# The Superoxide Synthases of Plasma Membrane Preparations from Cultured Rose Cells

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Preparations of plasma membranes isolated from cultured rose (Rosa damascena Mill. cv Gloire de Guilan) cells synthesized O<sub>2</sub><sup>-</sup> when incubated with either NADH or NADPH, as measured by an O<sub>2</sub><sup>--</sup>specific assay based on the chemiluminescence of lucigenin. The activities were strongly dependent on the presence of Triton X-100. The  $K_m$  for NADH was 159  $\mu$ M; that for NADPH was 19  $\mu$ M. Neither NADH- nor NADPH-dependent activity was inhibited by azide, an inhibitor of peroxidase, nor by antimycin A, an inhibitor of mitochondrial electron transport; both activities were inhibited by 30 to 100 nm diphenylene iodonium, an inhibitor of the mammalian NADPH oxidase. The NADH- and NADPH-dependent activities could be distinguished by detergent solubilization and ultracentrifugation: the NADH-dependent activity sedimented more easily, whereas the NADPH-dependent activity remained in suspension. One or both of these enzymes may provide the O<sub>2</sub><sup>-</sup> seen when plant cells are exposed to pathogens or pathogen-associated elicitors; however, plasma membranes from rose cells treated with a Phytophthora elicitor had the same activity as control cells.

The term "active oxygen" refers to chemically reactive species of oxygen, including singlet oxygen, hydroxyl radical,  $O_2^-$ , and  $H_2O_2$ . The generation of these species occurs as a byproduct of several natural processes, such as photosynthesis and respiration. Since the products can destroy sensitive functional molecules, plants and other organisms have evolved detoxifying mechanisms (Foyer et al., 1994).

Active oxygen species also play a positive role in normal plant development and in a plant's response to stress. In normal development,  $H_2O_2$  serves as a substrate for the polymerization of lignin, suberin, and possibly other cell-wall components (Hahlbrock and Griesbach, 1979; Olson and Varner, 1993).  $O_2^-$  and  $H_2O_2$ , produced in response to bacterial or fungal attack, are thought to kill the attacking cells and also adjacent host cells, thus limiting the spread of the pathogen. Active oxygen species have been suggested to play a signaling role in inducing other defenses to pathogens (Levine et al., 1994; Price et al., 1994).

In past publications, our laboratory has reported that suspension-cultured cells of rose (*Rosa damascena*) produce a burst of  $H_2O_2$  after they have been treated with UV-C radiation (Murphy and Huerta, 1990). They also produce the same response when they are challenged with a chem-

ical elicitor, a cell-wall component isolated from a culture of a *Phytophthora* species originally identified as a parasite on rose (Arnott and Murphy, 1991).

The mechanism of synthesis of active oxygen species, and particularly of metabolically required H<sub>2</sub>O<sub>2</sub>, is a topic of active discussion. In rose cells, the elicitor-induced H<sub>2</sub>O<sub>2</sub> is synthesized exclusively from O<sub>2</sub><sup>-</sup>, as judged by the ability of N,N-diethyldithiocarbamate, an inhibitor of superoxide dismutase, to block the appearance of  $H_2O_2$  in elicitor-stimulated cells (Auh and Murphy, 1995). It has been proposed that apoplastic H2O2 synthesized by horseradish roots occurs by dismutation of the O2- generated through a complex cycle involving extracellular NADH, phenolic compounds, and peroxidase (Elstner and Heupel, 1976; Gross et al., 1977; Halliwell, 1978). The extracellular NADH is said to be supplied by cell-wall-bound malate dehydrogenase, with the substrate malate provided by a malate-oxalacetate shuttle in the plasma membrane. An alternative source of O<sub>2</sub><sup>-</sup> in plant cells has been proposed by Doke (1985): a plasma membrane-associated oxidase that transfers electrons from NAD(P)H in the cytoplasm to oxygen in the apoplast. In rose cells the synthesis of  $H_2O_2$ is stimulated by added NADH and phenolics (or salicylhydroxamic acid), but it is neither stimulated by malate nor inhibited by oxalacetate (Murphy and Huerta, 1990), as it would be if a mechanism similar to that postulated for horseradish roots were active. Furthermore, the synthesis of O<sub>2</sub><sup>-</sup> and H<sub>2</sub>O<sub>2</sub> by elicitor-stimulated rose cells is inhibited by DPI, an inhibitor of the plasma membrane NADPH oxidase of mammalian neutrophils, consistent with the presence of such an enzyme in the rose cell plasma membranes (Auh and Murphy, 1995). In this paper, we describe NADH- and NADPH-oxidases assayed by the appearance of  $O_2^-$  ( $O_2^-$  synthases) in rose cell plasma membrane preparations.

### MATERIALS AND METHODS

### Cells, Plasma Membranes

Suspension-cultured cells of *Rosa damascena* Mill. cv Gloire de Guilan were grown as described by Murphy et al. (1979). Plasma membranes were prepared from 4-d-old cultures by the aqueous two-phase method of Larsson et al. (1983) as previously described (Murphy and Auh, 1992). With this method, we disrupted the cells by sonication in a buffer containing 25 mM Tris, 25 mM Mes, 3 mM EDTA, 2.5

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Abbreviations: DPI, diphenylene iodonium;  $O_2^-$ , superoxide.

mM DTT, and 250 mM Suc, pelleted dense material by centrifugation at 18,800g for 15 min, and then collected microsomal membranes by centrifugation at 190,000g for 40 min. We resuspended the microsomes and subjected them to three successive partitions in a 12-g mixture containing 6.3% PEG, 6.3% dextran, 4 mM KCl, and 250 mM Suc in 5 mM potassium phosphate buffer, pH 7.6. The membranes from the third upper (PEG) phase were diluted in 1 mM Tris-Mes buffer, pH 6.5, containing 1 mM MgSO<sub>4</sub> and 20% Suc, pelleted by centrifugation, and stored at  $-70^{\circ}$ C until used.

An analysis of marker enzymes in the final preparation is given in Table I. K<sup>+</sup>-stimulated ATPase (plasma membrane), IDPase (dictyosome), Cyt c oxidase (mitochondrion), and Cyt c reductase (ER) were assayed as described by Hodges and Leonard (1974), except that the mitochondrial Cyt c oxidase was estimated by the difference in activity with and without 1 mM KCN and the Cyt c reductase was measured in the presence of 1  $\mu$ M antimycin A rather than KCN. Peroxidase (soluble) was measured by the oxidation of guaiacol according to the method of Castillo et al. (1984). The specific activity of the ATPase in the plasma membrane preparation was high and within the normal range. The purification procedure removed more than 99.7% of the total Cyt c oxidase, Cyt c reductase, IDPase, and peroxidase. However, a measurable amount of the oxidase, reductase, and peroxidase remained in the plasma membrane preparation (Table I).

## Assay of O<sub>2</sub><sup>-</sup> Synthesis

The assay of  $O_2^-$  was based on the chemiluminescence of lucigenin (bis-*N*-methylacridinium nitrate, Sigma) as previously described (Auh and Murphy, 1995) but with modifications. A standard assay mixture contained 1.5 to 4 µg of plasma membrane protein, 100 µM NADH or 30 µM NADPH, 0.02% (w/w) Triton X-100, and 0.4 mM lucigenin,

**Table 1.** Marker enzymes in a representative plasma membranepreparation

The crude extract and the final plasma membrane fraction prepared from 20 g of washed 4-d-cultured rose cells were assayed for ATPase, KCN-sensitive Cyt *c* oxidase, antimycin A-resistant Cyt *c* reductase, IDPase, guaiacol peroxidase, and NADH- and NADPHdependent  $O_2^{--}$  synthases. The homogenate (Homog) contained 3.07 mg mL<sup>-1</sup> protein in 107 mL; the membrane fraction (U3) contained 0.57 mg mL<sup>-1</sup> protein in 1.2 mL. NT, Not tested. ND, None detected.

<b>F</b>	Total Activity		Specific Activity	
Enzyme	Homog	U3	Homog	U3
	nmol min <sup>-1</sup>		nmol min <sup>-1</sup> mg <sup>-1</sup>	
ATPase	NT	650	NT	950
Cyt c oxidase	2100	6.0	6.5	8.7
Cyt c reductase	2400	1.9	7.4	2.8
IDPase	$3.2  imes 10^4$	ND	98	ND
Guaiacol POD	8400	4	26	5.8
$O_2^-$ synthase				
NADH	5029ª	53	15ª	77
NADPH	94ª	18	0.29 <sup>a</sup>	27
30 7 1	AL 14 1 1 1 1		<ul> <li>to tate 10</li> </ul>	and the last

 ${}^{a}O_{2}^{-}$  synthase activity in homogenate was initially rapid but decreased below background within 20 min.

plus sufficient buffer (0.1 M Gly brought to pH 9 with NaOH plus 1 mm EDTA) to make a total volume of 1 mL. The components were combined in a 4-mL disposable plastic scintillation vial, and the chemiluminescence was measured by counting the mixture for 1 min in a liquid scintillation counter (Beckman model 9800) operated without the coincidence circuit (single-photon counting mode).

#### Assay Calibration with Xanthine and Xanthine Oxidase

A sample of xanthine oxidase (grade I or grade III, Sigma) was assayed by the oxidation of xanthine: 10 to 200  $\mu$ L of a xanthine oxidase dilution (nominally 0.3 unit  $mL^{-1}$ ) and 0.05 mM xanthine were combined with the Gly-NaOH-EDTA buffer described above, and the rate of increase in  $A_{290}$ , representing the oxidation of xanthine to uric acid, was measured in a spectrophotometer over 1 to 3 min. The rate of the reaction (mol  $O_2^{-}$  min<sup>-1</sup>) was calculated by multiplying the rate of increase of  $A_{290}$  (min<sup>-1</sup>) by the reaction volume (3 mL) and dividing the result by the absorbance constant of uric acid (extinction coefficient at 290 nm =  $1.22 \times 10^7$  mL mol<sup>-1</sup> cm<sup>-1</sup>) and the path length (1 cm) and then multiplying the result by an assumed 2 mol O<sub>2</sub><sup>-</sup> produced for each 1 mol xanthine oxidized (Corbisier et al., 1987). The same sample of xanthine oxidase was then further diluted by a factor of 100, and aliquots were combined with 0.05 mм xanthine, 0.4 mм lucigenin, and 1 mм N,N-diethyldithiocarbamate in sufficient Gly-NaOH-EDTA buffer to total 1 mL; the enzyme-stimulated chemiluminescence of the mixture was measured. Both the rate of xanthine oxidation and the chemiluminescence increased proportionally with increasing enzyme. By assuming that the proportionality held for comparisons between the assays, we could calculate the reaction rate equivalent to a given level of chemiluminescence. One million counts per minute  $(10^6 \text{ cpm in the chemiluminescence assay})$  was equivalent to 6.3  $\pm$  1.6 pmol O<sub>2</sub><sup>-</sup> produced min<sup>-1</sup> (mean  $\pm$  sE, five experiments). This value was used to convert chemiluminescence data to reaction rates for presentation in the tables and figures below.

### Solubilization and Differential Centrifugation

Samples of plasma membrane were solubilized by diluting them into buffer (15 mM Tris, pH 7.4, 2 mM MgSO<sub>4</sub>, 330 mM Suc) containing 0.75 mg mL<sup>-1</sup> Triton X-100, 2.5 mg mL<sup>-1</sup> sodium deoxycholate, or 0.75 mg mL<sup>-1</sup> dodecylmaltoside and incubating them for at least 10 min at 0°C. To sediment unsolubilized membranes, samples were centrifuged for 60 min at 100,000g. The supernatants were removed, and the pellets were resuspended in the buffer described above (minus detergent).

## RESULTS

## **Basic Characteristics of the Enzyme**

In the presence of 100  $\mu$ M NADH or 30  $\mu$ M NADPH and 0.02% Triton X-100, 0.4 mM lucigenin produced a level of chemiluminescence equivalent to 6 to 20 pmol min<sup>-1</sup>, the level depending on the preparation of dinucleotide; this

chemiluminescence was strongly stimulated by preparations containing plasma membrane. Figure 1 shows the development and persistence of chemiluminescence in typical plasma membrane-containing reaction mixtures. An initial increase peaking at 20 to 40 min was followed by a regular, exponential decay. We assumed that the chemiluminescence closely reflected the rate at which O<sub>2</sub><sup>-</sup> was produced, i.e. that the lucigenin reacted with the  $O_2^-$  as quickly as it was formed and there was no long-term accumulation of  $O_2^{-}$ . In confirmation of this assumption, we observed that the chemiluminescence of a fixed amount of  $O_2^{-}$ , prepared by incubating DMSO with NaOH (Hyland et al., 1983) and added to lucigenin in buffer, was lost exponentially with a half-life of 0.11 min (data not shown). This observation implies that the assay responded quickly to changes in  $O_2^-$  production rate. It also implies that the slow increase and decrease of chemiluminescence shown in Figure 1 represented activation and decay of the enzyme activity. Because initial counts were more variable (affected by room light and other factors), all reaction mixtures were counted six times over a period of at least 75 min, and measurements were based on data taken on the fourth, fifth, and sixth counts. The results from experimental treatments were always compared to those of untreated controls counted at the same times.

The rate of chemiluminescence was proportional to the concentration of lucigenin, up to a concentration of 0.4 to 0.5 mM (Fig. 2A). We used 0.4 mM lucigenin, a near-satu-



**Figure 1.** Time course for lucigenin chemiluminescence. Reaction mixtures of 1 mL contained 0.1  $\mu$  Gly-NaOH buffer, pH 9, 0.02% Triton X-100, 0.4 mm lucigenin, and 100  $\mu$ m NADH (top) or 30  $\mu$ m NADPH (bottom), either without (open symbols) or with (filled symbols) plasma membrane (top, 0.84  $\mu$ g of protein; bottom, 6.2  $\mu$ g of protein). Different closed symbols represent replicate tubes.



Figure 2. Characteristics of plasma membrane-catalyzed O<sub>2</sub><sup>-</sup> synthesis. Except where noted, assays contained 100 µM NADH, 0.02% Triton X-100, and 0.4 mm lucigenin. A, Dependence on lucigenin concentration. Each tube received 0.84 µg of plasma membrane protein, NADH, Triton X-100, as noted above, and the concentration of lucigenin given on the abscissa. B, Dependence on concentration of Triton X-100. Tubes received 0.84, 1.7, 2.2, or 4.1 µg of plasma membrane protein (from different preparations), NADH and lucigenin as noted above, and the concentration of Triton X-100 given on the abscissa. The background luminescence observed with no plasma membrane was subtracted from each value. Within each experiment, values were normalized to the value obtained at 0.08% Triton X-100. C, Dependence on concentration of enzyme. NADH, Triton X-100, and lucigenin as noted above were combined with the amounts of plasma membrane given in B and with measured fractions of those amounts. The background luminescence observed with no plasma membrane was subtracted from each value; the differences were normalized by dividing by the value given by the standard amount ("1" on the abscissa) to correct for the different activities of the enzyme preparations. The data in A come from a single experiment. The data in B and C represent means  $\pm$  sE (n = 4).

rating amount, for all of the experiments reported in this paper. Lucigenin chemiluminescence has been described as specific for  $O_2^-$  (Corbisier et al., 1987), and the inhibition by superoxide dismutase of the chemiluminescence produced by rose plasma membrane in the presence of NADH or NADPH (Fig. 3) is consistent with the interpretation that



**Figure 3.** The effect of  $O_2^{-}$  dismutase (SOD) on plasma membranestimulated lucigenin chemiluminescence. Mixtures containing buffer, 100  $\mu$ M NADH or 30  $\mu$ M NADPH, 0.02% Triton X-100, and 0.4 mM lucigenin, with and without 10<sup>4</sup> units (nominal) of SOD were counted by scintillation spectrometry for 75 min; the value at -5 min shows the last count. At time 0, 0.84  $\mu$ g of plasma membrane protein was added, and mixtures were counted periodically during the next 75 min. The figure shows the results from a representative experiment; the experiment was performed three times with similar results.

in this system chemiluminescence was at least 90% mediated by  $O_2^{-}$ .

The synthesis of  $O_2^-$  by plasma membrane acting with NADH or NADPH was strongly dependent on the inclusion of Triton X-100 in the assay medium (Fig. 2B). The latency (1 – [activity without Triton]/[maximum activity with Triton]) was 1.0 ± 0.06, averaged over four independent plasma membrane preparations. Triton X-100 also had an effect on the chemiluminescence of lucigenin, both alone and in combination with xanthine plus xanthine oxidase. The standard concentration of Triton X-100 (0.02%) produced an increase of 25% in chemiluminescence (data not shown). This effect contributed to the apparent stimulation of plasma membrane-catalyzed chemiluminescence by the detergent.

The rate of chemiluminescence increased proportionally with an increase in plasma membrane preparation (Fig. 2C), confirming the utility of the assay as a measure of  $O_2^-$  synthase activity.

The purification procedure resulted in a membrane preparation with  $O_2^-$  synthase specific activity that was substantially higher than the activities of Cyt *c* oxidase, Cyt *c* reductase, and peroxidase (Table I). The purification and recovery of the  $O_2^-$  synthases were difficult to determine, because the chemiluminescence stimulated by the homogenate decayed more rapidly than that by the plasma membrane and may not have been an accurate measure of the amount of enzyme in the homogenate. The highest specific activity of a plasma membrane preparation was 170 nmol  $O_2^-$  min<sup>-1</sup> mg<sup>-1</sup> protein, although the value shown in Table I is more typical. Other preparations had specific activity of different preparations may have been caused by different levels of contaminants in the preparation, different

ent responses of the membranes to the Triton X-100, different degrees of loss of cofactor (see below), and different durations of storage (some preparations had been stored for as long as 4 years at  $-70^{\circ}$ C). All experiments characterizing the enzymes were performed on at least three independent preparations with similar results.

## Substrate and Cofactor Dependence

The dependence of the  $O_2^-$  synthetic reaction on substrate concentration was different for NADH and NADPH, although in both cases the kinetics could be modeled by the Michaelis-Menten formula (Fig. 4). With NADH as substrate, the  $K_m$  was 159  $\mu$ M; with NADPH as substrate, the  $K_m$  was 19  $\mu$ M. Although the use of different preparations of plasma membrane, with different specific activities, made it difficult to compare  $V_{max}$  for the two substrates, three of the preparations were used with both substrates. The ratios of  $V_{max}$  for NADH:NADPH demonstrated by



**Figure 4.** Dependence of  $O_2^-$  synthase activity on substrate concentration. All assays contained 0.02% Triton X-100 and 0.4 mm lucigenin. Data were derived from 10 experiments using 4 different enzyme preparations (top, NADH) or from 3 experiments with 3 different enzyme preparations (bottom, NADPH). Data obtained from preparations with different specific activities were normalized to a standard value (averaged from several experiments) at 0.15 mm NADH and 0.016 mm NADPH to obtain more accurate estimates of the  $K_m$ . Curves represent best least-squares fits of the data to a Michaelis-Menten equation by a nonlinear regression procedure.

these three preparations were 3.8, 1.3, and 2.5. Thus, the  $V_{\rm max}$  for NADH was generally higher than for NADPH, but there was substantial variation in these ratios among the preparations.

The activities of both NADH- and NADPH-dependent enzymes were stimulated by FAD (Fig. 5). For both enzymes, the maximum degree of stimulation over the activity observed without added FAD was approximately 2-fold, reached with 0.3  $\mu$ M FAD using NADH as substrate and 1  $\mu$ M FAD using NADPH as substrate. There was a loss of stimulation at 10  $\mu$ M FAD with NADH, but not NADPH, as substrate. FMN gave no significant increase in NADHdependent activity, but it did give a significant (P < 5%) stimulation of NADPH-dependent activity. This stimulation reached a maximum of 1.6-fold over the control activity at 6  $\mu$ M FMN (data not shown).

## Inhibitors

The effects of various compounds were investigated to test the possible relationship of the rose cell  $O_2^-$  synthases to other electron transport activities (see "Discussion"). Azide is an inhibitor of peroxidases, with 3.3 mM N<sub>3</sub><sup>-</sup> giving 50% inhibition of the rose cell peroxidases secreted into the growth medium (data not shown). Antimycin A is an inhibitor of the mitochondrial electron transport chain. Quinacrine, imidazole, and DPI are inhibitors of mammalian  $O_2^-$  synthase. Arg is a substrate for mammalian nitric oxide synthase, an enzyme also reported to synthesize  $O_2^-$ , and *N*-nitroarginine methyl ester is an inhibitor of  $O_2^$ synthesis by that enzyme (Pou et al., 1992). Auxins (2,4-D, naphthaleneacetic acid) stimulate NADH oxidases in cer-



**Figure 5.** Stimulation of  $O_2^-$  synthase activity by FAD. All assays contained 0.02% Triton X-100, 0.4 mM lucigenin, 100  $\mu$ M NADH or 30  $\mu$ M NADPH, plasma membrane protein, and the concentration of FAD given on the abscissa. Data represent means  $\pm$  sE from three experiments using three different enzyme preparations. Background values (without enzyme or FAD) were subtracted, and the differences were related to the value obtained with enzyme but without added FAD. Control activities ranged from 79 to 134 (NADH) and 32 to 36 (NADPH) pmol min<sup>-1</sup>.

**Table II.** Effects of putative inhibitory or activating compounds on  $O_2^-$  synthase activity

All assays contained 0.02% Triton X-100, 0.4 mM lucigenin, either 100  $\mu$ M NADH or 30  $\mu$ M NADPH, and 1.5 to 4  $\mu$ g of plasma membrane yielding an activity in control mixtures of 28 to 160 (NADH) or 12 to 70 (NADPH) pmol min<sup>-1</sup>. Values are activities in reaction mixtures containing inhibitor relative to control mixtures without inhibitor. A value of 1.0 indicates that there was no effect. Background activities (without enzyme) were subtracted: a negative value indicates that the luminescence with enzyme was less than the luminescence without enzyme. Means and SE were calculated from assays using three independently prepared plasma membrane preparations.

ta bilata a	Substrate			
minutor	NADH	NADPH		
NaN <sub>3</sub> , 3.3 mм	$0.98 \pm 0.02$	$0.91 \pm 0.10$		
Antimycin A, 10 µм	$0.97 \pm 0.05$	$1.09 \pm 0.03$		
Arginine, 1 mм	$1.37 \pm 0.26$	$1.02 \pm 0.06$		
N-nitroarginine methyl ester, 1 тм	$1.07 \pm 0.17$	$0.94 \pm 0.02$		
2,4-D, 10 µм	$0.99 \pm 0.03$	$1.10 \pm 0.12$		
Naphthaleneacetic acid, 10 µм	$1.03 \pm 0.06$	$1.09 \pm 0.14$		
Quinacrine, 500 μм	$0.73 \pm 0.20$	$0.52 \pm 0.01^{a}$		
Imidazole, 10 mм	$0.82 \pm 0.19$	$1.05 \pm 0.22$		
DPI, 15 μм	$-0.02 \pm 0.05^{a}$	$-0.02 \pm 0.02^{a}$		
<sup>a</sup> Values significantly differer	t from 1.0 ( $P < 0$ .	05 by <i>t</i> test).		

tain plant plasma membranes (Morré and Brightman, 1991).

The  $O_2^-$  synthetic reaction showed no sensitivity to azide, antimycin A, Arg, *N*-nitroarginine methyl ester, or auxins and little sensitivity to quinacrine and imidazole when assayed either with NADH or NADPH. However, it was completely inhibited by DPI (Table II). The sensitivity to DPI depended on the substrate provided: with NADH, the reaction was inhibited 50% by 100 nm DPI; with NADPH, the reaction was more sensitive, with 50% inhibition requiring approximately 30 nm DPI (Fig. 6).

## Solubilization and Differential Centrifugation

Although the differences in reaction characteristics ( $K_{\rm m}$ and sensitivity to DPI) seen with NADH and NADPH as substrates could be explained by the specificity of the active site of a single enzyme, it is also possible that separate enzymes could be transferring electrons from the two electron donors to O2. A series of solubilization/sedimentation tests were conducted to determine whether the NADH- and NADPH-dependent O<sub>2</sub><sup>-</sup> synthases could be separated from each other and to provide initial information needed for the purification of the enzymes (Fig. 7). The NADH-dependent O<sub>2</sub><sup>-</sup> synthase was easily sedimented at 100,000g for 1 h in the Suc-Tris medium; it was partially solubilized by 0.75 mg mL<sup>-1</sup> Triton X-100 and 0.75 mg  $mL^{-1}$  dodecylmaltoside but not by 2.5 mg  $mL^{-1}$  sodium deoxycholate. The NADPH-dependent O<sub>2</sub><sup>-</sup> synthase was not as easily sedimented, even in the absence of detergent, and was more completely solubilized by the detergents. After they were added to reaction medium (a dilution of



**Figure 6.** Inhibition of  $O_2^-$  synthase activity by DPI. All assays contained 0.02% Triton X-100, 0.4 mm lucigenin, either 100  $\mu$ m NADH or 30  $\mu$ m NADPH, and plasma membrane. Values are activities in tubes containing inhibitor relative to control tubes without inhibitor. Background activities without enzyme were subtracted. Control activities for the three experiments ranged from 23 to 158 (NADH) and 23 to 52 (NADPH) pmol min<sup>-1</sup>. The data points indicate means from three independent experiments. A two-way factorial analysis of variance indicated that the results for NADH and NADPH were significantly different (P < 0.02).

50-fold), the activities of NADH- and NADPH-dependent enzymes in the supernatants retained their ability to be stimulated by addition of 0.02% Triton X-100 to the assay medium.

## Lack of Activation by Elicitor

The synthesis of  $O_2^-$  by rose cells in vivo was stimulated by an elicitor preparation from a *Phytophthora* species (Auh and Murphy, 1995). However, the specific activity in vitro of  $O_2^-$  synthase in plasma membranes prepared from elicitor-treated cells was the same as that from control cells, whether NADH or NADPH was used as substrate (Table III).

### DISCUSSION

The data presented demonstrate that plasma membrane preparations from suspension-cultured rose cells catalyze the synthesis of  $O_2^-$  in the presence of the electron donors NADH and NADPH. Rose cell extracellular peroxidases are sensitive to azide (50% inhibition at 3.3 mM) and not to DPI (Auh and Murphy, 1995; T.M. Murphy and C.-K. Auh, unpublished experiments), whereas the synthesis of  $O_2^-$  by our membrane preparations was sensitive to DPI and not to azide (Table II). The activity of guaiacol peroxidase in the plasma membrane preparation was very low (Table I), and there were no added phenolics, compounds required for the formation of  $O_2^-$  by peroxidase, in our assay mixtures. We thus suggest that peroxidase cannot be responsible for  $O_2^-$  synthesis in this system. This conclusion is consistent with the results of Doke and Miura (1995).

The observation that the  $O_2^-$  synthesis of plasma membrane vesicles is strongly stimulated by Triton X-100 is consistent with the idea that the oxidizing sites of the  $O_2^$ synthase(s) are on the cytoplasmic side of the vesicles, unavailable to the substrate NADH or NADPH in the absence of a detergent. However, lipid or detergent may stimulate plasma membrane enzymes by mechanisms in addition to permeabilization of vesicles (for example, see Sandstrom et al., 1987; Bourdil et al., 1990). A direct effect on the  $O_2^-$  synthase(s) is not unlikely, since their degree of stimulation by Triton X-100 was greater than that generally measured for ATPase in plasma membranes prepared by the same method.

## Evidence for Two O<sub>2</sub><sup>-</sup> Synthases

Several observations suggest that the NADH- and NADPH-dependent  $O_2^-$  synthase activities represent different enzyme systems. These include the variable ratio of



**Figure 7.** Solubilization and centrifugation of NADH- and NADPHdependent  $O_2^-$  synthase. Aliquots of plasma membrane preparation (8.4 µg of protein) were treated for 10 min at 4°C in 200 µL of 15 mM Tris-Cl buffer, pH 7.4, 2 mM MgSO<sub>4</sub>, 330 mM Suc, containing 0.75 mg mL<sup>-1</sup> Triton X-100 (Tri), 0.75 mg mL<sup>-1</sup> dodecylmaltoside (DM), or 2.5 mg mL<sup>-1</sup> sodium deoxycholate (DC), or in buffer lacking detergent, and then centrifuged for 1 h at 100,000g. Supernatants were removed and the pellets were resuspended overnight in the original volume of buffer. Supernatants and pellets were assayed for  $O_2^-$  synthase with NADH (top) or NADPH (bottom) as substrate and in the presence or absence of 0.02% Triton X-100 in the reaction mixture. The upper and lower parts of each bar show the activity in the supernatant and pellet, respectively. Values represent means from three independent experiments.

# **Table III.** Comparison of $O_2^-$ synthesis by plasma membranes prepared from elicitor-treated and untreated rose cells

Equal amounts of plasma membrane protein (2.2  $\mu$ g in one experiment; 3.4  $\mu$ g in another) from untreated cells and cells treated with elicitor for 30 or 60 min were compared in reaction mixtures containing 0.02% Triton X-100, 0.4 mm lucigenin, and either 100  $\mu$ m NADH or 30  $\mu$ m NADPH. Within each experiment, rates for plasma membranes from treated cells were related to those from untreated cells; the activities of membranes from the untreated cells were 76 and 103 (NADH) and 52 and 69 (NADPH) pmol min<sup>-1</sup>. Data are means  $\pm$  se.

Substrate	Control	Time of Treatment		
		30 min	60 min	
NADH	1.0	$1.11 \pm 0.18$	$0.92 \pm 0.03$	
NADPH	1.0	$1.15 \pm 0.07$	$1.04 \pm 0.03$	

NADH- to NADPH-dependent activity in different preparations, the great difference in  $K_{\rm m}$  between the two activities, the difference in concentration of DPI needed to inhibit the two activities by 50%, and the different sedimentation properties of the two activities. The low degree of sedimentation of the NADPH-dependent activity indicates that it is either on regions of the plasma membrane that are smaller or less dense than the regions holding the NADH-dependent activity or it is on a different type of membrane. Although the two-phase isolation procedure is recognized as highly specific for plasma membrane, some contamination of the preparation by other membranes occurs (Table I). The presence of Cyt c oxidase suggested that mitochondrial membrane might contribute to the NADH-dependent  $O_2^-$  synthase activity. However, the insensitivity of the activity to antimycin A argues against the idea that the full mitochondrial electron transport system is involved.

## Relation to O<sub>2</sub><sup>-</sup> Synthesis in Vivo

The sensitivity of the lucigenin assay was sufficiently great, and the conditions of the plasma membrane assay and intact-cell assay (Auh and Murphy, 1995) were sufficiently different, that we cannot conclude with certainty that the  $O_2^-$  measured in the present plasma membrane experiments came from the same enzyme system that produces  $O_2^-$  in intact cells. However, in both cases the  $O_2^$ synthesis was not inhibited by azide or cyanide and was inhibited by low concentrations of DPI (Table II; Auh and Murphy, 1995). The difference in sensitivity to DPI (50% inhibition at 30-100 nM DPI for plasma membranes, at 13  $\mu$ M for intact cells) can be explained by limitations in the uptake of DPI by cells and the presence of Triton X-100 in the plasma membrane assay. The fact that the sensitivity of the intact-cell system to quinacrine and imidazole (Auh and Murphy, 1995) was greater than that of the plasma membrane system (Table II) suggests that there were components in intact cells that were essential for elicitor-induced O<sub>2</sub><sup>-</sup> synthesis and that were lost in the isolation of plasma membranes.

Which of the two substrates is responsible for the synthesis of  $O_2^-$  in intact cells? The low  $K_m$  of the NADPH-specific system and the supposition that NADH is primarily oxidized in living cytoplasm whereas NADPH is primarily reduced suggest that the NADPH system is the most likely candidate. This suggestion is consistent with an earlier suggestion concerning ferricyanide reduction by *Zea mays* roots (Qui et al., 1985). However, the high  $V_{max}$  of the NADH-specific system suggests that its potential contribution should not be ignored.

Doke and his colleagues have extensively investigated O<sub>2</sub><sup>-</sup> synthesis by *Phytophthora*-infested potato tuber tissue. Recently, Doke and Miura (1995) reported that NADPH oxidase is associated with plasma membrane isolated from tuber tissue inoculated with an incompatible Phytophthora strain or with a cell-free elicitor preparation but not with plasma membrane from uninoculated tubers or from tubers inoculated with a compatible strain (Doke and Miura, 1995). We did not observe such a distinction between plasma membrane preparations from control and elicitortreated cells (Table III), possibly because of changes in the enzymes during the purification procedure. The fact that the plasma membranes from control rose cells had O<sub>2</sub><sup>-</sup> synthesis activity is consistent with our observation (Auh and Murphy, 1995) that control cells produce  $O_2^-$  at a level as high as one-fourth that of elicited cells. Doke and Miura (1995) also reported that O<sub>2</sub><sup>-</sup> synthesis by isolated potato plasma membrane could be stimulated in vitro by a Phytophthora elicitor in the presence of a 100,000g supernatant fraction and other components. We have not yet attempted to duplicate these findings.

## Relation to Other Aspects of Plasma Membrane Redox Activity

Redox activities of plant plasma membranes have been studied in several other contexts. Electrons, presumably from cytoplasmic donors, can be accepted by extracellular ferricyanide ion (Møller and Lin, 1986). This may be similar to trans-plasma membrane electron transport in plasma membrane  $O_2^-$  synthesis. However, there are reasons to believe that the enzyme systems differ in at least some aspects. For instance, the application of a fungal elicitor, which stimulates  $O_2^-$  synthesis in rose cells (Auh and Murphy, 1995), inhibits ferricyanide reductase in the same cells (Murphy and Auh, 1992).

Plasma membranes isolated from soybean hypocotyls demonstrate an auxin-stimulated NADH oxidase activity (Morré and Brightman, 1991). The fact that this activity is stimulated under the same conditions in which growth is stimulated and only by those auxin analogs that also stimulate cell elongation suggests that the activity has some role in the growth process. This enzyme is separable from the ferricyanide reductase. The reduced product of this activity has not been determined, but oxygen appears to be the acceptor and  $O_2^-$  may be formed. However, it is not clear how  $O_2^-$  would participate in the stimulation of growth. In fact there are reasons to believe that its appearance in the apoplast might have the opposite effect through cross-linking cell-wall components (Fry and Miller, 1987).

The auxin-stimulated NADH oxidase shows a low degree of detergent stimulation (Morré and Brightman, 1991), one characteristic that distinguishes it from the  $O_2^-$  synthases described in the present paper. Furthermore, the rose cell  $O_2^-$  synthases were not stimulated by 10  $\mu$ M 2,4-D or naphthaleneacetic acid (Table II).

Investigations of blue-light phenomena in plant development have led to the elucidation of a plasma membrane electron transport chain containing a flavin and a Cyt. Irradiation with blue photons leads to changes in absorbance spectra that indicate an initial reduction of the flavin and subsequent reduction of the Cyt. Although measurement of the spectral changes is possible in solutions from which oxygen has not been excluded, the reduction of the Cyt is increased by the inclusion of oxygen-scavenging systems like Glc plus Glc oxidase (Asard et al., 1995). Thus, it is possible that  $O_2$  accepts single electrons from the reduced Cyt and that these studies represent another view of one or both of the  $O_2^-$  synthases.

Neutrophils from mammalian blood produce  $O_2^{-}$  as part of their scavenging function, and this ability is centered in the plasma membrane. The composition of the mammalian NADPH oxidase has been determined: there is a flavocytochrome *b* integral to the membrane and three cytoplasmic proteins, including a *rac*-type GTP-binding protein, that are essential for activity (Abo et al., 1992; Segal and Abo, 1993). Like the rose enzyme, the neutrophil enzyme is inhibited by DPI, which binds to the flavocytochrome component (Cross and Jones, 1986).

It has also been reported that the mammalian plasma membrane enzyme nitric oxide synthase can form  $O_2^-$  under some conditions (Pou et al., 1992; Culcasi et al., 1994). Nitric oxide synthase is, like the plant  $O_2^-$  synthase reported here, inhibited by low concentrations of DPI, suggesting that there is a possible relationship between the two electron transport systems. However, neither Arg nor *N*-nitroarginine methyl ester had any effect on rose cell  $O_2^-$  synthase, and we have no other evidence connecting either rose cell  $O_2^-$  synthase with nitric oxide synthase.

The use of the lucigenin assay has allowed us to detect  $O_2^-$  synthases that appear to differ from other plant plasma membrane redox systems. The possibility that there are two separate  $O_2^-$ -generating systems must be considered in further studies of  $O_2^-$  synthesis by normal and stressed plant cells.

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