

Crystal structure of d(GCGAAAGCT) containing a parallel-stranded duplex with homo base pairs and an anti-parallel duplex with Watson–Crick base pairs

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ABSTRACT

A DNA fragment d(GCGAAAGCT), known to adopt a stable mini-hairpin structure in solution, has been crystallized in the space group $I4_122$ with the unit-cell dimensions $a = b = 53.4 \text{ \AA}$ and $c = 54.0 \text{ \AA}$, and the crystal structure has been determined at 2.5 Å resolution. The four nucleotide residues CGAA of the first half of the oligomer form a parallel duplex with another half through the homo base pairs, $C_2:C_2^+$ (singly-protonated between the Watson–Crick sites), $G_3:G_3$ (between the minor groove sites), $A_4:A_4$ (between the major groove sites) and $A_5:A_5$ (between the Watson–Crick sites). The two strands remaining in the half of the parallel duplex are split away in different directions, and they pair in an anti-parallel *B*-form duplex with the second half extending from a neighboring parallel duplex, so that an infinite column is formed in a head-to-tail fashion along the *c*-axis. It seems that a hexammine cobalt cation supports such a branched and bent conformation of the oligomer. One end of the parallel duplex is stacked on the corresponding end of the adjacent parallel duplex; between them, the guanine base of the first residue is stacked on the fourth ribose of another duplex.

INTRODUCTION

Since Watson and Crick (1) proposed a structural model for the double helix of DNA from their interpretation of available fiber-diffraction data, DNA has been assumed to always adopt the *B*-form conformation *in vivo*, and this assumption has been verified by many X-ray analyses of DNA fragments. An alternative conformation is the *A*-form (2), which is observed predominantly in RNA structures, but DNA can also have this conformation at low-humidity (3). In either form, the two DNA strands are positioned to form a right-handed helix.

Another discovery was left-handed *Z*-DNA, which was found to form in alternating purine–pyrimidine sequences, $(CG)_n$ (4,5). This conformation is also involved in some biological systems (6). All three major duplex structures, however, consist of two sugar-phosphate backbones aligned in anti-parallel fashion with one strand in the 5'→3' and the other strand in the 3'→5' orientation. These two strands are bound to each other through Watson–Crick base pairings. In the case of RNAs, too, the alignment of the two chains is anti-parallel.

Homo pairs formed between the same kinds of bases were found in the X-ray crystal structures of d(CG) (7) and d(CA) (8) at acidic pH, the two strands being aligned in parallel fashion with the same strand polarity. These findings stimulated extensive structural studies on parallel duplexes. Robinson *et al.* (9,10) showed from NMR studies that the d(CGATCG) sequence forms a *B*-form anti-parallel duplex at neutral pH, as well as a parallel duplex with homo base pairs at low pH (<5), and that the d(CGA) sequence forms such a motif. The sequence d(TCGA) also forms a parallel-stranded duplex, but the first T residue does not interact directly with another T (11). The parallel-aligned strands have been observed in the structures of DNA duplexes with reverse Watson–Crick A:T base pairs (12), triplexes (13), tetraplexes (14) and the *i*-motif (15,16).

In general, the parallel duplex structure is difficult to find in natural DNA, because every system *in vivo* is established by using the Watson–Crick-type anti-parallel duplex between complementary strands. In some cases, as reported by Tchurikov and co-workers (17–19), however, parallel duplex formation is possible for specific functions in local regions of DNA. Such a parallel duplex has been also detected in *Escherichia coli* mRNA (20). To understand the biological functions of parallel duplexes, it is necessary to clarify their structural properties. Despite various indirect evidence for parallel-stranded duplexes, detailed structures have not yet been fully investigated in the crystalline state.

From extensive studies on the nucleotide sequence d(GCGAAAGC), we found that the DNA fragment has extraordinary properties with (i) abnormal mobility in electrophoresis (21), (ii) high thermo-stability (22), (iii) an

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unusual CD spectrum different from expected patterns (22), and (iv) robustness against nuclease digestion (23). The biological functions of such a fragment were discussed for G4 phage (24). For this specific sequence, a mini-hairpin structure was postulated from NMR and CD spectroscopy (22). The present nonamer, containing a thymidine residue at the ninth position, shows very similar properties but a higher thermostability when compared with the octamer (22), and was also expected to have a similar hairpin structure in solution (22). In contrast, the crystal structures of the halogeno-derivatives of the nonamer containing 5-bromo- or 5-iodo-cytidines as the second residue show a zipper-like duplex (25,26). To investigate the structural versatility of such a specific sequence in relation to its physicochemical properties, we determined the crystal structure of native d(GCGAAAGCT) by X-ray diffraction, and found not only that this sequence adopts another type of structure, but also, surprisingly, that half of the sequence forms a parallel duplex, whereas the other half forms an anti-parallel duplex. Here we describe the structural details of the novel parallel duplex and compare its conformation with the standard anti-parallel duplex.

MATERIALS AND METHODS

Synthesis and crystallization

A DNA oligomer with the sequence d(GCGAAAGCT) was synthesized on a DNA synthesizer. The nonamer was purified by HPLC, and converted to the Na⁺ salt on a cation-exchange column. Finally, the cation was removed by gel-filtration chromatography. Crystallization conditions were surveyed at room temperature by the hanging-drop vapor diffusion method with a screening kit reported by Berger *et al.* (27) for nucleic acids, each droplet (4 μ l initial volume) being equilibrated with 700 μ l of reservoir solution. Although some crystals only appeared under conditions containing Co(NH₃)₆Cl₃ with Mg acetate and NaCl, their size was too small for X-ray analysis. We found that adding 2,6-diaminopyridine and a small quantity of *n*-decanoyl-*N*-methylglucamide, purchased from DOJINDO Laboratories Co., Ltd, facilitated crystal growth. After further optimization of crystallization conditions, suitable crystals for X-ray experiments were obtained at room temperature in the following conditions. A droplet was prepared by merging 2 μ l of a 3 mM DNA solution including 24 mM 2,6-diaminopyridine and 2 μ l of a reservoir solution (pH 6.0) containing 20 mM MgCl₂, 80 mM NaCl, 30 mM Co(NH₃)₆Cl₃, 25% 2-methyl-2,4-pentanediol and 0.17% *n*-decanoyl-*N*-methylglucamide in 40 mM Na-cacodylate buffer. Crystals were grown up to 0.15 \times 0.15 mm in size for 2 weeks. Fresh crystals were mounted in nylon cryoloops (Hampton Research) with the reservoir solution containing 30% (v/v) 2-methyl-2,4-pentanediol as a cryoprotectant and stocked in liquid nitrogen.

Data collection

For MAD phasing using the anomalous scattering of cobalt atoms, X-ray data were taken at BL18b of Photon Factory in Tsukuba. Three different wavelengths, 1.00, 1.6049 and 1.6054 \AA , were employed, based on XAFS measurements. A crystal specimen was cooled at 100 K and X-ray diffraction patterns were collected on a CCD detector (Quantum 4z)

positioned 250 mm (for 1.00 \AA) or 150 mm (for 1.6 \AA) from the crystal. Diffraction patterns, using 4 or 5 $^\circ$ oscillation, with 90 s exposure per frame, and a total range 180 $^\circ$, were processed at 2.9 or 3.1 \AA resolution by the program DPS/MOSFLM (28–31). The three data sets were scaled separately using the programs SCALA, TRUNCATE and SCALEIT of the CCP4 suite (32).

Another data set was collected for structure refinement at 100 K with synchrotron radiation ($\lambda = 0.900 \text{\AA}$) at BL44XU of SPring-8 in Harima. Intensity data were recorded on a 3 \times 3-arrayed CCD detector (PX210; Oxford Inst. Co.) positioned 243 mm from the crystal. Each frame was taken with 1 $^\circ$ oscillation and 15 s exposure. In total, 180 frames were processed at 2.4 \AA resolution with the same program mentioned above. To compensate for overloaded reflections, the data set was merged with the data taken with $\lambda = 1.00 \text{\AA}$. 1676 unique reflections with 100% completeness were obtained with $R_{\text{merge}} = 6.8\%$. The statistics of the data collection and crystal data are summarized in Table 1.

Structure determination and refinement

From the three data sets with different wavelengths, a cobalt atom was uniquely found on the crystallographic 2-fold axis. Phases were estimated by the MAD method using the program SOLVE (33), the figure-of-merit being 0.64. Electron density was modified by a solvent flattening technique (solvent content 60.9%) with the program CNS (34). In the modified map, the phosphate-ribose backbone with the individual bases was easily traced. The molecular structure was constructed on a graphic workstation with the program QUANTA (Molecular Simulation Inc.).

The atomic parameters were refined with the program CNS (34) through a combination of rigid-body, simulated-annealing, crystallographic-conjugate-gradient-minimization refinements and *B*-factor refinements, followed by interpretation of an omit map at every nucleotide residue. No restraints were applied between paired nucleotides or to the sugar puckering. Only the 5'-terminal G₁ residue was difficult to assign due to rather poor density, but after several steps of refinement, it appeared in an $F_o - F_c$ map. Once the two magnesium atoms were found, one at a general position and the other on the 2-fold axis, these were added to the refinement along with several water molecules. By taking hydrogen-bond interactions with surrounding atoms into consideration, water molecules octahedrally coordinated to the two magnesium atoms and ammonium cations coordinated around the cobalt atom could be reasonably assigned. In the final refinement, reflections in the outer shell beyond 2.5 \AA resolution were suppressed due to extremely low I/σ values (<3). The statistics of the structure refinement are summarized in Table 1. Figure 1 shows local $2F_o - F_c$ maps for the homo base pairs. All local helical parameters, including torsion angles and pseudorotation phase angles of ribose rings, were calculated by using the program NUPARAM (35). Figure 1 was drawn with the program O (36), Figures 2 and 5 with the program MOLSCRIPT (37) and Figure 4 with the program RASMOL (38).

RESULTS

Crystal packing of the two duplex regions

The Mathew's criterion for protein crystals (39) is also applicable to nucleic acid crystals as a general rule (40); from

Table 1. Crystal data, statistics of data collection, and statistics of structure refinement

X-ray source	Synchrotron radiation at Photon Factory			Synchrotron radiation at SPring-8	Structure refinement
Space group	I_1422			$I4_122$	Resolution range (Å)
Unit cell (Å)	$a = b = 53.3, c = 53.9$			$a = b = 53.4, c = 54.0$	Used reflections
Asymmetric unit ^a	1			1	R-factor (%) ^b
Wavelength (Å)	1.00 (remote)	1.6049 (peak)	1.6054 (edge)	0.900	R_{free} (%) ^c
Resolution (Å)	38–2.9	38–3.1	38–3.1	27–2.4	Number of DNA atoms
Observed reflections	12 067	9902	9904	20 224	Number of waters
Unique reflections	973	806	809	1647	Number of magnesium atoms
Completeness (%)	100	99.9	100	98.3	Number of cobalt hexamine molecules
In the outer shell (%)	100 (3.06–2.9)	100 (3.27–3.1)	100 (3.27–3.1)	100 (2.53–2.4)	RMSD from ideal geometry
R_{merge} (%) ^d	6.4	8.4	7.7	5.4	Bond lengths (Å)
R_{anom} (%) ^e	2.3	5.3	4.0	–	Bond angles (°)
Theoretical f'/f''	0.16/1.78	–5.94/3.92	–6.09/3.93	–	Improper angles (°)
Refined f'/f''	–0.339/1.241	–6.007/3.329	–7.201/2.031	–	

^aA single strand d(GCGAAAGCT).

^bR-factor = $100 \times \sum ||F_o| - |F_c|| / \sum |F_o|$, where $|F_o|$ and $|F_c|$ are the observed and calculated structure factor amplitudes, respectively.

^cCalculated using a random set containing 10% of observations that were not included during refinement (55).

^d $R_{\text{merge}} = 100 \times \sum_{hklj} |I_{hklj} - \langle I_{hkl} \rangle| / \sum_{hklj} I_{hklj}$.

^e $R_{\text{anom}} = 100 \times \sum_{hklj} |I_{hklj}(+) - I_{hklj}(-)| / \sum_{hklj} [I_{hklj}(+) + I_{hklj}(-)]$.

this, the number of nonamers in the asymmetric unit is estimated to be 1. Initially the structure was expected to adopt one of two possible structures, hairpin-type (stem-loop) or zipper-like duplex, because these have been already assigned from NMR studies (22) and X-ray analysis of a bromo-derivative (25). However, an electron density map of the present nonamer clearly shows that the conformation of the phosphate backbone is quite different (see Fig. 1). The first 5'-terminal guanosine residue G_1 protrudes into the solvent region and its density is slightly broadened even on the final $2F_o - F_c$ map, suggesting its flexibility or disorder in the crystal. Figure 2 shows that the next four residues, C_2 , G_3 , A_4 and A_5 , form a right-handed helix with those of another strand, related by a crystallographic 2-fold symmetry along the face-diagonal axis of the tetragonal lattice. The two chains are associated and oriented in parallel alignment. The remaining halves of the two strands, containing the A_6 , G_7 , C_8 and T_9 residues, split away in opposite directions along the c -axis. Around this axis, one half is further coupled with another half of the adjacent parallel duplex, related by a crystallographic 4_1 symmetry, as shown schematically in Figure 3A. These two halves are associated in an anti-parallel alignment to form another duplex through normal, Watson-Crick base pairs. This coupling is repeated in a head-to-tail manner to form a long column along the c -axis, so that the parallel-stranded regions extrude at right angles from the surface of the column, like knobs. The anti-parallel columns are linked together by facing the flat ends of the parallel-stranded regions, and forming base-base stacking interactions between C_2 residues, as shown in Figure 3B. This linkage is further stabilized by hydrophobic interactions of the 5'-terminal G_1 residues. The guanine moiety of the major conformer, which is stacked on the ribose ring of the fourth residue A_4 , covering the hydrophobic face of the ring in a hydrophilic environment. Such an interaction has already been observed in nucleic acid crystals (41). Between the four columns of anti-parallel duplexes, there is a long channel for solvent molecules,

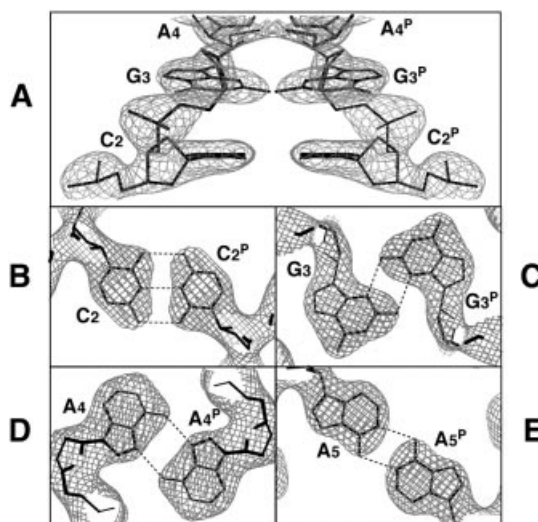


Figure 1. Local $2F_o - F_c$ maps for phosphate backbones (A) and homo base pairs (B–E). Densities are contoured at 2σ level for (A), 1.5σ for (B and C), and 0.8σ for (D and E). Broken lines indicate possible hydrogen bonds.

extending along the c -axis, as shown in Figure 3B. The water molecules in this channel were difficult to locate.

Base-pair formation

Figures 1B–E and 4 show the homo base pairs found in the parallel duplex region; the hydrogen bond distances and angles are also indicated. These base pairings stabilize the parallel duplex region. The first C:C⁺P pair involves three hydrogen bonds between N4 and O2^P, between N3 and N3^P, and between O2 and N4^P (P indicates the parallel strand). In this case, the C:C⁺ pair must be protonated at N3 of either C residue, with disordering between N3 and N3^P. The second G:G^P pair involves N2–H···N3^P and N2^P–H^P···N3 hydrogen bonds between the minor groove sites. These two base pairs

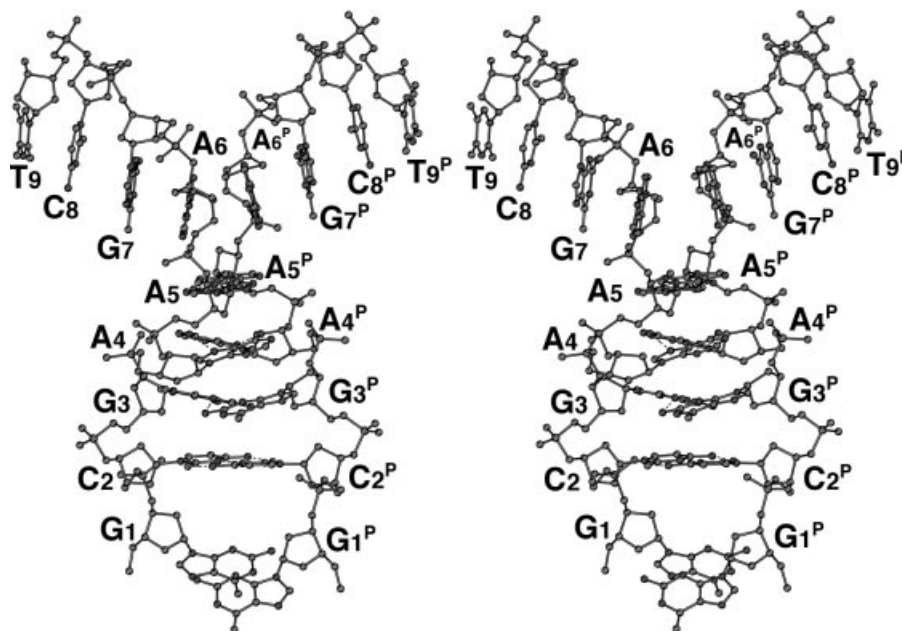


Figure 2. A stereo-pair diagram of the parallel duplex with homo base pairs. The two halves of the strands branch away from the parallel region in different directions. The superscript P indicates the counter strand aligned in parallel fashion.

are of the same types as those reported in several X-ray structures (7,42,43). The third A:A^P pair involves the two N6–H···N7 hydrogen bonds between major sites. This pairing is known to occur between the protonated adenines (44,45) and between the neutral forms (46). The latter case is detected in d(TCGA) at pH 4.0 by NMR measurement (11). Since the present crystal was obtained at pH 6.0, the adenine moiety of the A₃ residue is not protonated. The three base-pair formations, described above, are identical to those derived from NMR analyses for d(CGA) (10) and d(TCGA) (11).

In the fourth A:A^P pair, the pairing occurs between the Watson–Crick sites, through two N6–H···N1 hydrogen bonds. This pairing is also the same as that found in crystals of small molecules (47) and oligo-nucleotides (48). In all base pairings of the parallel duplex region, the two bases or two nucleotide residues are images of each other under the crystallographic 2-fold symmetry. On the other hand, the anti-parallel duplex region is formed by four base pairs, A₆:T₉^A, G₇:C₈^A, C₈:G₇^A and T₉:A₆^A, with the normal Watson–Crick types (^A indicates the anti-parallel strand); the distances and angles for hydrogen bond formations are in standard ranges (49).

Conformation of the parallel duplex

A defining feature of the parallel-stranded duplex is the backbone conformation, forming a right-handed helix. Table 2 shows values of representative helical parameters. The average values of the twist angle, D_z , and the C1···C1 distance in the parallel region seem close to those of the anti-parallel region, which has a typical B-form conformation (49). However, the individual values fluctuate highly in the parallel region depending on the pairing types. Both the C₂:C₂⁺ and A₅:A₅ homo pairs are formed at the Watson–Crick pairing sites, resulting in a shorter C1···C1 distance at the C₂:C₂⁺ pair and a longer one at the A₅:A₅ pair. The second G₃:G₃ pair occurs in the minor groove sites so that the C1···C1 distance is

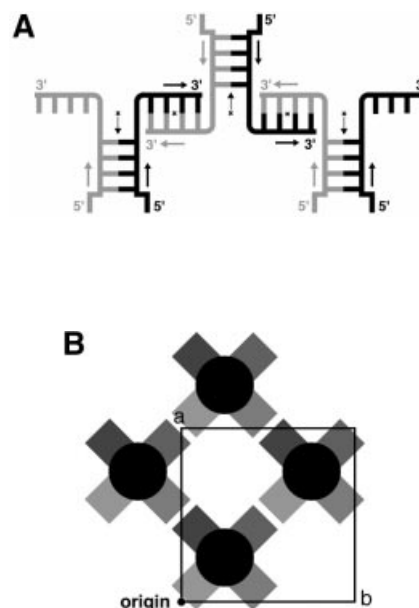


Figure 3. A coupling motif formed by the two arms extending out from the parallel duplexes, to form a long column of anti-parallel duplexes (A) and a large channel extending along the *c*-axis, presumably occupied by water (B). The crystallographic 2-fold axes are indicated with an arrow and a cross. The parallel duplex regions protrude from the column like knobs at right angles. The flat end of each parallel duplex is stacked with another flat end of the adjacent parallel duplex.

the shortest. The C1···C1 distance of the A₄:A₄ pair formed in the major groove sites is similar to the normal distance, 10.8 Å. From the first pair to the second pair, the twist angle is only 13°, reflecting a small deviation from the Watson–Crick pairing site to the minor groove site. But the largest movement

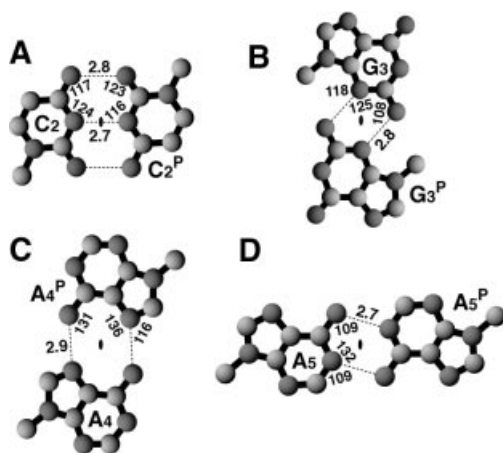


Figure 4. Hydrogen bond formation (broken lines) between the homo bases. Values indicate atomic distances (Å) and angles (°). The crystallographic 2-fold axis is indicated with a black spot at the center of the base pair.

is found in going from the second step to the third, since the pairing changes from the minor groove site to the major groove site. The fourth pair requires a small twist, due to the change from the major site to the Watson–Crick site. Depending on the pairing sites, the stacking distances are also changed. The largest differences are found in the opening angles. In the parallel region, the paired bases must occupy a *trans* position with respect to each other. Furthermore, the two bases in the present crystal are exactly related by the crystallographic 2-fold symmetry so that the opening angles should be just 180°. Large propeller twists occur at the G:G pair in the minor site and at the A:A pair in the major site, but they are not abnormal values. Such a twisting is required to release the close contacts of other regions and the geometrical constraints for hydrogen bonding (41).

The torsion angles in the ribose phosphate backbone are indicated in Table 3. They fluctuate within an allowed range,

with a similar behavior in the parallel and the anti-parallel regions. The conformational feature of the parallel region is similar to those of the NMR structures (10,11), including some differences at both ends of the duplexes; the root mean square deviations (RMSDs) are within 1.4 Å when the corresponding parts are superimposed on each other. In the anti-parallel region, every residue, except for the terminal T₉ residue, adopts a C2'-*endo* conformation, with a normal fluctuation (41). The T₉ residue adopts a C4'-*exo* conformer, due to a close contact with a hydrated magnesium cation, as shown in Figure 5A. Even in the parallel region, however, the second and third residues (forming the first and second pairs) adopt a C2'-*endo* conformation. In contrast, the fourth and fifth residues adopt a C3'-*endo* pucker. This drastic change occurs at the junction between the two regions, parallel and anti-parallel, at a point where they are largely bent, as shown in Figure 5A and B. The two A₆ residues are split away from the parallel region to form Watson–Crick pairs with the neighboring T₉ residues, which are almost perpendicular to the A₅:A₅ pair.

The ribose ring of the first residue G₁ adopts a C4'-*endo* conformation and the C5'–C4' bond adopts an unusual torsion angle in order to extend the guanine moiety for stacking on the ribose ring of the neighboring A₄ residue.

Ionic effect for stabilizing the structure

The two arms containing the A₆, G₇, C₈ and T₉ residues, which form an anti-parallel duplex as described above, are further related by the crystallographic 2-fold symmetry. On this symmetry axis, a hexa-ammine cobalt cation is bound to the center of the anti-parallel duplex, as seen in Figure 5B. The N7 and O6 atoms of the G₇ residue are directly hydrogen bonded to two different ammonium cations separately. A water molecule, which is bound to the third ammonium cation, is hydrogen bonded to the phosphate oxygen of the A₄ residue and to the N7 and N6 atoms of the A₅ residue. Another water bridges the phosphate oxygen of the A₆ residue and the third

Table 2. Local helical parameters and intra base-pair parameters

Base pair	Inclin ^a	Tip	Twist	D _z ^b	Prop ^a	Buck ^a	Open ^a	C1...C1 ^b	
Parallel									
2	C:C	0	0	13	3.6	-6	4	180	9.5
3	G:G	0	0	80	3.2	17	-23	180	7.4
4	A:A	0	0	21	3.8	27	27	180	10.8
5	A:A	0	0	21	3.8	-2	-1	180	13.3
Anti-parallel									
6	A:T	0	-4	36	3.4	-18	-2	15	10.4
7	G:C	0	-4	36	3.4	-15	-3	21	10.8
8	C:G	-9	0	42	3.2	-15	3	21	10.8
9	T:A	0	4	36	3.4	-18	2	15	10.4
Average value									
Parallel		0	0	38	3.5	19	2	180	10.3
Anti-parallel		-3	0	38	3.3	-16	0	18	10.6

^aInclin, inclination angle (°); Prop, propeller twist angle (°); Buck, buckle angle (°); Open, opening angle (°) (35).

^bDistance in Å.

Table 3. Torsion angles ($^{\circ}$) and sugar conformation

Sequence	P-O5' α	O5'-C5' β	C5'-C4' γ	C4'-C3' δ	C3'-O3' ϵ	O3'-P ζ	C1'-N χ	Phase ^a θ	Pucker
G	-	-	160	149	-134	156	-124	-133	C4'-endo
Parallel									
C	-57	165	57	137	-174	-69	-120	150	C2'-endo
G	-74	-176	49	139	-155	154	-82	154	C2'-endo
A	-59	135	55	95	-136	-65	-175	31	C3'-endo
A	-67	172	43	87	-145	-69	-124	15	C3'-endo
Anti-parallel									
A	-74	-156	51	147	-162	-98	-114	166	C2'-endo
G	-62	170	39	150	-124	174	-86	153	C2'-endo
C	-49	132	45	130	-173	-110	-110	141	C1'-exo
T	-68	173	49	88	-	-	-124	44	C4'-exo

^aPseudo-rotation phase angle of ribose ring.

ammonium cation. With these interactions, the hexa-ammine cobalt cation thus supports bending of the nonamer at the junction between the parallel region and the anti-parallel region. The C3'-endo pucker of the A₄ and A₅ residues may be required for this bending.

The two hexa-hydrated magnesium cations were located. One is bound to the G_{3'} and C_{2'} residues through hydrogen bonds with coordinated water molecules, as shown in Figure 5C. The primed residues are those of the parallel duplex, stacked on the original duplex. The guanine moiety of the flipped G₁ residue, which is covered by the ribose ring of A_{4'}, is bound to the coordinated waters. Another magnesium cation is located on the 2-fold axis at the branching point of the two arms of the parallel duplex, where the two other arms, extending from different parallel duplexes, come together to form the two anti-parallel duplex regions, as shown in Figure 5A. The cation is bound to the A₆:T_{9'}^P and T_{9'}:A₆^P pairs, the two hydrated water molecules being hydrogen bonded to the two O₂ atoms of these T₉ residues. Thus, the stacking between the two base pairs is further stabilized by these interactions.

DISCUSSION

From NMR and CD measurements, the present d(GCGAAAGCT) nonamer shows extraordinary properties, similar to those of the d(GCGAAAGC) octamer (22). Therefore, it was expected to adopt a similar mini-hairpin structure in solution (22). To investigate the structure of the nonamer, we first crystallized its iodo-derivative d(GC^XGAAAGCT) (C^X = 2'-deoxy-5-iodocytidine) at pH 6.0 (50). Initially, this crystal structure was solved at 1.9 Å resolution, but it was difficult to distinguish between the two possible structures, hairpin dimer or duplex dimer. Recently, we have re-determined its structure at a higher resolution, 1.4 Å (26); this structure clearly indicates that the derivatives form a duplex dimer with a zipper-like structure, similar to the bromo-derivative solved earlier (25). The crystal structure of d(GCGAAAGC), which has been recently determined at a high 1.6 Å resolution (26), also shows that the octamer forms a zipper-like duplex when crystallized at pH 7.0. In either structure, the mini-hairpin and the zipper, however, the stem and the duplex parts are formed in anti-parallel alignment, as is usual in B-DNA. Although the present crystals were prepared under acidic conditions

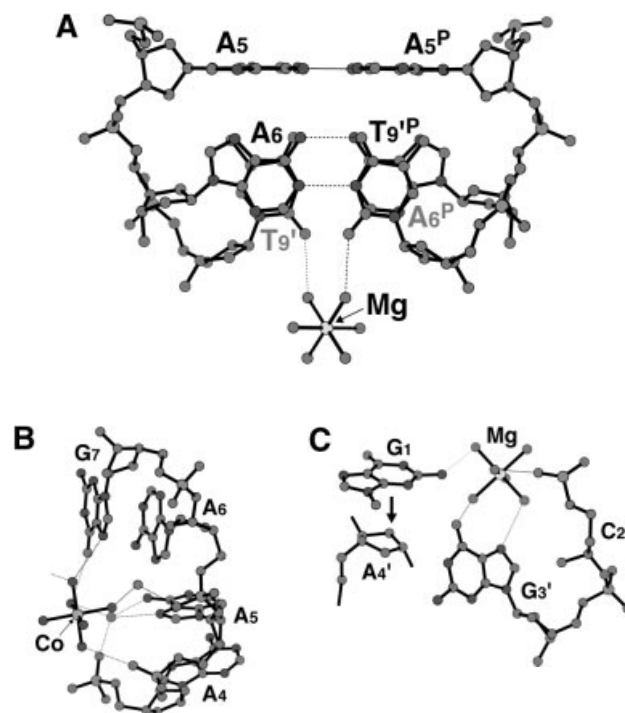


Figure 5. Bound cations stabilizing the d(GCGAAAGCT) structure. (A) A junction between the parallel and the anti-parallel duplexes. A hydrated magnesium cation is bound to the two T₉ residues to stabilize their stacking interaction. Broken lines indicate possible hydrogen bonds. (B) A hexa-ammine cobalt cation bound to the bent corner joining the parallel and the anti-parallel duplexes. Broken lines indicate possible hydrogen bonds. This diagram shows half of the interactions. The other half is related by the 2-fold symmetry, so that the cobalt atom is centered in the spherical hole of the two bent strands. (C) Stacking interaction of the G₁ residue with the ribose moieties of the neighboring A₄ residue. The hydrated magnesium cation bridges the G₁, G_{3'} and C_{2'} residues of different strands to stabilize the stacking of the two parallel duplexes.

(at pH 6.0 in crystallization droplets) slightly above the proton dissociation constant of cytosine [$pK_a \approx 4.5$ (51)], hemi-protonation is induced on one cytosine residue, because the C:C pair has a negative hole at the center, which can more easily accept a proton to stabilize the pairing. Such phenomena have been observed in several cases (52). It is necessary to note that one (C₂) of the two cytosine residues is hemi-protonated and the other cytosine residue C₈ is not protonated

at all under such mild acidic conditions. Therefore, it is reasonable to suppose that two types of half-duplexes are in equilibrium in solution: one is a parallel duplex between CGAA parts and the other is an anti-parallel duplex between AGCT parts. Once a parallel half duplex is formed, the conformation with two extended arms could be preferred due to base–base interactions. In crystallization, the parallel duplexes easily associate to form a long column, with the anti-parallel duplex associated in a head-to-tail manner. In addition, these columns are laterally bound through stacking interactions between the parallel duplexes. It will be interesting to clarify which structure occurs when these arms are missing. The crystal structure of the tetramer d(CGAA) will give an answer to this question.

Another question is whether the parallel duplex structure is involved in some biological function. As the major part of living systems is established by using the Watson–Crick pairing between complementary strands, it may be difficult to find such a parallel duplex (*P*-DNA) in the stable structure of a double helix. Once *Z*-DNA was discovered, it took 20 years to find *Z*-DNA-binding proteins (6). For the parallel duplex, too, it is expected that when DNA is unwound to perform a function in a single-stranded state and slightly acidic conditions, *P*-DNA will occur as a local structure, as reported by Tchurikov and co-workers (17–19). As single-strand DNAs from DNA viruses or phages can adopt many types of structures (like RNA), sometimes these could have a specific function. A typical example is a DNA enzyme derived by *in vitro* selection (53). In practice, functional DNAs are being developed for several purposes (54), because DNA is more stable thermally and chemically than RNA. The present finding of *P*-DNA will be a useful structural guide for designing such functional DNAs. Furthermore, the structural features found in the present investigation may be useful for controlling DNA function by manipulating the pH.

DATA BANK ACCESSION CODES

The atomic coordinates have been deposited in the Protein Data Bank with the ID 1ixj.

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