. After growth, cell samples were taken for ICP-MS metal content measurements, and the magnet swarf was removed from the bioreactor, rinsed 4 times with ddH₂O, and dried at 65 °C for 72 h before reuse. Bars show the means of cellular REE content normalized to dry weight (DW) for two replicate experiments. A distinct swarf batch was used for each individual experiment. Error bars indicate standard deviations.

in a single fed-batch process, reaching a culture OD of 20 when 5 mM citrate is added to increase leaching. With a second consecutive process run, using the same swarf batch as that of the first run, bioaccumulation of Nd accounted for another $7.75 \pm 0.12\%$ of the original amount (Figure 6). These promising results suggest that a continuous process run could feasibly be used to recover an even higher proportion of REE in a swarf batch.

A recent techno-economic analysis reported the potential economic feasibility of using biolixivants to increase REE leaching from waste materials and found that the carbon substrate (glucose) is a major contributor to cost.²⁰ Our results showed that addition of low molarity citrate significantly

increased REE leaching and bioaccumulation from the Nd magnet swarf. Citrate can be added directly to the bioreactor, or in the future, process development could be linked via coculture with a heterotrophic acid-producing microbe, such as *Glucanobacter oxidans*. However, to truly limit the costs of an REE recovery process, an acid-independent approach would be preferred, especially in areas in which secondary hazardous waste streams are of concern.

Approaches for Enhanced REE-Specific Bioleaching and Bioaccumulation. The genetic tractability and availability of genetic tools for strain engineering make *M. extorquens* AM1 a uniquely suitable microbial platform for REE recovery. A biosynthetic pathway for an REE-chelator (lanthanophore) has been previously reported and was named *mll* after its product methylolanthanin.⁶⁴ Expression of the *mll* has already been shown to increase Nd uptake in *M. extorquens* AM1 from both soluble chloride and poorly soluble oxide forms in a highly specific manner.⁶⁴ We tested the impact of expressing the *mll* in trans during growth with magnet swarfs and saw increases in REE bioaccumulation. Nd, Pr, and Dy accumulation increased by more than 3-fold with *mll* expression, reaching 80 mg of Nd/g of DW, 15 mg of Pr/g of DW, and 8 mg of Dy/g of DW (Figure 7A), while Fe

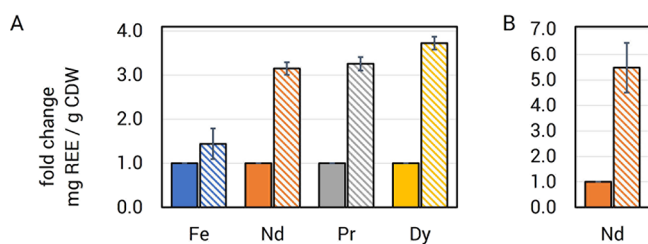


Figure 7. Strain engineering for increased REE bioaccumulation. (A), in trans expression of the *mll* enhances REE (Nd, neodymium; Pr, praseodymium; Dy, dysprosium) bioaccumulation from magnet swarf in *M. extorquens* AM1, with minimal Fe uptake. $\Delta mxaF$, solid bars; $\Delta mxaF/pAZ1$, hatched bars. A detailed description of pAZ1 can be found in the Supporting Information (Supplemental Methods and Figure S9). (B), deletion of the exopolyphosphatase-encoding *ppx* gene results in higher Nd bioaccumulation. $\Delta mxaF$, solid bars; Δppx , hatched bars. For A and B, the metal content was determined by ICP-MS and normalized to cell dry weight. Values represent the mean of samples from 3 independent cultures grown in shake flasks. Error bars reflect standard deviations.

accumulation increased only marginally. Notably, expression of the *mll* was under control of the *lac* promoter, a comparatively weak promoter in *M. extorquens* AM1.⁶⁵ Generation of a stronger expression system is underway and in theory should boost REE bioaccumulation even further.

Pyroloquinoline quinone (PQQ) has been shown to directly bind REEs⁶⁶ and has been implicated in microbial REE solubilization.⁶² We recently reported *M. extorquens* AM1 strain evo-HLn⁴⁰ that exhibited enhanced PQQ production in the presence of the heavy REE, gadolinium. We measured Nd bioaccumulation in evo-HLn and saw a 53% increase compared to $\Delta mxaF$.

REEs are stored in intracellular polyphosphate granules in *M. extorquens* AM1. Depolymerization of polyphosphate is catalyzed by exopolyphosphatase activity. We hypothesized that limiting the cellular capacity for polyphosphate depolymerization could generate higher levels of REE bioaccumulation. A *ppx* (encoding exopolyphosphatase)

deletion strain was generated and assessed for REE bioaccumulation, showing a ~ 5.5 -fold increase in Nd levels reaching 202 mg Nd/g DW (Figure 7B).

The results shown herein demonstrate that selective targeting of specific processes for REE metabolism, generated by a detailed understanding of the physiology of *M. extorquens* AM1, is an effective strategy for engineering enhanced REE storage. Reduction of inorganic phosphate in the growth medium, likely allowing for increased uptake and granulation with REE, resulted in a nearly 4-fold enhancement of Nd yields over baseline levels (Figure 8). Identification of REE-binding

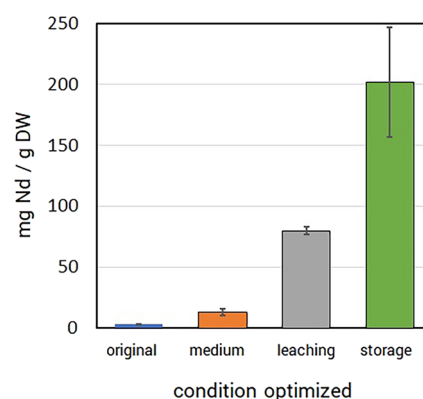


Figure 8. Process performance enhancement. Chemical and genetic targeting of specific metabolic processes generates improved REE bioaccumulation relative to the original (baseline) measurements. Manipulations to growth medium inorganic phosphate (medium), lanthanophore production (leaching), and REE/phosphate granulation (storage) each generated elevated REE bioaccumulation yields. Bioaccumulation was measured by ICP-MS and is normalized to dry weight. Error bars represent standard deviations of biological triplicates.

ligands like methylolanthanin allows for nonacidic, highly specific leaching from feedstocks. Genetic engineering to increase methylolanthanin levels produced Nd yields more than 20-fold higher than baseline values (Figure 7). Likewise, detailed knowledge of REE storage granulation and storage processes has allowed for genetic manipulation and engineering to further boost intracellular REE content. By engineering reduced capacity for granule deconstruction, we were able to increase the Nd storage capacity over 50-fold (Figure 7) relative to the baseline condition with $\Delta mxaF$. These improved yields, combined with a culture OD of 20 in a 0.75 L bioreactor, would allow Nd recovery of 1.3–2.1 g of Nd/L, representing a recovery between 65 and 100% of Nd in a single process run when using 1% Nd magnet swarf pulp density.

ENVIRONMENTAL IMPLICATIONS

Current REE mining and refinement practices are costly, environmentally destructive, and unsustainable.⁶⁷ Safer, cleaner alternatives are needed. Microbes have emerged as a promising biological alternative to harsh chemical extraction methods for leaching REE from complex sources and waste streams, opening new avenues for processing feedstocks and recycling these critical metals. *M. extorquens* AM1, a bacterium with the natural ability to solubilize and accumulate REE, provides a unique pathway toward selective REE bioleaching, bioaccumulation, and recovery that is scalable and does not rely on extremophile organisms and their inherent limitations. Future

work will investigate this further with a focus on end-to-end conversion of feedstock to scalable rare earth oxide products.

We have reported optimization of several parameters to increase REE recovery yields in a 10 L bioreactor format, reaching yields of over 1.3% dry weight with the potential for up to 20% dry weight with further strain engineering. Our yields rival current state-of-the-art technologies when benchmarked against conditions when REE leaching is not limited (with citrate or with *mll* overexpression),⁴³ and our process adds the major advantage of comparatively high REE uptake and storage from solid feedstocks. The potential of *M. extorquens* AM1 to recover REE is not limited to E-waste, as our process is compatible with even the most unrefined sources, such as monazite and bastnäsite ores, and pulverized smartphones (Table S2). *M. extorquens* AM1 is uniquely suited biologically for REE recovery through its production of lanthanophores, dedicated REE transport machinery, and its capability for producing REE-phosphate granules that will facilitate purification. The potential reduction in REE mining and refinement pollution and hazardous exposure make *M. extorquens* AM1 an attractive, green alternative to current chemical methodologies.

■ ASSOCIATED CONTENT

SI Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.est.3c06775>.

Supplemental methods; impact of inorganic phosphate on strain performance; effect of calcium in growth medium on Nd uptake; abiotic leaching of REE; growth rates of *M. extorquens* AM1 with organic acid addition; bioleaching and bioaccumulation measurements after growth with organic acids; effect of citrate on bioleaching and bioaccumulation; selective bioconcentration of REE; iterative bioaccumulation from a single swarf batch; plasmid map of pAZ1 for *mll* expression in trans; ratio of metals chemically leached from magnet swarf by organic acids; growth of *M. extorquens* AM1 with pure and complex REE sources; and flexibility of *M. extorquens* for REE bioleaching and bioaccumulation from complex sources (PDF)

■ AUTHOR INFORMATION

Corresponding Author

Norma Cecilia Martinez-Gomez – Department of Plant and Microbial Biology, University of California, Berkeley, Berkeley, California 94720, United States; orcid.org/0000-0002-4071-7204; Email: cecimartinez@berkeley.edu

Authors

Nathan M. Good – Department of Plant and Microbial Biology, University of California, Berkeley, Berkeley, California 94720, United States; orcid.org/0000-0002-9490-5913
Christina S. Kang-Yun – Physical and Life Sciences Directorate, Lawrence Livermore National Laboratory, Livermore, California 94550, United States; orcid.org/0000-0001-5536-3253
Morgan Z. Su – Department of Plant and Microbial Biology, University of California, Berkeley, Berkeley, California 94720, United States

Alexa M. Zytneck – Department of Plant and Microbial Biology, University of California, Berkeley, Berkeley, California 94720, United States

Colin C. Barber – Department of Plant and Microbial Biology, University of California, Berkeley, Berkeley, California 94720, United States

Huong N. Vu – Department of Biological Sciences, San José State University, San José, California 95192, United States

Joseph M. Grace – Department of Biological Sciences, San José State University, San José, California 95192, United States

Hoang H. Nguyen – Department of Biological Sciences, San José State University, San José, California 95192, United States

Wenjun Zhang – Department of Chemical and Biomolecular Engineering, University of California, Berkeley, Berkeley, California 94720, United States; orcid.org/0000-0001-9348-972X

Elizabeth Skovran – Department of Biological Sciences, San José State University, San José, California 95192, United States; orcid.org/0000-0002-1737-5621

Maohong Fan – Department of Chemical and Biomedical Engineering, University of Wyoming, Laramie, Wyoming 82071, United States; orcid.org/0000-0003-1334-7292

Dan M. Park – Physical and Life Sciences Directorate, Lawrence Livermore National Laboratory, Livermore, California 94550, United States

Complete contact information is available at:

<https://pubs.acs.org/doi/10.1021/acs.est.3c06775>

Author Contributions

The manuscript was written through the contributions of all authors. All authors have approved the final version of the manuscript.

Funding

The information, data, or work presented herein was funded in part by the Advanced Research Projects Agency-Energy, United States Department of Energy, under Award Number DE-AR0001337.

Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

We would like to thank Yi Yao and Samir Budhathoki from the University of Wyoming for their assistance with some ICP-MS sample analysis, and Bang Luong, Caitlin Hoerber, Clarisse Hufana, Deval Patel, James Cai, Kim Luu, Krishna Gupta, Mariam Jacob, Nicholas Lien, Alice Huang, and Jennifer Lepe-Rodriguez for assistance with CFU measurements for smartphone growth curves. Part of this work was carried out under the auspices of the DOE by Lawrence Livermore National Laboratory under contract DEAC52-07NA27344 (LLNL-JRNL-851205).

■ REFERENCES

- (1) Summaries. United States Geological Survey: 2016; p 202. <https://d9-wret.s3.us-west-2.amazonaws.com/assets/palladium/production/mineral-pubs/mcs/mcs2016.pdf>.
- (2) Alonso, E.; Sherman, A. M.; Wallington, T. J.; Everson, M. P.; Field, F. R.; Roth, R.; Kirchain, R. E. Evaluating Rare Earth Element Availability: A Case with Revolutionary Demand from Clean Technologies. *Environ. Sci. Technol.* **2012**, *46* (6), 3406–3414.
- (3) Kingsnorth. *Meeting the Challenges of Rare Earths Supply in the next Decade*; Industrial Minerals Company of Australia Pty Ltd.: 2010.

- (4) Binnemans, K.; Jones, P. T.; Blanpain, B.; Van Gerven, T.; Yang, Y.; Walton, A.; Buchert, M. Recycling of Rare Earths: A Critical Review. *J. Clean. Prod.* **2013**, *51*, 1–22.
- (5) Graedel, T. E.; Harper, E. M.; Nassar, N. T.; Reck, B. K. On the Materials Basis of Modern Society. *Proc. Natl. Acad. Sci. U. S. A.* **2015**, *112* (20), 6295–6300.
- (6) Doe, U. S. Department of Energy. *Critical Materials Strategy* (2010); Department of Energy (DOE): Washington, United States of America: 2011.
- (7) Mancheri, N. A.; Sprecher, B.; Bailey, G.; Ge, J.; Tukker, A. Effect of Chinese Policies on Rare Earth Supply Chain Resilience. *Resour. Conserv. Recycl.* **2019**, *142*, 101–112.
- (8) Zhang, S. Study on Economic Significance of Rare Earth Mineral Resources Development Based on Goal Programming and Few-Shot Learning. *Comput. Intell. Neurosci.* **2022**, 2022, No. 7002249.
- (9) Office of Research; Development. *Rare Earth Elements: A Review of Production, Processing, Recycling, and Associated Environmental Issues*. 2013.
- (10) Balam, V. Rare Earth Elements: A Review of Applications, Occurrence, Exploration, Analysis, Recycling, and Environmental Impact. *Geoscience Frontiers* **2019**, *10* (4), 1285–1303.
- (11) Talan, D.; Huang, Q. A Review of Environmental Aspect of Rare Earth Element Extraction Processes and Solution Purification Techniques. *Miner. Eng.* **2022**, *179*, No. 107430.
- (12) Sahoo, P. K.; Kim, K.; Powell, M. A.; Equeenuddin, S. Recovery of Metals and Other Beneficial Products from Coal Fly Ash: A Sustainable Approach for Fly Ash Management. *International Journal of Coal Science & Technology* **2016**, *3* (3), 267–283.
- (13) Kumari, A.; Panda, R.; Jha, M. K.; Kumar, J. R.; Lee, J. Y. Process Development to Recover Rare Earth Metals from Monazite Mineral: A Review. *Miner. Eng.* **2015**, *79*, 102–115.
- (14) Liang, T.; Li, K.; Wang, L. State of Rare Earth Elements in Different Environmental Components in Mining Areas of China. *Environ. Monit. Assess.* **2014**, *186* (3), 1499–1513.
- (15) Zhou, L.; Li, Z.; Liu, W.; Liu, S.; Zhang, L.; Zhong, L.; Luo, X.; Liang, H. Restoration of Rare Earth Mine Areas: Organic Amendments and Phytoremediation. *Environ. Sci. Pollut. Res. Int.* **2015**, *22* (21), 17151–17160.
- (16) Khawassek, Y. M.; Eliwa, A. A.; Gawad, E. A.; Abdo, S. M. Recovery of Rare Earth Elements from El-Sela Effluent Solutions. *Journal of Radiation Research and Applied Sciences* **2015**, *8* (4), 583–589.
- (17) Van Gosen, B. S.; Verplanck, P. L.; Seal, R. R., II; Long, K. R.; Gambogi, J. *Rare-Earth Elements*; US Geological Survey: 2017. DOI: 10.3133/pp18020.
- (18) Brisson, V. L.; Zhuang, W.-Q.; Alvarez-Cohen, L. Bioleaching of Rare Earth Elements from Monazite Sand. *Biotechnol. Bioeng.* **2016**, *113* (2), 339–348.
- (19) Reed, D. W.; Fujita, Y.; Daubaras, D. L.; Jiao, Y.; Thompson, V. S. Bioleaching of Rare Earth Elements from Waste Phosphors and Cracking Catalysts. *Hydrometallurgy* **2016**, *166*, 34–40.
- (20) Thompson, V. S.; Gupta, M.; Jin, H.; Vahidi, E.; Yim, M.; Jindra, M. A.; Nguyen, V.; Fujita, Y.; Sutherland, J. W.; Jiao, Y.; Reed, D. W. Techno-Economic and Life Cycle Analysis for Bioleaching Rare-Earth Elements from Waste Materials. *ACS Sustainable Chem. Eng.* **2018**, *6* (2), 1602–1609.
- (21) Jin, H.; Reed, D. W.; Thompson, V. S.; Fujita, Y.; Jiao, Y.; Crain-Zamora, M.; Fisher, J.; Scalzone, K.; Griffel, M.; Hartley, D.; Sutherland, J. W. Sustainable Bioleaching of Rare Earth Elements from Industrial Waste Materials Using Agricultural Wastes. *ACS Sustainable Chem. Eng.* **2019**, *7* (18), 15311–15319.
- (22) Park, D. M.; Brewer, A.; Reed, D. W.; Lammers, L. N.; Jiao, Y. Recovery of Rare Earth Elements from Low-Grade Feedstock Leachates Using Engineered Bacteria. *Environ. Sci. Technol.* **2017**, *51* (22), 13471–13480.
- (23) Cížková, M.; Mezricky, D.; Rucki, M.; Tóth, T. M.; Náhlík, V.; Lanta, V.; Bišová, K.; Zachleder, V.; Vítová, M. Bio-Mining of Lanthanides from Red Mud by Green Microalgae. *Molecules* **2019**, *24* (7), No. 071356.
- (24) Das, N.; Das, D. Recovery of Rare Earth Metals through Biosorption: An Overview. *J. Rare Earths* **2013**, *31* (10), 933–943.
- (25) Moriawaki, H.; Masuda, R.; Yamazaki, Y.; Horiuchi, K.; Miyashita, M.; Kasahara, J.; Tanaka, T.; Yamamoto, H. Application of Freeze-Dried Powders of Genetically Engineered Microbial Strains as Adsorbents for Rare Earth Metal Ions. *ACS Appl. Mater. Interfaces* **2016**, *8* (40), 26524–26531.
- (26) Alipanah, M.; Park, D. M.; Middleton, A.; Dong, Z.; Hsu-Kim, H.; Jiao, Y.; Jin, H. Techno-Economic and Life Cycle Assessments for Sustainable Rare Earth Recovery from Coal Byproducts Using Biosorption. *ACS Sustainable Chem. Eng.* **2020**, *8* (49), 17914–17922.
- (27) Jin, H.; Park, D. M.; Gupta, M.; Brewer, A. W.; Ho, L.; Singer, S. L.; Bourcier, W. L.; Woods, S.; Reed, D. W.; Lammers, L. N.; Sutherland, J. W.; Jiao, Y. Techno-Economic Assessment for Integrating Biosorption into Rare Earth Recovery Process. *ACS Sustainable Chem. Eng.* **2017**, *5* (11), 10148–10155.
- (28) Kiyono, M.; Omura, H.; Omura, T.; Murata, S.; Pan-Hou, H. Removal of Inorganic and Organic Mercurials by Immobilized Bacteria Having *mer-ppk* Fusion Plasmids. *Appl. Microbiol. Biotechnol.* **2003**, *62* (2–3), 274–278.
- (29) Johnston, C. W.; Wyatt, M. A.; Li, X.; Ibrahim, A.; Shuster, J.; Southam, G.; Magarvey, N. A. Gold Biomineralization by a Metallophore from a Gold-Associated Microbe. *Nat. Chem. Biol.* **2013**, *9* (4), 241–243.
- (30) Reith, F.; Etschmann, B.; Grosse, C.; Moors, H.; Benotmane, M. A.; Monsieurs, P.; Grass, G.; Doonan, C.; Vogt, S.; Lai, B.; Martinez-Criado, G.; George, G. N.; Nies, D. H.; Mergeay, M.; Pring, A.; Southam, G.; Brugger, J. Mechanisms of Gold Biomineralization in the Bacterium *Cupriavidus metallidurans*. *Proc. Natl. Acad. Sci. U. S. A.* **2009**, *106* (42), 17757–17762.
- (31) Kwok, R. Inner Workings: How Bacteria Could Help Recycle Electronic Waste. *Proc. Natl. Acad. Sci. U. S. A.* **2019**, *116* (3), 711–713.
- (32) Kulkarni, S.; Misra, C. S.; Gupta, A.; Ballal, A.; Apte, S. K. Interaction of Uranium with Bacterial Cell Surfaces: Inferences from Phosphatase-Mediated Uranium Precipitation. *Appl. Environ. Microbiol.* **2016**, *82* (16), 4965–4974.
- (33) Basnakova, G.; Stephens, E. R.; Thaller, M. C.; Rossolini, G. M.; Macaskie, L. E. The Use of *Escherichia coli* Bearing a *phoN* Gene for the Removal of Uranium and Nickel from Aqueous Flows. *Appl. Microbiol. Biotechnol.* **1998**, *50* (2), 266–272.
- (34) Roszczenko-Jasińska, P.; Vu, H. N.; Subuyuj, G. A.; Crisostomo, R. V.; Cai, J.; Lien, N. F.; Clippard, E. J.; Ayala, E. M.; Ngo, R. T.; Yarza, F.; Wingett, J. P.; Raghuraman, C.; Hoerber, C. A.; Martinez-Gomez, N. C.; Skovran, E. Gene Products and Processes Contributing to Lanthanide Homeostasis and Methanol Metabolism in *Methylorubrum extorquens* AM1. *Sci. Rep.* **2020**, *10* (1), 12663.
- (35) Skovran, E.; Martinez-Gomez, N. C. Microbiology. Just Add Lanthanides. *Science* **2015**, *348* (6237), 862–863.
- (36) Martinez-Gomez, N. C.; Vu, H. N.; Skovran, E. Lanthanide Chemistry: From Coordination in Chemical Complexes Shaping Our Technology to Coordination in Enzymes Shaping Bacterial Metabolism. *Inorg. Chem.* **2016**, *55* (20), 10083–10089.
- (37) Maroufi, S.; Nekouei, R. K.; Hossain, R.; Assefi, M.; Sahajwalla, V. Recovery of Rare Earth (i.e., La, Ce, Nd, and Pr) Oxides from End-of-Life Ni-MH Battery via Thermal Isolation. *ACS Sustainable Chem. Eng.* **2018**, *6* (9), 11811–11818.
- (38) Deshmane, V. G.; Islam, S. Z.; Bhavne, R. R. Selective Recovery of Rare Earth Elements from a Wide Range of E-Waste and Process Scalability of Membrane Solvent Extraction. *Environ. Sci. Technol.* **2020**, *54* (1), 550–558.
- (39) Peelman, S.; Kooijman, D.; Sietsma, J.; Yang, Y. Hydro-metallurgical Recovery of Rare Earth Elements from Mine Tailings and WEEE. *Journal of Sustainable Metallurgy* **2018**, *4* (3), 367–377.
- (40) Good, N. M.; Lee, H. D.; Hawker, E. R.; Su, M. Z.; Gilad, A. A.; Martinez-Gomez, N. C. Hyperaccumulation of Gadolinium by *Methylorubrum extorquens* AM1 Reveals Impacts of Lanthanides on Cellular Processes Beyond Methylophagy. *Front. Microbiol.* **2022**, *13*, No. 820327.

- (41) Vu, H. N.; Subuyuj, G. A.; Vijayakumar, S.; Good, N. M.; Martinez-Gomez, N. C.; Skovran, E. Lanthanide-Dependent Regulation of Methanol Oxidation Systems in *Methylobacterium extorquens* AM1 and Their Contribution to Methanol Growth. *J. Bacteriol.* **2016**, *198* (8), 1250–1259.
- (42) Delaney, N. F.; Kaczmarek, M. E.; Ward, L. M.; Swanson, P. K.; Lee, M.-C.; Marx, C. J. Development of an Optimized Medium, Strain and High-Throughput Culturing Methods for *Methylobacterium extorquens*. *PLoS One* **2013**, *8* (4), No. e62957.
- (43) Diep, P.; Mahadevan, R.; Yakunin, A. F. Heavy Metal Removal by Bioaccumulation Using Genetically Engineered Microorganisms. *Front Bioeng Biotechnol* **2018**, *6*, 157.
- (44) Hogendoorn, C.; Roszczenko-Jasińska, P.; Martinez-Gomez, N. C.; de Graaff, J.; Grassl, P.; Pol, A.; Op den Camp, H. J. M.; Daumann, L. J. Facile Arsenazo III-Based Assay for Monitoring Rare Earth Element Depletion from Cultivation Media for Methanotrophic and Methylophilic Bacteria. *Appl. Environ. Microbiol.* **2018**, *84* (8), No. e02887.
- (45) Petzoldt, T. *Growthrates: R Package Growthrates*; Github.
- (46) Firsching, F. H.; Brune, S. N. Solubility Products of the Trivalent Rare-Earth Phosphates. *J. Chem. Eng. Data* **1991**, *36* (1), 93–95.
- (47) Chowdhury, N. A.; Deng, S.; Jin, H.; Prodius, D.; Sutherland, J. W.; Nlebedim, I. C. Sustainable Recycling of Rare-Earth Elements from NdFeB Magnet Swarf: Techno-Economic and Environmental Perspectives. *ACS Sustainable Chem. Eng.* **2021**, *9* (47), 15915–15924.
- (48) Binnemans, K.; Jones, P. T. Rare Earths and the Balance Problem. *Journal of Sustainable Metallurgy* **2015**, *1* (1), 29–38.
- (49) Yang, Y.; Walton, A.; Sheridan, R.; Güth, K.; Gauß, R.; Gutfleisch, O.; Buchert, M.; Steenari, B.-M.; Van Gerven, T.; Jones, P. T.; Binnemans, K. REE Recovery from End-of-Life NdFeB Permanent Magnet Scrap: A Critical Review. *Journal of Sustainable Metallurgy* **2017**, *3* (1), 122–149.
- (50) Priya, A.; Hait, S. Comparative Assessment of Metallurgical Recovery of Metals from Electronic Waste with Special Emphasis on Bioleaching. *Environ. Sci. Pollut. Res. Int.* **2017**, *24* (8), 6989–7008.
- (51) Brierley, C. L.; Brierley, J. A. Progress in Bioleaching: Part B: Applications of Microbial Processes by the Minerals Industries. *Appl. Microbiol. Biotechnol.* **2013**, *97* (17), 7543–7552.
- (52) Zhuang, W.-Q.; Fitts, J. P.; Ajo-Franklin, C. M.; Maes, S.; Alvarez-Cohen, L.; Hennebel, T. Recovery of Critical Metals Using Biometallurgy. *Curr. Opin. Biotechnol.* **2015**, *33*, 327–335.
- (53) Jain, N.; Sharma, D. K. Biohydrometallurgy for Nonsulfidic Minerals—A Review. *Geomicrobiol. J.* **2004**, *21* (3), 135–144.
- (54) Kermer, R.; Hedrich, S.; Bellenberg, S.; Brett, B.; Schrader, D.; Schönherr, P.; Köpcke, M.; Siewert, K.; Günther, N.; Gehrke, T.; Sand, W.; Rächle, K.; Bertau, M.; Heide, G.; Weitkämper, L.; Wotruba, H.; Ludwig, H.-M.; Partusch, R.; Schippers, A.; Reichel, S.; Glombitza, F.; Janneck, E. Lignite Ash: Waste Material or Potential Resource - Investigation of Metal Recovery and Utilization Options. *Hydrometallurgy* **2017**, *C* (168), 141–152.
- (55) Glombitza, F.; Reichel, S. Metal-Containing Residues from Industry and in the Environment: Geobiotechnological Urban Mining. *Adv. Biochem. Eng. Biotechnol.* **2013**, *141*, 49–107.
- (56) Shin, D.; Kim, J.; Kim, B.-S.; Jeong, J.; Lee, J.-C. Use of Phosphate Solubilizing Bacteria to Leach Rare Earth Elements from Monazite-Bearing Ore. *Minerals* **2015**, *5* (2), 189–202.
- (57) Hassanien, W. A. G.; Desouky, O. A. N.; Hussien, S. S. E. Bioleaching of Some Rare Earth Elements from Egyptian Monazite Using *Aspergillus ficuum* and *Pseudomonas aeruginosa*. *Walailak J. Sci. Technol.* **2014**, *11* (9), 809–823.
- (58) Amiri, F.; Yaghmaei, S.; Mousavi, S. M. Bioleaching of Tungsten-Rich Spent Hydrocracking Catalyst Using *Penicillium simplicissimum*. *Bioresour. Technol.* **2011**, *102* (2), 1567–1573.
- (59) Qu, Y.; Lian, B. Bioleaching of Rare Earth and Radioactive Elements from Red Mud Using *Penicillium Tricolor* RM-10. *Bioresour. Technol.* **2013**, *136*, 16–23.
- (60) Maes, S.; Zhuang, W.-Q.; Rabaey, K.; Alvarez-Cohen, L.; Hennebel, T. Concomitant Leaching and Electrochemical Extraction of Rare Earth Elements from Monazite. *Environ. Sci. Technol.* **2017**, *51* (3), 1654–1661.
- (61) Corbett, M. K.; Eksteen, J. J.; Niu, X.-Z.; Croue, J.-P.; Watkin, E. L. J. Interactions of Phosphate Solubilising Microorganisms with Natural Rare-Earth Phosphate Minerals: A Study Utilizing Western Australian Monazite. *Bioprocess Biosyst. Eng.* **2017**, *40* (6), 929–942.
- (62) Schmitz, A. M.; Pian, B.; Medin, S.; Reid, M. C.; Wu, M.; Gazel, E.; Barstow, B. Generation of a *Gluconobacter oxydans* Knockout Collection for Improved Extraction of Rare Earth Elements. *Nat. Commun.* **2021**, *12* (1), 6693.
- (63) Xiong, Y. Organic Species of Lanthanum in Natural Environments: Implications to Mobility of Rare Earth Elements in Low Temperature Environments. *Appl. Geochem.* **2011**, *26* (7), 1130–1137.
- (64) Zytneck, A. M.; Gutenthaler, S. M.; Aron, A. T.; Reitz, Z. L.; Phi, M. T.; Good, N. M.; Petras, D.; Daumann, L. J.; Martinez-Gomez, N. C. Discovery and Characterization of the First Known Biological Lanthanide Chelator. *bioRxiv* **2023**, No. 476857.
- (65) Marx, C. J.; Lidstrom, M. E. Development of Improved Versatile Broad-Host-Range Vectors for Use in Methylophilic and Other Gram-Negative Bacteria. *Microbiology* **2001**, *147* (Pt 8), 2065–2075.
- (66) Lumpe, H.; Daumann, L. J. Studies of Redox Cofactor Pyrroloquinoline Quinone and Its Interaction with Lanthanides(III) and Calcium(II). *Inorg. Chem.* **2019**, *58* (13), 8432–8441.
- (67) Arshi, P. S.; Vahidi, E.; Zhao, F. Behind the Scenes of Clean Energy: The Environmental Footprint of Rare Earth Products. *ACS Sustainable Chem. Eng.* **2018**, *6* (3), 3311–3320.