Inhibition of the B-Lactamases of *Escherichia coli* and Klebsiella aerogenes by Semi-Synthetic Penicillins

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1. A new automated micro-iodometric method is described for screening compounds for inhibitory action against β -lactamase enzymes. 2. Over 1000 semi-synthetic penicillins were tested for inhibitory activity against the β -lactamase of *Escherichia coli* B 11 and 18 showed a fractional inhibition similar to or higher than that of methicillin. 3. The best inhibitors were alkoxy- and halogen-substituted phenyl-, naphthyl- or quinolyl-penicillins. 2-Isopropoxy-1-naphthylpenicillin (BRL 1437) was clearly the best and had a K_i value about 1% of that of methicillin. 4. The inhibition of the β -lactamase of E. coli B 11 by BRL 1437 was shown to be reversible and competitive. The K_t was 0.004 μ M and K_t/K_m with ampicillin and p-hydroxyampicillin (BRL 2333) was about 0.0001. The K_m and V_{max} , values were determined for the β -lactamases of E. coli B 11 and Klebsiella aerogenes A against a variety of penicillins. Cell-bound and solubilized enzymes gave similar K_t and K_m values. 5. BRL 1437 was superior to cloxacillin and methicillin for inhibition of the β -lactamase of live, fully grown cultures of several strains of E. coli and K. aerogenes. Of a group of inhibitors BRL 1437 was the most stable to the β -lactamase of E. coli B11.

Many Gram-negative bacteria produce the enzyme β -lactamase (penicillin amido- β -lactam-hydrolase, EC 3.5.2.6), and this has been shown to be associated with resistance to penicillins, although it is not always the sole cause of resistance (Ayliffe, 1963; Cole & Sutherland, 1966; Percival et al., 1963; Sutherland, 1964; Sabath & Finland, 1968). This association has created interest in inhibitors of the enzyme as penicillin-protecting agents (synergists) or as tools for studying the mechanism of bacterial resistance.

There have been several reports of the inhibition of Gram-negative β -lactamases by methicillin (BRL) 1241), oxacillin(BRL 1400) or cloxacillin(BRL 1621), all compounds selected for their stability to staphylococcal β -lactamase. Initially Jago et al. (1963) and Hamilton-Miller (1963) reported inhibition of enzymes from Pseudomonas pyocyanea and Klebsiella aerogenes respectively, and later Sutherland & Batchelor (1964) and Hamilton-Miller & Smith (1964) demonstrated inhibition of enzymes from strains of Klebsiella, Aerobacter, Escherichia, Pseudomonas and Proteus. Dicloxacillin has been shown to be a good inhibitor of the β -lactamases of Shigella flexneri by Bach et al. (1967), and substituted benzamidocephalosporanic acids and the cephalosporin analogue of cloxacillin have been shown to be inhibitors of the fi-lactamases of Aerobacter aerogenes, Proteus morganii and Enterobacter cloacae (O'Callaghan et al., 1967, 1969).

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It has frequently been reported that high concentrations of these inhibitors were required for good antibacterial synergism. Sutherland & Batchelor (1964) reported that the degree of synergism between their β -lactamase inhibitors and penicillins was inadequate for clinical application, although Sabath et al. (1967) have suggested that the high concentrations of penicillins obtainable in urine may be sufficient for synergism.

There is thus considerable room for improvement on the existing combinations of β -lactamase inhibitors with penicillins or cephalosporins, and it was the object of the present work to see if superior inhibitors could be found. For this purpose large numbers of semi-synthetic penicillins were examined for ability to inhibit cell-free preparations of the β -lactamases of Escherichia coli and K. aerogenes. Because previously available methods were not satisfactory for testing and comparing the inhibitory activities of large numbers of compounds at substrate concentrations that are therapeutically attainable, a new automated micro-iodine procedure was devised.

Materials and Methods

Cultures, penicillins and reagents

The E. coli strains were: Bll and B41 (infantile gastroenteritis, serotype 055, Public Health Labs., Cork, Ireland), 83 (wound swab, Middlesex Hospital), T506 and T507 (urine, St. Thomas's Hospital) and NCIB9465. The K. aerogenes strains were:

A (Beecham typical Klebsiella laboratory strain), B (strain from Middlesex Hospital), C (Imperial College), 1281 (urine, West Middlesex Hospital), R129 (urine, Redhill General Hospital) and C978 (Middlesex Hospital). The Klebsiella oxytoca strains were T200 (St. Thomas's Hospital) and F (Beecham Laboratory strain). Proteus vulgaris A and Pseudomonas aeruginosa A were Beecham Laboratory strains. Bacillus cereus 569/H was NCIB8933. Staphylococcus pyogenes (var. aureus) A was ^a Beecham Laboratory strain and strain H was from Redhill General Hospital.

All novel semi-synthetic penicillins and derivatives were sodium salts synthesized in the Chemistry Department of Beecham Research Laboratories. Sodium salts of benzylpenicillin, ampicillin (BRL 1341), carbenicillin (BRL 2064), methicillin (BRL 1241) and cloxacillin (BRL 1621) were commercial materials available from Beecham Research Laboratories. Sodium nafcillin (BRL 1383) was obtained from Wyeth Laboratories Inc., Philadelphia, Pa., U.S.A. Penicilloic acids were prepared by NaOH hydrolysis of the corresponding penicillins and freeze-drying (Cole & Sutherland, 1966).

Stock phosphate buffer (pH7.0, 0.5M) was prepared by dissolving $24.3g$ of KH_2PO_4 and $56.0g$ of K_2HPO_4 in 1 litre of water containing 1 ml of butyl acetate as preservative.

Laboratory chemicals were purchased from British Drug Houses Ltd., Poole, Dorset, U.K., and bacterial cultures were grown in nutrient broth no. 2, prepared as directed by the manufacturer (Oxoid Ltd., London S.E.1, U.K.).

Preparation of β -lactamases

Small quantities of cultures were grown in 100ml volumes of nutrient broth in 500 ml conical flasks on a rotary shaker (280rev./min, 1 in radius) at 37° C for 18h and cells were collected by centrifugation at 38000g for 15min.

Larger quantities of E . coli B11 were prepared as follows. A portion (1Oml) of an overnight nutrientbroth culture was used to inoculate 3 litres of sterile nutrient broth. This was incubated for 22.5h at 37°C in a 5-litre conical flask. During growth, the culture was aerated at 3 litres/min, 0.2 ml of 10% (v/v) Pluronic L.81 antifoam in soya-bean oil (Jacobson, Van den Burg and Co. Ltd., London W.C.1., U.K.) being used to prevent foaming. Cells were harvested at ²¹ OOOg with an MSE continuous-action rotor at 5°C and a flow rate of 60ml/min. Cells were removed from the collection basket as a slurry and spun down at 38000g for 15min.

Cells from large- and small-scale preparations were washed by resuspending in 0.02vol. of 0.05Mpotassium phosphate buffer, pH7, and re-centrifuging for 15min at 38000g. Finally the cells were resuspended in 0.02vol. of water, chilled in an ice bath and subjected to ultrasonic treatment for 10-20min. Alternate 2.5min periods were used at the two frequencies of maximum amplitude (i.e. 21 and ²⁶ kHz) of an MSE 100W ultrasonic disintegrator. Cell debris was finally spun down by centrifugation for 30min at 38000g and the supernatant was used as stock enzyme preparation, being deep frozen when not in use.

In a reaction mixture containing a final dilution of 1 in 2500 of the E. coli B11 stock enzyme preparation and 40μ g of ampicillin/ml the rate of hydrolysis was $0.55 \,\mu$ g/min per ml at 37°C and pH7.0 as determined by the starch-iodometric assay. The velocity for the K. aerogenes A stock enzyme preparation at ^a final dilution of 1 in 750 was 0.46 μ g/min per ml in a similar reaction mixture. The stock enzyme preparations of E. coli B11 and K. aerogenes A had activities of 4.5 and 1.14 units/ml respectively; one unit being the amount that will catalyse the transformation of 1μ mol of ampicillin/min under the above conditions.

Hydroxylamine assay

The hydroxylamine assay described by Batchelor et al. (1961) was used to determine the purity of test penicillins before evaluation. Purity was expressed in terms of sodium benzylpenicillin, with correction for molecular weight. Some of the inhibitors were very old preparations and gave poor purity on the above basis, but molar extinction does vary with structure. e.g. ampicillin gives a high assay and carbenicillin a low one in terms of benzylpenicillin.

Bioassay and determination of minimum inhibitory concentrations

Bioassay of penicillins was carried out as described by Sutherland (1964). Minimum inhibitory concentration values were determined by Mr. R. Sutherland of these laboratories using nutrient agar.

Starch-iodometric assay

The method, as described by Novick (1962), modified for use with the Unicam SR. 800 double-beam spectrophotometer, was used for measurement of the rate of formation of penicilloic acid. Typical reaction mixtures consisted of 1.Oml of starchiodine reagent $+1.0$ ml of ampicillin solution (in the range $0.25-25 \mu g/ml$ final concentration)+0.1ml of enzyme solution+0.1 ml of inhibitor solution or 0.1 ml of buffer, and the volume was made up to 2.5 ml with buffer. Reaction blanks were set up by replacing the enzyme with buffer, because ampicillin solutions absorb iodine very much more than the enzyme does. Starch-iodine reagent, ampicillin and buffer were left in the constant-temperature cell housing for 5min to

warm up to a reaction temperature of 37° C before addition of the enzyme. Reactions were followed at 590nm over a 10min period. All solutions were made up in the 0.05M-phosphate buffer.

The starch-iodine solution was prepared as described by Novick (1962), except that BDH AnalaR soluble starch was used. Experiments showed that the starch-iodine complex had an absorption maximum at 590 nm and a molar extinction coefficient (ϵ) at this wavelength of 29200. By using this value and also the number of equivalents of iodine reacting with 1 mol of α -aminobenzylpenicilloic acid ($n = 6.7$, measured by titration), reaction velocities were transformed from a change in extinction per unit time $(\Delta E/\Delta t)$ into μ g of ampicillin/ml hydrolysed per unit time (v), by using the expression $v = \frac{M \cdot \Delta E \times 10^3}{n \cdot \epsilon \cdot l \cdot \Delta t}$, where $M =$ molecular weight of ampicillin = 371.4 and $l =$ optical path length = 1 cm. Values of *n* for the penicilloic acids of other penicillins were found to be as follows: benzylpenicillin 6.4, carbenicillin 9.3, p -hydroxy- α -aminobenzylpenicillin (BRL 2333) 6.85, α -carboxy-3-thienylmethylpenicillin (BRL 2288) 8.4.

K_m and K_l values

For determinations of K_m and V_{max} , eight substrate concentrations [SI were used, generally in the range $1-100\,\mu$ g/ml. The initial velocity (v) of reaction was determined at pH7 and 37°C by using the above starch-iodometric method and expressed as ΔE_{590} / unit time. The top velocity for ampicillin was about $1 E_{590}$ unit/200s, or 0.57 μ g/min per ml. The intercept on the abscissa of a graph of $1/v$ against $1/[S]$ gave $-1/K_m$ and the intercept on the ordinate gave $1/V_{\text{max}}$.

For K_i determinations two inhibitor concentrations were used each with four substrate concentrations. The substrate concentrations were in the range $8-40 \mu$ g of ampicillin/ml and the inhibitor concentrations, differing by a factor of two, were chosen to give inhibition in the region of $30-60\%$ as judged by preliminary experiment. The enzyme concentration was kept constant and was such as to give the velocity stated above. Initial velocity, v_i , at pH7 and 37°C was determined by the above starch-iodometric method. For each inhibitor concentration [I] a graph was plotted of $1/v_i$ against 1/[S] and the intercept on the abscissa, $-1/K_m$ $(1 + [I]/K_i)$ used to calculate K_i . The lines for the two inhibitor concentrations intersected on the ordinate and gave K_t values usually within 10% of one another.

Automated micro iodine assay

Direct adaption of the above starch-iodometric assay for use with a Technicon AutoAnalyzer proved

Fig. 1. Flow diagram and manifold design for automated micro iodine method used to screen for β -lactamase inhibitors and measure fractional inhibition

The trace shows compounds acting as substrates, non-substrates and inhibitors of E. coli B11 β -lactamase with ampicillin (10 μ g/ml) as substrate. E_{288} was 0.9 without added enzyme and 0.28 after β -lactamase action. In the presence of inhibitor a higher final value of E_{288} was observed, and a peak appears on the trace. Sample numbers 1-11 were 1μ g/ml of the following compounds (structures in Table 1): methicillin (1), benzylpenicillin (2), BRL ¹²²² (3), BRL ¹⁰⁷¹ (4), methicillin (5), BRL ¹²³⁴ (6), BRL ⁹⁵⁷ (7), BRL ²⁸⁵² (8), methicillin (9), BRL ¹³³⁶ (10), BRL ¹⁷⁰⁷ (11). Samples 12, ¹³ and ¹⁴ were water and samples 15-20 were 3-isopropoxy-4-quinolylpenicillin (BRL 3215) at the concentrations 0.05, 0.1, 0.25, 0.50, 0.75 and 1.0μ g/ml respectively. For experimental details see the Materials and Methods section.

unsatisfactory because incomplete mixing between the iodine solution and the colloidal starch solution resulted in non-stoicheiometric reaction. However, an absorption spectrum of iodine solution in potassium iodide (tri-iodide ion) was found to possess two distinct peaks, with λ_{max} at 288nm and 350nm. Graphs of extinction against iodine concentration at these wavelengths were straight lines passing through the origin, and gave molar extinction coefficients of 3.33×10^4 at 288nm and 2.38×10^4 at 350nm. The height of the 288 nm absorption peak was used to measure the iodine remaining after reaction with penicilloic acid and this was made the basis for an automated screen for E. coli B11 β -lactamase inhibitors, the manifold design and flow diagram for which are shown in Fig. 1.

Solutions of compounds under test as inhibitors were automatically sampled and diluted before being mixed with enzyme and substrate. The final concentrations of test compound (purity not corrected) and substrate were respectively 1 and $10\,\mu$ g/ml. The final enzyme concentration was sufficient to hydrolyse about 1μ g of ampicillin/min. After incubation for 7min at 37°C iodine was fed into the system to react with the penicilloic acid formed. The iodine reagent consisted of about 0.04mm-iodine in 4% (w/v) KI and 0.05 M-potassium phosphate buffer.

A test compound acting as an inhibitor decreases the amount of penicilloic acid formed in the reaction between ampicillin and enzyme, causing an increase in residual iodine. The changes in concentration of iodine were recorded continuously at 288nm by passing the reaction stream through a flow cell in a Beckman DB spectrophotometer connected to ^a stripchart recorder. Compounds acting as inhibitors showed up as a series of peaks on the strip chart. We found it convenient to alternate each test compound with a water-wash sample and to use methicillin as a marker after every fourth test sample. This facilitated identification of test compounds and indicated any drift in output signal. Compounds giving higher inhibition peaks than methicillin were selected for further study.

An example of a screen trace is shown in Fig. 2. The trace starts at E_{288} 0.9, which was set by the concentration of the iodine reagent. When enzyme was added to the reaction stream the extinction decreased. This change in extinction is proportional to the amount of α -aminobenzylpenicilloic acid formed during the reaction period of 7min and is a measure of the initial velocity of the reaction (v) . When an inhibitor was fed into the system the extinction increased as a result of decreased penicilloic acid formation. The difference in extinction between the inhibitor peak and the original base-line gives a measure of the inhibited velocity (v_i) .

Results

$Screen for \, \beta$ -lactamase inhibitors

Over 1000 semi-synthetic penicillins and their derivatives were screened by using the automated micro iodine assay with ampicillin as substrate and the β -lactamase of E. coli B11. Some 64 compounds showed various degrees of inhibitory activity. Fig. 2 is the trace obtained for a small collection of such compounds and illustrates the fractional inhibition (i) given by different substances. Fractional inhibition is defined as

$$
i=\frac{v-v_i}{v}
$$

where v and v_i are respectively initial reaction rates in the absence and presence of test inhibitors. Thus methicillin causes a change in extinction from 0.29 to 0.66 and hence has a fractional inhibition of 0.61. Benzylpenicillin was next tested and the trace shows a decrease in extinction to 0.26, indicating that this substance is acting as a secondary substrate and not an inhibitor. l-Ethyl-1-phenylpropylpenicillin (BRL 1222) was the third test compound and shows slight inhibition. This was followed by triphenylmethylpenicillin (BRL 1071) and the trace registers no signal, signifying that this compound is neither an inhibitor nor a substrate for the enzyme. The structures of these and subsequent test inhibitors are given in Tables ¹ and 4. The three methicillin reference samples give an indication of the accuracy and reproducibility of the method. Fig. 2 also shows (samples 15-20) the relationship between fractional inhibition and inhibitor concentration for 3-isopropoxy-4-quinolylpenicillin (BRL 3215) from which the concentration giving 50% inhibition (I_{50}) may be determined.

Structure-inhibition relationships

The compounds in Table ¹ were selected to illustrate the relationship between degree of inhibition and chemical structure. Analogues of BRL 2852, BRL ¹³³⁶ and BRL ¹⁷⁰⁷ were examined in more detail for their inhibitory activity against β -lactamase from E. coli B11 and K. aerogenes A and the K_i results, with ampicillin as substrate, are given in Tables 2, 3 and 4.

Penicilloic acids derived from 6-aminopenicillanic acid, BRL 1437, benzylpenicillin, ampicillin, methicillin or cloxacillin had no inhibitory activity at a concentration of $1.0\,\mu$ g/ml. Similarly, the following analogues of penicillin had no inhibitory activity: DL-alanyl-DL-alanine, benzoxycarbonyl derivatives of valine, valylvaline, valylvalylvaline and valylvalylvalylvaline, D-biotin, 8-hydroxypenillic acid (NN'-dibenzylethylenediamine salt), 2,6-dimethoxy-3,5-di-iodobenzoic acid (side chain of BRL 1540).

Nature of the inhibition

Experiments with the starch-iodometric assay showed that a constant degree of inhibition of the hydrolysis of ampicillin by E. coli B11 β -lactamase with BRL ¹⁴³⁷ was attained within 100s. Further experiments showed that this inhibition could be overcome by addition of excess of ampicillin. Such results led us to believe that inhibition is reversible and probably competitive in nature. The reversibility of the inhibition was confirmed by a dialysis experiment (Table 5).

The competitive nature of the inhibition was proved by classical kinetic methods. The effect of a range of ampicillin concentrations on the inhibition brought about by 2-isopropoxy-1-naphthylpenicillin (BRL 1437) and methicillin was investigated by the starchiodometric method with E. coli B11 enzyme. Fig. ³ shows that BRL ¹⁴³⁷ acted as ^a competitive inhibitor, since the maximum reaction velocity was unaffected by the presence of BRL 1437. The V_{max} . with methicillin was slightly decreased, but inhibition was mainly competitive.

Kinetic constants

The kinetic constants K_m , V_{max} , V_{max}/K_m and K_l (BRL 1437)/ K_m for E. coli B11 and K. aerogenes A β -lactamases and various substrates are shown in Table 6. BRL ²³³³ has ^a similar antibacterial spectrum to ampicillin (Sutherland & Rolinson, 1971) but is much better absorbed in man (Croydon & Sutherland, 1971) and BRL 2288 is more active than carbenicillin against strains of Pseudomonas (Sutherland et al., 1971). For each β -lactamase the sameenzyme concentration was used for all substrates, thus making the V_{max} , values comparable. The ratio V_{max}/K_m is a measure of the physiological efficiency of enzymes (Pollock, 1965) and is calculated from the values in Table 6. As the enzyme preparations for the two organisms were made in a similar fashion the physiological efficiencies can be compared.

A comparison of the K_m values for ampicillin and the β -lactamases of the cell-bound and cell-free enzyme from E. coli T506 showed that they were similar, being respectively 40.5 μ M and 54 μ M. For the cell-bound enzyme a washed whole-cell suspension

Table 1. Structure-inhibition relationships of some penicillins for E. coli B11 β -lactamase

Fractional inhibition was measured by using the automated micro iodine assay as described in the text. Test compounds were used at a concentration of 2.7μ M (approx. 1 μ g/ml) to inhibit the hydrolysis of 27 μ M-sodium ampicillin (10 μ g/ml) by a cell-free preparation of E. coli B 11 β -lactamase. The reaction system was incubated at pH7.0 and 37°C for 7min.

Table 2. Inhibition of β -lactamase by derivatives of phenylpenicillin

 K_t values were measured by using the starch-iodometric assay as described in the Materials and Methods section.

was used and for the cell-free enzyme a sample of the latter suspension was sonicated (see the Materials and Methods section) and the debris removed by centrifugation. The velocities of hydrolysis of ampicillin by these two preparations were within a factor of 2. The addition of 7.5 ng of BRL 1437/ml to the two preparations gave a similar degree of inhibition.

Effect of 2-isopropoxy-1-naphthylpenicillin (BRL 1437) on β -lactamase preparations from various bacteria

BRL ¹⁴³⁷ was selected from a large number of compounds as the most effective inhibitor of E. coli B11 and K. aerogenes A β -lactamases. The effect on β -lactamases of other organisms was tested by

measuring the decrease in reaction velocity against benzylpenicillin when the relatively high concentration of 1μ g of BRL 1437/ml was added to the reaction mixtures. The results are shown in Table 7.

Effect of inhibitors on the β -lactamases of viable cultures

All experiments described above have employed β -lactamases released from the cells of E. coli or

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K. aerogenes. To demonstrate the effect of a range of good and poor inhibitors on the β -lactamase of live cultures of recently isolated ampicillin-resistant strains the cultures were incubated with $100 \mu g$ of ampicillin/ml in the presence or absence of 5μ g of test compound/ml. The residual ampicillin was determined by bioassay and the results are shown in Table 8. The experiment was repeated by using the preferred inhibitor BRL ¹⁴³⁷ but with the ampicillin at $5\mu g/ml$ and the inhibitor at 0.1 or 0.5 $\mu g/ml$ to

Reaction rate

Table 5. Reversible inhibition of E. coli B11 β -lactamase by BRL 1437

To 5ml portions of B11 enzyme was added either 1 ml of water or 1 ml of a 2.5 μ g/ml solution of BRL 1437. Portions (3 ml) of these solutions were separately dialysed overnight in 1 litre of water at 4^oC in Visking tubing, and the remaining 3 ml portions were stored at 4°C overnight. The next morning each sample was diluted to 3.5 ml (to allow for any volume change) and the enzymic activity of each fraction assayed by the starch-iodometric assay with 10μ g of ampicillin/ml as substrate. The BRL 1437 diffusate was evaporated to dryness on a rotary evaporator, redissolved in 1 ml of water and tested for its ability to inhibit the hydrolysis of 10μ g of ampicillin/ml by fresh enzyme.

Fig. 3. Lineweaver-Burk plots showing the determination of K_m value for ampicillin with two concentrations of E. coli B11 β -lactamase, and the competitive inhibition of the hydrolysis by this enzyme with BRL ¹⁴³⁷ and methicillin

Reaction mixtures for K_m determination consisted of 6.0-22.0 μ g of ampicillin/ml plus high (Δ) or low (one-third) (A) concentration of enzyme. Reaction mixtures for inhibition experiments consisted of $6.0-20.0\,\mu$ g of ampicillin/ml plus the low concentration of enzyme plus 0.17μ g of methicillin/ml (\blacksquare) or 0.019μ g of BRL 1437/ml (\bullet) (all final concentrations). Reaction velocities were measured by the starchiodometric assay over the initial 5min period of reaction at pH7.0 and 37°C.

simulate possible clinical conditions. The results are presented in Table 9.

Stability of inhibitors and related penicillins to $E.$ coli B 11 β -lactamase

As all of the inhibitors described above are penicillins, the possibility existed that the poorer ones might be substrates for β -lactamase. A range of compounds each at $100 \,\mu$ g/ml was therefore incubated with various concentrations of cell-free E. coli B11 β -lactamase. The remaining compound was bioassayed and the results are presented in Table 10. As with the results in Tables 8 and 9 the possibility exists for continued action of β -lactamase during the early part of the incubation of the bioassay plate.

Discussion

Examination of a large number of semi-synthetic penicillins for their ability to inhibit the hydrolysis of ampicillin by E. coli B11 β -lactamase has revealed several new potent inhibitors and has provided information on structure-activity relationships. The results in Table ¹ illustrate that disubstitution at the a-carbon atom of the substrate molecule benzylpenicillin (or n-butylpenicillin as in BRL 1201) confers weak inhibitory activity. However, if these substituents are too bulky, as in triphenylmethylpenicillin (BRL 1071), the compound does not act as an inhibitor and is a poor substrate, suggesting that this compound cannot enter the catalytic site. Oxacillin and cloxacillin behave similarly. Insertion of a β -CH₂ group into the BRL 1071 molecule (BRL 1307) deflects the phenyl group attached to it and allows the molecule access to the catalytic site and to regain inhibitory properties. 2,6-Dichloro-substitution of a-methoxybenzylpenicillin converts the compound

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into a fairly good inhibitor (BRL 2072) as does incorporation of the α -side-chain carbon atom of benzylpenicillin into a ring system such as cyclohexyl (BRL 1234).

Like benzylpenicillin, phenyl- (BRL 957), naphthyl- (BRL 1371) and quinolyl- (BRL1 559) penicillins are substrates for the E . *coli* B11 enzyme (Table 1), but substitution with an alkoxy group in the 2-position in all three compounds converts them into inhibitors.

Results in Table 2 illustrate further the effect of alkoxy substitution in phenylpenicillins. Insertion of alkoxy groups at both 2- and 6-positions gives better inhibitors, the best substituent group being isopropoxy as in BRL 1593. Insertion of halogen atoms in the 3- and 5-, or 4-positions also had significant effect; 3,5-dichloro substitution also gave a good inhibitor.

Substitution by an alkoxy group in the 2-position of the side-chain of naphthylpenicillin (Table 3) or quinolylpenicillin (Table 4) also confers inhibitory activity, which again increases as this homologous series is ascended. The optimum substituent for maximum inhibitory activity is isopropoxy, as was found in the substituted phenylpenicillin series.

These observations suggest that inhibitory activity conferred on these penicillins is not merely due to bulk of the side chains attached to 6-aminopenicillanic acid but is probably also due to more specific electronic effects. Thus alkoxy substituents in aromatic ring systems will tend to polarize this part of the penicillin molecule, so that there will be a greater electron density in the region of the meta-positions of phenylpenicillins and the 3-position of naphthylpenicillins. These electron-dense areas could be envisaged as constituting a secondary binding with adjacent positively charged areas on the enzyme molecule.

The most active inhibitor against the β -lactamase of E. coli Bi1, 2-isopropoxy-l-naphthylpenicillin (BRL 1437) was also the most active against the β -lactamase of K. *aerogenes* A although inhibitor structure-activity profiles were not identical for the two organisms, the K. aerogenes enzyme being readily inhibited by a wide range of structures.

Experiments with BRL ¹⁴³⁷ revealed that it was ^a reversible (Table 5) and competitive (Fig. 3) inhibitor. The graphs used to obtain K_i values for other inhibitors suggest that they also act competitively. The stability results in Table ¹⁰ show that BRL ³²¹⁵ and BRL 1437, respectively the o -isopropoxy-substituted quinolyl- and naphthyl-penicillins, are highlyresistant to the β -lactamase of E. coli B11. As cloxacillin and triphenylmethylpenicillin have some stability but are not inhibitors and methicillin is a poor inhibitor but has fair stability, there does not seem to be any correlation between stability and effectiveness as an inhibitor, although clearly an effective inhibitor must be stable.

INHIBITION OF β -LACTAMASES

Table 7. Effect of BRL 1437 on the β -lactamases of various bacteria

Suitable dilutions of the enzyme preparations were prepared so that they hydrolysed 10μ g of benzylpenicillin/ml at an initial rate of about 0.1-0.3 μ g/min per ml. The cell-free extracts of cells were prepared by sonicating cells as described in the Materials and Methods section and centrifuging off cell debris. The enzyme preparations were incubated with 10μ g of benzylpenicillin/ml at pH7 (0.05M-phosphate buffer) and 37°C in the presence and absence of 1μ g of BRL 1437/ml. The reaction velocities were determined by the starch-iodometric method.

Table 8. Inhibition of ampicillin destruction with live cultures of Escherichia and Klebsiella

Bacterial cultures were grown for 18h in test-tubes of sterile Oxoid no. ² nutrient broth at 37°C. A sterile solution of ampicillin was added to each culture to give a final concentration of $100 \mu g/ml$. Sterile solutions of β -lactamase inhibitors were added where indicated to give final concentrations of $5 \mu g/m$. The tubes were incubated for a further 4h at 37°C before bioassay of the residual ampicillin. A certain amount of further destruction of ampicillin by β -lactamase would have occurred during the bioassay. Test inhibitors at 5 μ g/ml concentrations have no antibacterial activity against the bioassay organism.

Compound BRL ¹⁴³⁷ also has inhibitory action against the β -lactamases of Pseudomonas and Proteus, as shown in Table 7. It is without effect on staphylococcal β -lactamase and has weak effect on that of B. cereus. Like Nafcillin (2-ethoxy-1-naphthylpenicillin), which is used clinically for penicillinresistant staphylococci, BRL ¹⁴³⁷ is stable to staphylococcal β -lactamase and is active against staphylococci.

The fact that compounds such as BRL ¹⁴³⁷ can have good inhibitory activity against the β -lactamases of live cultures of various ampicillin-resistant strains of E. coli and K. aerogenes is revealed in Table 8, where the inhibition was good at the low concentration of 5μ g of inhibitor/ml with ampicillin concentrations of $100\,\mu$ g/ml. BRL 1437 would appear to be

Table 9. Inhibition of β -lactamase by live fully grown bacterial cultures at low concentrations of ampicillin and BRL ¹⁴³⁷

The method was as described in Table 7 except that the final concentration of ampicillin was $5 \mu g/ml$. The β -lactamase inhibitor BRL 1437 was added to give final concentrations of 0.5 or 0.1 μ g/ml and the residual ampicillin determined by bioassay after 4h at 37°C.

superior to BRL ³²¹⁵ and methicillin (BRL 1241) whereas cloxacillin (BRL 1621) was ineffective, in agreement with the result in Table 1. Inhibition was also obtained at 0.1 and 0.5μ g of BRL 1437/ml with ampicillin at $5 \mu g$ /ml (Table 9). Inhibition was most marked for Klebsiella, but E. coli B11, which produced the highest enzyme activity, was not inhibited in this stringent test. These concentrations are in the clinically obtainable range and suggest that this combination could have useful synergistic effects in vivo where ampicillin destruction was responsible for bacterial resistance. None of the inhibitors in Table 8 was effective against the β -lactamase of K. oxytoca, suggesting that this organism produces a different enzyme.

The Klebsiella A enzyme has both K_m and V_{max} . values lower than that of E . coli B11 against all substrates, and the V_{max}/K_m ratios are therefore similar for the two enzymes (Table 6) indicating similar physiological efficiencies (Pollock, 1965). The relatively low V_{max}/K_m ratio for BRL 2288 and carbenicillin indicates that these compounds have some general stability to the β -lactamases of E. coli and K. aerogenes, but these compounds do not differ from the others by more than a factor of five. The lower V_{max} of carbenicillin and BRL 2288 is somewhat compensated by the higher affinity.

From the K_l/K_m ratios for BRL 1437 it is clear that for all compounds in Table 6 the relative affinity of BRL ¹⁴³⁷ for both enzymes is over ¹⁰⁰⁰ times the affinity of the benzylpenicillin, carbenicillin and BRL ²²⁸⁸ for the enzyme. With ampicillin and BRL ²³³³ as substrate the relative affinity for BRL ¹⁴³⁷ was 10000 times greater. Thus with ampicillin and BRL ²³³³ as substrates BRL ¹⁴³⁷ would be expected to be very effective at protecting these compounds from destruction by E. coli and K. aerogenes β lactamases and therefore to act synergistically. With

Table 10. Stability of inhibitors and related penicillins to β -lactamase of E. coli B11

Test compounds at a final concentration of 100 μ g/ml (purity corrected) were incubated at pH7 at 37°C for 4h in buffer or in the presence of various dilutions of the stock enzyme preparation of E. coli B11. The remaining penicillin was bioassayed against the relevant standard by using B . *subtilis* and expressed as $\%$ of compound remaining, after correction for the buffer control.

a K_i/K_m ratio of 2×10^{-4} for dicloxacillin and ampicillin, Bach et al. (1967) obtained good antibacterial synergism for Shigella flexneri. Of course, antibacterial synergism cannot be expected where thebacterium is insensitive to ampicillin (i.e. intrinsically resistant), even though its β -lactamase is effectively inhibited. This appears to be the case with E . coli B11 (R. Sutherland, personal communication), which was chosen for our preliminary screen mainly because it produced a high β -lactamase activity.

O'Callaghan et al. (1969) have examined many, cephalosporins for stability to the β -lactamase of *Ent*. cloacae and found that the most stable were inhibitors of the action of this enzyme on cephaloridine. We initially looked for β -lactamase-stable (*E. coli*) penicillins as a route to inhibitors but abandoned it because of the questionable relationship between stability and ability to inhibit the enzymes. In the present paper this is illustrated by triphenylmethylpenicillin (BRL 1071), which is stable but not an inhibitor.

Hamilton-Miller et al. (1965) have reported that cloxacillin is a better inhibitor than methicillin for hydrolysis of benzylpenicillin by E . coli β -lactamase (ampicillin was not hydrolysed). This is contrary to the results obtained in the present paper for cultures which hydrolyse ampicillin, and it may be that their E. coli strain 419 was atypical and more like our strain 83 (Table 8). Hamilton-Miller et al. (1965) reported that methicillin was a better inhibitor than cloxacillin for two strains of Klebsiella, which is in agreement with our findings.

There is now no shortage of good inhibitors of the β -lactamases of E. coli and K. aerogenes, and it is to be hoped that the frequency and degree of synergism between penieillins and the new inhibitors for a collection of isolates might be higher than obtained with previously available inhibitors (Sutherland & Batchelor, 1964).

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