



Characterization of two *Bunodosoma granulifera* toxins active on cardiac sodium channels

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1 Two sodium channel toxins, *BgII* and *BgIII*, have been isolated and purified from the sea anemone *Bunodosoma granulifera*. Combining different techniques, we have investigated the electrophysiological properties of these toxins.

2 We examined the effect of *BgII* and *BgIII* on rat ventricular strips. These toxins prolong action potentials with EC_{50} values of 60 and 660 nM and modify the resting potentials.

3 The effect on Na^+ currents in rat cardiomyocytes was studied using the patch-clamp technique. *BgII* and *BgIII* slow the rapid inactivation process and increase the current density with EC_{50} values of 58 and 78 nM, respectively.

4 On the cloned hH1 cardiac Na^+ channel expressed in *Xenopus laevis* oocytes, *BgII* and *BgIII* slow the inactivation process of Na^+ currents (respective EC_{50} values of 0.38 and 7.8 μ M), shift the steady-state activation and inactivation parameters to more positive potentials and the reversal potential to more negative potentials.

5 The amino acid sequences of these toxins are almost identical except for an asparagine at position 16 in *BgII* which is replaced by an aspartic acid in *BgIII*. In all experiments, *BgII* was more potent than *BgIII* suggesting that this conservative residue is important for the toxicity of sea anemone toxins.

6 We conclude that *BgII* and *BgIII*, generally known as neurotoxins, are also cardiotoxic and combine the classical effects of sea anemone Na^+ channels toxins (slowing of inactivation kinetics, shift of steady-state activation and inactivation parameters) with a striking decrease on the ionic selectivity of Na^+ channels.

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Abbreviations: ApA and ApB, *Anthopleura xanthogrammica* toxins A and B; ATX I and ATX II, *Anemonia sulcata* toxins I and II, *BgII* and *BgIII*, *Bunodosoma granulifera* toxins II and III; HEPES, 2-[4-(2-hydroxyethyl)-1-piperazinyl] ethanesulphonic acid; Sh I, *Stygodactyla helianthus* toxin I; Tris, Tris(hydroxymethyl)aminomethane; TTX, Tetrodotoxin; VGSC, Voltage-Gated Sodium Channel

Introduction

As key elements of signal transduction, voltage-gated sodium channels (VGSCs) are the target of toxins of various origins and chemical structures. The different effects of these toxins range from blocking the pore of the channel (e.g. tetrodotoxin, μ -conotoxin) to modifying its gating and permeation characteristics (e.g. sea anemone toxins, scorpions α - and β -toxins, batrachotoxin). These natural compounds are powerful tools for understanding the physiological contribution of VGSCs to cell and organ behaviour and for probing and correlating ion channel structure and function (Catterall, 1995; 2000; Cestele & Catterall, 2000; Denac *et al.*, 2000; Lazdunski *et al.*, 1986; Renaud *et al.*, 1986). Moreover, elucidation of the mechanisms of action of toxins, knowledge of their 3D-structures and the discovery of common scaffolds

between toxins open wide perspectives in designing various drugs (Ménez, 1998).

VGSCs are composed of a pore-forming α subunit of approximately 260 kDa flanked by two auxiliary β subunits, a non covalently associated β_1 subunit of 36 kDa and a β_2 subunit of 33 kDa linked by a disulphide bond. In heterologous systems, expression of the α subunit alone is sufficient to form a functional Na^+ channel regulated by voltage. Membrane depolarization causes a voltage-dependent conformational change of the channel that induces an increase of its permeability to Na^+ . This is followed by an inactivation process, wherein the channel closes and the permeability to Na^+ is shut off (Hille, 1992). The β subunits, which do not form the pore, modulate the level of expression, channel gating properties, and interaction with cytoskeleton proteins (Catterall, 2000). Sodium channels can be differentiated by their primary structure, current kinetics and relative sensitivity to the neurotoxin tetrodotoxin (TTX). They have been cloned in

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different tissues of various species. The human cardiac isoform (hH1) of the VSGC (Gellens *et al.*, 1992), or Na_v 1.5 according to the recent nomenclature of VGSCs (Goldin *et al.*, 2000), is a TTX-resistant channel expressed in cardiomyocytes where it plays a key role in the generation and propagation of the cardiac action potential (Balsler, 1999).

Sea anemones produce several polypeptide toxins, mainly active on ionic channels of excitable membranes, to capture their prey. Sea anemone toxins acting on Na⁺ channel have been intensively studied over the past years. They are both cardio- and neurotoxins and their main effects are to slow the inactivation process of Na⁺ currents and to prolong the action potentials. These polypeptides cross-linked by three disulphide bridges and of molecular masses of approximately 5 kDa have been broadly categorized as type 1 (e.g. ATXI and ATXII from *Anemonia sulcata* (Wunderer & Eulitz, 1978; Wunderer *et al.*, 1976), ApA and ApB from *Anthopleura xanthogrammica* (Norton, 1981; Reimer *et al.*, 1985; Tanaka *et al.*, 1977)) and type 2 (e.g. ShI from *Stichodactyla helianthus* (Kem *et al.*, 1989; Wilcox *et al.*, 1993)), according to amino acid sequence similarity and immunological cross-reactivity (Figure 2; see (Norton, 1991; Norton, 1998) for reviews). Toxins acting on Na⁺ channels bind to at least six receptor sites (Catterall, 2000; Denac *et al.*, 2000). Sea anemone toxins bind to receptor site 3, located in the extracellular linker between segments S3 and S4, in the fourth domain (D4). They share their receptor site with scorpion α -toxins (Gordon *et al.*, 1998; Possani *et al.*, 1999; Rogers *et al.*, 1996) and funnel web spider toxins (Nicholson *et al.*, 1994; 1998). The common point between these toxins is that they induce a slowing of the inactivation process of Na⁺

channels. From a mechanical point of view, the Na⁺ channel inactivation process derives mainly from the voltage-dependent coupling of activation driven by the transmembrane movement of the voltage sensor (S4 segments). By binding to site 3, toxins prevent the normal gating movements of the IVS4 segment and thus uncouple Na⁺ channel activation from inactivation (Catterall, 2000).

Bunodosoma granulifera is a common anemone of Cuban sea shores. Several active compounds with different pharmacological actions have been isolated from its secretions (Aneiros *et al.*, 1993; Loret *et al.*, 1994; Salinas *et al.*, 1997). BgII and BgIII are two toxins that have been purified and sequenced from this sea anemone originally by Loret *et al.* (1994). They consist in two peptides of 48 amino acids each containing six cysteine residues that form three disulphide bridges. The amino acid sequences of these two toxins possess a higher similarity with type 1 sea anemone toxins, like ApA and ApB or ATX II, than with the toxins of type 2, like ShI (Figure 2). The sequences of BgII and BgIII are almost identical, only differing by a single amino acid. In BgIII, at position 16, an aspartic acid replaces the asparagine of BgII (Figure 2). Despite their resemblance, the toxicity and the binding to rat brain synaptosomes of BgII is interestingly significantly higher than that of BgIII. However, up to now, electrophysiological characterization of these two toxins was still completely lacking.

In the present study, combining different approaches, we have therefore investigated the electrophysiological properties of BgII and BgIII toxins. Firstly, we have examined their effect on rat ventricular strips and observed that both toxins are able to prolong action potentials. Then, using the patch-clamp technique, we have studied the effect of BgII and BgIII on Na⁺ currents in cardiomyocytes revealing that these toxins slowed the rapid inactivation process of Na⁺ currents and increased the current density. Finally, in order to refine our study, we have investigated the influence of these toxins on the cloned cardiac sodium channel hH1 expressed in *Xenopus laevis* oocytes, using the two-electrode voltage clamp technique. These experiments confirmed that BgII and BgIII markedly slow the inactivation process of Na⁺ currents in a dose-dependent way and showed that both steady-state inactivation and activation are shifted to more positive potentials. It was also revealed that toxin-modified channels displayed a decreased ionic selectivity. Moreover, effects of BgII are more marked than effects of BgIII emphasizing the importance of the amino acid in position 16 for the biological effects of these two *Bunodosoma granulifera* toxins.

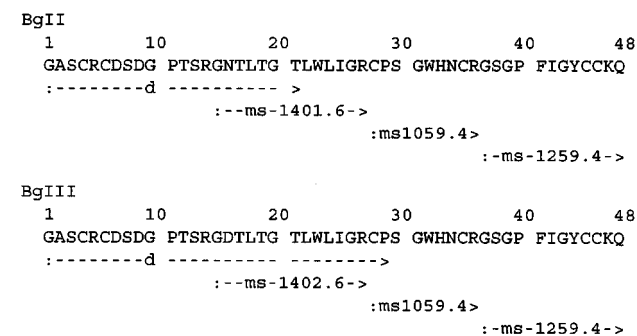


Figure 1 Amino acid sequence determination of toxins. BgII: Direct sequencing by Edman degradation of reduced and alkylated toxin provided unequivocal identification of the 21 first amino acids, shown by underlining (-d->). Three additional fragments were obtained by HPLC separation of toxin BgII digested with Arg-C endopeptidase, which gave peptides with molecular masses of 1401.6, 1059.4 and 1259.4, respectively. The sequences were obtained by ionization, using MSIMS mass spectrometry, and correspond to sequence from positions 15 to 27, 28 to 36 and 37 to 48. The entire sequence corresponds exactly to that reported earlier (Loret *et al.*, 1994). BgIII: Direct sequencing of reduced and alkylated toxin provided the 29 first amino acid residues, as indicated by the underlining (-d->). Three additional overlapping peptides were obtained by Arg-C endopeptidase digestion conducted as mentioned above for BgII, in which the only difference was found in the peptide corresponding to positions 15 to 27 (molecular mass 1402.6) where an aspartic acid substitutes an asparagine in position 16. The MSIMS spectrometry analysis confirmed the sequences shown, which also corresponds exactly to those earlier reported (Loret *et al.*, 1994). The numbers on top of the sequences indicate the positions of the amino acids in the sequences.

	1	10	20	30	40	50	%id.																																										
BgII	G	A	S	C	R	C	D	S	D	G	P	T	S	R	G	N	T	L	T	G	T	L	W	L	I	G	R	C	P	S	G	W	H	N	C	R	G	S	G	P	F	I	G	Y	C	C	K	Q	---
BgIII	G	A	S	C	R	C	D	S	D	G	P	T	S	R	G	D	T	L	T	G	T	L	W	L	I	G	R	C	P	S	G	W	H	N	C	R	G	S	G	P	F	I	G	Y	C	C	K	Q	---
ApA	G	V	S	C	R	C	D	S	D	G	P	S	V	R	G	N	T	L	S	C	T	L	W	L	I	F	P	S	C	P	S	G	W	H	N	C	K	A	G	S	P	I	G	W	C	K	---		
ApB	G	V	P	L	C	D	S	D	G	P	R	P	R	G	N	T	L	S	C	T	L	W	L	I	F	P	S	C	P	S	G	W	H	N	C	K	A	G	S	P	I	G	W	C	K	---			
ATX II	G	V	P	L	C	D	S	D	G	P	S	V	R	G	N	T	L	S	C	T	L	W	L	I	F	P	S	C	P	S	G	W	H	N	C	K	A	G	S	P	I	G	W	C	K	---			
ShI	A	A	C	K	C	D	E	G	E	D	I	R	T	A	L	T	C	T	V	L	G	S	-	C	N	A	G	W	K	C	A	S	Y	T	I	L	A	D	C	R	K	K	K	K	K	---			

Figure 2 Comparison of the amino acid sequences of BgII, BgIII with other sea anemone toxins. BgII and BgIII have been purified from the sea anemone *Bunodosoma granulifera* (Loret *et al.*, 1994), ATXII from *Anemonia sulcata* (Wunderer *et al.*, 1976), ApA and ApB from *Anthopleura xanthogrammica* (Norton, 1981; Reimer *et al.*, 1985; Tanaka *et al.*, 1977) and ShI from *Stichodactyla helianthus* (Kem *et al.*, 1989; Wilcox *et al.*, 1993). Identical amino acids are indicated with a black background, homologous amino acids are indicated with a gray background and '%id' stands for the percentage of identity in comparison to BgII. Dashes represent gaps.

Methods

Toxin extraction and isolation

Sea anemone *Bunodosoma granulifera* was collected along the north coast of Havana province, Cuba and transported to the laboratory in containers filled with sea water.

To prepare crude extract, in brief, 1 kg of *B. granulifera* was homogenized in 1 l of ethanol adjusted to pH 5.4 with acetic acid. After centrifugation (30 min, 4°C, 15,000 × g), the supernatant was separated and precipitated with 10 volumes of acetone (4°C, overnight). The precipitate was recovered, diluted in water, concentrated under reduced pressure to eliminate acetone and freeze dried later. Lyophilized crude extract was dissolved in 0.10 M ammonium acetate pH 6.7 and gel filtered on Sephadex G-50 m (4.5 × 132 cm) as previously described (Aneiros *et al.*, 1993; Salinas *et al.*, 1997).

The toxic peak was submitted to chromatography on the cation-exchanger Fractogel EMD SO₃-650 M (1.2 × 32 cm) eluted with a linear gradient of ammonium acetate (pH 5.4) from 0.01 to 1.0 M. Toxins were rechromatographed on Fractogel EMD SO₃-650 M under analogous conditions, and desalted on Sephadex G-25. Further purification of toxins was performed by reversed-phase HPLC.

We used the toxicity test in crab to follow the toxic activity of chromatographic fractions during the purification procedures. The toxicity of the different fractions was evaluated as paralysing activity after injection on shore crabs (*Uca thayeri* and *Carcinus maenas*).

Mass spectrometry analysis

The masses and sequences were obtained using a Finnigan LCQ-Duo ion trap mass spectrometer, equipped with an electrospray ion source. The analysis was performed as recently reviewed (Dongre *et al.*, 1997).

Right ventricular strips

Small right ventricular strips were dissected out from adult rat hearts fixed to a bath chamber and continuously superfused (10 ml min⁻¹) with normal Tyrode solution (NaCl, 140; KCl, 2.5; CaCl₂, 2; MgCl₂, 0.5; hydroxyethylpiperazine-N'-2-ethanesulfonic acid (HEPES), 10 and glucose, 5 (pH = 7.4, gassed with O₂)) at 35°C and field stimulated with 2 ms pulses at a frequency of 1.25 Hz (cycle length, 800 ms). Standard fine-tipped microelectrodes (tip resistance, 15 to 20 MΩ) were used to record action potentials (Alvarez *et al.*, 1981). Action potential duration was measured at 0 (D₀) and -60 (D₋₆₀) mV.

Isolation of adult ventricular cardiomyocytes

Single rat ventricular cells were dispersed by a collagenase/trypsin enzymatic method similar to that previously described (Galan *et al.*, 1998). In brief, rat hearts were cannulated through the aorta and perfused with a physiological solution for 5 min (in mM): NaCl 112.0, KCl 2.5, CaCl₂ 1.8, MgCl₂ 0.6, HEPES 10.0, glucose 5.0, Na₂-pyruvate 5.0, and pH adjusted to 7.4. Thereafter, the hearts were perfused with a low-Ca²⁺ (CaCl₂ 30 μM) solution for 5 min. The hearts were

then perfused for 30 min with the same solution containing collagenase (Boehringer Mannheim) and trypsin (Merck) at 1.5 mg ml⁻¹ and 0.4 mg ml⁻¹ respectively and supplemented with minimum essential medium (1 μl ml⁻¹ MEM; Sigma), creatine (5 mM) and taurine (20 mM). At the end of this period, the right ventricle was cut off and gently shaken in the same solution without enzymes. Isolated myocytes were kept in this physiological solution (Ca²⁺ = 1 mM) at room temperature (21–23°C) and used within 6–8 h.

Patch-clamp recordings

The 'whole-cell' variant of the patch-clamp method was used (Hamill *et al.*, 1981). Recordings were made using an RK-300 (Biologic, Claix, France) patch-clamp amplifier. Patch pipettes were made from borosilicate glass and had tip resistances between 1.0 and 1.2 MΩ. The pipette 'intracellular' solution contained (mM): CsCl 100, TEA-Cl 20, MgCl₂ 4.0, Na₂-GTP 0.5, Na₂-ATP 3, ethyleneglycol-bis(-aminoethyl ether) N,N,N',N'-tetraacetic acid (EGTA) 10, HEPES 10, the pH was adjusted to 7.2 with CsOH. For recording the Na⁺ current (I_{Na}), Ca²⁺ and K⁺ currents were blocked by CoCl₂ (extracellular, 3 mM), tetraethylammonium chloride (TEA-Cl) and CsCl (intracellular and extracellular; see below), respectively. A cell aliquot was put in a Petri dish containing the control solution (mM): NaCl 112, KCl 2.5, CaCl₂ 1.8, MgCl₂ 0.6, HEPES 10, glucose 5.0, Na₂-pyruvate 5.0; pH was adjusted to 7.4 with NaOH. Experiments were performed at room temperature (22 ± 2°C).

After achieving the whole-cell patch-clamp configuration, each cell was exposed to different extracellular solutions by positioning it at the extremity of one of six capillaries (inner diameter of each capillary was 250 μm). Such a system allowed rapid changes of solution (<2 s). Currents were routinely evoked by 50 ms voltage-clamp pulses to -40 mV from a holding potential of -100 mV at a frequency of 0.25 Hz (cycle length 4 s). The current amplitude was estimated as the difference between the peak inward current and the current level at the end of a 50 ms pulse. The composition of the extracellular solution in these experiments was (in mM): NaCl 5, TEA-Cl 110, CaCl₂ 1.8, MgCl₂ 2, glucose 10, HEPES 10, pH was adjusted to 7.4 at 21°C. Pulse generation and data acquisition were done, using computer facilities and ACQUIS1 (CNRS License, France) software. Currents were filtered at 3 KHz and digitized at 20 KHz with a LabMaster DMA (model TM 125, Scientific Solutions, U.S.A.).

Current to voltage (I-V) relationships and availability curves were constructed using a standard double pulse voltage protocol. From a holding potential of -100 mV, a 50 ms test pulse to -40 mV was preceded by 50 ms prepulses to various membrane potentials. The time interval between pulses was 5 ms. I-V relationships were constructed from currents elicited by prepulse potentials. Normalization of current amplitudes at a given test pulse by the maximal current recorded as a function of prepulse potential gave the availability curve. The experimental points were fitted to a Boltzmann function:

$$I/I_{\max} = (1 + \exp((V_p - V_{1/2})/s))^{-1} \quad (1)$$

where V_p is the prepulse potential, V_{1/2} is the potential for half-availability and s a slope factor.

The inactivation time course of currents was described by fitting the current traces between the inward peak and the end of the pulse using the fitting procedures of ACQUIS1 software. Recovery from inactivation was studied by a two pulse protocol. From a holding potential of -100 mV, a pair of 50 ms pulses to -40 mV was applied; the interval between the two pulses was varied from 5 to 650 ms. Recovery from inactivation (or reactivation) was obtained by the ratio of current measured at the second pulse to the current measured at the first pulse as a function of the time interval. Experimentally obtained means of the availability curves were fitted to a Boltzmann function to obtain the potential for half inactivation and slope factors.

Values are expressed as means \pm s.e.means.

Expression in *Xenopus oocytes*

For the expression in *Xenopus oocytes*, the hH1 gene was subcloned into pSP64T (Gellens *et al.*, 1992). For *in vitro* transcription, hH1/pSP64T has been first linearized by *Xba*I. Then, capped cRNAs were synthesized from the linearized plasmid using the large-scale SP6 mMMESSAGE-mMACHINE transcription kit (Ambion, U.S.A.). The *in vitro* synthesis of cRNA encoding hH1 and isolation of *Xenopus oocytes* was as previously described (Liman *et al.*, 1992). Oocytes were injected with 50 nl of hH1 cRNA at a concentration of 1 ng nl⁻¹ using a Drummond microinjector (U.S.A.). The solution used for incubating the oocytes contained (in mM): NaCl 96, KCl 2, CaCl₂ 1.8, MgCl₂ 2 and HEPES 5 (pH 7.4), supplemented with 50 mg l⁻¹ gentamicin sulphate.

Electrophysiological recordings in *Xenopus oocytes*

Whole-cell currents from oocytes were recorded from 1 to 3 days after injection using the two-electrode voltage clamp technique. Voltage and current electrodes were filled with 3 M KCl. Resistances of both electrodes were kept as low as possible (*ca.* 0.1–0.2 M Ω). During the experiments, the external NaCl concentration has been reduced in order to facilitate the clamp. Bath solution composition was (in mM): NaCl 40, KCl 2, CaCl₂ 1.8, MgCl₂ 2 and Tris-HCl 60 (pH 7.4). Experiments were performed using a GeneClamp 500 amplifier (Axon instruments, U.S.A.) controlled by a pClamp data acquisition system (Axon instruments, U.S.A.). Using a four-pole low-pass Bessel filter, currents were filtered at 5 kHz and sampled at 10 kHz. Digital leak subtraction of the current records was carried out using a P/4 protocol.

Current to voltage (*I*–*V*) relationships and steady-state activation curves were constructed using a standard single pulse voltage protocol. From a fixed holding potential of -90 mV, 25 ms test pulses to membrane potentials varying from -70 to $+45$ mV by 5 mV increments were performed. *I*–*V* relationships were constructed from peak currents elicited at different potentials. Normalization of conductance (g/g_{\max}) as a function of the test pulse voltage gave the steady-state activation curve. The experimental points were fitted to a Boltzmann function:

$$g/g_{\max} = (1 + \exp((V_p - V_{1/2act})/s))^{-1} \quad (2)$$

where V_p is the pulse potential, $V_{1/2act}$ is the potential for half-activation and s a slope factor. Steady-state inactivation

curves were constructed using a standard double pulse voltage protocol. From a fixed holding potential of -90 mV, a 50 ms test pulse to -20 mV was preceded by 50 ms prepulses to membrane potentials varying from -120 to 30 mV.

Normalization of the current amplitude at the test pulse by the maximal current recorded (I/I_{\max}) as a function of prepulse potential gave the steady-state inactivation curve. The experimental points were fitted to a Boltzmann function:

$$I/I_{\max} = (1 + \exp((V_p - V_{1/2inact})/s))^{-1} \quad (3)$$

where V_p is the prepulse potential, $V_{1/2inact}$ is the potential for half-inactivation and s a slope factor. The permeability ratio of Na⁺ as compared to K⁺ (P_{Na}/P_K) has been calculated using the Goldman-Hodgkin-Katz (GHK) equation:

$$E_{rev} = (RT/zF) \ln ((P_{Na}[Na]_o + P_K[K]_o)/(P_{Na}[Na]_i + P_K[K]_i)) \quad (4)$$

where E_{rev} is the reversal potential determined from the *I*–*V* relationship, P_{Na} and P_K are the permeabilities of the channel for Na⁺ and K⁺, respectively. The external concentrations of Na⁺, $[Na]_o$, and K⁺, $[K]_o$, were 40 and 2 mM, respectively. The internal concentrations of Na⁺, $[Na]_i$, and K⁺, $[K]_i$, have been estimated to be 6 and 92 mM, respectively (Dascal, 1987).

The inactivation time course of Na⁺ currents was described by fitting the current traces between the inward peak and the end of the pulse using the fitting procedures of pClamp6 software (Axon instruments, U.S.A.).

Data are presented as mean \pm s.e.mean. All experiments were performed at room temperature ($20 \pm 2^\circ\text{C}$).

Results

Toxin identification

After gel filtration chromatography on Sephadex G-50, only fraction three eluting between 70–80% of one whole column volume was found to be toxic in the crab bioassay. This fraction was submitted to a cation exchange chromatography on Fractogel EMD SO₃-650M. The fractions eluting in the gradient at ammonium acetate concentrations between 0.1 and 0.2 M were further chromatographed under the same conditions to give the toxins (*Bg*III and *Bg*II respectively) that were earlier reported by Loret *et al.* (1994).

For *Bg*II toxin, a direct sequencing by Edman degradation of reduced and alkylated toxin provided an unequivocal identification of the 21 first amino acids. Then, following a digestion of this toxin with Arg-C endopeptidase a HPLC separation gave three additional fragments with molecular masses of 1401.6, 1059.4 and 1259.4, respectively. The sequences were obtained by ionization, using MS/MS mass spectrometry, and correspond to sequences from position 15 to 27, 28 to 36 and 37 to 48. The entire sequence corresponds exactly to that earlier reported by Loret *et al.*, 1994 (Figure 1).

For *Bg*III toxin, a direct sequencing by Edman degradation of reduced and alkylated toxin provided an unequivocal identification of the 29 first amino acids. Then, overlapping

peptides were obtained by Arg-C endopeptidase digestion conducted as mentioned above for BgII, in which the only difference was found in the peptide corresponding to positions 15 to 27 (molecular mass 1402.6) where an aspartic acid substitutes an asparagine in position 16. The MS/MS spectrometry analysis confirmed the sequences shown, which also corresponds exactly to those earlier reported by Loret *et al.*, 1994 (Figure 1).

BgII and BgIII consist in two peptides of 48 amino acids which each contain six cysteine residues forming three disulphide bridges. The measured masses for BgII and BgIII, 5072.7 Da and 5073.1 Da, respectively, are almost identical to the theoretical masses determined according to the amino acid sequences, 5071.7 Da and 5072.7 Da for BgII and BgIII, respectively. In Figure 2, amino acid sequences of BgII and BgIII are shown and compared with other sea anemone Na⁺ channel toxins: ATXII from *Anemonia sulcata* (Wunderer *et al.*, 1976), ApA and ApB from *Anthopleura xanthogrammica* (Norton, 1981; Reimer *et al.*, 1985; Tanaka *et al.*, 1977) and ShI from *Stichodactyla helianthus* (Kem *et al.*, 1989; Wilcox *et al.*, 1993). The amino acid sequence of BgII possesses a higher similarity with type 1 sea anemone toxins (72.9%, 64.5% and 70.8% identity with ApA, ApB and ATX II, respectively) than with the toxins of type 2 (42.6% identity with ShI).

Effects of *Bunodosoma granulifera* toxins on action potentials of rat ventricular strip

Superfusion of rat ventricular strips with 10 μ M BgII or BgIII toxins induced a fast increase in action potential duration (Figure 3). Within 1 min, action potential duration at 0 mV (D_0) was increased to 160 and 130% of control in the presence of BgII and BgIII, respectively. Action potential duration at -60 mV (D_{-60}) was increased to 230 and 185% in the presence of BgII and BgIII, respectively. During this initial period, no major changes were observed in action potential amplitude or resting membrane potential. At longer superfusion times (5 min), action potential duration was further increased and the resting potential decreased. Interestingly, these changes were more marked with BgII which increased D_0 to 560% of control and decreased resting potential from -92 to -45 mV. Under the action of BgIII, D_0 was only further increased to 216% of control while the resting potential was decreased from -93 to -87 mV.

The increase of the action potential duration induced by BgII and BgIII is dose-dependent. The effect of BgII and BgIII has been studied at concentrations ranging from 0.01 to 10 μ M and revealed that BgII is more potent than BgIII. The toxin concentration producing the half maximum effect (EC_{50}) on the action potential duration at 0 mV determined by a sigmoidal fit is 60 ± 4 nM for BgII ($n=3$) and 663 ± 380 nM for BgIII ($n=3$).

Effects of BgII and BgIII on Na⁺ current of rat ventricular cardiomyocytes

To study the effect of *Bunodosoma granulifera* toxins on the Na⁺ current (I_{Na}), rat ventricular myocytes were superfused with an extracellular solution containing 5 mM NaCl so as to decrease I_{Na} amplitude and to obtain a better voltage-clamp

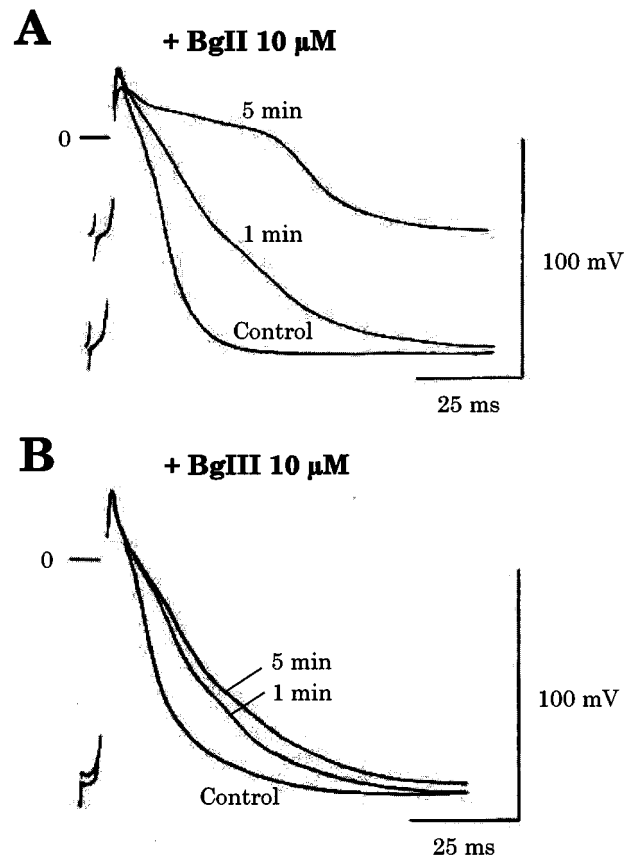


Figure 3 Effects of *Bunodosoma granulifera* toxins on the resting and action potential characteristics of rat ventricular strips. Both BgII (A) and BgIII (B) were applied at a concentration of 10 μ M. Within 1 min, both toxins markedly increased the action potential duration. After a longer time (5 min) the resting potential was decreased. Note that the increase in action potential duration and the decrease in resting potential were more marked with BgII.

Table 1 Effects of BgII and BgIII on I_{Na} of rat ventricular cardiomyocytes (values are expressed as per cent increase over control value)

BgII molar concentration	10^{-9}	10^{-8}	10^{-7}	10^{-6}	10^{-5}
I_{Na} density	0	15 ± 10	$100 \pm 8^*$	$152 \pm 17^*$	$144 \pm 12^*$
$T_{50\%}$	0	10 ± 8	$273 \pm 5^*$	$246 \pm 6^*$	$301 \pm 10^*$
Number of cells	5	5	6	7	6
BgIII molar concentration	10^{-9}	10^{-8}	10^{-7}	10^{-6}	10^{-5}
I_{Na} density	0	9 ± 8	$92 \pm 9^*$	$167 \pm 15^*$	$127 \pm 10^*$
$T_{50\%}$	0	22 ± 9	$261 \pm 12^*$	$234 \pm 10^*$	$281 \pm 12^*$
Number of cells	4	5	5	6	5

* $P < 0.05$ respect to control; $T_{50\%}$: time for half inactivation of I_{Na} (I_{Na} was evoked at -40 mV from a holding potential of -100 mV)

control. Myocytes were held at -100 mV and routinely depolarized to -40 mV with 50 ms voltage-clamp pulses to monitor I_{Na} . Membrane capacity was determined in voltage-clamped cells (Galan *et al.*, 1998) and the mean value obtained from 27 cells was 124.5 ± 18.3 pF. Ionic current amplitudes were normalized to membrane capacity and are expressed as current density (pA/pF). Under these experi-

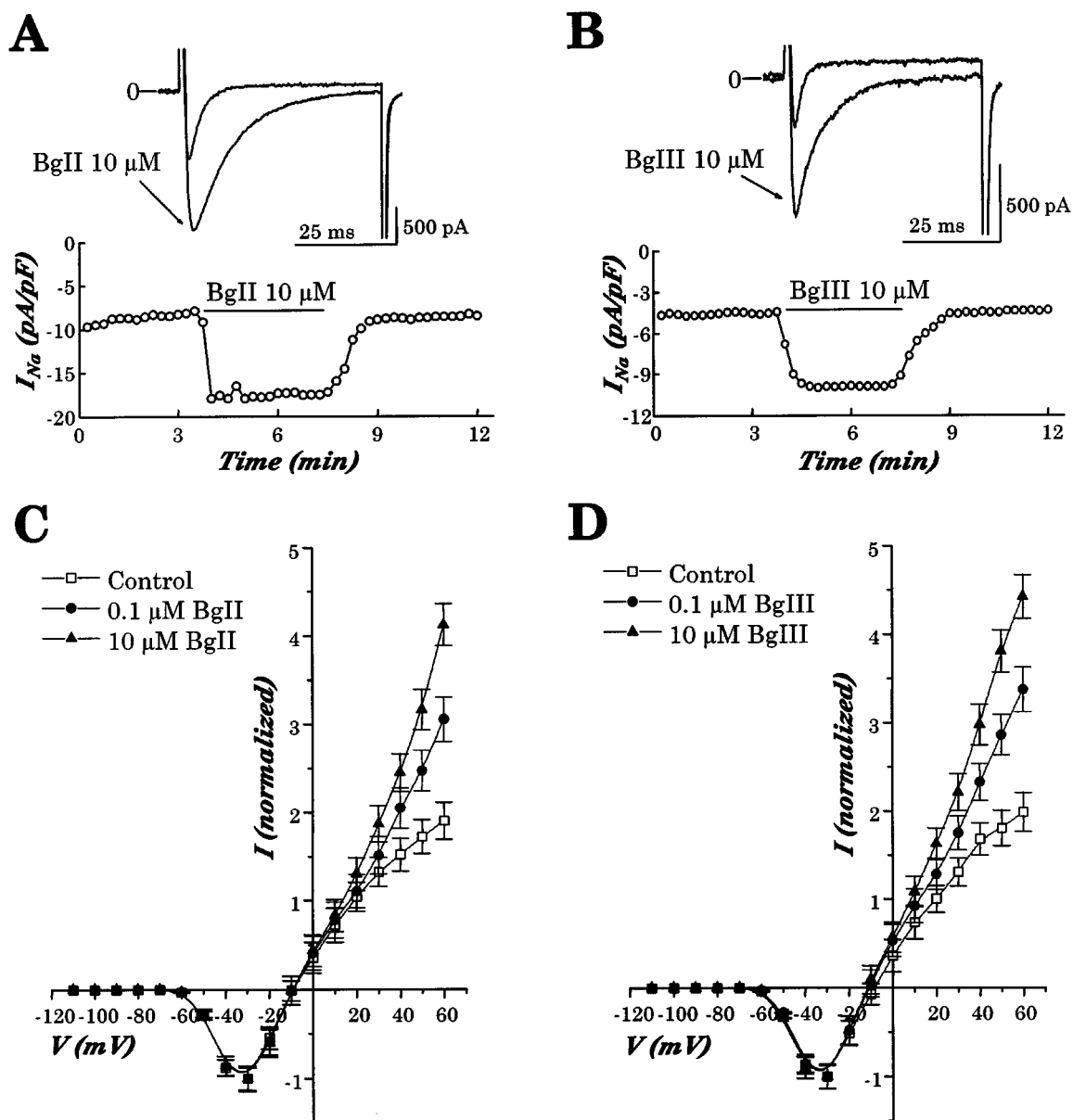


Figure 4 Effect of *BgII* and *BgIII* toxins on Na⁺ currents in rat ventricular cardiomyocytes. (A,B) Time course of the effects of 10 μ M *BgII* or 10 μ M *BgIII* on I_{Na} of a rat ventricular cardiomyocyte. Points represent I_{Na} density values (pA/pF) at -40 mV. The holding potential was -100 mV. The thick horizontal line indicates the period during which the cell was superfused with the toxin. The inset on the top shows current traces recorded in control condition and during the action of toxins. Note in these cells, the marked increase in current amplitude and the slowing of inactivation. (C,D) Averaged normalized current-voltage relationship of I_{Na} established in six cells under control conditions and in the presence of 0.1 and 10 μ M *BgII* (A) or 0.1 10 μ M *BgIII* (B).

mental conditions I_{Na} at -40 mV had a density of -7.4 ± 0.9 pA/pF ($n = 27$).

Both *BgII* and *BgIII* exerted concentration-dependent effects on the I_{Na} amplitude and the inactivation time course. At low concentrations (10^{-9} , 10^{-8} M), no significant effects were seen. At higher concentrations (10^{-7} to 10^{-5} M), both toxins increased the peak I_{Na} density at -40 mV and delayed its inactivation time course. Table 1 summarizes the obtained results.

In the present experimental conditions (i.e. 5 mM external Na⁺ concentration), the inactivation time course of I_{Na} could be described by a double exponential function in about 50% of the cells. In the other half, a best fit was

obtained by a single exponential function. In addition, in most cells, the inactivation time course in the presence of toxin was best described by a single exponential. For these reasons, we chose to measure the half inactivation time of I_{Na} ($T_{50\%}$) as an indication of the slowing of the inactivation time course induced by *BgII* and *BgIII* (Table 1).

The effects of both *BgII* and *BgIII* on I_{Na} were rapidly achieved (Figure 4A, B). Steady-state effects were commonly reached in 10–15 s and were stable throughout the 4–5 min period of exposure to the toxins. Washout of toxin effect was complete for both *BgII* and *BgIII* and took approximately 30 s (Figure 4A,B). The EC_{50} of the toxins on I_{Na}

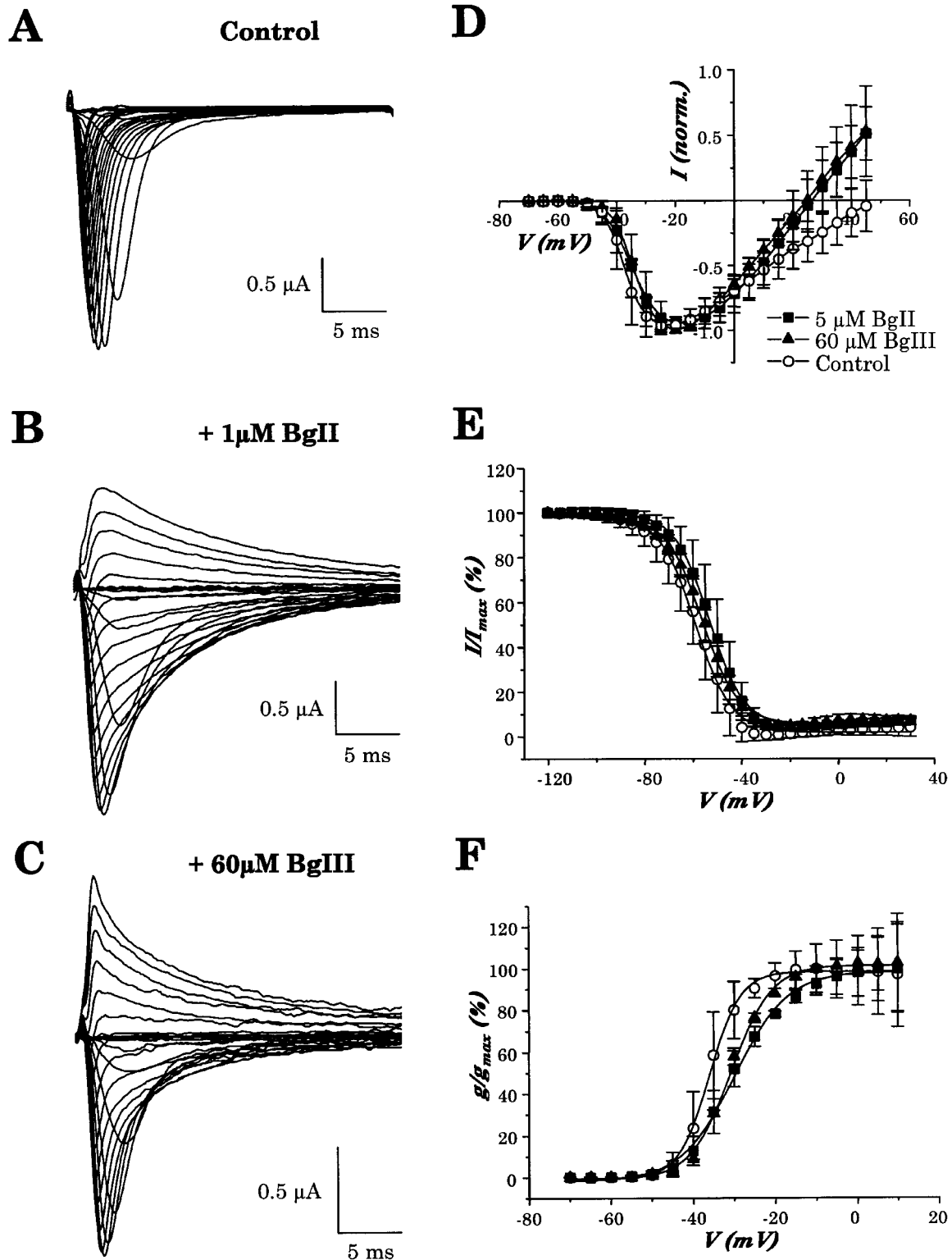


Figure 5 Effect of BgII and BgIII toxins on the cloned hH1 channel expressed in *Xenopus* oocytes. (A–C) Representative family of current traces evoked in an oocyte expressing hH1 channel by depolarizations ranging between -70 to $+45$ mV, using 5 mV increments, from a holding potential of -90 mV, in control conditions (A), in the presence of $1 \mu\text{M}$ BgII toxin (B) and in the presence of $60 \mu\text{M}$ BgIII toxin (C). Note that the slowing of inactivation is more marked with BgII than with BgIII and also the emergence of outward currents in the presence of both toxins. (D) Averaged normalized current-voltage relationship of hH1 in control conditions, in the presence of $5 \mu\text{M}$ of BgII toxin or $60 \mu\text{M}$ of BgIII toxin. (E) Averaged normalized steady-state inactivation in control condition, in the presence of $5 \mu\text{M}$ of BgII toxin or $60 \mu\text{M}$ of BgIII toxin. (F) Average normalized steady-state activation in the absence of toxins, in the presence of $5 \mu\text{M}$ of BgII toxin or $60 \mu\text{M}$ of BgIII toxin. In (D–F), data are mean \pm s.e. mean of $n=20$, $n=6$ and $n=4$ experiments, respectively.

determined by a sigmoidal fit is 58 ± 8 nM for *BgII* ($n=5$) and 78 ± 48 nM for *BgIII* ($n=5$). Moreover, these actions of *BgII* and *BgIII* on I_{Na} were voltage-dependent, the effects being more marked at positive potentials. Normalized current to voltage relationships were established under control condition and after superfusion of the myocytes with 0.1 and 10 μ M *BgII* or *BgIII* (Figure 4C,D). Under non-normalized conditions, 10 μ M *BgII* and *BgIII* increased I_{Na} at -40 mV to $144.3 \pm 12.0\%$ and $127.4 \pm 10.2\%$, of controls, respectively. At $+60$ mV, the large outward I_{Na} was increased to $306.2 \pm 15.2\%$ and $304.6 \pm 16.5\%$ of controls, respectively.

The actions of *BgII* and *BgIII* on I_{Na} were further characterized at a concentration of 10 μ M by establishing availability and recovery from inactivation (reactivation) curves (see Methods). Under control condition, best fittings of the experimental means to a Boltzmann function gave a half inactivation potential and slope factor of -80 ± 2 mV and 6.6 ± 0.3 mV, respectively. After application of 10 μ M *BgII* ($n=6$) and *BgIII* ($n=5$) they were -83 ± 3 mV and 7.8 ± 0.4 mV and -77 ± 3 mV and 6.2 ± 0.4 mV, respectively. Except for the change in the slope factor in the presence of *BgII* ($P < 0.05$), these values were not significantly different from the ones measured in control. However, it should be emphasized that both *BgII* or *BgIII* increased I_{Na} availability at potentials between -60 and -40 mV with respect to control. For example, at -40 mV availability increased from 0 to 0.13 ± 0.01 and to 0.10 ± 0.01 with *BgII* and *BgIII* respectively (data not shown).

Recovery from inactivation (reactivation) was established in control condition and after superfusion of the cells with 10 μ M *BgII* and *BgIII*. In control condition, the time to reach the half reactivation level of I_{Na} was 35.2 ± 6.0 ms. Neither *BgII* ($n=4$) nor *BgIII* ($n=4$) exerted significant effects on I_{Na} recovery from inactivation; times for half reactivation were 32.5 ± 5.3 ms and 31.8 ± 7.3 ms, respectively.

Effect of BgII and BgIII toxins on cloned hH1 Na⁺ channel expressed in Xenopus oocytes

Families of Na⁺ current recorded using the two-electrode voltage-clamp technique on *Xenopus* oocytes expressing hH1 channels are shown in Figure 5. Currents traces were evoked by step depolarizations of 25 ms ranging from -70 to $+45$ mV, using intervals of 5 mV, from a holding potential of -90 mV. While in control conditions (Figure 5A), Na⁺ currents inactivated rapidly, the inactivation kinetics strongly slowed down at the different voltages in the presence of 1 μ M *BgII* (Figure 5B) or 60 μ M *BgIII* (Figure 5C) toxin. The slowing of the inactivation process induced by 1 μ M of *BgII* was stronger than the one induced by 60 μ M of *BgIII*. While almost all the channels closed after 25 ms in control conditions, a bigger proportion of channels remained open and led to a persistent current in the presence of the toxins. Another interesting observation is that, although in control conditions only inward Na⁺ currents can be seen in the voltage range of -70 to $+45$ mV, clear outward Na⁺ currents appeared at the most positive voltages, in the presence of 1 μ M *BgII* or 60 μ M *BgIII* toxin.

Current to voltage (I-V) relationships are displayed in the Figure 5D, in the absence and presence of 5 μ M and 60 μ M of *BgII* and *BgIII*, respectively. The I-V curves reveal that *BgII*

and *BgIII* induced a depolarizing shift in the threshold of activation of the hH1 channel. This shift is confirmed by the voltage dependence of steady-state activation shown in Figure 5F. In control conditions, the half activation voltage ($V_{1/2act}$) and the slope factor measured by a Boltzmann fit of the activation curve are -37.0 ± 0.2 mV and 3.8 ± 0.2 mV ($n=20$), respectively. In the presence of 5 μ M *BgII*, $V_{1/2act}$ is -30.0 ± 0.4 mV and the slope factor is 6.5 ± 0.4 mV ($n=6$). In the presence of 60 μ M *BgIII*, $V_{1/2act}$ is -30.8 ± 0.3 mV and the slope factor is 5.0 ± 0.3 mV ($n=4$). Thus, *BgII* and *BgIII* induce a relatively similar shift of $+7.0$ mV and $+6.2$ mV of the $V_{1/2act}$ values, respectively but *BgII* is more potent than *BgIII*. Moreover, as can be anticipated from the emergence of outward currents upon toxin application in Figure 5A–C, the reversal potential (E_{rev}) changes in the presence of toxins (Figure 5D). In control conditions, E_{rev} is 44.4 ± 7.4 mV ($n=20$) while E_{rev} measured in the presence of 5 μ M *BgII* and 60 μ M *BgIII* is 26.3 ± 6.4 mV ($n=6$) and 27.0 ± 7.4 mV ($n=4$), respectively. This will be further discussed below.

Figure 5E shows the voltage dependence of steady-state sodium channel inactivation in the absence and presence of 5 and 60 μ M of *BgII* and *BgIII*, respectively. Measurements of steady-state inactivation were made following a standard two-pulse protocol. A 50 ms conditioning pulse ranging from -120 to 30 mV was followed by a 50 ms test pulse at -20 mV. In control conditions, a prepulse of -120 mV forces all the channels in the resting state and make these fully available for activation during the test pulse. In contrast to this, a prepulse of -30 mV moves all the channels in the inactivated state and thus non-available for activation during the test pulse. The voltage at which half of the channels are inactivated ($V_{1/2inact}$) and the slope factor determined by a Boltzmann fit of the inactivation curve are -58.9 ± 0.4 and 7.4 ± 0.3 mV, respectively. *BgII* and *BgIII* induce a depolarizing shift in the steady-state inactivation of the hH1 channel. In the presence of 5 μ M *BgII*, $V_{1/2inact}$ and the slope factor are -53.2 ± 0.3 mV and 7.1 ± 0.2 mV ($n=6$), respectively. In the presence of 60 μ M *BgIII*, $V_{1/2inact}$ and the slope factor are -56.1 ± 0.3 mV and 7.3 ± 0.3 mV ($n=4$), respectively. Thus, saturating concentrations of *BgII* and *BgIII* cause a shift of $+5.7$ mV and $+3.8$ mV, respectively, in the voltage at which half the hH1 Na⁺ channels are inactivated.

The inactivation time course of hH1 increased with the concentration of toxin (Figure 6). Figure 6A,B display the effect of different concentrations of *BgII* and *BgIII* on Na⁺ currents carried by hH1 and evoked by a step depolarization to 0 mV for 25 ms from a holding potential of -90 mV. As can be seen on these different current traces, the slowing of the inactivation induced by *BgII* and *BgIII* is clearly concentration-dependent, with an effect more pronounced for *BgII* than *BgIII*. The inactivation process can be fitted satisfactorily by a single exponential. The relationship between *BgII* and *BgIII* concentration and the obtained τ -values are displayed in Figure 6C,D, respectively. It can be seen that τ increased as a function of the concentration of the toxins. In control condition, τ was 1.0 ± 0.1 ms ($n=17$). In the presence of 1 μ M *BgII* or 100 μ M *BgIII*, τ increased to 5.8 ± 0.6 ($n=5$) and 4.6 ± 0.4 ms ($n=3$), respectively. The increase of τ is thus approximately 25% more important in the presence of *BgII* than in the presence of *BgIII*. The concentration of the toxin producing the half maximum effect

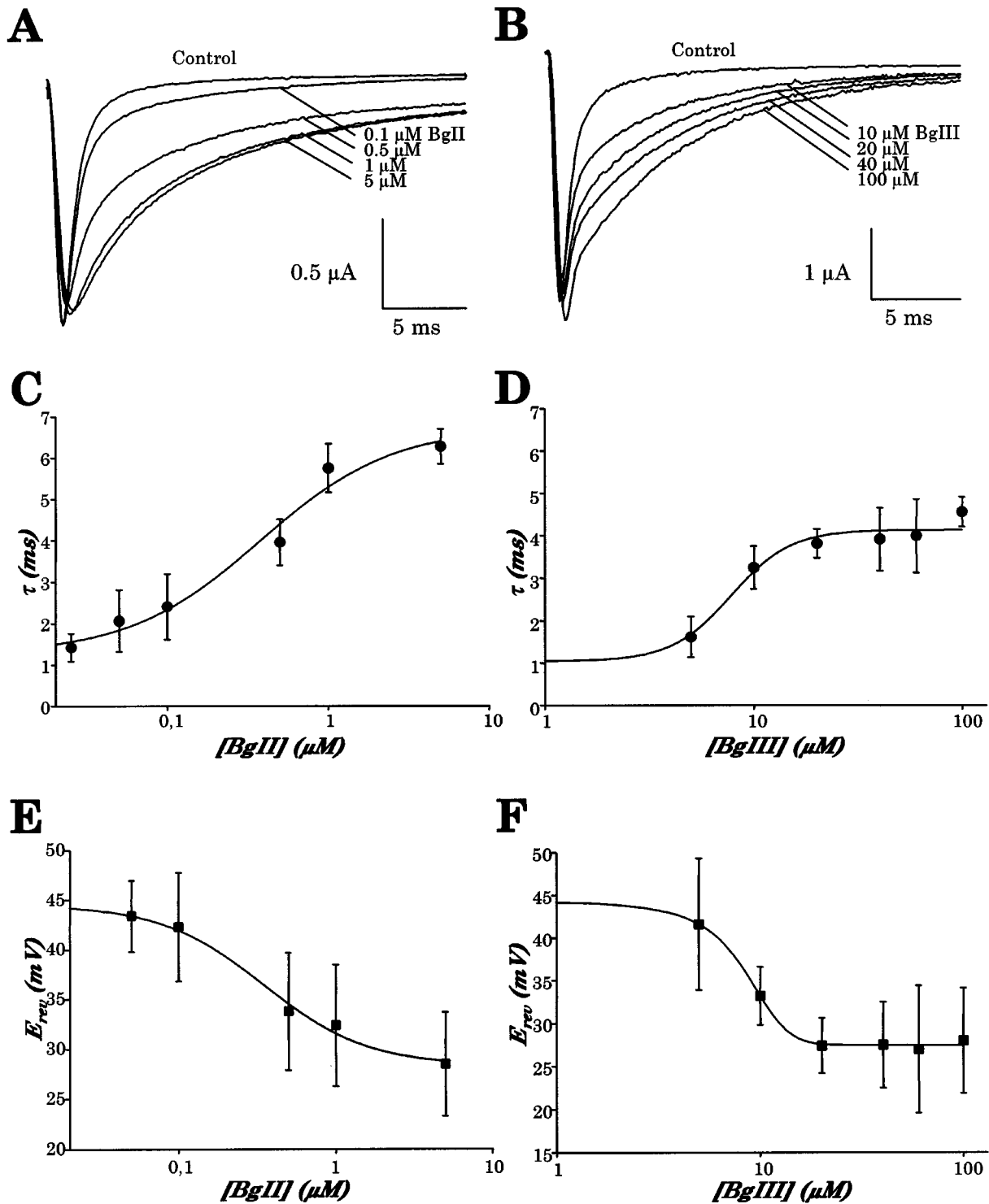


Figure 6 Concentration dependence of the slowing of inactivation and of the modification of the reversal potential induced by *BgII* and *BgIII* toxins on hH1 Na⁺ channels. (A) Current traces were evoked by a step depolarization to 0 mV lasting 25 ms from a holding potential of -90 mV, in the absence (control) and in the presence of increasing concentrations of *BgII* toxin (as indicated), using the same oocyte. (B) Same protocol as in (A) but in the absence (control) and in the presence of increasing concentrations of *BgIII* toxin (as indicated), in the same oocyte. (C) Averaged time constant of inactivation (τ) plotted vs concentration of *BgII* toxin. Time constants of inactivation were calculated from a first order exponential fit of current traces evoked by a step depolarization to 0 mV from a holding potential of -90 mV. The EC₅₀ value determined by a sigmoidal fit is 382 ± 140 nM. Data are the mean \pm s.e. mean of at least three experiments. (D) Averaged time constant inactivation (τ) plotted vs concentration of *BgIII* toxin. Time constants of inactivation were calculated as in (C). The EC₅₀ value as determined by a sigmoidal fit of τ is 7.8 ± 1.2 μM . (E) Averaged reversal potential plotted vs concentration of *BgII* toxin. The EC₅₀ value determined by a sigmoidal fit is 340 ± 70 nM. (F) Averaged reversal potential plotted vs concentration of *BgIII* toxin. The EC₅₀ value determined by a sigmoidal fit is 8.4 ± 0.3 μM . Data are the mean \pm s.e. mean of at least three experiments at each concentration.

on the inactivation time constant (EC_{50}) determined by a sigmoidal fit is 382 ± 140 nM for *BgII* and 7.8 ± 1.2 μ M for *BgIII*. From this follows that *BgII* is approximately 20 times more potent than *BgIII*.

As can be seen in the current-voltage relationships in Figure 5D, *BgII* and *BgIII* induce a shift of the reversal potential. In control conditions, the reversal potential of Na⁺ currents generated by hH1 channels is 44.4 ± 7.4 mV ($n=20$). In the presence of 5 μ M *BgII* and 100 μ M *BgIII*, E_{rev} is 28.5 ± 5.2 mV and 28.0 ± 6.1 mV, respectively ($n=4$ and 3). The maximal shifts of E_{rev} as induced by *BgII* and *BgIII* are very similar, -15.9 and -16.4 mV, respectively. A statistical analysis shows that the differences of E_{rev} in the absence or in the presence of toxins are significant (Student's *t*-test, $P < 0.05$). The permeability ratio of Na⁺ over K⁺ (P_{Na}/P_K) determined from the measured E_{rev} using the GHK equation (see Methods) decreases from 84 in control condition to 12.3 and 12.7 in the presence of 5 μ M *BgII* and 100 μ M *BgIII*, respectively. This represents a 6.8 and 6.6 fold decrease in the ionic selectivity for Na⁺ over K⁺ of the hH1 channel in the presence of the toxins. Similarly, the modification of the reversal potential induced by *BgII* and *BgIII* observed in a different external concentration of Na⁺ (96 mM) resulted in a 7 and 6.7 fold decrease in the ionic selectivity for Na⁺ over K⁺ in the presence of 5 μ M *BgII* and 100 μ M *BgIII*, respectively (data not shown). In the same experimental conditions (40 mM external Na⁺), the reversal potential of hH1 is not modified in the presence of 1 μ M ATX II (data not shown).

Figure 6E,F display the concentration dependence of the reversal potential of hH1 channels as a function of the concentration of *BgII* or *BgIII* toxins. The EC_{50} determined by a sigmoidal fit is 340 ± 70 nM for *BgII* and 8.4 ± 0.3 μ M for *BgIII*. These EC_{50} values are very comparable with the ones obtained by fitting the toxin concentration dependence of the time constant of inactivation. In both cases, i.e. E_{rev} and time constant of inactivation measurements, it is clear that *BgII* is more potent than *BgIII*.

Discussion

In rat ventricular strips, *BgII* and *BgIII* significantly prolong the action potentials and induce a decrease of the resting potential in a concentration-dependent way. The half maximum effect (EC_{50}) on the action potential duration at 0 mV is 60 nM for *BgII* and 663 nM for *BgIII*. In rat cardiac myocytes, patch-clamp studies point out that *BgII* and *BgIII*, at concentrations ranging from 0.1 to 10 μ M, increased Na⁺ current density (with an effect more pronounced at positive voltages) and delayed its inactivation time course (Figure 4). The electrophysiological effect of *BgII* and *BgIII* relates to the mechanisms of action of other Na⁺ channels toxins from sea anemone, for example the well known ATX II from *Anemonia sulcata* (Alsen *et al.*, 1981; Chahine *et al.*, 1996; Ravens, 1976; Ulbricht & Schmidtmayer, 1981), ApA and ApB from *Anthopleura xanthogrammica* (Benzinger *et al.*, 1997; Kudo & Shibata, 1980; Shibata *et al.*, 1976), or the recently isolated APE 1.1, APE 1.2, APE 2.1, APE 2.2 and APE 5.3 from *Anthopleura elegantissima* (Bruhn *et al.*, 2001). Similarly to *BgII* and *BgIII*, these toxins inhibit the inactivation process of Na⁺ channels which leads to

cardiotoxic (arrhythmia) and neurotoxic (repetitive firing) effects. In comparison, the EC_{50} values of the Na⁺ influx increase induced by ApA, ApB and ATXII in rat cardiac tissues are 3, 2 and 15 nM, respectively (Schweitz *et al.*, 1981).

Although in this study, we focused on the cardiotoxicity of these toxins, Loret *et al.* (1994) have shown in their experiments that *BgII* and *BgIII* are also potent neurotoxins. When injected intracerebroventricularly to mice, *BgII* and *BgIII* present LD_{50} values of 0.4 and 21 μ g kg⁻¹, respectively. Moreover, binding competition assays of these toxins with AahII, a scorpion toxin also acting on site 3 of Na⁺ channels, resulted in a K_d of 9 nM for *BgII* and 72 nM for *BgIII*. Likewise, recent patch-clamp experiments in cultured neurons have shown that *BgII* is able to interact with neuronal Na⁺ channels (E. Salceda, personal communication).

In order to deepen our comprehension on the molecular mechanisms of action of *BgII* and *BgIII* toxins, we performed voltage-clamp studies of the cloned cardiac Na⁺ channel hH1 expressed in *Xenopus* oocytes. Our experiments show that the main effect induced by *BgII* and *BgIII* is a marked and concentration-dependent slowing of the inactivation process of Na⁺ current associated with a decrease of the ionic selectivity of the channel. Interestingly, in comparison to the data obtained in cardiomyocytes, no clear effects on the current amplitude were observed. Our results indicate that *BgII* is about 20 times more potent than *BgIII*. The concentrations for which the half maximal effect on the inactivation process are observed (EC_{50}) are 0.38 and 7.8 μ M for *BgII* and *BgIII*, respectively. In comparison, the EC_{50} of ATX II on cloned hH1 channels expressed in mammalian cells is 11 nM (Chahine *et al.*, 1996). *BgII*, and even more *BgIII*, thus appear to be less powerful inhibitors of the Na⁺ channel inactivation process than ATX II. Nevertheless, this conclusion has to be considered, at least in part, by the fact that the pharmacological characteristics of the heterologous expression system used (i.e. mammalian cells *vs Xenopus* oocytes) may be different.

The difference in efficiency between *BgII* and *BgIII* could be surprising regarding the almost identical amino acid sequences of these two toxins. However, our findings are consistent with the results of a previous study by Loret *et al.* (1994) who have shown that the toxicity of *BgII* injected intracerebroventricularly in mice is more than 50 times superior than the toxicity of *BgIII*. In the same study, the higher toxicity of *BgII* is correlated with a higher competition binding with the scorpion α -toxin Aah II to rat brain synaptosomes. All together, these data strongly suggest that the asparagine in position 16, which is a very conservative residue among type 1 toxins, is an important residue for the toxicity of these toxins.

In parallel to the action on the inactivation process of Na⁺ currents, our study also reveals that *BgII* and *BgIII* induce several other modifications of the electrophysiological properties of hH1 channels. Firstly, the steady-state activation and inactivation parameters are shifted to more positive values by the toxins. In comparison, the steady-state activation of hH1 expressed in mammalian cells is not affected by ATX II while the steady-state inactivation is shifted to a more positive potential by about 3 mV (Chahine *et al.*, 1996).

The most intriguing observation, in the presence of both toxins, is the shift of the reversal potential of Na⁺ current generated by hH1. The measured shifts correspond to a 7

fold decrease of the ionic selectivity for Na⁺ over K⁺. In comparison, in myocytes, no modifications of the E_{rev} are observed upon addition of BgII and BgIII. In control conditions and in the presence of toxins, E_{rev} remains ~ -10 mV. In these experiments, the internal concentration of Na⁺ is higher than the external one (7 mM and 5 mM, respectively) and there is no K⁺ in solution since it has been replaced by TEA⁺ and Cs⁺. One explanation could be that the E_{rev} in the presence of the toxin is not shifted, because, even if the selectivity of the channel is modified, K⁺ is not present and, except Na⁺, no other monovalent cations are able to pass through the channel. The EC₅₀ values for the observed changes in E_{rev} are 0.34 and 8.4 μM for BgII and BgIII respectively. Here again the potency of BgII is higher than the potency of BgIII, confirming the importance of the asparagine in position 16 for the binding and the toxicity of these toxins. Interestingly, the efficacy of BgII and BgIII on the change in reversal potential is very similar (a shift of approximately -16 mV in both cases), suggesting that the mechanism by which BgII and BgIII affect the ionic selectivity of hH1 is different from the one affecting the inactivation time course or steady-state activation. To the best of our knowledge, the modification of the ionic selectivity of Na⁺ channel has never been observed with any other sea anemone toxins. Nonetheless, several other toxins have been reported to affect the ionic selectivity of Na⁺ channels. For example, two alkaloid toxins (batrachotoxin isolated from the skin of the Colombian arrow poison frog of the genera *Dendrobates* and *Phylllobates*, as well as aconitine from the plant *Aconitium napellus*) are able to decrease the ionic selectivity of Na⁺ channels (Denac *et al.*, 2000; Hille, 1992). It is thought that these toxins act by a widening of the selectivity filter. The receptor site of these lipid-soluble toxins is localized within the plasma membrane (receptor site 2) while BgII and BgIII bind to a site localized in the extracellular loop between segments S3 and S4, in the fourth domain (D4) (receptor site 3) (Loret *et al.*, 1994). It should be mentioned that another toxin which binds to site 3,

the α-toxin AaH II from the scorpion *Androctonus australis* Hector, has also been reported to modify the reversal potential of Na⁺ currents (Benoit & Dubois, 1987). On the one hand, it remains difficult to explain how toxin binding to an external site not localized within the pore could decrease the ionic selectivity of the channel. On the other hand, given the negatively charged residues in the outer mouth of rat sodium channels controlling the ion flux and selectivity (Chiamvimonvat *et al.*, 1996), it could be hypothesized that the apparent decrease in ionic selectivity of the channel could result from a modification of the surface charges of the external part of the channel due to the binding of the toxin. In this view, the toxin can be considered as a pre-filter interfering with the access of ions to the pore.

In the present study, we have characterized and compared the mode of action of two toxins isolated from the sea anemone *Bunodosoma granulifera*, BgII and BgIII, on action potentials of rat ventricular strips, on Na⁺ currents in rat cardiomyocytes and on the cloned cardiac Na⁺ channel hH1 expressed in *Xenopus* oocytes. This work is the first electrophysiological characterization of these toxins and reveals that BgII and BgIII interact with Na⁺ channels in cardiac cells, pointing out that these toxins, generally known as neurotoxins, are also cardiotoxic. From a mechanical point of view, our work discloses that, beside the 'classical' effects of sea anemone Na⁺ channels toxins (slowing of inactivation kinetics, shift of steady-state activation and inactivation parameters), BgII and BgIII also induce a decrease of the ionic selectivity of Na⁺ channels.

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