

# Microbial community in a sediment-hosted CO<sub>2</sub> lake of the southern Okinawa Trough hydrothermal system

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Increasing levels of CO<sub>2</sub> in the atmosphere are expected to cause climatic change with negative effects on the earth's ecosystems and human society. Consequently, a variety of CO<sub>2</sub> disposal options are discussed, including injection into the deep ocean. Because the dissolution of CO<sub>2</sub> in seawater will decrease ambient pH considerably, negative consequences for deep-water ecosystems have been predicted. Hence, ecosystems associated with natural CO<sub>2</sub> reservoirs in the deep sea, and the dynamics of gaseous, liquid, and solid CO<sub>2</sub> in such environments, are of great interest to science and society. We report here a biogeochemical and microbiological characterization of a microbial community inhabiting deep-sea sediments overlying a natural CO<sub>2</sub> lake at the Yonaguni Knoll IV hydrothermal field, southern Okinawa Trough. We found high abundances (>10<sup>9</sup> cm<sup>-3</sup>) of microbial cells in sediment pavements above the CO<sub>2</sub> lake, decreasing to strikingly low cell numbers (10<sup>7</sup> cm<sup>-3</sup>) at the liquid CO<sub>2</sub>/CO<sub>2</sub>-hydrate interface. The key groups in these sediments were as follows: (i) the anaerobic methanotrophic archaea ANME-2c and the Eel-2 group of Deltaproteobacteria and (ii) sulfur-metabolizing chemolithotrophs within the Gamma- and Epsilonproteobacteria. The detection of functional genes related to one-carbon assimilation and the presence of highly <sup>13</sup>C-depleted archaeal and bacterial lipid biomarkers suggest that microorganisms assimilating CO<sub>2</sub> and/or CH<sub>4</sub> dominate the liquid CO<sub>2</sub> and CO<sub>2</sub>-hydrate-bearing sediments. Clearly, the Yonaguni Knoll is an exceptional natural laboratory for the study of consequences of CO<sub>2</sub> disposal as well as of natural CO<sub>2</sub> reservoirs as potential microbial habitats on early Earth and other celestial bodies.

anaerobic oxidation of methane | chemolithotroph | CO<sub>2</sub> disposal | CO<sub>2</sub> hydrate | liquid CO<sub>2</sub>

The release of one-carbon compounds (i.e., CO<sub>2</sub> and methane) into the atmosphere due to human activities has been recognized as a major factor causing dramatic climatic change on the Earth. In this century, the increasing concentrations of greenhouse gases are expected to cause warmer surface temperatures at an accelerating rate and subsequent alternation of ecosystems and biogeochemical cycles (1). The emission of massive amounts of CO<sub>2</sub> already has altered the chemistry of surface seawater worldwide (2), and its current concentration exceeds the amount of CO<sub>2</sub> that can be absorbed by the ocean (3). To reduce CO<sub>2</sub> emissions into the atmosphere, a variety of options have been discussed, including the disposal of CO<sub>2</sub> into the deep sea (4, 5). Pilot studies have been carried out to test the long-term disposal of CO<sub>2</sub> as liquid and/or hydrates in the deep sea (6–8). However, the impact of CO<sub>2</sub> disposal on deep-sea ecosystems remains largely unknown. Negative consequences on diversity, abundance, and functioning of deep-water communities have been predicted because the dissolution of CO<sub>2</sub> in seawater will decrease pH considerably (4, 9). It remains unknown whether physicochemical or geobiological sinks for excess

CO<sub>2</sub> may control the pH effect, such as the formation and burial of CO<sub>2</sub> hydrates and the assimilation of CO<sub>2</sub> into microbial biomass.

In deep-sea hydrothermal vent and cold-seep ecosystems, CO<sub>2</sub>, methane, and sulfide emissions support unique microbial ecosystems (e.g., refs. 10–13). Some types of magmatism cause a dramatic CO<sub>2</sub> enrichment in hydrothermal fluids (14). Natural emission of liquid and hydrated CO<sub>2</sub> droplets was first observed at the Izena Hole of the Mid-Okinawa Trough backarc hydrothermal system (15). Buoyant fluids containing 86–91% CO<sub>2</sub>, with H<sub>2</sub>S and CH<sub>4</sub> as residual gases, were released from the seafloor at 1,400 m depth and a bottom water temperature of 3.8°C (15). However, fishery equipment covering the Izena Hole has prevented further investigations of this intriguing site. Very recently, venting of liquid CO<sub>2</sub> (98% CO<sub>2</sub> in droplets) has been observed at the northwest Eifuku submarine volcanoes of the northern Mariana Arc (16). The composition of microbial communities of methane hydrate-bearing sediments has been investigated (e.g., refs. 10 and 11), nothing is known regarding the presence of microbial communities in liquid CO<sub>2</sub> and CO<sub>2</sub>-hydrate environments. Because liquid CO<sub>2</sub> is widely used for dry cleaning clothes for its organic solvent-like properties, the sediments vented by liquid CO<sub>2</sub> may not be habitable. However, astrobiologists have speculated that life could be hosted by and interact with liquid and/or solid CO<sub>2</sub> on Earth and other celestial bodies. Recently, the southern pole on Mars was found to be mainly composed of ice and solid CO<sub>2</sub> (17) and is considered as a potential habitable zone for life on Mars because the extremely high levels of biologically damaging solar UV radiation may be significantly reduced at the polar caps (18). Here we report studies of the diversity and biogeochemistry of the microbial community inhabiting liquid CO<sub>2</sub>/CO<sub>2</sub>-hydrate-bearing deep-sea sediments at the Yonaguni Knoll IV hydrothermal field of the southern Okinawa Trough.

## Results and Discussion

### Sample Collection at the Yonaguni Knoll IV Hydrothermal Field.

During the YK03-05 and YK04-05 expeditions in 2003 and 2004, we investigated the Yonaguni Knoll IV hydrothermal field (24°50.7'N, 122°42.1'E; 1,370–1,385 m water depths) by using the *R/V Yokosuka* and the manned submersible *Shinkai 6500* (Fig. 1A). The hydrothermal field is characterized by two active

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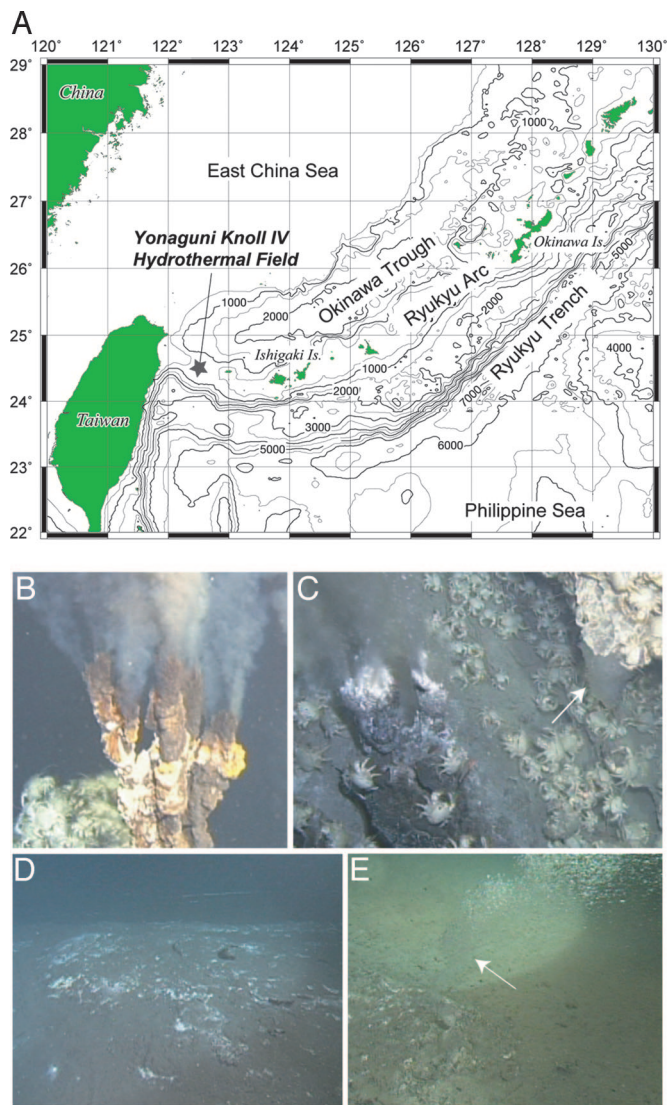
Abbreviation: AOM, anaerobic oxidation of methane.

Data deposition: The 16S rRNA, *mcrA*, and *cbbL* gene sequences reported in this paper have been deposited in the DNA Data Bank of Japan/European Molecular Biology Laboratory/GenBank databases (accession nos. AB252422–AB252455).

See Commentary on page 13903.

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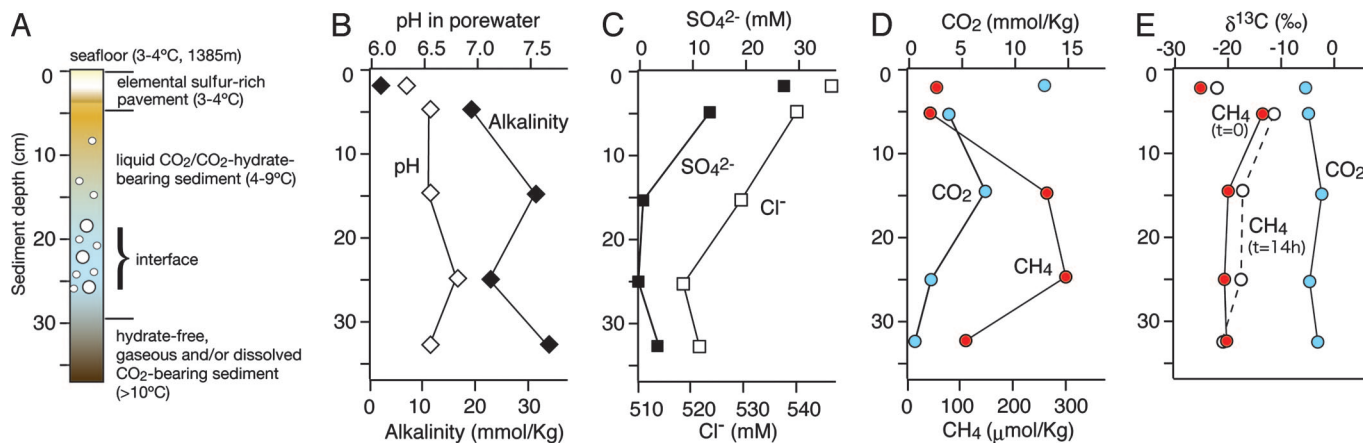
**Fig. 1.** Overview of the Yonaguni Knoll IV hydrothermal field at the southern Okinawa Trough. (A) Location map of the Yonaguni Knoll IV. (B) “Lion chimney,” one of the most active black smoker vents in this field. (C) “Crystal chimney,” one of the vapor-rich clear smoker vents adjacent to the Tiger black smoker. The emission of liquid  $\text{CO}_2$  droplets was observed in close proximity. (D) White patchy area, “ $\text{CO}_2$ -hydrate zone,”  $\approx 50$  m southward from the Tiger chimney. (E) Continuous emission of liquid  $\text{CO}_2$  droplets from the “ $\text{CO}_2$  lake” overlying hydrates. White arrows in C and E indicate the emission of liquid  $\text{CO}_2$  droplets. The video of these hydrothermal events recorded by DSV *Shinkai 6500* is available as Movie 1, which is published as supporting information on the PNAS web site.

black smoker vents (Tiger and Lion chimneys) (Fig. 1B), with vent fluid temperatures up to  $323^\circ\text{C}$ . Near the black smokers, vapor-rich clear venting fluids and small liquid  $\text{CO}_2$  droplets were observed emanating from the sediment (Fig. 1C). Approximately 50 m southward from the hydrothermal vents, down a gentle slope, we found a liquid  $\text{CO}_2$  lake below a 20–40 cm thick cover of sediments (Movie 1). On this seafloor structure, we observed white patches and pavements from which small droplets of liquid  $\text{CO}_2$  were leaking, so-called “ $\text{CO}_2$ -hydrate zone” ( $>200$  m $^2$ , Fig. 1D). Benthic fauna was rarely observed. After the retrieval of a sediment core from one of the white patches, the continuous emission of liquid  $\text{CO}_2$  through the cored hole was observed (Fig. 1E). The *in situ* temperature in this area increased from  $3.9^\circ\text{C}$  in the overlying bottom water to  $9.9^\circ\text{C}$  at

$35$  cm sediment depth, indicating the migration of hydrothermal fluids (Fig. 5, which is published as supporting information on the PNAS web site). At  $1,380$  m depth, liquid  $\text{CO}_2$  is less dense than water (6). Hence, the question arises as to how gaseous/liquid  $\text{CO}_2$  can accumulate into the lake. At  $13.5$  MPa pressure, hydrate stability is reached at temperatures  $<10^\circ\text{C}$  (6, 9, 15). Hence, at *in situ* bottom temperatures of  $4^\circ\text{C}$ , liquid  $\text{CO}_2$  may react with seawater to form a solid ice-like hydrate ( $\text{CO}_2\cdot 6\text{H}_2\text{O}$ ) that, because of its higher density, can cap the liquid  $\text{CO}_2$  reservoir (15). Indeed, our visual inspection indicated a layer of  $\text{CO}_2$  hydrates ( $<10$  cm thickness) below the sediment cover, which likely acts as a cap structure for the underlying liquid  $\text{CO}_2$  lake (Fig. 2A and Movie 1). To investigate the distribution and composition of microbial communities in the liquid  $\text{CO}_2$ / $\text{CO}_2$ -hydrate-bearing sediments, representative samples of sediments (core 819NK, 35 cm in length) and seafloor pavement (819p, 10 cm in thickness), covering the liquid  $\text{CO}_2$ / $\text{CO}_2$ -hydrate transition zone (Fig. 2A), were collected for geochemical and microbiological analyses. Additionally, a sediment core was taken 50 m away from the  $\text{CO}_2$ -hydrate zone as a reference (core 818).

**Chemical Characteristics.** The pore water chemistry of the cores collected from the sediments overlying the  $\text{CO}_2$  lake showed distinct differences to sediments taken outside the hydrate/liquid  $\text{CO}_2$  zone. Alkalinity in core 819NK ( $>20$  mmol/kg) was one order of magnitude higher than that in the reference cores ( $<3$  mmol/kg) (Fig. 2B), reflecting the migration of  $\text{CO}_2$ -rich fluids into the sediment cover of the  $\text{CO}_2$  lake. Also, the concurrent decrease in sulfate and chloride pore water concentrations of core 819NK with depth (Fig. 2B) indicates upward or lateral fluid migration but also could be due to melting of  $\text{CO}_2$  hydrates upon recovery. The pH measured onboard in pore water of core 819NK and pavement samples were 6.6 and 6.3, respectively (Fig. 2B), in contrast to a pH of  $\approx 7.3$  outside of the  $\text{CO}_2$ -hydrate zone in the Yonaguni Knoll IV hydrothermal field (data not shown). However, *in situ* pH at the liquid  $\text{CO}_2$  interface below the sediment cover may be considerably lower than onboard results and theoretically may reach pH 4 taking into account the alkalinity,  $\text{CO}_2$  concentration, and ambient pressure of the system (19). We measured  $>300$  mM  $\text{CO}_2$  in vent fluids and  $>20$  mM  $\text{CO}_2$  in pore water of sediments at the Yonaguni Knoll hydrothermal system (ref. 20; see Fig. 2D). The emanating liquid  $\text{CO}_2$  droplets collected *in situ* by a gas-tight water sampling system (21) were composed mostly of  $\text{CO}_2$  containing a small amount of methane ( $\text{CO}_2$ ,  $85.10 \pm 2.12\%$ ;  $\text{CH}_4$ ,  $13.95 \pm 2.05\%$ ;  $n = 2$ ), with hydrogen concentrations below the detection limit ( $<0.01$  mM). The stable carbon isotopic compositions of  $\text{CO}_2$  and  $\text{CH}_4$  in these droplets were  $-7.0\text{‰}$  and  $-26.4\text{‰}$ , respectively, which were very similar to those from the Tiger chimney vent fluids ( $-7.6\text{‰}$  and  $-26.3\text{‰}$ , respectively). These similar isotopic values indicate that both  $\text{CO}_2$  and  $\text{CH}_4$  in the  $\text{CO}_2$ -hydrate zone have the same origin as in the hydrothermal fluids and that they have not been altered significantly by microbial activities [e.g., by methanogenesis, and/or anaerobic oxidation of methane (AOM)].

X-ray diffraction (XRD) analysis showed that the sediments and pavements in the  $\text{CO}_2$ -hydrate zone consisted mainly of elemental sulfur, quartz, and montmorillonite. They were devoid of carbonates, probably due to the low pH of pore water in sediments of the  $\text{CO}_2$ -hydrate zone. Average organic carbon content in pavement and sediment core samples was  $1.12 \pm 0.30\%$  and  $0.76 \pm 0.02\%$  [wt/wt], respectively. XRD analysis and carbon, nitrogen, and sulfur analysis with an elemental analyzer showed large amounts of elemental sulfur [ $15 \approx 60\%$  (wt/wt) as total sulfur compounds; Fig. 5] in the samples, which may have been deposited by hydrothermal processes and associated  $\text{CO}_2$  migration. The microbial oxidation of hydrothermally derived hydrogen sulfide also may contribute to sulfur deposition in the



**Fig. 2.** Scheme of the examined sediment sample from the “CO<sub>2</sub>-hydrate zone” and related geochemical depth profiles. (A) Location of liquid CO<sub>2</sub>/CO<sub>2</sub>-hydrate interface, sediment cover, and pavement in core 819NK, and *in situ* temperatures. (B) Profiles of pH (open diamonds) and alkalinity (as CaCO<sub>3</sub>, filled diamonds) in pore waters. (C) Concentrations of SO<sub>4</sub><sup>2-</sup> (filled squares) and Cl<sup>-</sup> (open squares) in pore water. (D) Concentration of methane and CO<sub>2</sub> in headspace. (E) Carbon isotopic compositions of methane (red circles, *t* = 0; white circles, *t* = 14 h) and CO<sub>2</sub> (light blue circles) in headspace. The data from top layer (single points) and sediment column (connected by lines) correspond to 819 pavement and 819NK sediment core, respectively, both obtained from the CO<sub>2</sub>-hydrate zone.

sediment cover above the CO<sub>2</sub> reservoir. Further investigations are needed to understand whether microbial processes contribute to sediment consolidation and pavement formation and, hence, to the capping of the liquid CO<sub>2</sub> reservoir in this zone.

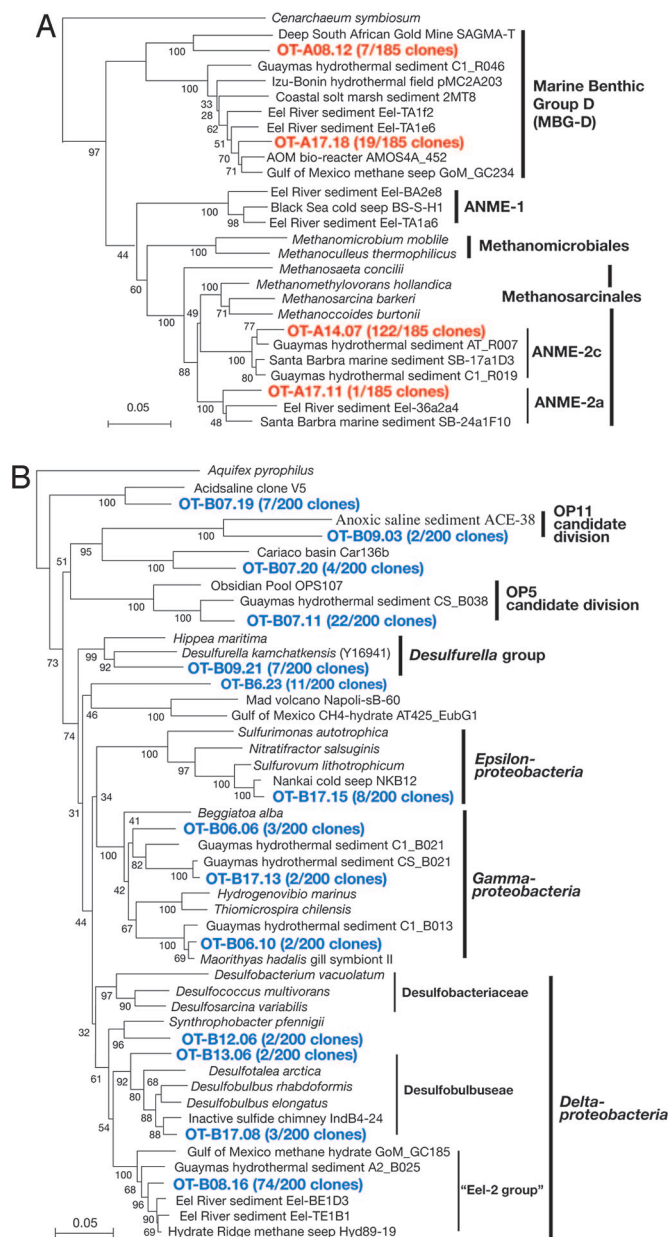
**Cell Abundance and Archaea Population.** Acridine orange direct counts of microbial cells revealed that the pavement sample harbored >10<sup>9</sup> cells·cm<sup>-3</sup> integrated over a 10-cm sediment depth. In contrast, sediments at the liquid CO<sub>2</sub>/CO<sub>2</sub>-hydrate interface in core 819NK contained only ≈10<sup>7</sup> cells·cm<sup>-3</sup>. Notably, all microbial parameters indicate a decline toward the liquid CO<sub>2</sub> interface (Fig. 5). Quantitative real-time PCR analysis of 16S rRNA genes by using a specific probe and primer set for Archaea (22) indicated that archaeal 16S rRNA genes were abundant in the upper core 819NK sediment and pavement samples (maximum 30% of total 16S rRNA genes; see Fig. 5) but decreased considerably toward the interface with liquid CO<sub>2</sub>/CO<sub>2</sub> hydrates.

**Diversity of Microbial Community.** Sequence analysis revealed that the most frequently detected phylotypes belonged to the ANME-2 group archaea and sulfate-reducing bacteria (SRB) belonging to a cluster of Deltaproteobacteria previously found associated with cold methane-seep environments (10, 12, 23, 24). The representative sequence of OT-A14.07 (122 of 185 total archaea clones with similarity cutoff at 97%) was affiliated with the ANME-2c group and that of OT-B08.16 (74 of 200 total bacteria clones with similarity cut off at 97%) belonged to the Eel-2 group (alternatively classified as Seep SRB-2 group) within the Deltaproteobacteria (ref. 24; Fig. 3). Sequences related to the genus *Desulfobulbus*, which contains potential AOM-associated sulfate-reducing bacterias (25), also were detected, albeit as minor percentages of the total bacterial sequences. Interestingly, the composition of the sulfate reducer guild in sediments containing liquid CO<sub>2</sub>/CO<sub>2</sub> hydrate substantially was different from that previously reported for cold-seep sediments accompanied with high methane fluxes. For example, the members of the Eel-2 group, which dominated here have been only minor components of cold-seep communities (25–27), and sequences affiliated with the *Desulfococcus/Desulfosarcina* group, which is the typical consortium partner of ANME I and II cells (10, 12, 23–27), never were detected.

We also studied the methyl coenzyme M reductase  $\alpha$ -subunit gene (*mcrA*), a key gene of methanogenesis and anaerobic meth-

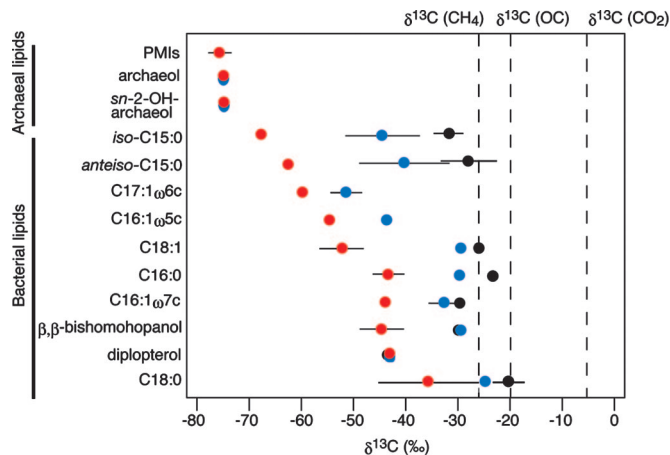
anotrophy (12, 28). Compared with the 16S rRNA gene library, the analysis of the *mcrA* gene clone library revealed a higher diversity of ANME phylotypes, most of which belonged to ANME-2 *mcrA* group-d and -c (ref. 28; see phylogenetic tree in Fig. 6, which is published as supporting information on the PNAS web site). Neither 16S rRNA nor *mcrA* sequences related to hydrogenotrophic methanogens were detected from the CO<sub>2</sub>-hydrate zone. We tried to cultivate hydrogenotrophic methanogens; however, no growth was observed. These microbiological results are consistent with the fact that the hydrogen concentration in the liquid CO<sub>2</sub> droplets was less than the detection limit.

A number of other phylotypes representing relatively minor components of the microbial community were retrieved from the liquid CO<sub>2</sub>/CO<sub>2</sub>-hydrate-bearing sediments. The second most abundant archaeal phylotype based on 16S rRNA gene sequences belonged to the Marine Benthic Group-D (MBG-D) (Fig. 3A), which is a phylotype that is commonly found to co-occur with ANME sequences in methane-driven seep systems. However, the function of members of this group remains unknown (10, 12, 25). In the bacterial clone libraries, sequences of Gamma- and Epsilonproteobacteria, the OP5 and OP11 candidate divisions, and a few unknown groups were detected. Some of these phylotypes are closely related to previously cultured chemolithotrophs (Fig. 3B). For example, the Epsilonproteobacteria sequences were related closely to sulfur-oxidizing chemolithoautotrophic bacterium *Sulfurovum lithotrophicum* (29) that uses the reductive tricarboxylic acid cycle for carbon assimilation (30, 31). Gammaproteobacteria sequences were related to sulfur- and/or hydrogen-oxidizing chemolithoautotrophs (or chemolithomixotrophs) such as *Beggiatoa*, *Hydrogenovibrio*, and *Thiomicrospira*. In addition, the ribulose 1,5-bisphosphate carboxylase large subunit gene (*cbbL*), which is a key gene for chemoautotrophic CO<sub>2</sub> assimilation via the Calvin-Benson cycle, was amplified successfully from surface sediments of core 819NK sediment and pavement samples. The sequence of OT-cbbL1.07 was the most abundant *cbbL* phylotype (24 of 32 total *cbbL* clones), which is related to the obligatory chemolithoautotrophic sulfur-oxidizing Gammaproteobacteria *Thioalkalivibrio thiocyanoxidans* and *Hydrogenovibrio marinus* (Fig. 7, which is published as supporting information on the PNAS web site). These molecular results suggest the co-occurrence of a diverse group of sulfur-metabolizing chemolithotrophs with the AOM community in the liquid CO<sub>2</sub>/CO<sub>2</sub>-hydrate-bearing marine sediments.



**Fig. 3.** Phylogenetic trees of archaeal (A) and bacterial (B) 16S rRNA gene sequences from the liquid CO<sub>2</sub>-hydrate hydrothermal system as marked in red and blue, respectively. The trees were inferred by neighbor-joining analysis by using manually aligned homologous positions of 16S rRNA sequences. The number of related clones with a 97% similarity sequence cutoff are indicated in parentheses. The results of 100 bootstrap trials are shown at each phylogenetic branch. (Scale bars: 0.05-nt substitution per sequence position.)

**AOM Activity.** To determine whether methane is being produced or consumed, we anaerobically incubated freshly collected sediments from the CO<sub>2</sub>-hydrate zone on board and monitored changes in carbon isotopic composition ( $\delta^{13}\text{C}$ ) of methane over time. In both 819NK sediment core and pavement samples, the headspace methane became slightly enriched in  $^{13}\text{C}$  ( $-1.12 \pm 1.15\text{‰}$ ;  $n = 5$ ) after a 14-h incubation at 4°C (Fig. 2E), indicating that methane is oxidized anaerobically in these sediments. We also measured *in vitro* anaerobic methane consumption coupled to sulfate reduction by incubating sediment slurries diluted with methane-enriched medium for sulfate-reducing bacteria with  $^{35}\text{SO}_4$  and  $^{14}\text{CH}_4$  tracers at 4°C (32). Significant but



**Fig. 4.** Carbon isotopic compositions of archaeal and bacterial lipids extracted from 819p pavements (red) and 819NK core sediments (blue) in the CO<sub>2</sub>-hydrate zone and 818 core sediments (black) obtained from outside of the CO<sub>2</sub>-hydrate zone (>50 m). Carbon isotopic compositions of CO<sub>2</sub> and methane were obtained from liquid CO<sub>2</sub> droplets collected *in situ* by the gas-tight sampler. The  $\delta^{13}\text{C}$  value of algal organic carbon (OC) was calculated from the  $\delta^{13}\text{C}$  value of phytol from core 818, assuming an isotopic offset between phytoplanktonic biomass and phytol of +4‰ (33). Bars indicate SDs ( $n = 2-6$ ).

low AOM [ $1-8 \text{ nmol}\cdot\text{g}$  of dry weight ( $\text{gdw}^{-1}\cdot\text{d}^{-1}$ ) and sulfate reduction rates ( $1-15 \text{ nmol}\cdot\text{gdw}^{-1}\cdot\text{d}^{-1}$ ) were measured in the examined sediments. These combined results indicate that methane oxidation indeed occurs in liquid CO<sub>2</sub>/CO<sub>2</sub>-hydrate-bearing sediments. The effect of pH and CO<sub>2</sub>-fluid migration on AOM and the growth and distribution of methanotrophic consortia above the CO<sub>2</sub> lake needs further investigation.

**Carbon Isotopic Composition of Lipid Biomarkers.** Further evidence for the *in situ* assimilation of CH<sub>4</sub>/CO<sub>2</sub> by the microbial community inhabiting the liquid CO<sub>2</sub>/CO<sub>2</sub>-hydrate-bearing sediments was provided by the stable carbon isotopic composition of specific lipid biomarkers. The  $\delta^{13}\text{C}$  values of components of archaeal origin such as archaeol, *sn*-2-hydroxyarchaeol and 2,6,10,15,19-pentamethylicosanes were strongly  $^{13}\text{C}$ -depleted (Fig. 4), indicating that methane plays a major role for the carbon and energy metabolisms of archaea in this habitat (10, 23, 27, 34-36). A similar  $^{13}\text{C}$  depletion, albeit slightly smaller, also was observed for the  $\delta^{13}\text{C}$  values of *i*-C<sub>15</sub>, *ai*-C<sub>15</sub>, and C<sub>17</sub> fatty acids (Fig. 4), which are known to be abundant in sulfate-reducing bacteria (37, 38). Matching the occurrence of *Desulfobulbus* sequences, we also detected their typical biomarker C<sub>17:1ω6c</sub> (37) in the pavement samples with  $\delta^{13}\text{C}$  values of  $-59.8\text{‰}$ . In accordance with the depth profiles of bacterial counts by acridine orange direct counts, the concentrations of lipid biomarkers derived from organisms associated with AOM was significantly higher in the pavement samples than in the underlying liquid CO<sub>2</sub>-bearing sediments (Fig. 8, which is published as supporting information on the PNAS web site). This decrease in abundance of AOM-related lipid biomarkers could be related to decreasing sulfate availability but also to other factors such as decreasing pH levels in the liquid CO<sub>2</sub>-vented sediments.

Diplopterol, a bacterial biomarker, was detected as a major component in all sediment samples. The  $\delta^{13}\text{C}$  values of diplopterol were  $\approx -43\text{‰}$  (Fig. 4), which is  $^{13}\text{C}$ -depleted relative to *in situ* CO<sub>2</sub> and methane by 36‰ and 17‰, respectively. The  $^{13}\text{C}$ -depleted diplopterol could be derived either from aerobic methanotrophs or yet-unidentified chemolithotrophs (39). However, an aerobic methanotrophic origin for diplopterol is unlikely based on the genetic and organic geochemical data. No phylo-

types capable of aerobic methane oxidation were observed, and no sequences related to the methane monoxygenase (*pmoA* and *mmoX*) genes (15), the key genes in aerobic methanotrophy, were detected. Consequently, we infer that the diplopterol is most likely derived from yet-identified chemolithotrophs. Even C<sub>16:0</sub>, which is a major fatty acid in many bacteria and eukaryotes, was depleted in <sup>13</sup>C (−45‰) relative to CO<sub>2</sub> and methane (Fig. 4). These isotopic values were notably different from those in sediments obtained from outside of the CO<sub>2</sub>-hydrate zone (core 818, see Fig. 4). The strong <sup>13</sup>C depletion of archaeal and bacterial lipids indicates the dominance of microbial populations adapted to exploiting hydrothermal carbon sources in the form of CH<sub>4</sub> and CO<sub>2</sub>.

### Conclusion and Prospects

This investigation shows microbial diversity and function at the interface of liquid CO<sub>2</sub> reservoir covered by CO<sub>2</sub> hydrates in deep-sea sediments. Our findings demonstrate that this extreme habitat can be populated by an indigenous community of microbes assimilating one-carbon compounds and suggest that methane and sulfur oxidations are the main energy providing processes in the methane-containing liquid CO<sub>2</sub>/CO<sub>2</sub>-hydrate system. However, we observed a strong decline in cell numbers and abundance of specific lipid biomarkers toward the liquid CO<sub>2</sub> interface on a scale of decimeters, indicating that these communities and their functioning may be affected by variations in CO<sub>2</sub> concentrations and *in situ* pH. Critical parameters for further investigations are the stability of the liquid/solid CO<sub>2</sub> interface and the role of fluid flow in pH, mass transport, and mineralogy of this hydrothermal system.

### Materials and Methods

**Sample Collection.** During the YK04-05 cruise in May 2004, the liquid CO<sub>2</sub>/CO<sub>2</sub>-hydrate-bearing sediments were collected by 35- or 50-cm-long push cores equipped with multiple temperature probes (interval 5 cm). Release of CO<sub>2</sub> gas bubbles from the sediment cores was observed during the submersible rise to the sea surface as previously described by Sakai *et al.* (15). Emission of liquid CO<sub>2</sub> was observed from a crack in the white-yellowish pavement and was collected by the submersible by using a WHATS gas-tight bottle sampler (21). Sediment cores and pavement samples were subsampled immediately by using 50-ml tip-cut sterilized syringes at 5-cm depth intervals and then prepared for microbiological and geochemical studies in the onboard laboratory.

**Characterization of Chemistry in Pore Water and Liquid CO<sub>2</sub>.** Pore water was obtained from sediments within 3 h after recovery. The innermost part of a sediment core was transferred immediately into airtight 50-ml plastic syringes, and then pore water was extracted by pressure filtration through a 0.45- $\mu$ m pore-size filter by using a stainless steel clamp at 4°C. Cl<sup>−</sup> and SO<sub>4</sub><sup>2−</sup> were measured with an ion chromatograph as described in refs. 12 and 15. Compound-specific concentrations and stable carbon isotopic compositions of CO<sub>2</sub> and methane were measured by an isotope ratio-monitoring gas chromatograph-mass spectrometer (MAT252; Thermo Finnigan, Bremen, Germany) as described in ref. 40.

For incubation of core sediments,  $\approx$ 3 cm<sup>3</sup> sediment samples

were collected by tip-cut syringe into 69 cm<sup>3</sup> glass bottles with N<sub>2</sub> gas in headspace. Each bottle then was sealed with a rubber stopper and incubated at 4°C for 14 h. Before and after incubation, the samples were poisoned with HgCl<sub>2</sub> (6% wt/vol) and stored at 4°C until the measurement of the content and carbon isotopic composition of methane in the headspace.

**DNA Extraction and Molecular Analyses.** DNA was extracted from 10 g of wet sediment by using MoBio Mega-prep Soil DNA Kit (Mo Bio Laboratories, Inc., Solana Beach, CA) according to the manufacturer's instructions and then purified (11). 16S rRNA, *mcrA*, *cbfL*, *pmoA*, and *mmoX* genes were amplified by PCR, and then phylogenetic analyses were carried out (11, 12, 30). The relative abundances of archaeal 16S rRNA genes among total prokaryotic 16S rRNA genes were estimated by the quantitative real-time PCR with domain Archaea-specific TaqMan probe and primer sets according to the protocol described in ref. 22.

**AOM Activity.** To confirm AOM activity in sediments, we measured AOM rates *in vitro* by using anoxic sediment slurries stored at 4°C for a year. The stored slurry samples [ $\approx$ 10–15 ml containing 10% (vol/vol) sediment] were amended with anoxic artificial seawater containing 28 mM sulfate and 300 KPa methane in the headspace and then horizontally incubated at 4°C. After 1 month of preincubation, a significant production of sulfide (0.3–2.3 mM) was observed. The activated slurry samples then were subjected to the measurement of AOM and sulfate reduction rates according to protocols described in refs. 32 and 41. The rates were calculated based on one to six replicates depending on the total slurry volume available.

**Total Organic Carbon, Total Sulfur, and Lipid Analyses.** Contents of total organic carbon and total sulfur were determined by using a carbon, nitrogen, and sulfur analyzer (Carlo Erba, Milan, Italy). The freeze-dried sediments were extracted ultrasonically with 2 $\times$  methanol/2 $\times$  dichloromethane:methanol [1:1(vol/vol)]/2 $\times$  dichloromethane to obtain the total lipid extracts. Aliquots of the total extracts were saponified after the addition of an internal standard (containing 1-nonadecanol, nonadecanoic acid, 5-cholestane, and hexatriacontane) with aqueous 0.5 M KOH in methanol (3 h at 80°C). Nonsaponifiable (neutral lipids) and acid fractions were sequentially extracted with hexane at pH  $\approx$ 14 and 2, respectively. The neutral fractions were silylated with *N,O*-bis(trimethylsilyl)trifluoroacetamide in pyridine and analyzed by GC-MS (Trace GC-MS; Thermo Finnigan) for lipid identification and quantification. Repeated concentration measurements were within  $\pm$ 10%. Compound-specific  $\delta^{13}$ C analyses were performed by using an isotope ratio-monitoring GC-MS system (Delta Plus XP; Thermo Finnigan). The  $\delta^{13}$ C values for individual compounds are the means of duplicate runs ( $\sigma = \pm$ 0.3 to 0.6) expressed versus VPDB.

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