

A Calcium Ion-Dependent Adenosine Triphosphate Pyrophosphohydrolase in Plasma Membrane from Rat Liver

DEMONSTRATION THAT THE ADENOSINE TRIPHOSPHATE ANALOGUES
ADENOSINE 5'-[β -IMIDO]TRIPHOSPHATE AND ADENOSINE 5'-[β -METHYLENE]-
TRIPHOSPHATE ARE SUBSTRATES FOR THE ENZYME

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An ATP pyrophosphohydrolase in a rat liver plasma-membrane subfraction was studied with respect to specific Ca^{2+} activation of the β -phosphate bond hydrolysis. ATP and, in addition, adenosine 5'-[β -imido]triphosphate and adenosine 5'-[β -methylene]triphosphate were substrates for Ca^{2+} -stimulated enzymic hydrolysis of the β -phosphate bond. A 15-fold activation was observed by raising the free Ca^{2+} concentration from 10^{-7} to 10^{-5} M. Mg^{2+} had little effect. Solubilization in 1% deoxycholate and partial purification on a sucrose density gradient resulted in a 5-fold increase in specific activity with unaltered Ca^{2+} -stimulation pattern. The possible importance of the enzyme in Ca^{2+} transport is discussed.

In 1967 Lieberman *et al.* reported the presence of an enzyme activity named nucleoside triphosphate pyrophosphohydrolase in rat liver plasma membranes. Very active ATP pyrophosphohydrolase activities associated with liver plasma membrane have later been observed by several authors (Ray *et al.*, 1970; Franklin & Trams, 1971; House *et al.*, 1972).

No function has yet been established for this enzyme in the membrane. Franklin & Trams (1971) found that the enzyme is not associated with acid-CoA ligase (EC 6.2.1.3) in plasma membranes.

It has previously been shown that ΔG for the hydrolysis of PP_i is near zero in the cell (Flodgaard & Fleron, 1974; Flodgaard, 1976). Consequently it can be deduced that ΔG for hydrolysis of ATP to AMP and PP_i in the cell is approx. 84 kJ/mol (20 kcal/mol) (Lawson *et al.*, 1976), and twice as large as ΔG for the hydrolysis of ATP to ADP and P_i . It is therefore tempting to associate the membrane-bound ATP pyrophosphohydrolase with reactions requiring more free-energy input than can be supplied by the hydrolysis of ATP to ADP and P_i . Extrusion of Ca^{2+} across the plasma membrane of the cell could be a candidate for such a reaction. In fact, on the assumption of an intracellular Ca^{2+} concentration of 10^{-7} to

10^{-8} M (Baker, 1976), an extracellular Ca^{2+} concentration of 10^{-3} M and a membrane potential of 70 mV, the extrusion of 1 mol of Ca^{2+} against the steady-state concentration and electrochemical gradients requires 38–42 kJ (9–10 kcal) [$\Delta G = RT \ln(c_{in}/c_{out}) + |z|F\phi$]. This is equal to or very close to the free energy of hydrolysis of ATP to ADP and P_i . Thus, with a stoichiometry of 1 Ca^{2+} ion transported to 1 ATP molecule hydrolysed to ADP and P_i , an efficiency of the pump near 100% is required.

The object of the present investigation was to study the ATP pyrophosphohydrolase with respect to an activation by Ca^{2+} . To exclude interference by ATPases and adenylate kinase the enzyme was also studied by using ATP analogues not attacked by these enzymes (Yount *et al.*, 1971).

Experimental

Chemicals

[8- ^3H]Adenosine 5'-[β -imido]triphosphate (ammonium salt; specific radioactivity 25 Ci/mmol), [8- ^3H]adenosine 5'-[β -methylene]triphosphate (ammonium salt; specific radioactivity 23 Ci/mmol), 5'-[γ - ^{32}P]ATP (sodium salt; specific radioactivity 3 Ci/mmol), [^3H]AMP and [^3H]UTP were from The Radiochemical Centre, Amersham, Bucks., U.K. The purity of the ^3H -labelled ATP analogues was tested with respect to contamination with ATP by t.l.c. on poly(ethyleneimine)-cellulose in 2M-sodium for-

Abbreviations used: p[N]ppA, adenosine [β -imido]-triphosphate; p[CH₂]ppA, adenosine [β -methylene]-triphosphate; Tes, 2-[[2-hydroxy-1,1-bis(hydroxymethyl)-ethyl]amino]ethanesulphonic acid; ATPase, adenosine triphosphatase.

mate buffer, pH 3.4. ATP contamination was less than 1%. Unlabelled ATP (disodium salt) and Tes were from Sigma Chemical Co., St. Louis, MO, U.S.A. Unlabelled p[NH]ppA, p[CH₂]ppA (both 98% pure, Boehringer certificate of analysis), AMP, UTP (sodium salts), adenosine, inorganic pyrophosphatase (EC 3.6.1.1), phosphoglucomutase (EC 2.7.5.1), glucose 6-phosphate dehydrogenase (EC 1.1.1.49), and UDP-glucose pyrophosphohydrolase (EC 2.7.7.9) were from Boehringer, Mannheim, Germany. Lumagel was from Lumac LSC Chemicals, Basel, Switzerland. Poly(ethyleneimine)-cellulose thin-layer sheets were product no. 5579 from Merck, Darmstadt, Germany. All other chemicals were of analytical grade.

Preparation of plasma membranes

House *et al.* (1972) described a procedure for the isolation of three subfractions of plasma membranes from rat liver. Only the low-density subfraction having a sucrose buoyant density of 1.13 g/ml and being free of desmosomes, junctional complexes and bile canaliculi was used in the present investigation. Mitochondrial contamination as measured by monoamine oxidase activity with the method of Wurtman & Axelrod (1963) was found to be negligible, and the glucose 6-phosphatase activity (EC 3.1.3.9) measured by the method of Baginski *et al.* (1970) showed less than 10% contamination with microsomal particles.

Measurement of ATP pyrophosphohydrolase activity with ATP analogues as substrates

Incubation was carried out at 37°C in 20 µl of the various buffers described in Table 1 containing ³H-labelled analogue (2.5 mM), diluted to a specific radioactivity of 2 mCi/mmol after the removal of ethanol by freeze-drying. Then 1–10 µg of protein was added and incubation carried out for 60 min. After the end of the incubation 50 nmol each of AMP and adenosine was added in a volume of 10 µl. A 10 µl portion was immediately spotted on poly(ethyleneimine)-cellulose thin-layer sheets, and the chromatogram was developed in 0.5 M-LiCl for 10 cm. The AMP+adenosine spot was cut out and extracted for 15 min in 1 ml of 4 M-NH₃. The ³H radioactivity was counted in 10 ml of Lumagel in a Packard model 2425 Tri-Carb liquid-scintillation counter. The measured radioactivity was proportional to the amount of enzyme added in the assay and with incubation time for at least 90 min. It was found that 100% of substrate could be hydrolysed to AMP+adenosine with excess of enzyme.

Measurements of ATP pyrophosphohydrolase activity in the solubilized subfraction freed from inorganic pyrophosphatase, and with ATP as substrate

The initial incubation procedure was the same as described above, except for the addition of [γ -³²P]ATP

Table 1. Composition of Ca²⁺- and Mg²⁺-containing buffers

The buffers were prepared in 20 mM-Tes, pH 8.0, containing 5 mM-EDTA, 5 mM-EGTA and 2.5 mM of either ATP, p[N]ppA or p[CH₂]ppA. The total amounts of CaCl₂ and MgCl₂, to be added, to give the required free concentrations were calculated from the appropriate acid dissociation constants and complex-formation constants for all the components present in the buffers. The constants for ATP, EDTA, and EGTA were taken from Sillén (1971), and those for the two ATP analogues from Yount *et al.* (1971). The calculations were done on a Univac 1100 digital computer by using an ALGOL program described earlier (Flodgaard & Fleron, 1974). Ca²⁺ and Mg²⁺ binding to the various dissociated forms of the analogues is stronger than binding to ATP. The pK values for the terminal phosphate group in the analogues, however, is higher than for ATP, and thus replacement of ATP constants with analogue constants in the computer program did not alter the calculated total concentrations significantly.

Free concentrations (M)		Total concentrations (mM)	
Ca ²⁺	Mg ²⁺	CaCl ₂	MgCl ₂
10 ⁻³	0	15.3	0
10 ⁻⁴	0	11.9	0
10 ⁻⁵	0	10.4	0
0	10 ⁻⁵	0	5.38
10 ⁻⁵	10 ⁻³	6.83	6.62
10 ⁻⁶	0	9.7	0
10 ⁻⁷	0	7.8	0
10 ⁻⁸	0	3.9	0

instead of the ³H-labelled analogue. The reaction was stopped by boiling for 5 min on a water bath. After cooling, 10 nmol of carrier PP_i was added in a volume of 10 µl. The [³²P]PP_i formed was incorporated into [³H,³²P]UTP as described by Flodgaard & Fleron (1974). The nucleotides were separated by the method of Randerath & Randerath (1964), by using poly(ethyleneimine)-cellulose thin-layer sheets. The ATP and UTP spots were cut out and counted for radioactivity as described above; the latter was counted for both ³H and ³²P radioactivity. From the ratio between the ³H and ³²P radioactivities in the UTP spot the amount of PP_i produced from ATP was calculated. Enzyme activity was proportional to the amount of enzyme added, and with time for at least 1 h.

Inorganic pyrophosphatase activity was measured under the same conditions as described above for the measurements of ATP pyrophosphohydrolase except for replacement of [γ -³²P]ATP with [³²P]PP_i at concentrations of 200, 100 and 10 µM. After incubation for 1 h with solubilized membranes residual [³²P]PP_i was measured as described.

Adenosine triphosphatase was measured as described by Formby *et al.* (1976).

Protein was estimated by the method of Lowry *et al.* (1951) with the modifications of Lous *et al.* (1956).

Solubilization of the plasma membranes and partial purification of the ATP pyrophosphohydrolase

The plasma membranes were solubilized in 1% deoxycholate in 25 mM-Tes, pH 7.5, for 1 h (0.75 mg of protein/ml of solution) at 0°C. After dialysis overnight against excess of 25 mM-Tes, pH 7.5, the solubilized membranes were placed on linear 3–30% sucrose gradients in a Beckman SW 27 rotor (0.75 mg of protein per gradient) and centrifuged at 100000g for 24 h at 4°C. The gradients were emptied through the top with a Beckman gradient fraction-recovery system, and 1 ml fractions were collected. The peak fractions containing most of the enzyme activity (15–21% sucrose) were pooled, dialysed and ultrafiltered against excess of 25 mM-Tes, pH 7.5, to a final volume of about 125 μ l/0.75 mg of membrane protein.

Results

A specific activity in the range 1.6–2.3 μ mol of AMP+adenosine/h per mg of protein was obtained with four individual preparations of plasma membrane subfraction with p[CH₂]ppA as substrate and a Ca²⁺ concentration of 10⁻⁵ M, with no Mg²⁺, and at pH 8.0.

The membrane nature of the enzyme was checked by sonificating 330 μ g of membrane in 500 μ l of 50 mM-Tes, pH 8.5, containing 1.0 M-NaCl for 10 s at 50 W. Some heating was noticed, and enzymic activity fell to 63% of initial value ([Ca²⁺] = 10⁻⁵ M, no Mg²⁺, p[CH₂]ppA as substrate). After 1 h at room temperature (20°C) with occasional shaking, the preparation was centrifuged at 40000g for 15 min at 4°C. All residual activity (47% of initial) was recovered in the pellet, although 20% of the initial protein was extracted in the supernatant. This result is in agreement with the findings of Lieberman *et al.* (1967).

A 15-fold stimulation of the specific activity of the solubilized enzyme with ATP as substrate is seen by raising the Ca²⁺ concentration from 10⁻⁷ to 10⁻⁵ M in the absence of Mg²⁺ (Fig. 1; for composition of buffers, see Table 1). A very small but reproducible (three experiments) stimulation is seen when the Ca²⁺ concentration is raised from 10⁻⁵ to 10⁻⁶ M. Another 2-fold increase in activity is observed by raising the Ca²⁺ concentration to 10⁻³ M (Fig. 1). The curves with p[N]ppA and p[CH₂]ppA as substrate show similar stimulation, but Ca²⁺ above 10⁻⁵ M inhibits. With p[CH₂]ppA as substrate, replacement of the maximum stimulatory Ca²⁺ concentration at 10⁻⁵ M with Mg²⁺ at 10⁻⁵ M and with no Ca²⁺ (all other conditions identical with those described in Fig. 1) results in near-zero activity (0.73 μ mol/h per mg of protein). At 10⁻⁵ M-Ca²⁺ and a physiological Mg²⁺ concentration (10⁻³ M), with

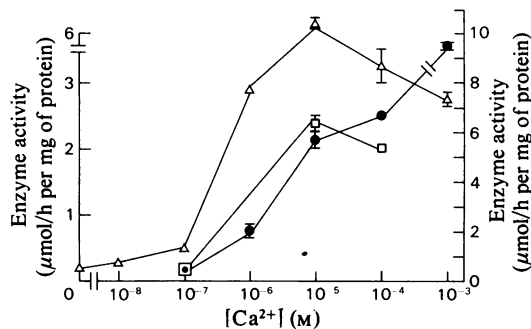


Fig. 1. Specific activity of solubilized ATP pyrophosphohydrolase versus [Ca²⁺]

The assays were done at pH 8.0, 37°C for 60 min in 20 μ l of 20 mM-Tes buffer containing 2.5 mM of either [³²P]ATP (●) (left ordinate), p[N]pp[³H]A (□) (left ordinate) or p[CH₂]pp[³H]A (Δ) (right ordinate), EDTA, EGTA (both at 5 mM) and 1.2 μ g of solubilized enzyme. The total amounts of CaCl₂ added to give the Ca²⁺ concentration were calculated on a digital computer by using appropriate acid dissociation and complex-formation constants (see Table 1). [³²P]PP_i or [³H]AMP+[³H]adenosine were isolated at the end of incubation, and the specific radioactivities were calculated. The experiments were carried out in duplicate, and have been repeated at least three times with different enzyme concentrations. The Figure shows means of duplicate determinations in one experiment. Errors calculated from the duplicated determinations are indicated with bars when greater than the marks used.

p[CH₂]ppA as substrate a 25% decrease of the specific activity was observed compared with the same conditions without Mg²⁺. Replacement of Ca²⁺ with Sr²⁺ or Ba²⁺ at free concentrations of 1.93 × 10⁻³ M and with no Mg²⁺ decreases the specific activity of the solubilized enzyme with p[CH₂]ppA as substrate at pH 8.0 to 14.0 and 15.4% of the activity obtained with 10⁻³ M-Ca²⁺ respectively. From the maximum activity of between 8.2 and 10.4 μ mol/h per mg of protein from three individual preparations of solubilized and purified enzyme with 10⁻⁵ M-Ca²⁺ with p[CH₂]ppA as substrate a 4.5–5-fold increase in specific activity by solubilization and partial purification can be deduced.

The solubilization rests on the criterion that no enzyme activity could be pelleted by centrifugation at 100000g for 1 h at the end of the purification procedure (see the Experimental section). The preparation was completely free of inorganic pyrophosphatase (see the Experimental section), but some ATPase activity was still detectable in the presence of Mg²⁺. However, in the absence of Mg²⁺ the only reactions taking place were the splitting of ATP to AMP+adenosine and PP_i, as the recovery of ³²P in the sum of [³²P]ATP and [³²P]PP_i was 100% (99.3 ± 1.3%,

mean \pm S.E.M., $n=8$). The recovery of ^3H radioactivity in the sum of ^3H -labelled analogue and ^3H -labelled AMP+adenosine isolated from the assays was near 100% (for p[CH₂]ppA 97.4 \pm 1.1%, $n=15$; for p[N]ppA 93.7 \pm 2.4%, $n=10$), indicating selective hydrolysis at the β -phosphate position. p[CH₂]p and p[N]p were not isolated. Total recovery of enzyme activity after solubilization, gradient centrifugation, dialysis and ultrafiltration varied between 40 and 50% in three experiments, the greatest loss observed during the ultrafiltration and dialysis. The enzyme seems to be purified partly together with 5'-nucleotidase since a threefold increase in this activity in the final preparation was observed (10 $\mu\text{mol/h}$ per mg of protein, measured at 30°C, pH 7.5 and with 0.1 mM-[^3H]AMP as substrate).

The high specific activity of the ATP pyrophosphohydrolase compared with the very low specific activity observed for adenylate cyclase in these membranes (House *et al.*, 1972) makes it unlikely that this enzyme should interfere. Measurement of ATP pyrophosphohydrolase in the presence of 1 mM-3-isobutyl-1-methylxanthine (a phosphodiesterase inhibitor) showed negligible ^3H radioactivity in the cyclic AMP isolated.

Discussion

The present investigation has shown that an ATP pyrophosphohydrolase in rat liver plasma membranes was activated by Ca²⁺ in the physiological range of free concentrations from 10⁻⁸ to 10⁻⁴ M.

The results with the analogues of ATP are surprising. Although it has been shown that p[N]ppA can act as substrate in some reactions using the β -phosphate bond, breakdown of the much more unphysiological p[CH₂]ppA analogue at the β -position has never been reported before. Therefore the widely accepted inertness of this compound can no longer be sustained. In the present work it was shown to be even a better substrate than ATP (Fig. 1). The Ca²⁺-stimulation pattern with this analogue supports the contention in the experiments with authentic ATP that it is the breakdown of the β -phosphate bond that is stimulated by Ca²⁺.

Lieberman *et al.* (1967) measured the ATP pyrophosphohydrolase in plasma membranes at pH 12. They did not investigate Ca²⁺-dependency, but observed a Mg²⁺-dependence. They found a pH optimum at 10, and, since we found a higher specific activity at pH 9 compared with pH 7.5 (10.0 versus 6.22 $\mu\text{mol/h}$ per mg of protein in one experiment with ATP as substrate), it is probably the same enzyme we have studied. The experimental conditions were,

however, very different. Furthermore, the nearly zero activity at pH 7.5 observed by these authors may be caused by an Mg²⁺-dependent inorganic pyrophosphatase. Our substantial activity observed at physiological pH with ATP as substrate was measured with a pyrophosphatase-free preparation.

The function of the enzyme is still unknown, and it is too early to conclude anything from the Ca²⁺ activation. The ability of the enzyme to break down ATP analogues may be of great help in evaluating the function, since ubiquitous inorganic pyrophosphatases and adenylate kinase will obscure the distinction between an ATPase and an ATP pyrophosphohydrolase with authentic ATP as substrate, but not with the analogues as substrates.

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References

- Baginski, E. S., Foa, P. P. & Zak, B. (1970) in *Methoden der Enzymatischen Analyse* (Bergmeyer, H. U., ed.), Vol. 1, pp. 839–843, Verlag Chemie, Weinheim/Bergstr.
- Baker, P. F. (1976) in *Calcium in Biological Systems* (Duncan, C. J., ed.), pp. 67–89, Cambridge University Press, London, New York and Melbourne
- Flodgaard, H. (1976) Ph.D. Thesis, University of Copenhagen
- Flodgaard, H. & Fleron, P. (1974) *J. Biol. Chem.* **249**, 3465–3474
- Formby, B., Capito, K., Egeberg, J. & Hedeskov, C. J. (1976) *Am. J. Physiol.* **230**, 441–448
- Franklin, J. E. & Trams, E. G. (1971) *Biochim. Biophys. Acta* **230**, 105–116
- House, P. D. R., Poulis, P. & Weidemann, M. J. (1972) *Eur. J. Biochem.* **24**, 429–437
- Lawson, J. W. R., Guynn, R. W., Cornell, N. & Veech, R. L. (1976) in *Gluconeogenesis: Its Regulation in Mammalian Species* (Hanson, R. W. & Mehlman, M. A., eds.), pp. 481–512, John Wiley and Sons, New York and London
- Lieberman, I., Lansing, A. I. & Lynch, W. E. (1967) *J. Biol. Chem.* **242**, 736–739
- Lous, P., Plum, C. M. & Schou, M. (1956) *Nord. Med.* **55**, 693–695
- Lowry, O. H., Rosebrough, N. J., Farr, A. L. & Randall, R. J. (1951) *J. Biol. Chem.* **193**, 265–275
- Randerath, E. & Randerath, K. (1964) *J. Chromatogr.* **16**, 126–129
- Ray, T. K., Tomasi, V. & Marinetti, G. V. (1970) *Biochim. Biophys. Acta* **211**, 20–30
- Sillén, L. G. (1971) *Chem. Soc. Spec. Publ.* **25**
- Wurtman, R. J. & Axelrod, J. (1963) *Biochem. Pharmacol.* **12**, 1439–1440
- Yount, R. G., Babcock, D., Ballantyne, W. & Ojala, D. (1971) *Biochemistry* **10**, 2484–2489