Short Communication

## Sterols of Chlorella. II. The Occurrence of an Unusual Sterol Mixture in Chlorella vulgaris<sup>1, 2</sup>

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The first identification of a sterol in Chlorella was by Klosty and Bergmann (5) in 1952 when they isolated ergosterol from Chlorella pyrenoidosa. More recently, Otsuka (6) identified ergosterol and a  $\Delta^5$ sterol in Chlorella ellipsoidea. Ergosterol was the principal component of the mixture. Recent work in this laboratory has been centered about the identification of sterols from other species of Chlorella. A mixture of the  $\triangle^5$  sterols, poriferasterol, clionasterol, and 22-dihydrobrassicasterol was found (7) in C. ellipsoidea and C. saccharophila, but no ergosterol was detected. Chlorella vulgaris was shown to contain chondrillasterol as its principal sterol component, and although 2 other sterols were present in significant quantities, they were not isolated. Since plant sterols are known to have an effect on animal metabolism, and algae are being considered as future sources of food, it is important that we know the sterol composition of these algae. This paper describes the isolation and identification of each of the 2 unknown sterols occurring in C. vulgaris.

Cells of Chlorella vulgaris Beyer., Emerson's strain, were grown heterotrophically on basal inorganic medium containing 0.5 % glucose in 15-liter carboys equipped with bubbling tubes for air (3). The cells were harvested in a Sharples Super Centrifuge and freeze-dried before extraction. Average yield was 3 grams dry weight per liter. Lipid material was extracted from the cells with acetone in a soxhlet apparatus, saponified under nitrogen, and the nonsaponifiable matter extracted with ether in a liquidliquid extraction apparatus. The non-saponifiable lipid was fractionated as described by Hoftmann et al. (3) on Woelm Grade III neutral alumina. The fraction containing sterols was acetylated and rechromatographed under the same conditions described above. The fraction containing the mixture of sterol acetates was subjected to column chromatography on Anasil B which was added to the 3 cm  $\times$  40 cm column in a slurry of n-hexane. The sterol acetates were added to the column in a minimum amount of

*n*-hexane and eluted with 2% ether in *n*-hexane. Gas chromatography was used to assay the 15-ml fractions which were collected. The 3 sterols isolated were named sterol 1, sterol 2, and sterol 3 in the order of their elution from the Anasil column. Pure sterol 1 was obtained from the Anasil column only with great difficulty. It was more readily obtained by preparative gas chromatography of the unresolved sterol 1-sterol 2 mixture. The trapped sterol was purified by alumina fractionation, digitonin precipitation and regeneration of the free sterol followed by recrystallization in methanol and acetone. Subsequent gas chromatographic analysis showed no impurities even when sterol 1 was injected in amounts sufficient to produce full-scale deflection of the recorder.

Nearly 100 mg of the sterol mixture was obtained from 100 g of C. vulgaris. The approximate composition of the sterol mixture was sterol 1, 10 %; sterol 2, 25 %; and sterol 3, 65 %. Sterol 3 has previously been identified as  $24\beta$ -ethyl- $\Delta^{7, 22}$  cholestadienol or chondrillasterol (7), a sterol which also occurs in another unicellular green alga, Scenedesmus obliquus (2). Sterols 1 and 2 also gave rapid Liebermann-Burchard reactions indicative of  $\wedge^{7}$  double bond and failed to show the typical ultraviolet absorption spectrum of a  $\triangle^{5, 7}$  sterol. Thus these sterols, like chondrillasterol, are  $\triangle^7$  sterols. The infrared spectra of sterols 1 and 2 were essentially identical. They were also similar to the spectrum of chondrillasterol except for the absence of the strong band at  $10.3\mu$ . The presence of this absorption band indicates the presence of a trans double bond at C-22 (4). The infrared spectrum of  $\triangle^7$  chondrillastenol, produced by Raney Nickel hydrogenation of chondrillasterol, was identical to that of sterol 1.

Gas chromatographic retention times (table I) indicate that sterol 1 is a  $\triangle^7$  C-29 sterol and sterol 2 is a  $\triangle^7$  C-28 sterol since the GLC retention times of sterol 1 coincide with those of  $24\beta$ -ethyl- $\triangle^7$  cholestenol ( $\triangle^7$ -chondrillastenol) and the retention times of sterol 2 coincide with those of  $24\beta$ -methyl  $\triangle^7$  cholestenol ( $\triangle^7$  ergostenol).

In table II, melting point and optical rotation data of *Chlorella* sterols are compared with values obtained from the literature for  $\triangle^{\tau}$  ergostenol (1) (also called fungisterol) and those obtained for  $\triangle^{\tau}$  chondrillastenol synthesized from chondrillasterol. The values

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	Relative	time*	
Compound	SE-30**	QF-1***	NGS†
Chondrillasterol	3.06	4.00	11.8
$\Delta^7$ Chondrillastenol Sterol 1	3.47 3.47	4.48 4.46	14.1 14.2
∆ <sup>7</sup> Ergostenol Sterol 2	2.82 2.82	3.80 3.80	$\begin{array}{c} 11.0\\ 11.1 \end{array}$

 
 Table I. GLC Relative Retention Times of Chlorella Sterols and Certain Known Sterols

\* Relative to cholestane.

- \*\* Column 6 ft × 3.4 mm ID, 2 % SE-30 on 100-140 mesh Gas-chrom Q, 20 psi, 236°, cholestane time 7 min.
- \*\*\* Column 6 ft × 3.4 mm ID, 1 % QF-1 on 100-140 mesh Gas-chrom P, 20 psi, 236°, cholestane time 3 min.
  - † Column 6 ft  $\times$  3.4 mm ID. 1 % NGS on 100-140 mesh Gas-chrom P, 20 psi, 216°, cholestane time 4 min.

obtained for sterol 2 are essentially identical to those found in the literature for  $\triangle^{\tau}$  ergostenol. Agreement is also excellent in the comparison of melting points of sterol 1 and  $\triangle^{\tau}$  chondrillasterol.

A mass spectrum of the acetate of sterol 1 was obtained. The experimental molecular weight determination of 456 corresponds to that calculated for  $\Delta^{\tau}$  chondrillasteny! acetate. Therefore, sterol 1 is  $24\beta$ -ethyl  $\Delta^{\tau}$  cholestenol ( $\Delta^{\tau}$  chondrillastenol) and sterol 2 is identical to  $\Delta^{\tau}$  ergostenol ( $24\beta$ -methyl- $\Delta^{\tau}$  cholestenol).

This is the first reported isolation of  $\triangle^{\tau}$  chondrillastenol from natural sources. This is also the first reported isolation of  $\triangle^{\tau}$  ergostenol from a green plant. It is of interest to note that gas chromatographic analysis of the sterol fraction from *Scenedesmus obliquus*, in which chondrillasterol was first isolated from a plant (2), in addition shows peaks with identical relative retention times to that of  $\triangle^{\tau}$ ergostenol and  $\triangle^{\tau}$  chondrillastenol. It seems quite probable that these 2 sterols are also companions of chondrillasterol in *Scenedesmus*.

The sterols of *Chlorella vulgaris* (fig 1) are identical to those previously isolated from *Chlorella* 

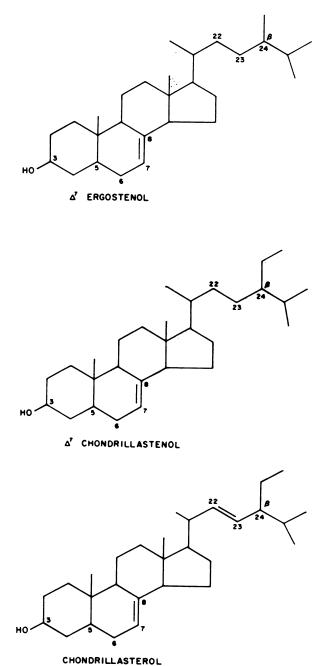


Table II. Melting Point and Optical Rotation Data for Chlorella Sterols and Certain Other Sterols

Sterol	% of Total Sterol	M.P. Sterol	M.P. Acetate	Optical rotation (at 25 C in CHCl <sub>3</sub> ) Sterol
Chondrillasterol (Sterol 3)	65	169–70	174–5	$\pm 0$
$\Delta^7$ Chondrillastenol Sterol 1	10	140–1 140–2	160.5–162 161–163	+ 5
$\Delta^7$ Ergostenol Sterol 2	25	148 148–9	160 159–161	0 2

ellipsoidea except that C. vulgaris contains  $\triangle^7$  sterols while those of C. ellipsoidea are  $\triangle^5$  sterols. In all living organisms studied which synthesize  $\triangle^5$  sterols,  $\triangle^7$  sterols are precursors in the biosynthesis of the  $\triangle^5$  sterols. Chlorella vulgaris differs from C. ellipsoidea in its apparent inability to convert the steroid  $\triangle^7$  double bond to the  $\triangle^5$  double bond as most plants do.

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