Is N-acetylation of 5-aminosalicylic acid reversible in man?

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In two healthy male subjects the disposition of deuterated N-(²H₃) acetyl-5-aminosalicylic acid (d₃-ac-5-AS) was investigated after a single rectal dose of 500 mg d₃-ac-5-AS. Urine and plasma were analysed by h.p.l.c. and gas chromatography massspectrometry. Peak concentrations of around 0.5 µg/ml occurred within 6 h and plasma concentrations declined thereafter with a half-life ($t_{1/2}$) of about 6 h which was confirmed by urinary excretion data. Renal clearance of d₃-ac-5-AS ranged between 200 and 300 ml/min and only 4.4–11.2% of the dose could be recovered in the 48 h urine. Since no undeuterated ac-5-AS could be detected in any of the plasma and urine samples an irreversible acetylation of 5-AS is assumed in man.

Keywords 5-aminosalicylic acid acetylated metabolite deacetylation

Introduction

Several studies indicate that 5-aminosalicylic acid (5-AS) is the active moiety of sulphasalazine (SZ) in the treatment of Crohn's disease and ulcerative colitis (Khan et al., 1977; Klotz et al., 1980: van Hees et al., 1980; Campieri et al., 1981). 5-AS is liberated from SZ by bacterial azo cleavage of SZ in the colon and the absorbed 5-AS is mainly N-acetylated by capacity-limited gut wall and hepatic metabolism to N-acetyl-5-aminosalicylic acid (ac-5-AS) which is subsequently excreted into the urine (Das & Dubin, 1976; Pieniaszek & Bates, 1979; Fischer et al., 1983). Fast and slow acetylators have not been described. However, there are conflicting data as to whether ac-5-AS might contribute to the therapeutic effect (Willoughby et al., 1980; Binder et al., 1981). These contradictory findings might be due to differences in study design and/or patient variability in drug disposition. Since acetylation is the major pathway for the elimination of 5-AS we investigated if an equilibrium exists between its acetylation and deacetylation to see whether reversible N-acetylation might contribute to the action of 5-AS. Such reversibility of acetylation has been observed with some sulphonamides (Schröder, 1973; Vree et al., 1981, 1983) and might also occur with other acetylated drugs.

Methods

The synthesis and analysis of d_3 -ac-5-AS (including d_0 -ac-5-AS) in plasma and urine by h.p.l.c. and GC/MS have been described in detail elsewhere (Fischer *et al.*, 1981, 1984). The disposition and excretion of d_3 -ac-5-AS was studied in two healthy male volunteers with normal kidney and liver function following a single rectal dose of 500 mg in the form of two suppositories. Blood samples (10 ml) were drawn after 0.25, 0.5, 0.75, 1, 1.5, 2, 2.5, 3, 4, 5, 6, 8 and 10 h. Urine was collected in fractions between 0-6 h, 6-12 h, 12-24 h and 24-48 h. Plasma and aliquots of urine were kept frozen (-25°C) until analysis.

The elimination half-life $(t_{1/2})$ of ac-5-AS was calculated from the log/linear decline of the plasma drug concentration (6–10 h postdosing) by linear regression analysis and from urinary excretion data by the rate and sigma-minus methods. The renal clearance (CL_R) of d₃-ac-5-AS was calculated from plasma and urine drug concentrations.

Results

Following rectal administration d₃-ac-5-AS was detected in the plasma of both subjects after 1 h

and its concentration peaked at around $0.5 \,\mu g/$ ml within 6 h. This was followed by a monoexponential decline. The low plasma concentrations (range 0.05 to 0.5 µg/ml) suggested incomplete absorption and this was supported by cumulative urinary recoveries of 22.1 mg and 55.9 mg which accounted for 4.4 and 11.2% of the dose, respectively in each subject. The decline in plasma drug concentrations might have been influenced by a slow absorption process. The plasma $t_{1/3}$ of 6 h and 6.3 h for d₃ac-5-AS were in reasonable agreement with values derived from the rate of urinary excretion (5.3 h and 9.4 h, respectively). The renal plasma clearance of ac-5-AS averaged 199 ml/ min and 300 ml/min in the two subjects studied (Table 1).

GC/MS analysis of urine (1 day collection) and plasma samples (h.p.l.c. concentrations between 0.1 and 0.5 μ g/ml) demonstrated in both subjects that the measured concentrations of ac-5-AS consisted only of d₃-ac-5-AS (Figure 1). No d₀-ac-5-AS was detected in any sample above the limit of background noise or natural content of d₃-ac-5-AS (purity \ge 99.7%).

Discussion

There is conflicting information as to whether or not ac-5-AS has a therapeutic effect in the treatment of ulcerative colitis (Willoughby *et al.*, 1980; Binder *et al.*, 1981). Therefore, we investigated the possibility that acetylation of 5-AS might be reversible as a partial explanation of this discrepancy. The existence of such an acetylation-deacetylation equilibrium could be shown if a distinction can be made between ac-5-AS as administered drug and as apparent 'drug'- after deacetylation and reacetylation. The analytical problem can be resolved by using labelled ac-5-AS. Therefore, we used h.p.l.c. for the determination of total concentration of labelled and non-labelled ac-5-AS, whereas d₃-ac-5-AS was discriminated from ac-5-AS by GC/MS.

Since ac-5-AS was administered as retention enema in the studies of Willoughby *et al.* (1980) and Binder et al. (1981), we gave d₃-ac-5-AS in a similar form (suppositories). Owing to a limited availability of labelled ac-5-AS the supposed 'therapeutic' dose of 500 mg could be given only to two subjects. The GC/MS results (Figure 1) indicated that ac-5-AS is not deacetylated to a measurable extent. Similar results have been found with N-acetylprocainamide (NAPA) where only 2.8% of the administered NAPA-¹³C was metabolized by deacetylation (Stec et al., 1980). In contrast, the existence of a deacetylation-acetylation equilibrium in man has been demonstrated with a sulphonamide, N_4 -acetyl-sulphamerazine (Vree *et al.*, 1983). The success in demonstrating such an equilibrium depends on the position of the equilibrium. However, in man acetylation of 5-AS is fast, as indicated by a short $t_{1/2}$ of 0.6 to 1.4 h (unpublished data), whereas elimination of ac-5-AS is relatively slow ($t_{1/2}$ 5–11 h, Table 1). Based on this pharmacokinetic comparison one may assume that the formation of ac-5-AS is rapid such that its renal clearance becomes the rate-limiting step in the overall elimination process of 5-AS. Similar rate limiting steps in the metabolite kinetics of 5-AS have been reported in rats (Houston & Cassidy, 1982). Since plasma ac-5-AS concentrations exceed those of 5-AS following multiple dosing with 5-AS (Fischer et al., 1983) sufficient amounts of the metabolite would be available for deacetylation. However, the latter process could not be

	Subject 1 (60 kg, 37 years)	Subject 2 (67 kg, 40 years)
a: Plasma data		
Maximal concentration, (µg/ml)	0.45	0.55
Time of peak (h)	6	6
Apparent $t_{1/2}$ (h)	6.3	6.0
b: Urine data		
t_{16} (h)	9.4ª	5.3ª
· · · ·	11.4 ^b	8.1 ^b
CL _R (ml/min)	199	300
Ae (48 h) (mg)	55.9	22.1

Table 1Disposition of d_3 -ac-5-AS in two healthy subjects followingrectal administration of 500 mg

^aCalculated by the rate method

^bCalculated by the sigma minus method

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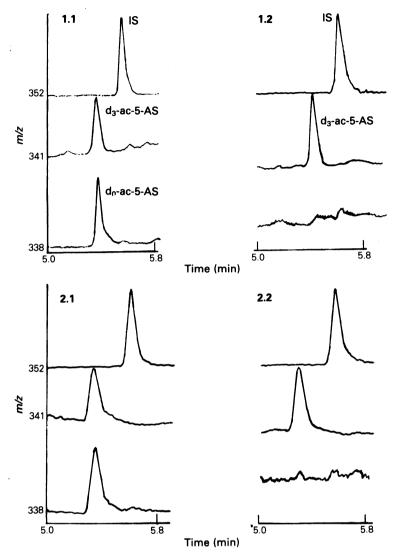


Figure 1 Typical GC/MS-chromatograms of plasma and urine extracts. Electron impact mode 70 eV, selected ion monitoring m/z 352, 341 and 338 for the derivatives methylester- bis (trimethylsilyl) ether of internal standard, d_3 -ac-5-AS and d_0 -ac-5-AS, respectively.

- 1.1 Calibration: blank plasma spiked with non-deuterated and deuterated ac-5-AS (100 ng of each/1 ml diluted plasma).
- 1.2 Plasma extract from one of the volunteers at 6 h after drug administration.
- 2.1 Calibration: blank urine spiked with non-deuterated and deuterated ac-5-AS (100 ng of each/1 ml diluted urine)
- 2.2 Urine extract from one of the volunteers, 0-6 h urine collection after drug administration.
- To each of the samples 400 ng of propionyl-5-AS as internal standard (IS) was added prior to analysis.

verified by our d_0/d_3 -ac-5-AS measurements. Therefore, although we have studied only two individuals, it appears that the discrepancy in the results of investigations of the therapeutic efficacy of ac-5-AS (Willoughby *et al.*, 1980; Binder *et al.*, 1981) may not be explained by a metabolic phenomenon. The excellent technical assistance of Mrs A. Kopp and the secretarial help of Mrs B. Grözinger are gratefully acknowledged. One of us (U.K.) was supported by the Deutsche Forschungsgemeinschaft Bonn (Project Kl 436/2). The work was also supported by the Robert Bosch Foundation, Stuttgart.

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