



Published in final edited form as:

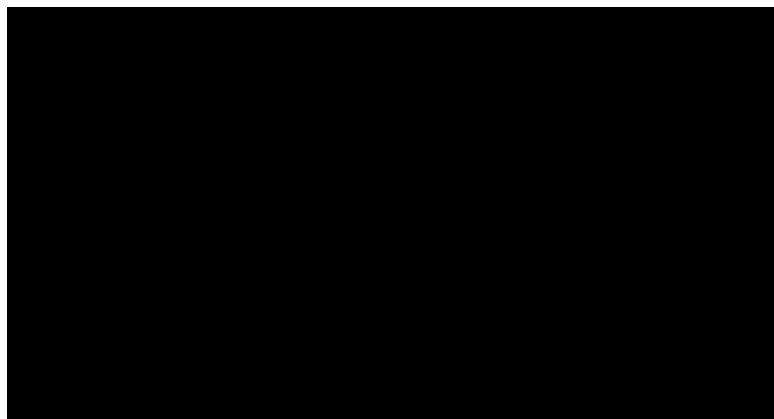
Org Lett. 2002 April 4; 4(7): 1189–1192.

Pyran Annulation: Asymmetric Synthesis of 2,6-Disubstituted-4-methylene Tetrahydropyrans

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Abstract



A reaction process for the asymmetric construction of a variety of *cis* or *trans* disubstituted pyrans is described. This sequence allows for the asymmetric convergent union of two aldehydes with silyl-stannane reagent 1 in a two-step process: catalytic asymmetric allylation of the first aldehyde using 1 with a BITIP catalyst, followed by reaction of the alcohol so obtained with a second aldehyde and TMSOTf.

An ever increasing number of biologically significant natural products, especially those of marine origin, possess functionalized tetrahydropyran rings as critical substructural motifs. Prominent examples of considerable current interest would include bryostatin 1 and phorbaxazole (Figure 1).¹⁻² Our interest in these structures, particularly bryostatin 1 and other potential therapeutic agents based upon this structural platform,³ has led us to investigate new methods for the asymmetric synthesis of pyrans bearing the substitution patterns commonly encountered in these natural products. The occurrence of such pyran motifs in a number of biologically active natural products has inspired the development of many clever methods for obtaining these oxacycles. The intramolecular attack of an olefin on an oxocarbenium ion, or intramolecular Prins reaction, although one of the established methods, has been underutilized in this capacity. This is in part due to the paucity of methods for obtaining suitable substrates to provide appropriately functionalized *nonracemic* tetrahydropyrans.

There exists extensive precedent for the construction of oxygen-containing heterocycles, including tetrahydropyrans, by approaches based upon intramolecular versions of the basic Prins reaction.⁴ Previous work has also documented the utility and enhanced reactivity of allylsilanes as intramolecular traps for oxocarbenium ions generated in various ways.⁵ Kang

has described reactions of racemic hydroxy allylsilanes of general structure **2** with vinyl ethers and α -halo acetals to give acetals that were subsequently cyclized to afford pyrans.⁶ Marko has developed an efficient pyran synthesis that combines the Noyori method⁷ for generation of oxocarbenium ions with intramolecular allylsilane trapping and has coined the term ISMS (intramolecular silyl modified Sakurai) to describe this reaction (eq 1, Scheme 1).⁸ Studies by Marko have also documented the use of racemic hydroxy allylsilanes in sequences leading to pyrans via an initial ene reaction; Marko has suggested the term “IMSC” (intramolecular Sakurai cyclization) to refer to the cyclization step of the process (eq 2, Scheme 1).⁹ Very recently, during the preparation of our manuscript, Leroy and Marko reported on the use of a functionalized allylstannane in $\text{BF}_3 \cdot \text{OEt}_2$ -promoted reactions with aldehydes, followed by $\text{Bi}(\text{OTf})_3$ -promoted cyclocondensation with a second aldehyde, to yield differentially substituted pyrans as shown in Scheme 1 below (eq 3).¹⁰

Rychnovsky has also developed a process termed “segment coupling Prins cyclization”¹¹ and applied it to an extremely concise construction of the C_{20} – C_{27} pyran subunit of phorbaxazole.¹² Rychnovsky and Kopecky have also recently reported a tandem Mukaiyama aldol–Prins sequence utilizing allylsilanes, which bears similarity to the process described herein.¹³ Both of these methods access nonracemic pyrans. Smith has utilized the Petasis–Ferrier rearrangement in efficient syntheses of nonracemic tetrahydropyrans, and Panek has employed nonracemic allylsilanes to access dihydropyrans.¹⁴

Herein we report an exceptionally facile enantioselective synthesis of 2,6-*cis*-disubstituted-4-methylenetetrahydropyran systems. These are obtained in a very brief *three steps* from commercially available starting materials. As indicated in Scheme 2, the overall two-step reaction process results in the convergent union of two aldehyde components with a four-carbon unit derived from the known 2-(trimethylsilylmethyl)allyltri-*n*-butylstannane reagent **1**.¹⁵ For purposes of discussion, it is convenient to refer to these as the first aldehyde, R_1CHO , and the second aldehyde, R_2CHO .

The first step in the overall process is the preparation of the requisite hydroxy allylsilanes in nonracemic form. This was achieved for the present examples by catalytic asymmetric allylation of the first aldehyde R_1CHO with stannane **1**.¹⁶ These reactions were found to proceed well using our previously described BINOL titanium tetraisopropoxide (BITIP) catalyst;¹⁷ in general, the hydroxy allylsilanes were obtained in high yields and with high levels of enantioselectivity. (Table 1).

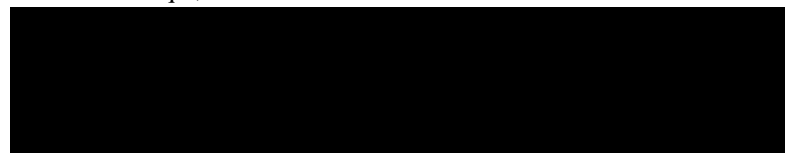
The second step of the process, the TMSOTf-promoted annulation of such silanes with the second aldehyde R_2CHO , was found to occur rapidly (generally within 15 min at -78°C) to provide the 2,6-*cis*-tetrahydropyran containing an *exo*-methylene in the 4-position. The *trans* diastereomer was not detected. The stereochemical outcome is in accord with previous observations and with the expectation that cyclization should occur via a chairlike transition state with R_1 and R_2 equatorially disposed. The results for several examples are provided in Table 2.

We have noticed an interesting combination of solvent effects and substrate dependence with respect to this annulation reaction. The reaction generally proceeded well in $\text{CH}_2\text{-Cl}_2$, but in all cases at least a trace amount of an *endocyclic* olefin that was inseparable from the desired product was obtained. In the cases where the cyclization substrate is **2c**, this problem was greatly magnified. Previous attempts by Marko to cyclize similar preformed TMS ethers also resulted in mixtures of *endo* and *exo* olefins, which were very difficult to separate.^{8a,18} However, noting that the substrates that performed best in this reaction contained α or β -alkoxy substituents, diethyl ether was also examined as solvent. A dramatic improvement was found

with this oxygenated solvent as no endocyclic olefins were obtained regardless of the substrate employed.

Another problem observed in these reactions is related to the use of substrates **2d** and **2e** (Table 1). In CH₂Cl₂, these substrates decomposed immediately upon exposure to TMSOTf. In ether at -78 °C no decomposition was observed; however, no cyclized product was obtained either. Under these conditions the alcohol is rapidly converted to the TMS ether, but the reaction stalls at this juncture.¹⁹ We are currently still examining this issue.

It is also possible to use this pyran annulation to obtain access to 2,6-*trans*-disubstituted pyrans. As shown in eq 4,



the annulation reaction works very well when an ortho ester is used as the electrophile.²⁰ Thus, the reaction of **2a** with trimethylorthoformate smoothly provides methylacetal **4** in excellent yield (eq 4). Although **4** is obtained as an epimeric mixture, this is of little consequence. Reaction of a storable nucleophile (allylsilane, allylstannane, or enol silane) with the oxocarbenium ion generated upon treatment of such acetals with a Lewis acid is well-known to afford the 2,6-*trans* isomer with high selectivity as a consequence of stereoelectronically preferred axial addition to the intermediate oxocarbenium ion.²¹

This annulation process also lends itself to iterative applications that allow for the rapid assembly of structures containing multiple pyran rings, such as the bis pyran subunit of phorboxazole. Thus, if a pyran constructed using this process contains a latent or protected aldehyde, exposure of the aldehyde function and reaction with an appropriate hydroxy allylsilane in a second annulation affords a bis pyran. An example of this iterative annulation is shown in Scheme 3. In this case, bis pyran **6** is assembled from stannane **1**, α -benzyloxyacetaldehyde, and β -OTBDPS propionaldehyde in five linear steps. The compatibility of both α - and β -alkoxy aldehydes in these reactions thus easily provides pyrans containing three differentiated and highly malleable functional groups on the THP ring. Finally, this pyran synthesis is quite flexible since the annulation can be approached from two different directions for any given pyran. This should increase its utility and help to promote its application.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

Acknowledgment

Financial support from the National Institutes of Health (through GM-28961) and by Pfizer, Inc. is gratefully acknowledged.

References

1. (a) Kageyama M, Tamura T, Nantz MH, Roberts JC, Somfai P, Whritenour DC, Masamune S. J. Am. Chem. Soc 1990;112:7407. For total syntheses of bryostatin and leading references to other work, see: (b) Evans DA, Carter PH, Carreira EM, Charette AB, Prunet JA, Lautens M. J. Am. Chem. Soc 1999;121:7540. (c) Ohmori K, Ogawa Y, Obitsu T, Ishikawa Y, Nishiyama S, Yamamura S. Angew. Chem., Int. Ed 2000;39:2290.
2. (a) Forsyth CJ, Ahmed F, Cink RD, Lee CS. J. Am. Chem. Soc 1998;120:5597. For total syntheses of phorboxazole and leading references to other work, see: (b) Evans DA, Fitch DM, Smith TE, Cee VJ.

- J. Am. Chem. Soc 2000;122:10033. (c) Smith AB III, Verhoest PR, Minbiole KP, Schelhaas M. J. Am. Chem. Soc 2001;123:4834. [PubMed: 11457294]
3. (a) Wender PA, De Brabander J, Harran PG, Jimenez J-M, Koehler MFT, Lippa B, Park C-M, Shiozaki S. J. Am. Chem. Soc 1998;120:4534. (b) Wender PA, De Brabander J, Harran PG, Jimenez JM, Koehler MF, Lippa B, Park CM, Siedenbiedel C, Pettit GR. Proc. Natl. Acad. Sci. U.S.A 1998;95:6624. [PubMed: 9618462] (c) Wender PA, Blaise L. Tetrahedron Lett 2000;41:1007. (d) Wender PA, Hinkle KW, Koehler MFT, Lippa B. Med. Res. Rev 1999;19:388. [PubMed: 10502742]
 4. (a) Adams DR, Bhaynagar SD. Synthesis 1977:661. Snider, BB. Comprehensive Organic Synthesis. Trost, BM., editor. 2. Pergamon Press; Oxford, U.K.: 1991. p. 527 (c) Cloninger MJ, Overman LE. J. Am. Chem. Soc 1999;121:1092. (d) Kozmin SA. Org. Lett 2001;3:755. [PubMed: 11259054]
 5. (a) Mohr P. Tetrahedron Lett 1993;34:6251. (b) Paquette LA, Tae J. J. Org. Chem 1996;61:7860. [PubMed: 11667744] (c) Sano T, Oriyama T. Synlett 1997:716. (d) Suginome M, Iwanami T, Ito Y. J. Org. Chem 1998;63:6096. [PubMed: 11672232] (e) Chen C, Mariano PS. J. Org. Chem 2000;65:3252. [PubMed: 10814229]
 6. Sung TM, Kwak WY, Kang K-T. Bull. Korean Chem. Soc 1998;19:862.
 7. Murata S, Suzuki M, Noyori R. Tetrahedron 1988;44:4259.
 8. (a) Mekhalfia A, Marko IE. Tetrahedron Lett 1991;32:4779. (b) Mekhalfia A, Marko IE, Adams H. Tetrahedron Lett 1991;32:4783.
 9. (a) Marko IE, Bayston DJ. Tetrahedron Lett 1993;34:6595. (b) Marko IE, Bayston DJ. Tetrahedron 1994;50:7141. (c) Marko IE, Mekhalfia A, Murphy F, Bayston DJ, Bailey M, Janousek Z, Dolan S. Pure Appl. Chem 1997;69:565. (d) Marko IE, Plancher J-M. Tetrahedron Lett 1999;40:5259.
 10. Leroy B, Marko IE. Tetrahedron Lett 2001;41:8685.
 11. Jaber JJ, Mitsui K, Rychnovsky SD. J. Org. Chem 2001;66:4679. [PubMed: 11421792]
 12. Rychnovsky SD, Thomas CR. Org. Lett 2000;2:1217. [PubMed: 10810711]
 13. Kopecky DJ, Rychnovsky SD. J. Am. Chem. Soc 2001;123:8420. [PubMed: 11516301]
 14. (a) Petasis NA, Lu S-P. Tetrahedron Lett 1996;36:141. For other recent asymmetric approaches to pyrans, see: (b) Smith AB III, Verhoest PR, Minbiole KP, Lim JJ. Org. Lett 1999;1:909. [PubMed: 10823221] (c) Smith AB III, Minbiole KP, Verhoest PR, Beauchamp TJ. Org. Lett 1999;1:913. [PubMed: 10823222] (d) Huang H, Panek JS. J. Am. Chem. Soc 2000;122:9836.
 15. (a) Kang K-T, Hwang SS, Kwak WY, Yoon UC. Bull. Korean Chem. Soc 1999;20:801. (b) Clive DLJ, Paul CC, Wang Z. J. Org. Chem 1997;62:7028. The 2-(chloromethyl)allylsilane precursor to **1** is commercially available (Aldrich)
 16. For an alternative preparation of such hydroxy silanes, by Noyori reduction of β -ketoesters followed by application of the Brunelle allylsilane synthesis, see ref ¹³.
 17. Keck GE, Tarbet KH, Geraci LS. J. Am. Chem. Soc 1993;115:8467. The "method B" catalyst preparation was found to be the most suitable for the reaction of stannane **1** with aldehydes. Experimental procedures may be found in Supporting Information
 18. Marko IE, Bayston DJ, Mekhalfia A, Adams H. Bull. Soc. Chim. Belg 1993;102:655. Marko subsequently reported a solution to this problem through the use of trimethylsilyl ethers of simple alcohols as an additive:
 19. Both Kang and Marko have also noticed greatly diminished yields using certain aromatic or heteroaromatic materials in their acetal cyclizations and ene reactions. Note refs ⁶ and ^{9a}.
 20. (a) Marko IE, Mekhalfia A. Tetrahedron Lett 1992;33:1799. For previous uses of acetals, ketals, and ortho esters in this context, note: (b) Marko IE, Mekhalfia A, Bayston DJ, Adams H. J. Org. Chem 1992;57:2211. Note also ref ^{8b}
 21. For two recent examples, see refs ^{2b} and ^{14b}.

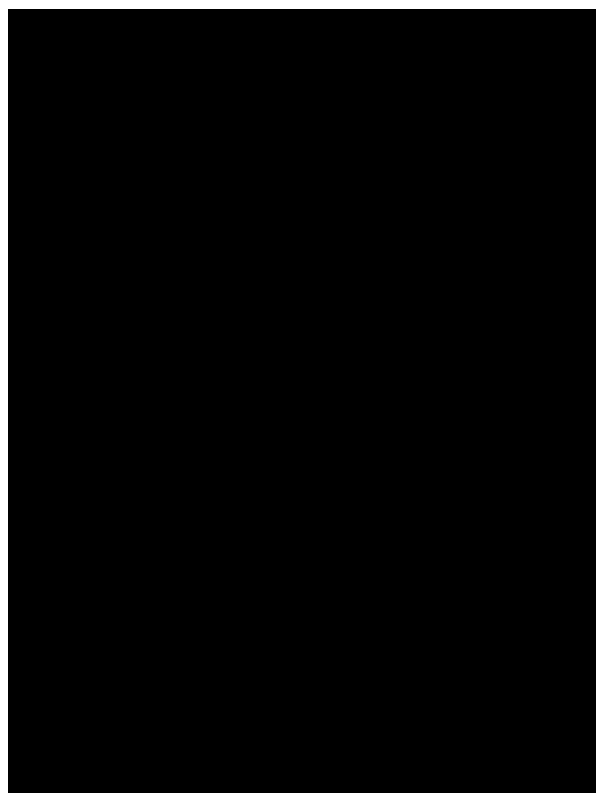
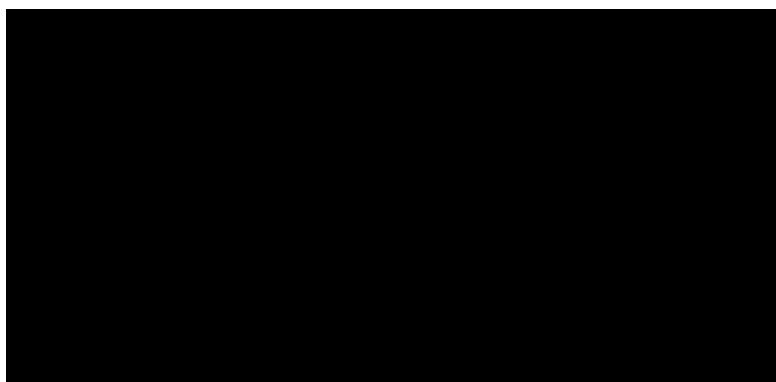


Figure 1.
Structures of bryostatin 1 and phorboxazole.



Scheme 1.
Related Racemic Pyran Syntheses



Scheme 2.
Overall Pyran Annulation Process

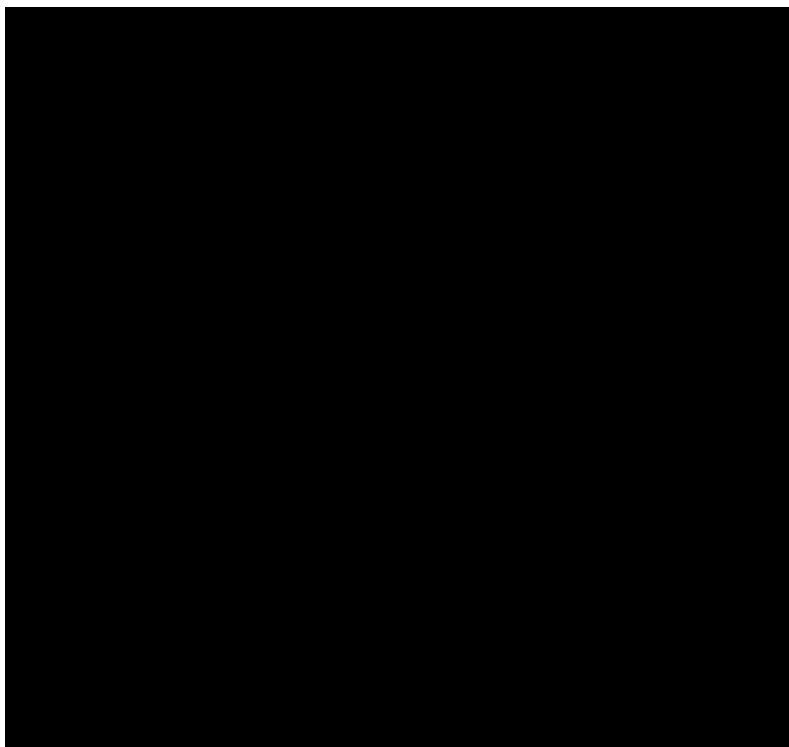


Table 1.
Isolated Yields and Enantiomeric Excesses for the Reaction of **1** with Aldehydes

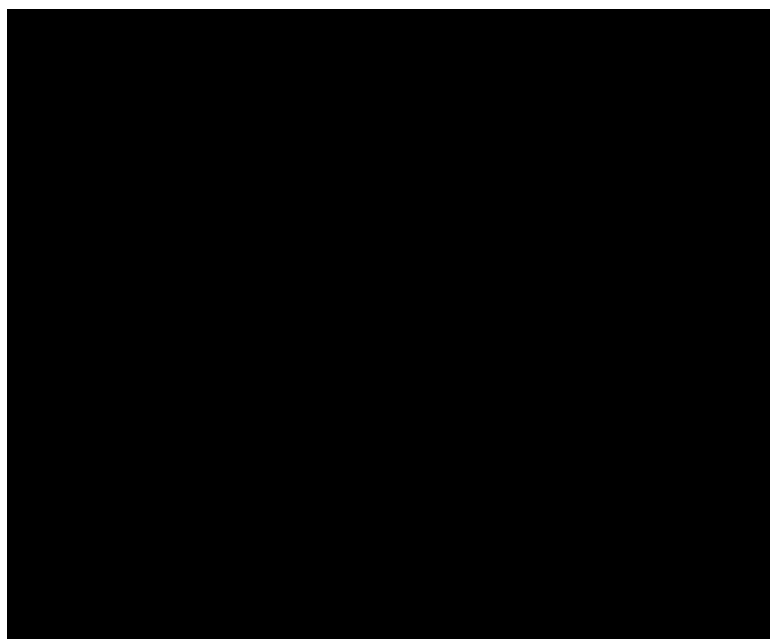
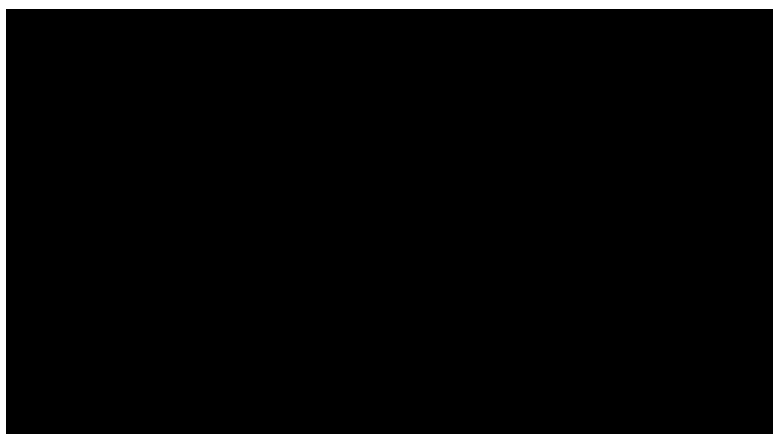


Table 2.
Isolated Yields for TMSOTf-Promoted Annulations



Scheme 3.
Iterative Pyran Annulation