

Calcium binding, structural stability and guanylate cyclase activation in GCAP1 variants associated with human cone dystrophy

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Abstract Guanylate cyclase activating protein 1 (GCAP1) is a neuronal Ca^{2+} sensor (NCS) that regulates the activation of rod outer segment guanylate cyclases (ROS-GCs) in photoreceptors. In this study, we investigated the Ca^{2+} -induced effects on the conformation and the thermal stability of four GCAP1 variants associated with hereditary human cone dystrophies. Ca^{2+} binding stabilized the conformation of all the GCAP1 variants independent of myristoylation. The myristoylated wild-type GCAP1 was found to have the highest Ca^{2+} affinity and thermal stability, whereas all the mutants showed decreased Ca^{2+} affinity and significantly lower thermal stability in both apo and Ca^{2+} -loaded forms. No apparent cooperativity of Ca^{2+} binding was detected for any variant. Finally, the non-myristoylated mutants were still capable of activating ROS-GC1, but the measured cyclase activity was shifted toward high, nonphysiological Ca^{2+} concentrations. Thus, we conclude that distorted Ca^{2+} -sensor properties could lead to cone dysfunction.

Keywords Neuronal calcium sensor · GCAP · Photoreceptor · Cone dystrophy

Introduction

The concentration of cytoplasmic Ca^{2+} in photoreceptor cells changes dynamically upon light exposure, ranging from 500 to 600 nM in dark-adapted cells to 50–100 nM in the light.

Guanylate cyclase-activating proteins (GCAPs) are expressed in retinal rod and cone cells and function as neuronal calcium sensors (NCS) [1] that detect changes in Ca^{2+} concentration and modulate the activity of membrane-bound photoreceptor-specific guanylate cyclases (ROS-GCs) in a Ca^{2+} -dependent manner [2–4]. Ca^{2+} /GCAP-dependent regulation of ROS-GC activity forms a powerful feedback mechanism that sets the overall photoresponse and is highly involved in adaptation phenomena [5, 6]. Among GCAPs, GCAP1 shows stronger immunoreactivity in cones than in rods [7] and was found to be capable of restoring recovery of rod and cone photoresponses in the absence of GCAP2 [3, 8]. However, recent studies showed that the kinetics of recovery from light flashes is slowed down in mice lacking GCAP2, suggesting a role for GCAP2 in adjusting the operating range of rods to higher light intensities [9].

In a very recent work, three novel (i.e., E89K, D100E, G159V) [10] and one previously identified (i.e., L151F) [11, 12] mutations in the gene encoding GCAP1 were found in patients suffering from different forms of cone dystrophies, and the corresponding heterologously expressed proteins were investigated in their capability of activating ROS-GC1 [10]. The disease-associated GCAP1 variants showed as a common hallmark the persistent stimulation of ROS-GC1 at physiological calcium [10]. Moreover, the GCAP1 conformational change upon Ca^{2+} binding was monitored for each variant by fluorescence spectroscopy and gel electrophoresis, highlighting differences between the wild-type (WT) and the mutated forms [10].

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In this study, we investigate whether the persistent stimulation of ROS-GC1 at physiological Ca^{2+} of disease-associated GCAP1 mutants is related to altered Ca^{2+} affinity and/or altered Ca^{2+} -induced response of the protein conformation and stability. We thus present a biophysical and functional characterization of four disease-associated GCAP1 mutants and compare their structural stability with that of the WT in the presence and in the absence of Ca^{2+} . We further investigate the influence of the myristoyl group on stability, conformation and function in these GCAP1 variants. The equilibrium Ca^{2+} -binding constants were measured for each GCAP1 variant, and our data are consistent with non-cooperative sequential binding that is strongly influenced both by myristoylation and by the point mutations.

Materials and methods

Building the homology model of human myristoylated GCAP1

The homology model of human myristoylated (myr) GCAP1 was built as follows. The FASTA sequence of human GCAP1 was retrieved from the ExpASY Proteomics Server (<http://www.expasy.ch>) and used as an input for the Modeller web server for protein structure modeling (ModWeb server: <http://modbase.compbio.ucsf.edu/ModWeb20-html/modweb.html>). Twenty-six models were retained out of 213 calculated by the server, which were ranked according to pre-defined indexes based on the reliability index and a composite score calculated with respect to the X-ray structure of chicken GCAP1 [13] (PDB entry: 2R2I) used as a template (sequence identity: 85.00%). The models were checked with the Protein Health module in the QUANTA2006 package (<http://www.accelrys.com>), and the highest score model passed all the quality checks in

terms of main and side chain conformations and atomic contacts. Superimposition of the modeled human GCAP1 with the template (chicken GCAP1) led to a 0.34 Å C α -root mean-square deviation, indicating high structural similarity in particular in the EF-hand loop regions. The three Ca^{2+} ions and the myristoyl chain were manually placed, followed by rotamer adjustment of the surrounding side chains. The resulting three-dimensional (3D) model is shown in Fig. 1.

Expression and purification of GCAP1 variants

Wild-type and mutants of GCAP1 in their nonmyristoylated and myristoylated forms were obtained as described previously [10]. In brief, by employing the same vectors pET21-bGCAP1 and pET21-bGCAP1-D6S as described in [10], the GCAP1 variants were obtained by overexpression in BL21 *E. coli* cells. The expressed GCAP1 forms were extracted from the insoluble fraction of the inclusion bodies by homogenization in 6 M guanidine hydrochloride and dialyzed twice against 3 l of Tris buffer (20 mM Tris-HCl, 150 mM NaCl, 1 mM DTT pH 7.4), and insoluble fractions were removed by centrifugation (100,000 $\times g$ for 30 min). The supernatant was concentrated to 5 ml, and the GCAP variants were purified in the presence of 1 mM EDTA and 1 mM EGTA by a combination of size exclusion and ion-exchange chromatography using a Superdex75 column and a UnoQ-column, respectively (BioRad Life Science Group, Hercules, CA) [14]. The degree of myristoylation was analyzed by reversed-phase high-performance liquid chromatography [15].

Protein sample preparation and circular dichroism spectroscopy

For CD spectroscopy and UV monitored Ca^{2+} titrations in the presence of chelator, proteins were dissolved in a

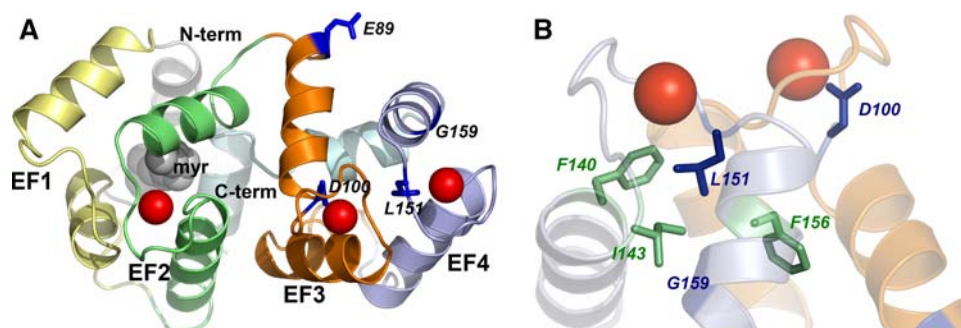


Fig. 1 Three-dimensional homology model of human myrGCAP1. **a** Cartoon representation of the myrGCAP1 overall fold. EF1 is colored in yellow, EF2 in green, EF3 in orange and EF4 in light blue. N- and C-terminals are colored in pale grey and pale cyan, respectively, and the myristoyl group is represented by grey spheres. Calcium ions are represented by red spheres, while residue targets of

cone dystrophy-associated mutations are labeled and shown by dark blue sticks. **b** Details of the EF3-EF4 domain shown from a different angle. Some key hydrophobic residues (see “Discussion”) are shown by green sticks and labeled; all the other structural features are represented in transparency with the same colors as in **a**

decalcified Tris-buffer (5 mM Tris-HCl, 150 mM KCl, pH 7.5). Decalcification was achieved by insertion of a five-times boiled dialysis tube (3,500 MW cutoff, 18 mm width, 11.5 mm diameter, 1 ml/cm capacity) filled with Chelex-100 resin (Bio-Rad) in the plastic bottle containing the buffer, and left on a rocker for few days. Adjustment of the pH following the decalcification process was performed with particular care (inserting the pH electrode in a withdrawn fraction) to avoid calcium contamination of the buffer. For each preparation, MilliQ water was employed (Millipore Corporation).

All circular dichromism (CD) experiments were carried out using a JASCO J-815 spectropolarimeter (JASCO, Easton, MD) with a Peltier type thermostated cell holder. Far-UV CD spectra (200–250 nm) were recorded at 37°C with a scan rate of 20 nm/min, a band width of 4 nm and a digital integration time of 16 s. Four spectra were accumulated and averaged for each sample. The spectrum of the buffer alone was subtracted from the spectrum of each protein sample. The protein concentration was 19–25 μM in 0.1-cm cuvettes. The spectra of the apo-GCAP1 forms were first recorded in the presence of 100 μM EDTA (Fig. 2, black dashed line). CaCl_2 was then added to a final concentration of 200 μM , and the spectra of the Ca^{2+} -loaded forms of the GCAP1 variants were recorded (Fig. 2, black solid line).

The thermal denaturation of all the GCAP1 variants in both their apo- and Ca^{2+} -loaded forms was monitored between 4 and 96°C using the same conditions and concentration range as for the far-UV spectra. The ellipticity at 222 nm (θ_{222}) was recorded with a scan rate of 1°C/min and a response time of 16 s using 0.1 cm quartz cuvettes with a Teflon stopper. For the Ca^{2+} -loaded form of each variant, a 250–200 nm CD spectrum was recorded 5–30 min after reaching the maximum unfolding temperature ($T = 96^\circ\text{C}$) (red dashed lines in Fig. 2). The denatured samples were then cooled to 4°C to check for reversibility, and 5–30 min after reaching the minimum temperature another CD spectrum was recorded (blue dashed line in Fig. 2).

Analysis of thermal denaturation data

The analysis of thermal denaturation curves was performed for each variant assuming a two-state transition. If equilibrium were reached between a fully folded and a fully unfolded state, the θ_{222} as a function of the temperature could be described quantitatively by accounting for the fraction of folded and unfolded protein. However, in our experiments no complete unfolding was achieved for any variant and in any condition, as is clearly shown by the CD signals that remain negative even at 96°C. The “unfolded” state indeed refers to the denatured state observed in our

experiments, for which residual structure is clearly present. Hence, the theoretical fitting curve was used as an ideal approximation for assessing the apparent transition temperature (T_i):

$$\theta_{222}(T) = \frac{(b_n + k_n T) + (b_u + k_u T) \exp[-\Delta G_{nu}(T)]}{1 + \exp\left(\frac{-\Delta G_{nu}(T)}{RT}\right)} \quad (1)$$

where n indicates the native, folded state and u the (ideal) unfolded state; b is the baseline value and k the slope; T is the temperature and ΔG_{nu} is the Gibbs free energy for the folded–unfolded transition and is given by:

$$\Delta G_{nu}(T) = -\left(\Delta H\left(1 - \frac{T}{T_i}\right)\right) + \Delta C_p\left(T - T_i - T \ln \frac{T}{T_i}\right) \quad (2)$$

where ΔH and ΔC_p respectively indicate the change in enthalpy and heat capacity upon denaturation at constant pressure. We point out that the only parameter obtained from the fitting procedure that was actually employed in comparison between the GCAP1 variants was the apparent transition temperature T_i , since the lack of a real equilibrium between the native and fully unfolded state in our experiments made a detailed thermodynamic characterization of the process unfeasible.

While the thermal denaturation curves of the GCAP1 apo-forms were in every case fitted well by the two-state unfolding curve (Fig. 2, right columns panels, full red lines), the Ca^{2+} -loaded forms showed in general a much more gradual transition, indicating a denaturation process of low cooperativity, and within the tested temperature range a quantitative fitting was not always possible. The experimental setting in fact limits the study to temperatures $<100^\circ\text{C}$. Table 1 reports the fitted or the estimated values of T_i for each GCAP1 variant.

Determination of Ca^{2+} -binding constants using a chromophoric chelator

Calcium binding to GCAP1 variants was monitored by an accurate titration method described previously [16–18], which is based on the competition for Ca^{2+} between the protein and a chromophoric chelator whose absorbance changes upon Ca^{2+} binding. The precision of the method is very high for the determination of the equilibrium binding constants, especially when a set of proteins, i.e., different mutants, are compared [16, 17]. Approximately equal amounts (10–50 μM) of chelator and protein are titrated with Ca^{2+} , and the binding of the cation to the chelator is monitored by UV absorbance at 263 nm, allowing the extraction of the binding constants by computer fitting the absorbance as a function of the total Ca^{2+} concentration. In detail, a Ca^{2+} -free solution of the 5,5'-Br₂-BAPTA chelator

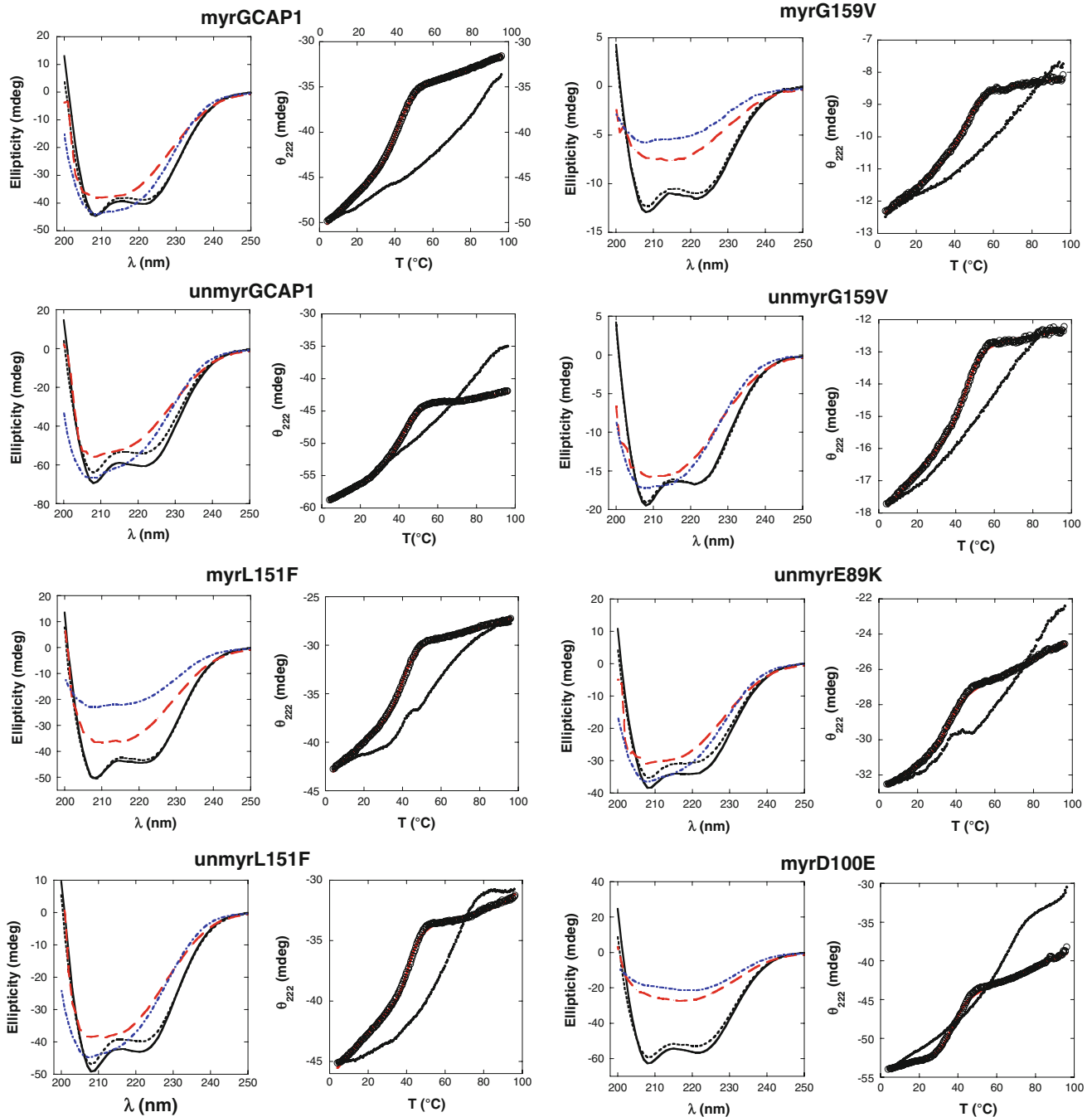


Fig. 2 Data from CD spectroscopy performed on GCAP1 variants. For each GCAP1 variant, the panel on the left reports the far UV-CD spectra recorded in the 200–250 nm range for the apo form at $T = 37^\circ\text{C}$ (black dashed line), the Ca^{2+} form at $T = 37^\circ\text{C}$ (solid black line), the Ca^{2+} form after thermal denaturation ($T = 96^\circ\text{C}$, red dashed line) and the Ca^{2+} form after partial renaturation ($T = 4^\circ\text{C}$, blue dashed line). In the right panels, the thermal unfolding curves are

was prepared by dissolving the chelator in the decalcified Tris-buffer described above, and the exact chelator concentration C_Q was spectroscopically obtained by measuring the absorbance at 239.5 nm after addition of saturating

reported for each variant in terms of CD signal at 222 nm as a function of temperature in the 4–96°C range. Data points are reported in the absence (open circles) and in the presence (filled circles) of saturating Ca^{2+} . For the apo cases, computer-fitted curves to apo-protein data by the two-state model (Eq. 1) are shown by solid red lines

Ca^{2+} . It was found to range between 19 and 21 μM . The initial concentration of Ca^{2+} in the chelator solution [Ca_Q] was also spectroscopically determined to be between 1.6 and 3 μM for all samples. For each titration experiment,

Table 1 Results from far-UV CD spectroscopy, thermal denaturation and Ca^{2+} -binding experiments with GCAP1 variants

	Relative $\Delta\theta_{222}^a$	$T_t^{\text{apo}} (\text{°C})^b$	$T_t^{\text{Ca}} (\text{°C})^b$	$\log K_1^c$	$\log K_2^c$	$\log K_3^c$	$\log(K_1 K_2)/2$	$\text{IC}_{50} (\mu\text{M})^d$
myrGCAP1	0.047	50.1 ± 1.5	>70	8.4 ± 0.4	7.51 ± 0.07	5.9 ± 0.4	7.9 ± 0.3	0.8 ± 0.2^e
nonmyrGCAP1	0.12	42.7 ± 0.2	>60	7.8 ± 0.5	6.7 ± 0.3	3.3 ± 3.0	7.2 ± 0.4	2.2 ± 0.8^f
myrG159V	0.044	48.4 ± 0.2	>60	6.7 ± 0.2	4.7 ± 0.5	3.3 ± 1.9	5.6 ± 0.2	14.5 ± 1.9^e
nonmyrG159V	0.006	47.1 ± 0.2	63 ± 8	6.7 ± 0.3	5.81 ± 0.16	3.6 ± 1.7	6.2 ± 0.2	40.3 ± 1.0
myrL151F	0.023	43.0 ± 0.3	63 ± 3	7.3 ± 0.2	4.0 ± 1.6	5.1 ± 0.7	5.6 ± 0.8	39 ± 5
nonmyrL151F	0.075	43.5 ± 0.3	65.3 ± 0.7	6.9 ± 0.9	6.6 ± 0.2	5.0 ± 0.3	6.8 ± 0.5	576 ± 131
myrD100E	0.073	37.5 ± 0.2	66.1 ± 0.6	6.37 ± 0.13	5.0 ± 0.5	5.1 ± 0.3	5.6 ± 0.3	21.9 ± 1.3^e
nonmyrE89K	0.12	36.9 ± 0.3	74.4 ± 0.7	7.0 ± 0.2	6.4 ± 0.5	5.0 ± 0.2	6.6 ± 0.3	40 ± 9

^a Measured at $T = 37^\circ\text{C}$

^b T_t represents the transition temperature according to the two-state native-unfolded transition model (see Eqs. 1 and 2); apo refers to the calcium-free form, whereas Ca to the calcium-loaded form

^c K_1 , K_2 and K_3 are the three macroscopic equilibrium constants for calcium binding to GCAP1 variants

^d IC_{50} is the calcium concentration at which the GC activity is half maximal

^e Data from ref. [10]

^f Data from ref. [14]

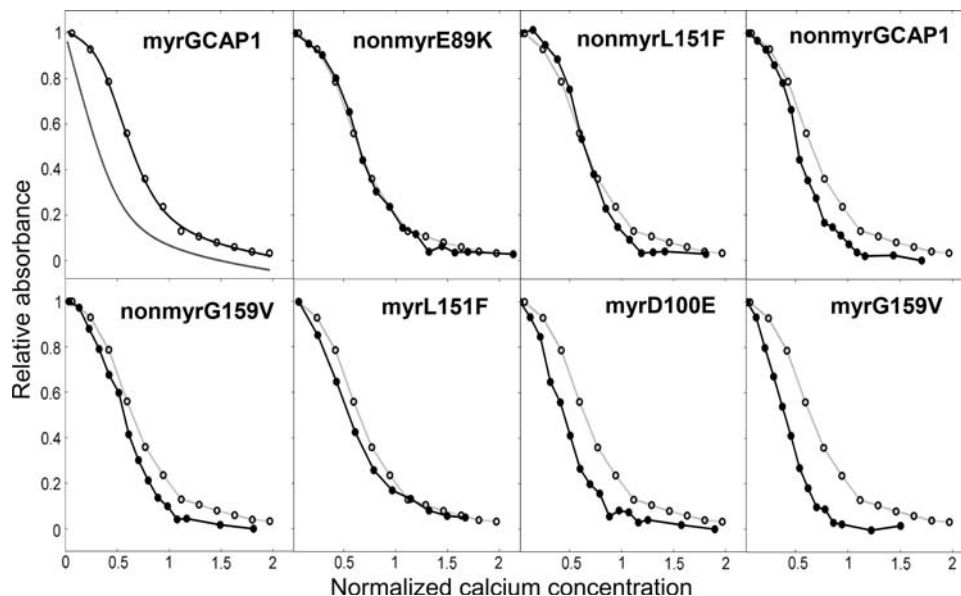


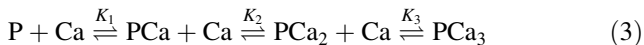
Fig. 3 Ca^{2+} -titration curves for the eight variants of GCAP1. In the upper left panel, the experimental points for myrGCAP1 (empty circles) are shown together with the optimal curve by computer fitting (see “Materials and methods”) and the theoretical (simulated) curve for a protein that binds Ca^{2+} several orders of magnitude lower than the chelator. The curves include the effects of dilution upon titration. In the other panels, empty circles and grey curves refer to myrGCAP1, reported for comparison, while filled circles and black curves refer to one representative set of titration data obtained for that

lyophilized proteins were dissolved in the decalcified chelator solution to obtain a similar concentration to that of the chelator. The precise concentration of each protein stock C_p was determined by amino acid analysis following acid hydrolysis (Biomedical Center, Uppsala, Sweden). In each experiment, 1 ml of chelator-protein solution was

variant. The axes have been normalized in the following way: $Y = \frac{A_{263} - A_{Ca}}{A_0 - A_{Ca}}$, where A_{263} is the absorbance read at 263 nm, A_{Ca} is the fitted absorbance at saturating Ca^{2+} , and A_0 is the fitted absorbance in the absence of Ca^{2+} additions, $X = \frac{[\text{Ca}^{2+}]_{\text{tot}}}{C_0 + 3C_p}$, where $[\text{Ca}^{2+}]_{\text{tot}}$ is the total Ca^{2+} concentration (μM) after each addition, C_0 is the total concentration of the chelator (μM), C_p the total concentration of the protein (μM), and 3 is the number of Ca^{2+} -binding sites per protein variant

inserted in a 1-cm cuvette, and the absorbance at 263 nm (λ_{max} for the Ca^{2+} -free chelator) was monitored upon titration with fixed amounts (2.5 μl) of 3 mM CaCl_2 . Titrations were performed until no significant change in the absorbance was detected. Two or three replications of each titration experiment were performed, and the data (Fig. 3)

were fitted using a Newton-Raphson direct least-square fitting procedure implemented in the CaLigator software [16]. In brief, the model used for the fitting allows determining the macroscopic binding constants, the response signals being dependent only on the stoichiometric number of bound Ca^{2+} . Therefore, since GCAP1 has three binding sites, the involved equilibria concerning the protein are:



where P denotes the protein (GCAP1 variant), and the K 's are the macroscopic equilibrium constants. The analysis of the equilibria including the one with the chelator leads to the following equation for the total Ca^{2+} :

$$[\text{Ca}_{\text{tot}}] = [\text{Ca}]_{\text{free}} + \frac{[\text{Ca}]_{\text{free}} C_Q}{K_d + [\text{Ca}]_{\text{free}}} + \frac{C_p \left(K_1 [\text{Ca}]_{\text{free}} + 2K_1 K_2 [\text{Ca}]_{\text{free}}^2 + 3K_1 K_2 K_3 [\text{Ca}]_{\text{free}}^3 \right)}{1 + K_1 [\text{Ca}]_{\text{free}} + K_1 K_2 [\text{Ca}]_{\text{free}}^2 + K_1 K_2 K_3 [\text{Ca}]_{\text{free}}^3} \quad (4)$$

where K_d is the dissociation constant for the chelator- Ca^{2+} complex, which is 2.3 μM for 5,5'-Br₂-BAPTA in Tris 2 mM, 150 mM KCl, at pH 7.5 [17]. The absorbance at each titration point is calculated as:

$$a_{\text{calc},i} = \left(a_{\text{free}} + (a_{\text{bound}} - a_{\text{free}}) \frac{[\text{Ca}]_{\text{free}}}{K_d + [\text{Ca}]_{\text{free}}} \right) \frac{C_{Q_i}}{C_{Q_0}} \quad (5)$$

where a_{free} is the absorbance when no Ca^{2+} is added, a_{bound} is the final absorbance in saturating conditions, when no more Ca^{2+} is added and C_{Q_0} and C_{Q_i} are the total chelator concentrations at the start and at the titration point i , respectively.

During the fitting procedure, $[\text{Ca}]_{\text{free}}$ in Eq. 5 is replaced by a numerical solution of Eq. 4. The sum of the squares of residuals χ^2 is obtained by summing over all points in the titration:

$$\chi^2 = \sum_n (a_{\text{calc}} - a_{\text{meas}})^2 \quad (6)$$

and the variable parameters are iterated in a separate procedure until an optimal fit, i.e., minimum χ^2 , is obtained. The fitting procedure is rendered automatically by the software CaLigator [16].

All the Ca^{2+} -binding experiments were performed at room temperature ($T \sim 21^\circ\text{C}$), which ensures thermal stability for all the variants analyzed in this study.

Guanylate cyclase assay

Guanylate cyclase activity was measured by using bovine rod outer segment membranes, of which ca. 96% of the cyclase activity originates from ROS-GC1 [19]. Purified GCAP1 forms were reconstituted with washed ROS

membranes devoid of intrinsic GCAP activity as described before [20]. The assay buffer was slightly modified to yield free Mg^{2+} of 1 mM. All incubation steps were performed under very dim red light, and incubation time was 5 min. Synthesis of cGMP was quantified by high performance liquid chromatography as described [20].

Results

Cone dystrophy-associated mutations in myrGCAP1: a structural perspective

Based on the available high-resolution structure of Ca^{2+} -bound myrGCAP1 from chicken [13], we built a reliable homology model of human myrGCAP1 to focus on the regions targeted by three novel and one known mutation recently found in patients suffering from different forms of cone dystrophy [10]. The model is shown in Fig. 1, where the EF-hand motifs (EF1: pseudo EF hand, in yellow; EF2, EF3 and EF4: Ca^{2+} -binding EF hands, in green, orange and blue, respectively) are shown together with the N-(grey) and C-terminal (pale cyan) helices, with the myristoyl group shown by grey spheres and the residues affected by point mutations shown by sticks, namely E89, G159, D100 and L151. A visual inspection of the homology model revealed a perturbation of helical structures in the EF-hand domains, as well as changes in Ca^{2+} coordination and in surface charges (for a detailed discussion of structural aspects, see below). Since it is known that protein surface charges can greatly affect the thermodynamics of Ca^{2+} binding, even if not directly involved in the metal coordination [21], and that they may also significantly influence the overall protein stability [22], we investigated the influences of the point mutations in GCAP1 on conformation and stability.

Far UV-CD spectroscopy of myrGCAP1 variants: effects of Ca^{2+} -binding

Far-UV CD spectra were recorded for all the GCAP1 variants in the absence and presence of Ca^{2+} , and the results are shown in Fig. 2. In the absence of Ca^{2+} all the GCAP1 variants clearly showed the typical CD spectrum of α -helical proteins, indicating that the mutations do not appreciably perturb the protein secondary structure. Moreover, the absence of the myristoyl moiety did not affect the shape of the spectra, indicating that the secondary structure was in no case significantly perturbed by the acyl modification (Fig. 2).

Similarly to what has been previously observed for WT GCAP2 [23], addition of Ca^{2+} leads to an increase of the CD signal for all the variants (Fig. 2). Depending on the GCAP1 variant, the change in ellipticity at 222 nm ranges

from no change (0.6%) to 12% (Table 1). The detected changes are similar to or lower than the 14% increase observed for myrGCAP2 upon Ca^{2+} binding [23].

Effects of Ca^{2+} and myristoylation on the stability of GCAP1 variants

Thermal denaturation was investigated in the 4–96°C range for all the GCAP1 variants by monitoring the ellipticity at 222 nm, in the absence and presence of Ca^{2+} . The raw data are shown in Fig. 2, and the values of the transition temperature T_t obtained by fitting a two-state transition model to these data (see “Materials and methods”) are reported in Table 1.

When denaturation occurred in the absence of Ca^{2+} , a clear transition between two structural states could be observed, allowing for an accurate fitting (Fig. 2). However, addition of Ca^{2+} shifted the transition to higher temperatures, and the denatured state in some cases was not detectable (Fig. 2), thus making the estimate of T_t only approximate. The stabilizing effect of Ca^{2+} was apparent for all the GCAP1 variants as the transition toward denatured states always occurred at higher temperatures compared to the apo-forms (Fig. 2).

Wild-type myrGCAP1 showed the highest thermal stability in both the apo ($T_t = 50.1 \pm 1.5^\circ\text{C}$) and Ca^{2+} -loaded forms ($T_t > 70^\circ\text{C}$) compared to all other variants. These estimates are consistent with the very recent differential scanning calorimetry (DSC) experiments by Lim et al. [24], who measured a transition temperature of 52°C for apo-WT myrGCAP1 and of 103°C for the Ca^{2+} -bound form. A slightly lower transition temperature ($T_t = 42.7 \pm 0.2^\circ\text{C}$) was instead obtained in this study for the apo-nonmyrGCAP1 compared to the value obtained by DSC measurements ($T_t = 47^\circ\text{C}$, [24]). Nevertheless, in both experiments the stabilizing effect of the myristoyl group on the WT structure of GCAP1 was confirmed. However, the stabilizing effect of myristoylation was not observed for the mutants G159V and L151F for which we detected only minor differences in T_t compared to nonmyristoylated forms both in the absence and in the presence of Ca^{2+} (Table 1). Apparently, the destabilizing effect of the mutations was not compensated by the presence of the myristoyl group.

Among all tested GCAP1 variants, myrD100E ($T_t = 37.5 \pm 0.2^\circ\text{C}$) and nonmyrE89K ($T_t = 36.9 \pm 0.3^\circ\text{C}$) were found to be the most unstable in the absence of Ca^{2+} . When saturating Ca^{2+} was added, all GCAP1 variants became more stable than their apo-forms as shown by a shift to higher T_t s that largely exceeded 15°C in most cases (Table 1). Particularly impressive is the Ca^{2+} -induced stabilization of nonmyrE89K that exhibited a native-like stability and a difference in T_t between the apo and the Ca^{2+} -bound form of $>37^\circ\text{C}$.

We did not observe reversible unfolding for any of the GCAP1 samples similarly to recently reported DSC experiments [24]. In order to assess the protein secondary structure after denaturation, we recorded the far UV-CD spectrum for all the Ca^{2+} -loaded variants at 96°C (Fig. 2, red dashed lines). While some secondary structure was clearly retained, the spectra demonstrated a loss of α -helical content when compared to the ones recorded at 37°C . In an attempt to follow protein refolding, the samples were cooled back to 4°C (Fig. 2; blue dashed lines). With respect to the spectra at 96°C , some CD signal is recovered for most of the variants, indicative of a partial recovery of secondary structure. However, the spectral shape is still far from the native, purely α -helical one. Renaturation curves were incompatible with the two-state transition model and did not overlap with the corresponding denaturation curves (results not shown).

Ca^{2+} binding to GCAP1 variants

Competition with the 5,5'-Br₂-BAPTA chelator was monitored spectroscopically to follow the binding of Ca^{2+} ions to GCAP1. The titration curves obtained for all the variants are shown in Fig. 3. Overall, Ca^{2+} binding to GCAP1 appeared to be sequential and to occur with no apparent cooperativity. All the variants were found to bind Ca^{2+} to some extent, as is clearly shown by comparing to the simulated curve for a case where the Ca^{2+} affinity is several orders of magnitude lower for the proteins than for the chelator (Fig. 3, upper panel). All data sets were reasonably fitted by a model in which the protein has three Ca^{2+} -binding sites (see “Materials and methods”), and the value of the binding constants could hence be obtained by computer fitting (Table 1). The working conditions under which the Ca^{2+} binding experiments were performed ensure robust and reliable determinations. In fact, the experiments were performed at protein concentrations significantly lower than $100 \mu\text{M}$, which is the lowest concentration at which GCAP1 dimerization was observed [24]. On the other hand, preliminary studies showed that some mutations may change the monomer/dimer equilibrium, but do not cause aggregation of the proteins under these conditions (manuscript in preparation).

Our results are consistent with myrGCAP1 having two high affinity Ca^{2+} binding sites (represented by K_1 and K_2 in Table 1) and one site with much lower affinity (K_3 in Table 1). Indeed, the detected affinity for the third binding site was in most cases similar to or even below the experimental detection limit using 5,5'-Br₂-BAPTA ($\log K_{\text{chelator}} = 5.6$, $K_{\text{Dchel}} = 2.3 \mu\text{M}$). The method measures macroscopic binding constants, and for a case with two sites with identical high affinity sites and one site with much lower affinity, $\log K_1$ will be 0.6 higher than $\log K_2$ [25]. A larger difference

is seen for all variants except one, suggesting that the two high affinity sites vary somewhat in affinity, although the error limits for the individual $\log K$ s are generally larger than for their product, which can be determined with considerably higher accuracy. Hence, we report also as an apparent affinity the average of the logarithms of the first two macroscopic binding constants ($(\log K_1 + \log K_2)/2$). Given the much lower affinity for the third site, this apparent affinity is not affected by the poorly accurate information on the low affinity binding site (i.e., $\log K_3$).

Among the GCAP1 variants, the WT myr form showed the highest affinity for Ca^{2+} (Table 1), followed by the nonmyr form, which is observed to bind Ca^{2+} with a ~ 5 -fold lower affinity. Such a difference concerns both high-affinity binding sites, as the first macroscopic dissociation constant (K_{d1}) increased from 4 nM for the myr form to 16 nM for the nonmyr form, while K_{d2} increased from 31 nM for the myr form to 200 nM. All the GCAP1 mutants associated with cone dystrophy showed impaired apparent Ca^{2+} affinity, which for some variants (myrG159V, myrD100E, myrL151F) is even two log units lower, i.e., 100-fold lower affinity than WT (Table 1). The most apparent effect on K_{d1} was observed for the myrD100E mutant ($K_{d1} = 426$ nM), whereas K_{d2} is particularly affected by the myrG159V ($K_{d2} = 20$ μM) and myrD100E ($K_{d2} = 10$ μM) mutations.

Myristoylation affected Ca^{2+} binding to all variants, in some cases enhancing the affinity (WT GCAP1), whereas in some others decreasing it (G159V, L151F).

GCAP1-dependent regulation of GC activity

In order to investigate the functionality of GCAP1 variants and their Ca^{2+} -dependent activation of photoreceptor GC, we performed GC activity assays for the cases that had not been previously analyzed (Fig. 4). These include various mutant forms recently found to be associated with cone dystrophy, i.e., nonmyrG159V, myrL151F, nonmyrL151F and nonmyrE189K. Table 1 compares the data of the present with the previous study in terms of IC_{50} , i.e., the Ca^{2+} -concentration at which the GC activity is half maximal. The tested variants showed IC_{50} values significantly shifted to high Ca^{2+} concentrations compared to the native myrGCAP1 (Table 1, Fig. 4). In particular, nonmyrG159V, myrL151F and nonmyr E89K showed a ~ 50 -fold shift in Ca^{2+} sensitivity, whereas nonmyrL151F showed an impressive ~ 720 -fold shift.

Discussion

Gathering detailed biochemical and biophysical information on the molecular determinants of disease-associated states is essential in order to be able to quantitatively

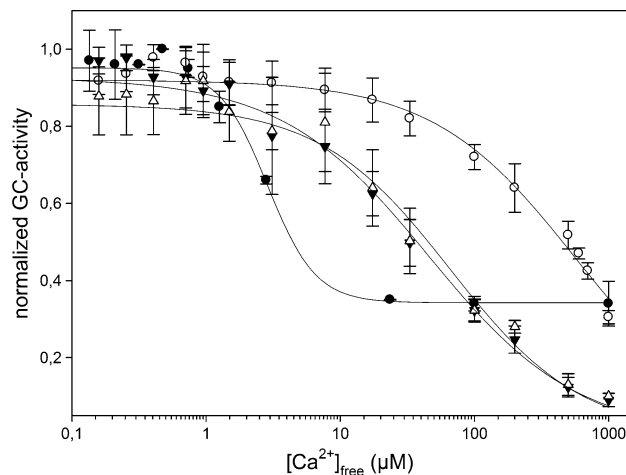


Fig. 4 Regulation of ROS-GC1 by nonmyrGCAP1 variants. Washed bovine rod outer segment membranes containing ROS-GC1 were reconstituted with 10 μM nonmyrGCAP1 (filled circle), nonmyrL151F (open circle), nonmyrE89K (filled inverted triangle) and nonmyrG159V (open triangle) at varying Ca^{2+} concentrations. Guanylate cyclase activity of nonmyrGCAP1 was measured as in ref. [19] and that of the mutants as described in the “Materials and methods.” Activity was calculated as nmol cGMP/min/mg rhodopsin and is expressed as normalized activity

describe phototransduction events in normal and altered conditions, and hence to design effective interventions [6]. In this study we attempted to characterize the effects of Ca^{2+} binding to some GCAP1 variants associated with human cone dystrophies and to investigate the role of myristoylation in both protein stability and GCAP1-mediated activation of GC. To date, ten mutations have been found in the gene encoding GCAP1 that are related to autosomal dominant cone dystrophy (adCD), autosomal dominant rod-cone dystrophy (adCRD) and macular dystrophy (adMD), which have been characterized in previous works [12, 26–32]. Among these mutants, four were recently investigated in a combined clinical and molecular characterization [10]. The same variants of GCAP1 have been further analyzed here in order to highlight common disease-associated hallmarks from a biophysical and functional perspective.

Interestingly, all four mutations analyzed in this study, which are involved in cone dystrophies, are located in the EF3-EF4 domain of GCAP1 (Fig. 1). This region is known to undergo major conformational changes upon Ca^{2+} binding [33], and we followed the structural effects of Ca^{2+} binding by CD spectroscopy. The results showed some common features shared by WT and mutated GCAP1s (Fig. 2). The apo form of each variant exhibited a typical α -helical pattern, indicative of a well-defined secondary structure. Addition of saturating Ca^{2+} increased the intensity of the CD signal, as it was quantified by positive $\Delta\theta_{222\text{s}}$ (Table 1), without altering the shape of the

spectrum. This behavior suggests that in the absence of Ca^{2+} , all GCAP1 variants form a flexible molten globule state, with fully native secondary structure content but lack of stable tertiary structure. Differently from the interpretation of enhanced helical content proposed for GCAP2 [23], we suggest that the increase observed in the CD signal upon addition of Ca^{2+} is a consequence of the protein transition to a more rigid and compact state. The same behavior was observed for other Ca^{2+} -binding proteins, including GCAP2 [34]. Furthermore, this interpretation is compatible with very recent two-dimensional NMR data [24] that show a typical molten-globule spectrum for the apo form and the clear appearance of characteristic EF-hand peaks upon Ca^{2+} saturation. CD spectra also suggest that the presence of the myristoyl group affected the flexibility of the protein in a manner that depended on the point mutation.

The idea that Ca^{2+} -loaded forms are more stable than the respective apo forms was further confirmed by thermal denaturation studies (Fig. 2; Table 1). In every case, but to a different extent depending on the GCAP1 variant, the transition to a less folded state is shifted toward higher temperatures in the presence of Ca^{2+} . This result is in agreement with recent DSC observations by Lim. et al. [24] on WT GCAP1 being stabilized by metal binding. Likewise, we also noticed that myristoylation at the N-terminus had a stabilizing effect on the WT form both in the presence and in the absence of Ca^{2+} (Table 1). In addition, we proved that the effect of acylation on protein stability nevertheless depends on the GCAP1 variant, as both G159V and L151F showed virtually unaltered stabilities independent of myristoylation (Table 1). Interestingly, the change in the transition temperature upon addition of Ca^{2+} , quantitatively determined for most of the variants (Table 1), correlated well with the change in CD signal detected at 222 nm (correlation coefficient $R = 0.92$; see Table 1). This seems to further support our interpretation that the increased signal upon Ca^{2+} saturation is due to GCAP1 gaining a more compact and overall stable conformation rather than to undergoing a change in its secondary structure as postulated before for GCAP2 [23].

In the present study, we measured for WT myrGCAP1 a slightly higher affinity for Ca^{2+} compared to that recently measured by Lim et al. [24] by isothermal titration calorimetry (ITC). The ITC data of Lim et al. [24] are compatible with two nanomolar affinity binding sites ($K_{\text{dEF3}} = 80$ nM and $K_{\text{dEF4}} = 200$ nM) and one micromolar affinity site ($K_{\text{dEF2}} = 0.9$ μM) when measured in the absence of Mg^{2+} . When the constants measured in the present study for WT myrGCAP1 are given as macroscopic K_{d} s, the values are $K_{\text{d}_1} = 4$ nM, $K_{\text{d}_2} = 31$ nM and $K_{\text{d}_3} = 1.2$ μM . The fact that we measured macroscopic binding constants as well as differences in the running buffer and

the method employed to measure the initial protein concentration could account for the slight divergence with respect to ITC determinations. One of the advantages offered by the chelator method employed in this study is its proven precision when comparing between protein sets (e.g., WT and mutants) [16, 17]. Hence, the differences in apparent affinities for the GCAP1 variants listed in Table 1 are expected to be highly reliable.

The impact of disease-related mutations was further analyzed by referring to the modeled protein structure of myr human GCAP1. Thus, we restrict this aspect of our discussion to the myristoylated GCAP1 mutants. When the analysis is limited to the myristoylated forms, the impaired ability of binding Ca^{2+} observed experimentally for all the cone dystrophy-associated mutants for WT GCAP1 could be partially explained by analyzing the modeled protein structure (Fig. 1B). The 3D structure of the Ca^{2+} -loaded myristoylated form of chicken GCAP1 was resolved recently by X-ray crystallography at 2.0 Å resolution (PDB entry: 2R2I). In contrast to recoverin (Rec), which upon Ca^{2+} -binding exposes the acyl group to the solvent, GCAP1 keeps the myristoyl group buried inside the protein milieu, hence providing evidence for no Ca^{2+} -mediated switching [13]. This result confirms previous surface plasmon resonance studies that highlighted an influence of the myristoyl group on the Ca^{2+} -dependent regulation of ROS-GC1 by GCAP1, but no Rec-like switching for GCAP1 and GCAP2 [14]. A structural comparison between myrGCAP1, nonmyrGAP2 and nonmyrGAP3 [35] reveals that the overall architecture and conformation of the Ca^{2+} -binding EF hands are conserved independent of the acylation, but the structure of the N- and C-terminal helices is severely affected in the nonmyristoylated proteins [13]. In detail, the absence of the myristoyl group was found to cause a severe destabilization of the C-terminal helix in nonmyrGAP2, which could not be resolved by NMR spectroscopy, as well as a considerably increased flexibility of the N-terminal region [34]. Overall, these structural findings suggest that the N-terminal myristoylation of GCAPs is important for maintaining the physiological conformation of the N- and C-terminal helices, which is finally required for functional interaction with ROS-GC1 [14, 36].

The analysis of protein structure further reveals some specific features of the sites that are targeted by mutagenesis in the cone-dystrophy conditions. In this respect, G159 was found to be disease-associated when mutated into V [10]. The native G159 lies at the border between the two helices making up the EF4 motif (Fig. 1b), and a mutation to V is expected to perturb sterically the tight interaction with I143 in the facing helix. This perturbation might propagate and eventually affect the Ca^{2+} -binding process. L151 is tightly packed in the core between EF3 and EF4 (Fig. 1b), in a hydrophobic cleft formed by F156, F140 and

again I143. The observed mutation to F is likely to modify the hydrophobic interactions and finally destabilize the whole EF3-EF4 domain as F140 would need to drastically change its conformation in order to accommodate the new side chain. The perturbation would likely affect EF4 wholly, with a possible repercussion on the Ca^{2+} binding in its loop. D100 is directly involved in Ca^{2+} coordination in the loop of EF3 (Fig. 1). While conserving the net charge, mutation to E is expected to affect the Ca^{2+} binding. The site has evolved to accommodate D, and the larger E may be sterically disfavored, leading to suboptimal Ca^{2+} coordination. Finally, E89 is totally solvent-exposed (Fig. 1a) and is not involved in significant short-range electrostatic interactions. The closest intra-helical interaction is with R172, which is stabilized by a stronger coulombic interaction with D168; such interaction is in fact expected to be weak (E89-R172 distance = 13 Å). The effect on Ca^{2+} binding, structure and stability is therefore expected to be limited and exerted through purely electrostatic effects.

The spectroscopic titration method to follow the competition with a chelator also allows very accurate detection of cooperativity mechanisms in Ca^{2+} -binding proteins (see, for example, [17, 18, 37, 38]). The unique advantage of this technique is that, different from other approaches that are based on chemical modifications of the protein, it allows detecting and quantifying cooperativity mechanisms without perturbing the system. The shape of the titration curve indeed contains the information on interaction between binding sites and hence on potential cooperativity. It has been demonstrated that positive cooperativity and sequential binding result in S-shaped curves of opposite curvature [37], hence making a clear distinction between the two mechanisms possible. In none of the cases tested here did we observe apparent cooperativity for Ca^{2+} binding to GCAP1 (Fig. 3), but rather our data are consistent with sequential binding. This is in contrast with what has been very recently reported by Lim et al. [24], who observed positive cooperativity in Ca^{2+} binding between EF3 and EF4, differently from a previous work in which the overall Ca^{2+} -binding isotherm of GCAP1 obtained by analysis of fluorescent chelator binding showed no evidence of cooperativity [39]. The observed difference might be due to the mutants employed in the study to disable specific binding sites, which may perturb the system and affect the measured change in the free energy of Ca^{2+} binding [24]. Double mutants that selectively impair Ca^{2+} binding to EF2, EF3 or EF4 could have unpredictable structural and thermodynamic effects on the other EF hands and affect their Ca^{2+} affinity, as the present study clearly shows. Therefore, while the Ca^{2+} -mediated activation process of ROS-GC1 by GCAP1 definitively shows the features of cooperativity in Ca^{2+} binding [40], our data strongly suggest that metal binding to GCAP1 alone is not cooperative.

Surprisingly, we could not find any significant correlation between the apparent Ca^{2+} affinity and IC_{50} values for the set of mutants studied here. No correlation was found also when focusing on individual binding constants, even when the set of previously analyzed mutants (see ref. [10]) was included in the analysis. The interpretation of this finding is not trivial, as the information contained in the IC_{50} values summarizes a rather complex series of events. As discussed and suggested previously [24, 39, 41], the activation step of ROS-GC1 by GCAP1 would include: (1) dissociation of Ca^{2+} from GCAP1 while it is bound to ROS-GC1; (2) an induced conformational change in GCAP1 that is transferred to the catalytic domain of ROS-GC1; (3) a rearrangement of the dimer interface in ROS-GC1 leading to an accelerated cGMP synthesis. The concerted action of these sequential steps is mirrored in the cooperativity of ROS-GC1 activation [40], but Ca^{2+} binding to isolated GCAP1 occurs without cooperativity. Furthermore, GCAP1 might display different Ca^{2+} affinities when GCAP1 is bound to ROS-GC1. This phenomenon is not uncommon and has been observed with calmodulin and other Ca^{2+} -sensor proteins and their corresponding targets [42]. These issues, however, deserve future investigation.

Finally, the minor stabilizing effect of myristoylation in WT GCAP1 (Fig. 2; Table 1) is in agreement with previous studies [13] as well as with the effect of acylation on the control of the Ca^{2+} sensitivity [14, 15, 43]. These results suggest an allosteric mechanism of communication between the Ca^{2+} -binding sites and the myristoyl moiety without a switch mechanism [14].

In conclusion, the results of this study support the idea that disease-associated forms of GCAP1 are less stable and have lower affinity for Ca^{2+} than the native form. Although they can be still functional, they would require a much higher Ca^{2+} concentration to inhibit the cyclase, a phenomenon that is even more apparent for the nonmyristoylated forms. Hence, in vivo, these forms are expected to maintain ROS-GC1 in a state of constitutive activity, eventually driving the photoreceptor to degeneration.

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