

## Antibacterial Activity of Licochalcone A against Spore-Forming Bacteria

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Received 12 November 2001/Returned for modification 24 December 2001/Accepted 31 January 2002

**Licochalcone A was isolated from the roots of licorice, *Glycyrrhiza inflata*, which has various uses in the food and pharmaceutical industries; isolation was followed by extraction with ethanol and column chromatography with silica gel. In this study, the activities of licochalcone A against some food contaminant microorganisms were evaluated in vitro. The vegetative cell growth of *Bacillus subtilis* was inhibited in a licochalcone A concentration-dependent manner and was completely prevented by 3 µg of licochalcone A/ml. Licochalcone A showed a high level of resistance to heating at 80 to 121°C for 15 min. Licochalcone A did not inhibit the germination of heat-treated spores of *B. subtilis* induced by L-alanine. Licochalcone A showed effects against all gram-positive bacteria tested and especially was effective against all *Bacillus* spp. tested, with MICs of 2 to 3 µg/ml, but was not effective against gram-negative bacteria or eukaryotes at 50 µg/ml. Although the cationic antimicrobial peptides protamine and ε-poly-L-lysine resulted in the loss of antimicrobial activity in the presence of either 3% (wt/vol) NaCl or protease at 20 µg/ml, the antibacterial activity of licochalcone A was resistant to these conditions. Thus, licochalcone A could be a useful compound for the development of antibacterial agents for the preservation of foods containing high concentrations of salts and proteases, in which cationic peptides might be less effective.**

The bacterial spores of the genera *Bacillus* and *Clostridium* act as a survival stage, which is characterized by a high level of resistance to heat and other adverse conditions typically used to kill vegetative cells and other microorganisms (2). In food, the spores themselves do not represent a hazard. However, despite being metabolically dormant, the spores have a functional environmental sensory mechanism that can trigger germination under favorable conditions (22). Thus, the processes of germination, outgrowth (vegetative growth after germination), and/or toxin formation can result in spoilage and/or food poisoning (17). In practice, conditions for outgrowth in many foods are suboptimal due to the presence of a combination of factors, such as water activity, reduced pH, preservatives, organic acids or salt, and pasteurization steps (2).

Generally, the surface charge of microbial cells is anionic at a neutral pH because of the dissociation of acidic groups, such as phosphate and carboxylate, on the cell surface (16). Recently, a great deal of attention has been paid to cationic antimicrobial peptides, such as protamine and ε-poly-L-lysine, as natural food preservatives with a low toxicity. Protamine found in salmon spermatozoan nuclei (salmine) is a basic peptide of 32 amino acids, of which 21 are arginine (6, 10). Protamine has activity against a broad spectrum of bacteria, yeasts, and fungi, and this activity is considered to be the result of its polycationic nature (6, 10, 16). The broad antimicrobial spectrum of protamine and the fact that protamine is naturally occurring and nontoxic to humans make it a promising biological alternative to chemical preservatives and disinfectants (6, 10, 16). Alternatively, ε-poly-L-lysine from culture filtrates of

*Streptomyces albus* also shows antimicrobial activity with a broad spectrum (14, 16, 20, 23). These peptides have the common features of being highly basic due to the presence of multiple arginine and lysine residues and of forming amphipathic structures (14, 16). However, the growth inhibitory effects of these peptides against food contaminant microorganisms are repressed by the presence of a high concentration of salt, which could interfere with the binding of cationic peptides to the cell surface (10, 16). Although these peptides carry many cationic residues required for antimicrobial activity, peptides bound to the cell surface are subject to protease digestion (16). In terms of practical applications, cationic antimicrobial peptides would be less effective in the preservation of salted foods and are not available for the preservation of foods with protease activity.

Licorice, the root and rhizome of the *Glycyrrhiza* spp. *Glycyrrhiza uralensis* (9, 19), *Glycyrrhiza glabra* (3, 5, 13, 15, 18, 19), and *Glycyrrhiza inflata* (7, 8, 11, 18, 19), is currently used in the tobacco, food, and pharmaceutical industries. It has been used for centuries as a medicine because of its wide-ranging therapeutic properties, including relief of rheumatic and other pain, and its healing effect on gastric ulcers (19). The crude extract of licorice has also found commercial use as a food additive in Japan, as it contains the sweetening principal glycyrrhizin (19). Chemical investigations have revealed the presence of a wide variety of bioactive phenolic constituents in licorice; these have attracted attention as a potential source of chemical leads (19). *G. inflata* is one of the main botanical sources of licorice and is chemically characterized by the presence of retrochalcones, which are distinguished from ordinary chalcones by the absence of oxygen functionality at position 2 (7, 8, 11, 18, 19). Five retrochalcones, licochalcones A, B, C, and D and echinatin, were isolated from *G. inflata* roots and characterized (7, 8, 11, 18); the content of licochalcone A (Fig. 1) was found to

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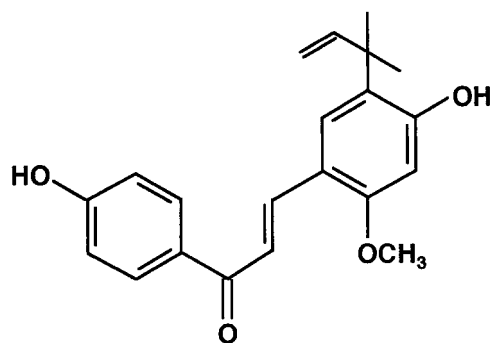


FIG. 1. Structure of licochalcone A.

be very high (19). We chose to investigate licochalcone A simply because it was the most abundant component of the naturally occurring mixture.

Recently, from among the five retrochalcones, various biological activities of licochalcone A were reported, e.g., antiprotazoal (1), anti-inflammatory (19), anti-tumor promoting (19), antioxidative (7, 18), and antimicrobial (8, 18) activities, but little information has been reported for the other chalcones. Furthermore, the antimicrobial activity of licochalcone A has not been thoroughly investigated, and little is known about its activity against food contaminant bacteria, including spore-forming bacteria, such as the genera *Bacillus* and *Clostridium*, and toxin-producing bacteria, such as *Bacillus cereus* and *Staphylococcus aureus*. In this study, we purified licochalcone A from the roots of *G. inflata* and evaluated its in vitro activities against some food contaminant microorganisms and its effects in the presence of a combination of factors, such as pH, salt, protease, and heat, as food preservation conditions.

#### MATERIALS AND METHODS

**Microorganisms.** The following 17 microorganisms, obtained from the stock culture collection at the Institute for Fermentation (Osaka, Japan), were used in this study: *Escherichia coli* K-12 IFO 3301, *Pseudomonas aeruginosa* IFO 3923, *B. cereus* IFO 3514, *Bacillus coagulans* IFO 12583, *Bacillus subtilis* IFO 3007, *Bacillus stearothermophilus* IFO 12550, *Clostridium sporogenes* IFO 13950, *Enterococcus faecalis* IFO 3989, *Enterococcus faecium* IFO 3826, *Lactobacillus acidophilus* IFO 13951, *Lactobacillus plantarum* IFO 12519, *Leuconostoc mesenteroides* IFO 3832, *S. aureus* 209P IFO 12732, *Streptococcus lactis* IFO 12546, *Streptococcus mutans* IFO 13955, *Aspergillus niger* IFO 6341, and *Saccharomyces cerevisiae* IFO 0203.

**Extraction and isolation of licochalcone A.** The roots of licorice (*G. inflata*) used in this study were obtained from Kobayashi-Kei Co., Ltd. (Kobe, Japan). Licochalcone A was extracted and isolated from the crushed roots as described previously (7, 8, 11, 18). To remove water-soluble fractions from the roots of licorice, the roots were extracted with a 10-fold volume of hot water at 80°C for 3 h, and then the residue was dried at room temperature. The dried roots (1 kg) were extracted with 5 liters of ethanol to provide the extract (56 g). Silica gel (100 g, BW-200; Fuji Silysia Chemical Ltd., Kasugai, Japan) was added to the extract (56 g), and the mixture was concentrated in vacuo. Silica gel absorbed with the extract was added to a column (30 by 6.0 cm) and eluted with 2 liters of 70% (vol/vol) ethyl acetate in *n*-hexane. The concentrated eluent (30 g) in vacuo was chromatographed on a silica gel column (BW-200, 100 by 4.2 cm); elution was done with 0 to 10% (vol/vol) methanol (MeOH) in  $\text{CHCl}_3$  (a total of 4 liters) as the eluent to give five fractions, 1 (2 g), 2 (5 g), 3 (13 g), 4 (10 g), and 5 (5 g). The absorption at 380 nm was monitored by thin-layer chromatography (TLC) (silica gel, high-performance TLC plate, 0.2 mm; Merck) with a solvent system of 3% (vol/vol) MeOH in  $\text{CHCl}_3$ . Fraction 3 (13 g) was rechromatographed on a silica gel column (BW-200, 100 by 4.2 cm); gradual elution was done with 0 to 10% (vol/vol) MeOH in  $\text{CHCl}_3$  (a total of 4 liters) and then 0 to 30% (vol/vol) acetone

in *n*-hexane (a total of 4 liters). Each fraction was subjected to TLC (high-performance TLC plate, 0.2 mm; Merck) with a solvent system of 3% (vol/vol) MeOH in  $\text{CHCl}_3$ . The absorption at 380 nm was monitored; subsequent recovery, evaporation, and recrystallization from MeOH- $\text{H}_2\text{O}$  furnished pure licochalcone A (3 g).

The crystals of licochalcone A were orange needles, and the melting point was 101°C. The main absorption maxima of licochalcone A in MeOH were 254, 312, and 377 nm. Mass spectrometry of purified licochalcone A was carried out on a JMS-AM II apparatus (Joel Ltd., Tokyo, Japan) recording at 70 eV and with a source temperature of 250°C: *m/z* 338 ( $\text{M}^+$ , 17%), 323 (4%), 307 (100%), and 121 (88%). The purity of licochalcone A was examined by high-pressure liquid chromatography and mass spectrometry as described previously (11, 18). Purified licochalcone A was dissolved in the mobile-phase solvent and applied to an octyldecyl silane column (YMC-Pack ODS-A312, 5- $\mu\text{m}$  particle size, 6 by 150 mm; YMC Co. Ltd., Kyoto, Japan). Licochalcone A was eluted with 65 to 100% (vol/vol) MeOH at 1.5 ml/min and analyzed with a UV detector (SPD-6AU; Shimadzu, Kyoto, Japan) at 310 nm. Pure licochalcone A (>99%) was dissolved in dimethyl sulfoxide (DMSO) at 5% (wt/vol), and the licochalcone A solution was used as an antimicrobial agent in the experiments described below.

**Activity against *B. subtilis*.** *B. subtilis* IFO 3007 was inoculated into nutrient broth (NB) (pH 7.0; Difco) in test tubes, grown with shaking for 24 h at 30°C, and then subcultured as vegetative cultures in fresh NB with shaking for 18 h at 30°C to approximately  $10^9$  cells/ml (2, 22). To determine the activity of licochalcone A against vegetative cells of *B. subtilis*, aliquots of 10  $\mu\text{l}$  of the cultures were inoculated into 100 ml of fresh NB containing licochalcone A at 0, 1.0, 2.0, 3.0, 4.0, or 5.0  $\mu\text{g/ml}$  in 200-ml Erlenmeyer flasks (12). The 5% (wt/vol) licochalcone A solution in DMSO was diluted in medium and added to the flasks. DMSO controls were included, although no adverse effects were observed at the highest concentration used. Culturing was continued at 30°C, and at each incubation time point, aliquots of the broth containing licochalcone A were inoculated into 10 ml of fresh NB agar on plates to determine viable cell counts in the broth. The plates were incubated for 48 h at 30°C, and CFU were counted as described previously (12). All analyses were performed with triplicate cultures, and values are given as means of three measurements.

To prepare spores of *B. subtilis* (2, 22), the vegetative cultures were spread onto NB agar supplemented with 50  $\mu\text{g}$  of  $\text{MnSO}_4/\text{ml}$  (21); the plates were incubated at 30°C for 7 days, until at least 95% of the population consisted of phase-bright spores under a phase-contrast light microscope. For harvesting of spores, the plates were washed with cold, sterile distilled water (4°C) to detach the spores, followed by centrifugation at  $4,000 \times g$  for 20 min at 4°C (2). Spore suspensions were cleaned of vegetative cells and debris by discarding the uppermost layers of the pellets obtained by centrifugation (22). The clean spore suspensions were stored at -20°C. A spore suspension (1 ml;  $10^9$  to  $10^{10}$  cells/ml) was diluted in 3 ml of 0.1 M phosphate buffer ( $\text{Na}_2\text{HPO}_4\text{-KH}_2\text{PO}_4$ ; pH 7.0) and activated by heating in a water bath at 70°C for 30 min (2, 22). Aliquots of the heat-activated spore suspension were transferred to fresh buffer in a final volume of 3 ml to adjust the turbidity at 610 nm to 1.0 (22). After both 10 mM L-alanine and 10 mM D-glucose, inducers of germination, were added to the buffer (2, 22), the germination of spores was determined by measuring the turbidity at 610 nm. To examine the effect of licochalcone A on the germination of *B. subtilis*, 10  $\mu\text{g}$  of licochalcone A/ml was added to the spores with both L-alanine and D-glucose. All analyses were performed with triplicate cultures, and values are given as means of three measurements.

**Determination of MICs against food microorganisms.** The media used for food contaminant microorganisms in this study were as follows: NB (pH 7.0; Difco) for *E. coli* K-12 IFO 3301, *P. aeruginosa* IFO 3923, *B. cereus* IFO 3514, *B. coagulans* IFO 12583, *B. subtilis* IFO 3007, *B. stearothermophilus* IFO 12550, and *S. aureus* 209P IFO 12732; brain heart infusion broth (pH 6.0; Difco) for *E. faecalis* IFO 3989, *E. faecium* IFO 3826, *L. acidophilus* IFO 13951, *L. plantarum* IFO 12519, *L. mesenteroides* IFO 3832, *S. lactis* IFO 12546, and *S. mutans* IFO 13955; thioglycolate medium (pH 7.0; Wako, Osaka, Japan) for *C. sporogenes* IFO 13950; and yeast extract-peptone-dextrose (pH 5.0) for *A. niger* IFO 6341 and *S. cerevisiae* IFO 0203. All microorganisms except for *C. sporogenes* IFO 13950 were inoculated into medium in test tubes and grown to stationary phase for 24 h at 30°C to up to  $2.0 \times 10^8$  to  $1.0 \times 10^9$  CFU/ml. *C. sporogenes* IFO 13950 was grown anaerobically for 72 h at 30°C to up to  $10^8$  CFU/ml.

MICs were estimated as described by Katsura et al. (12). The culture of each food microorganism was diluted 10-fold with sterile water. Aliquots of 100  $\mu\text{l}$  of the diluted culture of each food microorganism were inoculated at  $10^5$  to  $10^6$  CFU/ml into 10 ml of each medium containing licochalcone A at 1.0, 1.5, 2.0, 2.5, 3.0, 3.5, 4.0, 4.5, 5.0, 6.0, 7.0, 8.0, 9.0, 10, 15, 20, 25, or 50  $\mu\text{g/ml}$  in test tubes, serially diluted licochalcone A, and the respective growth medium. The MICs of licochalcone A against all microorganisms except for *C. sporogenes* IFO 13950

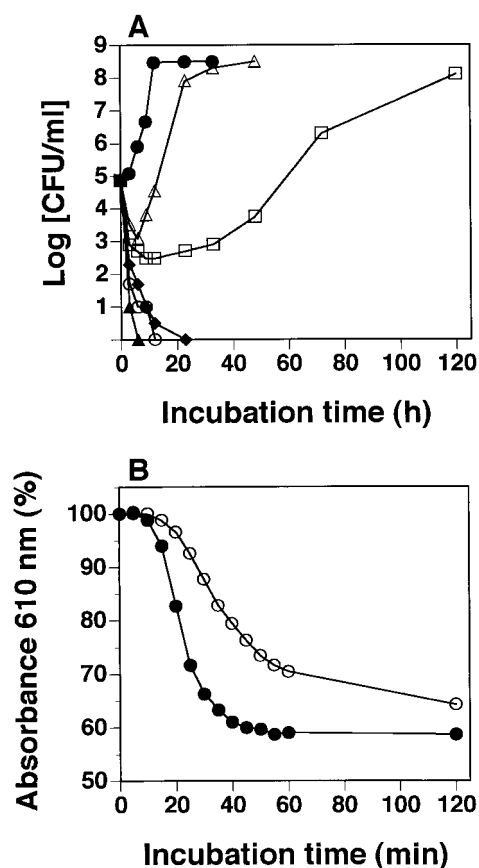


FIG. 2. Viable counts of vegetative cells (A) and germination of spores (B) of *B. subtilis* IFO 3007 in the presence and absence of licochalcone A. Symbols (licochalcone A concentration): for panel A,  $\Delta$ , 1  $\mu\text{g/ml}$ ;  $\square$ , 2  $\mu\text{g/ml}$ ;  $\blacklozenge$ , 3  $\mu\text{g/ml}$ ;  $\circ$ , 4  $\mu\text{g/ml}$ ;  $\blacktriangle$ , 5  $\mu\text{g/ml}$ ; and  $\bullet$ , none; and for panel B,  $\circ$ , 10  $\mu\text{g/ml}$ ; and  $\bullet$ , none.

were determined by measuring the turbidity at 610 nm after 48 h of incubation at 30°C. The MIC against *C. sporogenes* IFO 13950 was determined by measuring the turbidity at 610 nm after 7 days of incubation at 30°C. The MIC for each microorganism was defined as the minimum concentration of licochalcone A limiting turbidity to <0.05 absorbance unit at 610 nm (12). All analyses were performed with triplicate cultures, and values are given as means of three measurements.

**Stability of antimicrobial activity.** The MIC of licochalcone A against *B. subtilis* IFO 3007 was determined under the following conditions: temperature, 80, 100, and 121°C for 15 min; pH, 5.0 to 7.0; NaCl, 0 to 3.0% (wt/vol); and protease digestion, 20  $\mu\text{g/ml}$  (Pronase-P; Nacalai Tesque, Kyoto, Japan). To evaluate the antimicrobial activity of licochalcone A, the antimicrobial peptides protamine (salmine; Wako) and  $\epsilon$ -poly-L-lysine (Wako) were used for control experiments. All analyses were performed with triplicate cultures, and values are given as means of three measurements.

## RESULTS

**Activity against *B. subtilis*.** Figure 2A shows the inhibitory effect of licochalcone A on the vegetative growth of *B. subtilis* IFO 3007. Vegetative cell growth was inhibited in a licochalcone A concentration-dependent manner. At 2  $\mu\text{g/ml}$ , licochalcone A inhibited the growth of *B. subtilis* IFO 3007 for 12 h before cell growth started. At 3  $\mu\text{g/ml}$ , licochalcone A completely inhibited cell growth.

To examine the effects of licochalcone A on the germination

TABLE 1. Activity of licochalcone A against food contaminant microorganisms

Microorganism	MIC ( $\mu\text{g/ml}$ )
<i>Escherichia coli</i> K-12 IFO 3301	>50
<i>Pseudomonas aeruginosa</i> IFO 3923	>50
<i>Bacillus cereus</i> IFO 3514	3.0
<i>Bacillus coagulans</i> IFO 12583	2.0
<i>Bacillus subtilis</i> IFO 3007	2.0
<i>Bacillus stearothermophilus</i> IFO 12550	2.0
<i>Clostridium sporogenes</i> IFO 13950	8.0
<i>Enterococcus faecalis</i> IFO 3989	6.0
<i>Enterococcus faecium</i> IFO 3826	6.0
<i>Lactobacillus acidophilus</i> IFO 13951	5.0
<i>Lactobacillus plantarum</i> IFO 12519	5.0
<i>Leuconostoc mesenteroides</i> IFO 3832	15.0
<i>Staphylococcus aureus</i> 209P IFO 12732	3.0
<i>Streptococcus lactis</i> IFO 12546	8.0
<i>Streptococcus mutans</i> IFO 13955	5.0
<i>Aspergillus niger</i> IFO 6341	>50
<i>Saccharomyces cerevisiae</i> IFO 0203	>50

of *B. subtilis* IFO 3007, 10  $\mu\text{g}$  of licochalcone A/ml was added to heat-treated spores with both 10 mM L-alanine and 10 mM D-glucose (Fig. 2B). Heat-treated spores of *B. subtilis* IFO 3007 germinated to vegetative cells after the addition of L-alanine as an inducer of germination. Germination was slightly inhibited by licochalcone A but reached the same level as in controls without licochalcone A after 120 min of incubation. Therefore, licochalcone A did not inhibit germination, as shown in Fig. 2B, but inhibited vegetative growth after the germination of *B. subtilis* spores, as shown in Fig. 2A.

**MICs of licochalcone A against food microorganisms.** To evaluate the activity of licochalcone A against food contaminant microorganisms, the MICs were determined (Table 1). The bacteriostatic effects of licochalcone A were shown by MICs of 2 to 15  $\mu\text{g/ml}$  for all gram-positive bacteria tested, including spore-forming bacteria, such as the genera *Bacillus* and *Clostridium*, and toxin-producing bacteria, such as *B. cereus* and *S. aureus*. In particular, licochalcone A was effective for all tested *Bacillus* spp., with MICs of 2 to 3  $\mu\text{g/ml}$ . However, licochalcone A was not effective for gram-negative bacteria, such as *E. coli* and *P. aeruginosa*, or eukaryotes, such as fungi, at 50  $\mu\text{g/ml}$ .

From the roots of *G. inflata*, we also purified two chalcones as reported by Hatano et al. (9) and examined the MICs against *B. subtilis* IFO 3007; these were 40  $\mu\text{g/ml}$  for isoliquiritin and 50  $\mu\text{g/ml}$  for echinatin.

**Stability of antimicrobial activity.** Figure 3 shows the heat stability of the effect of licochalcone A against *B. subtilis* IFO 3007 after holding at 80, 100, or 121°C for 15 min at a pH of between 5.0 and 7.0, conditions representative of the practical conditions encountered in food processing. The bacteriostatic activity of licochalcone A was resistant to heating at 80 to 121°C for 15 min and stable from pH 5.0 to pH 7.0, although the antibacterial activity at an acidic pH was higher than that at a neutral or an alkaline pH.

To compare the antibacterial activity of licochalcone A in the presence of NaCl or protease with those of two cationic antimicrobial peptides, protamine and  $\epsilon$ -poly-L-lysine, the inhibitory actions of these compounds on *B. subtilis* IFO 3007 and *S. aureus* IFO 12732 were examined by using media sup-

DISCUSSION

A recent trend in food processing is to avoid the use of chemical preservatives; thus, natural antimicrobial alternatives are required. It is of interest to determine if spores occurring in foods will germinate after heat treatment and if they will grow during storage. The heat resistance of spores is influenced by the environment and varies for different *Bacillus* species (17). Generally, food manufacturers rely on preservation by moist heat to produce food products stable under ambient conditions (2). However, if heat activation of germination of the spores were to occur in this process, subsequent outgrowth might result in spoilage and/or food poisoning during preservation (2). Therefore, it is important to evaluate activity against spore-forming bacteria to determine whether antimicrobial compounds inhibit spore germination or vegetative growth. In particular, food preservatives would need to have antibacterial effects against outgrowth after pasteurization. Licochalcone A did not inhibit germination but completely inhibited the outgrowth of *B. subtilis* spores (Fig. 2). Two cationic antimicrobial peptides, protamine and  $\epsilon$ -poly-L-lysine, showed similar effects (14, 16, 20), but oleuropein purified from olive extract inhibited both the germination and the subsequent outgrowth of spores of *B. cereus* (22).

The cationic antimicrobial peptides caused the release of calcium from the spores of *Bacillus* spp. (16), and the growth inhibitory effect was repressed by the addition of calcium salts (10) and NaCl, as shown in Table 2. These findings suggested that the presence of a high concentration of salt could interfere with the binding of cationic peptides to the cell surface. Furthermore, the peptides were not resistant to protease digestion; thus, the macromolecules carried many cationic residues required for activity (16). Recently, it was reported that a series of hybrid cationic peptides were resistant to salt and had activity against *P. aeruginosa* in the presence of up to 300 mM NaCl (4). The ability to resist salt (NaCl is the most predominant salt in vivo) is important for cationic peptides to function under physiological conditions. Interactions with food components, such as salts and proteases of endogenous or microbial origin, might limit the scope of the use of cationic antimicrobial peptides as preservatives in foods. The antibacterial activ-

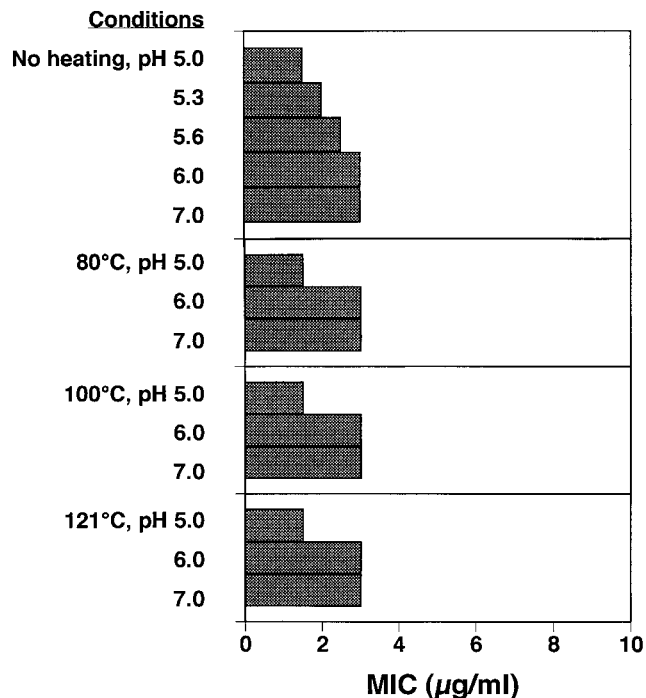


FIG. 3. Activity of licochalcone A against *B. subtilis* IFO 3007 after heating at 80, 100, or 121°C for 15 min at a pH of between 5.0 and 7.0.

plemented with NaCl or protease (Table 2). The MICs of protamine and  $\epsilon$ -poly-L-lysine for the two bacterial strains increased in both an NaCl concentration-dependent manner and an incubation time-dependent manner. After only 5 h of incubation of the two antimicrobial peptides with 3% (wt/vol) NaCl, the MICs of these peptides increased rapidly (data not shown). In contrast, the MIC of licochalcone A was stable even in the presence of 3% (wt/vol) NaCl after 7 days. Moreover, digestion of the two antimicrobial peptides with protease resulted in the partial loss of antimicrobial activities, but licochalcone A was resistant to protease digestion.

TABLE 2. MICs of licochalcone A, protamine, and  $\epsilon$ -poly-L-lysine against food contaminant bacteria in the presence of either NaCl or protease

Compound	Incubation time (days) <sup>a</sup>	MIC (µg/ml) against the following organism under the indicated conditions:									
		<i>B. subtilis</i> IFO 3007					<i>S. aureus</i> IFO 12732				
		% NaCl (wt/vol)				Protease at 20 µg/ml	% NaCl (wt/vol)				Protease at 20 µg/ml
0	1	2	3	0	1		2	3			
Licochalcone A	2	2.0	2.0	2.0	2.0	2.0	3.0	3.0	3.0	3.0	3.0
	7	2.0	2.0	2.0	2.0	2.0	3.0	3.0	3.0	3.0	3.0
Protamine	2	6.0	15.0	100	300	25.0	3.0	6.0	6.0	600	6.0
	7	6.0	25.0	300	300	25.0	3.0	15.0	300	>600	25.0
$\epsilon$ -Poly-L-lysine	2	3.0	3.0	6.0	15.0	6.0	2.0	2.0	15.0	300	2.0
	7	3.0	3.0	6.0	100	6.0	3.0	6.0	600	>600	2.0

<sup>a</sup> MICs were determined by measuring the turbidity at 610 nm after 2 or 7 days of incubation at 30°C.

ity of licochalcone A was stable even in the presence of 3% (wt/vol) NaCl, and this compound was resistant to protease digestion. Therefore, licochalcone A would be a useful compound for the development of antibacterial agents for the preservation of foods containing high concentrations of salt, such as soup, and proteases, such as fermentative foods, in which the cationic peptides might be less effective.

Some antimicrobial flavonoids from the licorice species *G. uralensis* (9), *G. glabra* (3, 5, 13, 15, 18), and *G. inflata* (8, 18) were reported previously. The content of licochalcone A in *G. inflata* was the highest (ca. 0.8%) in these three licorice species (19). We examined the MICs of chalcones purified from the roots of *G. inflata* against *B. subtilis* IFO 3007; the MICs (in micrograms per milliliter) were as follows: licochalcone A, 2.0; isoliquiritigenin, 40; and echinatin, 50. Among these compounds, licochalcone A showed the strongest antibacterial activity. Furthermore, licochalcone A also showed potent effects on both methicillin-resistant *S. aureus* and methicillin-sensitive *S. aureus*, with an MIC of 16 µg/ml (9). It was reported that licochalcone A inhibited oxygen consumption in *Micrococcus luteus* cells, and the site of inhibition was thought to be between CoQ and cytochrome *c* in the bacterial respiratory electron transport chain (8). The presence of a hydrophobic prenyl moiety in licochalcone A would be important, since it would provide sufficient hydrophobicity for molecules to penetrate the cell membrane. Recently, Chen et al. (1) reported that chalcones exhibited potent activities (antileishmanial and antimalarial) against *Leishmania major* and *Leishmania donovani* and that licochalcone A selectively inhibited fumarate reductase in the respiratory chain of the parasites.

In the present study, salt-, heat-, and protease-resistant licochalcone A was suggested to be a promising lead compound for the development of agents against spore-forming bacteria, such as those of the genera *Bacillus* and *Clostridium*. The oil extract from *G. inflata* in which licochalcone A is the main constituent has been used for the preservation of foods. Higashimaru Shoyu Co., Ltd. (Tatsuno, Japan) is the only commercial source for the extract containing licochalcone A as an antibacterial agent because it holds the patent (22a). The mechanism of the antibacterial activity of licochalcone A and its application as a food additive are currently under investigation in our laboratory.

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