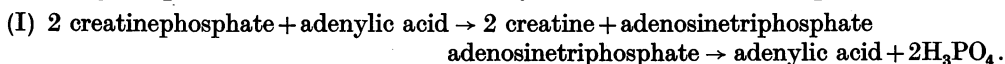


14. The Adenosinetriphosphatase Activity of Myosin Preparations

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It has been known for many years that muscle contraction is accompanied by appearance of equivalent amounts of inorganic phosphate and free creatine, creatinephosphate disappearing. Until the work of Lohmann [1934] using muscle extract, this process was regarded as an enzymic hydrolysis of creatinephosphate, going on with liberation of energy. But Lohmann showed that the breakdown of creatinephosphate is a composite process, requiring the presence in the extract of adenylic acid and adenosinetriphosphatase:



Thus free energy of creatinephosphate breakdown is used for resynthesis of adenosinetriphosphate (ATP). Further work in several laboratories has since shown that the energy-rich phosphorylated intermediates in carbohydrate breakdown, phosphopyruvic acid [Ostern *et al.* 1935; Needham & van Heyningen, 1935; Meyerhof & Lehmann, 1935; Lehmann, 1935] and diphosphoglyceric acid [Warburg & Christian, 1939] are similarly used in the muscle to rebuild ATP (see Fig. 1). The central importance of ATP breakdown

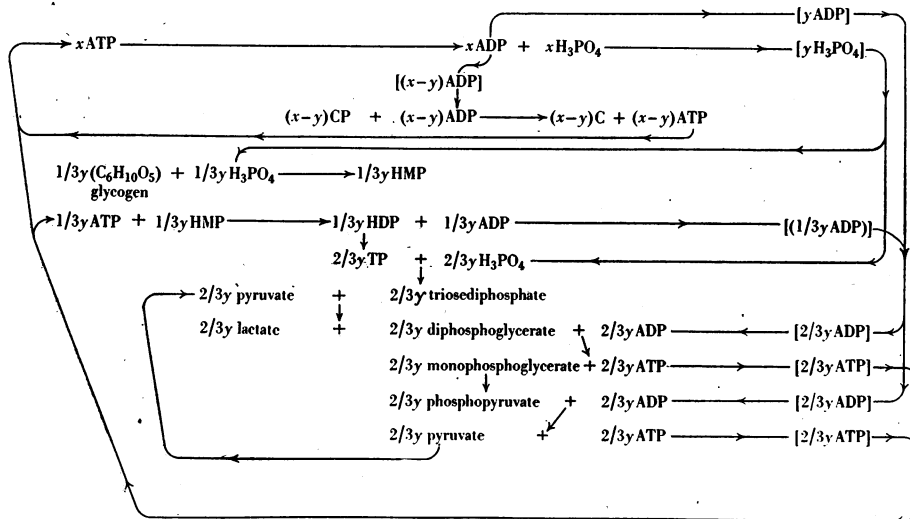


Fig. 1. Diagram showing chemical changes in muscle contraction. ATP=adenosinetriphosphate; ADP=adenosinediphosphate; CP=creatinephosphate; C=creatine; HMP=hexosemonophosphate; HDP=hexosediphosphate; TP=triosephosphate.

as (1) the only reaction going on with liberation of energy available for the process of contraction, (2) the first chemical reaction demonstrable upon stimulation, thus becomes obvious [see Needham, 1937; 1938].

The correctness of the view that the inorganic phosphate formed on contraction has its origin in ATP is borne out by the fact that muscle brei shows high phosphatase activity with ATP as substrate, two phosphate groups being split off, but that other phosphatases are very feeble or absent.

Lohmann [1935, 1, 2] showed that in the formation of adenylic acid from ATP, adenosinediphosphate (ADP) occurs as an intermediate, and indications have accumulated that probably in the muscle itself reaction (I) runs



dephosphorylation of ATP perhaps usually not proceeding past the loss of one phosphate group.

The important work of Engelhardt & Liubimova [1939] and of Liubimova & Engelhardt [1939] showed that the adenosinetriphosphatase activity of muscle is associated with the myosin fraction of the muscle proteins. After purification of the myosin by repeated reprecipitation, the protein was still just as active in splitting off one phosphate group from ATP, but had lost entirely the power, shown by the brei or by once-precipitated myosin, of converting ADP into adenylic acid [Liubimova & Pevsner, 1941; Bailey, 1942].

The question of the identity of adenosinetriphosphatase and myosin has been considered by Liubimova & Engelhardt [1939] and by Bailey [1942]. If the protein myosin, of which the contractile fibrils are composed, can indeed interact enzymically with the substance ATP, whose breakdown provides the energy of contraction, then the significance of this fact in the mechanism of contraction must be very great. The mode of interaction between the myosin and the ATP, and the ways in which this interaction might affect the length of the fibrils, has been considered by Engelhardt & Liubimova [1939], Kalekar [1941; 1942], Needham *et al.* [1941] and Bailey [1942].

Most of the experiments described below were done in 1939 and early 1940. It was impossible to continue them, but the work has been continued and extended by Dr K. Bailey.

Experimental methods

(1) *Preparation of myosin.* The myosin was made from rabbit muscle and from the muscle of *Rana esculenta*, usually by the method of Greenstein & Edsall [1940], sometimes by that of Weber & Meyer [1933]. On one occasion, a LiCl extract of rabbit muscle was made [Smith, 1934], and the myosin prepared from this had the same degree of activity as the myosin made from muscle of the same animal by Greenstein & Edsall's method. The myosin was usually precipitated twice. The myosin content was assessed by estimating the protein N.

(2) *Preparation of ATP.* The ATP was prepared as the Ba salt by the method in use in Prof. Parnas's laboratory at Lwów: I am indebted to Dr C. Lutwak-Mann for details [see also Lohmann & Schuster, 1935].

A rabbit is killed, bled and skinned. The carcass is cooled in ice and then the muscle is rapidly removed. It is kept cold and minced through a cooled mincer. It is then mixed with an equal weight of ice-cold 10% trichloroacetic acid and well stirred. After a few minutes the mixture is strained through muslin, the residue is squeezed as dry as possible and again extracted with an equal weight of 4% trichloroacetic acid. After straining again, the united filtrates are centrifuged. The supernatant liquid is brought with 40% NaOH to pH 6.8, and 50% Ba acetate is added until no more precipitate is formed. After 1-2 hr. at 0°, when the precipitate (Ba salts of ATP and inorganic phosphate) has settled, the supernatant liquid is decanted off, the precipitate is centrifuged down and washed with water. The precipitate is treated with 0.2 N HNO₃ until the reaction is faintly blue to Congo-red paper and the small amount of insoluble material is rejected. The ATP is then precipitated as the Hg salt by means of Lohmann's reagent (100 g. HgNO₃, 8H₂O dissolved in a mixture of 25 ml. HNO₃, sp. gr. 1.4, and 25 ml. water [Lohmann, 1931]) added at 0°, about 3-5 ml. of the reagent being needed per kg. of muscle; after precipitation the mixture is left at 0° for 15 min. and then spun. The precipitate is suspended in a small

volume of water made faintly acid with HNO_3 , and the Hg is removed by H_2S and filtering. H_2S is removed from the filtrate by aeration, and the reaction is brought to pH 6.8 with dilute NaOH. 25 % Ba acetate is added until precipitation is just complete, excess being avoided. The precipitate of Ba salt is filtered off on a small Büchner funnel; it is washed well with 1 % Ba acetate, then with 50 %, 75 % and 97 % alcohol and finally with ether. For use in the experiments, Ba was removed by dissolving the salt in the minimum amount of *N* HCl, diluting and adding sat. Na_2SO_4 solution, avoiding excess, until no further precipitate was obtained. The BaSO_4 precipitate was spun off and washed, the combined supernatant liquids were neutralized and made up to the required volume.

Two phosphate groups of the ATP constitute the pyrophosphate-P, and are hydrolysable by *N* acid in 7 min. at 100° ('7' P').

(3) *Method of testing the enzyme activity.* The myosin gel, after precipitating for the second time and centrifuging down, was suitably diluted with distilled water and carefully mixed to give a uniform suspension. 1 ml. of this suspension was pipetted into each of a series of test-tubes and 0.5 ml. of the ATP solution was added. This contained about 0.3 mg. 7' P/ml. The pH was adjusted by means of NaHCO_3 solution, as in the experiments of Liubimova & Engelhardt [1939], 0.1 ml. of the required concentration being added. In some of the later experiments, NaHCO_3 - Na_2CO_3 buffer, as recommended by Bailey [1942], or Michaelis veronal buffer was used. MgCl_2 and CaCl_2 , 0.05 ml. of a 0.1 *M* solution, were used as activators; the metal was thus present in amount approximately equivalent to the pyro-P. The contents of one tube were precipitated at once by adding 1.5 ml. of 10 % trichloroacetic acid; the others were incubated at 37° for varying lengths of time before addition of trichloroacetic acid. The myosin clot was filtered off, the tubes and filter were washed with 4 % trichloroacetic acid, and the inorganic P was estimated in the filtrate by Fiske & Subarrow's method.

RESULTS

Distribution of adenosinetriphosphatase activity between aqueous extract of muscle and muscle residue. 5 g. of muscle were ground very thoroughly with sand and 7.5 ml. distilled water. The brei was allowed to stand at room temperature for 4 hr. in order that processes leading to resynthesis of ATP (and thus likely to obscure adenosinetriphosphatase activity) might be concluded. After this time, about 20 ml. more water were added; the whole was dialysed for 1.5 hr. against running water, strained through muslin and centrifuged. The extract was made up to 50 ml.; the residue was ground with more water and also made up to 50 ml. Tests showed that in 5 min. 0.5 ml. of the extract split off 0.004 mg. pyro-P, while 0.5 ml. of the residue suspension split off 0.043 mg. pyro-P; i.e. the activity of the water-soluble material in a given quantity of muscle is less than 10 % of that of the water-insoluble residue.

Adenosinetriphosphatase activity of myosin preparations. In the experiments with myosin from rabbit muscle it was found that (a) only one phosphate group was split off; or (b) one group was rapidly split off, and that thereafter a very slow liberation of phosphate continued. Two typical experiments are shown in Fig. 2. In the case of frog muscle, some splitting off of the second group was always observed (Fig. 3).

Activation by Mg^{++} and Ca^{++} . The effect of Mg^{++} upon the rate of dephosphorylation was first tried, as this ion is known to activate many phosphate transfer reactions in which ATP is concerned. Later, after the appearance of the paper of Liubimova & Engelhardt [1939], from which it seemed likely that Ca^{++} also must be able to serve as an activator, the effect of Ca^{++} was also tried (Table 1). These results, with two different preparations of rabbit myosin, illustrated the fact that about 30-90 % activation of inorganic P formation is obtained by adding Mg ions, and that under these conditions Ca ions have a similar effect.

When a much more alkaline medium was used, as in the experiments of Liubimova & Engelhardt, it was found that the effect of Ca ions is much greater, that of Mg ions much less (Table 2).

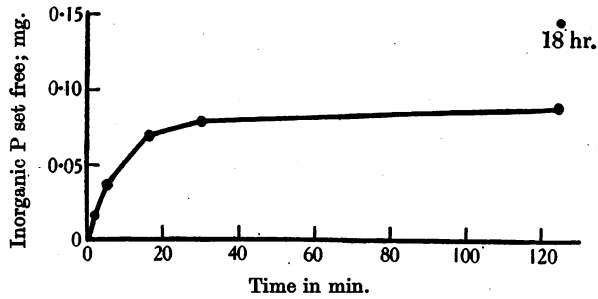


Fig. 2. Adenosinetriphosphatase activity of twice-precipitated rabbit myosin. Myosin suspension 1.85 mg. N/ml.; pH 7. ATP (0.15 mg. 7' P) added.

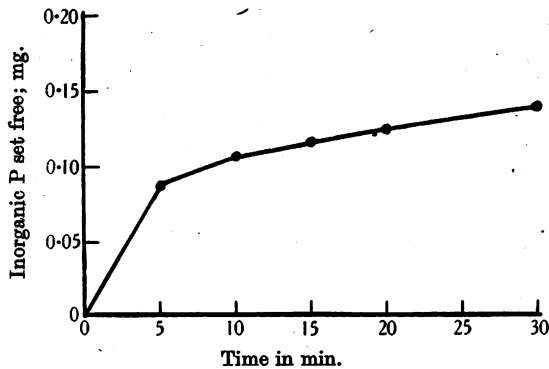


Fig. 3. Adenosinetriphosphatase activity of twice-precipitated frog myosin. Myosin suspension 0.34 mg. N/ml.; pH 7.5. ATP (0.15 mg. 7' P) added.

Table 1

The reaction mixture contained 1 ml. myosin suspension (1.5 mg. protein) 0.5 ml. ATP (0.3 mg. 7' P) and 0.1 ml. 1% NaHCO₃.

| | No activator | +0.05 ml. 0.1 M MgCl ₂ | +0.05 ml. 0.1 M CaCl ₂ |
|----------------------------|--------------|-----------------------------------|-----------------------------------|
| Phosphate set free (mg. P) | 0.062 | 0.082 | — |
| | 0.017 | 0.033 | 0.033 |

Table 2

The reaction mixture contained 1 ml. myosin suspension (2.4 mg. protein) and 0.5 ml. ATP (0.3 mg. 7' P).

| | Phosphate set free, mg. P |
|---|---------------------------|
| 0.05 ml. 0.1 M MgCl ₂ ; 0.1 ml. 2% NaHCO ₃ | 0.073 |
| 0.05 ml. 0.1 M CaCl ₂ ; 0.1 ml. 2% NaHCO ₃ | 0.073 |
| 0.05 ml. 0.1 M MgCl ₂ ; 0.1 ml. 10% NaHCO ₃ | 0.033 |
| 0.05 ml. 0.1 M CaCl ₂ ; 0.1 ml. 10% NaHCO ₃ | 0.109 |

From the later experiments of Bailey, it is clear that Ca ions have a much greater effect than any other on adenosinetriphosphatase. It is likely that in the above experiments, where the protein concentration was relatively high, that enough Ca ions were present in the myosin suspension itself to activate fairly well at the low pH, but not at the high pH, where probably decreased Ca solubility has an effect.

Since the experiments of Bailey were all carried out between pH 8 and 9, and since the pH curve of Liubimova & Engelhardt, between pH 7.4 and 9, was determined for Ca activation only, it seemed worth while to see whether with thrice-precipitated myosin and low myosin concentration (i.e. using enzyme as free as possible from Ca⁺⁺) Mg⁺⁺ might activate at lower pH in a manner comparable with Ca⁺⁺. The following experiment shows that this is not the case. Veronal buffer was used and since this is a weaker buffer than bicarbonate a larger volume had to be used (Table 3).

Table 3

The reaction mixture contained 2.5 ml. of a myosin suspension (about 1 mg. protein) in *M*/35 veronal buffer, pH 7.4, and 0.5 ml. ATP (0.2 mg. 7' P).

| Addition | Time (min.) | Phosphate set free, mg. P |
|---|-------------|---------------------------|
| 0.05 ml. 0.1 <i>M</i> MgCl ₂ | 5 | Trace |
| | 10 | 0.01 |
| | 45 | 0.05 |
| 0.05 ml. 0.1 <i>M</i> CaCl ₂ | 5 | 0.081 |
| | 10 | 0.093 |
| | 45 | 0.097 |

Effect of iodoacetate and of NH₄ salts. It has been shown that myosin in the undenatured state contains —SH groups, and that these can react with iodoacetate; they also disappear upon treatment with NH₄ salts [see e.g. Mirsky, 1936; Greenstein & Edsall, 1940]. To see if the presence of these —SH groups is essential for adenosinetriphosphatase activity, the following experiments were made.

(a) The reaction between —SH compounds and iodoacetate is slow, and the myosin must therefore be treated with the iodoacetate for some time before the substrate is added. 1 ml. samples of myosin (containing about 1 mg. protein) were incubated with *M*/300 iodoacetate for 15 min. at 37°; substrate, buffer and activator were then added, and the activity was compared with that of myosin similarly incubated without iodoacetate. It was found, however, that practically all activity had been lost in both cases; this inactivation by warming in absence of substrate (especially if the protein concentration is very low, as here) had been observed by Liubimova & Engelhardt. The treatment with iodoacetate was therefore carried out at room temperature and in a higher protein concentration.

10 ml. of myosin gel (containing about 10 mg. protein per ml.) were treated with 1 ml. *M*/30 iodoacetate. Portions were removed at intervals for the nitroprusside test; this was still faintly positive after 1 hr., but completely negative after 2 hr. 1 ml. of the iodoacetate-myosin was mixed with 9 ml. of carbonate-bicarbonate buffer, pH 9.1, final concentration 0.2%, and the activity was compared with that of a sample of myosin which had been similarly treated, but to which water was added instead of iodoacetate (Table 4). There is thus no inhibition; experiments with *M*/300 iodoacetamide gave a similar result.

Table 4

| Exp. | Conditions | Phosphate set free, mg. P |
|------|-------------------------------|---------------------------|
| 1 | Normal myosin, in 7 min. | 0.079 |
| | Iodoacetate myosin, in 7 min. | 0.077 |
| 2 | Normal myosin, in 5 min. | 0.041 |
| | Iodoacetate myosin, in 5 min. | 0.042 |

(b) 10 ml. of myosin gel were treated with NH₄Cl to give a final concentration of 2*N*. The sol was dialysed with frequent changes of water for some hours, and the precipitated myosin was spun down. The activity was compared before and after this treatment (Table 5). Here again there is no inactivation. That the —SH groups are inessential is

also shown by the fact that the enzymic activity persists in glycine buffer [Bailey, 1942] although Greenstein & Edsall [1940] found that treatment with glycine abolished the nitroprusside reaction.

Table 5

| | Phosphate set free, mg. P per mg. N in 10 min. |
|---|--|
| Before NH_4Cl treatment | 0.032 |
| After NH_4Cl treatment | 0.040 |

Specificity of the enzymic activity of myosin. (a) It has been conclusively shown by Liubimova & Engelhardt [1941] and by Bailey [1942] that myosin has no adenosine-diphosphatase activity; Bailey also showed that it does not hydrolyse inorganic pyrophosphate, although muscle brei does. In the present work its specificity was further tested upon α -glycerophosphate and hexosediphosphate. The latter is of special interest since it seems to be the only metabolite besides ATP to be directly dephosphorylated at all readily by muscle [see Pillai, 1938]. Myosin precipitated three times was used, and was tested at pH 9.1 (carbonate-bicarbonate buffer) and pH 6.1 (veronal buffer), the substrate concentration being $M/75$. Activation was tried with both Mg^{++} and Ca^{++} , but even in the presence of Mg^{++} not more than 1-2% of the phosphate of α -glycerophosphate, or 3-4% of that of hexosediphosphate, was hydrolysed in 3 hr. at 37°.

(b) Although myosin so readily catalyses the transfer of phosphate from ATP to water, it has no action in bringing about transfer of phosphate from ATP to fructose-6-phosphate (Neuberg ester). This was shown in the following experiments, carried out both at pH 9.1 and pH 7.4 (Table 6). In another experiment the possibility of transesterification

Table 6

The reaction mixture contained 1 ml. of myosin suspension (about 1 mg. protein) in carbonate-bicarbonate buffer, pH 9.1, 0.5 ml. ATP (0.2 mg. 7' P) and 0.05 ml. 0.1 M CaCl_2 .

| Addition | P set free, mg. | |
|-----------------------------------|-----------------|------------|
| | In 7 min. | In 20 min. |
| 0.3 ml. H_2O | 0.075 | 0.098 |
| 0.3 ml. Neuberg ester (0.2 mg. P) | 0.075 | 0.098 |

at pH 7.4 in presence of Mg^{++} was tested. In this case the P hydrolysable in 10 min. in N HCl at 100° was also estimated: in this hydrolysis time, two phosphate groups are split off from ATP, about 7% of the P of fructose-6-phosphate and about 37% of the hexosediphosphate-P. If now some phosphate were transferred from ATP to the Neuberg ester, with formation of hexosediphosphate, there would be a decrease in the amount of P hydrolysable in 10 min. It can be calculated that if only 10% of the pyrophosphate-P of ATP were thus transferred, the decrease would amount to 7%, which is well outside the experimental error (Table 7).

Table 7

The reaction mixture contained 1 ml. myosin suspension (1 mg. protein) in water, 0.5 ml. ATP, pH adjusted to 7.4 (0.22 mg. 7' P) and 0.05 ml. 0.1 M MgCl_2 . 30 min. at 37°.

| Addition | Inorg. P | 10' P |
|--|----------|-------|
| + 0.3 ml. Neuberg ester (containing 0.2 mg. P) | 0.026 | 0.207 |
| + 0.3 ml. H_2O ; 0.3 ml. Neuberg ester added after trichloroacetic precipitation | 0.027 | 0.206 |

(A separate test with this myosin showed that with Ca activation it had the usual high adenosinetriphosphatase activity at pH 9.1, setting free 0.05 mg. P in 5 min.)

Thus just as much P is liberated as inorganic P in the presence as in the absence of Neuberg ester, and there is no indication of any hexosediphosphate formation from this ester. This result is in harmony with the finding by Sakov [1941] of an enzyme in aqueous extract of muscle which brings about phosphorylation of Neuberg ester by ATP.

DISCUSSION

The diagram (Fig. 1) is an attempt to summarize the chemical events during the actual stimulation period under anaerobic conditions. Hydrolysis of ATP is presented as the essential reaction supplying free energy to the contracting fibrils. The resynthesis of part of this ATP (an unknown proportion) by reaction of the ADP with creatinephosphate is shown, and the resynthesis of the rest by means of phosphorylated carbohydrate derivatives. The diagram makes clear (a) how the ATP concentration is maintained constant in spite of its continual breakdown; (b) how the free phosphate and free creatine formed come to be equivalent to the creatinephosphate disappearing, although not formed from it by direct hydrolysis.

When glycogen is phosphorylated by inorganic phosphate in muscle extract, the first product is the Cori ester, glucose-1-phosphate, and this is transformed into the Robison ester, glucose-6-phosphate, which is in equilibrium with fructose-6-phosphate. It is the last, the Neuberg ester which is phosphorylated by ATP to give fructosediphosphate, the Harden & Young ester.

It should be noticed that the reaction between ATP and Neuberg ester to form hexosediphosphate is accompanied by about the same large heat output as accompanies the hydrolysis of ATP, and it seems likely that a similar amount of free energy may be produced in the two reactions. Nevertheless, as the transfer of phosphate to the Neuberg ester is catalysed by a soluble enzyme, and not by the myosin fraction, it seems unlikely that this energy would be available to the fibrils. It would form part of the waste heat which always accompanies contraction. The diagram shows, however, that during anaerobic stimulation, a relatively small part of the ATP is broken down by transesterification, much the greater part by hydrolysis under the influence of adenosinetriphosphatase.

SUMMARY

1. The finding of Engelhardt & Liubimova was confirmed that myosin preparations have adenosinetriphosphatase activity, splitting off one phosphate group rapidly, the other very slowly or not at all. More than 90 % of the adenosinetriphosphatase activity of muscle is to be found in the washed brei, less than 10 % in the aqueous muscle extract.
2. Abolition of the —SH groups of myosin has no effect on the adenosinetriphosphatase activity.
3. Myosin preparations show no α -glycerophosphatase or hexosediphosphatase activity, either at pH 9.1 or pH 6.0; nor have they any power to transfer phosphate from ATP to fructose-6-phosphate.

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