Faecal unconjugated bile acids in patients with colorectal cancer or polyps

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Abstract

The unconjugated faecal bile acid profiles of 14 patients with colorectal cancer, nine patients with polyps and 10 controls were compared using gas liquid chromatography, controlling for such confounding variables as cholecystectomy, gall stones and hepatic function. Patients with adenomatous polyps had a higher concentration of faecal bile acids (5.23 µmol/ g, 2·16-13·67 (median, range) v 1·96, 0·91-6.97; p=0.016) lithocholic acid (2.41, 0.88-3.22 v 1.07, 0.38-2.03; p=0.013) and total secondary bile acids (5.23, 2.16-13.4 v 1.96, 0.73-6.63; p=0.02) compared with control subjects. Patients with colorectal cancer had an increased (p=0.029) proportion of secondary faecal bile acids (mol%) compared with controls (100, 96.5-100 v 95.19, 81.73-100) and the ratios of the primary bile acids, cholic and chenodeoxycholic acid, to their respective derivatives (secondary bile acids) were significantly lower in cancer patients compared with control and patients with polyps (p=0.034 to 0.004). This study lends further support to the theory that bile acids may play a role in the development of polyps and colorectal cancer.

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Epidemiological studies have implicated dietary fat of animal origin as an important aetiological factor in the development of colorectal cancer¹² which may be related to altered faecal bile acids.³ Bile acids given either orally,⁴ or rectally,⁵⁶ have been shown to have a tumour promoting effect in rat models for colorectal cancer.

The mechanisms by which bile acids may produce these effects are uncertain. Bile acids may be mutagenic,⁷⁸ but some of the techniques used have been questioned.⁹⁻¹¹ Bile acids may directly damage DNA,¹² or chromatin.¹³ Lithocholic acid has been shown to produce DNA strand breaks¹⁴ and both chenodeoxycholic acid and lithocholic acid induce mitotic aneuploidy in yeast cells.¹⁵ Possible mechanisms which may explain an increase in colonic cell proliferation caused by bile acids, include effects on prostaglandin synthesis,¹⁶ arachidonic acid release,¹⁷ and protein kinase C activity.¹⁸

Populations at greater risk of colorectal cancer have been found to excrete more faecal bile acids,^{3 19 20} but others have not shown this.²¹ The correlation between dietary habits and faecal bile acid excretion, is also supported by differences found in the faecal bile acid profiles of certain population subgroups,^{20 22} or different socioeconomic groups.²³ Furthermore, dietary manipulation studies have shown that high fat intake can increase faecal bile acid excretion.2425

Two early studies showed an increased faecal bile acid excretion in patients with colonic cancer,^{26,27} but later studies have been unable to confirm this.²⁸⁻³⁴ Only one of four studies in polyps has shown any increase in faecal bile acid excretion in this group.^{26,31,35,36} Differences in patient selection, faecal collection, and analytical methodology may account for these apparently conflicting results.²¹

More recently Hill and Owen, unable to show their previous findings of a significant increase in faecal bile acid excretion, have proposed instead that the ratio of lithocholic acid: deoxycholic acid and lithocholic acid: deoxycholic acid×total faecal bile acid concentration may be important markers in patients with colorectal cancer.^{37 38}

The role of the colon in modifying biliary bile acid composition,³⁹ has prompted several groups⁴⁰⁻⁴³ to examine the biliary bile acid profiles of patients with colorectal cancer or adenomatous polyps with conflicting results. Again differences in analytical techniques, patient selection, and other variables makes the interpretation of some of these studies difficult.

In this study we report the results of detailed analysis of the faecal unconjugated bile acid profiles of patients with colorectal cancer or polyps in comparison with controls and in which a careful attempt at controlling for confounding factors has been made.

Methods

PATIENTS AND STUDY DESIGN

Patients with colorectal cancer, adenomatous polyps or controls were recruited from the general surgical outpatients' department of Dudley Road Hospital.

Patients with colorectal cancer (14) or adenomatous polyps (nine) entered the study after diagnosis. None of these patients were obstructed. The control group (10) comprised patients with non-specific abdominal symptoms other than constipation. Control subjects as well as patients with cancer or polyps underwent rigid sigmoidoscopy and colonoscopy or double contrast barium enema.

All groups underwent abdominal ultrasound scan to determine the presence of gall stones and exclude liver metastases in the cancer group, as well as biochemical tests to assess hepatic and renal function. Weight was measured and recorded for each patient. All were white caucasians on an *ad libitum* omnivorous diet.

Patients with abnormal liver function tests, gall stones, previous cholecystectomy, or hepatic metastases were excluded. No patient had a

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Accepted for publication 13 January 1992 coexisting medical complaint likely to affect bile acid metabolism nor had any undergone previous bowel surgery other than appendicectomy. Any patients who had taken antibiotics within the three months before entering the study, or other medication likely to affect bile acid metabolism, were also excluded.

The histological examination of carcinomas and polyps was carried out by the Pathology Department of the Dudley Road Hospital.

FAECAL COLLECTION

Complete five day faecal collections were obtained from patients, at home or place of work and taking their normal diet. This was achieved by delivering small 48 1 (Frigidaire, FV 602) freezers to the homes of the participants so that stools could be frozen immediately at -22° C. At the same time patients carried out a mouth-to-anus transit time study using radioopaque markers (Portex UK Ltd).⁴⁴ Subjects were asked to collect every stool for five days and record the time each was passed. After five days the freezers were collected and returned to the laboratory. Faecal samples were stored at -20° C before being radiographed (for transit time studies), homogenisation and freeze drying.

HOMOGENISATION

Each five day stool collection was rapidly thawed and stools were weighed before the whole collection was pooled.

The pooled samples were homogenised for 20 minutes in five times the faecal weight of cold water using a Stomacher 3500 (Seward, London, UK). A portion of the homogenate was then transferred to a beaker containing a magnetic stirrer (to ensure continuous mixing) and 10 ml aliquots were transferred to 20 ml glass vials, freeze dried (Edwards Modulyo freeze drier with three shelf closing accessory) for four days (with heating after 24 hours) and sealed under vacuum. The lyophylised samples were stored until required for analysis.

BILE ACID EXTRACTION

Bile acids were extracted according to methods previously described by Setchell *et al*,^{45 46} with a recent modification.⁴⁷

Duplicate samples of 10 mg freeze dried faeces were taken, 100 μ l ³H cholic acid (17 nCi=0.68 pmol) and 0.25 μ mol nordeoxycholic acid were added as recovery standards.

The faecal sample was transferred to a 250 ml round bottomed flask with 10 ml 90% ethanol and sonicated for 15 minutes before addition of 20 ml hexane. The flask was swirled and a bilayer allowed to form. The hexane (upper layer) was then decanted into a clean flask. The hexane step was repeated, the decanted hexane pooled, dried on a rotary evaporator and stored. A further 35 ml 90% ethanol was added to the flask containing the ethanolic faecal suspension, which was then refluxed gently for two hours. After reflux the sample was cooled on ice and centrifuged at 1500 g for 15 minutes. The supernatant was removed and saved. The faecal pellet was resus-

pended in 45 ml 80% ethanol and refluxed for a further two hours. The sample was centrifuged and the supernatant retained. The faecal pellet was then resuspended in 45 ml chloroformmethanol (1:1; vol:vol) and refluxed for a further one hour. Again the sample was centrifuged and the supernatant saved. Finally the pellet was resuspended in 45 ml chloroform:methanol and sonicated (to minimise any losses) before being centrifuged and the supernatant collected. The extracted supernatants were pooled, in a clean round bottomed flask and dried using a rotary evaporator.

The dried sample was resuspended in 5 ml acid water (pH 3–4) which was then sonicated for five minutes. The suspension was then filtered through a 10×1 cm column of Lipidex 1000. The column was converted to the aqueous form with 40 ml distilled water before use. The eluate was collected in a conical flask together with a column wash of 40 ml water (aqueous extract). Bile acids were recovered from the Lipidex 1000 by eluting the column with 40 ml 100% methanol into the original round bottomed flask.

The aqueous extract was then passed through a Sep Pak C18 cartridge (500 mg) prepared by previous washing with 5 ml 100% methanol followed by 10 ml water. Bile acids were eluted from the cartridge with 5 ml 100% methanol. The methanolic eluates were pooled in the original round bottomed flask and dried on a rotary evaporator.

The extracted bile acids were resuspended in 20 ml 72% ethanol. The pooled 'hexane' fraction (stored earlier) was resuspended in 10 ml 72% ethanol. The ethanolic suspensions were then filtered through a column of Amberlyst A-15 $(2 \text{ cm} \times 10 \text{ cm})$, the column then being washed with a further 20 ml of 72% ethanol and the samples pooled. The column was prepared by suspending the Amberlyst A-15 in 72% ethanol for 24 hours before slurrying it into columns and allowing it to settle by gravity. This was then treated with 50 ml 2 M NaOH in 72% ethanol and washed to neutrality with 200 ml 72% ethanol. It was converted to the H^+ form by washing with 50 ml 2 M HCl in 72% ethanol, followed by 72% ethanol until the eluate was neutral. After use, the Amberlyst was resuspended in 72% ethanol until required when it was reactivated as described. The Amberlyst resin was changed after every 10 extractions.

GROUP SEPARATION

Group separation was achieved by applying the effluent from the Amberlyst to a column of Lipidex DEAP (dietylaminohydroxypropyl Sephadex LH-20) as described by Almé et al.48 This was prepared in the acetate form by washing 100 g Lipidex-DEAP on a sintered glass funnel with 500 ml potassium acetate-KOH 0.5 M in 72% ethanol and then with further volumes of 72% ethanol until the eluate was neutral. It was then washed with 500 ml 0.1 M acetic acid in 72% ethanol and washed to neutrality again with 72% ethanol. The gel was slurried into glass columns and allowed to settle by gravity (\approx 30 minutes) to give a column size of 0.5×14 cm. After the addition of the sample, it

was eluted under pressure using nitrogen $(0.5-1.0 \text{ Kg cm}^{-2})$ to give a flow rate of 50–80 drops per minute. The column was washed with 9 ml 72% ethanol (pH 7) to give a fraction comprising neutral compounds and any esterfied bile acids. Unconjugated bile acids were obtained by elution with 7.5 ml 0.15 M acetate in 72% ethanol (apparent pH 3.8, adjusted with ammonium hydroxide) followed by 4.5 ml 72% ethanol. The remaining glycine conjugated, taurine conjugated and sulphated bile acid fractions were obtained by continued stepwise elutions with acetate buffers, but are not further considered in this report.

The unconjugated fractions were dried on a rotary evaporator and taken up in methanol as required.

DERIVITISATION

The internal standard (62.5 nmol coprostanol) was added to the sample. Diazomethane was freshly prepared using $\bar{0.54}$ g KOH dissolved in 14 ml 90% methanol and 2.86 g N-methyl-nnitroso-p-toluenesulphonamide in 40 ml diethyl ether in a round bottom flask. After five minutes condensed diazomethane was collected in an ice cooled tube. The bile acid mixture was transferred to a screw top reaction vial containing 0.5 ml methanol and 1 ml diazomethane. Methylated bile acids were produced by reaction at room temperature for one hour and subsequently evaporated to dryness under nitrogen. The trimethylsilyl ester derivatives were obtained by reacting the dried bile acid sample with 200 μ l of the silanisation mixture (pyridine: hexamethyldisilazine : trimethylchlorosilane; 3:2:1; vol:vol:vol) at 55°C for one hour. The derivative was dried and extracted with 3×200 µl hexane and centrifuged in a microtube at 13400 g for 10 minutes. The supernatant was transferred to a glass gas liquid chromatography vial (Chromacol Ltd, London, UK) dried and resuspended in 200 µl hexane.

CHROMATOGRAPHY

A 1 µl volume was injected onto the 25 m DB1 capillary tube column with a 1 m retention gap (Jones Chromatography, Hengoed, Wales) of a PU4400 gas chromatography with a PU4700 auto injector (Phillips Scientific, Cambridge, UK). The sample had a 170:1 split, at an initial temperature of 220°C, ramping at 2°C/minute to 280°C at which it was kept for 15 minutes, using helium as the carrier gas. Detection was by flame ionisation using hydrogen and air at 290°C together with helium as a make up gas to improve the detection limit. The integration system was a PU3202 computer utilising PU6000 software (Phillips Scientific). Ultra pure gases were used and were further purified before entry into the chromatograph using the following filters; oxyfilter, moisture filter and charcoal filter ×2 (Chrompack, London, UK).

The lower detection limit for bile acids was 0.3 pmol. Putative bile acids were identified by comparing peaks with the retention times of authentic standards and subsequently confirmed by gas liquid chromatography mass spectro-

metry⁴⁹ at the Clinical Research Centre's Unit of Mass Spectrometry. Relative retention times and response factors were calculated by comparison of authentic standards to coprostanol. Concentrations of the individual bile acids were calculated by comparing their peak areas with that of the internal standard coprostanol. The values from at least two chromatograms were used to obtain a mean; the percentage individual bile acid present was calculated from the micromolar concentrations. Total bile acid concentration was calculated as the sum of all gas liquid chromatography detected individual bile acids.

STATISTICAL ANALYSIS

All results are expressed as the median value with the range to indicate dispersion.

Comparisons between the three groups was undertaken using the Kruskall-Wallis test adjusted for ties; if significant (p<0.05) subsequent analysis between two groups was undertaken using the two-tailed Mann-Whitney U test. A χ^2 test was used to evaluate differences in sex distribution and frequency of occurrence of individual bile acids. Tests of correlation were undertaken using Spearman's rank correlation test (two-tailed). Significance was taken as p<0.05.

ETHICAL CONSIDERATIONS

All patients gave informed consent. The study was approved by the Ethical Committee of the West Birmingham Health Authority.

MATERIALS

Reference bile acids were obtained from Steraloids Ltd (Croydon, UK) and Sigma UK Ltd (Poole, UK). ³H cholic acid was obtained from Amersham International PLC (Aylesbury, UK). All solvents were AR grade and were obtained from Fisons along with pyridine (Loughborough, UK). All other reagents were purchased from Sigma UK Ltd. Lipidex-DEAP was obtained from Packard United Technologies (Pangbourne, UK). Sep-pak C-18 cartridges (500 mg) were purchased from Millipore-Waters (Watford, UK). All water used was double distilled and deionised using the Milli-Q system (Millipore, Watford, UK). Helium CP grade (>99.999% pure), Hydrogen hydrocarbon zero grade (>99.999% pure), and air BTCA 74 grade were supplied by BOC-Special Gases (Wolverhampton, UK).

Results

PATIENTS

There were no significant differences in the ages, weights, or stool frequency between the three groups. Transit times and daily stool weights are shown in Table I.

All the cancer patients had adenocarcinomas: two had Dukes's stage A, seven had B, and five had C. The polyps varied in size from 0.5 cm to 1.0 cm and five showed dysplastic changes (one mild and four moderate dysplasia).

BILE ACIDS

The mean recovery of the radiolabelled standard was 92.9% (7.11) (SD). A typical chromatogram is shown in the Figure; 16 unconjugated bile acids were identified during this study (Table II). From our own observations (unpublished) and those of others,45 sulphated and conjugated bile acids represent less than 10% of the total bile

TABLE I Patient details

	Controls (n=10)	Cancers (n=14)	Polyps (n=9)
Sex Male	7	8	9*
Female	3	6	0
Age (yr)	68.5	67.5	65
8- ()-/	60-81	47-79	27—73
Weight (kg)	65.5	71	70.5
	53-98	44-92	62-92
Stools/day	1.3	1.7	1.8
	0.8-5	0.4-7	1-2.8
Wet wt/day (g)	149 ^a	86 ^b	161°
	81-308	12-150	94-337
Transit time (hr)	68 ^d	114 ^e	52 ^f
	15-114	42-114	11-94

Upper figures indicate median, lower figures in italics indicate range. * χ^2 (df=2) 0.1>p>0.05

range. * χ^2 (df=2) 0·1>p>0·05. Kruskall-Wallis (H=12·3, df=2) a v v v p=0·002 Kruskall-Wallis (H=10·7, df=2) d v v p=0·005 Mann Whitney a v p=0·007 b v p p=0·002 a v p p=0·033 c v p p=0·033 c v p=0·002 d v n n=0·31

v fp=0.31



TABLE II Faecal bile acid concentration (µmol/g dry faeces) in control, cancer, and polyp patients

Bile acid	Control (n=10)	Cancer $(n=14)$	Polyp (n=9)
	(10 10)	((
Lithocholate	$1.07 (10)^{a}$	1·56 (<i>14</i>) ⁶	2·41 (9) ^c
3α hydroxy-5 β cholanoate	0.38-2.03	0.37-3.83	0.88-3.22
Deoxycholate	0.61(10)	1·17 (<i>14</i>)	1.53 (9)
3α , 12α dihydroxy- 5β cholanoate	0.18-2.78	0.09-2.53	0.8–0.89
Isolithocholate	0·35 (<i>10</i>)	0·42 (<i>14</i>)	0·42 (9)
3β hydroxy-5β cholanoate	0.07–1.03	0.14-0.8	0·18-0·96
Trans-isodeoxycholate	0.07 (9)	0·13(11)	0·24 (9)
3α,12β dihydroxy-5α cholanoate	0-0.36	00.52	0.06-0.69
12-oxo-lithocholate	0.09 (7)	0·18 (10)	0.14(7)
3α hydroxy-12oxo-5β cholanoate	0-0.35	0-1.61	0-0.82
Isodeoxycholate	0.03(6)	0.06(9)	0.23 (8)
3α , 12 β dihydroxy-5 β cholanoate	00·18	00:36	00·À9´
Trans-isolithocholate	0.00(4)	0.00(4)	0.00(3)
3β hvdroxy-5α cholanoate	0-0.17	0-0.12	0-0.2
Chenodeoxycholate	$0.05(6)^{d}$	$0.00(1)^{e}$	$0.07(5)^{f}$
3α , 7α dihydroxy-56 cholanoate	0-0.23	$0 - 0 \cdot 12'$	0-0.78
Cholate	0.05(6)	0.00(4)	0.1(6)
3α , 7α , 12α tribydroxy-56 cholanoate	0-0.14	0-0.12	0-1.06
Hyodeoxycholate	0.00(4)	0.00(6)	0.00(2)
3a 6a dihydroxy, 58 cholanoate	0-0.15	0.00(0)	0.00(2)
Ursodeovycholate	0.00(4)	0.00(3)	0.00(4)
3g7B dihydroxy-5B cholanoate	0.0.24	0.00(5)	0.0.97
Surp university - Sp cholanoate	0-00(4)	0.07(11)	0-0 67
3a 78 128 tribudrovy-58 chalapoate	0.0.22	0.0.46	0.3.37
7-ovo-lithocholate	0.00(2)	0.00(2)	0.00(1)
3 a hudrovy-70yo-58 cholanoate	0.00(2)	0.00(2)	0.1.02
7 ovo deovucholate	0-0 32	0-0+3	0-1.02
2 12 dibudrowy 7 avo 58 abolamonto	ND	0.00(2)	ND
12 and abaradaamabalata	ND 0.00(1)	0-00(1)	ND
2 7 7 dibudrowy 50 shalanaata	0.00(1)	0.00(1)	ND
50,70 dinydroxy-5p cholanoate	0-0-1	0.00(1)	ND
2 1 1 7 12 1 50 1 1		0.00(1)	
ou nyuroxy-/,12 dioxo-op choianoate		0-0.00	
SUM 2° BA	1.90°	3.93"	5.23
	0.74-6.63	0.8/-8.28	2.16-13.40

Upper figures indicate median values, lower figures in italics indicate range, figures in brackets indicate number of patients in which bile acid detected. ND=not detected. Frequency of detection of chenodeoxycholate: ${}^{d} v v {}^{f} \chi' (df=2) 0.02 > p > 0.01$ Kruskall-Wallis (H=8·54, df=2) ${}^{a} v {}^{b} v {}^{c} p = 0.034$ Kruskall-Wallis (H=8·54, df=2) ${}^{a} v {}^{b} v {}^{c} p = 0.014$ Kruskall-Wallis (H=8·53, df=2) ${}^{g} v {}^{h} v {}^{i} p = 0.043$ Mann Whitney ${}^{a} v {}^{b} p = 0.17$ ${}^{b} v {}^{c} p = 0.027$ ${}^{a} v {}^{c} p = 0.008$ p=0.008 p = 0.01p=0.77 p=0.25 $\dot{\mathbf{p}} = 0.1$

p = 0.02

Unconjugated faecal bile acids

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 $\begin{array}{l} 4=3\alpha \ hydroxy-5\beta \ cholanoate (lithocholate) \\ 5=3\alpha, I2\beta \ dihydroxy-5\beta \ cholanoate (isodeoxycholate) \\ 6=3\alpha, I2\beta \ dihydroxy-5\alpha \ cholanoate (allo-isodeoxycholate) \\ \end{array}$ $7=3\alpha$, 12α dihydroxy- 5β cholanoate (deoxycholate) $8=3\alpha$, 7α , 12α trihydroxy-5 β cholanoate (cholate) $9=3\alpha$ hydroxy-120x0-5\beta cholanoate (12-0x0-lithocholate)

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acids excreted in faeces and are not further considered in this report.

The faecal bile acid concentration (µmol/g dry weight) was greater in the polyp group when compared with controls. The daily faecal bile acid excretion was found to be significantly higher in the polyp group compared with cancer patients, while the difference between the polyp group and controls was of borderline significance (Table III).

Of the individual bile acids identified, a significantly lower concentration of the primary bile acid chenodeoxycholate was found in the faeces of cancer patients, indeed it was detected significantly less frequently in cancer patients (Table II). The concentration of lithocholic acid was found to be higher in patients with polyps than controls as was the concentration of total secondary bile acids (Table II).

TABLE III Concentration and daily excretion of faecal bile acids

	Controls (n=10)	Cancers (n=14)	Polyp (n=9)
Bile acids (µmol/g dry	1·96 ^a	3·97 ^b	5·23 ^c
weight faeces)	0·91–6·97	0·87–8·41	2·16–13·67
Bile acid daily	71·1 ^d	50·6 ^e	146·5 ^f
excretion (µmol)	13·4–303·1	3·5–209·7	71·8–327

Upper figures indicate median values, lower figures in italics **AUSKAII-Wallis** (H=6·77, df=2) ${}^{a}v^{b}v^{c}p=0.034$ Kruskall-Wallis (H=6·77, df=2) ${}^{d}v^{c}v^{f}p=0.045$ Mann Whitney ${}^{a}v^{b}p=0.33$ ${}^{a}v^{c}p=0.016$ ${}^{b}n^{c}p=0.016$ p=0.75 p=0.067 p=0.018

TABLE IV Chenodeoxycholate (CDCA) and cholate (CA) derivatives (mol%), ratios of CDCA:total CDCA derivatives and CA:total CA derivatives

	Controls (n=10)	Cancers (n=14)	$\begin{array}{c} Polyps\\ (n=9) \end{array}$
CDCA derivatives	50.27	51.84	47.08
	44.77-82.48	36.68-93.49	38.29-76.64
CA derivatives	49.51	45.23	52.92
	11.04-55.23	6.57-63.32	23.36-61.32
CDCA:CDCA derivatives	0.022	0.000	0.042
	0-0.18	0-0.04	0-0.21
CA:CA derivatives	0.037ª	0.00e	0.032 ^r
	<i>0–0</i> · <i>1</i> 7	0-0.04	00.23

Upper figures indicate medians, lower figures indicate range. Kruskall-Wallis (H=9.56, df=2) ${}^{a}v{}^{b}v{}^{c}p=0.009$ Kruskall-Wallis (H=6.29, df=2) ${}^{d}v{}^{c}v{}^{f}p=0.043$ Mann Whitney ${}^{b}v{}^{c}p=0.008$ ${}^{a}v{}^{c}p=0.09$ ${}^{d}v{}^{c}p=0.034$ ${}^{d}v{}^{c}p=0.037$

p = 0.97

 $v^{f} p = 0.034$

There were no significant differences in the proportions of total chenodeoxycholic acid or cholic acid derivatives between the three groups (Table IV). The proportion of metabolites of chenodeoxycholic acid and cholate was, however, higher in the cancer group, as reflected by lower ratios of chenodeoxycholic acid and cholate to their total derivatives (Table IV). The overall proportion of the secondary bile acids was highest among the cancer patients when compared with both polyp patients and controls (Table V). No significant differences were found in lithocholic acid:deoxycholic acid ratio or the lithocholic acid:deoxycholic acid×total faecal bile acid index when the groups were compared (Table V).

The proportion of secondary bile acids in each group was found to correlate strongly with both transit time and stool weight ($r_s = 0.717$, p<0.01 and $r_s = -0.527$, p<0.01 respectively).

Discussion

One of the key elements of this study was the exercise of tight control over variables which may influence faecal bile acid profiles; age, transit time, gall stones, cholecystectomy, hepatic function, hepatic metastases, previous surgery, and antibiotics.

Not only is there a significant day to day variation in faecal bile acid excretion, but faecal bile acid concentrations may vary between different areas of the same stool.46 A minimum complete three day stool collection with homogenisation has been shown to be necessary to minimise variation and most previous studies,

TABLE V Faecal bile acids: lithocholic acid: deoxycholic acid ratios, proportions of secondary bile acids

Bile acid	Controls (n=10)	Cancers (n=14)	Polyps (n=9)
LCA:DCA	1·49 0·97_7·47	1.84	1.31
LCA:DCA X TFBA	6·21	7.9	8·19
	1·67-12·41	3.76-19.52	2·83_21·91
%2° bile acids (mol%)	95·19ª	100^{b}	97·19°
	81·73–100	96.85–100	78·18–100

Upper figures indicate median values, lower figures in italics indicate range. LCA=lithocholic acid; DCA=deoxycholic acid; TFBA=total faecal bile acid concentration LCA= nucleon $z_{\text{intro}(p)}$ (µmol/g). Kruskall-Wallis (H=7·1, df=2) $v^{b}v^{c}p=0.029$ Mann Whitney $v^{b}p=0.029$ $v^{c}p=0.02$ $v^{c}p=0.02$

 $v^{c} p = 0.90$

faecal bile acids. Over a five day period patients with adenomatous polyps had a higher daily faecal bile acid excretion compared with cancers and higher total faecal bile acid and secondary bile acid concentrations, compared with controls. Patients with established colorectal cancer showed increased proportions of secondary bile acids in their faeces.

possibility of bacterial biotransformation of

Similar findings of an increased faecal bile acid concentration and increased secondary bile acid concentration in patients with adenomatous polyps were reported by Reddy and Wynder.²⁶ Although stool collection in their study was only carried out for 24-48 hours, it was done at home. Breuer et al³⁶ studied 12 patients with adenomatous polyps and found no differences in faecal bile acid excretion; the participants in this study were, however, hospitalised and a change in diet may be of relevance to their findings. Furthermore, it is not clear whether or not the control group was screened for polyps. Because polyps are a relatively common finding among a population of a similar age to their patients,⁵⁰ this is a significant confounding factor. No attention was paid to previous cholecystectomy nor to the presence of gall stones. Both factors have been shown by several groups^{42 51-53} to be associated with changes in biliary bile acid composition. Similar difficulties arise in the interpretation of another study of polyp patients that showed no differences in faecal bile acid excretion.35 Furthermore, this latter study was carried out on native Japanese, a population with a considerably lower incidence of colorectal cancer and probably not comparable with our population.

A significantly higher concentration of lithocholic acid was found in polyp patients compared with controls. This may be important as lithocholic acid has been shown to be genotoxic in in vitro studies.14 15

Patients with established colorectal cancer had a lower faecal concentration of the primary bile acid chenodeoxycholate and were found to excrete a higher proportion of secondary bile acids, compared with both controls and polyps. This was reflected by the increased proportions of the biotransformation products of both cholate and chenodeoxycholic acid seen, and is in accordance with two previous studies.26 27 Our own recent studies⁴³ have shown an increased proportion of secondary biliary bile acids in patients with colorectal cancer. The changes we observed may have occurred as a result of an increased absorption of secondary bile acids, previously shown in patients with large and severely dysplastic polyps.^{40 54} This, together with feedback inhibition of primary bile acid synthesis, resulting from increased enterohepatic cycling of secondary bile acids,55 may explain the differences seen. In our studies therefore, faecal and biliary bile acid findings have been consistent. Although there were no differences in faecal bile acid concentration in those patients

with colorectal cancer, total daily excretion was significently lower than in patients with polyps, an increased colonic absorption of bile acids may in part account for this. Further studies comparing the bile acid absorption in polyp patients with that in patients with established colorectal cancer, may help to clarify this issue.

A prolonged colonic transit time may be expected to result in increased bacterial metabolism of bile acids, and the observation that an increased proportion of secondary bile acids was strongly correlated with a prolonged bowel transit time is therefore of interest. Yet the cancer patients, who had significantly longer mouth-to-anus transit times, were not obstructed clinically or radiologically and stool frequencies were similar in all groups. Nagengast et al,⁵⁶ using a similar radioopaque marker method to our own, could find no relationship between faecal bile acid concentration or pattern and transit time in healthy controls. A recent study,⁵⁷ however, showed an association between diets high in refined sugar and prolonged mouth-to-anus transit time (radioopaque markers), despite a reduced mouth-to-caecum transit time (H₂ breath test). This was coupled with an increased faecal bile acid concentration and proportion of secondary faecal bile acids. These findings would suggest that an interplay of various intraluminal factors, including bile acids themselves, may affect colonic function. Clearly the nature of this relationship requires further investigation, particularly as no other case controlled studies have reported on transit time in relation to faecal bile acid concentration.

As little is known about the distribution of bile acids between the water or solid fractions of faeces and their possible cellular effects, the observed differences in faecal bile acids may not totally reflect their significance. Indeed the presentation of bile acids to the colonic mucosa is likely to be dependant upon the concentration of bile acids in the water fraction of faeces.58 The role played by dietary fibre is of importance to this compartmentalisation because it binds bile acids.²¹ Moreover, the fermentation of fibre will have effects on colonic pH.59 This may not only affect 7-alpha-dehydroxylase activity (responsible for secondary bile acid formation),⁶⁰ but also the ionic composition and the solubility of bile acids in the water compartment of faeces.61 These factors and their interrelationships require elucidation in these patients. Nevertheless, we propose that the increased concentration of bile acids and increased daily excretion may predispose to the development of polyps. This could result from bile acid mediated effects on cellular proliferation through several possible pathways as already discussed.4 16-18 The differences observed in cancer patients are more difficult to explain, as the faecal bile acid profiles of cancer patients in our study were different from those of polyps and controls. We would suggest that an altered metabolism (possibly related to gut transit time) may occur at the large polyp or cancer stage. This, with or without an increased absorption of faecal secondary bile acids may account for the differences observed in both faecal and biliary bile acids.

Neither the ratio of lithocholic acid: deoxycholic acid, or lithocholic acid:deoxycholic acid×total faecal bile acid were significantly increased in patients with colorectal cancer or polyps. This suggests that such indices are unlikely to provide useful markers for the identification of colorectal cancer risk as previously proposed.37 38

In conclusion we have shown a significantly increased faecal bile acid concentration in patients with adenomatous polyps and an increased proportion of secondary bile acids in the faeces of patients with established colorectal cancer. This study lends further support to the theory that these compounds may have some role in the pathogenesis of polyps and colorectal cancer.

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