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Unlocking the pharmacological potential of Brennnesselwurzel (*Urtica dioica L.*): an in-depth study on multifaceted biological activities

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Abstract

Brennnesselwurzel (Urtica dioica L.) is recognized for its diverse pharmacological properties. With a range of chemical constituents, such as vitamins, minerals, phenolic compounds, fibers, and amino acids, Brennnesselwurzel (BWE) has a long history of traditional medicinal use in Europe and Asia. The correlation between a plant's metabolite composition and its activity can vary depending on considerations such as geographic location, environmental conditions, and genetic variations. In the present study, we explore the phytochemical profile and biological activity of the 70% acetone extract of the BWE plant. The chemical profile of the BWE extract was explored using several techniques, including amino acid analyzer, HPLC, GC–MS, and other colorimetric analysis. The antioxidant activity of the BWE extract was assessed by evaluating the total antioxidant, free radical scavenging activity (DPPH, ABTS, H₂O₂), and metal chelating scavenging activity (FRAP, CUPRAC, metal chelating). Furthermore, we assessed the antimicrobial and antiproliferation activities of the BWE extract against 29 microbial strains and 15 cell lines, respectively. Our phytochemical analyzes revealed that the BWE extract has a unique profile of metabolites including amino acids, flavonoids, phenolics, volatile oils, lipids, and vitamins. The BWE extract showed a total antioxidant capacity of 30.94 ± 1.58 mg GAE/g, together with potential free radical scavenging activity towards ABTS $(IC_{50} = 153.51 \pm 3.97 \ \mu g/ml)$, DPPH $(IC_{50} = 195.75 \pm 5.91 \ \mu g/ml)$, and H_2O_2 $(IC_{50} = 230.67 \pm 5.98 \ \mu g/ml)$. Although the BWE extract showed no significant antifungal activity, our findings revealed that the BWE extract possesses substantial antibacterial activity against Staphylococcus epidermidi, Streptococcus mutants, Enterococcus faecalis, Micrococcus sp., Klebsiella pneumonia and Porphyromonas gingivalis. Furthermore, the BWE extract demonstrated potential antiproliferative activity toward a panel of cancer cell lines with a high selectivity index. Among the cells examined, the BWE extract exhibited significant cytotoxic activity toward HCT-116, A-549, MDA-MB-231 cells with IC_{so} of 15.11, 15.32, 15.79 μ g/mL, respectively, while it possessed no significant cytotoxic activity towards WI-38 cells (IC₅₀ 119.62 μ g/mL). Taken together, our findings reveal that BWE extract possesses a wide spectrum of biological activities, including antioxidant, antibacterial, and antitumor activities, and could be considered for further research to explore its potential as a natural plant-based supplement for human diseases.

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Keywords Antibacterial activity, Antifungal activity, Antioxidant activity, Antitumor activity, *Brennnesselwurzel*, GC–MS analysis, HPLC analysis, Medicinal plants, Phytochemical analysis

Introduction

Drug discovery is a complex and challenging process involving the identification and development of new medications for treating various diseases [1, 2]. Meeting regulatory requirements and obtaining approval from health authorities, such as the U.S. Food and Drug Administration (FDA) or the European Medicines Agency (EMA), is a crucial and complex step in drug discovery [3-5]. Historically, drugs based on natural products have been used in traditional medicine [6-9]. Currently, these drugs play a remarkable role in drug discovery, as they offer a vast source of chemical diversity, with complex structures and unique biological activities [10]. Natural product-based drugs often exhibit remarkable efficacy, high selectivity, good tolerability, and low toxicity [11–15]. Therefore, natural product-based drugs with potential medicinal benefits continue to be an essential area of research. In this context, medicinal plants have become an important source of bioactive compound identification and have played a vital role in human health and well-being since ancient times [16, 17]. With an estimated 85% of the population in Asia and the Middle East relying on traditional medicine for their primary healthcare needs [18], the exploration of medicinal plants holds significant promise for the discovery of novel drugs and therapeutic interventions including inflammation, hypercholesterolemia, atherosclerosis, endothelial dysfunction, asthma, and cancer progression [19]. Most of the medications used in traditional medicine originate from natural products such as plants, and then clinical trials have been conducted to distinguish valuable plants with beneficial impacts [20]. These plants have been used in several contexts as traditional medicines [21], including Ayurveda, traditional Chinese medicine (TCM), and traditional Indian medicine (TIM) [22, 23]. Nowadays, several pharmaceutical drugs are derived from plant sources or are inspired by plant-based compounds. The discovery and development of new drugs often involve investigating the chemical components of plants and their potential therapeutic applications [24-26]. Medicinal plants produce a wide range of bioactive metabolites, including alkaloids, flavonoids, terpenoids, and phenolics, which contribute to their therapeutic effects [27]. These compounds can have various biological effects, such as anti-inflammatory, antimicrobial, and antioxidant activities [28–31].

Among the vast array of medicinal plants, *Brennnesselwurzel* (the German term for stinging nettle, *Urtica*

dioica L.) has gained significant recognition for its diverse pharmacological properties [32, 33]. According to previous reports, Urtica dioica has a wide range of chemical constituents that contribute to its pharmacological activities, such as vitamins, minerals, phenolic compounds, fibers, and amino acids [32, 34]. In addition, its leaves are rich in carbohydrates, fats, vitamins, chlorophylls, carotenoids, and minerals [32, 35]. Therefore, stinging nettle (Urtica dioica) has been used extensively for several centuries as a traditional medicine, particularly in Europe and Asia [36]. Urtica dioica has been extensively studied for its antioxidant properties, as it contains flavonoids and phenolic chemicals [35]. The antioxidant defense mechanism protects cells from the harmful effects of reactive oxygen species (ROS), preventing lipid peroxidation and DNA damage [37, 38]. In addition to the antioxidant potential of Urtica dioica, the plant also possesses significant antimicrobial activity against both gram-negative and gram-positive bacteria, including Escherichia coli, Bacillus subtilis, Lactobacillus plantarum, and Pseudomonas aeruginosa [39-41]. Urtica dioica extracts have also shown antifungal activity against different fungal species, especially Candida albicans, which is considered a common fungal pathogen responsible for yeast infections. Urtica dioica exhibits various bioactive metabolites, including flavonoids, lectins, and polysaccharides, which may interfere with the integrity of the fungal cell membrane by changing the permeability and fluidity of the cell membrane, which can lead to leakage of intracellular contents [36, 42, 43]. Furthermore, several studies have reported the cytotoxic and antiproliferative activities of Urtica dioica extracts against various cancer cell lines such as prostate and breast cancers, suggesting its potential as a natural anticancer agent. Despite that, the potential anticancer activity of Urtica dioica is an area of ongoing research, and the exact mode of action is not yet fully understood. The reported studies suggested that the antioxidant activity of Urtica dioica extracts, which is attributed to its bioactive compounds such as flavonoids, phenolic compounds, and polysaccharides, may help protect cells from oxidative damage and potentially inhibit tumor growth [44-46].

The correlation between metabolite composition and activity of a plant can vary depending on factors such as geographic location, environmental conditions, and genetic variations. For example, different populations

of the same plant species, such as Urtica dioica, found in different regions can exhibit variations in their metabolite profiles and biological activities [36]. These variations can be attributed to various environmental factors, soil composition, climate, and genetic differences among the populations [47]. These differences can potentially lead to variations in the medicinal properties of the plant, including its antimicrobial, antioxidant, or anticancer effects. Based on these facts and our continuous efforts in discovering novel bioactive agents [48-53], the presented research paper aimed to comprehensively explore the phytochemical composition of the 70% acetone extract of the Brennnesselwurzel plant from Germany. In addition, we intended to gain insight into the antioxidant, antimicrobial, and anticancer potency of the extract. The findings of this research will provide valuable information on the therapeutic potential of Brennnesselwurzel extract in the context of antioxidant supplementation and cancer treatment, which will pave the way for further research and development of natural products derived from this medicinal plant.

Materials and methods

Plant material and extraction

The Brennnesselwurzel plant (Urtica dioica) was acquired from Vita Ideal (Germany). The institutional botanist identified the plants and confirmed them by Dr. Ahmed Elbanhawy, Department of Botany, Suez Canal University, Egypt, as Pulicaria Undulata. At least five representative samples have been vouched and deposited at SCUI herbarium under the accession numbers AE_19812023/10. The acquired plant was pulverized using a mixer mill (Retsch MM400, Germany). The powdered Brennnesselwurzel plant was mixed with a solution containing 70% acetone and deionized water (240 ml) and placed at 28° C for 3 days on a rotary shaker for extraction (Clim-O-Shake system Kuhner IRC-1-U). The resulting mixture was double-filtered using filter paper and then dried with a rotary evaporator to obtain the dry lyophilized extract.

Phytochemical characterization

Assessment of the total primary phytochemical metabolites

The evaluation of total primary phytochemical metabolites was carried out using the reported chemical and colorimetric analysis including the Folin-Ciocalteu assay (total phenolic content) [54], Aluminum chloride colorimetric assay (total flavonoid content) [55], Bromocresol greenbased colorimetric assay (total alkaloid content) [56], Conventional assay (total lipid content) [57], Bradford assay (total protein content) [58], Modified Vanillin-Sulphuric Acid assay (total saponin content) [59], Phenol–sulphuric acid assay (total carbohydrate content) [60], and acidified-vanillin assay (total tannin content) [61].

GCMS analysis

Analysis using Thermo Scientific's GC-MS-Orbitrap Q Exactive. A vial containing 4 ml of plant leaf extract or 50 µM standard solutions was spun dry at 39° C using an insert. Twenty ml of methoxyamine (20 mg/mL; 90 min at 37°C) and 30 ml of MSTFA were added to the samples automatically using a preparative robot and left for 30 min at 37 °C. To calculate the retention index, each sample received 5 μ l of an alkane mix (3 μ g. μ l⁻¹) containing 14 alkanes ranging from C9-C36 (Connecticut n-hydrocarbon mix, Supelco). The analysis consisted of injecting 1 µl in split mode at 230° C using a Trace 1300 Series GC equipped with a TG-5 SILMS column (30 m×0.25 mm×0.25 µm; Thermo Scientific). Helium was continuously flowed at 1 ml/min. First, the GC oven reached 70° C for one minute. It was heated to 325° C at 15°C/minute for four minutes. Full MS scan mode with positive polarity, filament delay 4.12 min, 50-750 m/z mass scan range, 60,000 resolution, AGC target 1E6, and 300° C MS transfer line were used for MS research. Ion source electron impact ionisation at 250° C (70 eV). TraceFinder® (Thermo Scientific) automatically recognized analytes using retention time, a primary characteristic fragment (m / z ion), and a confirmation fragment with a maximum tolerance of 0.00007 Da through focussed screening [62-64].

Analysis of amino acids

Employing a Shimadzu Prominence xR UFLC system in conjunction with a SCIEX hybrid triple guadrupolelinear ion trap MS featuring a Turbo VTM ESI source, we conducted the separation and analysis of all samples. Cysteinic acid was negatively ionized, but all other amino acids were positive. The elution of amino acids was achieved by a 20 mM ammonium formate gradient in water (A) and acetonitrile: water (90:10) with a pH of 3.0, incorporating a concentration that is increasing steadily. A 3-µl sample was introduced into the Infinity Lab 120 Z-HILIC Poroshell column (2.7 µm, 100×2.1 mm; Agilent Technologies, Santa Clara, CA, USA). A continuous 0.25 ml/min flow was used for gradient elution of amino acids. After two minutes of 100% to 90% B, six minutes of 50% B and 30 s of 100% B, the column was reequilibrated for 6.5 min. The ESI+and ESI ion spray voltage was 4.5 kV; ion source temperature was 400 °C; and the curtain gas concentrations were 1, 45, 2, 40, and 35. Targeted MRM was utilized to detect and monitor ions. The Quantitation Wizard in Analyst (v. 1.6.2) (SCIEX, Concord, Canada) analyzed the data.

To determine the quantities and losses of amino acids, the peak regions were compared directly with isotope-labeled internal standards [65-67].

Analysis of flavonoids and phenolic acids

Flavonoids and phenolic compounds were analyzed using a Waters 2695 Alliance HPLC system (Waters Inc., Milford, CT, USA) and a UV-Vis DAD. Waters SunfireTM C18 reverse phase chromatography column (250 mm length, 4.6 mm width, 5 µm particle size) was used for separation. The equipment received standard phenolic solutions and mixtures via auto-injector. To find an effective standard separation method, gradient, and isocratic mobile phases were investigated at different column temperatures and flow rates. After initial investigations, the gradient method was chosen using acetonitrile (mobile phase A, HPLC grade \geq 99.9%; Honeywell Seelze, Germany) and phosphoric acid (mobile phase B), prepared by dropwise adding 85% orthophosphoric acid (Sigma-Aldrich, Merck, Darmstadt, Germany) to HPLC grade water (Carlo Erba) until pH=2. The 60-min run-time of the method adjusted the concentration gradient as follows: A) 5% A and 95% B at first; b) 35% A and 65% B for 15 min; c) 35% A and 65% B for 20 min; d) 40% A and 60% B for 30 min; e) 40% A and 60% B for 35 min; f) 50% A and 50% B for 40 min; g) 70% A and 30% B for 52 min; and h) 0.5 mL/min flow rate and 5° C temperature maintained. This study chose three wavelengths: 210, 280, and 360 nm for the HPLC-DAD analysis after examining the UV-vis spectra of the phenolic standards' UV-Vis spectra [68, 69].

Analysis of total fats and vitamins

Fat methyl esters were evaluated according to Tsiplakou et al. [70]. Overall, a mixture of 15 ml of isopropanol and 8.5 ml of the sample was combined in a separate funnel. Subsequently, hexane (11.25 ml) was added in 3-increments and the mixture was shaken once more. To separate and evaporate the hexane layer (20 mL), $(2520 \times g)$ at 5° C for 5 min) was centrifuged and 7.5 mL 0.47 M aqueous Na₂SO₄. 5 ml of hexane was added to 100 mg of lipid. After adding 0.2 ml of 11.2 g of KOH to 100 ml of methanol, the mixture was vortexed and left to react for five minutes. After adding 0.5 g of NaHSO₄·×H₂O, the mixture was centrifuged at 25° C for 5 min at $5000 \times g$. Analysis of the supernatant layer was carried out following the method outlined by Gortzi et al. [71], then a fifth of the sample split. For five minutes, the column temperature increased 20° C every minute from 70° C to 160° C for three minutes. At 5° C each minute, the temperature reached 240° C. The compounds were identified using Sigma FAME standards in established amounts.

We triple-checked every sample. Following Kondyli et al.. [72], fat-soluble vitamins were detected. A straightforward procedure involved shaking the sample (2 ml) with absolute ethanol containing 0.1% (w/v) ascorbic acid (5 ml), along with 50% KOH (2 ml). The resulting mixture was then allowed to react at 80 degrees Celsius for a duration of 20 min. Subsequently, petroleum ether: diethyl ether (1: 1) (20 ml) with 0.01% butylated hydroxytoluene (BHT) was introduced after cooling the sample. After vortexing and adding 15 ml of water at 2 °C, the mixture was centrifuged at 2000×g for 15 min. After splitting, the biological layer vanished. Vitamin E analysis was done according to the method of Gortzi et al. (52). Additionally, an l-Porasil column (125, 10 μ m, 3.9 \times 300 mm; Waters Corp., Waltham, MA, USA) was used. Initially, lipid (1 g) was dissolved in n-hexane (5 ml) and then introduced into the HPLC system. The mobile phase was 2: 97.5: 0.5 2-propanol, n-hexane, and 100% ethanol added at 1 ml / min. Furthermore, vitamin A analysis was determined using Mendoza et al. method [73]. The analysis was carried out using a Shimadzu CBM-20A liquid chromatograph (Shimadzu Europa GmbH, Duisburg, Germany) equipped with a 300° C column oven and SIL-20AC autosampler. The detector was a 325 nm Shimadzu SPDM20A diode array detector. Using a Luna C18 column (Phenomenex, Torrance, CA, USA), 250×4.6 mm, 5 μ m. The mobile phase was 100% methanol at 1 ml / min. Three times each sample was analyzed.

Assessment of antioxidant activity Evaluation of free radical scavenging activity

ABTS method Two solutions of distilled water were prepared: one with a concentration of 7 mM for ABTS and another with a concentration of 2.45 mM for potassium persulfate. Additionally, a diluted ABTS reagent was used to create a functioning ABTS reagent. Then. Trolox was synthesized as standard solutions with 0.1 mg / ml in a solution of water: ascorbic acid (1: 2) and 1 mg / ml of alpha-tocopherol in hexane. Standard solutions were used to create batch dilutions and then the ABTS working reagent was added. The absorbances were tested after shaking the contents of the tube and letting it sit at room temperature for seven minutes. This approach involved the creation of calibration curves for the reference solutions. The produced samples' hydrophilic and lipophilic extracts underwent identical analyses. At 734 nm, the absorbances were measured [74]. The following formula was used to determine the capacity of extracts to ABTS radical scavenging [75]:

ABTS radical scavenging effect (%) = ((AB - AA)/ AB) \times 100

Where, AB is the absorbance of ABTS radical+control and AA is the absorbance of ABTS radical+sample extract.

DPPH method A 1 mmol/L concentration of DPPH standard solution was produced in solutions of ethanol and methanol. The standard solution of alpha-tocopherol (1 mg / ml) was prepared in hexane. Subsequently, a sequence of dilutions was created, covering a concentration range of 2 to 5 μ g/mL. In addition, a standard Trolox solution was formulated, comprising 0.1 mg / ml of the compound dissolved in a solution of water: ascorbic acid (1:2). This solution was diluted several times, with concentrations ranging from 0.4 to 20 µg/mL. Calibration curves were generated for the dilution series of standard alpha-tocopherol and Trolox solutions, after incorporation of the DPPH solution into each prepared solution. Both lipophilic and hydrophilic extracts were prepared, with an equivalent volume of DPPH solution added to each. The material was shaken and exposed to darkness for three hours and thirty minutes. At 517 nm, the absorbances of the samples and standards were determined [74]. The capacity of the extracts to remove DPPH radicals was determined using the formula below [76]:

DPPH radical scavenging effect (%) = $((A0 - A1)/A0) \times 100$

where: A0 = the control absorbance, A1 = the sample absorbance.

 H_2O_2 method Following the Ruch et al. [77] method, an aliquot of extracts/ ascorbic acid (0.1 ml, 25–400 µg/ mL) was transferred to Eppendorf tubes. The volume was set at 0.4 ml with 50 mM phosphate buffer (pH 7.4) and then a solution of H2O2 (0.6 ml, 2 mM) was added. After the reaction mixture was vortexing and allowed to react for 10 min, the absorbance at 230 nm was measured. The ability to eliminate H_2O_2 was evaluated using the following formula [78]:

 H_2O_2 scavenging effect % = [(A0 A1)/ A0] × 100

Where: A0 = Absorbance of the control, A1 = Absorbance of a sample.

Evaluation of metal-chelating scavenging activity

FRAP method FRAP was determined in this study according to Akinmoladun et al. [79] that used a method designed by Benzie and Strain (1996) to quantify the ability to decrease ferric ions. FRAP reagent was made by combining 300 mM sodium acetate buffer (pH 3.6), 10.0 mM TPTZ solution (tripyridyl triazine) and 20.0 mM

FeCl3.6H2O solution in a volume ratio of 10: 1: 1. Subsequently, different amounts of leaf extract (2, 3.9, 7.8, 15.6, 31.25, 62.5, 125, 250, 500, and 1000 μ g/ml) were added to 3 ml of FRAP reagent and the reaction mixture was incubated at 37 C for 30 min. At 593 nm, an increase in absorbance was measured [80].

Cupric ion reducing antioxidant (CUPRAC) method Duplicates of all samples and standards were prepared using a mixture consisting of 150 μ L each of 7.5 mM neocuproine, 1 M ammonium acetate, and 10 mM copper chloride dihydrate (II). Ascorbic acid (1g) was used to generate the standard curve. The concentrations included a blank (EDTA) and 1000, 500, 250, 125, 62.5, 31.25, 15.6, 7.8, 3.9, and 2 μ g/ml. After adding neocuproine, ammonium acetate, and copper (II) chloride dihydrate to both samples and standards, the test tubes were incubated at room temperature for thirty minutes. Subsequently, the test tubes were examined at 450 nm using a Hitachi U-2001 Spectrophotometer [81].

Metal chelating activity method Two working solutions were prepared: one for 0.25 mM ferrosine and the other for 0.1 mM ferrous sulfate (II). A 100 mg/L concentration of a standard Na2EDTA solution was obtained. This standard solution was used to create a range of dilutions with concentrations ranging from 1 to 20 mg/L. Reagents that had already been produced were added to the prepared solutions to create the calibration curve. After a ten-minute incubation, the contents of the tubes were shaken and absorbances were measured. The methodology used to generate the EDTA calibration curve was applied similarly to prepare extracts from hydrophilic and lipophilic extracts. At 562 nm, absorbances were measured [74].

Assessment of antimicrobial activity Microbial strains

A total of 29 pathogenic microbes were used in the study divided into 13 fungal strains, 8 Gram-positive bacteria and 8 Gram-negative bacteria. Fungal strains included *Aspergillus fumigatus* (RCMB 002008), *A. niger* (RCMB 002005), *Candida albicans* RCMB 005003 (1) ATCC 10231, *Penicillium aurantiogriseum* IMI 89372, *Syncephalastrum racemosum* RCMB 016001 (1), *P. marneffeii* (RCMB 001022), *Cryptococcus neoformas* RCMB 0049001, *Candida lipolytica* RCMB 005007(1), *P. expansum* RCMB 001001 (1) IMI 28169, *P. italicum* RCMB 001018 (1) IMI 193019, *Fusarium moniliform* (RCMB 008005), *Trichophyton rubrum* (RCMB 025002), and *Geotrichum candidum* (RCMB 041001). Gram-positive bacteria included *Staphylococcus aureus* ATCC 25923,

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Bacillus subtilis RCMB 015 (1) NRRL B-543, B. cereus RCMB 027 (1), Staphylococcus epidermidis RCMB 009 (2), Micrococcus sp. RCMB 028 (1), Streptococcus mutants RCMB 017 (1) ATCC 25175, Methicillin resistant Staphylococcus aureus (MRSA) ATCC 4330, and Enterococcus faecalis (ATCC 29212). Gram-negative bacteria were Enterobacter cloacae RCMB 001 (1) ATCC 23355, Salmonella typhimurium RCMB 006 (1) ATCC 14028, Escherichia coli ATCC 25922, Klebsiella pneumonia RCMB 003 (1) ATCC 13883, Proteus vulgaris RCMB 004 (1) ATCC 13315, Serratia marcenscens 007001, Pseudomonas aeruginosa ATCC 27853, and Porphyromonas gingivalis RCMB 022001 (1) EMCC 1699.

Antimicrobial activity screening

The diffusion agar technique (bacterial suspension containing 106 CFU/ml of the bacterial test strain disseminated over Muller Hinton agar) was used to verify the antibacterial properties of the plant extracts. Each extract was placed on 5 µg/disc of sterilized filter paper. gentamycin was used as a positive control. These discs were placed on agar plates and left to incubate for 24 h at 37° C. Using a Vernier caliper, the inhibition zones were measured. For antifungal activity, the plant extracts were tested using a well diffusion technique. A 0.2 mL saline inoculum, a fungal strain, was applied to an agar plate. In every plate, five 4 mm ditches were created. 100% ethanolic plant extracts (50 mg/ml) were made and 50 µl of the extracts were added to each well. The positive control in this experiment was ketoconazole (100 μ g/ml). Then, 24 h were spent incubating the plates at 37 C. Bacterial and fungal species were acquired from the Regional Center for Mycology and Biotechnology (RCMB) in Egypt.

Assessment of antiproliferation activity Cell culture

The American Type Culture Collection (ATCC, Rockville, MD) provided the following human cancer cell lines. HepG-2 (human hepatocellular cancer cell line), HCT-116 (human colon cancer cell line), MCF-7 (human breast cancer cell line), HT-29 (human colon

WI-38, or normal human lung fibroblast cells, were donated by the American Type Culture Collection (ATCC, Rockville, MD). In addition, dimethylsulfoxide (DMSO), MTT, and trypan blue dye were supplied by Sigma (St. Louis, MO, USA). Lonza (Belgium) provided fetal bovine serum, gentamycin, L-glutamine, RPMI-1640, HEPES buffer solution and 0.25% trypsin-EDTA. Cells were grown in RPMI-1640 medium supplemented with gentamycin (50µg/ml) and inactivated foetal calf serum (10%). Cell subculture was performed two or three times a week, and cells were kept at 37° C in a humidified atmosphere with 5% CO2. Cells for WI-38 (Normal Human Cell Lines) were grown in Dulbecco's Modified Eagle Medium (DMEM) supplemented with L-glutamine (1%), heat-inactivated Foetal Bovine Serum (10%), and HEPES (50µg/ml) [82–84].

Cytotoxicity assay

The MTT cell viability assay was used in 15 different cell lines to assess the activity of the Brennnesselwurzel (nettle root) extract. Cells were distributed in Corning[®] 96-well tissue culture plates, suspended in media at a concentration of 5×10^4 cells/well, and exposed to varying concentrations of brennnesselwurzel extract (nettle root) extract $(0-500 \,\mu\text{g/mL})$. Subsequently, they were incubated in a humidified atmosphere of 5% CO2 at 37° C for 24 h. Following the incubation period, 12 mM MTT stock solution (10 µL, 5 mg MTT/ 1 mL PBS) was added per well, along with fresh RPMI 1640 medium (100 µL) without phenol red (untreated controls). The plate was then shaken and further incubated at 37° C in a humidified environment with 5% CO2 for an additional 4 h. An aliquot of the medium (85 μ L) was removed from the wells and DMSO (50 µL) was introduced. Each well was thoroughly mixed using a pipette and the resulting mixture was then incubated at 37° C for 10 min. Subsequently, the reduction of MTT was evaluated by assessing absorbance using a microplate reader (SunRise, TECAN, Inc., USA) at 590 nm. For each concentration, the average of three replicates was calculated [85-87]. The percentage of viability was then computed and expressed as

Percentage of viability = (Aborbance of treated cells/ Aborbance of control cells) $\times 100\%$

cancer cell line), MDA-MB-231 (human breast cancer cell line), PC-3 (human prostate cancer cell line), A-549 (human lung cancer cell line), A-431 (human skin cancer cell line), Panc-1 (human pancreatic cancer cell line), CACO2 (human intestinal ca Furthermore,

Subsequently, using GraphPad Prism software (San Diego, CA, USA), the 50% inhibitory concentration (IC50), representing the concentration required to induce detrimental effects in 50% of intact cells, was determined from graphs of the dose–response curve for each concentration [88].

Statistical analysis

Microsoft Excel was used to assess the free radical DPPH, the ABTS radical, H₂O₂-induced oxidative stress, the ferric reducing antioxidant power (FRAP), CUPRAC, and the metal-chelating scavenging activity. Furthermore, all data were presented as mean values with standard deviations in triplicate. One-way ANOVA was used to assess the mean values. There were significant variations in parameter means (*p* < 0.05) [89, 90].

Results

Phytochemical characterization

After the plant material was collected cautiously, the extraction process was applied using 70% acetone in a water solution. The Brennnesselwurzel (BWE) plant extract was tested for its primary phytochemical content and secondary metabolites, including amino acids, flavonoids, phenolics, lipids, and vitamins. Furthermore, detailed gas chromatography was employed in association with mass spectrometry to gain more insight into the prominent bioactive metabolites in the BWE extract.

Analysis of total nutrients

In the analysis of total nutrients of the BWE extract, we evaluated the quantitative composition of essential primary metabolites present in the plant. The analysis was carried out using validated spectrophotometric and chromatographic methods, ensuring reliable and accurate results. Our findings revealed that the BWE extract is a valuable source of various bioactive nutrients (Fig. 1). Our results showed that the BWE extract consists of a wide range of primary metabolites, including flavonoids (4.53 mg/g), phenolic acids (81.47 mg/g), alkaloids (0.15 μ g/g), lipids (0.12 mg/g), ash (9.31 mg/g), carbohydrates (96.15 mg/g), proteins (92 mg/g), saponin (37.62 mg / g) and tannins (2.62 mg/g). The analysis revealed that the BWE extract has high concentrations of phenolic compounds, carbohydrates, proteins, and saponin. These findings indicate that BWE has significant antioxidant properties due to high levels of phenolic compounds, providing potential health benefits. In addition, significant amounts of carbohydrates and proteins suggest that BWE could serve as a valuable nutritional source. The presence of saponin further implies the chemical diversity of the BWE extract, pointing toward its possible use in therapeutic contexts. On the other hand, the considerable levels of flavonoids and alkaloids suggest the possible antioxidant potential and medicinal properties. Furthermore, the presence of ash and tanning further suggests the mineral content of BWE extract, which is significant



Fig. 1 Total primary metabolites detected in BWE extract by phytochemical screening assays. Data were presented as mean values with standard deviations in triplicate



Fig. 2 Amino acids detected and their concentrations in the BWE extract by an amino acid analyzer

both for nutritional and medicinal aspects. These findings provide valuable information on the phytochemical profile of BWE and further indicate its possible potential applications in various contexts.

Amino acid analysis

An amino acid analyzer was used to determine the amino acid profile of the BWE extract. Our analysis provided a comprehensive overview of 18 amino acids and related amines, along with their respective retention time, response, and amount. Our results revealed that the BWE extract exhibits a wide range of different amino acids in considerable concentrations, suggesting the BWE extract as a potential source of essential amino acids (Fig. 2, Table S1). Among the amino acids detected, the BWE extract possesses high concentrations of certain amino acids, including glutamic acid (GLU), aspartic acid (ASP), leucine (LEU) and lysine (LYS) (>20 PPM). The substantial concentration of glutamic acid and aspartic acid in the BWE extract indicates the possible effect of the extract on protein metabolism and synthesis. The existence of leucine and lysine, essential for muscle health and overall physiological functions, further emphasizes the nutritional richness of the BWE extract. These findings underscore the potential significance of BWE extract in supporting protein synthesis and highlight its nutritional value, making it a significant topic in dietary.

Analysis of phenolic acids and flavonoids

Then, the content of flavonoids and phenolic acids in the BWE extract was assessed using targeted High Performance Liquid Chromatography (HPLC) analysis (Fig. 3). HPLC analysis unraveled the presence of various flavonoid and phenolic metabolites in considerable concentrations. Among the identified flavonoid metabolites, catechin and luteolin exhibit appreciable concentrations $(>10\mu g/g)$, further underscoring the possible potential of pf BWE extract as an antioxidant and anti-inflammatory supplement. Additionally, other bioactive flavonoid metabolites were detected at considerable levels, including kampferol, apigenin, quercetin, and rutin (Table S2). On the other hand, the evaluation of phenolic metabolites revealed that catechol and syringic are present in substantial concentrations (14.22 μ g/g, 11.38 μ g/g, respectively), signifying the possible therapeutic potential of BWE extract as an antitumor and antioxidant supplement. In addition, other phenolic acid metabolites were detected in compounds, such as salicylic, pyrogallol, cinnamic, caffeic and ferulic, which were identified in considerable concentration in the BWE extract, ranging from 2.10 to 7.02 μ g/g (Table S3). Taken together, the substantial flavonoids detected in the BWE extract suggest the antioxidant and antitumor characteristics of the BWE extract. The presence of different flavonoid metabolites such as kampferol, apigenin, quercetin, and rutin contributes to the overall chemical profile of the extract. Furthermore, phenolic acids possess several bioactivities including antioxidant, anticancer, and anti-inflammatory effects, suggesting that BWE may provide potential therapeutic benefits.

Analysis of total vitamins

Furthermore, we explored the phytochemical profile of the BWE extract employing HPLC analysis to determine



Fig. 3 HPLC chromatogram and concentration of the detected flavonoids (A-B) and phenolic acids (C-D) metabolites in the BWE extract

the total vitamin profile. The analysis can be divided into two groups according to solubility, namely fat-soluble vitamers (FSV) and water-soluble vitamers (WSV) as shown in Fig. 4 [91]. The results of the HPLC analysis revealed distinctive profiles for both FSV and WSV. Fat-soluble vitamers, including Vitamin D, Vitamin K, and Retinol, demonstrated varying concentrations, with Vitamin K exhibiting the highest concentration at 12.56 µg/gm. These fat-soluble compounds are integral to nutritional and health considerations, and their presence in the BWE extract underscores the potential health benefits associated with the consumption of this botanical extract (Table S4). However, water-soluble vitamers showed a diverse range of concentrations, with Vitamin B2 leading at 11.20 μ g/g, followed by Vitamin B3 at 9.80 µg/g. Variable concentrations of water-soluble vitamers, such as Vitamin C, Vitamin B1, and Vitamin B6, point to the diverse nutritional composition of the BWE extract. These water-soluble compounds are essential for physiological functions and are indicative of the potential role of the extract in the health context. Collectively, the HPLC results provide a complete illustration of the vitamin composition of the BWE extract, highlighting its nutritional richness and potential health-promoting properties (Table S5).

Gas Chromatography (GC) analysis

The identification of secondary metabolites in BWE extract was further achieved by conducting GC-MS analysis with Thermo Scientific's GC-MS-Orbitrap Q Exactive system. In this technique, the plant extracts or standard solutions were dried, then treated with methoxyamine and MSTFA using a preparative robot and subjected to retention index calculation with an alkane mix. The samples were then injected into a Trace 1300 Series GC equipped with a TG-5 SILMS column and the resulting chromatogram was analyzed with a full MS scan mode, offering a mass scan range of 50-750 m/z, a resolution of 60,000, and an AGC target. The ion source operated at 250° C with electron impact ionization (70 eV). Thermo Scientific's TraceFinder[®] software was employed for automatic analyze recognition based on retention time, a primary characteristic fragment, and a confirmation fragment. The detected metabolites, retention time, peak area, and their molecular formula are summarized in Table 1.

The GC–MS chromatogram (Fig. 5 and Table 1) identified a total of 41 secondary metabolites in the BWE acetone extract. The identified metabolites were mainly fatty acids and aromatic metabolites. This categorization showed that BWE extract shows 18 different fatty acid



Fig. 4 HPLC chromatogram and concentration of detected (A, B) FSV and (C, D) WSV metabolites in the BWE extract

metabolites, indicating the presence of metabolites with long hydrocarbon chains. aqMoreover, various fatty acids and their derivatives play essential roles in energy storage, signaling, and structural components of cells and are associated with lipid metabolism. While, the aromatic class of metabolites exhibited 10 different recognized metabolites, including benzoic acid methyl ester and. psi,. psi-, 1,1,2,2'-tetrahydro-1,1'-dimethoxy- Carotene with the highest peak area % (2.89% and 3.78%, respectively). Aromatic metabolites often contribute to diverse biological activities. Further analysis also revealed the presence of various compounds such as cyclopropaneoctanoic acid, 2-octyl, methyl ester (0.27%), ethanol, 2- (9-octadecenyloxy) -, (Z)-(0.26%) and 2,2,4,4-Tetramethyl-6-(1oxo-3-phenylprop-2-enyl)-cyclohexane(1aà,1bá,4aá,5á ,7aà,7bà,8à,9á,9aà)]- (0.19%), which belong to classes of cyclopropanes, alcohols and terpenes, respectively. These different metabolites possess various biological activities and are often involved in metabolic processes. In particular, the main compounds identified were fatty acid metabolites that included hexadecanoic acid methyl ester and 8,11-octadecadienoic acid methyl ester, with the highest area percentages of 20.85% and 15.22%, respectively (Figure S1).

We also used GC-MS analysis with Thermo Scientific's GC-MS-Orbitrap Q Exactive system technique in our study to detect the essential oils and other volatile metabolites in the BWE extract. The GC-MS analysis is effective in separating volatile and semivolatile compounds, as it can separate individual volatile compounds on the basis of their boiling points and chemical properties. The mass spectrometer then identifies these compounds by their unique mass-to-charge ratios, providing an accurate and reliable identification of the components present in the oil. GC-MS analysis of the BWE extract identified a total of 30 essential oils and volatile metabolites (Fig. 6 and Table 2). Among these, fatty acids (FA) were predominant (7 metabolites), including n-hexadecanoic acid (8.81%), 9,12,15-octadecatrienoic acid, (Z,Z,Z) -2,3-dihydroxypropyl ester (6.03%) and 6,9,12-octadecatrienoic acid methyl ester (2.73%) with the highest peak area. These findings suggest a substantial presence of FAs in the BWE extract, which suggests their various roles in biological processes. The most prominent other essential oils include allo-ocimene, cis-4-thujanol, caryophyllene and sabinene with area percentages of 12.61%, 10.56%, 5.66% and 4.74%, respectively, which belong

Number	Compound Name	Chemical Formula	Retention Time (RT)	Area%	R.I
1	cis-13-Octadecenoic acid	C ₁₈ H ₃₄ O ₂	5.13	0.33	2178
2	(2-phenyl-1,3-dioxolan-4-yl)methyl ester, cis-9-Octadecenoic acid	C ₂₈ H ₄₄ O ₄	5.23	1.20	3246
3	Benzoic acid methyl ester	C ₈ H ₈ O ₂	6.27	2.89	1084
4	Estragole	C ₁₀ H ₁₂ O	8.86	0.79	1177
5	1-Methoxy-4-(1-propenyl)-benzene	C ₁₀ H ₁₂ O	8.91	0.41	1264
6	Hexadecanoic acid-phenylmethyl ester	C ₂₃ H ₃₈ O ₂	15.87	0.41	2527
7	3-Amino-4-[(1-benzyl-2-methoxy-2-oxoethyl)amino]-4-Oxobutanoic acid	C ₁₄ H ₁₈ N ₂ O ₅	15.98	0.44	2528
8	2H-Indeno[1,2-b]furan-2-one, 3,3a,4,5,6,7,8,8b-octahydro-8,8-dimethyl	C ₁₃ H ₁₈ O ₂	16.66	1.56	1642
9	Cholestan -3-ol,2-methylene, (3ß,5α)	C ₂₈ H ₄₈ O	16.84	0.26	2647
10	methyl 8-(2-octylcyclopropyl)octanoate	C ₂₀ H ₃₈ O ₂	21.52	0.27	2276
11	12-Methyl-tetradecanoic acid-methyl ester	C1 ₆ H ₃₂ O ₂	23.02	0.53	959
12	14-Hydroxy-14-methyl-hexadec-15-enoic acid, methyl ester	C ₁₈ H ₃₄ O ₃	23.02	0.53	2060
13	, 2-(9-octadecenyloxy)-, (Z)- Ethanol	C ₂₀ H ₄₀ O ₂	24.16	0.26	2640
14	9-Hexadecenoic acid, methyl ester, (Z)-	C ₁₇ H ₃₂ O ₂	25.11	0.49	1885
15	Benzenepropanoic acid, 3,5-bis(1,1-dimethylethyl)-4-hydroxy-, methyl ester	C ₁₅ H ₁₆ O ₆	25.46	1.78	1329
16	Hexadecanoic acid, methyl ester	C ₁₇ H ₃₄ O ₂	25.62	20.85	1908
17	1,25-Dihydroxyvitamin D3, TMS derivative	C ₃₀ H ₅₂ O ₃ Si	26.97	0.43	3150
18	Glycidyl oleate	C ₂₁ H ₃₈ O ₃	27.34	0.40	2374
19	Cyclopropanebutanoic acid, 2-[[2-[[2-[(2-pentylcyclopropyl)methyl]cyclopropyl]methyl] cyclopropyl]methyl]-, methyl ester	C2 ₅ H ₄₂ O ₂	27.53	0.46	2528
20	8,11-Octadecadienoic acid methyl ester	C ₁₉ H ₃₄ O ₂	28.61	15.22	2439
21	10-Octadecenoic acid methyl ester	C ₁₉ H ₃₆ O ₂	28.80	5.86	2100
22	trans-13-Octadecenoic acid methyl ester	C ₁₉ H ₃₆ O ₂	28.80	4.86	2163
23	16-Octadecenoic acid, methyl ester	C ₁₉ H ₃₆ O ₂	28.90	1.89	2431
24	Dasycarpidan-1-methanol, acetate (ester)	C ₂₀ H ₂₆ N ₂ O ₂	29.12	2.74	2453
25	Methyl stearate	C ₁₉ H ₃₈ O ₂	29.36	2.64	2133
26	Oleic acid	C ₁₈ H ₃₄ O ₂	29.63	0.26	2113
27	<i>n</i> -Propyl-11-octadecenoate	C ₂₁ H ₄₀ O ₂	30.04	0.42	2284
28	trans-9-Octadecenoic acid, (2-phenyl-1,3-dioxolan-4-yl)methyl ester	C ₂₈ H ₄₄ O ₄	30.08	1.46	2175
29	Norgestrel trimethylsilyl ether	C ₂₄ H ₃₆ O ₂ Si	30.15	0.30	2393
30	(3ß,24S)-Stigmast-5-en-3-ol	C ₂₉ H ₅₀ O	30.54	5.47	2657
31	1H-Cyclopropa[3,4]benz[1,2-e]azulene-5,7b,9,9a-tetrol, 1a,1b,4,4a,5,7a,8,9-octahydro- 3-(hydroxymethyl)-1,1,6,8-tetramethyl-, 5,9,9a-triacetate	$C_{26}H_{36}O_8$	31.27	0.44	3399
32	Champanone A	C ₁₉ H ₂₀ O ₄	31.80	0.19	2470
33	1-Cyclopentyl-4-(3-cyclopentylpropyl)-dodecane	C ₂₅ H ₄₈	32.11	0.99	2785
34	Eicosanoic acid methyl ester	C ₂₁ H ₄₂ O ₂	32.80	4.01	2311
35	Linoleic acid ethyl ester	C ₂₀ H ₃₆ O ₂	33.38	5.82	2139
36	4-Hexyl-1-(7-methoxycarbonylheptyl)bicyclo[4.4.0]deca-2,5,7-triene	C ₂₅ H ₄₀ O ₂	33.95	0.83	2670
37	Dimethyl 2,3-di(8-nonen-1-yl)succinate	C ₂₄ H ₄₂ O ₄	34.02	0.31	2593
38	4-Hydroxy-octadecanoic acid methyl ester	C ₁₉ H ₃₈ O ₃	34.41	2.86	1461
39	3-Ethyl-5-(2-ethylbutyl)-Octadecane,	C ₂₆ H ₅₄	34.51	1.80	2413
40	1,1',2,2'-tetrahydro-1,1'-dimethoxypsi,psiCarotene	C ₄₂ H ₆₄ O ₂	37.93	3.78	3918
41	Icosyl (Z)-octadec-9-enoate	C ₃₈ H ₇₄ O ₂	38.40	1.86	2121

Table 1 List of the main secondary phytochemical metabolites identified in the acetone extract of BWE

to monoterpenes, oxygenated monoterpenes and sesquiterpenes (Figure S1). The existence of monoterpenes and oxygenated monoterpene metabolites could help in understanding the characteristic of essential oils, as monoterpenes often contribute to the characteristic smell of essential oils, as well as knowing the biological activities of the plant extract, as the additional oxygen atoms in monoterpenes could contribute to the therapeutic properties of the extract. Furthermore, the GC–MS analysis revealed the presence of other bioactive



Fig. 5 GC–MS chromatogram of the 70% acetone BWE extract showing the detected secondary metabolites



Fig. 6 GC-MS chromatogram of the 70% acetone BWE extract showing the essential oils and volatile secondary metabolites

metabolites, highlighting the complexity of the composition of the BWE extract and supporting its potential as a natural source of plant-based supplement (Table 2).

Evaluation of antioxidants

Initially, we evaluated the total antioxidant capacity (TAC) of BWE extract, which estimates the total amount of antioxidants in the sample compared to ascorbic acid as a positive control. Our analysis revealed that the BWE extract has a TCA of 30.94±1.58 mg GAE/g, compared to ascorbic acid (72.68 ± 3.74 mg GAE/g). Based on these findings, we encourage you to gain more information about the antioxidant activity of BWE extract by evaluating its ability to remove free radicals and metals. In this regard, various methods were used, including the free radical scavenging activity (ABTS assay, DPPH assay, H₂O₂ assay) and metal scavenging activity (FRAP scavenging assay, cupric iron-reducing antioxidant (CUPRAC) assay, metal chelating activity). These assays provide quantitative measures of the ability of BWE extracts or metabolites to eliminate free radicals and reduce oxidative stress. In these assessments, ascorbic acid and EDTA were used as standards for the assays carried out.

Evaluation of free radical scavenging activity

ABTS radical scavenging activity The ability of BWE extract to scavenge ABTS radicals was assessed by using an ABTS (2,2'-Azinobis [3-ethylbenzothiazoline-6-sulfonic acid]-diammonium salt) radical scavenging assay. In this assay, the ABTS+cation radical was produced by the reaction between a 7 mM ABTS solution and a 2.4 mM potassium persulfate (1:1) solution [92]. The antioxidant activity of the BWE extract was evaluated by assessing the ability to eliminate the ABTS radical, compared to ascorbic acid (Fig. 7). Our results revealed that BWE extract exhibits a dose-dependent capability to scavenge ABTS radical with highest scavenging activity at 500 µg/ml and 1000 µg/ml (76.98%, and 88.39%, respectively). The lowest percentage of scavenging activity was observed at 2 μ g/ml with 1.98%. Under the experimental conditions given, the BWE extract exhibited antioxidant activity with an IC₅₀ value of $153.51 \pm 3.97 \mu g/ml$,

Number	Compound Name	Chemical Formula	Retention Time (RT)	Area %	R.I
1	(+)-3-Carene	C ₁₀ H ₁₆	4.21	1.13	948
2	Allo-Ocimene	C ₁₀ H ₁₆	5.53	12.61	993
3	p-Mentha-1,4-dien-7-ol	C ₁₀ H ₁₆	5.80	0.73	1240
4	Sabinene	C ₁₀ H ₁₆	6.09	4.74	897
5	4- cis-(.±)-Thujanol	C ₁₀ H ₁₈ O	6.55	0.67	1311
6	4- cis-Thujanol	C ₁₀ H ₁₈ O	6.74	10.56	1311
7	Terpinen-4-ol	C ₁₀ H ₁₈ O	8.52	5.97	1137
8	(S)-(-)-α-Terpineol	C ₁₀ H ₁₈ O	8.78	2.49	1158
9	1-methyl-4-(1-methylethylidene)- cyclohexanol	C ₁₀ H ₁₈ O	10.28	4.88	1346
10	cis-Sabinene hydrate acetate	C ₁₂ H ₂₀ O ₂	10.33	4.41	1041
11	1,6-Octadien-3-ol, 3,7-dimethyl-, acetate	C ₁₂ H ₂₀ O ₂	10.47	0.93	1082
12	iso-3-Thujyl acetate	C ₁₂ H ₂₀ O ₂	11.46	1.98	1219
13	Caryophyllene	C ₁₅ H ₂₄	14.59	5.66	1494
14	γ-Elemene	C1 ₅ H ₂₄	16.37	3.29	1431
15	(-)-Spathulenol	C ₁₅ H ₂₄ O	18.05	1.49	1536
16	(Z)-2-(9-Octadecenyloxy)-ethanol	C ₂₀ H ₄₀ O ₂	24.16	2.09	2068
17	13-Heptadecyn-1-ol	C ₁₇ H ₃₂ O	24.98	0.79	1971
18	Palmitic acid methyl ester	C ₁₇ H ₃₄ O ₂	25.62	1.51	1878
19	n-Hexadecanoic acid	C ₁₆ H ₃₂ O ₂	26.38	8.81	1968
20	Hexadecanoic acid ethyl ester	C ₁₈ H ₃₆ O ₂	26.95	1.25	1968
21	Dasycarpidan-1-methanol acetate	C ₂₀ H ₂₆ N ₂ O ₂	27.72	1.66	2453
22	Oleic Acid	C ₁₈ H ₃₄ O ₂	27.94	1.15	2175
23	6,9,12-Octadecatrienoic acid methyl ester	C ₁₉ H ₃₂ O ₂	29.43	2.73	2447
24	(Z,Z,Z)-9,12,15-Octadecatrienoic acid, 2,3-dihydroxypro- pyl ester	C ₂₁ H ₃₆ O ₄	29.52	6.03	2586
25	Octadecanoic acid, 2,3-dihydroxypropyl ester	C ₂₁ H ₄₂ O ₄	30.03	2.70	2681
26	Retinol	C ₂₀ H ₃₀ O	30.96	2.82	2238
28	(3ß,24S)-Stigmast-5-en-3-ol,	C ₂₉ H ₅₀ O	33.96	1.20	2657
29	1,2-Benzenedicarboxylic acid	C ₂₄ H ₃₈ O ₄	35.80	2.19	2586
30	3-Hydroxyspirost-8-en-11-one	$C_{27}H_{40}O_4$	37.69	1.16	3044

Table 2 List of the main essential oils and volatile secondary metabolites in BWE extract

as compared to the standard (IC₅₀=10.66±0.89 µg/mL). The BWE extract demonstrated an ABTS scavenging activity of 88.39% at 1000 µg/ml, ascorbic acid showed an activity of 95.46% at the same concentration. These results indicate that the BWE extract possesses considerable scavenging activity toward the ABTS radical, as compared to the ascorbic acid standard.

DPPH radical scavenging activity The DPPH (2,2-Diphenyl-1-picrylhydrazyl) scavenging assay is a robust assay to evaluate antioxidant activity by eliminating the DPPH radical. The DPPH free radical interacts with an odd electron to give a noticeable absorbance at a specific wavelength (517 nm). As shown in Fig. 7, the BWE extract demonstrated considerable and dose-dependent scavenging activity toward the DPPH radical, with the highest activity of 87.13% at 1000 μ g/ml. On the contrary, the lowest percentage of inhibition is

observed at a concentration of 2 µg/ml, with a value of 0.97%. In this assay, the BWE extract exhibited antioxidant activity with an IC_{50} of 195.75 ± 5.91 µg/ml, as compared to ascorbic acid with an IC_{50} of 10.21 ± 0.77 µg/ml. Our findings indicated that the BWE extract has DPPH scavenging activity compared to that of ascorbic acid, as indicated by the activity at 1000 µg/ml (87.13% and 98.65%, respectively). Furthermore, at lower concentration (3.9 µg/ml), the BWE extract showed a percentage of DPPH scavenging of 2.65%, while ascorbic acid showed 1.04% activity. These findings indicate that the BWE extract exhibits considerable antioxidant activity by scavenging the DPPH radical.

 H_2O_2 radical scavenging activity The H_2O_2 scavenging assay was also applied to test the antioxidant activity of the BWE extract at different concentrations (2, 3.9, 7.8, 31.25, 62.5, 250, 500, and 1000 µg/ml) by neutralizing hydrogen



Fig. 7 The dose-dependent activity of BWE extract to scavenge the radical of DPPH (**A**), ABTS (**B**) and H_2O_2 (**C**). The radical scavenging activity (IC_{50}) of the BWE extract (**D**). Data were presented as mean values with standard deviations in triplicate

peroxide, a reactive oxygen species that can cause oxidative damage. As indicated in Fig. 7, our analysis indicated that the BWE extract shows a moderate and dose-dependent radical scavenging activity against H2O2 radicals. At high concentrations (1000 µg/ml, 500 µg/ml, and 250 µg/ ml), BWE extract possessed considerably high percentages of scavenging H2O2 with 75.42%, 63.14%, and 52.95%, respectively. However, at low concentration (2 μ g/ml), the percentage of H₂O₂ scavenging decreased significantly to 0.85%. IC50 assessment revealed that BWE exhibits an IC_{50} of 230.67±5.98 µg/ml, while ascorbic acid showed an IC_{50} of 14.77 ± 0.69 µg/ml, at similar conditions. Compared to ascorbic acid, the BWE extract showed moderate H_2O_2 scavenging activity, as indicated by the activity at 1000 μ g/ ml (75.42%, compared to 95.73% for ascorbic acid). These findings further support the antioxidant potential of BWE extract by scavenging for radicals.

Assessment of metal scavenging activity

FRAP scavenging assay The FRAP (Ferric reducing antioxidant power) scavenging assay was performed to test the antioxidant activity of BWE extract. As depicted in Fig. 8, the BWE extract exhibited considerable FRAP scavenging activity, with the highest activity recorded at 1000 µg/ml and 500 µg/ml (63.89% and 54.93%, respectively). On the other hand, the lowest activity was observed at 2 µg/ml, with a minimal scavenging activity of 0.51%. These results underscore the concentration-dependent nature of the scavenging activity of the plant extract. Under the experimental conditions specified, the BWE extract showed antioxidant activity with an IC₅₀ of 425.43 ± 9.61 µg/ml, as compared to ascorbic acid (IC₅₀=20.89 ± 1.25 µg/ml). These results were supported by comparing activity at high concentration (1000 µg/



Concentration [µg/mL]

Fig. 8 The dose-dependent activity of BWE extract to chelate the metal of FRAP (A), CUPRAC (B), and metal chelating activity (C). Data were presented as mean values with standard deviations in triplicate

ml). While the BWE extract showed a FRAP scavenging percentage of 63.89%, ascorbic acid showed 94.19% activity. These results indicate that the BWE extract has moderate metal scavenging activity.

Cupric Iron Reducing Antioxidant (CUPRAC) assay Next, we evaluated the metal scavenging activity of the BWE extract by conducting the CUPRAC assay. As shown in Fig. 8, the CUPRAC assay revealed that the BWE extract exhibits the highest scavenging activity (34.81%) observed at a concentration of 1000 μ g/ ml, while the lowest percentage of scavenging activity (0.09%) was observed at 2 μ g/ml. These results indicate a comparatively weak metal chelating activity of BWE extract under the given experimental conditions. On the contrary, EDTA (positive control) demonstrated significant metal chelating activity with an IC₅₀ of 232.28 ± 4.73 µg/ml, under the same experimental conditions. At a concentration of 1000 µg/ml, EDTA showed a chelating activity of 69.32%, while the BWE extract showed 34.81% at the same concentration. These findings provide a comparative perspective on the moderate ability of the BWE extract to scavenge metals.

Metal chelating activity assay Finally, the evaluation of the antioxidant activity of the BWE extract was conducted by applying the metal chelating activity assay.

As shown in Fig. 8, the metal chelating activity assay revealed the nonsignificant activity of the BWE extract. Although the BWE extract exhibited dose-dependent activity, the highest activity observed was 29.76% at 1000 µg/ml concentration. These findings further suggest the relatively low metal chelating activity of the BWE extract, under the specified experimental conditions. On the contrary, EDTA showed dose-dependent and potential metal chelating activity with an IC_{50} of $21.16 \pm 2.08 \ \mu g/ml$. Furthermore, at a concentration of 1000 µg/ml, EDTA showed a metal chelating activity of 95.13%, while at a concentration of 2 µg/ml, 19.87% was observed. Taken together, the evaluation of antioxidant activity by several assays reveals that the BWE extract possesses potential antioxidant activity, which could be mainly attributed to its ability to scavenge free radicals.

Antimicrobial assessment

Despite the increasing availability of antimicrobials, the rate at which microorganisms develop drug resistance is concerning. Therefore, more research is needed to identify alternative treatments that are both more efficient and safer than the drugs currently available (30). Toward this, we assessed the antimicrobial activity of BWE extract against 29 microbial strains mainly (13 fungal strains, 8 Gram positive bacterial strains, and 8 Gram negative bacterial strains).

Evaluation of antifungal activity

Assessment of antifungal activity was performed using the well-established diffusion agar technique. The antifungal activity of the BWE extract was evaluated against

Table 3 The antifungal activity of BWE extract at 10mg / mlagainst the growth of 13 fungal stains

Fungal strain	Inhibition Zone (mm)			
	BWE extract	Ketoconazole		
Aspergillus fumigatus	NA	17		
Aspergillus niger	NA	15		
Candida albicans	NA	20		
Penicillium aurantiogriseum	NA	25		
Syncephalastrum racemosum	NA	15		
Penicillium marneffeii	NA	13		
Cryptococcus neoformas	NA	25		
Candida lipolytica	NA	18		
Penicillium expansum	NA	17		
Penicillium italicum	NA	18		
Fusarium moniliform	NA	21		
Trichophyton rubrum	NA	22		
Geotrichum candidum	NA	14		

the growth of 13 different fungal strains at a concentration of 10 mg / ml. In our assays, ketoconazole was used as a positive antifungal drug at 100 μ g/mL concentration. As shown in Table 3, the BWE extract displayed a non-significant antifungal effect against the examined strains at the applied concentrations. These findings indicate that the BWE extract has no considerable activity toward the microorganisms tested for fungi.

Evaluation of antibacterial activity

The diffusion agar technique was also utilized to evaluate the antibacterial activity of the BWE extract. In this regard, the BWE sample was tested at 10 mg/ml concentration, while gentamycin was used as a positive control at 4µg/ml concentration. As shown in Table 4, the BWE extract exhibited considerable antibacterial activity against a set of Gram-positive bacteria, including Enterococcus faecalis, Streptococcus mutants, Staphylococcus epidermidis, and Micrococcus sp. strains. However, BWE extract did not show significant antibacterial activity against the strains of Staphylococcus aureus, Bacillus subtilis, Bacillus cereus and MRSA. These results indicate that BWE extract could be a natural medicinal extract available for the treatment of infections caused by susceptible bacterial strains, particularly Enterococcus faecalis, Streptococcus mutants, Staphylococcus epidermidis, and Micrococcus sp..

Furthermore, the BWE extract showed notable antibacterial efficacy against the growth of Gramnegative bacteria, including *Klebsiella pneumonia* and *Porphyromonas gingivalis* (Table 5). In particular, BWE demonstrates a substantial capacity to prevent the growth of *Porphyromonas gingivalis* bacteria, surpassing the effectiveness of the antibiotic *Gentamycin*. *Porphyromonas gingivalis* is recognized as a

Table 4 Antibacterial activity of BWE extract against a panel of

 Gram-positive bacterial strains
 Figure 1

Gram Positive Bacteria	Inhibition Zone (mm)			
	BWE extract	Gentamycin		
Staphylococcus aureus	NA	24		
Bacillus subtilis	NA	26		
Bacillus cereus	NA	25		
Staphylococcus epidermidis	9	28		
Micrococcus sp.	8	21		
Streptococcus mutants	9	22		
Methicillin-Resistant Staphylococ- cus aureus (MRSA)	NA	15		
Enterococcus faecalis	9	26		

Table 5 The antibacterial activity of BWE extract against a panel of Gram-negative bacterial strains

Gram Negative Bacteria	Inhibition Zone (mm)			
	BWE extract	Gentamycin		
Enterobacter cloacae	NA	30		
Salmonella typhimurium	NA	17		
Escherichia coli	NA	30		
Klebsiella pneumonia	8	21		
Proteus vulgaris	NA	25		
Serratia marcenscens	NA	25		
Pseudomonas aeruginosa	NA	27		
Porphyromonas gingivalis	17	18		

causative agent of periodontal diseases. Our findings underscore the considerable antibacterial potential of the BWE extract against the *Porphyromonas gingivalis* strain (Fig. 9).

Evaluation of antitumor activity

Finally, we assessed the antitumor activity of the BWE extract by screening its antiproliferation activity against a panel of 15 cell lines. The activity of the BWE extract was evaluated in triplicate at different concentrations, including 0.25, 0.5, 1, 2, 3.9, 780, 15.60, 31.25, 62.50, 125, 250, and 500 μ g/mL. As shown in Fig. 10, the BWE extract exhibited considerable and dose-dependent antiproliferative activity toward the panel of texted cancer

cell lines. The BWE extract demonstrated moderate cytotoxic activity towards SKOV, RD, HeLa and Hep2 cells with $IC_{50} > 100 \mu g/mL$, while showing considerable activity toward PC-3, A-431 and Panc-1 cells with an IC_{50} of 55-70µg/mL. The evaluation of the selectivity index indicated that the extract displays a selectivity index of 1 towards SKOV, RD, HeLa, and Hep2 cancer cells, compared to WI-38 cells, indicating that the BWE extract does not possess significant selectivity toward these cell lines. On the other hand, BWE extract displayed selectivity index>2 toward PC-3, A-431, and Panc-1 cells. Furthermore, the BWE extract demonstrated substantial antiproliferative activity toward the growth of HCT-116, A-549, MDA-MB-231, HepG-2, MCF-7, CACO2 and HT-29 cell lines with IC₅₀ ranging from 15–30 μ g/mL, and selectivity index>3, as compared to WI-38 cells. Among the cells examined, the BWE extract exhibited the most cytotoxic activity toward HCT-116, A-549, MDA-MB-231cells with an IC₅₀ of 15.11, 15.32, 15.79 µg/mL, respectively, and a selectivity index ~ 8. Furthermore, the evaluation of the cytotoxicity of the BWE extract on WI-38 cells (normal cells) indicated that the BWE extract possesses a nonsignificant cytotoxic activity on WI-38 cells with IC₅₀ 119.62 µg/mL. Collectively, these findings indicate that the BWE extract possesses potential antitumor activity and could be considered as a natural plant-based antitumor supplement.



Fig. 9 The growth inhibition zones of Staphylococcus epidermidi. (A), Streptococcus mutants (B), Enterococcus faecalis (C), Micrococcus sp. (D) Klebsiella pneumonia (E) and Porphyromonas gingivalis (F) after treatment with BWE extract



Fig. 10 A Evaluation of the antiproliferative activity of BWE extract against a panel of 14 cancer cell lines and normal human lung fibroblast WI-38 cells. B IC_{50} values of the antiproliferative activity of the BWE extract against examined cell lines. C The selectivity index of the BWE extract toward cancer cells compared to WI-38 cells. Data were presented as mean values with standard deviations in triplicate

Discussion

Various global markets have used various medicinal plants in the treatment of health conditions such as hepatotoxicity, cardiovascular disease, neurotoxicity, gastrointestinal diseases, and even cancer [93]. The various effects of the therapeutic effectiveness of most medicinal plants are believed to be caused by the complex chemical composition of the active phytochemical components [94, 95]. The presented study investigated the phytochemical profile of the acetone extract of BWE plant from the plant of Germany. We extensively explored the bioactivity of the BWE extract by assessing antioxidant, antimicrobial, and anticancer activity. In agreement with Khan et al., the GC–MS analysis of BWE extract demonstrated a set of metabolites, including: (-)-Spathulenol, Cholestan-3-ol, 2-methylene-, ($3\dot{a},5\dot{a}$)-, Hexadecanoic acid, ethyl ester, Ethanol, 2-(9-octadecenyloxy)-, (Z)-, Hexadecanoic acid, methyl ester and Linoleic acid ethyl ester [96]. On the other hand, Khan et al. study detected other unique metabolites, including carophyllen oxide, β -Himachalenoxide, 3, 7, 11, 15-Tetramethyl-2-hexadecen-1-ol and 9, 12-Octadecadienoic acid, ethyl ester, despite sharing the same chemical formula as our compounds. Differences in the metabolites identified through gas chromatography (GC) analysis can stem from various factors, even when substances with identical chemical formulas are examined. These factors include column temperature, mobile phase flow rate, column length, and polarity of the compounds [97]. In addition, GC–MS analysis revealed a wide spectrum of metabolites within the BWE extract, each of which possesses different chemical structures. In particular, the presence of metabolites such as eicosanoic acid, methyl ester (also known as methyl eicosanoate), oleic acid, linoleic acid ethyl ester, and caryophyllene suggests the potential occurrence of bioactive molecules with known anticancer activities. For cis-13-octadecenoic acid, 9-octadecenoic acid (cis and trans isomers), hexadecanoic acid, methyl ester, methyl stearate, and terpinen-4-ol have been reported to possess antimicrobial properties. Furthermore, metabolites such as (-)-alpha-terpineol, (-)-spathulenol, and certain cyclopropane derivatives have been investigated for their antioxidant properties. It is essential to recognize that the biological activities of these metabolites are contextdependent and influenced by factors such as concentration, formulation, and the specific biological model used for the assessment.

Furthermore, phytochemical screening investigations have been carried out to gain more information on the categories of metabolites that are present in BWE, such as amino acids, flavonoids and phenolic acids, vitamins, and total nutrients. According to our results, the BWE extract displayed significant concentrations of the following metabolites: amino acids (especially glutamic acid (GLU) and aspartic acid (ASP)), flavonoids (such as catechin and luteolin), phenolic acids (such as catechol and syringenic), vitamins, fatty acids, saponins, tannins, carbohydrates, and protein. Joshi et al. 2014 showed that flavonoids, tannins, volatile chemicals, fatty acids, polysaccharides, isolectins, sterols, terpenes, protein, vitamins, and minerals are among the primary chemical components of Urtica dioica [98]. Furthermore, Kasouni et al. 2021 showed that the phytochemical analysis of the BWE extract demonstrated the presence of the following metabolites: coumarins, ketoses, carbohydrates, flavonoids, saponins, and resins. On the other hand, the plant extract under scrutiny revealed the absence of several metabolites, including steroids, proteins, tannins, alkaloids, and terpenoids [99]. Minor variations in the composition of the phytochemical profile may depend on the specific part of the plant subjected to extraction and the geographical location (demography) where the plant was harvested.

Our phytochemical analyses suggest potential biological activities of the BWE extract, including antioxidant, antimicrobial, and anticancer activities. The assessment of TAC revealed that the BWE extract possesses considerable antioxidant activity. Accordingly, we extended our investigation to explore the free radical scavenging activity and the metal chelating activity of BWE extract. The BWE extract showed significant scavenging activity for the free radicals ABTS, DPPH, and H_2O_2 , compared to the ascorbic acid standard. Khan et al. 2023 [96] and Yousuf et al. 2022 [100] reported similar activity for the DPPH scavenging assay. Furthermore, Yousuf et al. 2022 (78) demonstrated strong scavenging activity for the ABTS assay. Also, Carvalho et al. [101] and Kukrić et al. 2012 [102] showed that Urtica dioica has strong antioxidant activity using ABTS, DPPH, and FRAP assays. The effective antioxidant activity of the BWE extract could be correlated with the inclusion of phenolic metabolites and their analogues [98, 103–105]. According to previous reports, the strong antioxidant activity of plant extract could also be associated with the presence of ascorbic acid, chlorophylls, carotenoids, and other metabolites [106–108]. In addition, the variation in ROS scavenging activity observed in different research studies may be ascribed to differences in pre- and post-harvest conditions. Variables such as soil types, organic manure types, weather conditions, and genetic differences can markedly affect variability in nutritional accumulation, consequently influencing biological activity [106]. For example, various agroecological conditions led to the identification of fluctuating amounts of nutrients in nearly identical species of Urtica dioica [103]. Although the BWE extract exhibited lower percentages compared to the reference EDTA in the CUPRAC and metal chelating activity assays, these values are not considered insignificant for a natural product [109]. Beyond mitigating oxidative stress by inhibiting the Fenton reaction, the chelating activity of plant extracts serves as a preventive measure against bacterial infections. The chelating properties of the extract induce bacteriostasis by extracting iron from the wound site since iron is essential for bacterial growth. The lactoferrin protein shows a comparable mode of action demonstrated by the lactoferrin protein [110].

Recently, microorganisms have begun to exhibit resistance to existing antimicrobials. Consequently, additional studies are imperative to identify alternative medicines that are not only more effective but also secure [111, 112]. Our results showed that the BWE extract does not have a significant antifungal impact against the 13 fungal microorganisms tested, compared to the reference Ketoconazole drug. On the other hand, Taheri et al. 2022 [36] and Gulçin et al. 2004 [113] showed that Urtica dioica extract has considerable antifungal activity against Candida albicans. The Brennnesselwurzel plant extract has also shown antibacterial activity against Gram-positive and Gram-negative bacteria. Our results showed that the BWE extract exhibits significant antibacterial activity against Staphylococcus epidermidi, Streptococcus mutants, Enterococcus faecalis, Micrococcus sp., Klebsiella pneumonia and Porphyromonas gingivalis. Contrary to our findings, the study reported by Zeković et al. 2017 [114] confirmed that Urtica dioica extract has considerable antibacterial activity with a MIC

of 9.76µg/ml against the methicillin-resistant staphylococcus aureus (MRSA) strain. Zenão et al. 2017 [115] revealed that these observed effects were related to their high amount of flavonoids (quercetin) and hydroxycinnamic acids/phenolic acids (chlorogenic, caffeic and rosmarinic acids). Furthermore, Ghaima et al. 2013 [116] also evaluated the antibacterial potential of Urtica dioica extract against various bacteria strains, including Salmonella typhi, Escherichia coli, Staphylococcus aureus, and Bacillus cereus, using cephalothin as standard drug. Their findings showed that the zone of inhibition of the plant extract against Salmonella typhi, Escherichia coli, Staphylococcus aureus, and Bacillus cereus was found to be 22, 10, 20, and 24 mm, respectively. On the contrary, the results of our plant extract did not demonstrate a significant impact against the same bacterial strains. Additionally, Modarresi-Chahardehi et al. 2012 [40] investigated the antibacterial activity of Urtica dioica against Methicillin-resistant Staphylococcus aureus (MRSA) and Bacillus cereus and showed that BWE exhibits MIC values of 16.33 and 8.33 mg/mL, respectively. On the other hand, our findings showed an insignificant impact against both strains. Further, Gulcin et al. 2004 [113] demonstrated that the BWE extract exhibits antibacterial effects against Escherichia coli, Staphylococcus epidermidis and Micrococcus luteus. The disparity in results could be attributed to the preparation of the Urtica dioica extract using different solvents, namely butanol, chloroform, diethyl ether, ethyl acetate, hexane, and methanol. Khan et al. 2023 [96] also tested the antibacterial activity of Urtica dioica against different bacterial strains, such as Escherichia coli, Bacillus subtilis, Staphylococcus aureus and Salmonella typhimurium. Their results showed that BWE has respectable antibacterial activity (1000 g/mL) against the tested bacterial strains when compared to ofloxacin $(0.25 \ \mu g/mL)$. These findings are consistent with Jyoti et al. 2016 [117] and Zeroual et al. 2021 [118], both of whom reported similar results against the selected pathogens. Motamedi et al. 2012 [41] elucidated that the resistance of Gram-negative bacteria, particularly Escherichia coli and Salmonella typhimurium, can be attributed to their distinctive bacterial structure. They possess an outer membrane envelope that functions as a barrier that limits the entry of certain substances. Moreover, their porins control the type and size of molecules allowed to enter their cytoplasm. Some of these resilient species can also produce polysaccharide capsules, which hinder the entry of specific chemicals into the cytoplasm. These findings also implied that Urtica dioica extracts show significant antibacterial activity against Gram-positive species. This suggests the potential use of the plant as a selective treatment for infections, particularly the methicillin-resistant strain of S. aureus (MRSA), which poses a significant threat to human health. Furthermore, Dogruoz et al. 2008 [119] reported that the efficacy of bacteria inhibition or eradication can be affected by the specific plant extract, the solvent used, and the organisms under examination.

In addition to antioxidant assays, we also performed a cytotoxicity assay on 15 different cell lines to test the antitumor activity of BWE extract. The dose-dependent curve for the different cell lines showed that the BWE extract significantly reduces the viability percentage of the cancerous cell lines. In particular, the BWE extract showed substantial cytotoxic activity against HCT-116 (colorectal cancer), A-549 (lung cancer), MDA-MB-231 (breast cancer), HepG-2 (liver cancer), MCF-7 (breast cancer), CACO2 (colorectal cancer) and HT-29 (colorectal cancer) human cell lines, as indicated by IC₅₀ values (15.11, 15.32, 15.79, 21.73, 24.9, 27.68, and 29.89 µg/ ml, respectively). Interestingly, the BWE extract demonstrated a high selectivity index for cancer cell lines compared to WI-38 cells. In line with our findings, Mansoori et al. 2017 [120] showed that the Urtica dioica extract significantly suppressed the cancer cell line with an IC₅₀ value of 31.37 mg/ml against the MCF-7 cell line and 38.14 mg/ml mg/ml against the MDA-MB-231 cell line. They concluded that according to QRT-PCR, Urtica dioica extracts reduced cell migration by upregulating the expression of E-cadherin and downregulating the expression of miR-21, matrix metalloproteinase (MMP) 1, MMP9, and MMP13. Moreover, Yousuf et al. 2022 [100] reported that BWE extract exhibits an IC_{50} of 114 μ g/ml against the HepG-2 cell line. However, our results showed an IC₅₀ of 21.73 μ g/ml against the same cell line, suggesting that the current BWE extract possesses a unique set of secondary metabolites with antitumor characteristics. The difference in anticancer activity could be due to the different solvents that have been utilized in the extraction methodology. The extraction solvent may affect the phytochemical profile of the BWE extract, including flavonoids and tannin [121]. Ghasemi et al. 2016 [122] showed that the BWE extract has antiproliferation activity against human colon cancer HT-29 with an IC₅₀ value of 24.7 μ g/ml. Furthermore, Mohammadi et al. 2016 [123] revealed that the BWE extract exhibits cytotoxic activity against PC-3 with an IC₅₀ value of $15.54 \,\mu$ g/ml. In alignment with these studies, our findings revealed that the BWE extract has IC_{50} of 29.89 µg/ml against HT-29 cells and IC_{50} of 70.24 $\mu g/ml$ against PC-3 cells. Variation in antitumor activity of the BWE extract could be attributed to differences in demography and duration of treatment. In the Ghasemi et al. study, HT-29 cells were treated for 72 h, while in the study by Mohammadi et al. the PC-3 cells were treated for 48 h. Furthermore, Karakol et al. 2022 [46] detected cytotoxic activity of BWE extract against MCF-7 and MDA-MB-231 cells,

as the IC₅₀ of 18 and 14 μ g/ml, respectively. Our results indicated a significant impact against MCF-7 and MDA-MB-231 cells with IC₅₀ of 24.9 and 15.79 µg/ml, respectively. Cytotoxic activity may be due to polyphenolic chemicals, such as flavonoids, which may have anticancer effects through a variety of mechanisms, including antioxidant activity, activation of apoptosis, reduction of cell growth, and cell migration (42). Taking into account our results presented, the acetone BWE extract seems to exhibit considerable antioxidant, antibacterial, and antitumor activities. These activities could be associated with a wide variety of metabolites detected, including amino acids, flavonoids, phenolic acids, vitamins, fatty acids, tannins, carbohydrates, and proteins. More research is needed to explore the therapeutic potential of BWE extract in the animal disease model and to gain more insight into the possible applicability as a medicinal plant supplement against human diseases.

Conclusions

In summary, our study delved into the complete exploration of the 70% acetone extract of Brennnesselwurzel (Urtica dioica L.), revealing its diverse pharmacological potential. The phytochemical analysis showed a complex profile encompassing amino acids, flavonoids, phenolics, volatile oils, lipids, and vitamins. This intricate composition likely contributes to the observed biological activities of the extract. It should be noted that among its attributes, the BWE extract demonstrated robust antioxidant capabilities, as reflected in its substantial total antioxidant capacity and effective free radical scavenging against ABTS, DPPH, and H_2O_2 . The potent antioxidant properties suggest a potential role in mitigating oxidative stress-related conditions. Furthermore, the extract demonstrated pronounced antibacterial efficacy against a variety of microbial strains, including Staphylococcus epidermidi, Streptococcus mutants, Enterococcus faecalis, Micrococcus sp., Klebsiella pneumonia and Porphyromonas gingivalis. This broad-spectrum antibacterial activity positions the BWE extract as a promising candidate for further exploration in the realm of antimicrobial agents. Furthermore, the study revealed compelling antiproliferative effects of the BWE extract against various cancer cell lines. The significant cytotoxic activity of HCT-116, A-549, and MDA-MB-231 cells was particularly striking, emphasizing its potential as an antitumor agent. Importantly, the extract showed a favorable selectivity index, indicating its ability to target cancer cells while sparing normal cells, as evidenced by its nonsignificant cytotoxicity towards WI-38 cells. Collectively, these findings underscore the multifaceted biological activities of BWE extract, which span antioxidant,

antibacterial, and antitumor properties. The observed effects provide a strong reason for further investigations, including in vivo studies, to elucidate the full therapeutic potential of BWE as a natural plant-based supplement for human diseases.

Supplementary Information

The online version contains supplementary material available at https://doi. org/10.1186/s12906-024-04709-6.

Supplementary Material 1: Figure S1: Chromatogram of amino acids detected in the BWE extract by amino acid analyzer; Figure S2: Representative metabolites predominantly detected in the BWE extract by GC-MS analysis; Table S1: List of amino acid detected in BWE extract by Amino Acid Analyzer; Table S2: List of flavonoids detected by HPLC analysis in BWE extract; Table S3: List of phenolics detected by HPLC analysis in BWE extract; Table S4: List of fat-soluble vitamins detected by HPLC analysis in BWE extract; Table S5: List of water-soluble vitamins detected by HPLC analysis in BWE extract; Table S5: List of water-soluble vitamins detected by HPLC analysis in BWE extract; Table S5: List of water-soluble vitamins detected by HPLC analysis in BWE extract.

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Authors' contributions

Amr Elkelish, Essa M. Saied, Sulaiman Lakoh, Mai Labib conceptualization, Supervision, Writing - Original Draft, Writing - Review & Editing, Data Curation, software and resourses. Abdelghafar M. Abu-Elsaoud, Alaa M. Alqahtani;Norah Al harthi, Najwa Al harthi,: Methodology, Investigation, Validation, Data Curation, Writing - Review & Editing,; Methodology; Mohammad El-Nablaway: Project administration; Funding, Project, Review & Editing, Data Curation. All authors have read and agreed to the published version of the manuscript.

Data availability

No datasets were generated or analysed during the current study.

Declarations

Ethics approval and consent to participate Not applicable.

Consent for publication

Not applicable.

Competing interests

The authors declare no competing interests.

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