

Proton exit from the heme–copper oxidase of *Escherichia coli*

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ABSTRACT Pathways of proton entry have been identified in the proton-translocating heme–copper oxidases, but the proton exit pathway is unknown. Here we report experiments with cytochrome *bo*₃ in *Escherichia coli* cells that may identify the beginning of the exit pathway. Systematic mutations of arginines 438 and 439 (R481 and R482 in the *E. coli* enzyme), numbering as in cytochrome *aa*₃ from bovine heart mitochondria, which interact with the ring D propionates of the two heme groups, reveal that the D propionate of the oxygen-binding heme is involved in proton pumping; its anionic form must be stabilized in order for proton translocation to occur. This may locate the beginning of the pathway by which pumped protons exit from the enzyme structure.

The structurally and functionally related respiratory heme–copper oxidases are redox-linked proton translocators (1–5) that convert free energy from the reduction of O₂ to an electrochemical proton gradient across mitochondrial or bacterial membranes, to be subsequently used for ATP synthesis by the membrane-bound H⁺-ATP synthase (6). The mechanism of proton translocation by this class of enzymes is intensively studied at present, and much of this research has recently been devoted to locating the pathways in the protein structure by which protons are transferred during translocation. Mutagenesis experiments with heme–copper oxidases from *Escherichia coli* (cytochrome *bo*₃; refs. 7 and 8), *Rhodobacter sphaeroides* (cytochrome *aa*₃; ref. 9) and *Paracoccus denitrificans* (cytochrome *aa*₃; ref. 10) have established that protons to be pumped are transferred into the protein via the so-called D-pathway, which has been identified in the crystal structures of cytochrome *aa*₃ from *P. denitrificans* and bovine heart mitochondria (refs. 11–13; Fig. 1). After this, the protons reach the invariant E242 residue[†] (14) in a hydrophobic cavity near the middle of the membrane, probably via bound water molecules (14–16). Little is known about proton conduction beyond E242, except that it may require isomerization of the side chain of this residue (15–17). However, because the D-pathway also conducts at least two of the four protons consumed in O₂ reduction to water at the binuclear heme–copper center, there must also be connectivity to this site from E242, again possibly via bound water molecules (refs. 15 and 16; see Fig. 1). The pathway of the pumped protons beyond E242 is not known, including the exit path to the opposite side of the membrane. Proton transfer via the D-channel to E242 is believed to be largely passive. Hence, a structural assignment of the beginning of the proton exit path may be valuable for the elucidation of the mechanism of translocation, which must include the structural and functional details of how this process is linked to the chemistry of O₂ reduction. It is conceivable that the “molecular machinery” responsible for this linkage should reside between the residue E242 and the beginning of the path by which protons exit the

membrane. Experiments reported here may help to define the latter.

MATERIALS AND METHODS

Bacterial growth conditions and purification of histidine-tagged cytochrome *bo*₃ enzymes from *E. coli* were as described previously (18). Site-directed mutagenesis was performed according to published methods (8) and was confirmed by DNA sequencing (ALFexpress DNA Sequencer, Pharmacia). This confirmation was also done for all mutant cells from fermentor cultivations as well as for the cells used in proton translocation measurements.

Ubiquinol oxidase (cytochrome *bo*₃) activity in the mutant and wild-type bacterial membranes and isolated enzymes were measured as described previously (7). Proton translocation in cells was determined by the oxygen pulse method (7, 8).

RESULTS AND DISCUSSION

A domain on the proton output side of the ring D propionates of the two heme groups (Fig. 1) contains several polar and charged residues and a cluster of bound water molecules, as revealed by the x-ray structures of the cytochrome *aa*₃ enzymes from *P. denitrificans* and bovine heart mitochondria (12, 13), as well as from calculations based on these structures (16, 19). This domain, which appears well connected protonically to the outside of the membrane, has indeed been implicated in proton exit (11, 13, 19), but functional support for this route is lacking because no mutations in this domain have been found to affect proton translocation specifically (7, 14, 20). Electrostatic calculations based on the crystal structures of the oxidized enzyme suggest that all four heme propionates are stabilized in their anionic state (21). The D-propionate of the oxygen-binding heme (heme *a*₃ or *o*₃) is stabilized by charge interactions with the conserved arginines 438 and 439, as well as by hydrogen bonds from R438, W126 (11, 12), and perhaps from a water molecule between this propionate and the Cu_B ligand H291 (ref. 13; Fig. 1). The D-propionate of the low-spin heme (heme *a* in *aa*₃ enzymes; heme *b* in *bo*₃ enzymes) is also stabilized by charge interactions with R438 and R439 (21), and by hydrogen bonds from the εNH group of R439 and the backbone NH of W126 (12, 13).

To explore a possible role in proton translocation of the conserved arginines and the two D-propionates with which they interact, we have systematically mutagenized the two arginines. In agreement with the report by Kawasaki *et al.* (22), changing the very well conserved R438 (R481 in cytochrome *bo*₃ from *E. coli* used here) to glutamine decreased enzyme turnover only a little. There was also no effect on proton translocation efficiency, as measured in multiturnover O₂-pulse experiments with *E. coli* cells (Table 1). However, changing R438 to either asparagine or leucine abolished

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[†]The amino acid numbering is for subunit I of cytochrome *aa*₃ from bovine heart mitochondria if not indicated otherwise.

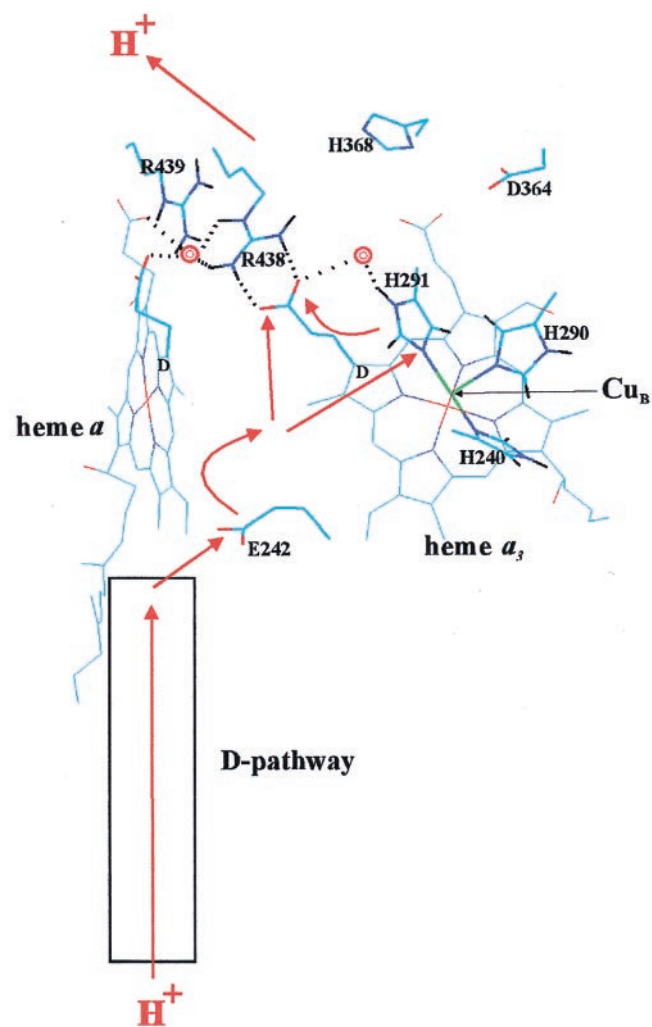


FIG. 1. Proton transfer in the heme-copper oxidases. The view is approximately in the membrane plane with the proton input and output sides of the membrane below and above, respectively. The figure is based on the x-ray coordinates of cytochrome aa_3 from *Paracoccus denitrificans* (ref. 13; Protein Data Bank identification code 1AR1), drawn by using the program HYPERCHEM 4.5 (Hypercube, Ontario, Canada). The D rings of hemes a and a_3 are marked and the D-propionate groups are highlighted. Cu_B in the foreground (green), and some key residues are both highlighted and marked. Two crystallographically visible water molecules (13) are shown as red double rings. Dotted lines indicate potential hydrogen bonds discussed in the text. The red arrows outline possible paths of pumped protons across the enzyme's membrane domain (see text).

proton translocation, although the turnover decreased only to $\approx 50\%$ (Table 1). In contrast, mutation of R438 to methionine caused almost complete loss of activity as well as major structural perturbations as indicated by altered optical heme spectra, which was not found with any of the other mutants studied here (not shown; cf. ref. 22).

On the other hand, while arginine-439 (R482 in cytochrome bo_3) could be changed to glutamine, asparagine, or even leucine without loss of proton translocation (Table 1), changing both arginines to glutamines abolished proton translocation even though either single mutation had no effect. Enzyme turnover decreased to $\approx 40\%$ in this double mutant—i.e., only slightly more than for each single mutation (Table 1).

Inspection of the structure suggests that the amide group of a glutamine in place of R438 (R438Q) could still maintain hydrogen bonding to the D-propionate of the oxygen-binding heme (heme a_3 in cytochrome aa_3 ; heme o_3 in cytochrome bo_3) without much change in the position of the propionate, but this

Table 1. Range of cytochrome bo_3 enzyme activity in wild-type (wt) and mutant *E. coli* membranes (m) and isolated enzyme (i), and proton translocation efficiency in cells

Strain (Cyt bo_3 numbering)	Enzyme location	Activity, % of wt	H^+/e^-
Wild type	m, i	100	1.6–2.2
R481Q	m	70–80	1.6–2.0
	i	54–64	
R481M	i	2–8	—
R481N	i	51–65	1.0–1.3
R481L	i	35–45	1.0–1.3
R482Q	m	75–95	1.7–2.1
	i	46–60	
R482N	m	48–58	1.7–2.2
	i	33–39	
R482L	m	75–85	1.7–2.0
R481Q/R482Q	i	35–45	0.9–1.2

Wild-type activity range: 400–750 (m) and 350–560 (i) electrons per sec per bo_3 . Methods for determining proton translocation and enzyme activity and for mutagenesis are described in refs. 7, 8, and 14. In the quinol oxidase cytochrome bo_3 1 H^+/e^- is released to the outside of the cells due to the oxidation of hydroquinone.

is not possible with asparagine (R438N) or leucine (R438L). The loss of proton translocation in the latter two mutants suggests that the R438/heme o_3 (a_3) D-propionate pair is critical for the proton-translocating mechanism. However, because proton translocation efficiency is normal in the R438Q mutant we conclude that this function does not absolutely demand a protonatable or a positively charged side chain in the 438 site. The loss of proton translocation may therefore be ascribed to changes induced in the properties of the D-propionate of heme o_3 (a_3) with which R438 interacts most strongly.

The anionic form of the D-propionate of heme o_3 (a_3) may be stabilized by ≈ 11.0 and 6.5 p K_a units because of charge interactions with R438 and R439, respectively (21). Further stabilization by hydrogen bonding from R438 may amount to ≈ 5 p K_a units. In the R438Q, N, and L mutants the stronger charge interaction is lost, but hydrogen bonding uniquely remains in R438Q. Interestingly, it seems that in this case this subtle difference determines whether proton translocation occurs. In the R439 mutants only the weaker charge interaction with the heme o_3 (a_3) D-propionate is lost and proton translocation remains. However, in the double mutant both charge interactions are abolished and destabilization of this propionate is expected to be considerable. Now proton pumping is lost despite the fact that the hydrogen bond from the 438 locus may be retained.

These predicted effects of the mutations on the heme o_3 (a_3) D-propionate are summarized in Table 2 and compared with the data on proton translocation. The listed interactions show only the general trend (21), and do not, for example, include possible compensatory effects that might occur in a mutant. At any rate, they suggest that loss of proton translocation correlates with the extent of destabilization of the anionic form of the heme o_3 (a_3) D-propionate. Such destabilization would favor the uncharged propionic acid state, which may be the cause of the observed decoupling of proton translocation from the chemistry of O_2 reduction.

The simplest rationale is that the D-propionate of heme o_3 (a_3) normally functions as a proton acceptor in a crucial step of the proton translocation mechanism, which would define the beginning of the exit path of pumped protons. However, our results do not exclude that proton transfer may also involve the neighboring D-propionate of the low-spin heme, which appears to be more strongly stabilized than the D-propionate of heme o_3 (a_3) by charge interactions with the arginines (21), and may thus not be sufficiently perturbed by the single arginine

Table 2. Predicted approximate stabilization of the anionic state of the D-propionate group of heme o_3 (a_3) by its charge interactions (21) with R438 (R481 in cytochrome bo_3) and R439 (R482 in bo_3) and its hydrogen bond interaction with R438 (or Q438)

Strain (Cyt bo_3 numbering)	Predicted stabilization, pK _a units				Proton translocation (see Table 1)
	Charge interaction		Hydrogen bond	Sum	
	With R481	With R482			
Wild type	11	6.5	5	22.5	Yes
R481Q	0	6.5	5	11.5	Yes
R481N, L	0	6.5	0	6.5	No
R482Q, N, L	11	0	5	16	Yes
R481Q/R482Q	0	0	5	5	No

mutations. In this connection it may be of special interest to note that a water molecule which lies within hydrogen-bonding distance from both propionates of the low-spin heme b (a), is also within hydrogen-bonding distance from the N^e and NH₂ nitrogens of R438 (ref. 13; Fig. 1). Since the latter is a hydrogen bond donor to the D-propionate of heme o_3 (a_3), both propionates of the low-spin heme are also at least potentially connected to the pathway of proton exit.

The D-propionate of heme o_3 (a_3) may accept protons from E242 either directly (11), via bound water molecules (16), or more indirectly via water molecules and the copper ligand H291 (Fig. 1) as proposed in a recent version (19) of the histidine cycle model of proton translocation (23). At any rate, our results suggest that a key problem of the proton translocation mechanism is now to elucidate how proton transfer is accomplished from E242 to the D-propionate of heme o_3 (a_3) in a way that is coupled to the oxygen reduction chemistry at the binuclear heme–copper center.

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