# The Effect of Ergosterol on Dipalmitoylphosphatidylcholine Bilayers: A Deuterium NMR and Calorimetric Study

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ABSTRACT We have studied the effect of ergosterol, an important component of fungal plasma membranes, on the physical properties of dipalmitoylphosphatidylcholine (DPPC) multibilayers using deuterium nuclear magnetic resonance (<sup>2</sup>H NMR) and differential scanning calorimetry (DSC). For the <sup>2</sup>H NMR experiments the sn-1 chain of DPPC was perdeuterated and NMR spectra were taken as a function of temperature and ergosterol concentration. The phase diagram, constructed from the NMR spectra and the DSC thermograms, exhibits both solid-ordered (so) + liquid-ordered (lo) and liquid-disordered (ld) + lo phase coexistence regions with a clear three-phase line. This is the first demonstration that lo domains exist in liquid crystalline membranes containing ergosterol. The domain sizes in the Id+Io phase coexistence region were estimated by analyzing the exchange of labeled DPPC between the two regions, and depend on ergosterol concentration. The DPPC-ergosterol phase diagram is similar to that of the DPPC-cholesterol multibilayer system except that the so+lo and Id+lo phase coexistence regions are considerably broader.

### INTRODUCTION

Sterols are essential components of eukaryotic cells both as structural membrane components and as initiators and regulators of biological processes. In particular, cholesterol is ubiquitous in mammalian cells and ergosterol is the major sterol in certain fungi and protozoans, where it fulfills several important functions. In the unicellular eukaryote, Saccharomyces cerevisiae, Rodriguez et al. (1985) have identified four of these functions, which include growth (sparking) and bulk requirements, the latter related to maintaining membrane integrity under physiological conditions. Recently Kuebler et al. (1996) found evidence for the presence of Triton X-100 insoluble fractions in S. cerevisiae plasma membrane lipids, which implied the existence of ''rafts'' in the plasma membrane. Bagnat et al. (2000) showed that the major components of rafts in the plasma membrane of S. cerevisiae are phosphoinositol-based sphingolipids and ergosterol. The same authors found that, in contrast to mammalian rafts, the assembly of rafts in *S. cerevisiae* starts in the endoplasmic reticulum (ER) and that glycophosphatidylinositol anchored proteins associate with rafts in the ER. Bagnat et al. (2001) found that the major yeast plasma membrane protein, plasma membrane ATPase (Pma1p), needs to be associated with rafts for delivery to the plasma membrane. In vitro studies by Xu et al. (2001) using ternary

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lipid mixtures containing ergosterol support the presence of rafts in yeast.

The lipid composition of the *S. cerevisiae* plasma membrane has been studied extensively (Longley et al., 1968; Bottema et al., 1985; Patton and Lester, 1991, Zinser et al., 1991, 1993; Tuller et al., 1999). Until recently, however, most investigations reported on the fatty acid and polar head moieties of the lipid molecules and the detailed composition of individual lipid molecules was not known. Schneiter et al. (1999) identified the specific lipid molecules of both the plasma membrane and the membranes of the organelles of the X2180-1F wild-type strain of S. cerevisiae using electrospray ionization tandem mass spectrometry. The results for the plasma membrane are as follows. Phosphoinositol-based sphingolipids form 30% of the total phospholipid content of the plasma membrane, the other 70% being composed of glycerophospholipids (GPLs) in agreement with the results of Patton and Lester (1991). Interestingly, the amide-linked chain of the sphingolipids is highly saturated and unusually long (C26:0) and free ceramide with the same chain length is also present in abundance. (It is not known, however, whether the ceramide is present as a precursor of the sphingolipids or whether it has a specific function in the plasma membrane.) In this context, the results of Eisenkolb et al. (2002) indicate that the (26:0) chains of the sphingolipids are vital for the establishment of raft association by Pma1p and the stability of this protein at the cell surface of yeast. These authors also established that the presence of ergosterol rather than cholesterol is a structural requirement for this association. The S. cerevisiae plasma membrane GPLs are chiefly composed of a saturated acyl chain (e.g., C16:0 or C18:0) and a monounsaturated acyl chain (e.g., C16:1). In contrast to the

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case of higher eukaryotic cells neither glycolipids nor GPLs with polyunsaturated acyl chains were found.

The range of polar heads of the GPLs had already been analyzed by Zinser et al. (1991, 1993) for the same strain of S. cerevisiae in the same growth medium. The major GPL species were found to be phosphatidylcholine (PC), phosphatidylinositol, phosphatidylserine, and phosphatidylethanolamine. The concentration of ergosterol in the S. cerevisiae plasma membrane has been somewhat controversial. For example, Patton and Lester (1991) reported a 1:1 molar ratio of ergosterol/phospholipid whereas Zinser at al. (1991, 1993) found an ergosterol/phospholipid ratio of 3.5:1. More recently, Schneiter et al. (1999) found an ergosterol/ phospholipid ratio of 0.46:1, which is, as they point out, comparable to the cholesterol/phospholipid ratio of 0.35: 0.53 obtained for higher eukaryotic cells.

Here we study the properties and phase behavior of multilamellar dispersions of dipalmitoylphosphatidylcholine (DPPC) and ergosterol. Saturated PCs are known to have very similar physical chemical behavior to that of saturatedchain sphingomyelins (Ohvo-Rekila¨ et al., 2002). Furthermore, sphingolipids with deuterated acyl chains are not available commercially at present. Therefore, because the presence of rafts in yeast membranes is conjectured to result from the interaction between ergosterol and lipid molecules with high melting temperatures, DPPC-ergosterol is a good model in which to study phase behavior in a comprehensive way. In this context, it is usually assumed that rafts are in a tightly packed liquid-ordered (lo) phase where the sterols order the chains of the high melting lipids but the rafts retain their lateral fluidity. It is of interest to note that Xu et al. (2001) have found that ergosterol promotes the formation of raft-like domains more strongly than cholesterol. Indeed Urbina et al. (1995) have shown that 30 mol% ergosterol orders the acyl chains of dimyristoylphosphatidylcholine (DMPC) more strongly than cholesterol. Endress et al. (2002) compare the effect of cholesterol and ergosterol on the mechanical properties of DPPC bilayers. These authors find that the area compressibility modulus of DPPC bilayers containing 40 mol% ergosterol at  $10^{\circ}$ C is a factor of 1.5 higher than in the case of cholesterol. This fits well with the <sup>2</sup>H NMR results of Urbina et al. (1995) for DMPC-sterol multibilayers.

The experimental methods used here to study DPPCergosterol membranes are <sup>2</sup>H NMR spectroscopy and DSC. Aqueous multilamellar dispersions (MLDs) of ergosterol and sn-1 chain perdeuterated DPPC were used to obtain the chain order parameter as a function of temperature, T, and ergosterol mol fraction, x. Analysis based on and extended from that of Vist and Davis (1990) was applied to both the <sup>2</sup>H NMR spectra and the DSC thermograms to determine the  $T-x$  phase diagram. The phases making up this phase diagram are the same as those found by Vist and Davis (1990) and the notation due to Ipsen et al. (1987) is used to characterize them in this study. The gel phase of pure phospholipid bilayers is referred to as the solid-ordered (so) phase, the liquid crystalline phase of pure lipid bilayers is referred to as the liquid-disordered  $(\mathbf{Id})$  phase, and the  $\beta$ -phase of Vist and Davis (1990), which is found at higher cholesterol concentrations, is referred to as the liquid-ordered (lo) phase. The terms solid and liquid are used to characterize the nature of the phase whereas the words ordered and disordered indicate the conformational nature of the lipid acyl chains.

## MATERIALS AND METHODS

DPPC-d31 and DPPC were obtained from Avanti Polar Lipids (Alabaster, AL). Ergosterol and cholesterol were obtained from Sigma-Aldrich Canada (Oakville, ON). Deuterium-depleted water was from Sigma-Aldrich Canada.

## <sup>2</sup>H NMR spectroscopy

DPPC-d31/ergosterol multilamellar dispersions were prepared for ergosterol concentrations of 0, 5, 10, 13, 16, 20, 25, 27.5, 30, 35, and 42 mol%. DPPC-d31 and ergosterol were mixed in the appropriate quantities, dissolved in benzene/methanol, 4:1 (v/v), and then freeze-dried. Samples were hydrated using a pH 7.4 buffer prepared in deuterium-depleted water containing 50 mM HEPES, 150 mM NaCl, 4 mM EDTA. Hydration was performed by freeze-thaw-vortex cycling five times between liquid nitrogen temperature and 50°C. <sup>2</sup>H NMR experiments were performed on a locally built spectrometer at 46.8 MHz using the quadrupolar echo technique (Davis et al., 1976). The typical spectrum resulted from 10,000–15,000 repetitions of the two-pulse sequence with  $90^{\circ}$  pulse lengths of 3.95  $\mu$ s, interpulse spacing of 40  $\mu$ s, and dwell time 2  $\mu$ s. The delay between acquisitions was 300 ms and data were collected in quadrature with Cyclops eight-cycle phase cycling. The spectra were depaked using the procedure described by Lafleur et al. (1989). The spin-spin relaxation time,  $T_{2e}$ , was measured by varying the interpulse spacing from 40 to 100  $\mu$ s and taking the initial slope of the echo peak signal versus echo time. The sample was heated from 15 to  $60^{\circ}$ C. At each temperature, the sample was allowed to equilibrate for 20 min before a measurement. The first moment,  $M_1$ , was calculated using

$$
M_1=\frac{1}{A}\sum_{\omega=-x}^x|\omega|f(\omega),
$$

where  $\omega$  is the frequency shift from the central (Larmor) frequency,  $f(\omega)$  is the spectral intensity, and  $A = \sum_{\omega=-x}^{x} f(\omega)$ .

The boundaries of the  $so+lo$  coexistence region are determined by the spectral subtraction method (Vist and Davis, 1990; Thewalt and Bloom, 1992). Within the two-phase region, the <sup>2</sup>H NMR spectrum S is a superposition of the weighted so and lo spectra:

$$
S(x_A, T) = f_A S_f(x_f, T) + (1 - f_A) S_s(x_s, T)
$$
 (1)

$$
S(x_{\rm B}, T) = f_{\rm B} S_{\rm f}(x_{\rm f}, T) + (1 - f_{\rm B}) S_{\rm s}(x_{\rm s}, T), \tag{2}
$$

where  $x_A$  and  $x_B$  are the ergosterol concentrations of two DPPC-d31/ ergosterol MLDs,  $x_s$  and  $x_f$  are the ergosterol concentrations at the so and lo phase boundary,  $f_A$  and  $f_B$  are the fractions of lo phospholipid in the two samples, and  $S_s$  and  $S_f$  are the end-point spectra characteristic of so- and lophase domains, respectively, at the so and lo phase boundaries. Given two area-normalized <sup>2</sup>H NMR spectra  $S(x_A, T)$  and  $S(x_B, T)$ , the so (or lo) spectrum can be obtained by subtracting a fraction  $K$  (or  $K'$ ) of one spectrum from the other.  $K$  is the ratio of **lo**-phase phospholipid fractions in the two samples,  $f_A/f_B$ , and K' is the ratio of so phase phospholipid fractions in the two samples  $(1 - f_B)/(1 - f_A)$ . Using the K and K' values determined from the spectral subtraction, the phase boundaries,  $x_s$  and  $x_f$ , can then be calculated:

$$
x_{\rm s} = \frac{(1 - x_{\rm B})x_{\rm A} - K(1 - x_{\rm A})x_{\rm B}}{(1 - x_{\rm B}) - K(1 - x_{\rm A})}
$$
(3)

$$
x_{\rm f} = \frac{(1 - x_{\rm A})x_{\rm B} - K'(1 - x_{\rm B})x_{\rm A}}{(1 - x_{\rm A}) - K'(1 - x_{\rm B})}.
$$
 (4)

The spectral subtraction method is valid as long as certain assumptions hold. The spectra of the two phases,  $S_s$  and  $S_f$ , have to be sufficiently different that one can easily distinguish and carry out the subtraction procedure. The exchange of labeled lipid between two kinds of domains must be slow on the NMR timescale so that it can be neglected. In addition, the domain must be sufficiently large, so that the signal from the lipid on the boundary of the domains is negligible. Also, this method assumes that both phases have the same relaxation time  $T_{2e}$ , which is not true in this case. We found that the lo phase has a  $T_{2e}$  one to six times larger than the so phase  $T_{2e}$  (depending on the temperature), hence the so component decays with time faster than the lo component (the signal  $\propto e^{-t/T_{2c}}$ ). Thus, at any given quadrupolar echo time  $2\tau$ , the <sup>2</sup>H NMR spectrum will contain a smaller so component S<sub>s</sub> than is representative of the sample.  $f_A$  and  $f_B$  in Eqs. 1 and 2, which should be denoted as  $f_A(t = 2\tau)$  and  $f_B(t = 2\tau)$ , do not reflect the actual fraction of fluid phospholipid in the samples due to this  $T_{2e}$  effect. Thus, the K and K' (which should be denoted as  $K(t = 2\tau)$  and  $K'(t = 2\tau)$ , respectively) determined from the spectral subtraction will not be correct, leading to a deviation of  $x_s$  and  $x_f$ from the actual values. To eliminate this  $T_{2e}$  effect, corrected  $f_A$  and  $f_B$  values (i.e.,  $f_A(t = 0)$ ) and  $f_B(t = 0)$ ) are calculated by extrapolating the height of the respective echo signal back to  $t = 0$  using the measured  $T_{2e}$  for a given temperature, and then the corrected K and K' values (i.e.,  $K(t = 0)$  and  $K'(t = 0)$  $(0)$ ) can be derived and expressed in terms of the experimentally determined  $K$ , K' (i.e.,  $K(t = 2\tau)$ ,  $K'(t = 2\tau)$ ), and  $T_{2e}$ s. Using Eqs. 3 and 4 with these corrected K and K' values, the  $T_{2e}$ -corrected values of  $x_s$  and  $x_f$  are obtained.

#### Differential scanning calorimetry

DSC measurements were performed on DPPC/ergosterol MLDs at ergosterol concentration of 0, 5, 10, 13, 20, 25, 30, and 40 mol%. Suspensions of multilamellar liposomes (5 mM) were prepared. After



weighing, the dry lipids were cosolubilized in a chloroform/methanol 2:1 mixture (or hydrated immediately in the case of pure DPPC). Chloroform/ methanol was driven off by a stream of nitrogen, and the samples were stored under low pressure for at least 72 h. The resulting dry lipid films were then dispersed in pure water. The temperature was kept at  $60^{\circ}$ C for 1 h during which the suspensions were shaken vigorously several times. The DSC scans were made using a DSC Nanocal (Calorimetry Science, Provo, UT). No baseline correction was applied to the data.

## RESULTS AND DISCUSSION

<sup>2</sup>H NMR spectra were collected from 15 to 60 $^{\circ}$ C. Fig. 1 A shows that pure DPPC-d31 MLDs undergo a gel (so) to liquid-disordered (**ld**) phase transition near  $40^{\circ}$ C ( $T_{\text{m}}$ ) as the temperature is raised. The spectra obtained above  $41^{\circ}$ C display pure ld phases. The spectrum is a superposition of Pake doublets, indicating that the membrane is in the ld phase where the acyl chain undergoes rapid, axially symmetric reorientation about the bilayer normal. The spectra obtained below 39°C are characteristic of a pure so phase membrane. Deuterons in so membranes do not undergo axially symmetric motion on the NMR timescale because the acyl chains are closely packed. The spectrum obtained at  $40^{\circ}$ C is composed of both so and ld components. The spectrum of the 95:5 DPPC-d31/ergosterol MLD at 40°C also displays both so and  $Id$  components (Fig. 1  $B$ ), but with the  $Id$ component dominating. The residual gel component can be observed by magnifying the tail of the spectrum, which extends beyond 50 kHz (see  $arrow$  in Fig. 1 B). This indicates that the so to ld transition occurs at a lower temperature in the presence of 5 mol% ergosterol.





Adding ergosterol affects DPPC-d31 membranes in both the ld and so states. Fig. 2 shows the spectrum of DPPC-d31/ ergosterol dispersions at 42°C (i.e., above  $T<sub>m</sub>$ ) as a function of ergosterol concentration. DPPC-d31 MLDs with up to 5 mol% ergosterol display pure ld phase spectra. With increasing ergosterol concentration the average spectral width increases and the individual peaks broaden. At 27.5 mol% ergosterol, the individual peaks become sharper and remain sharp at higher ergosterol concentration. These spectra are the typical pure lo phase spectra: a superposition of Pake doublets having an average spectral width (e.g.,  $107,000 \text{ s}^{-1}$  for 27.5 mol% ergosterol) nearly twice that of the **ld** phase (e.g.,  $56,000 \text{ s}^{-1}$  for pure DPPC-d31) due to the highly conformationally ordered lipid chains. For ergosterol concentrations between 10 and 25 mol%, the broadening of the individual doublets in the spectrum suggests an  $\mathbf{Id} + \mathbf{Io}$ 

domains in DPPC-d31 bilayers. Fig. 3 shows the spectrum of DPPC-d31/ergosterol as a function of ergosterol concentration at 38°C (i.e., below  $T<sub>m</sub>$ ). The 0 mol% ergosterol and 5 mol% ergosterol samples display so phase spectra. For the 10 mol% ergosterol sample, a small lo component appears in addition to the so spectrum, as indicated by the emergence of "peaks" near  $\pm 15$  and  $\pm$ 20 kHz. The coexistence of so and lo phases indicates that ergosterol induces lo domains in the DPPC-d31 bilayers. The

phase coexistence. Therefore, ergosterol induces lo-phase



FIGURE 2  ${}^{2}H$  NMR spectra of DPPC-d31/ergosterol as a function of ergosterol concentration at  $T = 42^{\circ}$ C.



FIGURE 3  ${}^{2}$ H NMR spectra of DPPC-d31/ergosterol as a function of ergosterol concentration at  $T = 38^{\circ}$ C.

proportion of lo component increases as the ergosterol concentration increases (see spectra at 13, 16, and 20 mol%). Concurrently, the proportion of so component decreases. Therefore, adding ergosterol decreases the propensity of DPPC-d31 membranes to form the so phase. As the ergosterol concentration reaches 25 mol% and higher, the DPPCd31 membranes are entirely in the lo phase and exhibit lo phase spectra.

The spectra as a function of temperature for 84:16 DPPCd31/ergosterol are shown in Fig. 4. The spectra of 26, 30, 34, and  $38^{\circ}$ C show that there is so +lo phase coexistence in the DPPC/ergosterol membrane. At  $26^{\circ}$ C, the spectrum consists mostly of the so component with a very small amount of the lo component as indicated by the emergence of Pake doublet peaks within the central part of the spectrum  $(\pm 20 \text{ kHz})$ . These Pake doublet peaks become more prominent as temperature increases and the so component is reduced. At 38°C, the lo component dominates the spectrum. At 42, 46, and  $50^{\circ}$ C, lo+ld phase coexistence is observed. This is found by examining the individual peaks of the depaked spectra. These peaks are broadened, indicating the coexistence of ld and lo components (see discussion of Fig. 7 below). At 55 and  $60^{\circ}$ C, a pure liquid crystalline phase is observed, where the individual peaks of the depaked spectra become sharp.



FIGURE  $4 \text{ }^2H$  NMR spectra as a function of temperature for 84:16 DPPCd31/ergosterol.

The first moment,  $M_1$ , measures the average spectral width. Each phase state has a distinct <sup>2</sup>H NMR spectrum and thus a distinct  $M_1$  value. A large variation of  $M_1$  with temperature can be used to pinpoint phase transition temperatures in the membrane. Fig. 5 shows  $M_1$  as a function of temperature for all DPPC-d31/ergosterol MLDs. Pure DPPC-d31 undergoes a pretransition at 31 °C from the  $L_{\beta'}$ 

phase to the  $P_{\beta'}$  phase and a main transition at 40°C from the  $P_{\beta}$  phase to the  $L_{\alpha}$  phase as the temperature is raised. Note that the  $L_{\beta}$  phase and the P<sub> $\beta$ </sub> phase are so phases and the  $L_{\alpha}$ phase is synonymous with the **ld** phase. The  $M_1(T)$  curve plunges less dramatically at  $T<sub>m</sub>$  as more ergosterol is added: below  $T_{\text{m}}$   $M_1(T)$  decreases with increasing ergosterol concentration, whereas above  $T_m$   $M_1(T)$  increases with increasing ergosterol concentration.  $M_1$  is proportional to the average order parameter in the liquid crystalline phase. Therefore, adding ergosterol increases the DPPC-d31 chain ordering above  $T<sub>m</sub>$ . The so to ld transition for DPPC-d31/ ergosterol MLDs containing 10–20 mol% ergosterol occurs at a constant temperature,  $T_m = 39.5 \pm 0.5^{\circ}$ C, implying the existence of a three-phase line in the phase diagram. For DPPC-d31/ergosterol MLDs containing ergosterol concentrations of 25 mol% and above, the so to ld phase transition is absent.

To further explore the phase behavior of DPPC-d31/ ergosterol MLDs in the liquid crystalline phase, we have plotted  $M_1$  as a function of ergosterol concentration at various temperatures above  $T_{\text{m}}$  (Fig. 6). Because  $M_1$  is proportional to the average order parameter, each curve in Fig. 6 displays the progression of the average order parameter with increasing ergosterol concentration at a given temperature. The average order parameter of MLDs containing coexisting lo and ld phases is expected to be most sensitive to increasing ergosterol concentration because a proportion of ld phase will be converted to the much more ordered lo phase upon ergosterol addition. For lo-phase MLDs the effect of added ergosterol will be more modest.



100 M,  $(x10^3 sec^{-1})$ 80 60 40  $\,0\,$  $10$ 20 30 40 ergosterol conc. (mol%)

FIGURE 5 The temperature dependence of  $M_1$  for DPPC-d31/ergosterol at various ergosterol concentrations: 0 mol% ( $\blacktriangledown$ ); 5 mol% ( $\triangle$ ); 10 mol% (n); 13 mol% (O); 16 mol% ( $\bullet$ ); 20 mol% ( $\nabla$ ); 25 mol% ( $\bullet$ ); 27.5 mol%  $(\diamondsuit)$ ; 30 mol% ( $\blacktriangle)$ ; 35 mol% ( $\square$ ); 42 mol% ( $\oplus$ ).

FIGURE 6  $M_1$  as a function of ergosterol concentration for DPPC-d31/ ergosterol at: 41°C ( $\bullet$ ); 42°C ( $\diamondsuit$ ); 43°C ( $\bullet$ ); 44°C ( $\otimes$ ); 45°C ( $\blacktriangle$ ); 47°C (O); 48<sup>o</sup>C ( $\blacktriangledown$ ); 49<sup>o</sup>C ( $\square$ ); 50<sup>o</sup>C ( $\blacklozenge$ ); 53<sup>o</sup>C ( $\triangle$ ); 57<sup>o</sup>C ( $\oplus$ ). The lines are guides to the eye.

The  $M_1$ (ergosterol) curves, except at 57°C, exhibit two distinct ergosterol concentration-dependent behaviors: below 27.5 mol% ergosterol the average order parameter increases rapidly as ergosterol concentration increases, whereas above 27.5 mol% ergosterol the average order parameter increases slowly or levels off as ergosterol concentration increases. The NMR spectra suggest that at these temperatures the 75:25 DPPC-d31/ergosterol MLD has coexisting  $\text{Id} + \text{Io}$  phases whereas the 70:30 DPPC-d31/ ergosterol MLD is in the lo phase. Therefore, the changes of slope in the  $M_1$ (ergosterol) curve suggests a  $\text{Id} + \text{Io}$ /lo boundary near 27.5 mol% ergosterol. To determine this boundary, we drew a line through the points for 20 and 25 mol% ergosterol (or 25 and 27.5 mol%) and fitted another line to the points between 27 and 42 mol% ergosterol (or 30 and 42 mol%). The  $\text{Id} + \text{lo} / \text{lo}$  boundary is then obtained from the intercept of the two lines. In principle there should be a similar change in slope at low ergosterol concentrations, reflecting crossing the  $\frac{Id}{Id} + \frac{Io}{I}$  phase boundary. However, these slope changes are less obvious than those at the  $\frac{Id+I_0}{\sqrt{2}}$ lo boundary. Thus, the  $\frac{Id}{d}$  + lo phase boundary has instead been determined by direct examination of the spectra (see discussion of Fig. 8).

The  $T = 41^{\circ}\text{C}$  curve in Fig. 6 levels off at an ergosterol concentration of  $\sim$ 27.5 mol%, implying that at 41 °C the labeled DPPC chain has reached its maximum sterolenhanced order. As noted above, the  $T = 57^{\circ}$ C curve behaves differently from the others; in particular the value of  $M_1$  increases steadily with increasing ergosterol concentration all the way to 42 mol% ergosterol, implying no crossing of the phase boundary.

Another approach to determining the  $\text{Id} + \text{Io}$  /lo boundary is by examining the temperature-dependent variation in width of individual Pake doublet peaks. This is best done using depaked spectra. Fig. 7 shows the depaked spectra of 25 mol% ergosterol from 37°C, where the membrane is in the lo phase, to  $60^{\circ}$ C, where the membrane is in the **ld** phase. There is a slow decrease in the quadrupolar splittings from  $T = 37-$ 41°C. The individual peaks in the spectrum remain sharp. Above 41°C, the individual peaks broaden significantly and the quadrupolar splittings decrease faster as a function of temperature, indicating membrane heterogeneity in the form of  $Id + Io$  phase coexistence. Lipids diffusing between  $Id$  and lo domains with a rate faster than the NMR timescale will yield spectra having broadened individual peaks. As this MLD is heated, a phase boundary reflecting the onset of  $ld+lo$  phase coexistence occurs around 41 $°C$ . The broad individual peaks persist at higher temperatures until at 53°C the individual peaks become narrow again, implying that the membrane no longer displays  $\text{Id} + \text{Io}$  phase coexistence. The individual peaks remain narrow at higher temperatures and the rate of decrease in the quadrupolar splittings as a function of temperature slows down. Thus, a second phase boundary (from  $Id+Io$  to a single liquid crystalline phase) occurs around 53°C. Similar analysis of the 72.5:27.5 DPPC-d31/



FIGURE 7 The depaked spectra of 75:25 DPPC-d31/ergosterol as a function of temperature. Because the spectra are symmetrical, we show only the high-frequency half of each.

ergosterol MLD's spectra yielded phase boundaries for the  $\text{Id} + \text{Io}$  coexistence region at 43 and 50 $\textdegree$ C.

More insight into the  $ld+lo$  phase coexistence region is gained by examining the depaked spectra as a function of ergosterol concentration at a temperature above  $T_{\text{m}}$ . An example of the variations in spectral width and sharpness of individual peaks at  $45^{\circ}$ C is given in Fig. 8. At low ergosterol concentrations  $(\leq 10 \text{ mol\%)}$  the spectral lines are sharp. They blur from 13 to 27.5 mol% ergosterol (and the spectral width also increases rapidly for this concentration range) and are sharp again at ergosterol concentrations  $\geq 30$  mol%. These ''blurry'' spectra are further analyzed below. Thus, the  $\frac{Id}{d}$  +  $\frac{Id}{d}$  phase boundary lies between 10 and 13 mol% ergosterol and the  $\text{Id} + \text{lo} / \text{lo}$  phase boundary between 27.5 and 30 mol% ergosterol.

Fig. 9 shows the DSC scans of DPPC/ergosterol membranes. Pure DPPC displays a sharp main transition (so to ld transition) near  $41.5^{\circ}$ C. At 5 mol% ergosterol, the main transition becomes broadened and shifts toward lower temperature. The DSC behavior of the 0 and 5 mol% ergosterol MLDs are consistent with the NMR data discussed in the previous paragraph. The broad transition in the  $5 \text{ mol} \%$ ergosterol membrane indicates a so and ld phase coexistence region. As ergosterol concentration increases from 10 to 20 mol%, the intensity of the main peak decreases, but the peak position seems to remain unchanged. A broad shoulder appears on the high-temperature side of the sharp peak,



FIGURE 8 The depaked spectra of DPPC-d31/ergosterol as a function of ergosterol concentration at  $T = 45^{\circ}$ C.

suggesting a two-phase region. We obtained the  $\text{Id} + \text{Io}/\text{Id}$ phase boundary by finding the temperatures where the broad peak ends (Vist and Davis, 1990). At concentrations of ergosterol at 30 mol% and above a very broad residual endotherm is observed, similar to that seen in DPPC/ cholesterol MLDs.

The partial phase diagram for DPPC-d31/ergosterol determined from the characteristic phase transition temperatures and the corresponding lipid compositions as discussed above is shown in Fig. 10. The  $\frac{Id}{Id} + \frac{Io}{I}$  boundary (*open* diamonds), determined from the spectral changes described in Fig. 8 is consistent with that obtained from the DSC data (solid diamonds). The  $\text{Id} + \text{lo} / \text{lo}$  boundary in Fig. 10 is determined from the changes of slope in the  $M_1$  versus ergosterol concentration curves (Fig. 6), as well as by examining the temperature-dependent changes in the depaked spectra of 25 and 27.5 mol% ergosterol as discussed earlier (Fig. 7). These two sets of data are quite consistent and agree with the boundary obtained by examining the changes in the ergosterol-dependent depaked spectra discussed in Fig. 8. The boundaries of the  $so+lo$  phase coexistence region (solid squares) are obtained using the well-established spectral subtraction method, described in detail in Materials and Methods. The open triangles, obtained from the onset of the transition in the  $M_1(T)$  curves MLDs having high ergosterol concentrations in Fig. 5, define the  $so+lo/lo$  boundary, consistent with the data obtained from spectral subtraction.



FIGURE 9 DSC scans of DPPC/ergosterol. Note that to compare DSC with NMR data one must shift the DSC temperature by  $-2^{\circ}$ C, given the fact that DPPC chain deuteration reduces the transition temperature by  $2^{\circ}$ C.

We have identified three two-phase regions and a threephase line in the DPPC/ergosterol partial phase diagram, as shown in Fig. 10. There is a so  $+$ ld phase coexistence region at ~40°C between 0 and ~9 mol% ergosterol. There is a so+lo phase coexistence region below 39.5 $\degree$ C between  $\sim$ 9 mol $\%$ ergosterol and  $25-32$  mol% ergosterol and a  $ld+lo$  phase coexistence region above  $39.5^{\circ}$ C between  $10-15$  mol% ergosterol and 25–28 mol% ergosterol. The upper bound of the  $\mathbf{Id} + \mathbf{Io}$  phase coexistence region lies in the neighborhood of 53°C, consistent with the discussion regarding Fig. 6. The boundary separating the  $so+lo$  and  $ld+lo$  regions is a  $so+Id+Io$  three-phase coexistence line at 39.5 $°C$ .

We compare the partial phase diagram of DPPC-d31/ ergosterol with that of DPPC-d62/cholesterol by Vist and Davis (1990). Both diagrams contain a three-phase line and three two-phase regions. (Note that Huang et al., 1993 used  $^{13}$ C-NMR and DSC data to construct a similar phase diagram for PC-cholesterol but with a different interpretation for the phase behavior in terms of microdomain formation and continuous phase transitions.) The  $so/so+lo$  boundary lies close to 9 mol% for DPPC-d31/ergosterol and around 7 mol% for DPPC-d62/cholesterol, which are consistent within error. However, the  $so+lo/lo$  boundaries are quite different. The  $so+lo/lo$  boundary of DPPC-d62/cholesterol occurs around 22.5 mol% and is nearly vertical. The boundary of the  $so+lo/lo$  two-phase region characterizes the composition of lo domains within the two-phase region. A nearly vertical boundary indicates that the composition of lo domains depends only slightly on temperature. On the



FIGURE 10 Partial phase diagram of the DPPC/ergosterol membrane. Midpoint of the transition from  $M_1(T)$  curves (Fig. 5) ( $\bullet$ ); onset or end of transition in  $M_1(T)$  curves ( $\square$ ); obtained by inspection of the depaked spectra versus ergosterol concentration (Fig. 8)  $(\diamondsuit)$ ; end of transition in DSC data (shifted by  $-2^{\circ}\text{C}$ ) ( $\bullet$ ); obtained by inspection of the depaked spectra versus temperature (Fig. 7) ( $\blacktriangle$ ); obtained from  $M_1$ (ergosterol) curves (Fig. 6) ( $\bigcirc$ ); obtained from spectral subtraction  $(\blacksquare)$ ; the onset of transition in  $M_1(T)$  curves for MLDs having ergosterol concentrations of 25, 27.5, or 30 mol%  $(\triangle)$ .

other hand, because the  $so+lo/lo$  boundary of DPPC-d31/ ergosterol slopes from 25 mol% near  $T<sub>m</sub>$  toward 32 mol% at  $27^{\circ}$ C, the composition of lo domains depends strongly on temperature. The ergosterol concentration required to form lo-phase domains in the DPPC-d31/ergosterol membrane is greater than the cholesterol concentration required to form lo-phase domains in the DPPC-d62/cholesterol membrane, i.e., more ergosterol than cholesterol is needed to obtain the lo phase. Thus, per molecule, ergosterol is less effective than cholesterol in promoting the lo phase in so-phase DPPC bilayers.

The  $\text{Id} + \text{Io}$  phase coexistence region in DPPC/ergosterol is larger than that in DPPC/cholesterol. The  $\text{Id} + \text{Io}$  boundary of DPPC/ergosterol is located around 27.5 mol%, whereas that of DPPC/cholesterol lies around 22.5 mol%, implying that more ergosterol than cholesterol is needed to obtain lophase domains in the ld-phase DPPC bilayers. Thus ergosterol is less effective than cholesterol in promoting the lo phase in the ld-phase DPPC bilayers. Our observation of spectral line broadening in the  $\text{Id} + \text{Io}$  phase coexistence region of DPPC-d31/ergosterol (Fig. 8) indicates that the diffusing labeled lipids sample a heterogeneous membrane on the <sup>2</sup>H NMR timescale ( $\tau \approx 10 \mu s$ ). Taking the translational diffusion coefficient to be  $\sim 10^{-11}$  m<sup>2</sup>/s (based on measurements in DMPC/cholesterol by Filippov et al., 2003) the root mean squared displacement of the DPPC, calculated from

$$
\left(\langle \Delta x^2 \rangle \right)^{1/2} = \left( 4D\tau \right)^{1/2},\tag{5}
$$

in this time is 20 nm. This rough estimate implies nanoscale separation between sterol-rich and sterol-poor regions, as was proposed for cell membranes by Varma and Mayor (1998). Such small length scales are consistent with the lack of observation of micron-sized phase-separated domains in binary phospholipid/cholesterol giant unilamellar vesicles (Veatch and Keller, 2003).

We now estimate the size of the domains involved in this  $ld+lo$  nanoscale phase coexistence by applying the method commonly used to understand  $A \leftrightarrow B$  chemical exchange, replacing chemical shifts by quadrupole splittings. The lipid exchange time is

$$
\tau = 1/2k_{-1},\tag{6}
$$

where  $k_{-1}$  is the rate constant characterizing lipid exchange between **lo** and **ld** domains  $(k_{-1}$  is assumed to be equal to k). This rate constant is obtained from

$$
1/T_{2,obs} = f_{A}/T_{2A} + f_{B}/T_{2B} + \left[ (f_{A}^{2} f_{B}^{2}) 4\pi^{2} (\Delta \delta_{AB})^{2} \right] / k_{-1}, (7)
$$

(Carrington and McLachlan, 1967), where  $f_A$  and  $f_B$  are fractions of spins in the lo and ld domains, respectively,  $\Delta\delta_{AB} = \Delta\delta_A - \Delta\delta_B$  is the difference of the quadrupolar splittings in the two states, and  $T_{2, \text{obs}}$ ,  $T_{2A}$ , and  $T_{2B}$  can be obtained from the linewidths  $\Delta \nu$  of individual peaks (1/T<sub>2</sub> =  $\pi \Delta \nu$ ). We chose the C15 peaks for this calculation because they are well resolved. Thus,

$$
\Delta \delta_{AB} = \Delta \nu_{Q,lo} - \Delta \nu_{Q,ld},\tag{8}
$$

where  $\Delta v_{\text{Q, lo}}$  and  $\Delta v_{\text{Q, Id}}$  are the quadrupole splittings of the C15 doublets for the pure lo (30 mol% ergosterol) and pure **ld** (10 mol% ergosterol) spectra, respectively. For  $T = 45^{\circ}$ C,  $\Delta\delta_{AB} = 26,730 - 13,920 = 12,810$  Hz. The rate constants  $T_{2, \text{obs}}$ ,  $T_{2A}$ , and  $T_{2B}$  are calculated below:

$$
1/T_{2,\text{obs}} = \pi \Delta \nu_{\text{lo+ld}} \tag{9a}
$$

$$
1/T_{2A} = \pi \Delta \nu_{\rm lo} \tag{9b}
$$

$$
1/T_{2B} = \pi \Delta \nu_{\rm ld},\tag{9c}
$$

where  $\Delta v_{\text{lo+ld}}$ ,  $\Delta v_{\text{lo}}$ , and  $\Delta v_{\text{ld}}$  are the widths at half maximum height of the individual C15 peaks in the depaked  $\text{Io}+\text{Id}$  spectra (13, 16, 20, and 25 mol% ergosterol), pure lo spectrum(30 mol% ergosterol), and pure ld spectrum (10 mol% ergosterol), respectively. Thus, for  $T = 45^{\circ}$ C

$$
\Delta \nu_{\rm lo} = 1220 \,\text{Hz} (30 \,\text{mol\% ergosterol}) \tag{10a}
$$

$$
\Delta v_{\rm ld} = 1030 \,\text{Hz} (10 \,\text{mol\% ergosterol}), \qquad (10b)
$$

and Table 1 lists the results of these calculations for  $\text{Io}+\text{Id}$ membranes.

The root mean squared distance  $(\langle \Delta x^2 \rangle)^{1/2}$  traveled by DPPC-d31 before it senses lo/ld heterogeneity is plotted as a function of ergosterol concentration in Fig. 11. It is roughly U-shaped, which is reasonable given the likelihood of fewer lo/ld interfaces being found for membrane compositions near

TABLE 1 A list of  $\Delta\nu_{\text{lo+Id}}$ ,  $f_A$ ,  $f_B$  values and calculated results  $\tau$ ,  $(\langle\Delta\pmb{x}^2\rangle)^{1/2}$ 

$Mol\%$ ergosterol	$\Delta\nu_{\rm lo+1d}$ (Hz)	ŤА (i.e., f <sub>lo</sub> )	ŤВ $(i.e., f_{1d})$	$\tau$ (s)	$(\langle \Delta x^2 \rangle)^{1/2}$ (m)
13	1380	0.09	0.91	$1.20 \times 10^{-5}$ $2.2 \times 10^{-8}$	
16	2010	0.28	0.72	$5.53 \times 10^{-6}$ $1.5 \times 10^{-8}$	
20	2250	0.52	0.48	$4.36 \times 10^{-6}$ $1.3 \times 10^{-8}$	
25	4360	0.83	0.17	$3.86 \times 10^{-5}$	$3.9 \times 10^{-8}$

the boundaries of the  $\text{Io}+\text{Id}$  coexistence region of the phase diagram. Note that if a more detailed estimate of the mean lateral diffusion coefficient had been used in the final calculation of  $(\langle \Delta x^2 \rangle)^{1/2}$  the fact that the diffusion constant in the ld phase is greater than the diffusion constant in the lo phase would result in a larger root mean squared displacement for lower ergosterol concentrations and a more symmetric U-shape. The length scale characterizing the domains is estimated to be twice  $(\langle \Delta x^2 \rangle)^{1/2}$ , i.e., 30–80 nm, depending on sterol content. We have no information on domain shape, however, and because a highly convoluted domain is possible if the energy associated with interfacial lipids is negligible, we cannot estimate the number of lipids constituting a domain. Interestingly, Veatch et al. (2004) estimated domain sizes of  $\sim 80$  nm in dioleoylphosphatidylcholine/DPPC  $(1:1) + 30$  mol% cholesterol membranes displaying liquid/liquid coexistence at 25°C.

We now compare the effect of ergosterol and cholesterol on the acyl chain order parameter of liquid phase DPPC. This was accomplished by measuring the  $CD_3$  (C16) splitting of DPPC-d31/ergosterol and DPPC-d62/cholesterol. Fig. 12 shows that below 13 mol% sterol, where the membranes are



FIGURE 11 The root mean squared distance  $(\langle \Delta x^2 \rangle)^{1/2}$  diffused by DPPC-d31 as a function of ergosterol concentration at  $T = 45^{\circ}$ C. The lines are guides to the eye.



FIGURE 12 The quadrupolar splitting of the  $CD_3$  doublet as a function of sterol concentration. DPPC-d31/ergosterol at 42°C (D); DPPC-d62/cholesterol at  $40^{\circ}$ C ( $\Box$ ). The lines are guides to the eye.

in the ld phase, there is no difference in the ordering of the two DPPC/sterol membranes. This is consistent with the simulation results of Smondyrev and Berkowitz (2001) at 11 mol % sterol. On the other hand, at sterol concentrations  $>13$ mol%, the C16 splitting of DPPC-d31/ergosterol is greater than that of DPPC-d62/cholesterol. This indicates that the chain ordering of ergosterol-containing membranes is larger than that of cholesterol-containing membranes; the ergosterol structure inhibits DPPC chain conformational freedom even more than does cholesterol. Therefore, in the  $\mathbf{Id} + \mathbf{Io}$ phase coexistence and the lo phase states, ergosterol orders DPPC chains more effectively than cholesterol. This result is consistent with the findings of Urbina et al. (1995) using 70:30 DMPC/sterol.

## CONCLUSIONS

Similar to cholesterol, adding ergosterol to DPPC increases the average lipid chain order in the liquid crystalline phase and interferes with the membrane's ability to form the so (or gel) phase below  $T_{\text{m}}$ . In the **ld** phase, ergosterol is no more effective than cholesterol in ordering DPPC chains. In regions of  $\text{Id} + \text{Io}$  phase coexistence and in the lo phase, ergosterol orders DPPC chains more effectively than cholesterol. On the other hand, ergosterol is less effective than cholesterol in promoting lo-phase domains in ld (and so) DPPC bilayers. Therefore, compared to cholesterol, ergosterol is more effective in inducing chain conformational order although less effective in promoting lateral packing order in the liquid phase. It is of interest that the sterols affect conformational and lateral packing order differently; the interaction between sterol and lipid molecules is of dual nature.

A partial phase diagram was obtained from the NMR spectra and the DSC thermograms. It exhibits both  $so+lo$  and  $ld+lo$  coexistence regions with a clear three-phase line separating them. In contrast to earlier NMR phase diagram determinations, we have been able in this study to definitively locate and characterize the  $\mathbf{Id} + \mathbf{Io}$  phase coexistence region. Our observation of  $Id+Io$  phase coexistence provides evidence that two liquid crystalline phases can coexist even in model membranes containing no proteins. Thus, rafts in cell membranes may be strongly influenced by lipid/lipid interactions. Distances between ld/lo domain interfaces of 30–80 nm are typical of DPPC/ergosterol membranes. Such dimensions are also typical of ''rafts'' in cell membranes (Kusumi et al., 2004).

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