

## Time-Resolved Detection of Conformational Changes in Oat Phytochrome A: Time-Dependent Diffusion

Takeshi Eitoku,\* Xristo Zarate,<sup>†</sup> Gennady V. Kozhukh,<sup>†</sup> Jeong-Il Kim,<sup>†‡§</sup> Pill-Soon Song,<sup>†§¶</sup> and Masahide Terazima\*

\*Department of Chemistry, Graduate School of Science, Kyoto University, Japan; <sup>†</sup>Kumho Life & Environmental Science Laboratory, Gwangju, Korea; <sup>‡</sup>Department of Biotechnology (BK21 program), College of Agriculture and Life Sciences, Chonnam National University, Gwangju, Korea; <sup>§</sup>Environmental Biotechnology National Core Research Center, Gyeongsang National University, Jinju, Korea; and <sup>¶</sup>Faculty of Biotechnology, College of Applied Life Sciences, Cheju University, Jeju, Korea

**ABSTRACT** Conformational changes in oat phytochrome A (phy) in solution after photoexcitation of the red-absorbing form (Pr) were studied in time-domain by the pulsed laser-induced transient grating technique. It was found that the diffusion coefficient ( $D$ ) of far-red-absorbing form (Pfr) of large phy ( $1.3 \times 10^{-11} \text{ m}^2 \text{ s}^{-1}$ ) is markedly reduced compared with that of Pr ( $5.8 \times 10^{-11} \text{ m}^2 \text{ s}^{-1}$ ). This large reduction indicates that the conformation of Pfr is significantly changed from that of Pr, so that the intermolecular interaction with water molecules increases. This change completes within 1 ms after the photoexcitation. On the other hand,  $D$  of Pr of intact phy ( $4.1 \times 10^{-11} \text{ m}^2 \text{ s}^{-1}$ ) first decreases upon photoexcitation to  $0.89 \times 10^{-11} \text{ m}^2 \text{ s}^{-1}$  within 1 ms and then gradually increases with a time constant of 100 ms to the value of Pfr,  $1.7 \times 10^{-11} \text{ m}^2 \text{ s}^{-1}$ . This slower phase suggests that the conformation of the N-terminal region changes with 100 ms to decrease the intermolecular interaction with water after a global change in the large phy region. The increase of  $D$  was interpreted in terms of  $\alpha$ -helix formation in the Pfr form from the random coil structure in the Pr form.

### INTRODUCTION

Phytochrome (phy) is a light sensor protein that controls a variety of light-driven adaptations of green plants including seed germination and flowering (1–3). Phy exists in one of the two stable photochromic forms; one absorbing preferentially in the red region of the visible spectrum with an absorption maximum at  $\lambda_{\text{max}} = 666 \text{ nm}$  (Pr), and the other absorbing in the far-red with  $\lambda_{\text{max}} = 730 \text{ nm}$  (Pfr). Phy consists of the N-terminal domain including the amino-terminal extension sequence ( $\sim 6 \text{ kDa}$ ), the photosensory domain containing chromophore ( $\sim 64 \text{ kDa}$ ), and the C-terminal domain that contains two Per-Arnt-Sim related domains and one histidine kinase-related domain (4). Although intact phy regulates the photoresponse in plants, large phy lacking the N-terminal extension sequence is biologically inactive (5). Recently, by using a C-terminal-truncated phyB sample fused to the nuclear localization signal domain, it was shown that the C-terminal part is not directly involved in phy signal transduction (6). These facts suggest that the conformation of the N-terminal domain is physiologically important. Revealing the photoinduced conformational change upon excitation has been a target of extensive researches in chemistry, biophysics, and photobiology of phy.

The ternary and quaternary structures have been studied by the quasi-elastic light scattering (7), electron micrography (8–10), size exclusion column chromatography (11), circular dichroism (CD) (12–14), and small angle x-ray scattering (SAXS) techniques (15,16). CD spectra of Pr and Pfr states

showed that the  $\alpha$ -helix content of these states is nearly identical for large phy (12), but it increased by  $\sim 5\%$  upon the transformation from Pr to Pfr for intact phy (13). Recently, a SAXS measurement combined with a simulation method revealed a global structural change upon the Pr  $\rightarrow$  Pfr transformation (16). Besides these steady-state measurements, the dynamics of the photoresponse should be essential to understand the reaction. By monitoring the absorption changes upon photoexcitation of Pr, for example, several intermediates have been detected during the Pr  $\rightarrow$  Pfr phototransformation process, and several reaction schemes have been proposed (17–19). Although such information is valuable, we should always be careful with the fact that absorption changes reflect a conformational change only around the chromophore. Most of the conformational changes in the N-terminal or C-terminal regions should be spectrally silent and they are very difficult to be detected spectroscopically. One of the unique techniques is the time-resolved CD method. Indeed, the time-dependence of the CD intensity, which reflects the content of the secondary structure, was observed upon the photoexcitation of Pr (20,21).

Another useful physical property reflecting the conformation of a protein is the diffusion coefficient ( $D$ ), which is a physical quantity indicating the speed of the diffusion. According to the Stokes-Einstein relationship, under a given environment (temperature and viscosity),  $D$  of a molecule reflects the molecular size (22,23). More importantly, however, not only the size, but also the intermolecular interaction, affects the magnitude of  $D$ . If the friction between the diffusing solute and the solvent molecules increases,  $D$  should decrease. Therefore, measurements of  $D$  in time

Submitted July 7, 2006, and accepted for publication August 11, 2006.

Address reprint requests to M. Terazima, Tel.: 81-75-753-4026; E-mail: mterazima@kuchem.kyoto-u.ac.jp.

© 2006 by the Biophysical Society

0006-3495/06/11/3797/08 \$2.00

doi: 10.1529/biophysj.106.092882

domain would be a powerful technique to study the time development of the intermolecular interaction, which is spectrally silent. For example, it has been recently revealed that formation or unfolding of  $\alpha$ -helices change the diffusion of a protein significantly (24,25). Hence, if  $D$  can be detected in time domain, it provides us with the information on the dynamics of protein conformational changes. Although  $D$  has been considered to be a static physical property for a long time, the pulsed laser-induced transient grating (TG) method was recently proven to be a powerful technique to monitor the time-dependence of  $D$ . (24) It enables us to monitor the time-course of the  $D$ -changes after photoexcitation (25).

In this work we adopted TG method to study  $D$ -changes in large phy and intact phy (Fig. 1) upon photoexcitation of the Pr state. We found that  $D$  of the large Pfr-phy species is markedly reduced (by approximately one-quarter) compared with the Pr form. This large reduction was interpreted in terms of significant conformation changes, which increase the intermolecular hydrogen-bonding and changes in the surface roughness. This change completes within 1 ms after the photoexcitation. Very interestingly, we found that  $D$  of intact phy first decreases upon the photoexcitation within 1 ms and then gradually increases with a time constant of 100 ms. This result is interpreted in terms of the  $\alpha$ -helix formation in the N-terminal domain.

## PRINCIPLE

The time-resolved measurement of  $D$  was performed by the TG technique. In the TG method, a sample is photoexcited by an optical interference pattern that is created by the interference of two laser beams (24–29). When the photoexcitation induces a chemical reaction to produce a product from a reactant, the spatial modulations of the concentrations of the reactant and product species leads to a modulation of the refractive index ( $\delta n$ ) and the absorption sinusoidally. This modulation can be monitored by a diffraction of a continuous wave probe beam (TG signal). In this study, we used a wavelength of the probe light such that the change in absorption can be ignored and only the refractive index change is considered.

There are several origins of the refractive index change. Under a weak excitation condition with a nanosecond ex-

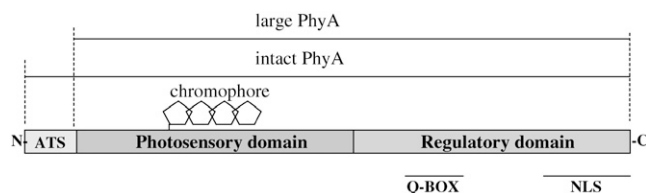


FIGURE 1 Schematic illustration of the intact phy and large phy we used. Some domains are described: ED, N-terminal extended domain; NLS domain, nuclear localization signal domain; and Q-BOX, quail box domain.

citation pulse, the refractive index change comes mainly from the thermal energy releasing (thermal grating:  $\delta n_{th}$ ) and created (or depleted) chemical species by the photoreaction (species grating). Under a weak diffraction condition, the grating intensity ( $I_{TG}$ ) is proportional to the square of the variation in the refractive index. Since the photoexcitation depletes a reactant and produces a reaction product, the phase of the spatial modulation of the product concentration is shifted  $180^\circ$  from that of the reactant. The grating intensity can then be written as

$$I_{TG} \cong \alpha(\delta n_{th}(t) + \delta n_p(t) - \delta n_r(t))^2, \quad (1)$$

where  $\alpha$  is a constant. The terms of  $\delta n_r(>0)$  and  $\delta n_p(>0)$  are, respectively, the refractive index changes due to the changes of the reactant and the product concentrations.

When a product ( $P$ ) is created from a reactant ( $R$ ) upon photoirradiation within a time-resolution of an experimental system,  $R \xrightarrow{h\nu} P$ , the temporal development and the spatial distribution of the concentrations of the reactant and the product can be calculated by solving diffusion equations. In this case, one may find that the time development of the species grating signal is expressed by a biexponential function as (25–27)

$$\delta n_p(t) - \delta n_r(t) = \delta n_r^0 \exp(-D_R q^2 t) - \delta n_p^0 \exp(-D_P q^2 t), \quad (2)$$

where  $q$  is the grating wavenumber, and  $D_R$  and  $D_P$  are the diffusion coefficients of the reactant and the product, respectively. Similarly, the thermal grating signal decays exponentially with a rate constant of  $D_{th} q^2$  ( $D_{th}$ ; thermal diffusivity of the solution). Since the grating wavenumber can be varied by varying the crossing angle of the excitation beams (26,27), one can determine these diffusion coefficients from the decay rates of the TG signals measured at different crossing angles.

When a reactant ( $R$ ) is converted to an intermediate species ( $I$ ) suddenly upon the photoexcitation and this intermediate is transformed to a product ( $P$ ) with a rate constant of  $k$ ,  $R \xrightarrow{h\nu} I \xrightarrow{k} P$ , and the diffusion coefficient of  $I$  ( $D_I$ ) and  $P$  ( $D_P$ ) are different, the averaged  $D$  (apparent  $D$ ) of the photoinduced species should change with a rate constant of  $k$ . The time dependence is expressed by the following equations (24):

$$\delta n_r(t) = \delta n_r^0 \exp(-D_R q^2 t), \quad (3a)$$

$$\delta n_p(t) = \frac{k}{(D_P - D_I) q^2 - k} \delta n_p^0 [\exp\{-(D_I q^2 + k)t\} - \exp(-D_P q^2 t)], \quad (3b)$$

$$\delta n_i(t) = \delta n_i^0 \exp\{-(D_I q^2 + k)t\}. \quad (3c)$$

Using these equations, one may find that the temporal profile of species grating signal can be expressed by

$$I_{TG}(t) = \alpha \left[ -\delta n_R^0 \exp(-D_R q^2 t) + \delta n_I^0 \exp\{-(D_I q^2 + k)t\} + \frac{k}{(D_P - D_I)q^2 - k} \delta n_P^0 [\exp\{-(D_I q^2 + k)t\} - \exp(-D_P q^2 t)] \right]^2. \quad (4)$$

## EXPERIMENTAL

The experimental setup was similar to that reported previously (26–28). Briefly, a laser pulse (wavelength = 610 nm) from a dye laser (HyperDye 300, GSI Lumonics, Watertown, MA) pumped by an excimer laser (Lambda Physik, Ft. Lauderdale, FL; XeCl operation, 308 nm) was used as an excitation beam and a diode laser (835 nm) as a probe beam. The excitation beam was split into two by a beam splitter, and crossed inside a sample cell. The sample is photoexcited by the created interference pattern to induce the refractive index modulation in the sample. A part of the probe beam was diffracted by the modulation (TG signal). The signal was isolated from the excitation laser beam with a glass filter and a pinhole, detected by a photomultiplier tube (model No. R1477, Hamamatsu Photonics, Hamamatsu City, Japan), and recorded by a digital oscilloscope. The spacing of the fringe was measured by the decay rate constant of the thermal grating signal from a calorimetric standard sample (aqueous solution of malachite green), which releases all the photon energy of the excitation as the thermal energy within a time response of our system. All measurements were carried out at room temperature ( $\sim 20^\circ\text{C}$ ). At the intervals of the TG data collection, the infrared light from a diode laser (wavelength = 785 nm) was irradiated to convert the sample state from Pfr to Pr.

Native (124 kDa) oat phytochrome A (phyA) was obtained from etiolated oat seedlings (*Avena sativa*, L. cv. Garry) with a specific absorbance ratio (SAR =  $A_{660}/A_{280}$  ratio) of over 1.0, as previously described (30). Large phytochrome lacking the N-terminal 65 amino-acid residues (114-kDa species) was obtained from the tryptic digests of Pr form of 124-kDa oat phyA (use 11). The digestion of phytochrome was performed at a trypsin/phytochrome ratio of 1:200 (w/w) in 10 mM Tris-HCl buffer (pH 8.0) containing 2 mM EDTA for 15 min reaction at  $37^\circ\text{C}$  under green safelight. Proteolysis was terminated with the addition of 4 mM PMSF. Then, we used the ammonium sulfate back-extraction procedure to separate the tryptic phytochrome species and the intact phytochrome as follows. The phytochrome digests were pelleted by 20 g/100 ml of ammonium sulfate. Using washing procedures with 100 mM phosphate buffer containing 5 mM EDTA, 5 mM 2-ME, and 2 mM PMSF (pH 7.8), we were able to separate the soluble 114-kDa species and insoluble native oat phyA. At the final step, size-exclusion chromatography on Superose 12 HR 10/30 column in 10 mM Tris-HCl, 150 mM NaCl, 2 mM DTT, and 1 mM EDTA (pH 7.8) at  $4^\circ\text{C}$  was performed to purify large phytochrome. The purity of 114-kDa large phytochrome was  $>95\%$  as judged by Coomassie Brilliant Blue staining after SDS-PAGE. The specific absorption ratio  $A_{666}/A_{280}$  was  $>1.0$ .

## RESULTS AND DISCUSSION

### Dynamics of large phy

The TG signal of large phy in the Tris buffer solution was measured upon photoexcitation at 610 nm. A typical signal measured at  $q^2 = 5.26 \times 10^{11} \text{ m}^{-2}$  is depicted in Fig. 2. The signal rose quickly with a time response of our system ( $\sim 20$  ns), decayed, and showed a weak rising component. After this rise, it decayed to the baseline and finally a relatively strong growth-decay component appeared. We found that this signal could be analyzed well with a sum of five exponential functions. For the assignment of these components, we

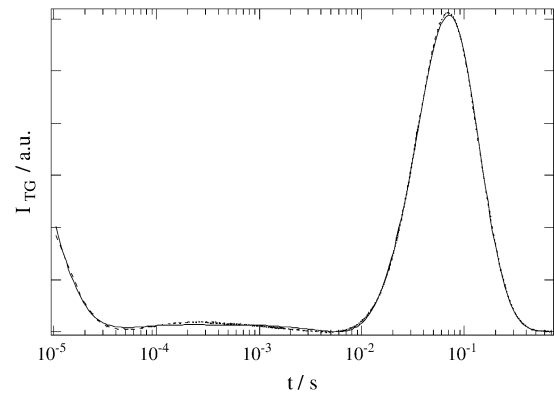


FIGURE 2 Observed TG signal (dashed line) after photoexcitation of large phy in the Tris buffer solution at  $q^2 = 5.3 \times 10^{11} \text{ m}^{-2}$ . The best fitted line with Eqs. 1 and 2 (i.e., Eq. 5) is shown by the solid line.

measured the signals under various grating wavenumbers ( $q^2$ ), and found that three rate constants depended on the  $q^2$ -value, whereas two of them were independent.

The  $q^2$ -dependence indicates that this time-dependence represents a diffusion process. By comparing with the thermal grating signal from a calorimetric reference sample (aqueous solution of malachite green), we found that one of the rate constants agreed with  $D_{th}q^2$  under this experimental condition. This fact indicates that this component is the thermal grating component created by the thermal energy due to the nonradiative transition from the excited state. The temporal profile of the TG signal can be expressed by

$$I_{TG}(t) = \alpha \{ \delta n_{th} \exp(-D_{th}q^2 t) + \delta n_1 \exp(-k_1 t) + \delta n_2 \exp(-k_2 t) + \delta n_3 \exp(-D_3 q^2 t) + \delta n_4 \exp(-D_4 q^2 t) \}^2, \quad (5)$$

where  $k_i$  ( $i = 1$  and  $2$ ) represents the  $q^2$ -independent rate constants ( $k_1 > k_2$ ),  $D_i$  ( $i = 3$  and  $4$ ) denotes the diffusion coefficient ( $D_3 > D_4$ ).

From the fitting of the signals at various  $q^2$ , the  $q^2$ -independent rate constants were determined to be  $k_1 = (40 \mu\text{s})^{-1}$  and  $k_2 = (10 \text{ ms})^{-1}$ . This  $q^2$ -independent dynamics reflect an intrinsic reaction of phy, not a diffusion process. There are two types of contributions in the species grating signal; the population grating, which is associated with an absorption spectrum change, and the volume grating, which comes from the density change due to a molecular volume change. The faster component with the  $40\text{-}\mu\text{s}$  lifetime is close to that reported before by using the beam diffraction method ( $43.4 \mu\text{s}$  at  $20^\circ\text{C}$ ) (31). This signal may originate mainly from the population grating but a possible volume grating contribution cannot be excluded. Indeed, a relatively large energetic relaxation and volume change associated with this process was reported (32).

The slower  $q^2$ -dependent components should represent the molecular diffusion processes. This last rise-decay curve (diffusion peak) was fitted by the last two exponential terms

of Eq. 5. From the  $q^2$ -plot of the rate constants (Fig. 3),  $D$  of the rise and decay components were determined to be  $D_3 = (5.8 \pm 0.2) \times 10^{-11} \text{ m}^2/\text{s}$ , and  $D_4 = (1.3 \pm 0.1) \times 10^{-11} \text{ m}^2/\text{s}$ . Furthermore, the preexponential factors were determined to be  $\delta n_3 < 0$ ,  $\delta n_4 > 0$ , and  $|\delta n_3| < |\delta n_4|$ . The signs of the preexponential factors were determined without ambiguity, using a fact that the sign of the thermal grating signal is negative ( $\delta n_{\text{th}} < 0$ ), and the diffusing species can be assigned from the signs of the preexponential factors and the theoretical equation (Eq. 2). On the basis of the signs, we can easily attribute the  $\delta n_3$  and  $\delta n_4$  terms of Eq. 5 to the  $\delta n_R$  and  $\delta n_P$  terms of Eq. 2, respectively. Hence,  $D_3$  and  $D_4$  should respectively correspond to the diffusion coefficients of the reactant ( $D_{\text{Pr}}^{\text{L}}$ ) and that of the product ( $D_{\text{Pfr}}^{\text{L}}$ ) for large phy.

It is significant that  $D_{\text{Pfr}}^{\text{L}}$  ( $1.3 \times 10^{-11} \text{ m}^2/\text{s}$ ) is drastically smaller than  $D_{\text{Pr}}^{\text{L}}$  ( $5.8 \times 10^{-11} \text{ m}^2/\text{s}$ ). In general,  $D$  is determined by the size and shape of a solute, and varieties of intermolecular interactions. At first, we tried to explain the observed  $D$  from the viewpoint of a radius change of the proteins. According to the Stokes-Einstein relationship,  $D$  is inversely proportional to the radius of the molecule if the other solution properties (e.g., viscosity or temperature) are identical (22). We examined whether or not the reduction in  $D$  can be explained in terms of an increase in molecular radius. There have been attempts to establish the relation between the radius of gyration ( $R_g$ ) and  $D$ . One of the recent successful attempts was reported by Hem and Niemyer (33). They proposed an empirical equation that can reproduce  $D$  (in  $\text{m}^2 \text{ s}^{-1}$ ) of a variety of native proteins from  $R_g$  (in  $\text{\AA}$ ) and the molecular weight ( $M$  in  $\text{g} \cdot \text{mol}^{-1}$ ) in a solution of temperature  $T$  (in K) and viscosity  $\eta$  (in Pa s),

$$D = 6.85 \times 10^{-15} T / \left( \eta \sqrt{M^{1/3} \cdot R_g} \right). \quad (6)$$

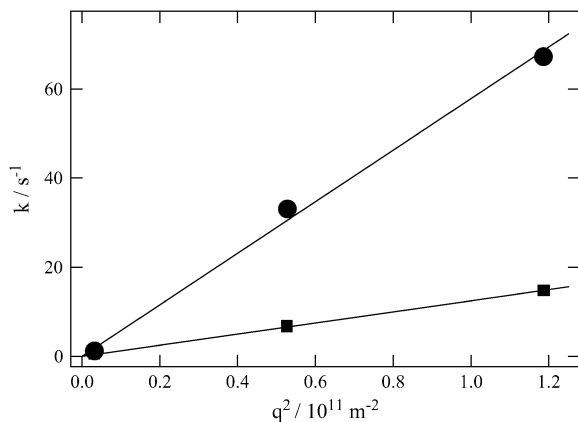


FIGURE 3 Plot of the rate constants ( $k$ ) from the diffusion signal against  $q^2$  (circles, reactant, Pr; squares, product, Pfr) for large phy. The solid lines are the best fitted line with  $k = Dq^2$ . The slopes of the fitted lines represent the diffusion coefficients.

Using the average value of the reported  $R_g$  for each state (5.71 nm for Pr and 6.26 nm for Pfr of Pea large phy) and the molecular weight of the dimer 236 kDa (phy exists as a dimer form) (16,33), one may calculate  $D$  of Pr and Pfr to be  $D_{\text{Pr}} = 3.4 \times 10^{-11} \text{ m}^2 \text{ s}^{-1}$ , and  $D_{\text{Pfr}} = 3.2 \times 10^{-11} \text{ m}^2 \text{ s}^{-1}$ . The difference in the calculated  $D$  between Pr and Pfr is much smaller than that of the experimentally observed value. Therefore, the large reduction in  $D$  upon photoexcitation is not due to the change in  $R_g$ .

A possible origin for the observed reduction in  $D$  may be the change in the protein-protein interaction to produce, e.g., a dimer of the reactant. This possibility was examined by measuring the concentration dependence of the TG signal. If the dimer or aggregates are formed by photoexcitation, the signal shape should be concentration-dependent, which was clearly displayed by other photosensitive proteins (34,35). However, in this phy case, the TG signal was independent of concentration in the range of  $10 \sim 70 \mu\text{M}$ . Therefore, we exclude the aggregation formation for explaining the  $D$ -reduction.

Previously, it was found that the enhanced intermolecular interaction between a protein (or a macromolecule) and water molecules reduced  $D$  significantly. For example, it was reported that  $D$  of the unfolded form of cytochrome c is nearly half of the native form (24). Similarly,  $D$  of polyglutamic acid with the secondary structure of  $\alpha$ -helices decreased upon unfolding of the  $\alpha$ -helices (36). These changes were explained as follows. When an  $\alpha$ -helix is unfolded, the intramolecular hydrogen bonding of the  $\alpha$ -helix is rearranged to the intermolecular one between the protein and water molecules. The intermolecular hydrogen bonding acts as an additional friction for the diffusion to reduce  $D$ . Hence, if the  $\alpha$ -helix content of Pr decreased upon Pfr formation, it could be a cause of the reduction of  $D$ . However, previously reported CD spectra of Pr and Pfr did not show significant differences in the  $\alpha$ -helix content (12). Therefore, the change in the extent of the  $\alpha$ -helix from Pr to Pfr may not be the cause of the  $D$ -change.

Finally, the other possible origin for the reduced  $D$ -value is the change in surface corrugation and roughness of proteins. When protein surface expands so that the solvent molecules interact, the protein molecule consequently feels more frictional drag in the solvent. Its dynamic flexibility is more restrained (37). The surface roughness of a protein molecule is related to  $D$ . Indeed, the larger the surface roughness, the smaller the diffusion coefficient. In this respect, the global structural changes calculated from the SAXS data showed that the Pfr state structure is more elongated and corrugated (16). Main reason for the corrugation is related to the exposure of the nuclear localization signal domain (6,38,39) and/or the structural change of the Quail box domain (40), both of which are thought to have major importance on the biological function of phy molecule.

Dynamical information on the conformation change in phy is important for studying the reaction mechanism of this

transformation. How fast does this change occur after the photoexcitation? It should be noted that the observed signal at various  $q^2$  can be reproduced very well by Eq. 5, which indicates that  $D_{Pr}^L$  and  $D_{Pfr}^L$  are time-independent within the observed time range; i.e., the  $D$ -change from Pr to Pfr has already completed before the diffusion signal appeared, which was  $\sim 1$  ms. This fast change in  $D$  after the photoexcitation can be further confirmed by the analysis described below.

The curvature of a biexponential function at the maximum intensity provides a geometric mean of reactant  $D_R$  and product  $D_P$  at this peak time ( $t_{peak}$ ) (41),

$$\sqrt{D_P D_R} = \frac{1}{q^2} \sqrt{\frac{-I_{TG}''(t_{peak})}{2I_{TG}(t_{peak})}}, \quad (7)$$

where  $I_{TG}(t_{peak})$  is the signal intensity at the peak and  $I_{TG}''(t_{peak})$  is the second derivative at this time. It should be noted that  $D_P$  in this equation does not necessarily mean  $D$  of the final product, but this  $D$  could be a species created from the reactant at this peak time. Hence, this value could be  $D$  of an intermediate species, if it exists. The value of  $(D_P D_R)^{1/2}$  calculated from Eq. 7 is plotted in Fig. 4 against the peak time. The plot shows that mean values are nearly constant during the observation time range indicating that  $D_P$  is time-independent. Therefore, we conclude that the  $D$ -change has already completed before the diffusion signal appeared.

### Dynamics of intact phy

The TG signal of intact phy was similar to that of large phy. The observed signal at  $q^2 = 4.87 \times 10^{11} \text{ m}^{-2}$  is shown in Fig. 5. The signal rose with the time-response of our system, decayed with a rate constant of  $D_{th} q^2$ , and showed a weak rising. Finally it exhibited a strong rise-decay component. From the TG signals at various  $q^2$ , the latest strong rise-decay curve was attributed to the signal representing molecular diffusion. The assignment of the signal component was

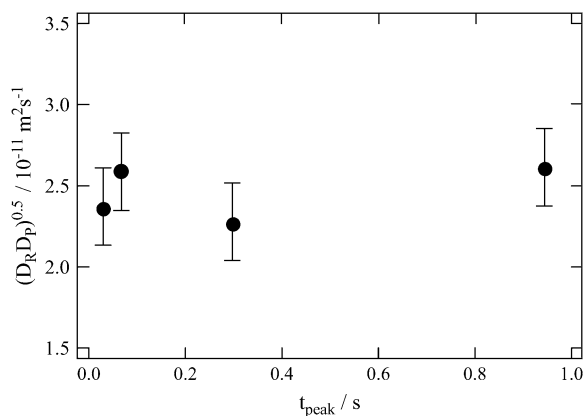


FIGURE 4 Plot of the geometric means of the diffusion coefficients  $(D_P D_R)^{1/2}$  calculated from the curvature of the diffusion peak (Eq. 7) versus the peak time ( $t_{peak}$ ).

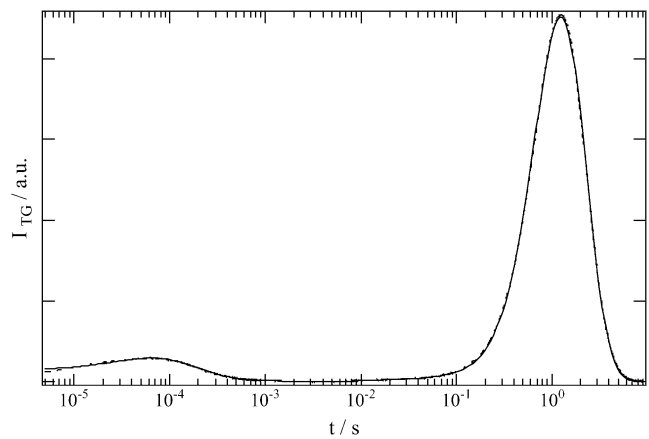


FIGURE 5 Observed TG signal (dotted line) after photoexcitation of intact phy in the Tris buffer at  $q^2 = 3.0 \times 10^{10} \text{ m}^{-2}$ . The best fitted line with Eq. 4 is shown by the solid line.

similar to that of large phy. The rise and decay of the diffusion peak represent the diffusion of Pr and Pfr, respectively. Therefore, the diffusion of Pfr also is slower than that of Pr of the intact phy.

Although these qualitative features of the signal are similar to those of large phy, we noted several differences. First, the diffusion peak intensity compared with the thermal grating or the species grating intensity of the  $40\text{-}\mu\text{s}$  component is much stronger than that of large phy. Second, most interestingly, the diffusion signal cannot be reproduced by Eq. 5. The fact that the diffusion peak cannot be fitted by the biexponential function may be explained in two ways: there may be more than two species involved in the diffusion signal, or apparent  $D$  is changing during the observation time range.

If the profile represents only the diffusion process, the time-dependence should be expressed by a combination of terms of  $\exp(-Dq^2 t)$  (e.g., Eq. 2). In this case, if the signals measured at various  $q^2$  are plotted against  $q^2 t$ , the signals should be identical. However, Fig. 6 shows that the signals are totally different depending on  $q^2$ -value. Therefore, the failure of the biexponential function cannot be explained by simply adding more diffusion terms; instead, it indicates that  $D$  is time-dependent.

The time-dependent  $D$  was further confirmed by the curvature analysis of the diffusion peak (Eq. 7). Fig. 7 depicts the geometric mean of  $D$  ( $(D_P D_R)^{1/2}$ ) against the peak time. As mentioned above, this  $D_P$  is  $D$  of the photoinduced species existing at this time. Interestingly, the mean value of  $D$  gradually increases with time. This feature strongly suggests that  $D$  first decreases upon photoexcitation and then gradually increases during the observation time range. This fact indicates that there is an intermediate species having different  $D$ -value.

The observed signal should be analyzed based on the time-dependent  $D$  model such as Eq. 4. However, since there are

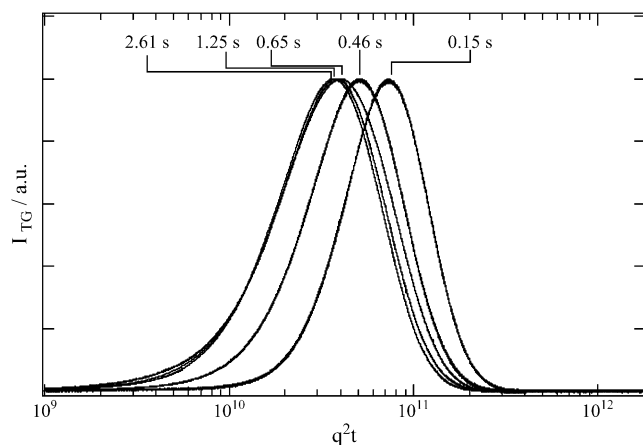


FIGURE 6 TG signals of intact phy measured at  $q^2$  of  $4.9 \times 10^{11}$ ,  $1.1 \times 10^{11}$ ,  $6.2 \times 10^{10}$ ,  $3.0 \times 10^{10}$ , and  $1.5 \times 10^{10} \text{ m}^{-2}$  (from right to left of the curves). The times of the peak after the photoexcitation are indicated in the figure. The signals are normalized at the peak intensity and plotted against  $q^2 t$  to show the temporal changes of  $D$ .

many parameters in Eq. 4, the parameters cannot be determined uniquely by the fitting. Hence, we analyzed the signal by the following way. First, it should be noted that the signals plotted against  $q^2 t$  are relatively close to each other for the signals measured at relatively small  $q^2$ . This fact indicates that  $D$  is almost time-independent in a longer time region. From this  $q^2 t$  plot, we first roughly estimated that the time dependence of  $D$  is almost negligible after 0.5 s from the photoexcitation. Hence, the diffusion signals after 0.5 s were analyzed by the time-independent equation, i.e., Eq. 3. From the slope of the  $q^2$  dependence of the rate constants from the biexponential fitting,  $D$  of the reactant (Intact Pr) and the final product (Pfr) are determined to be  $D_{\text{Pr}}^{\text{I}} = (4.1 \pm 0.2) \times 10^{-11} \text{ m}^2/\text{s}$ , and  $D_{\text{Pfr}}^{\text{I}} = (1.7 \pm 0.1) \times 10^{-11} \text{ m}^2/\text{s}$  (Fig. 7). Since the molecular size of intact phy is larger than

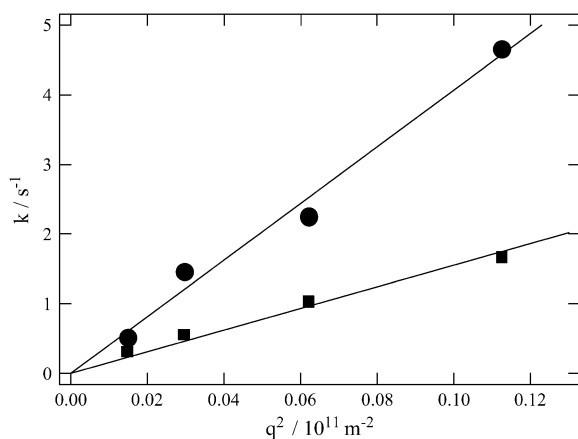


FIGURE 7 Plot of the rate constants ( $k$ ) from the diffusion signal against  $q^2$  (circles, reactant, Pr; squares, product, Pfr) for intact phy. The solid lines are best fitted lines with  $k = Dq^2$ . The slopes of the fitted lines represent the diffusion coefficients.

that of large phy, this slightly smaller  $D_{\text{Pr}}^{\text{I}}$  than  $D_{\text{Pr}}^{\text{L}}$  may be reasonable. Interestingly, on the other hand,  $D_{\text{Pfr}}^{\text{I}}$  is larger than  $D_{\text{Pfr}}^{\text{L}}$ . This difference will be explained later.

The time-dependence of  $D$  indicates that Pr does not convert to Pfr directly with respect to  $D$ , but an intermediate species is involved in the reaction path. If we refer this intermediate species as DI (diffusion detected intermediate), the reaction scheme is  $\text{Pr} \xrightarrow{h\nu} \text{DI} \xrightarrow{k} \text{Pfr}$ .

The peak profile analysis (Fig. 8) indicates that  $D$  of the DI intermediate ( $D_{\text{DI}}^{\text{I}}$ ) is smaller than  $D$  of Pfr. Hence, we analyzed the temporal profile of the TG signal based on the two-state model by assuming  $D_{\text{Pr}}^{\text{I}} > D_{\text{Pfr}}^{\text{I}} > D_{\text{DI}}^{\text{I}}$ . The signal over a wide time range (10 ms–10 s) can be fitted consistently with this model and two adjustable parameters,  $D_{\text{DI}}^{\text{I}}$  and  $k$ . The following parameters were determined:  $D_{\text{DI}}^{\text{I}} = (0.89 \pm 0.09) \times 10^{-11} \text{ m}^2/\text{s}$ , and the rate constant  $k = 10 \pm 3 \text{ s}^{-1}$ . This time constant of 100 ms is sufficiently shorter than the time range used for the biexponential fitting to determine  $D_{\text{Pr}}^{\text{I}}$  and  $D_{\text{Pfr}}^{\text{I}}$  (longer than 0.5 s).

It is interesting to note that the factor of the reduction in  $D$  from Pr to Pfr of large phy ( $D_{\text{Pfr}}^{\text{L}}/D_{\text{Pr}}^{\text{L}} = 22\%$ ) is very close to that of intact phy ( $D_{\text{DI}}^{\text{I}}/D_{\text{Pr}}^{\text{I}} = 22\%$ ) (see Fig. 8). Furthermore, the time range, in which the reduction in  $D$  occurs for intact phy is faster than 1 ms, which agrees with the result of large phy. These facts suggest that the conformational change that causes the  $D$ -reduction is considered to be similar, probably localized in the large phy domain.

Previously, Nishida et al. demonstrated that  $D$  increases with the folding of the  $\alpha$ -helices of cytochrome *c* (24). The increase in  $D$  for the DI  $\rightarrow$  Pfr process suggests that the  $\alpha$ -helix is forming during this process. In the case of intact phy, the time-resolved CD analysis showed the secondary structure change ( $\alpha$ -helix generation) occurred at N-terminal domain and its time constant was 70–160 ms. We suggest that the cause of the  $D$ -increase ( $D_{\text{DI}}^{\text{I}}$  to  $D_{\text{Pfr}}^{\text{I}}$ ) with time constant 100 ms is due to the generation of the secondary structure, e.g.,  $\alpha$ -helices of this protein.

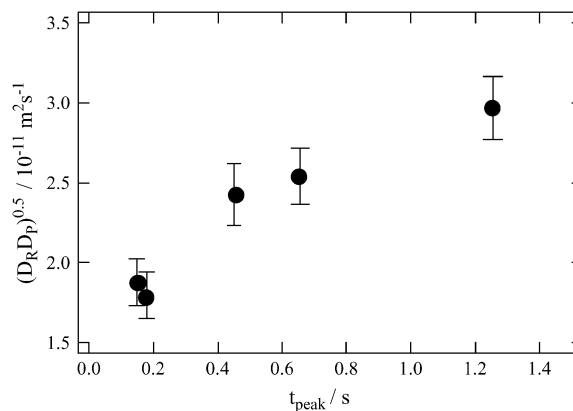


FIGURE 8 Plot of the geometric means of the diffusion coefficients  $(D_{\text{P}} D_{\text{R}})^{1/2}$  calculated from the curvature of the diffusion peak (Eq. 7) versus the peak time ( $t_{\text{peak}}$ ).

## Reaction dynamics of phy

The reaction scheme of phy has been investigated mainly by the transient absorption methods and a variety of schemes have been proposed. For example, Zhang et al. analyzed the data based on a sequential model with time constants of 7.4  $\mu$ s, 89.5  $\mu$ s, 7.6 ms, 42.4 ms, and >266 ms (18). Eilfeld et al. showed kinetics of 35.7  $\mu$ s, 2.7 ms, and 50 ms (17). Losi and Braslavsky analyzed the absorption changes by a parallel model showing time constants of (10  $\mu$ s, 7 ms, 400 ms), and (90  $\mu$ s, 50 ms, 3000 ms) (19). In our TG signal, we observed 40- $\mu$ s and 10-ms population grating kinetics at 20°C. Possible other dynamics could be masked by the strong diffusion signal of the TG method. Our unique observations here are the relatively fast conformational change of the large phy domain and 100-ms dynamics of the N-terminal region, both of which are spectrally silent. In particular, our observations such as the relatively fast (< 1 ms) conformational change and intermolecular interaction change in the large phy domain will be useful information for revealing the reaction dynamics of phy.

Previously,  $D$ -values of Pr and Pfr of large and intact phy were reported by using the quasi-elastic light scattering method at 14°C at relatively dilute condition (0.2 mg/ml = 1.6  $\mu$ M) (7). The results showed that  $D$ -values of Pr and Pfr for large phy were almost identical ( $2.7 \times 10^{-11}$  m<sup>2</sup>/s). The  $D$ -values of the intact phy-Pr and -Pfr species were also similar ( $1.76 \times 10^{-11}$  m<sup>2</sup>/s and  $1.82 \times 10^{-11}$  m<sup>2</sup>/s). The reason for the discrepancy between the previous light-scattering data and our TG data is not clear at present. However, we may consider that the difference could be due to the different principle of the measurements. In the previous light-scattering experiment,  $D$ -values were measured under a steady-state condition with illumination of red or IR light to the sample. If there is a meta-stable intermediate state, phy could equilibrate among various conformational states by the steady-state light illumination. The  $D$ -values measured by the quasi-elastic light-scattering method could represent such a mixture state of several species. On the other hand, our TG measurement detects light-induced  $D$  change directly. Hence, this is a more straightforward detection method. This contradiction should be examined in future by the quasi-elastic light-scattering measurement with highly controlled intensity and the wavelength of the red/IR light. Comparison of  $D$ -values from different methods under well-controlled conditions could reveal more detailed reaction properties of phy.

## CONCLUSION

Photoreactions of large and intact oat phytochrome A were investigated from a viewpoint of time-dependent diffusion coefficients. After the photoexcitation of large phy,  $D$  decreased quickly within 1 ms from  $5.8 \times 10^{-11}$  m<sup>2</sup> s<sup>-1</sup> (Pr) to  $1.3 \times 10^{-11}$  m<sup>2</sup> s<sup>-1</sup> (Pfr). This significant reduction in  $D$  was interpreted in terms of the change in the surface

roughness of phy. This result indicates that the global conformation change from Pr to Pfr completes within 1 ms after the photoexcitation. The TG signal of intact phy was different from large phy in several points. Most significantly,  $D$  of intact phy ( $4.1 \times 10^{-11}$  m<sup>2</sup> s<sup>-1</sup>) initially decreased to  $0.89 \times 10^{-11}$  m<sup>2</sup> s<sup>-1</sup> within 1 ms and gradually increased ( $1.7 \times 10^{-11}$  m<sup>2</sup> s<sup>-1</sup>) with a time constant of 100 ms. It is interesting to note that the initial  $D$  change is similar to that observed for large phy and the second phase is characteristic of intact phy. Hence the conformational change of the large phy domain in intact phy completes quickly within 1 ms. After this change, the increase in  $D$ , which is interpreted in terms of the  $\alpha$ -helix formation in the N-terminal region, takes place with the time constant of 100 ms.

This work is supported by Grants-in-Aid (Nos.13853002 and 15076204) from the Ministry of Education, Science, Sports and Culture in Japan. The work was also supported in part by the Korea Science and Engineering Foundation/Ministry of Science and Technology to the Environmental Biotechnology National Core Research Center (to J.-I. K. and P.-S.S.; grant No. R15-2003-012-01003-0).

## REFERENCES

- Briggs, W. R., and H. V. Rice. 1972. Phytochrome: chemical and physical properties and mechanism of action. *Annu. Rev. Plant Physiol.* 23:293–334.
- Furuya, M. 1987. Phytochrome and Photoregulation in Plants. Academic Press, New York.
- Sineshchekov, V. A. 1995. Photobiophysics and photobiochemistry of the heterogeneous phytochrome system. *Biochim. Biophys. Acta.* 1228: 125–164.
- Song, P.-S. 1988. The molecular topography of phytochrome: chromophore and apoprotein. *J. Photochem. Photobiol. B.* 2:43–57.
- Jordan, E. T., J. R. Cherry, J. M. Walker, and R. D. Viestra. 1996. The amino-terminus of phytochrome A contains two distinct functional domains. *Plant J.* 9:243–257.
- Matsushita, T., N. Mochizuki, and A. Nagatani. 2003. Dimers of the N-terminal domain of phytochrome B are functional in the nucleus. *Nature.* 424:571–574.
- Sarkar, H. K., D. K. Moon, P.-S. Song, T. Chang, and H. Yu. 1984. Tertiary structure of phytochrome probed by quasi-elastic light scattering and rotational relaxation time measurements. *Biochemistry.* 23: 1882–1888.
- Jones, A. M., and H. P. Erickson. 1989. Domain structure of 124-kiloDalton phytochrome from *Avena sativa* visualized by electron micrography. *Photochem. Photobiol.* 49:479–483.
- Tokutomi, S., M. Nakasako, J. Sakai, M. Kataoka, K. T. Yamamoto, M. Wada, F. Tokunaga, and M. Furuya. 1989. A model for the dimeric molecular structure of phytochrome based on small-angle x-ray scattering. *FEBS Lett.* 247:139–142.
- Nakasako, M., M. Wada, S. Tokutomi, K. T. Yamamoto, J. Sakai, M. Kataoka, F. Tokunaga, and M. Furuya. 1990. Quaternary structure of pea phytochrome I dimer studied with small-angle x-ray scattering and rotary-shadowing electron microscopy. *Photochem. Photobiol.* 52: 3–12.
- Lagarias, J. C., and F. M. Mercurio. 1985. Structure function studies on phytochrome. *J. Biol. Chem.* 260:2415–2423.
- Hunt, R. E., and L. H. Pratt. 1981. Physicochemical difference between the red- and the far-red-absorbing forms of phytochrome. *Biochemistry.* 20:941–945.
- Deforce, L., S. Tokutomi, and P.-S. Song. 1994. Phototransformation of pea phytochrome A induces an increase in  $\alpha$ -helical folding of the

- apoprotein: comparison with a Monocot phytochrome A and CD analysis by different methods. *Biochemistry*. 33:4918–4922.
14. Chai, Y. G., P.-S. Song, M. M. Cordonnier, and L. H. Pratt. 1987. A photoreversible circular dichroism spectral change in oat phytochrome is suppressed by a monoclonal antibody that binds near its N-terminus and by chromophore modification. *Biochemistry*. 26:4947–4952.
  15. Tokutomi, S., M. Kataoka, J. Sakai, M. Nakasako, F. Tokunaga, M. Tasumi, and M. Furuya. 1988. Small-angle x-ray scattering studies on the macromolecular structure of the red-light-absorbing form of 121 kDa pea oat phytochrome and its 114 kDa chromopeptide. *Biochim. Biophys. Acta*. 953:297–305.
  16. Nakasako, M., T. Iwata, K. Inoue, and S. Tokutomi. 2005. Light-induced global structural changes in phytochrome A regulating photomorphogenesis in plants. *FEBS J.* 272:603–612.
  17. Eilfeld, P., P. Eilfeld, J. Vogel, and R. Maurer. 1987. Evidence for a sequential pathway from Pr to Pfr of the phototransformation of 124-kDa oat phytochrome. *Photochem. Photobiol.* 45:825–830.
  18. Zhang, C. F., D. L. Farrens, S. C. Bjorling, P.-S. Song, and D. S. Kliger. 1992. Time-resolved absorption studies of native etiolated oat phytochrome. *J. Am. Chem. Soc.* 114:4569–4580.
  19. Losi, A., and S. E. Braslavsky. 2003. The time-resolved thermodynamics of the chromophore-protein interactions in biological photosensors as derived from photothermal measurements. *Phys. Chem. Chem. Phys.* 5:2739–2750.
  20. Chen, E. F., W. Parker, J. W. Lewis, P.-S. Song, and D. S. Kliger. 1993. Time-resolved UV circular dichroism of phytochrome A: folding of the N-terminal region. *J. Am. Chem. Soc.* 115:9854–9855.
  21. Chen, E. F., V. N. Lapko, P.-S. Song, and D. S. Kliger. 1997. Dynamics of the N-terminal  $\alpha$ -helix unfolding in the photoreversion reaction of phytochrome A. *Biochemistry*. 36:4903–4908.
  22. Cussler, E. L., 1984. *Diffusion, Mass Transfer in Fluid Systems*. Cambridge University Press, Cambridge, MA.
  23. Tyrrell, H. J. V., and K. R. Harris. 1984. *Diffusion in Liquids: A Theoretical and Experimental Study*. Butterworth, London, UK.
  24. Nishida, S., T. Nada, and M. Terazima. 2004. Kinetics of intermolecular interaction during protein folding of reduced cytochrome c. *Biophys. J.* 87:2663–2675.
  25. Eitoku, T., Y. Nakasone, D. Matsuoka, S. Tokutomi, and M. Terazima. 2005. Conformational dynamics of phototropin 2 LOV2 domain with the linker upon photoexcitation. *J. Am. Chem. Soc.* 127:13238–13244.
  26. Terazima, M., and N. Hirota. 1993. Translational diffusion of a transient radical studied by the transient grating method, pyrazinyl radical in 2-propanol. *J. Chem. Phys.* 98:6257–6262.
  27. Terazima, M., K. Okamoto, and N. Hirota. 1995. Translational diffusion of transient radicals created by the photoinduced hydrogen abstraction reaction in solution—anomalous size dependence in the radical diffusion. *J. Chem. Phys.* 102:2506–2515.
  28. Terazima, M. 2000. Is the translational diffusion of organic radicals different from that of closed-shell molecules? *Acc. Chem. Res.* 33: 687–694.
  29. Nada, T., and M. Terazima. 2003. A novel methods for study of protein folding kinetics by monitoring diffusion coefficient in time domain. *Biophys. J.* 85:1876–1881.
  30. Lapko, V. N., and P.-S. Song. 1995. A simple and improved method of isolation and purification for native oat phytochrome. *Photochem. Photobiol.* 62:194–198.
  31. Lagarias, J. C., and F. M. Mercurio. 1985. Structure function studies on phytochrome. *J. Biol. Chem.* 260:2415–2423.
  32. Michler, I., and S. E. Braslavsky. 2001. Time-resolved thermodynamic analysis of the oat phytochrome A phototransformation. A photothermal beam deflection study. *Photochem. Photobiol.* 74:624–635.
  33. Hem, L. H., and B. Niemeyer. 2003. A novel correlation for protein diffusion coefficients based on molecular weight and radius of gyration. *Biotechnol. Prog.* 19:544–548.
  34. Hazra, P., K. Inoue, W. Laan, J. Hellingwerf, and M. Terazima. 2006. Tetramer formation kinetics in the signaling state of AppA monitored by the time-resolved diffusion. *Biophys. J.* 91:654–661.
  35. Nakasone, Y., T. Eitoku, D. Matsuoka, S. Tokutomi, and M. Terazima. 2006. Kinetic measurement of transient dimerization and dissociation reactions of *Arabidopsis* phototropin 1 LOV2 domain. *Biophys. J.* 91: 645–653.
  36. Inoue, K., N. Baden, and M. Terazima. 2005. Diffusion coefficient and the secondary structure of poly-L-glutamic acid in aqueous solution. *J. Phys. Chem. B.* 109:22623–22628.
  37. Choi, J. H., H. Kim, and S. Lee. 1998. Correlation dimension as a measure of surface roughness of protein molecules. *J. Chem. Phys.* 109:7001–7004.
  38. Kircher, S., L. Kozma-Bognar, L. Kim, E. Adam, K. Harter, E. Schafer, and F. Nagy. 1999. Light quality-dependent nuclear import of the plant photoreceptors phytochrome A and B. *Plant Cell*. 11:1445–1456.
  39. Sakamoto, K., and A. Nagatani. 1996. Nuclear localization activity of phytochrome B. *Plant J.* 10:859–868.
  40. Quail, P. H., M. T. Boylan, B. M. Parks, T. W. Short, Y. Xu, and D. Wagner. 1995. Phytochromes—photosensory perception and signal-transduction. *Science*. 268:675–680.
  41. Spiegel, D. R., A. H. Marshall, N. T. Jukam, H. S. Park, and T. Chang. 1998. Measurement of mass diffusion coefficients using nonexponential forced Rayleigh scattering signals. *J. Chem. Phys.* 109:267–274.