Membrane Fluidity Is a Key Modulator of Membrane Binding, Insertion, and Activity of 5-Lipoxygenase

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ABSTRACT Mammalian 5-lipoxygenase (5-LO) catalyzes conversion of arachidonic acid to leukotrienes, potent mediators of inflammation and allergy. Upon cell stimulation, 5-LO selectively binds to nuclear membranes and becomes activated, yet the mechanism of recruitment of 5-LO to nuclear membranes and the mode of 5-LO-membrane interactions are poorly understood. Here we show that membrane fluidity is an important determinant of membrane binding strength of 5-LO, penetration into the membrane hydrophobic core, and activity of the enzyme. The membrane binding strength and activity of 5-LO increase with the degree of lipid acyl chain *cis*-unsaturation and reach a plateau with 1-palmitoyl-2-arachidonolyl-*sn*-glycero-3-phosphocholine (PAPC). A fraction of tryptophans of 5-LO penetrate into the hydrocarbon region of fluid PAPC membranes, but not into solid 1,2-dipalmitoyl-*sn*-glycero-3-phosphocholine membranes. Our data lead to a novel concept of membrane binding and activation of 5-LO, suggesting that arachidonic-acid-containing lipids, which are present in nuclear membranes at higher fractions than in other cellular membranes, may facilitate preferential membrane binding and insertion of 5-LO through increased membrane fluidity and may thereby modulate the activity of the enzyme. The data presented in this article and earlier data allow construction of a model for membrane-bound 5-LO, including the angular orientation and membrane insertion of the protein.

INTRODUCTION

Lipoxygenase (LO) is a nonheme, iron-containing enzyme that catalyzes oxygenation of polyunsaturated fatty acids (Kühn, 1999; Peters-Golden and Brock, 2003). Both plant and mammalian LOs have been shown to undergo functionally important, Ca²⁺-regulated binding to membranes, followed by production of lipid-derived bioactive mediators (Rouzer and Kargman, 1988; Brock et al., 1995; Tatulian et al., 1998; Hammarberg et al., 2000; Walther et al., 2004). Mammalian 5-lipoxygenase (5-LO) is of exceptional importance because it converts arachidonic acid (AA) to 5hydroperoxyeicosatetraenoic acid (5-HPETE) and then to leukotriene A₄, a key intermediate in biosynthesis of all leukotrienes that act as potent mediators of allergy, inflammation, apoptosis, and tumorigenesis (Ghosh and Myers, 1999; Rådmark, 2000; Chen et al., 2003; Goodarzi et al., 2003; Fan et al., 2004). Leukotriene production in stimulated myeloid cells is preceded by a Ca²⁺-mediated binding of 5-LO to nuclear membranes (Rouzer and Kargman, 1988; Wong et al., 1991; Malaviya et al., 1993; Woods et al., 1993; Brock et al., 1995; Pouliot et al., 1996). A Ca²⁺-independent, phosphorylation-mediated 5-LO translocation to the nuclear membrane and activation has also been documented (Werz et al., 2000, 2002). Although the role of membrane binding in 5-LO function is well established, the mode of membrane binding of 5-LO and the reason for localization to the nuclear membrane remain poorly understood.

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Preferential binding of 5-LO to nuclear membranes might be due to specific interactions between 5-LO and intrinsic proteins of nuclear membranes, such as 5-lipoxygenase activating protein (FLAP), an integral protein in nuclear membranes that significantly increases 5-LO activity (Reid et al., 1990; Abramovitz et al., 1993; Pouliot et al., 1996). However, 5-LO binding to nuclear membranes turned out to be FLAP-independent (Woods et al., 1995; Peters-Golden and Brock, 2003). In addition, artificial lipid membranes or cell plasma membranes were able to activate 5-LO in the absence of FLAP (Puustinen et al., 1988; Noguchi et al., 1994; Skorey and Gresser, 1998; Reddy et al., 2000; Pande et al., 2004), implying that FLAP is not a molecular determinant for nuclear membrane localization of 5-LO.

Among a large number of lipids examined, phosphatidylcholine (PC) turned out to be the most efficient naturally occurring lipid in terms of supporting 5-LO activity (Puustinen et al., 1988; Noguchi et al., 1994; Pande et al., 2004). The increased affinity of the N-terminal putative β -barrel domain of 5-LO for PC membranes compared to membranes containing phosphatidylglycerol (PG) or phosphatidylserine (PS) led Kulkarni et al. (2002) to conclude that high affinity of 5-LO for PC determines its selective translocation to PC-rich nuclear membranes. However, the facts that the β -barrel domain of 5-LO binds to PC more strongly than to anionic lipids and that PC supports 5-LO activity better than the anionic lipids cannot account for nuclear membrane targeting of 5-LO for two reasons. First, comparative analysis of the lipid composition of mammalian cell membranes indicated similar fractions of PC (42-48%) in the nuclear and other organellar membranes, including plasma membranes (Surette and Chilton, 1998; D'Antuono et al., 2000). Second, nuclear membranes contain large

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fractions of anionic lipids, including PS, PG, cardiolipin, phosphatidic acid, phosphatidylinositol and its mono- and bis-phosphates (Khandwala and Kasper, 1971; Neitcheva and Peeva, 1995; Surette and Chilton, 1998; D'Santos et al., 1999, 2000; D'Antuono et al., 2000). In fact, in human leukemia cells the (phosphatidylinositol + PS) fraction was higher in the nuclear membrane (25%) than in the whole-cell lipid extract (18%) (Surette and Chilton, 1998).

The effects of the lipid hydrocarbon chain composition and membrane fluidity on 5-LO binding and activity have not been studied. In this work, we demonstrate a significant stimulatory effect of membrane fluidity on membrane binding and activity of recombinant human 5-LO. Experiments with PCs in which the number of cis-unsaturated bonds in the sn-2 acyl chain increases from zero to six indicate that both membrane binding strength and 5-LO activity reach a maximum level with membranes composed of 1-palmitoyl-2-arachidonoyl-sn-glycero-3-phosphocholine (PAPC). Moreover, membrane depth-dependent tryptophan (Trp) fluorescence quenching by brominated lipids shows that side chains of certain tryptophans of 5-LO significantly penetrate into the interior of fluid PAPC membranes, whereas no penetration occurs into solid 1,2-dipalmitoyl-snglycero-3-phosphocholine (DPPC) membranes. Our results lead to a novel concept of membrane-mediated activation of 5-LO, implying that membrane fluidity supports 5-LO activity by facilitating insertion of the enzyme into the hydrocarbon region of the membrane. Considering that the nuclear membranes of various mammalian cells, including leukocytes, are enriched in AA-containing lipids more than other subcellular membranes (Neufeld et al., 1985; Capriotti et al., 1988; Albi et al., 1997; Surette and Chilton, 1998; D'Antuono et al., 2000), which results in increased fluidity of nuclear membranes (Yu et al., 1996; Albi et al., 1997; D'Antuono et al., 2000), our findings suggest that membrane fluidity may play a central role in nuclear membrane targeting and activation of 5-LO.

MATERIALS AND METHODS

Materials

Arachidonic acid was supplied by Cayman Chemical (Ann Arbor, MI). All lipids were synthetic, and cholesterol was purified from egg yolk; they all were purchased from Avanti Polar Lipids (Alabaster, AL). Most of the chemicals were purchased from Sigma (St. Louis, MO), unless specified otherwise, and the sources of other supplies are indicated below. The 5-LO plasmid, pT3-5LO, was kindly provided by Prof. Ying Yi Zhang (Boston University School of Medicine), and has been described by Zhang et al. (1992).

Expression and purification of 5-LO

Subcloning of the 5-LO gene into the expression vector pET-21a(+), as well as protein expression in *E. coli* BL21(DE3) cells and purification, were conducted as described previously (Pande et al., 2004). The protocol allows production of highly pure and active recombinant 5-LO that contains a 1:1

stoichiometric amount of iron cofactor. The pure protein eluted from the final size-exclusion Superdex-75 column was pooled, placed into a buffer containing 150 mM NaCl, 0.1 mM EGTA, 50 mM Tris-HCl (pH 7.5) using a G-25 desalting column, and used immediately for activity experiments. Alternatively, the protein was placed in pure deionized water by the same method, lyophilized, and stored at -80° C for biophysical experiments, which were carried out during the next 2 days. The activity of the lyophilized 5-LO was sustained at the level of $\sim 80\%$ over a period of several days at -80° C.

5-LO activity assay

5-LO activity was measured in an assay buffer containing 31.64 mM Na₂HPO₄, 5.4 mM KH₂PO₄, 0.2 mM ATP, 0.1 mM dithiothreitol, 0.1 mM EGTA, and 0.3 mM CaCl₂ (pH 7.5), in the presence of large (100 nm in diameter) unilamellar vesicles, at 22°C, essentially as described previously (Pande et al., 2004). Vesicles were prepared using a Liposofast extruder (Avestin, Ottawa, Canada), as described (Pande et al., 2004). 5-LO-catalyzed conversion of AA to 5-HPETE was monitored by recording time dependence of absorption at 238 nm, using a Cary 100 double-beam spectrophotometer (Varian, Palo Alto, CA). The specific activity of 5-LO was measured using an extinction coefficient of $\varepsilon_{238} = 23 \text{ mM}^{-1} \text{ cm}^{-1}$ (Percival, 1991; Skorey and Gresser, 1998). Protein concentration was measured by Bradford assay (Bradford, 1976).

Measurements of membrane fluidity

Membrane fluidity was evaluated by generalized polarization (GP) of 6lauroyl-2-(N,N-dimethylamino)-naphthalene (Laurdan) incorporated in vesicle membranes at 1 mol %. Laurdan can be excited at 350-360 nm, and its emission spectra shift with increasing membrane fluidity, resulting in a decrease in GP = $(F_{435} - F_{500})/(F_{435} + F_{500})$, where F_{435} and F_{500} are the fluorescence emission intensities at respective wavelengths (Parasassi et al., 1994; Harris et al., 2002; Nyholm et al., 2003). Steady-state fluorescence spectra were recorded on a Jasco-810 spectrofluoropolarimeter (Jasco, Tokyo, Japan). This is a spectropolarimeter with an additional photomultiplier tube mounted at 90° with respect to the incident light beam for fluorescence measurements. Fluorescence measurements were conducted using a 0.4-cm path-length quartz cuvette at 22°C, maintained by a Peltier temperature controller. The excitation and emission slits were 4 nm and 10 nm, respectively. The emission spectra of samples containing vesicles in a buffer of 0.1 mM EGTA, 0.3 mM CaCl₂, 50 mM Tris-HCl (pH 7.5) were measured between 390 and 560 nm, using excitation at 360 nm. The total lipid concentration was 0.1 mM. The emission spectra were used to calculate the Laurdan GP, which was used as a measure of membrane fluidity.

Membrane binding measurements by fluorescence spectroscopy

Binding of 5-LO to vesicle membranes was measured by resonance energy transfer (RET) from Trp residues of 5-LO to 1,2-dioleoyl-*sn*-glycero-3-phosphoethanolamine-*N*-(1-pyrenesulfonyl) (Py-PE), which was incorporated in vesicle membranes at 2 mol %. Large unilamellar vesicles were titrated into the protein solution, with continuous stirring, to yield a total lipid concentration from 4 to 760 μ M. The concentration of 5-LO was 0.2 μ M. After each addition of vesicles, the sample was equilibrated for 2 min and fluorescence emission spectra were recorded using a J-810 spectrofluoropolarimeter, described above. Temperature was maintained at 22°C. The excitation wavelength was 290 nm. Parallel control experiments were conducted in which vesicles without Py-PE were used to titrate protein solutions. In these control experiments, addition of lipid vesicles to the protein solution resulted in a decrease in the measured Trp fluorescence intensity due to sample dilution. In RET experiments with Py-PE-labeled

vesicles, the Trp emission intensity significantly decreased upon addition of vesicles due to protein-membrane interactions and energy transfer from Trp to Py-PE. Analysis of RET data and determination of membrane binding parameters of 5-LO were conducted as described (Qin et al., 2004). Briefly, changes in Trp emission intensity at 330 nm (ΔF) were measured at each lipid concentration, corrected for the changes in the fluorescence emission as measured in control experiments, and plotted against $\Delta F/[L]$, where [L] is the lipid concentration. Each of these Scatchard plots was extrapolated with a straight line, which was used to determine the saturating value of ΔF at high lipid concentrations (ΔF_{max}) and the lipid concentration corresponding to $\Delta F = \frac{1}{2}\Delta F_{\text{max}}$ ([L]_{1/2}), respectively from the ΔF -axis intercept and the slope of the plot. The experimental binding data were presented as the dependence of $\Delta F / \Delta F_{\text{max}}$ on [L]. We have shown previously (Qin et al., 2004) that a simple binding model, suggesting that the membrane surface contains a finite number of binding sites that can be either free or occupied by a protein molecule, leads to the following binding isotherm:

$$\Delta F_{\rm rel} = b - \sqrt{b^2 - \frac{\delta[L]}{N[P]}},\tag{1}$$

where $b \equiv \frac{1}{2} \left(1 + \frac{\delta[L]}{N[P]} + \frac{K_{\rm D}}{[P]} \right).$

In Eq. 1, $\Delta F_{\rm rel} \equiv \Delta F / \Delta F_{\rm max}$, $K_{\rm D}$ is the dissociation constant, N is the number of lipid molecules corresponding to a protein binding site, [P] is the total protein concentration, and δ is the fraction of protein-accessible lipid. It has also been shown in the same source that

$$K_{\rm D} = \frac{\delta}{N} [L]_{1/2} - \frac{[P]}{2}.$$
 (2)

In Eq. 2, the total protein concentration, [P], is known, $[L]_{1/2}$ can be determined experimentally, as described above, and the fraction of proteinaccessible lipid in the external leaflet of 100-nm (in diameter) vesicles with membrane thickness of ~4 nm (Vogel et al., 2000) is $\delta \approx 0.52$. Insertion of the expression for K_D from Eq. 2 into Eq. 1 yields an equation with only one unknown, i.e., *N*. The parameter *N* was varied within physically reasonable limits and theoretical binding isotherms were simulated through Eq. 1 until a best fit was achieved between the experimental and theoretical isotherms. This was followed by calculation of K_D through Eq. 2, using the best-fit value of *N*.

Tryptophan fluorescence quenching by brominated or nitroxide-labeled lipids

The degree of insertion of tryptophans into membranes was determined by measuring the quenching of Trp fluorescence of 5-LO by 1-palmitoyl-2stearoyl(dibromo)-sn-glycero-3-phosphocholines (Br2PCs) brominated at 6,7, or 9,10, or 11,12 positions of the acyl chains, or by 1-palmitoyl-2oleoyl-sn-glycero-3-phosphotempocholine (POPTC) nitroxide spin-labeled at the polar headgroup. First, the spectra of free 5-LO (0.24 μ M in a buffer containing 0.1 mM EGTA, 0.3 mM CaCl₂, and 50 mM Tris-HCl, pH 7.5) were measured between 300 and 400 nm, using a 0.4-cm optical path length quartz cuvette and an excitation wavelength of 290 nm. Then, large unilamellar vesicles were combined with 5-LO solutions to achieve the same 5-LO concentration and a final lipid concentration of 750 μ M, and fluorescence emission spectra were measured. Experiments were conducted using a fluid lipid, PAPC, and a solid lipid, DPPC. In each case, vesicles of five distinct lipid compositions were prepared, i.e., plain lipid without any quencher, vesicles containing Br₂PC brominated at 6,7, or 9,10, or 11,12 positions, and vesicles containing POPTC. Quenching efficiencies were calculated as F_0/F , where F_0 and F are peak fluorescence intensities in the absence and presence of the quencher, respectively. Brominated lipids were present at 25 mol %, and POPTC at 15 mol %. Spectra measured in the presence of 5-LO were corrected by subtracting the spectra measured under identical conditions but without 5-LO.

RESULTS

Dependence of 5-LO activity on membrane fluidity

Recently we have analyzed the effect of lipid polar headgroups on 5-LO-membrane interactions and 5-LO activity (Pande et al., 2004). The results suggested that the effects of different lipids on the membrane binding and activity of 5-LO might be exerted through modulation of both membrane surface properties and overall membrane structure, including lipid packing order. In this current work, we have tested the hypothesis that membrane fluidity and increased fractions of AA-containing lipids in membranes may promote 5-LO binding and activity. The activity of 5-LO was measured in the presence of PCs containing a palmitoyl residue at the sn-1 position and fatty acid residues with zero to six cisunsaturated bonds (n_{Δ}) at the sn-2 position, i.e., DPPC, 1palmitoyl-2-oleoyl-sn-glycero-3-phosphocholine (POPC), 1-palmitoyl-2-linoleoyl-sn-glycero-3-phosphocholine (PLPC), PAPC, and 1-palmitoyl-2-docosahexaenoyl-sn-glycero-3phosphocholine (PDPC). Membranes containing lipids with unsaturated hydrocarbon chains had a significant stimulatory effect on 5-LO activity. The activity of 5-LO, as measured using the initial slope of the time dependence of absorbance at 238 nm, due to 5-HPETE production (Percival, 1991; Skorey and Gresser, 1998), increased ~ 2.5 times just by incorporation of a single *cis*-unsaturated bond in the *sn*-2 acyl chain of the lipid, i.e., by replacing DPPC with POPC (Fig. 1 A). Phosphatidylcholines with 2, 4, and 6 cis-double bonds in their sn-2 acyl chains, i.e., PLPC, PAPC, and PDPC, resulted in further increase in 5-LO activity, indicating a correlation between the degree of lipid acyl chain unsaturation and 5-LO activity. It is known that cis-unsaturation of lipid acyl chains decreases lipid packing order in membranes and increases membrane fluidity (Stubbs et al., 1981; Keough et al., 1987; Cevc, 1991), and cholesterol is able to partially reverse this effect (de Almeida et al., 2003). Incorporation of 20 mol % cholesterol into POPC membranes significantly reduced 5-LO activity compared to that measured in the presence of pure POPC vesicles (Fig. 1 A), suggesting that the effect of lipid acyl chain cisunsaturation on 5-LO activity is probably exerted through membrane fluidity. The initial slopes of kinetic curves of 5-LO activity were used to calculate the amount of 5-HPETE production per mg of 5-LO per minute (Fig. 1 B). Data of Fig. 1, A and B, imply that 5-LO activity is significantly promoted in the presence of vesicles composed of lipids with an increasing degree of unsaturation.

It has been shown previously that membranes composed of anionic lipids failed to support 5-LO activity as efficiently as PC (Puustinen et al., 1988; Noguchi et al., 1994; Pande et al., 2004). Here we show that the effect of the anionic charge of membrane lipids on 5-LO activity depends on the degree of lipid acyl chain unsaturation. Data of Fig. 2 demonstrate that an increase in the fraction of anionic 1-palmitoyl-2-oleoyl-*sn*-glycero-3-phosphoglycerol (POPG)



FIGURE 1 Activity of 5-LO increases with an increasing degree of lipid cis-unsaturation and decreases in the presence of cholesterol in vesicle membranes. (A) Time dependence of conversion of AA to 5-HPETE, as measured by absorption at 238 nm. The buffer (specified in Materials and Methods) contains 100 µM AA and large unilamellar vesicles composed of 350 μ M DPPC (1), POPC + 20 mol % cholesterol (2), POPC (3), PLPC (4), PAPC (5), and PDPC (6). The reaction is initiated by adding $2.4 \,\mu$ g/ml 5-LO, as indicated by the arrow. Measurements are conducted at 22° C. (B) A bar graph showing the mean values and standard deviations of 5-LO activity in the presence of vesicles of different lipid compositions, as indicated. Experimental conditions are as in A. Data shown in B are averaged based on three independent experiments.

in zwitterionic POPC membranes from 0 to 1 causes less than a twofold impairment in 5-LO activity, whereas similar fractions of 1,2-dipalmitoyl-*sn*-glycero-3-phosphoglycerol (DPPG) in DPPC membranes exerts a fourfold inhibition of 5-LO. In effect, 5-LO activity in the presence of fully saturated DPPG is ~8-fold lower than in the presence of monounsaturated POPG. These data indicate that lipid acyl chain unsaturation is a key modulator of 5-LO activity both for zwitterionic and anionic lipid membranes.

To physically interpret the effects of lipids with varying degrees of *cis*-unsaturation and the presence of cholesterol in membranes on 5-LO activity, the fluidity of membranes was measured by GP of Laurdan incorporated in membranes at 1 mol %. Introduction of a single double bond in the *sn*-2 acyl chain of the lipid, i.e., replacement of DPPC with POPC,



FIGURE 2 Dependence of the specific activity of 5-LO on the content of anionic lipids in fluid (POPC/POPG) and solid (DPPC/DPPG) membranes. Total lipid concentration was 350 μ M in all cases, AA concentration was 100 μ M, and 5-LO was added to a final concentration of 2.4 μ g/ml. The buffer, method of 5-LO activity measurement, and other experimental conditions are described in Materials and Methods.

results in a sharp decrease in GP, indicating a significant increase in membrane fluidity (Fig. 3). Membrane fluidity further increases with increasing degrees of lipid acyl chain unsaturation, and reaches a plateau at four double bonds per sn-2 chain of the lipid, corresponding to PAPC. The GP value markedly increases upon introduction of 20 mol % cholesterol in POPC membranes, indicating a decrease in membrane fluidity. Data of Figs. 1–3 provide strong evidence for modulation of 5-LO activity by membrane fluidity.



FIGURE 3 Fluorescence emission spectra of 1 mol % Laurdan in vesicles composed of DPPC (1), POPC + 20 mol % cholesterol (2), POPC (3), PLPC (4), PAPC (5), and PDPC (6). The buffer contained 0.1 mM EGTA, 0.3 mM CaCl₂, and 50 mM Tris-HCl (pH 7.5). Total lipid concentration was 100 μ M in 0.4-cm path-length quartz cuvettes. The excitation wavelength was 360 nm, and the temperature was 22°C. The inset shows the dependence of GP on the lipid composition, as specified in the main figure. The values of GP were calculated as GP = (F₄₃₅ - F₅₀₀)/(F₄₃₅ + F₅₀₀), where F₄₃₅ and F₅₀₀ are the Laurdan fluorescence emission intensities at respective wavelengths.

Dependence of membrane binding strength of 5-LO on membrane fluidity

Our data indicate that 5-LO activity increases in the presence of membranes with increasing fluidity and reaches a saturating level with PAPC. To test the hypothesis that the increase in 5-LO activity results from stronger binding of the enzyme to more fluid membranes, we measured the binding of 5-LO to large unilamellar vesicles composed of DPPC, POPC, PLPC, PAPC, and PDPC. Membrane binding of 5-LO was studied using RET from tryptophans of 5-LO to 2 mol % Py-PE in the membranes. When 5-LO was titrated with phospholipid vesicles in the absence of an energy acceptor, Trp fluorescence decreased because of sample dilution (Fig. 4 A). Titration of 5-LO with Py-PE-containing vesicles resulted in a strong, lipid dose-dependent decrease in Trp fluorescence with concomitant increase in the pyrene emission, due to RET (Fig. 4 B). Since RET is based on short-range (28 Å for the Trp-pyrene pair) dipole-dipole interactions between energy donors and acceptors (Lakowicz, 1999), the observed effect indicates binding of 5-LO to vesicle membranes. Analysis of binding isotherms (Fig. 5) yielded the dissociation constants and the numbers of lipid molecules per protein binding site, as summarized in Table 1. These data indicate that membrane binding affinity of 5-LO increases with increasing degrees of lipid acyl chain cis-unsaturation, reaches the highest value for PAPC membranes $(n_{\Delta} = 4)$, and then slightly decreases upon further increase in the degree of lipid unsaturation, i.e., with PDPC membranes $(n_{\Delta} = 6)$. Altogether, our findings delineate a correlation between membrane fluidity, membrane binding strength, and activity of 5-LO.

The data of Table 1 indicate that the number of lipid molecules per 5-LO binding site, N, decreases from 184 to 114 with increasing membrane fluidity, which likely results from larger cross-sectional area of lipids (A_L) with higher degree of unsaturation (Stillwell and Wassall, 2003). Using

the values of $A_{\rm L}$ for all five lipids summarized in Table 1 (Stillwell and Wassall, 2003 and references therein), we estimated that the membrane surface area per 5-LO binding site is $N \times A_{\rm L} = 9600 \pm 1250$ Å².

Dependence of membrane insertion of 5-LO on membrane fluidity

The correlation between membrane fluidity, membrane binding affinity, and activity of 5-LO suggests that the enhancement of 5-LO activity by membrane fluidity may at least partly result from stronger binding of 5-LO to membranes of higher fluidity. However, other parameters of 5-LO-membrane interaction, such as the depth of insertion of 5-LO into the hydrophobic core of membranes, may also contribute to 5-LO function. We have estimated the depth of membrane insertion of the side chains of Trp residues of 5-LO by employing the quenching of Trp fluorescence by Br₂PCs brominated at 6,7, or 9,10, or 11,12 positions of the acyl chains, or by POPTC nitroxide spin-labeled at the polar headgroup. In the presence of fluid PAPC vesicles without quenchers, Trp fluorescence of 5-LO significantly increased (Fig. 6 A), indicating that some tryptophans of 5-LO experience less polar environment, apparently due to membrane binding of 5-LO. (Unlike the experiments presented in Fig. 4, in these experiments there was no dilution effect and the increase in Trp emission intensity in the presence of membranes could be seen directly.) Incorporation of Br₂PCs in PAPC vesicles resulted in quenching of 5-LO fluorescence by all three brominated lipids; the maximum quenching occurred with vesicles containing 9,10-Br₂PC (Fig. 6, A and C). The distances of bromine atoms in Br_2PCs brominated at 6,7, or 9,10, or 11,12 positions from the bilayer center have been estimated to be 11.0, 8.3, and 6.5 Å, respectively (McIntosh and Holloway, 1987). Therefore, maximum quenching of Trp fluorescence by 9,10-Br₂PC indicates that the side chains of certain Trp residues of 5-LO bound to



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FIGURE 4 Fluorescence emission spectra of tryptophans of 5-LO in the absence and presence of large unilamellar PAPC vesicles without (*A*) and with (*B*) 2 mol % Py-PE. In both panels, increasing darkness of the lines corresponds to an increase in total lipid concentration from zero to 760 μ M (see Fig. 5). Decrease in Trp emission intensity in *A* is due to dilution upon addition of stock vesicle suspension. In *B*, Trp fluorescence significantly decreases upon addition of Py-PE-containing vesicles due to energy transfer from Trp of 5-LO to Py-PE. The excitation wavelength was 290 nm, and buffer and temperature were as in Fig. 3.



FIGURE 5 Isotherms of 5-LO binding to vesicles composed of DPPC (\bigcirc), POPC (\blacksquare), PLPC (\blacktriangle), PAPC (\square), and PDPC (\triangle). The data points are obtained on the basis of the decrease in Trp fluorescence emission intensity due to energy transfer from Trp to Py-PE (2 mol % in vesicle membranes), as measured in RET experiments (e.g., Fig. 4). The theoretical curves are simulated through Eq. 1 using binding parameters summarized in Table 1.

PAPC membranes penetrate into the membrane hydrophobic core nearly halfway to the membrane center. When PAPC vesicles contained headgroup spin-labeled POPTC, Trp fluorescence was quenched to a larger extent than by any of the brominated lipids (Fig. 6, A and C). This result indicates that a larger fraction of tryptophans of 5-LO bound to PAPC membranes are located at the membrane-water interface compared to the fraction embedded into the lipid hydrocarbon region.

In the presence of solid DPPC vesicles containing brominated lipids, none of the Br₂PCs quenched the Trp fluorescence of 5-LO (Fig. 6, B and C). However, Trp fluorescence was quenched by POPTC even more efficiently than in the case of fluid PAPC membranes, presumably because tryptophans that embed into fluid PAPC membranes are located at the membrane-water interface in the case of solid DPPC membranes, thus increasing the total population of interfacially located tryptophans. Altogether, data of Fig. 6 suggest that 5-LO is able to partially intrude into the hydrocarbon region of fluid membranes, but in the case of solid membranes it interacts with the membrane surface without penetration into its hydrophobic core. In conjunction with our data that 5-LO has stronger membrane binding affinity and higher activity in the presence of fluid membranes, these data suggest that partial insertion of 5-LO into the membrane is likely to be important for the enzyme function.

A model for membrane-bound 5-LO

The available data on membrane binding of human 5-LO, its N-terminal putative β -barrel domain, and rabbit 15-LO allow

construction of a model of 5-LO bound to a phospholipid membrane in the fluid phase (Fig. 7). We have modeled the structure of 5-LO using the structure of rabbit reticulocyte 15-LO as a template (Protein Data Bank entry, 1lox), which shares 37% sequence identity and 57% similarity with 5-LO. Homology modeling was achieved using the internet-based SWISS-MODEL server (Schwede et al., 2003). The modeled structure of 5-LO was positioned at the membrane surface using the following criteria: 1), the symmetry axis of the N-terminal β -barrel is oriented at ~45° with respect to the membrane normal (Pande et al., 2004); 2), tryptophans 13, 75, and 102 likely interact with the membrane (Kulkarni et al., 2002); 3), Trp¹⁸¹ of rabbit 15-LO, which corresponds to Lys¹⁸³ in the aligned 5-LO sequence, is likely to interact with the membrane (Walther et al., 2002, 2004); and 4), at least one Trp penetrates into the membrane to a depth of 8–9 Å from the membrane center (this work). Positioning of the modeled 5-LO structure on the membrane surface according to these constraints indicates that Trp⁷⁵ likely penetrates into the membrane deeper than any other residue, Trp¹³ inserts into the hydrophobic part of the membrane, but not as deep as Trp⁷⁵, and Trp¹⁰² is closer to the membrane-water interface. Note that Kulkarni et al. (2002) have shown that Trp¹⁰² was more important for the interaction of the β -barrel domain of 5-LO with membranes than Trp¹³ or Trp⁷⁵. This is not necessarily in conflict with our model because tryptophans located at the polar headgroup/hydrocarbon boundary of membranes can facilitate protein-membrane interactions by means of both H-bonding and nonpolar interactions (Yau et al., 1998) and thus may contribute to the interaction more than totally embedded tryptophans that are not involved in H-bonding with lipid carbonyl oxygens. Also, the truncated β -barrel domain may not be oriented at the membrane surface as the full-length protein. Lys¹⁸³ is properly located to make ionic and/or H-bonding interaction with the phosphate group of membrane phospholipids. In Fig. 7, the residue Arg⁴¹¹ is highlighted to indicate the putative site of entrance of AA into the substrate binding cleft, because Gillmor et al. (1997) have suggested that the corresponding residue in the rabbit 15-LO sequence (Arg⁴⁰³) forms an ionic contact with the carboxyl group of arachidonate during the initial step

TABLE 1 Parameters characterizing the interaction of 5-LO with large unilamellar vesicles composed of PCs of varying fluidity

Lipid	$L_{1/2}$ (µM)	$K_{\rm D}~(\mu{\rm M})$	Ν
DPPC	443.5	1.15	184
POPC	297.8	0.78	176
PLPC	199.2	0.56	157
PAPC	83.2	0.24	127
PDPC	92.0	0.32	114

Parameters include lipid concentrations corresponding to binding of 50% of 5-LO ($L_{1/2}$), the dissociation constants (K_D), and the numbers of lipids per 5-LO binding site (N). Lipids were composed of phosphatidylcholines with varying degrees of *sn*-2 acyl chain unsaturation.



FIGURE 6 Fluorescence emission spectra of tryptophans of 5-LO in the absence and presence of large unilamellar vesicles composed of a fluid lipid, PAPC (*A*), or a solid lipid, DPPC (*B*). Only the top portions of spectra are shown to make differences between spectra more discernable. The spectra are numbered as follows: 1, free 5-LO in buffer; 2, 5-LO with large unilamellar vesicles of PAPC (*A*) or DPPC (*B*); 3, 6,7-Br₂PC-containing vesicles; 4, 9,10-Br₂PC-containing vesicles; 5, 11,12-Br₂PC-containing vesicles; and 6, POPTC-containing vesicles. Numbering in *A* also applies to the same line types in *B*. (*C*) Summary of the data obtained in three experiments (mean \pm SD). Open and solid bars apply to PAPC and DPPC membranes, respectively. When vesicles were present, the total lipid concentration was 750 μ M. Brominated lipids were present at 25 mol %, and POPTC at 15 mol %. Protein concentration was 0.24 μ M. Excitation was at 290 nm. The buffer and temperature were as in Fig. 3.

of enzyme-substrate interaction. Also shown are the ironcoordinating residues (histidines 367, 372, 550, and Ile^{673}) to identify the catalytic site of the enzyme (Zhang et al., 1992). The model indicates that most surface-exposed tryptophans of 5-LO are at the polar headgroup region of the membrane, whereas Trp⁷⁵ and Trp¹³ insert into the hydrophobic part of the membrane. This agrees with the data indicating stronger quenching of Trp fluorescence by headgroup nitroxidelabeled POPTC than by acyl chain-brominated lipids (Fig. 6). In the case of solid membranes, all tryptophans are presumably located at the membrane-water interface without membrane insertion, resulting in negligible quenching



FIGURE 7 A model for 5-LO bound to a phospholipid membrane in the fluid phase, constructed on the basis of homology modeling and data on membrane insertion and angular orientation of 5-LO. The protein structure is modeled using SWISS-MODEL on the basis of rabbit 15-LO crystal structure as a template, and is presented in a ribbon format. The N-terminal β -barrel and the catalytic domains are shown in plum and aqua, respectively. The mesh of gray spheres represents the plane of the phosphate groups of phospholipid molecules. Membrane-interacting and catalytically important residues are shown in the ball-and-stick format and are labeled by blue and red labels, respectively. Tryptophans that insert into the membrane hydrocarbon region are shown in yellow, and those located at the membrane-water interface are shown in orange. Phe¹⁹⁷ and Lys¹⁸³, which interact with the membrane by nonpolar and ionic interactions, are shown in green and purple, respectively. Arg⁴¹¹, which presumably forms an ionic contact with the carboxyl group of arachidonate during the initial step of enzyme-substrate interaction, is shown in magenta. Residues involved in coordination of the iron cofactor are colored according to the atom type. Trp¹³, and partially Trp⁵⁹⁹ and Lys¹⁸³, are underneath the polar groups of lipids and can hardly be seen. His³⁶⁷ is hidden behind the helical ribbon. According to our earlier results (Pande et al., 2004), the symmetry axis of

the N-terminal β -barrel domain is tilted at ~45° with respect to the membrane normal, and the current data suggest that at least one Trp, most likely the Trp⁷⁵, inserts into the membrane as deep as 8–9 Å from the membrane center.

by brominated lipids and stronger quenching by POPTC (Fig. 6). The more "peripheral" binding of 5-LO to solid membranes probably contributes to the decreased membrane affinity and lower activity of 5-LO (Figs. 1 and 5, Table 1).

DISCUSSION

Binding to intracellular membranes in vivo or to artificial lipid membranes in vitro is an essential step in activation of 5-LO. Despite this, little is known about the mechanism of interaction of 5-LO with membranes. No quantitative data on 5-LO-membrane interactions, such as binding constants or 5-LO-lipid binding stoichiometries, have been reported so far, and the effects of physical properties of membranes, such as membrane fluidity, on membrane binding and activity of 5-LO are poorly understood. Previously we have conducted a detailed analysis of the effects of lipid polar headgroups, as well as of Ca²⁺ and lipid concentrations, on 5-LO-membrane interactions and 5-LO activity (Pande et al., 2004). In this work, we concentrate on the effect of lipid acyl chain unsaturation and membrane fluidity on membrane binding strength, membrane insertion, and activity of 5-LO. Our results provide unprecedented evidence that membrane fluidity is an important determinant for the membrane-binding mode and activity of 5-LO.

To study the effect of lipid acyl chain unsaturation on the membrane-binding mode and activity of 5-LO, we have used PCs that contain a saturated palmitoyl chain at the sn-1 position, whereas the acyl chain at the sn-2 position was either a palmitoyl (16:0), oleoyl (18:1), linoleoyl (18:2), arachidonoyl (20:4), or docosahexaenoyl (22:6) residue (shown in parentheses are the ratios of carbon atoms to double bonds per chain). The choice of these lipids was based on the fact that in major lipids of mammalian cell membranes the sn-1 position is usually occupied by a saturated chain, such as a palmitoyl residue, and the sn-2 position is most frequently occupied by one of the unsaturated fatty acyl chains specified above (Albi et al., 1997; Surette and Chilton, 1998; D'Antuono et al., 2000). Replacement of fully saturated DPPC membranes by membranes containing lipids with *cis*-unsaturated acyl chains results in an \sim 3-fold increase in 5-LO activity. It is known that membranes composed of DPPC are in the solid gel phase at temperatures below 41°C, whereas membranes composed of POPC, PLPC, PAPC, or PDPC are in the fluid, liquid crystalline phase above 0°C (Stubbs et al., 1981; Litman et al., 1991; Stillwell and Wassall, 2003). Therefore, our data indicate that the membrane-binding mode and activity of 5-LO are controlled by membrane fluidity. Remarkably, the maximum stimulatory effect on 5-LO activity was reached with vesicles composed of PAPC. The fact that both the membrane-binding strength and activity of 5-LO correlate with Laurdan GP (Figs. 1, 3, and 5) indicates that the effect of PAPC on 5-LO function is exerted through variation of membrane fluidity rather than specific 5-LO-PAPC interactions. The activity of 5-LO in the presence of POPC membranes containing 20 mol % cholesterol, which consist of coexisting disordered and ordered phases (de Almeida et al., 2003), is between activities measured in the presence of solid (DPPC) and fluid membranes (e.g., POPC). This is consistent with the concept of modulation of 5-LO activity by membrane fluidity.

Our data indicate that the values of dissociation constants $(K_{\rm D})$ of 5-LO for phospholipid membranes decrease from 1.15 to 0.24 μ M upon an increase of the degree of lipid sn-2 acyl chain cis-unsaturation from zero to four, and undergoes little change upon further increase in n_{Δ} (Fig. 5 and Table 1). This resembles the increase in 5-LO activity as a function of membrane fluidity (Fig. 1). Although quantitative data on 5-LO binding to membranes are not available, surface plasmon resonance experiments yielded a $K_{\rm D}$ value of 0.6 nM for the binding of the putative β -barrel domain of 5-LO to POPC vesicles in the presence of 0.1 mM CaCl₂ (Kulkarni et al., 2002), which indicates much stronger binding of the β -barrel domain compared to that of the full-length 5-LO determined in this work (0.78 μ M). Differences in methods used by Kulkarni et al. (2002) and in this work might partially account for discrepancies in measured dissociation constants. On the other hand, this divergence is congruent with earlier data showing that intracellular translocation of the N-terminal β -barrel domain of 5-LO to the nuclear membrane was much faster than that of the full-length protein (Chen and Funk, 2001). Furthermore, stronger membrane binding by itself may not necessarily favor the enzyme function. It is possible that the catalytic domain, while reducing the binding strength, facilitates a defined, productive mode binding of the enzyme to the membrane surface.

The average area per 5-LO binding site at the membrane surface is estimated to be $A_{\rm bs} = 9600 \pm 1250 \text{ Å}^2$. Using the model of 5-LO (Fig. 7), we have determined that the protein molecule can be roughly presented as a prolate ellipsoid of revolution with semiaxes of 48 and 23 Å. This would correspond to a maximum cross-sectional area of $\pi \times 48 \times 23$ ≈ 3500 Å², implying that $A_{\rm bs}$ is ~ 2.7 times larger than the maximum cross-sectional area of the molecule. Because of the rotational freedom of membrane-bound protein molecules around the membrane normal, the effective area per membrane-bound 5-LO would be $\pi \times 48^2 \approx 7200 \text{ Å}^2$. On the other hand, even at the saturation of adsorption, there will be "empty" surface area between membrane-bound protein molecules. This means that upon binding of each protein molecule, an area of $(96 \text{ Å})^2 \approx 9200 \text{ Å}^2$ will become inaccessible to other protein molecules, which is in good agreement with the experimentally determined effective area per protein binding site of 9600 \pm 1250 Å².

Previous studies indicated that the N-terminal putative β -barrel domain of 5-LO had ~20 times stronger affinity (in terms of K_D) for PC membranes than for membranes containing 50 mol % of PG or PS (Kulkarni et al., 2002). This led to the conclusion that increased affinity of the

 β -barrel domain of 5-LO for PC versus anionic lipid membranes dictates the localization of 5-LO to PC-rich nuclear membranes (Kulkarni et al., 2002). However, the content of anionic lipids in nuclear membranes is comparable to or even higher than that of other cellular membranes (Khandwala and Kasper, 1971; Neitcheva and Peeva, 1995; Surette and Chilton, 1998; D'Santos et al., 1999, 2000; D'Antuono et al., 2000). Also, PC is abundant not only in nuclear but in many other membranes of eukaryotic cells (Hanahan, 1997; Surette and Chilton, 1998; Mathews et al., 1999; D'Antuono et al., 2000). For example, analysis of phospholipid content in the plasma membrane, endoplasmic reticulum, and the nuclear membrane of rat papillary cells indicated that PC was the most abundant lipid (42-48 mol %) in all three cases (D'Antuono et al., 2000). No significant difference was found in the PC content in nuclear membranes and wholecell lipid extract for human monocytes (Surette and Chilton, 1998). Therefore, preference of 5-LO for PC versus anionic lipids cannot account for selective recruitment of 5-LO to nuclear membranes.

Nuclear membranes are enriched in AA-containing lipids more than other cellular membranes (Neufeld et al., 1985; Capriotti et al., 1988; Albi et al., 1997; Surette and Chilton, 1998; D'Antuono et al., 2000). Thus, up to 48% of the total cellular AA-containing lipids were found in the nuclear membranes of various mammalian cell lines, including leukemia cells (Surette and Chilton, 1998). In rat renal papillary cells, the fraction of AA-containing lipids was 1.5 times higher, whereas the fraction of lipids with saturated chains was 1.6 times lower in nuclear than in endoplasmic reticulum membranes (D'Antuono et al., 2000). Large fractions of AA-containing lipids result in elevated fluidity of nuclear membranes (Yu et al., 1996; Albi et al., 1997; D'Antuono et al., 2000). This leads to a mechanism suggesting that increased membrane fluidity via high levels of AA-containing lipids may be an important factor for localization of 5-LO to nuclear membranes. Our data indeed provide support for this mechanism by indicating that among PCs with a palmitoyl residue in the *sn*-1 position and various acyl chains in the sn-2 position, PAPC was the most efficient lipid in terms of facilitating membrane binding, membrane insertion, and activity of 5-LO. Lipids with lower or higher degrees of acyl chain unsaturation were not as efficient as PAPC. Taking into account that cytosolic phospholipase A₂ (PLA₂), as well as FLAP, are also colocalized to nuclear membranes (Woods et al., 1993; Pouliot et al., 1996), this mechanism would lead to a congregation of all enzymes, regulators, and substrates for efficient metabolism of AAcontaining lipids to leukotrienes.

Our model of 5-LO binding to a membrane in the fluid phase (Fig. 7) shows that partial insertion of 5-LO into the hydrophobic core of the membrane may be required for membrane-proximal location of the orifice of the substratebinding pocket, marked by Arg⁴¹¹, and better access of 5-LO to membrane-residing AA. Partial membrane insertion of 5-LO may also be necessary for positioning 5-LO at the membrane surface in a productive mode, because the Ca^{2+} -mediated electrostatic mechanism alone is apparently not enough to do this.

Although details of membrane binding of 5-LO have not been described before, membrane-binding modes of several C2 domain-containing proteins or truncated C2 domains have been assessed in the past. Cytosolic PLA₂ α , which colocalizes with 5-LO at the nuclear membrane and produces AA for eicosanoid production, was shown to considerably penetrate into the hydrocarbon region of phospholipid monolayers (Lichtenbergova et al., 1998). Certain residues of the truncated C2-domain of cytosolic PLA₂ α penetrate into the hydrocarbon region of membranes as deep as 5-10 Å below the plane of the lipid phosphate groups (Ball et al., 1999; Malmberg et al., 2003). C2 domains of other membrane-binding proteins, e.g., protein kinase C_{α} and synaptotagmin I, penetrate into the polar headgroup region of membranes but not into the hydrocarbon region as deep as cytosolic PLA₂ α (Frazier et al., 2003; Kohout et al., 2003; Rufener et al., 2005). The hydrophobic residues Phe⁷⁰, Leu⁷¹, Trp¹⁸¹, and Leu¹⁹⁵ of rabbit 15-LO, which are clustered at the membrane-binding surface of 15-LO that contains the substrate-binding pocket, were found to play a major role in membrane anchoring of the protein, probably involving partial membrane insertion (Walther et al., 2002, 2004). It seems likely that hydrophobic protein-membrane interactions and partial membrane penetration constitute a common feature of C2-domain-containing proteins. This may be necessary to stabilize the membrane docking of these proteins. It should be noted that all experiments with cytosolic PLA₂ α , protein kinase C_{α}, and synaptotagmin I were conducted using fluid membranes composed of lipids containing an oleic acid moiety at the sn-2 or at both positions, and in experiments with 15-LO submitochondrial particles were used (Walther et al., 2002, 2004). Thus, it was not possible to assess the dependence of membrane insertion on membrane fluidity based on these studies. Our data, however, clearly demonstrate that membrane insertion of 5-LO only occurs with fluid, but not solid membranes, indicating an important role of membrane fluidity in modulating the function of 5-LO and probably other enzymes that undergo interfacial activation via membrane binding.

Finally, partial membrane insertion of 5-LO indicates a resemblance between the membrane binding modes of lipoxygenases and the other major class of enzymes involved in eicosanoid biosynthesis, cyclooxygenases. The latter enzymes are embedded in nuclear and perinuclear endoplasmic reticulum membranes as monotopic proteins (Spencer et al., 1998; Marvin et al., 2000; Picot and Garavito, 1994; Picot et al., 1997). Enrichment of these membranes with AAcontaining lipids probably facilitates membrane binding and insertion of both cyclooxygenases and 5-LO through increased membrane fluidity and thus plays a regulatory role in the biosynthesis of prostanoids and leukotrienes. We thank David Moe for expert technical assistance at the initial stages of this work and Dr. Jeffrey L. Urbauer for useful comments during preparation of the manuscript.

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