2. The increase in protein in the hypertrophy caused by hormones was confirmed for the levator ani muscle of the rat.

3. The percentage composition of the muscle was unchanged during the periods of hypertrophy and return to normal weight. With further loss of weight, the percentage of contractile proteins decreased and a corresponding increase in percentage of connective tissue occurred.

4. At 7 days after denervation a 36  $\%$  increase in absolute amount of sarcoplasmic protein accompanied by <sup>a</sup> <sup>27</sup> % increase in absolute amount of contractile protein was observed. At 45 days decreases in absolute amount of these two fractions of <sup>45</sup> and <sup>54</sup> % respectively had occurred.

5. There was no significant change in absolute amount of connective tissue during the period studied.

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# A Constituent of the Unsaponifiable Portion of Animal Tissue Lipids  $(\lambda_{\text{max}} 272 \text{ m}\mu.)$

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The distribution of 7-dehydrosteroid (i.e. material showing ultraviolet absorption similar to that of provitamin D) in animal tissues has been studied by Glover, Glover & Morton (1952) and by Festenstein & Morton (1955). The small intestine is a relatively rich source in some species (e.g. guinea pig) but not in others.

In the course of their work Festenstein & Morton noted the presence in the intestines of horse and pig (but not sheep or ox) of a constituent of the unsaponifiable fraction which showed an absorption maximum at  $272 \text{ m}\mu$ .

The selectively absorbing substance was concentrated by chromatography and characterized in terms of qualitative and quantitative ultraviolet absorption spectra in various solvents and by its adsorptive properties. Although horse intestine appeared to contain possibly 20 mg./kg. the yield of the richest concentrates was not very promising. A similar substance was found by Cain & Morton (1955) in horse liver but the investigation has been

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hampered by the presence of other very persistent congeners.

Some time later the '272  $m\mu$ . substance' was again encountered. Unsaponifiable fractions from various tissues of both normal and vitamin Adeficient rats were chromatographed on alumina and a 'rat  $272 \text{ m}\mu$ . material' was obtained and identified with the 'horse and pig  $272 \text{ m}\mu$ . substance'. More extensive study of various tissues from xerophthalmic rats and normal rats showed the substance to occur in kidney, liver, intestine and submaxillary gland, but to be absent from similarly separated fractions of bladder, testis and vagina. It is hoped in due course to comment upon the relative amounts of '272  $m\mu$ . material' in the tissues of control rats and vitamin A-deficient rats. It is clear, however, that the livers and kidneys of vitamin A-deficient rats are a fairly good source, and the same substance has been observed in hen liver and kidney. It also occurs (Kantiengar, Lowe, Morton & Pitt, 1955) in cockerel liver and kidney and in guinea pig liver (Leat, private communication). The present work records properties aiming at

characterizing the  $272 \,\mathrm{m}\mu$ . substance. It is proposed in what follows to use the abbreviation SA for the compound responsible for the absorption curve of Fig. 4 ( $\lambda_{\text{max}}$ , 272 m $\mu$ ., inflexion 330 m $\mu$ ., flat peak near  $410 \text{ m}\mu$ .).

#### EXPERIMENTAL

#### Materials

Fresh horse, sheep, ox and pig intestines were obtained from the Stanley Abattoir, Liverpool, and were treated in the manner described by Festenstein & Morton (1955).

Normal rat tissues were obtained from adult animals of the stock colony which had been reared on Animal diet no. 4 (British Extracting Co. Ltd., Bromborough Port, Cheshire) together with bread and milk. Vitamin A-deficient rats were produced by feeding weanlings the following diet: casein <sup>18</sup> % (either (a) 'Low vitamin content' from Genatosan Ltd., or (b) 'Fat-free and vitamin-free' from British Drug Houses Ltd.), rice starch 65%, arachis oil 5%, Marmite 8%, salt mixture (Osborne & Mendel, 1913)  $4\%$ . When the rats exhibited loss of weight and xerophthalmia they were anaesthetized and killed by cardiac puncture, using a syringe and collecting blood.

#### Methods

A few portions of intestinal tissue were ground with sharp sand and anhydrous  $\text{Na}_2\text{SO}_4$  and the lipid was extracted with peroxide-free ether. The dry lipid was chromatographed on alumina weakened by treatment with water and the fractions were examined spectroscopically, using cyclohexane as solvent. More usually, however, portions of tissue were subjected to alkali digestion and the unsaponifiable matter was extracted with light petroleum. The unsaponifiable fractions were chromatographed on alumina weakened with water (see below).

Rat tissues were treated as follows: to the moist tissue was added  $60\%$  (w/v) KOH (0.5 ml./g.) tissue and the mixture was heated on a water bath until it became homogeneous. Ethanol (2 vol.) was then added and boiling continued under reflux for 30-40 min., whereupon water was added and the mixture was extracted (five or six times) with ether (freshly distilled over reduced Fe). The combined ether extract was washed with water until free from alkali, dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, filtered through a sintered glass funnel and taken to dryness under  $N<sub>2</sub>$ .

Chromatographic separations were carried out on alumina (Grade 0. P. Spence and Co. Ltd., Widnes) weakened with  $2\%$  (v/w) water. Ether, light petroleum  $(A.R. b.p. 40-60^{\circ})$  and mixtures of these two solvents, were used as eluents. Both solvents were dried over Na wire and redistilled before use, the ether being redistilled over reduced Fe. Fractions were taken to dryness under nitrogen and examined spectroscopically.

#### RESULTS

Pig intestine. The unsaponifiable material from pig intestine was fractionally crystallized and a solid fraction showing a weak maximum at  $272 \text{ m}\mu$ . was chromatographed on  $5\%$  (v/w) watered alumina.  $5\%$  ether-light petroleum eluted a fraction (13 mg.) with  $E_{1 \text{ cm}}^{1 \text{ % } 272 \text{ m}\mu\text{.} 5 \cdot 0\text{.}$  This was rechromatographed and 2% ether-light petroleum eluted 4 mg. of material with  $E_{1 \text{ cm}}^{1 \text{ %}}$  272 m $\mu$ . 7.6 and 4% ether-light petroleum eluted 4 mg. of similar material with  $E_{1cm}^{1\%}$  $272 \text{ m}\mu$ . 4.2. The absorbing entity showed no selective absorption in the region 300-500 m $\mu$ . and was seen on the column as a faint yellow band.

Horse intestine. A  $12$  ft. long portion of fresh intestine was carefully trimmed from fatty tissue adhering to the outside, most of the contents (hereafter referred to as mucus) were expelled by pressure along the length and the remainder was collected after longitudinal sectioning.

The fatty tissue, the mucus and the intestine proper were saponified separately and the resulting unsaponifiable fractions were crystallized from methanol to remove most of the cholesterol. The residual unsaponifiable matter was then chromatographed on <sup>5</sup> % watered alumina in each case.

(a) Mucus. The unsaponifiable material from the first trial showed  $\lambda_{\text{max.}}$  270 m $\mu$ .  $E_{1 \text{ cm.}}^{1 \text{ % }}$  5.4 and that from the second trial (0.11%) showed a similar broad band  $E_{1 \text{ cm}}^{1 \text{ %}}$  4.7.

On chromatography an orange zone passed down the column fairly quickly with light petroleum as the developing solvent. In one case a fraction (18 mg.) showing  $E_{1 \text{ cm}}^{1 \text{ %}}$  22.5 at  $\lambda_{\text{max}}$  273 m $\mu$ . was obtained with 4% ether-light petroleum; in another experiment 2% ether-light petroleum eluted a substance with  $\lambda_{\text{max}}$  268 m $\mu$ . and 4% ether-light petroleum a substance with the whole curve displaced  $(\lambda_{\text{max}})$ 273 m $\mu$ .). The most strongly adsorbed materials included cholesterol, 7-dehydrocholesterol and hydroxylated carotenoids. The least strongly adsorbed material showed selective absorption near 257 m $\mu$ .

(b) Intestine. The unsaponifiable material  $(0.24\%$  of weight of tissue) showed  $\lambda_{\text{max}}$  270 m $\mu$ .  $E_{1 \text{ cm}}^{1 \text{ %}}$  4.3.

On chromatography, light petroleum eluted a small amount of carotene together with material transparent to ultraviolet light; <sup>2</sup> % ether-petroleum removed the remaining carotene and later gave a substance with  $\lambda_{\text{max.}}$  270 m $\mu$ .  $E_{1\text{ cm}}^{1\%}$  29 and  $\lambda_{\text{ max}}$ , 405 m $\mu$ .  $E_{1\text{ cm}}^{1\%}$ , 1.25 (probably SA).

In a second experiment 2% ether-light petroleum gave 0.14 g. of eluted material  $\lambda_{\text{max}}$  271 m $\mu$ .  $E_{1 \text{ cm}}^{1 \text{ %}}$  20, and  $4\%$ ether-light petroleum 0.092 g. of material with  $\lambda_{\text{max}}$ , 273 m $\mu$ .  $E_{1 \text{ cm}}^{1 \text{ %}}$  6.

(c) Mesenteric fat. The unsaponifiable fraction after chromatography showed clearly the presence of carotenoids but not of other substances showing characteristic ultraviolet bands.

(d) Variation of '270  $m\mu$ . substance' along the length of the intestine. Table <sup>1</sup> shows that the absorbing material is fairly evenly distributed and is not localized in the absorptive part of the gut.

Table 1. Variations in unsaponifiable matter along the length of a horse intestine

Values in parentheses refer to mucus, others to intestine.



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|                                                                                                                          |     | Mucus  | Intestine |
|--------------------------------------------------------------------------------------------------------------------------|-----|--------|-----------|
| Wet weight (g.)                                                                                                          |     | 190    | 500       |
| Lipid (%)                                                                                                                |     | 1.8    | 0.98      |
| Unsaponifiable matter, direct saponification $\langle \frac{0}{0} \rangle$                                               |     | 0.15   | 0.23      |
| $E_{1\ \text{cm}}^{1\ \text{%}}$ 270-272 m $\mu$ .:                                                                      |     |        |           |
| Measured on selectively absorbing fraction<br>$\bf (i)$<br>after chromatography of unsap. but<br>expressed on wet weight |     | 0.0038 | 0.0045    |
| (ii) Measured on two absorbing fractions after<br>chromatography of lipid but expressed on<br>wet weight                 | (a) | 0.0023 | 0.0067    |
|                                                                                                                          | (b) | 0.0012 | 0.0030    |
|                                                                                                                          | Sum | 0.0035 | 0.0097    |

Table 2. Horse intestine lipids and unsaponifiable matter



Fig. 1. 272 m $\mu$ . fractions from chromatography of horse intestine lipid or unsaponifiable fraction. -, a lipid fraction  $(0.097\%)$ ;  $\cdots$ , a fraction from unsaponifiable material  $(0.061\%)$ . Both fractions also show a weak band with  $\lambda_{\text{max}}$  405 m $\mu$ . (0.97 and 0.61% respectively).

(e) Occurrence of compound SA in intestinal lipids. The experiments detailed below were undertaken to test the possibility that SA is a saponification artifact.

A <sup>13</sup> ft. length of horse intestine was taken from the stomach end, the mucus expelled by squeezing and flushing with water. The mucus and intestine were each divided into two portions, one of which was saponified and the other used for extraction of lipid. Both lipids were chromatographed and the absorption spectra of the fractions were measured (Table 2).

Fractions showing a well-defined absorption minimum near 245 m $\mu$ ., a maximum near 272 m $\mu$ . and a much weaker peak near 405 m $\mu$ . were obtained by both methods (Fig. 1). The curves are identical in shape and it seems clear that the substance responsible is not a saponification artifact.

When the isolated lipid is saponified the yield of unsaponifiable matter confirms that obtained by direct digestion of mucus or tissue, but the  $272 \,\mathrm{m}\mu$ . maximum is not shown. The substance responsible is labile to alkali (Table 4), but is protected in some way during the alkali digestion of tissue. This result eliminates the possibility that the absorbing material is a saponification artifact.

 $Ox$  and sheep. Chromatography of unsaponifiable material from ox and sheep intestine revealed no substance absorbing selectively at  $270 \text{ m}\mu$ . Some carotene and 'xanthophylls' were, however, shown clearly in fractions from both mucus and intestinal tissue.

Horse stomach. The stomach and 10 ft. lengths of small intestine nearest to the stomach and caecum (including duodenum and ileum respectively) were obtained from the same animal. The stomach was cut open transversely and undigested food was removed. The oesophageal region (white and non-glandular) was separated from the darker glandular region where gastric juice is secreted (Sisson, 1943). The muscle layers could be peeled off easily using a scalpel. In each case the unsaponifiable fractions were chromatographed.

The '272  $m\mu$ . material' was found in all the tissues examined together with material showing appreciable absorption at wavelengths greater than  $320 \text{ m}\mu$ . Table 3 summarizes the results, and Figs. 2 and 3 indicate how the duodenal fraction shows the best  $270 \text{ m}\mu$ . curve, and the fraction from the outer layer of the oesophageal region the worst curve from the point of view of definition.

An unidentified carotenoid-like absorption spectrum  $(\lambda_{\text{max}}$ , 430 m $\mu$ ., inflexions 380, 410 and 460 m $\mu$ . in cyclohexane) was found in fractions from horse mesenteric fat, horse duodenum, ox and sheep intestine. It was not found in extracts from horse ileum or stomach. Zechmeister & Tuzson (1935) state that carotene is the chief carotenoid of the horse, but also report what seems to be the same absorbing entity as that mentioned here.

Rat liver. The absorption spectra shown by the unsaponifiable matter obtained from the livers of both normal and vitamin A-deficient rats showed  $\lambda_{\text{max}}$  270 m $\mu$ . In every case when these fractions were chromatographed on  $2\%$ watered alumina,  $4\%$  or  $6\%$  ether-light petroleum eluted a yellow fraction with  $\lambda_{\texttt{max.}}$  272-273 m $\mu$ .,  $\lambda_{\texttt{min.}}$  237m $\mu$ ., an inflexion about 330 m $\mu$ . and a very flat peak about 410 m $\mu$ . Compared with other tissues, rat liver proved a rich source of SA, fractions with  $E_{1 \text{ cm}}^{1 \text{ %}}$  up to 170 at 272 m $\mu$ . being obtained after a single chromatographic separation.

Rat kidney. The unsaponifiable materials from the kidney of normal and xerophthalmic rats were characterized by absorption spectra with  $\lambda_{\text{max}}$ , 270 m $\mu$ . When these fractions were chromatographed on 2% watered alumina, 4% etherlight petroleum always eluted a yellow band. The best fraction obtained was as follows  $(E_{1 \text{ cm}}^{1})$  in parentheses):  $\lambda_{\text{max}}$ . 273 m $\mu$ . (160.0),  $\lambda_{\text{min}}$ . 237 m $\mu$ . (42.5), inflexion,  $330 \text{ m}\mu$ . (11.5).

Rat intestine. The unsaponifiable materials from intestinal tissue of both control rats (small but definite liver stores of vitamin A) and frankly vitamin A-deficient rats, contained small amounts of SA (order 0.2-0.8% of unsap.). In each case a very clear-cut separation of SA from other substances was obtained using  $4\%$  ether-light petroleum for elution.

Rat submaxillary gland. A maximum at  $270 \text{ m}\mu$ . was characteristic of the unsaponifiable material from both control and vitamin A-deficient rat submaxillary glands. After chromatography on  $2\%$  watered alumina,  $4\%$  etherlight petroleum eluted a yellow zone. The best SA fraction obtained had  $\lambda_{\text{max}}$  271 m $\mu$ .  $E_{1 \text{ cm}}^{1 \text{ %}}$  83.8 and the inflexion near 330 m $\mu$ . was shown.

Other tissues of rat. Examination of bladder, testes and skin from around the eyes of both normal and vitamin Adeficient rats and the vaginas of vitamin A-deficient rats by similar methods, failed to show the presence of SA.

Hen liver. The unsaponifiable matter from hen liver showed  $\lambda_{\text{max}}$  271 m $\mu$ . Chromatography on 2% watered alumina again yielded <sup>a</sup> <sup>6</sup> % ether-light petroleum fraction characterized by  $\lambda_{\text{max.}}$  273 m $\mu$ ., the  $E_{1 \text{ cm.}}^{1 \text{ %}}$  was low (8.38) and definition poor.

Hen kidney. Again, the unsaponifiable fraction exhibited  $\lambda_{\text{max}}$  270 m $\mu$ . and on chromatography, 4% ether-light petroleum eluted a yellow oil with  $\lambda_{\text{max}}$  272-3 m $\mu$ .  $E_{1 \text{ cm}}^{1 \text{ %}}$ 30-5.









Fig. 2. ...., Duodenum fraction  $0.02\%$ ; ---, ileum Fig. 3. ...., Duodenum fraction  $0.2\%$ ; ---, ileum fraction  $0.015\%$ ; ---, outer layer of stomach glandular region lar region  $0.04\%$ ; ------, muscle layer  $0.05\%$ ;  $0.06\%$ ; ---------, muscle layer  $0.15\%$ ; -----, oesophageal stomach region outer layer 0-018%. layer of oesophageal stomach region 0-07%.  $-$ , oesophageal stomach region outer layer 0-018%.

fraction  $0.015\%$ ; ----, outer layer of stomach glandu-<br>lar region  $0.04\%$ ; ------, muscle layer  $0.05\%$ ;  $0.06\%$ ; --------, muscle layer  $0.15\%$ ; ----, outer

Figs. 2 and 3. Absorption curves of '272 m $\mu$ .' fractions obtained after chromatography of horse unsaponifiable material from different tissues (note change of scale at  $320$  m $\mu$ .).

#### Table 4. Absorption spectra of horse intestine and rat liver SA

Tests carried out on best available fractions-see text.



In conc.  $H_2SO_4$ , after 30 min. in dark at room temperature





Fig. 4. Absorption curves in cyclohexane for the best pre-<br>parations of SA from :  $\dots$ , horse intestine; and ——... rat parations of SA from:  $\cdots$ , horse intestine; and  $$ liver. When SA is obtained in a state of established purity  $\log E_{1\text{ cm}}^{1\%}$  values can be converted to  $\log \epsilon$  (molecular extinction coefficient).

#### Concentration of SA

The first experimentswere made on horse intestine material. Little success resulted from attempts to concentrate the material by partitioning between pairs of solvents. One fraction (75 mg.  $E_{1\,\text{em}}^{1\,\text{%}}$ , 70) was then shaken with five successive 5 ml. portions of cold methanol. The dissolved part (20 mg.) showed  $E_{1 \text{ cm}}^{1 \text{ %}}$  124 and the residue (59 mg.)  $E_{1 \text{ cm}}^{1 \text{ %}}$  58. The latter was, however, qualitatively the better in the sense that it had much less interfering absorption on the short wavelength side of the maximum.

Chromatography on 7% watered alumina showed the absorbing material to be slowly eluted with  $2\%$  ether-light petroleum. By combining

### Table 5. Fractions of SA obtained byrechromatography of two rich samples

Solvent for absorption spectra, cyclohexane.

Two samples bulked and placed on column:

(1) 17.5 mg.  $\lambda_{\text{max}}$ , 272 m $\mu$ .,  $E_{1 \text{ cm}}^{1 \text{ %}}$ , 172.  $P=3.71$ 

(2) 16.9 mg.  $\lambda_{\text{max}}$  272 m $\mu$ .,  $E_{1 \text{ cm}}^{\overline{1}}$ . 176.  $P = 3.78$ 

Fractions obtained on chromatography.



fractions showing the highest persistence  $P (=E_{\text{max}})$  $E_{\text{min}}$ ), and rechromatography, 10 mg. of a yellowish semi-solid material  $\lambda_{\text{max}}$ , 272 m $\mu$ .  $E_{1\text{ can}}^{1\text{%}}$ , 195 (cyclohexane) was obtained. This was chromatographed again and a very small amount of substance with slightly better persistence resulted. The spectroscopic properties of this material are compared with those of the best samples of rat liver SA in Table 4, while Fig. 4 compares their ultraviolet absorption spectra.

In view of the earlier experience with 'horse intestine SA', later attempts were made to concentrate the rat liver substance on watered alumina. It has been found that  $2\%$  water-weakened alumina is the most effective adsorbent for separation, SA being eluted with  $4-6\%$  ether-light petroleum mixtures. Using this procedure, two fractions totalling 34.4 mg. and with  $\lambda_{\text{max}}$  272-3 m $\mu$ .  $E_{1\text{cm}}^{1\%}$  172 (P=3.71) and 176 (P=3.78) were obtained. These were rechromatographed and three very viscous yellow samples of SA were obtained  $(a, b \text{ and } c)$ , all being eluted with  $4\frac{9}{6}$  ether-light petroleum (see Table 5).

Although rechromatography did not increase the  $E_{1}^{1\%}$ , value, it did raise the persistence of the best fraction (b) from 3-7 to 4-7. There are two possible explanations for this failure to raise the  $E_{1 \text{ cm}}^{1 \text{ %}}$  value by rechromatography; first, the material may be nearly pure and secondly, it is just possible that SA is mixed with a non-absorbing material from which it is inseparable chromatographically. This point will be mentioned later in the discussion.

Substance SA failed to give an insoluble digitonide.

Infrared absorption. The absorption curves of  $a, b$  and  $c$  (Table 5) were recorded in  $CS_2$  solution (Grubb-Parsons double-beam spectrometer). Samples a and b could not be distinguished; the principal maxiima were at 2915, 1656, 1608, 1379, 1289, 1259, 1171, 1096, 1024, 980, 840, 741 cm.-'. Sample  $c$  differed only from  $a$  and  $b$  in that the peak at 1171 had disappeared and maxima at 1156 and 1196 cm.-' were present. Fig. 5 shows the infrared absorption spectrum of fraction b.

Analysis. Fraction <sup>c</sup> contained no nitrogen and its molecular weight (Rast) was 430. Fraction b: C, 82 $\cdot$ 0; H, 10 $\cdot$ 25 $\%$ ; 0 (by difference), 7 $\cdot$ 75 $\%$ . This corresponds with  $C_{14}H_{21}O$  (requires C, 81.9; H,  $10.3\%$ ; mol.wt. 410) which from the mol.wt. indicates  $C_{28}H_{42}O_2$ .

Evidence for a second substance (with  $\lambda_{\max}$  ca. 266 m $\mu$ .). A large number of 'horse' preparations eluted before or with carotene were combined (0.43 g.) and chromatographed again on weakened alumina. Light petroleum eluted 0 34 g. of material absorbing weakly at  $257 \text{ m}\mu$ . Two small fractions land m were eluted when a little ether (5 and  $10\%$ ) was added to the solvent  $(E_{1 \text{ cm}}^{1 \text{ %}} 266 \text{ m}\mu$ . 33 and 64.  $P=1.05$  and  $1.19$  respectively, solvent: light petroleum).

Both fractions were examined in conc.  $H_2SO_4$  and showed a sharp band at  $273 \text{ m}\mu$ . (*l*) also showed an inflexion at  $355 \text{ m}\mu$ , which became a definite band in  $(m)$ . When the ultraviolet curves are corrected for end-absorption the rise in the 355 m $\mu$ . band (in conc.  $H<sub>a</sub>SO<sub>a</sub>$ ) is found to run exactly parallel to the increase in concentration of the  $266 \text{ m}\mu$ . substance in  $(m)$ .

Cholesta-3:5-dien-7-one shows  $\lambda_{\text{max}}$  269 m $\mu$ . in light petroleum and  $\lambda_{\text{max}}$  356 m $\mu$ . in conc. H<sub>2</sub>SO<sub>4</sub> at the correct relative intensities for identifying this substance. Although this is not conclusive it seems probable that the fractions  $l$  and  $m$  contain about 1.5 and  $4.5\%$  respectively of the dienone or a closely related substance.

Carotenoid. A carotenoid-like substance with  $\lambda_{\text{max}}$ , 430 m $\mu$ . (inflexions at 410 and 460 m $\mu$ .) was found in fractions from horse duodenum, mesenteric fat and ox and sheep intestine lipids. In all probability this is a breakdown product of  $\beta$ -carotene. In a forthcoming paper Dr J. Glover and Dr E. R. Redfearn will compare these bands with those of 10'-apocarotenal prepared by chemical methods from  $\beta$ -carotene. This compound could arise by  $\alpha$ xidative metabolism of  $\beta$ -carotene in the gut and its presence is the likeliest explanation of the observations here recorded.

Hydrocarbon from horse intestine. A colourless hydrocarbon m.p.  $57^{\circ}$  (not sharp) was isolated from the combined light petroleum eluates  $(C, 85.2\%;$ H,  $14.6\%$ ; mol.wt. (Rast) 503). This was probably a normal paraffin  $(C_{36}H_{74}$  requires C, 85.3; H,  $14.7\%$ ; mol.wt. 506).

#### DISCUSSION

Inspection of the spectroscopic characteristics, summarized in Table 4, and Fig. 4, together with the similar adsorptive properties, leaves little doubt that the horse intestine  $272 \text{ m}\mu$ . material and rat  $272 \text{ m}\mu$ . material are the same. Isolated from



either source, SA shows the same position of  $\lambda_{\text{max}}$ . in different solvents and the same loss of selective absorption in ethanolic KOH.

Fig. 4 shows that there is close agreement of the ultraviolet absorption spectra of the materials isolated from horse intestine and rat liver even when the spectra are subjected to the severe test of plotting  $\log E_{1 \text{cm}}^{1 \text{ %}}$  against wavelength. There is, however, a discrepancy between the sulphuric acid spectra; the horse intestine material shows a weak  $\lambda_{\text{max}}$  480 m $\mu$ . which is not exhibited by the rat liver material. As the  $E_{1 \text{ on}}^{1 \text{ %}}$  values at corresponding points are slightly higher for the rat SA than for the horse SA in sulphuric acid, it is of interest to compare them at  $480 \text{ m}\mu$ . Here the weak maximum of the horse material has  $E_{1 \text{ cm}}^{1 \text{ %}}$  60.0 and the smooth curve of the rat material  $E_{1 \text{ cm}}^{1 \text{ %}}$  46.0. Since, at this point, the relative intensities are reversed compared with other portions of the spectrum, it seems that the horse preparation contains an impurity not foundin the rat product. At this point, it should be mentioned that substances other than SA (to be discussed in later communications) occur in rat tissue which also show  $\lambda_{\text{max}}$  315 m $\mu$ . in sulphuric acid. Comparison of the ratio E 315 m $\mu$ . (H<sub>2</sub>SO<sub>4</sub>)/  $E_{272 \text{m}\mu}$  (cyclohexane) for the best fraction obtained from rat kidney (1-92) with that obtained from rat liver (1-99) suggests that these samples were free from such congeners.

The methods employed in the isolation of SA impose sharp limitations upon possible chemical structure. The fact that SA occurs in tissue unsaponifiable fractions implies that it could be a hydrocarbon, a steroid, a fat-soluble vitamin, a higher alcohol or perhaps a decomposition product. The absorption spectrum does not fit that of any known fat-soluble vitamin, and the fact that SA occurs in the tissues of xerophthalmic rats makes it unlikely to be a decomposition product of vitamin A. Analysis rules out a hydrocarbon, but the molecular formula points towards a steroid or similar lipid having two oxygen atoms per molecule. Compared with cholesterol, which is eluted from  $2\%$  watered alumina by  $8-10\%$  ether-light petroleum mixtures, SA must be relatively non-polar, because it is easily eluted from alumina of similar strength by <sup>4</sup> % ether-light petroleum mixtures. This fact suggests that the two oxygen atoms present in the molecule are more likely to be ketonic rather than hydroxyl groups, a conclusion supported by 'the infrared spectrum which shows no significant band in the region  $2.68-3.22 \mu$ , the wavelength region for the stretching vibrations of the hydroxyl groups.

The dominant  $270 \text{ m}\mu$ . band in the ultraviolet absorption spectrum suggests a chromophore consisting of either three conjugated double bonds or, of a keto group conjugated to two double bonds or an ene-dione.

At this point, the infrared spectrum may be helpful. The band at  $6.22 \mu$ . (1608 cm.<sup>-1</sup>) implies either a conjugated or unconjugated (C:C) double bond [if conjugated the band would be expected at  $6.21-6.33 \mu$ . (1610-1580 cm.<sup>-1</sup>) if unconjugated the range is  $6.06-6.25 \mu$ . (1650-1600 cm.<sup>-1</sup>) Miller, 1953]. The peak at  $6.04 \mu$ . (1656 cm.<sup>-1</sup>) agrees with the stretching vibration of a conjugated ketonic group (Jones & Dobriner, 1949). Of the other principal bands shown in the infrared spectrum that at  $3.43 \mu$ . (2915 cm.<sup>-1</sup>) corresponds to  $-CH_2$ stretching and that at  $7.25 \mu$ . (1379 cm.<sup>-1</sup>) to deformation of the methyl group of an isopropyl configuration, while that at  $8.54 \mu$ . (1171 cm.<sup>-1</sup>) agrees with the band due to the *iso*propyl group as a whole. The bands beyond  $8.5 \mu$ . (1176 cm.<sup>-1</sup>) are associated with vibrations subject to effects of interaction (fingerprint region) and interpretation of the bands occurring here is not obvious.

In view of the method of preparation of SA, and because of the absence of selective absorption in the region  $1720-1760$  cm.<sup>-1</sup>, the strong band at 1259 cm.<sup>-1</sup> (7.94  $\mu$ .) cannot be assigned to an ester; this band still requires explanation.

From the foregoing, SA is probably a steroid or similar lipid with a double bond conjugated with either one or two ketone groups. Furthermore, the steroid is likely to possess a cholesteryl side chain and angular methyl groups at  $C_{10}$  and  $C_{13}$  in order to account not only for the molecular formula but also the isopropyl and  $-C-CH<sub>a</sub>$  groups indicated in the infrared spectrum.

The ultraviolet absorption spectra of steroids have been summarized and classified in a recent comprehensive review by Dorfman (1953). The number of chromophores corresponding with selective absorption at  $270-275$  m $\mu$ . is not large (see Table 6), and the nearest behaviour to that of SA is shown by compounds containing the 8(9) ene-7: <sup>1</sup> 1-dione chromophore.



It is of interest to note that the carbonylstretching band for steroids with the  $8(9)$ -en-11-one configuration is at  $1660$  cm.<sup>-1</sup>, whereas for steroids with the 11-one group only it is at  $1710-1704$  cm.<sup>-1</sup> (Jones & Herling, 1954). The band at 1656 cm. $^{-1}$  in the infrared spectrum of fraction  $b$  (Table 5) has been assigned to the carbonyl groups. Unfortunately all the known substances which possess this chromophore also contain substituents at position 3. Table 6. Steroids showing  $\lambda_{\text{max}}$  near 270  $m \mu$ .



#### See Dorfman, 1953.

For 8(9)-ene-7:11-dione chromophore see also Fieser & Herz (1953). Fieser, Huang & Babcock (1953) and Fieser, Schneider & Huang (1953).

Furthermore, the tendency to report spectra in the steroid field in terms of  $\lambda_{\text{max}}$  and  $\epsilon_{\text{max}}$  only, means that inflexions are often not recorded in the literature. However, the curve for  $3\beta$ -acetoxyergosta - 8:22 - diene - 7:11 - dione (Heusser, Eichenberger, Kurath, Dallenbach & Jeger, 1951) shows a minimum near  $235 \text{ m}\mu$ . and apparently an inflexion in the region of  $330 \text{ m}\mu$ . The evidence is consistent with the presence of this chromophore in SA.



#### SUMMARY

1. A substance (SA) with an absorption peak at  $272 \text{ m}\mu$ . (in cyclohexane), an inflexion near 330 m $\mu$ . and a plateau near  $410 \text{ m}\mu$ .  $(E_{1\text{ cm}}^{1\text{%}} 180, 13 \text{ and }$ 8 approx. respectively) has been obtained by chromatography on alumina of various animal tissue unsaponifiable fractions.

2. SA is present in pig intestine, horse stomach and intestine, rat liver, kidney, intestine, and submaxillary gland whether from vitamin A-deficient or normal rats. SA was not found in unsaponifiable

fractions from ox or sheep intestine or from bladder, testis and vagina of vitamin A-deficient rats.

3. SA is not very strongly adsorbed on alumina, it contains two oxygen atoms (no hydroxyl) and a possible formula is  $C_{28}H_{42}O_2$  (molecular weight observed 430); it is alkali-labile, particularly so in the purer state. From the polarity, infrared and ultraviolet absorption and probable empirical formula, SA could be a steroid with a chromophoric grouping 8(9)-ene-7: <sup>1</sup> 1-dione.

4. Horse intestine unsaponifiable matter contained small quantities of material slightly less polar than SA with  $\lambda_{\text{max}}$  266 m $\mu$ . (possibly cholesta-3:5-dien-7-one) a carotenoid (probably 10'-apocarotenal) and a higher paraffin  $(m.p. 57^{\circ}, mol.wt.$ 503).

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# The Determination of Calcium in Serum by Flame Photometry

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Since 1948, when Riehm first demonstrated the feasibility of the direct flame-photometric method for the determination of calcium in dilute solutions, this principle has been applied to analysis of soil (Stanford & English, 1949), milk (Keirs & Speck, 1950) and serum (Severinghaus & Farabee, 1950). In the following year further reports of serum calcium determinations appeared from Leyton (1951) and from Mosher, Itano, Boyle, Myers & Iseri (1951), followed by those from Kapuscinski, Moss, Zak & Boyle (1952), Zak, Mosher & Boyle (1953), and Powell (1953). While the present account was being written, direct determinations on diluted serum have been described by Winer & Kubins (1953), Chen & Toribara (1953) and by McIntyre (1954). Several of the American authors used modified spectrophotometers, and in nearly all cases sera have been prepared for analysis by procedures involving oxidation, precipitation of calcium, or removal of sodium. The comparative advantages of simply diluting with water are evident. Although many forms of photometer, most of them elaborate, have been described, the desirability of using simple and relatively inexpensive equipment still remains.

The determination of calcium in blood by flame photometry presents peculiar difficulties. The concentration of calcium is small but that of sodium and potassium large, while the concentration range of calcium occurring in human sera is narrow, so that considerable accuracy is required. The normal range in human serum is  $9.0-11.5$  mg. Ca/100 ml., and even grossly pathological conditions extend this ambit only to the limits 4-16 mg./100 ml. If precipitation and other pretreatment of the sample are to be avoided, the serum must be diluted at least tenfold, so that the final concentration of calcium is about <sup>1</sup> mg./100 ml. This dilute calcium solution emits very little light from a flame. The use

of an interference filter permits the partial isolation of the calcium oxide red bands (approx.  $6200 \text{ Å}$ ), provided that didymium glass is used to reduce the excess ofsodiumlight also emitted. In an acetyleneair flame, maximum output of energy in the calcium bands is ensured, and the intensity of this light, although very low, can be measured either by the use of a photoconductive cell (Schwarz, 1948) as described below, or by means of photomultiplier apparatus.

The construction and use of an internal standard photometer which has proved satisfactory for the determinations of calcium, sodium and potassium in diluted serum and other material are described below.

### EXPERIMENTAL

#### De8cription of photometer

The general lay-out of the instrument is shown diagrammatically in Fig. 1.

The burner unit. Acetylene is burnt after admixture with the air-borne sample, and no atmospheric air enters the system. The brass Meker-type burner is shown in section in Fig. 2. The jet  $(A)$  is a standard welding component (British Oxygen Co. Ltd., London, type DH, 2 cu.ft./hr. i.e. 940 ml./ min.), and is screwed into position over a fibre washer (B). The top-plate  $(C)$ , of brass, is perforated as shown. Acetylene at a pressure only slightly above atmospheric enters at  $D$  and the air-mist through  $E$ . The burner is clamped in a boss attached to a plate  $(A, Fig. 3)$  bearing a light-trap  $(B)$  and bolted to a plate  $(C)$  of similar outline brazed to the chimney-tube  $(D)$  which it supports. This tube has diametrically opposed apertures  $(E, E')$  for exit of light, and a smaller hole  $(F)$  for insertion of a taper for lighting.

Gas control and atomizer. Both acetylene and air are taken from industrial cylinders fitted with two-stage regulators. The acetylene regulator is fitted with an outlet needle-valve which is used as on-off control to avoid readjustment of the succeeding valve. Fine needle-valves incorporated in the photometer are used for both gases. Acetylene from its