

Biosynthesis of agmatine in isolated mitochondria and perfused rat liver: studies with ^{15}N -labelled arginine

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An important but unresolved question is whether mammalian mitochondria metabolize arginine to agmatine by the ADC (arginine decarboxylase) reaction. ^{15}N -labelled arginine was used as a precursor to address this question and to determine the flux through the ADC reaction in isolated mitochondria obtained from rat liver. In addition, liver perfusion system was used to examine a possible action of insulin, glucagon or cAMP on a flux through the ADC reaction. In mitochondria and liver perfusion, ^{15}N -labelled agmatine was generated from external ^{15}N -labelled arginine. The production of ^{15}N -labelled agmatine was time- and dose-dependent. The time-course of $[\text{U-}^{15}\text{N}_4]\text{agmatine}$ formation from 2 mM $[\text{U-}^{15}\text{N}_4]\text{arginine}$ was best fitted to a one-phase exponential curve with a production rate of approx. $29 \text{ pmol} \cdot \text{min}^{-1} \cdot (\text{mg of protein})^{-1}$. Experiments with an increasing concentration (0–40 mM) of $[\text{guanidino-}^{15}\text{N}_2]\text{arginine}$ showed a Michaelis constant K_m for arginine of 46 mM and a V_{max} of

$3.7 \text{ nmol} \cdot \text{min}^{-1} \cdot (\text{mg of protein})^{-1}$ for flux through the ADC reaction. Experiments with broken mitochondria showed little changes in V_{max} or K_m values, suggesting that mitochondrial arginine uptake had little effect on the observed V_{max} or K_m values. Experiments with liver perfusion demonstrated that over 95% of the effluent agmatine was derived from perfusate $[\text{guanidino-}^{15}\text{N}_2]\text{arginine}$ regardless of the experimental condition. However, the output of ^{15}N -labelled agmatine ($\text{nmol} \cdot \text{min}^{-1} \cdot \text{g}^{-1}$) increased by approx. 2-fold ($P < 0.05$) in perfusions with cAMP. The findings of the present study provide compelling evidence that mitochondrial ADC is present in the rat liver, and suggest that cAMP may stimulate flux through this pathway.

Key words: arginase, arginine decarboxylase, cAMP, glucagon, insulin, ornithine.

INTRODUCTION

An important but unresolved question is whether agmatine is synthesized in mammalian mitochondria by the ADC (arginine decarboxylase) reaction. Agmatine, which is widely distributed in mammalian tissue [1–4], may have a role as a hormone and/or polyamine precursor for multiple metabolic functions. Numerous studies have shown that agmatine affects a number of physiological processes, including the regulation of intracellular polyamine levels and cellular proliferation, regulation of neurotransmitter receptors, inhibition of NO synthesis, release of insulin from β -cells and regulation of body nitrogen economy secondary to its action on hepatic ureagenesis [5–10]. Therefore it is important to determine whether agmatine is formed in mammalian tissue or exclusively derived from food intake and/or intestinal flora.

The presence of the ADC reaction has been well established in bacteria and plants [11,12]. A similar reaction in mammalian mitochondria was first identified in 1994 [5]. Since then, it has been shown that agmatine is present in several mammalian tissues and that the ADC reaction is active in various tissues and organs [2,3,5,6,10]. In addition, ADC has been partially cloned in the rat kidney [13], and a human cDNA clone exhibiting ADC activity has been described in [14]. Notwithstanding these reports demonstrating the presence of the ADC reaction in mammalian tissues, Pegg and co-workers [15] detected no ^{14}C -labelled agmatine after incubation of mouse or rat mitochondrial extract with $[\text{U-}^{14}\text{C}]\text{arginine}$. These authors argued that earlier studies demonstrating ADC activity were erroneous, and that the apparent presence of agmatine in mammalian tissue may correspond

to bacterial contamination, food intake and/or intestinal flora [15].

A major impediment in the study of ADC activity and/or measurement of [agmatine] has been the lack of a precise and specific tool for determining agmatine formation. The use of $^{14}\text{CO}_2$ release from ^{14}C -labelled arginine cannot differentiate between $^{14}\text{CO}_2$ released by the ADC reaction, the ornithine decarboxylase reaction, or mitochondrial oxidation of ornithine. Furthermore, the use of HPLC for determination of [agmatine] is somewhat problematic due to the low tissue [agmatine] and its poor separation from other amino acids present in the tissue [1,4]. The reported concentration of agmatine as determined by HPLC has ranged between pmol/g wet weight to $\mu\text{mol/g}$ wet weight [1–4,10]. However, the use of ^{15}N and GC–MS methodology allows determination of the true product of the ADC reaction as well as the precise tissue agmatine level. With ^{15}N -labelled arginine as a precursor and GC–MS as an analytical tool, one can precisely and specifically determine the production of ^{15}N -labelled agmatine by the ADC reaction. In addition, the isotope dilution approach allows for an accurate determination of [agmatine] in the picomolar range.

In the present study, we took advantage of ^{15}N and GC–MS methodology to characterize the kinetic parameters of ADC in isolated hepatic mitochondria of overnight fasted rats. In addition, a liver perfusion system was used to examine a possible action of insulin, glucagon or cAMP on flux through the ADC reaction. The results demonstrate that ^{15}N -labelled agmatine was formed in experiments with isolated mitochondria or during liver perfusions with ^{15}N -labelled arginine. The results suggest that cAMP may stimulate flux through the ADC reaction.

Abbreviations used: ADC, arginine decarboxylase; HFAA, hexafluoroacetylacetone; PKA, protein kinase A.

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MATERIALS AND METHODS

Experiments with isolated mitochondria

Mitochondria were isolated from the liver of overnight fasted rats by differential centrifugation as described previously [16]. Briefly, the liver of an anaesthetized rat was cannulated through the portal vein and rinsed with 0.9% NaCl solution (4 °C), excised and weighed. The minced liver was homogenized in a glass Potter–Elvehjem homogenizer with a Teflon® pestle in 12.5 volumes of cooled (4 °C) isolation buffer consisting of (mM) mannitol (225), sucrose (75), EGTA (1) and Hepes (5), pH 7.4. The mitochondrial pellet was gently resuspended with isolation medium to yield 50–80 mg of protein/ml. All manipulations were performed in a cold room and the mitochondrial suspension was kept on ice.

The basic incubation medium consisted of (mM) Tris (50), KCl (5), MgCl₂ (5), KHCO₃ (15), KH₂PO₄ (5), α -ketoglutarate (1), succinate (5), ATP (5), octanoic acid (2) and EDTA (2), pH 7.4. Mitochondrial suspensions (2 ml; ~3 mg of mitochondrial protein/ml) were incubated at 30 °C for the times indicated and with the addition of ¹⁵N-labelled arginine and other modulators. In the first series of experiments, we sought to determine the time-course of agmatine production from [U-¹⁵N₄]arginine. The level of arginine in mitochondria isolated from overnight fasted rats was 1.34 ± 0.39 nmol/mg of protein (mean \pm S.D.; $n = 13$ mitochondrial preparations from separate rats during a period of six months). Assuming a matrix volume of 1 μ l/mg of protein [17], mitochondrial [arginine] is approx. 1.34 ± 0.39 mM. Therefore whole mitochondria were incubated with 2 mM [U-¹⁵N₄]arginine, with or without 2 mM unlabelled ornithine, to determine a possible effect of cytosolic ornithine on mitochondrial production of agmatine, as previously indicated in [2].

To determine the Michaelis constant K_m for arginine and the V_{max} of the flux through the ADC reaction, we used the initial linear stage of ¹⁵N-labelled agmatine formation obtained in time-course experiments, indicated above. To this end, mitochondria were incubated for 7 min at 30 °C, with an increasing concentration (0–40 mM) of [guanidino-¹⁵N₂]arginine. The production of [guanidino-¹⁵N₂]agmatine was used to determine the K_m and the V_{max} values.

The above time-course incubations and experiments with an increasing concentration of [guanidino-¹⁵N₂]arginine were performed in the presence of 2 mM EDTA. However, it has been shown that free Ca²⁺ inhibits the ADC reaction [2]. Furthermore, arginine is a substrate for both ADC and arginase (arginase-I and/or arginase-II) [18,19]. Arginase may compete with ADC for arginine, and thus diminish the production of agmatine. It has been demonstrated that Mn²⁺ stimulates the activity of arginases (arginase-I and -II) [20]. Therefore to determine a possible effect of the mitochondrial bound arginase or mitochondrial free Ca²⁺ on the flux through the ADC reaction, an additional series of experiments was performed with 0.1 mM Mn²⁺ without EDTA. The addition of Mn²⁺ is expected to stimulate the arginase reaction and the omission of EDTA to preserve mitochondrial free [Ca²⁺].

From each of the incubations outlined above, an aliquot (100 μ l) was taken for protein determination and the incubation was stopped with 100–150 μ l of HClO₄ (60%). The deproteinized extracts were neutralized and assayed for agmatine.

Experiments with broken mitochondria

The measured K_m for arginine in experiments with intact mitochondria may be affected by mitochondrial uptake of arginine. To address this possibility, experiments with an increasing concentration of [guanidino-¹⁵N₂]arginine were repeated with broken mitochondria. Because mitochondrial matrix is required for ADC

activity [2] experiments with broken mitochondria will preserve mitochondrial matrix and ADC activity. Broken mitochondria were prepared from stock of intact mitochondrial pellet, subjected to three cycles of freezing in liquid nitrogen and thawing as described in [21]. The resulting mixture of matrix and mitochondrial membranes was suspended with a basic incubation medium (as indicated above), at a protein concentration of approx. 4 mg/ml. Then 2 ml of the suspension containing the matrix and mitochondrial membranes was incubated with an increasing concentration of [guanidino-¹⁵N₂]arginine similar to the experiments with whole mitochondria.

Measurement of mitochondrial respiration

Oxygen consumption in experiments with whole mitochondria was measured at 26 °C using a Clark electrode and the Oxygen Measuring System (Instech SYS203, Plymouth Meeting, PA, U.S.A.). Respiratory control was determined in each batch of mitochondrial preparation according to Chance and Williams [22]. An aliquot of mitochondrial preparation was added to a polarographic cell containing basic incubation medium with only 5 mM succinate and 1 mM α -ketoglutarate. The respiration rate at state two (V₂) was recorded when O₂ consumption became linear. Then 0.3 mM ADP was added to establish the respiration rate at state three (V₃). When most of ADP was converted to ATP, respiration rate at state four (V₄) was recorded. In most cases, oxygen consumption was 2–3, 9–12 and 2–3 nmol of O₂ · min⁻¹ · (mg of protein)⁻¹ for state two, three and four respectively, and V₃/V₂ or V₃/V₄ ratio was between 3 and 4. All experiments were performed with whole mitochondria having a V₃/V₂ ratio greater than 3. However, broken mitochondria had minimal or no oxygen consumption.

Experiments with liver perfusions

Livers from overnight fasted male rats were perfused in the non-recirculating mode as previously described in [10,23]. Briefly, the basic perfusion medium was Krebs saline continuously gassed with 95% O₂/5% CO₂ and containing lactate (2.1 mM) and pyruvate (0.3 mM) as metabolic fuels. The flow rate (3–3.5 ml · min⁻¹ · g⁻¹), pH and pO₂ (in influent and effluent media) were monitored throughout, and oxygen consumption was calculated. After 20 min of preperfusion, the basic perfusate was replaced by perfusate that contained, in addition to the lactate and pyruvate, precursors for urea nitrogen (i.e., 0.3 mM NH₄Cl and 1 mM glutamine), and 0.5 mM L-[guanidino-¹⁵N₂]arginine in the absence (control) or presence of 10⁻⁷ M insulin, glucagon or 10⁻⁴ M dibutyryl-cAMP, a permeable cAMP analogue. Samples were taken from the influent and effluent media for chemical and ¹⁵N-GC-MS analyses. At the end of the perfusion, the liver was freeze-clamped, treated with HClO₄, and metabolite measurements were done in neutralized extracts as indicated [10,23].

To determine the baseline concentration of hepatic agmatine, livers from a separate group of overnight fasted rats were rinsed for 5 min with cold (4 °C) saline, after which livers were freeze-clamped, treated with HClO₄ and [agmatine] was determined by spiking the extract (300 μ l) with [guanidino-¹⁵N₂]agmatine. The latter was prepared as described previously [10]. Determination of [agmatine] after liver perfusion with L-[guanidino-¹⁵N₂]arginine was accomplished as described below.

GC-MS methodology and determination of ¹⁵N-labelled agmatine

Isotopic enrichment in ¹⁵N-labelled agmatine was measured with GC-MS using a modification of a method described previously

[24]. To an aliquot of $\sim 300 \mu\text{l}$ of mitochondrial extract, $150 \mu\text{l}$ of $4 \text{ M NH}_4\text{OH}$ was added. Thereafter, the sample was loaded into an acetate 1-X8 Dowex resin column and agmatine was eluted with $3\text{--}4 \text{ ml}$ of H_2O and dried down under a stream of N_2 gas. Then $50 \mu\text{l}$ of HFAA (hexafluoroacetylacetone) was added to vials containing the dry sample, vortex-mixed and heated at 110°C for 60 min. The reaction mixture was cooled to room temperature ($\sim 22\text{--}25^\circ\text{C}$), dried under a stream of N_2 gas and dissolved with 1 ml of 3 M HCl . The HFAA derivative of agmatine was extracted with 1 ml of ethyl acetate. The extract was dried under a stream of N_2 gas and dissolved in $75 \mu\text{l}$ of ethyl acetate. Approximately $2\text{--}4 \mu\text{l}$ of ethyl acetate extract of the HFAA-agmatine derivative was injected into the GC-MS system for analysis.

Samples were analysed by either a GC-MS Agilent System (6890 GC-5973 Mass Selective Detector) or a Hewlett-Packard MSD (HP-5970). MS was performed using electron impact (EI) ionization with an ionizing voltage of -70 eV and an electron multiplier set to 2000 V . The GC injector temperature was set at 250°C and transfer line at 280°C . The usual GC temperature program was: 80°C for 2 min, and then $30^\circ\text{C}/\text{min}$ up to 300°C . The capillary column was a SUPELCOWAXTM-10, $15 \text{ m} \times 0.32 \text{ mm} \times 0.25 \mu\text{m}$ film thickness. ^{15}N enrichment in the HFAA derivative of [guanidino- $^{15}\text{N}_2$]agmatine was monitored using the m/z ratio 286/284 and, for [U- $^{15}\text{N}_4$]agmatine, using the m/z ratio 287/284.

Samples obtained from experiments with liver perfusions or mitochondrial incubations were analysed for [agmatine] by isotope dilution as follows. First, the initial (I_1) isotopic enrichment in ^{15}N -labelled agmatine was determined. Secondly, a parallel set of the same samples was spiked with a known amount of unlabelled agmatine (d), and a second measurement (I_2) of ^{15}N -labelled agmatine was taken. [Agmatine] was calculated as described in [25]. Amino acids were determined with HPLC using precolumn derivatization with *o*-phthalaldehyde [26].

Calculations and data analyses

Data obtained from mitochondrial incubation were analysed with GraphPad Prism-4 software for linear and nonlinear curve fitting. The production of [guanidino- $^{15}\text{N}_4$]agmatine during the course of incubations with L-[U- $^{15}\text{N}_4$]arginine was best fitted to a one-phase exponential association [$Y = Y_{\text{max}} \cdot (1 - e^{-kt})$], and the flux through the ADC reaction [$\text{nmol} \cdot \text{min}^{-1} \cdot (\text{mg of protein})^{-1}$] was given by the product of $Y_{\text{max}} \cdot k$. The GraphPad Prism-4 software was also used in experiments with an increasing concentration of arginine to determine the best curve fit and to calculate the V_{max} for the flux through the ADC reaction and the K_m for arginine.

During liver perfusions, the rate of agmatine output was determined by the measurement of agmatine concentration in the effluent (nmol/ml), normalized to the flow rate (ml/min) and liver wet weight, as described previously [10,22]. The output of [guanidino- $^{15}\text{N}_2$]agmatine was calculated from the product of ^{15}N enrichment, at.% excess/100 times concentration [$\text{nmol} \cdot \text{min}^{-1} \cdot (\text{g wet weight})^{-1}$] and is expressed ($\text{nmol of [guanidino-}^{15}\text{N}_2\text{]-agmatine} \cdot \text{min}^{-1} \cdot (\text{g wet weight})^{-1}$). Flux through the ADC reaction is represented by the output in the effluent of [guanidino- $^{15}\text{N}_2$]agmatine during the course of perfusion with L-[guanidino- $^{15}\text{N}_2$]arginine.

Each series of experiments with isolated mitochondria was repeated 4–6 times, and 3–4 times with liver perfusion, as outlined above. Statistical analysis was performed using In-STAT 1.14 software for the Macintosh. Student's *t* test or ANOVA was employed to compare two groups or differences among groups as needed. $P < 0.05$ was taken as indicating a statistically significant difference.

RESULTS

Measurement of agmatine: analytical consideration

Figure 1 illustrates a typical GC-MS chromatogram of the HFAA-agmatine derivative obtained from mitochondrial incubation or freeze-clamped liver extract. This representative chromatogram demonstrates an excellent separation of agmatine from other tissue constituents. There is a high signal-to-noise ratio for the ions at m/z 286 and 284 even with as little as 2 pmol of agmatine. The chromatogram in Figure 1(A) corresponds to the injection of $2\text{--}3 \mu\text{l}$ of the HFAA-derivative containing approx. $2\text{--}4 \text{ pmol}$ agmatine.

The EI ionization of the HFAA-agmatine derivative shows a major ion at m/z 284 for unlabelled agmatine. This ion contains three N due to the cleavage of one N (the original $\alpha\text{-N}$ of arginine) [25]. Thus [guanidino- $^{15}\text{N}_2$]agmatine and [U- $^{15}\text{N}_4$]agmatine are monitored at m/z 286 and 287 respectively. Multiple measurements indicated that the natural abundance of m/z 286/284 and 287/284 ratios is 6 and 0.9% respectively. This relatively low natural background value allows for precise measurement of ^{15}N enrichment ($< 1 \text{ at.}\%$ excess) in agmatine.

Production of agmatine in isolated mitochondria

To determine whether ADC activity exists in mitochondria isolated from rat liver, an initial series of experiments was designed to determine the time-course of ^{15}N -labelled agmatine production during incubations with a physiological level of mitochondrial arginine. Figure 2 illustrates the time-course of [U- $^{15}\text{N}_4$]agmatine production, which was best fitted to a one-phase exponential curve ($r^2 = 0.87$). Production of ^{15}N -labelled agmatine increased almost linearly up to 15 min and then reached a plateau between 15 and 30 min. Data points presented in this curve were used to calculate the rate constant (k) and flux through the ADC reaction. The means \pm S.E.M. for these parameters are: $0.11 \pm 0.05 \text{ min}^{-1}$ and $29 \pm 5.1 \text{ (pmol} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}\text{)}$ for k and flux through the ADC reaction respectively. Inclusion of 2 mM ornithine had little effect on the time-course of [U- $^{15}\text{N}_4$]agmatine production.

The next series of experiments was designed to characterize the K_m for arginine and V_{max} of the ADC reaction. Isolated mitochondria were incubated for 7 min with an increasing concentration of [guanidino- $^{15}\text{N}_2$]arginine (Figure 3A). Half-maximal arginine decarboxylation (the K_m for arginine) was achieved at $46 \pm 18.8 \text{ mM}$ with a V_{max} of $3.74 \pm 0.94 \text{ nmol} \cdot \text{min}^{-1} \cdot (\text{mg of protein})^{-1}$. The presence or absence of EDTA and/or Mn^{2+} had little effect on these kinetic parameters. In experiments with broken mitochondria, the V_{max} and K_m for arginine were $3.8 \pm 1.7 \text{ nmol} \cdot \text{min}^{-1} \cdot (\text{mg of protein})^{-1}$ and $28.2 \pm 17.8 \text{ mM}$ respectively (Figure 3B). Therefore arginine transport into mitochondria is not rate limiting in the production of agmatine, and may have little or no effect on the observed kinetic parameters of ADC in experiments with whole mitochondria.

Hormonal regulation of the ADC reaction in liver perfusion system

After the demonstration that agmatine is formed from arginine in mitochondria isolated from rat liver, experiments with the liver perfusion system were performed to determine the action of insulin, glucagon or dibutyryl-cAMP on the flux through the ADC reaction. Results in Figure 4 (panel A) demonstrate that over 95% of the effluent agmatine was derived from perfusate [guanidino- $^{15}\text{N}_2$]arginine, regardless of the experimental conditions. However, the output of ^{15}N -labelled agmatine (Figure 4, panel B) increased by approx. 30% ($P > 0.05$) in perfusions with glucagon or insulin, i.e., 7.9 ± 3.9 , 10.6 ± 5.5 and $13.3 \pm 5.3 \text{ nmol} \cdot \text{min}^{-1} \cdot \text{g}^{-1}$ in control, glucagon or insulin respectively (mean \pm S.D.), and

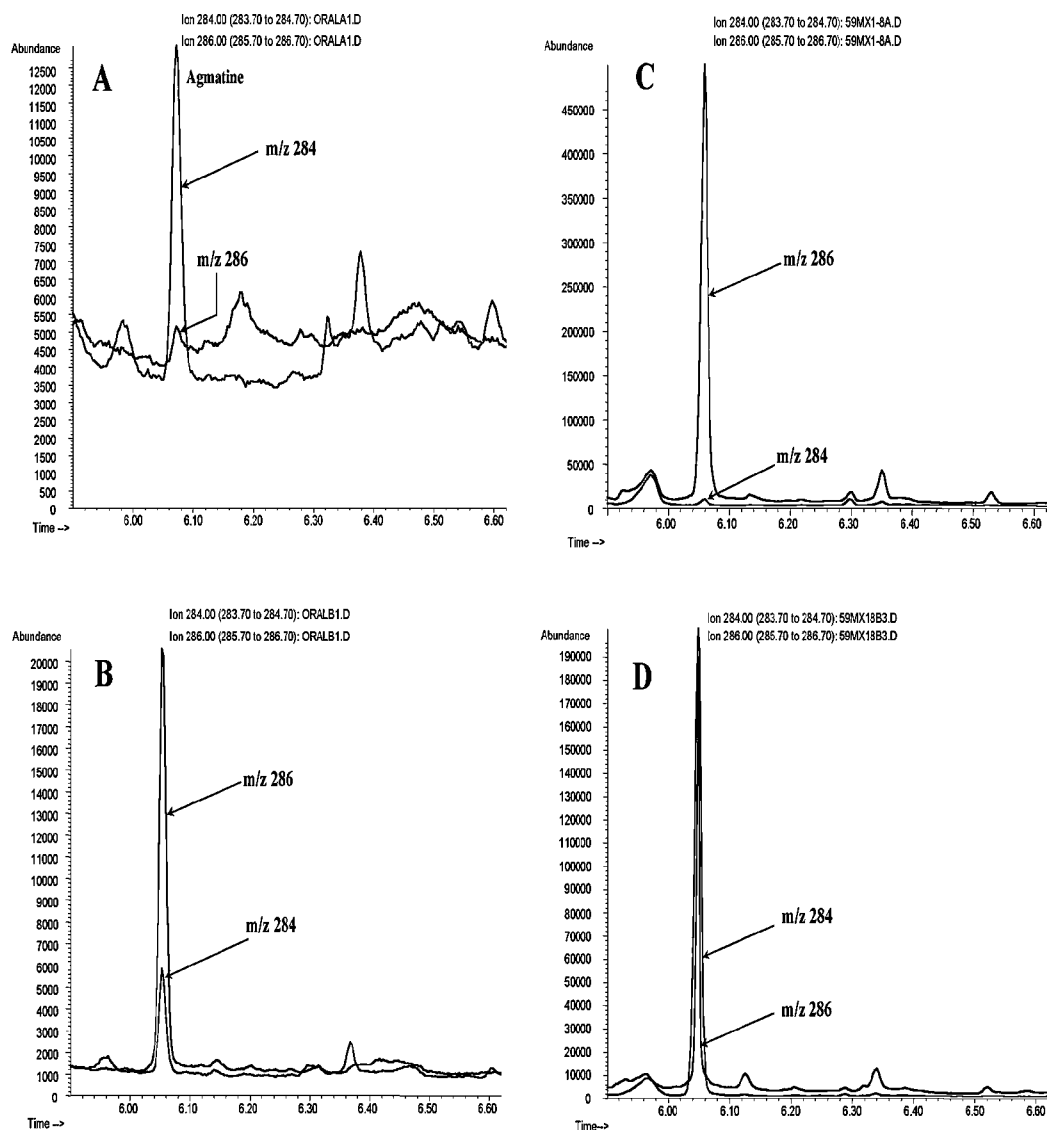


Figure 1 A representative GC–MS chromatogram of single ion monitoring of the HFAA derivative of agmatine

(A) A freeze-clamped liver extract from an overnight fasted rat. The chromatogram corresponds to approx. 2 pmol of agmatine (m/z 284). (B) Same extract shown in (A) after spiking with [guanidino- $^{15}\text{N}_2$]agmatine (2 nmol/g). (C) Mitochondrial extract after incubation for 7 min with 10 mM [guanidino- $^{15}\text{N}_2$]arginine. Peak corresponds to approx. 4 pmol of [guanidino- $^{15}\text{N}_2$]agmatine (m/z 286). (D) Same extract shown in (C) after the addition of unlabelled agmatine (~ 1.2 nmol/3 mg of protein). The m/z ratio 286/284 was used to determine the isotopic enrichment as well as the concentration of agmatine by isotope dilution.

significantly ($P < 0.05$) increased in perfusions with cAMP (19.03 ± 4.4 nmol \cdot min $^{-1}$ \cdot g $^{-1}$). These observations suggest that insulin and glucagon may have insignificant effect, whereas cAMP may stimulate the flux through the ADC reaction.

Measurement of ^{15}N enrichment in [guanidino- $^{15}\text{N}_2$]agmatine in freeze-clamped livers at the end of perfusions with [guanidino- $^{15}\text{N}_2$]arginine demonstrates that approx. 46, 54, 56 and 78 % of agmatine in the liver was in the form of [guanidino- $^{15}\text{N}_2$]agmatine, in control, perfusions with glucagon, insulin or dibutyl-cAMP respectively. Furthermore, the baseline agmatine content in freeze-clamped liver extracts obtained from overnight fasted rats (without perfusion) was 1.2 ± 0.6 nmol/g ($n = 3$). This value increased to 6.1 ± 2.2 , 9.1 ± 3.7 , 3.1 ± 1.9 and 8.2 ± 0.9 nmol/g in livers perfused with ^{15}N -labelled arginine without hormone ($n = 4$), in perfusion with insulin ($n = 3$), with dibutyl-cAMP ($n = 3$) or in perfusions with glucagon ($n = 3$) respectively. These observations illustrate that hepatic [agmatine] increased by approx.

4-fold after perfusion with 0.5 mM arginine. The apparent differences between control and perfusions with dibutyl-cAMP, insulin or glucagon are statistically insignificant. The formation of ^{15}N -labelled agmatine during liver perfusions and the increased [agmatine] in liver extract must have occurred after decarboxylation of the perfusate ^{15}N -labelled arginine. These findings, together with the above results obtained from isolated mitochondria, are in agreement with earlier studies demonstrating the presence of the ADC reaction in mammalian tissue [2,3,5,6].

DISCUSSION

The mitochondrial enzyme ADC (EC 4.1.1.19) has long been known to be present in bacteria, invertebrates and plants [11,12, 27]. ADC activity was first recognized in mammalian tissue in 1994 [5]. Thereafter, the activity of ADC has been demonstrated

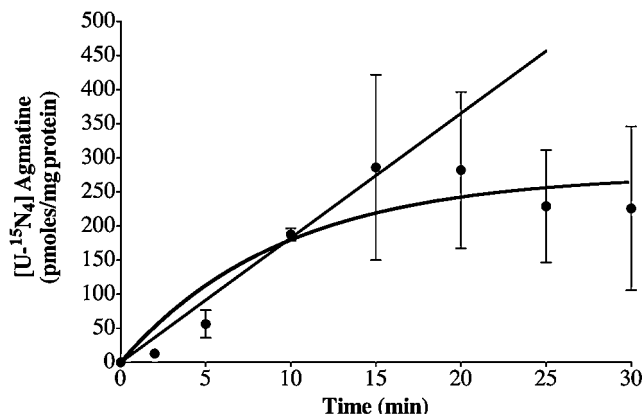


Figure 2 Time-course of ^{15}N -labelled agmatine production during the incubation of whole mitochondria with $[\text{U-}^{15}\text{N}_4]$ arginine

Experiments were performed with $[\text{U-}^{15}\text{N}_4]$ arginine (2 mM) and other metabolites as indicated under the Materials and methods section. The line shows the best fit ($r^2 = 0.87$) of a single-phase exponential association [$Y = Y_{\text{max}} \cdot (1 - e^{-kt})$]. The straight line corresponds to the linear stage of the reaction ($r^2 = 0.97$). ^{15}N -labelled agmatine was determined by the product of ^{15}N enrichment (at.% excess/100) times total amount (pmol/mg of protein). Results are expressed as means \pm S.D. for 4–6 independent experiments.

in a number of tissues and organs [2,3,5,9,10]. The findings of the present study (Figures 1–4) are in accordance with these earlier studies and provide compelling evidence that mitochondrial ADC is present in rat liver. Several observations presented here bear directly on this conclusion.

Mitochondrial incubations with either $[\text{U-}^{15}\text{N}_4]$ - or $[\text{guanidino-}^{15}\text{N}_2]$ -labelled arginine demonstrate: (i) the time-course of ^{15}N -labelled agmatine production (Figure 2); and (ii) the apparent Michaelis–Menten behaviour of ^{15}N -labelled agmatine production with increasing [arginine] (Figure 3). In the liver perfusion system the results demonstrate: (i) approximately 95–100 % of the effluent agmatine was in the form of $[\text{guanidino-}^{15}\text{N}_2]$ (Figure 4), indicating that the perfusate $[\text{guanidino-}^{15}\text{N}_2]$ arginine was the exclusive source of the effluent agmatine; (ii) approximately, 46, 54, 56 and 78 % of agmatine in the liver extracts was in the form of $[\text{guanidino-}^{15}\text{N}_2]$ agmatine, in control, perfusions with glucagon, insulin and dibutyryl-cAMP respectively. The formation of $[\text{guanidino-}^{15}\text{N}_2]$ agmatine can occur only after the decarboxylation of $[\text{guanidino-}^{15}\text{N}_2]$ arginine by the ADC reaction; and (iii) agmatine content in liver extracts at the end of perfusion with $[\text{guanidino-}^{15}\text{N}_2]$ arginine was approx. 3–4-fold higher than the baseline level.

The analyses of the present study were performed with the use of ^{15}N and GC–MS methodology. This approach provides an excellent separation, identification and accurate quantification of agmatine in a picomolar range (Figure 1). Analysis of agmatine with HPLC did not provide a complete separation of agmatine from other tissue constituents. In most cases, the peak of the *o*-phthaldialdehyde derivative of tryptophan overlapped the peak of agmatine (results not shown), resulting in an overestimation of tissue [agmatine]. Purification of the tissue as previously indicated in [1,4] resulted in a poor yield and an underestimation of tissue [agmatine].

In experiments with physiological [arginine] and whole mitochondria, the production of ^{15}N -labelled agmatine was linear between 0 and 15 min (Figure 2). The addition of 2 mM ornithine had little effect on flux through the ADC reaction. Similarly, the omission of EDTA when Mn^{2+} was added to the incubation had little effect on the K_m for arginine or the V_{max} of ADC. Incubations with Mn^{2+} significantly stimulated the production of

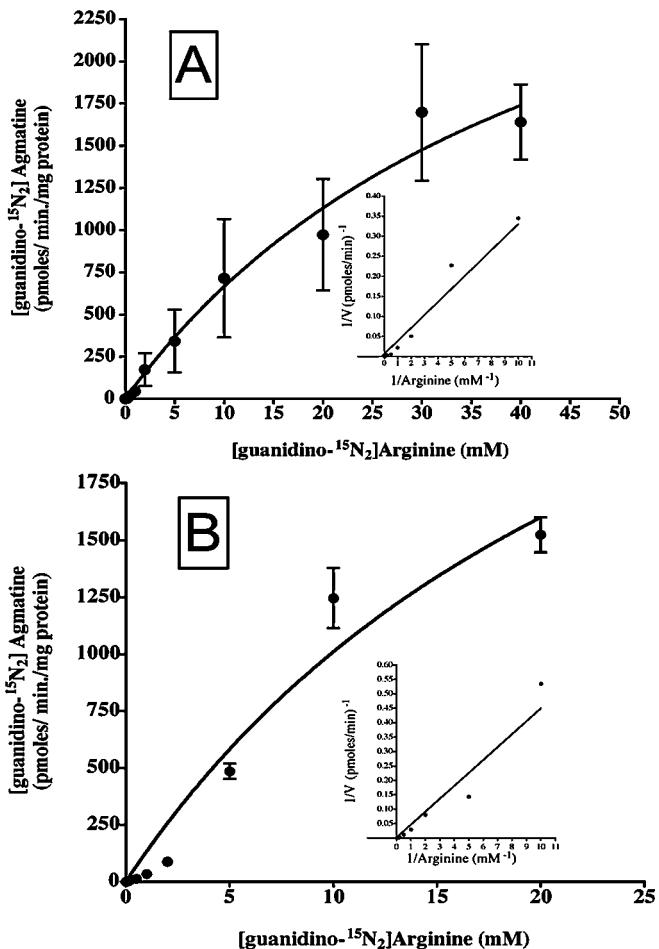


Figure 3 Saturation curves of ^{15}N -labelled agmatine production from $[\text{guanidino-}^{15}\text{N}_2]$ arginine

(A) Experiments with intact mitochondria and (B) experiments with broken mitochondria. Insets represent a Lineweaver–Burk plot of these data. Results were obtained after 7 min incubation of whole or broken mitochondria with increasing [arginine] as indicated. Results are the means ($n = 3$ –5) for the combined results obtained from experiments with the addition of Mn^{2+} (0.1 mM) and no EDTA or with EDTA (2 mM) and without the addition of Mn^{2+} . The lines represent the best fit ($r^2 > 0.90$). Results are expressed as means \pm S.D.

urea by the mitochondrial bound arginase (results not shown), with only little effect on the flux through the ADC reaction. However, the observations demonstrated that 70–80 % of arginine added to the incubation system was catabolized by arginase bound to the outer mitochondrial membrane, even without the addition of Mn^{2+} . Therefore the high value of K_m for arginine may be due to high catabolism of arginine by arginase bound to the outer mitochondrial membrane, thereby diminishing the availability of arginine for the intramitochondrial ADC reaction. The high K_m for arginine (15-fold greater than physiological [arginine]) cannot reflect arginine transport into mitochondria because the K_m was similar whether studies were performed with whole or disrupted mitochondria. Further evidence supporting this conclusion is provided by the lack of effect on the kinetic parameters of ADC in experiments with the addition of ornithine, which may inhibit mitochondrial arginine uptake [28,29]. Furthermore, a high-affinity mitochondrial transporter for arginine was found with a K_m of 0.08 mM and V_{max} of $1.89 \text{ nmol} \cdot \text{min}^{-1} \cdot (\text{mg of protein})^{-1}$ [28], indicating robust mitochondrial arginine uptake compared with the apparent low velocity of the ADC reaction. The present study suggests a low affinity of ADC for arginine, which

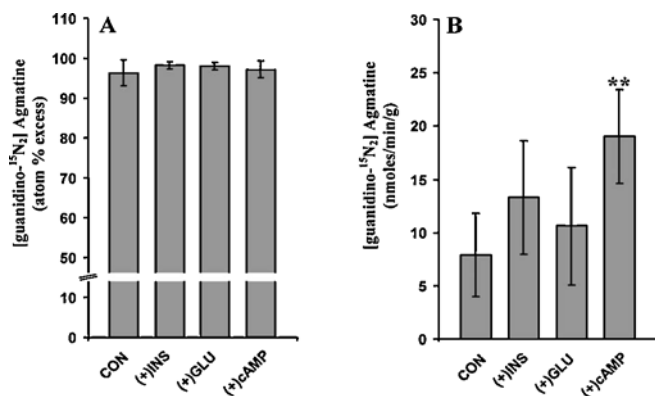


Figure 4 Production and output of ¹⁵N-labelled agmatine from livers perfused with [guanidino-¹⁵N₂]arginine

(A) ¹⁵N enrichment (at.% excess) in effluent agmatine. (B) Output (nmol · min⁻¹ · g⁻¹) of ¹⁵N-labelled agmatine as determined by the product of ¹⁵N enrichment (at.% excess/100) times total output (nmol · min⁻¹ · g⁻¹). Data were obtained from measurements done at the steady-state (i.e. 25–40 min after start of 0.5 mM [guanidino-¹⁵N₂]arginine infusion and other nutrients as indicated under the Materials and methods section without (CON, control) or with the addition of 10⁻⁷ M insulin (+INS), glucagon (+GLU) or 10⁻⁴ M dibutyryl-cAMP. Results are expressed as means ± S.D. for three livers. ***P* < 0.05.

suggests relatively low flux even when [arginine] is 15–20-fold higher than the physiological level. Therefore, with physiological [arginine], the production of agmatine may be negligible compared with the dietary intake of agmatine.

The findings of the present study somewhat differ from those previously reported in [2]. Using mitochondrial membranes obtained from the rat brain, Regunathan and Reis [2] found that the ADC reaction was linear up to 90 min with an initial *K_m* of 0.75 mM and *V_{max}* of 2.22 nmol · h⁻¹ · (mg of protein)⁻¹. A possible explanation for these differences is that in the previous study [2], the kinetic parameters were determined by the formation of ¹⁴CO₂ from [1-¹⁴C]arginine. This approach is less specific for measurement of the ADC activity compared with the ¹⁵N-GC-MS technique employed in the present study. This is especially true when taking into account the observation showing that the ADC reaction also produced ¹⁴CO₂ from [1-¹⁴C]ornithine [2]. In addition, a significant portion of [1-¹⁴C]arginine probably metabolized by mitochondrial bound arginase to form [1-¹⁴C]ornithine. The latter may be oxidized to form ¹⁴CO₂. The sum of ¹⁴CO₂ generated from [1-¹⁴C]arginine and [1-¹⁴C]ornithine may lead to a low *K_m* value for arginine and overestimation of the *V_{max}* for the ADC reaction. Thus a study of ADC kinetic parameters based on ¹⁴CO₂ production may not provide specific and accurate kinetic parameters for the ADC reaction. However, the use of ¹⁵N-labelled arginine as substrate and measurement of ¹⁵N-labelled agmatine would provide an accurate relationship between the [substrate] and [product], and thereby reliable values of *K_m* for arginine and *V_{max}* for the ADC pathway.

In addition, Regunathan and Reis [2] indicated that ADC was inhibited by ornithine or Ca²⁺. In the present study, the omission of EDTA or the presence of 2 mM ornithine had no significant effect on the observed values of *K_m* or *V_{max}*. However, an explanation for the apparent differences between the observations of the present study and previous results is not feasible because: (a) in the earlier study, the effect of Ca²⁺ was evaluated with the addition of superphysiological [Ca²⁺] in experiments with mitochondrial membrane preparations. The present study was performed using whole mitochondria without supplementation of external Ca²⁺. Most of the mitochondrial [Ca²⁺] is in the form of a phosphate salt. Thus the omission or addition of EDTA from the incubation

medium may have little effect on the mitochondrial Ca²⁺ pool when mitochondria were isolated with a medium containing 1 mM EGTA. (b) In the present study, 2 mM ornithine was added to the incubation medium in experiments with whole mitochondria. However, in a previous study [2], aside from the statement that ornithine inhibited ADC activity, there is no information regarding the [ornithine] used, the IC₅₀ for ornithine and the nature of this inhibition.

Similarly, observations of the present study are not in agreement with a recent study by Pegg and co-workers [15], who found only a release of ¹⁴CO₂ but failed to detect [U-¹⁴C]agmatine after incubation of mitochondrial extract with L-[U-¹⁴C]arginine. It was speculated that ¹⁴CO₂ was strictly generated by oxidation of [U-¹⁴C]ornithine produced by the arginase reaction. Although our experiments with [U-¹⁵N₄]arginine demonstrated that ¹⁵N-labelled urea and ornithine were the primary products of mitochondrial arginine catabolism (results not shown), the same experiments also documented the production of [U-¹⁵N₄]agmatine (Figures 2 and 3). As indicated above, the affinity of ADC for arginine is very low. Therefore it is possible that in the earlier study, the [arginine] used (not reported) was too low, and thus the production of [U-¹⁴C]agmatine was below the detection limit of the method used. The finding that “¹⁴CO₂ was produced in substantial amounts” [15], is expected because CO₂ may be formed by ornithine decarboxylase [32], the tricarboxylic acid cycle [30,31] and the ADC reaction [2]. The substantial generation of ¹⁴CO₂ would represent CO₂ generated through all these pathways, including the ADC reaction. Again, as indicated above, studies of ADC activity based on measurements of ¹⁴CO₂ generation may provide erroneous results. However, the use of ¹⁵N-labelled arginine as substrate and measurement of ¹⁵N-labelled agmatine is a direct, accurate and sensitive method for determination of the ADC activity.

Results of the present study indicate that the content of agmatine in freeze-clamped intact livers (baseline level without perfusion) and in control-perfused livers are 10–15-fold lower than what we had previously reported in [10]. This difference may be due to a lower concentration in livers obtained from overnight-fasted rats (the present study) versus fed rats [10]. Consumption of agmatine from dietary sources would be expected to increase hepatic [agmatine] in fed versus overnight-fasted rats. In addition, in previous studies [agmatine] was measured by HPLC [10], rather than the isotope dilution methodology used in the present study. It is possible that the measurement of agmatine levels by HPLC resulted in an overestimation of tissue [agmatine] due to an incomplete separation from other tissue constituents.

An observation of special interest is that insulin and glucagon had no significant effect on the output of [guanidino-¹⁵N₂]agmatine, whereas dibutyryl-cAMP significantly (*P* < 0.05) stimulated the output of [guanidino-¹⁵N₂]agmatine during liver perfusion with [guanidino-¹⁵N₂]arginine (Figure 4). Both glucagon and insulin are established regulators of hepatic amino acid uptake and metabolism. In most cases, the action of glucagon is brought about by cAMP [31,33]. Previous studies by O’Sullivan et al. [31] have shown that both glucagon and dibutyryl-cAMP stimulated the release of ¹⁴CO₂ during liver perfusion with [U-¹⁴C]arginine but not from [U-¹⁴C]ornithine. The observation of the present study showing that dibutyryl-cAMP stimulated the flux through the ADC reaction is in line with the stimulation of ¹⁴CO₂ release during liver perfusion with [U-¹⁴C]arginine [31].

cAMP regulates numerous physiological functions and enzyme activities, and mediates long- and short-term response mainly through the cAMP/PKA (protein kinase A) signalling pathway and phosphorylation [33,34]. An example is the down-regulation of NO production and the up-regulation of arginase [35]. It is

unlikely that the stimulation of arginase affected flux through the ADC reaction because the addition of Mn^{2+} , which significantly stimulated the production of urea by the arginase reaction (results not shown), had minimal effect on flux through ADC. An inhibition of cytosolic NO production by cAMP might affect mitochondrial ADC but it is most probable that cAMP stimulates flux through the ADC reaction by the cAMP/PKA signalling pathway. This possibility is in line with earlier studies suggesting that PKA activity is present in mitochondria [36], and affects the activity of various enzymes [33,34]. Nonetheless, further studies are required to explore the mechanism(s) of the cAMP action on flux through the ADC reaction.

In mammalian tissue, agmatine may be metabolized by the agmatinase reaction to form urea and putrescine or by the diamine oxidase reaction to form guanidinobutanoic acid [3,9,37]. Preliminary GC-MS screening indicates that ^{15}N -labelled putrescine was formed by the mitochondria (results not shown), but ^{15}N -labelled guanidinobutanoic acid was not detected. However, further qualitative and quantitative analyses, including specific GC-MS methods, are required to elucidate mitochondrial agmatine metabolism.

In summary, the findings of the present study demonstrate that ^{15}N -labelled agmatine was generated from exogenous ^{15}N -labelled arginine both in isolated mitochondria and in the liver perfusion system. The observations are in line with earlier studies demonstrating the activity of ADC in mammalian tissue [2,3,5,6,10]. The results suggest that cAMP may regulate the flux through the ADC reaction.

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