# Structural Basis for Thermostability of  $\beta$ -Glycosidase from the Thermophilic Eubacterium *Thermus nonproteolyticus* HG102

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The three-dimensional structure of a thermostable  $\beta$ -glycosidase (Gly<sub>Tn</sub>) from the thermophilic eubacterium *Thermus nonproteolyticus* **HG102 was determined at a resolution of 2.4 A˚. The core of the structure adopts the**  $(\beta \alpha)_8$  barrel fold. The sequence alignments and the positions of the two Glu residues in the active center indicate that Gly<sub>Tn</sub> belongs to the glycosyl hydrolases of retaining family 1. We have analyzed the structural features of Gly<sub>Tn</sub> related to the thermostability and compared its structure with those of other mesophilic **glycosidases from plants, eubacteria, and hyperthermophilic enzymes from archaea. Several possible features** contributing to the thermostability of Gly<sub>Tn</sub> were elucidated.

Glycosyl hydrolases catalyze the hydrolytic cleavage of the glycosidic bonds between two or more carbohydrates and between carbohydrate and noncarbohydrate moieties. They are ubiquitous enzymes that have been isolated and characterized from various organisms (archaea, eubacteria, and eukarya). The glycosyl hydrolases have been classified into over 80 families according to their amino acid sequence homology and structural similarities rather than substrate selectivity. Some of the families can be grouped into "clans," because the folds of their proteins are better conserved than their sequences (23). Families 1, 2, 5, 10, 17, 26, 30, 35, 39, 42, 51, 53, 59, 72, 79, and 86 are grouped into a superfamily, clan GH-A. Members of this superfamily adopt a  $(\beta \alpha)_8$  barrel fold (23). The  $\beta$ -glycosidases in family 1 constitute a major group among glycosyl hydrolases. They are characterized by broad substrate specificities, which make them potential tools for several applications  $(24)$ . In this regard,  $\beta$ -glycosidases from thermophilic sources are particularly attractive because of their biotechnological advantages for many stabilized biocatalysts. Furthermore, study of these  $\beta$ -glycosidases may contribute to a better understanding of the structure-function relationships of thermophilic enzymes by comparisons of their properties with those of mesophilic enzymes (47).

The thermostable  $\beta$ -glycosidase Gly<sub>Tn</sub> was produced by the thermophilic eubacterium *Thermus nonproteolyticus* HG102, which was isolated from a hot spring in Guangdong Province, southern China (8). The gene coding for Gly<sub>Tn</sub> in *T. nonproteolyticus* HG102 (GenBank accession number AF225213) has been cloned and expressed in *Escherichia coli*, and the recombinant enzyme was characterized (21).  $\text{Gly}_{\text{Th}}$  belongs to glycosyl hydrolase family 1. It has a broad β-glycosidase activity, and analysis of its substrate specificity revealed that it

prefers  $\beta$ -D-glucoside and  $\beta$ -D-fucoside to  $\beta$ -D-galactoside and  $\beta$ -D-mannoside. Gly<sub>Tn</sub> also has transglycosidic activity at high temperature. This enzyme shows optimum activity at 90°C and pH 5.6, with a half-life of 2.5 h at 90°C (21). Besides the --glycosidase gene in *T. nonproteolyticus* HG102, findings for other two β-glycosidase genes from *Thermus* sp. strain Z-1 (GenBank accession number AB034947) and *Thermus thermophilus* (GenBank accession number Y16753) have also been reported  $(14, 46)$ . These three  $\beta$ -glycosidases share high levels of sequence similarity, with nearly 95% of the residues identical.

The structures of  $\beta$ -glucosidase from white clover (Protein Data Bank [PDB] code 1CBG) and 6-phospho-β-galactosidase from *Lactococcus lactis* (PDB code 1PBG) were reported in 1995 (4, 53). Other structures of family 1  $\beta$ -glycosidases reported include those of myrosinase from *Sinapis alba* (PDB code 1MYR),  $\beta$ -glycosidase from *Sulfolobus solfataricus* (1GOW), β-glucosidase from *Bacillus polymyxa* (1BGA), β-glycosidase from *Thermosphaera aggregans* (1QVB), β-glucosidase from *Bacillus circulans* (1QOX), and β-glucosidase from maize (1E1E) (1, 7, 9, 11, 18, 43). All the structures have the same basic  $(\beta\alpha)_8$  barrel fold. The two hyperthermophilic structures 1GOW and 1QVB were from archaea.

The crystal structure of Gly<sub>Tn</sub> from *T. nonproteolyticus* HG102 described here was determined at a resolution of  $2.4 \text{ Å}$ . To our knowledge, this is the first  $\beta$ -glycosidase structure determined from a thermophilic eubacterium. It adopts a  $(\beta\alpha)_8$  barrel fold. The model of Gly<sub>Tn</sub> was compared with other mesophilic structures from plants and eubacteria and hyperthermophilic structures from archaea to elucidate the possible basis of its thermostability.

### **MATERIALS AND METHODS**

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**Expression, purification, and crystallization of Gly<sub>Tn</sub>.** The cloning, expression, purification, and crystallization of  $Gly_{Tn}$  were performed with the methods described before  $(22)$ .

**Data collection and structure refinement.** Using a Weissenberg camera (42), diffraction data were collected on a beamline BL-6B experimental station (Pho-

TABLE 1. Data collection and refinement statistics

Parameter	Value
	100
Unit cell parameters $(\check{A})$	
	66.7
	94.8
	176.7
	41,963
rms deviations from ideal geometry	
	0.007
	1.38
Average temperature factors $(A^2)$	
	29.3
	30.6
	31.4

ton Factory, Ibaraki, Japan). The data were processed with DENZO and SCALEPACK software (36), and the statistics are listed in Table 1.

The structure was determined by the molecular replacement method using the program Molrep in the CCP4 suite (9a). The positions of the two  $\text{Gly}_{\text{Tn}}$  molecules in the asymmetric unit were found with the model of *Bacillus polymyxa* --glucosidase structure (PDB code 1BGA) (43). Using the maximum-likelihood simulated annealing protocol and restraining the noncrystallographic symmetry, the initial model was refined with the program CNS (6). The proper  $\text{Gly}_{\text{Ta}}$ residues were built into the  $\sigma_A$ -weighed 2|Fo|-|Fc| electron density map with program O (25). After several rounds of refinement and model building, the R<sub>free</sub> and R factors dropped to 31.5 and 28.0% in the resolution range of 20.0 to 2.4 Å and there were two regions with poor densities: N terminal 1 to 4 and C terminal 431 to 436. The removal of the noncrystallographic symmetry in the following refinement was validated by the decrease of the  $R_{free}$  value. Water molecules were added to the model at locations with |Fo|-|Fc| densities higher than  $3\sigma$  and hydrogen-bonding stereochemistry. Inclusion of individual temperature factors was validated by a substantial decrease in the value of R<sub>free</sub>. At the end of refinement, the crystallographic R factor was  $23.0\%$ , with an R<sub>free</sub> value of 27.2%. The stereochemistry of the final model was analyzed with PRO-CHECK (29).

**Analysis of parameters affecting protein thermal stability. (i) Secondary** structure content. The  $\alpha$ -helix content of a protein is the percentage of residues that show  $\alpha$ -helical conformations. The  $\beta$ -sheet content is the percentage of residues that show  $\beta$ -sheet conformations. The corresponding dictionary of protein secondary structure file (26) was used to identify the residues with  $\alpha$ -helical and  $\beta$ -sheet conformations.

**(ii) Surface areas and cavities.** The total and hydrophobic accessible surface areas (ASA) were calculated with the program NACCESS (S. J. Hubbard and J. M. Thornton, NACCESS computer program, Department of Biochemistry and Molecular Biology, University College London, London, United Kingdom, 1993) with a probe radius of  $1.4 \text{ Å}$ . An output file containing summed atomic ASA over each residue was used. The number and volume of cavities were calculated with the program VOIDOO (26a).

**(iii) Hydrogen bonds.** The WHATIF program was used to identify all hydrogen bonds in the structures (51).

**(iv) Ion pairs.** Ion contacts were evaluated with the program CONTACT in CCP4 suite (9a). The ion pair was inferred when Asp or Glu side chain carbonyl oxygen atoms were found to be within 4.0 Å from the nitrogen atoms in Arg, Lys, and His side chains (3).

**PDB accession code.** The coordinates of the structure and the structure factor file have been deposited in the Protein Data Bank (PDB) under accession code 1NP2.

# **RESULTS AND DISCUSSION**

**Quality of the model.** The final model contains two molecules (A and B) in the asymmetric unit, including 6,824 nonhydrogen protein atoms and 334 water molecules. In both molecules A and B, the N-terminal 1 to 4 and C-terminal 431 to 436 residues were not defined in the election density maps, which means that these regions are disordered. The final R factor was 23.0% for reflections in the resolution range of 20.0 to 2.4 Å. The  $R_{\text{free}}$  value for 5% of the total reflections was 27.2%. The model has good stereochemistry, with root mean square (rms) deviations of  $0.007 \text{ Å}$  on bond length and  $1.38^{\circ}$  on bond angle (Table 1).

**Structural description and comparison with other family 1 -glycosidases.** Because the equivalent 426 Ca atoms of molecules A and B are in good agreement (rms value of  $0.53 \text{ Å}$ ) after superimposition, the following description and comparison were based on molecule A.  $Gly_{Tn}$  adopts the expected topology of a single  $(\beta/\alpha)_{8}$  barrel fold (TIM barrel) (Fig. 1). Additional secondary structure units were inserted into the connections between the  $\beta$ -strand and  $\alpha$ -helix in the  $(\beta/\alpha)$ repeat (Fig. 2). The longest connection (from residue 14 to 54 between  $\beta$ -strand 1 and  $\alpha$ -helix 1) contains several turns and two helices. The connections between  $(\beta/\alpha)$  repeats are rather short, with the exception of a 30-residue connection between



FIG. 1. The overall fold structure of Gly<sub>Tn</sub>. This figure and Fig. 3 were made with the programs MOLSCRIPT (28) and RASTER3D (33).



FIG. 2. Sequence alignments of the nine β-glycosidases from family 1 with known structures. The figure was produced with ESPript (17).

 $\alpha$ -helix 5 and  $\beta$ -strand 6, and there is a short  $\alpha$ -helix in the connection. Compared with the more compact bottom half of the barrel ( $\beta$ -strand N-terminus direction), the top half ( $\beta$ strand C-terminus direction) is loose and four loops on that side form the gate to the active site. These four loops are composed of residues 36 to 54, 175 to 184, 292 to 315, and 386 to 403.

The three-dimensional structure of  $\mathrm{Gly_{Tn}}$ , which is the first such structure determined from thermostable eubacteria, shows high similarities in overall structure with eight other members of the  $\beta$ -glycosidases of family 1, although the sequence identities between  $\text{Gly}_{\text{Tr}}$  and those eight members range from 26 to 47%. The TIM barrel folds in these structures were highly conserved (Table 2). The main differences among them are at the level of the connections that link the  $\beta$ -strand and  $\alpha$ -helix in the  $\beta/\alpha$  unit. It was also evident at the level of amino acid sequence when the sequences were aligned (Fig. 2). All eight family  $1 \beta$ -glycosidases with known structures were organized as dimers (1E1E, 1CBG, 1MYR, and 1PBG), tetramers (1QVB and 1GOW) or octamers (1BGA and 1QOX).  $Gly_{Tn}$  exists in the form of monomeric enzyme, which has been demonstrated by the molecular-mass estimation of about 50 kDa on the native enzyme by a gel filtration method (21). The surface area that is accessible to solvent  $(1.4 \text{ Å probe radius})$  of one Gly<sub>Tn</sub> molecule is about 16,000  $\AA^2$ . The surface area buried on the interface of the two molecules A and B in the asymmetric unit is about 1,200  $\AA$ <sup>2</sup> (600  $\AA$ <sup>2</sup> per molecule). It corresponds to about 3.7% of the solvent-exposed surface of one molecule.

**Active site.** Enzymatic hydrolysis of glycosidic bonds can be performed via two major mechanisms, giving rise to either an overall retention, or an overall inversion, of the anomeric configuration (12). Two critical residues (a proton donor and a nucleophile or base) are required for this reaction (12). Catal $y$ sis by retaining family 1  $\beta$ -glycosidases proceeds via a doubledisplacement mechanism, and the active site contains a pair of carboxylic acids that are about  $5.5 \text{ Å}$  apart. The two motifs  $T(F/L/M)NE(P/L/I)$  and  $-(I/V)TENG$  (involved in glycone binding and enzymatic hydrolysis of glycosidic bonds within the active site) are highly conserved (55). In the structure of  $\text{Gly}_{\text{Tn}}$ , the acid catalyst Glu164 in motif –TLNEP and the nucleophilic Glu338 in motif  $-ITERG$  were found at  $\beta$ -strands 4 and 7, respectively (Fig. 1, 2). The distance between the  $C_8$  atoms of

TABLE 2. Structural comparisons for TIM barrel folds

Enzyme	rms value $(\AA)^a$	Sequence identity $(\% )$				
1BGA	0.99	45				
1PBG	1.32	34				
1QOX	1.16	47				
1CBG	1.30	35				
1E1E	1.29	28				
1MYR	1.14	28				
1GOW	1.31	27				
1QVB	1.49	26				

*<sup>a</sup>* The rms values were calculated by a least-squares fit after superimposition of the 184 Ca atoms in the TIM barrel folds.

TABLE 3. Amino acid compositions and secondary structures

Enzyme		No. $(\%)$ of amino acids														
	Ala	<b>Cys</b>	Asp	Glu	Phe	Glv	His	<b>Ile</b>	Lvs	Leu	Met	Asn	Pro	Gln	Arg	Ser
$\mathrm{Gly}_{\mathrm{Tn}}$						56 (12.84) 2 (0.46) 22 (5.05) 34 (7.80) 19 (4.36) 40 (9.17) 12 (2.75) 14 (3.21) 4 (0.92) 44 (10.09)							$4(0.92)$ $9(2.06)$ $35(8.03)$ $7(1.61)$ $42(9.63)$ $13(2.98)$			
1BGA						25 (5.59) 5 (1.12) 28 (6.26) 27 (6.04) 23 (5.15) 39 (8.72) 17 (3.80) 30 (6.71) 8 (1.79) 31 (6.94)							12 (2.68) 25 (5.59) 18 (4.03) 24 (5.37) 25 (5.59) 21 (4.70)			
1PBG	33 (7.05)					$2(0.43)$ 40 $(8.55)$ 37 $(7.91)$ 25 $(5.34)$ 38 $(8.12)$		19 (4.06) 28 (5.98) 34 (7.26) 28 (5.98)					8 (1.71) 22 (4.70) 20 (4.27) 10 (2.14) 14 (2.29) 15 (3.21)			
	$100X$ 30 $(6.68)$					$5(1.11)$ 34 (7.57) 27 (6.01) 16 (3.56) 46 (10.24) 13 (2.90) 29 (6.46) 17 (3.79) 41 (9.13)							14 (3.12) 20 (4.45) 19 (4.23) 11 (2.45) 21 (4.68) 22 (4.90)			
1CBG	36 (7.35)					$5(1.02)$ 31 $(6.33)$ 24 $(4.90)$ 31 $(6.33)$ 37 $(7.55)$		13 (2.65) 23 (4.69) 32 (6.53) 44 (8.98)					$9(1.84)$ 28 (5.71) 26 (5.31) 9 (1.84) 26 (5.31) 30 (6.12)			
1E1E	29 (5.66)					5 (0.98) 35 (6.84) 31 (6.05) 26 (6.33) 42 (8.20) 10 (1.95) 28 (5.47) 33 (6.45) 39 (7.62)							13 (2.54) 34 (6.64) 32 (6.25) 11 (2.15) 22 (4.30) 33 (6.45)			
	1MYR 23 (4.59)					8 (1.60) 38 (7.58) 20 (3.99) 27 (5.39) 46 (9.18) 12 (2.40) 32 (6.39) 25 (4.99) 35 (6.99)							8 (1.60) 32 (6.13) 26 (5.19) 16 (3.19) 18 (3.59) 34 (6.79)			
	1GOW 28 (5.73)					$1(0.20)$ 32 (6.54) 30 (6.13) 23 (4.70) 40 (8.18)		13 (2.66) 20 (4.09) 23 (4.70) 36 (7.36)					$11$ (2.25) 30 (6.13) 26 (5.32) 11 (2.25) 32 (6.54) 32 (6.54)			
	$1QVB$ 29 (6.03)					$2(0.42)$ $29(6.03)$ $33(6.86)$ $22(4.57)$ $35(7.28)$		$12(2.49)$ $23(4.78)$ $29(6.03)$ $42(8.73)$					$11(2.29)$ $27(5.61)$ $30(6.24)$ $19(1.87)$ $24(4.99)$ $27(5.61)$			

*<sup>a</sup>* Charged residues, Arg, Lys, Asp, and Glu; hydrophobic residues, Ala, Val, Ile, Leu, Trp, Phe, Met, and Pro; thermolabile residues, Gln, Asn, Cys, and Met.

the two residues is 5.25 Å, which is consistent with the properties of retaining  $\beta$ -glycosidases.

**Structural basis for thermal stability.** Structural comparisons between thermophilic proteins from organisms living under extreme conditions and their mesophilic counterparts have been utilized to discover the possible thermostabilizing factors. The contributions of parameters (including ion pairs, hydrogen bonding, secondary structure, cavities, surface areas, amino acid composition, and flexibility) to the protein stability have been analyzed extensively (45, 48, 49). It seems that the thermal stability cannot be explained by a unique mechanism. Each thermostable protein uses one or a combination of the mechanisms elucidated in comparative studies to maintain its structure at high temperature.

The optimal growing temperature of *T. nonproteolyticus* HG102 (producing Gly<sub>Tn</sub>) is 65°C (8), while the two hyperthermophilic archaea *S. solfataricus* and *T. sphaera aggregans* (producing 1GOW and 1QVB, respectively) can grow at temperatures as high as 87 and 90°C (9, 54). Gly<sub>Tn</sub> displays optimal activity with a half-life of 2.5 h at 90°C. Enzyme 1GOW shows stronger thermostability, with a half-life of 48 h at 85°C (34); similarly, 1QVB can retain 95% of its activity after incubation at 80°C for 130 h (9). Other mesophilic glycosidases with known structures are from eubacteria or plants growing at normal temperature and should show weaker thermostability than  $\text{Gly}_{\text{Ta}}$ . For 1QOX, 80% of its activity remained after being heated at 50°C for 15 min in phosphate buffer, while 1% was left after 15 min at 60°C (37). The eubacterium *B. circulans* producing 1QOX cannot grow well at temperatures higher than 40°C (54). The halflife of 1BGA produced by *B. polymyxa* (whose highest growing temperature is 40°C [54]) is only 3.6 min at 48°C (31). Structural comparisons between  $Gly_{Tn}$  and these mesophilic and hyperthermophilic  $\beta$ -glycosidases would help to elucidate the possible structural determinants of its thermostability.

**Amino acid composition and the stabilization of secondary structure.** The amino acid compositions of the nine family 1 --glycosidases with known three-dimensional structures are listed in Table 3. The glycosidases 1GOW and 1QVB from hyperthermophilic archaea do not show significant changes in amino acid composition compared to the other six glycosidases from mesophiles. For  $Gly_{Tn}$ , the significant changes were (i) high content of Ala residues, (ii) high content of Pro residues, (iii) low content of thermolabile residues, and (iv) high Arg/ Lys ratio. Besides the high content of Ala in helices, the number of  $\beta$ -branched residues decreased significantly in  $\mathrm{Gly}_{\mathrm{Tn}}$ (Table 4). The secondary structure contents of the nine glyco-



FIG. 3. Stereo figures of an ion pair network acting as electrostatic links between two helices,  $\alpha$ 5 and  $\alpha$ 6.

TABLE 3—*Continued*

No. $(\%)$ of amino acids				Total no. of	$%$ Charged	$%$ Hydrophobic	$%$ Pro	$%$ Ala	% Thermolabile	Arg/Lys	$\%$	$\%$	
Thr	Val	Trp	Tyr	amino acids	amino acids	amino acids	amino acid	amino acid	amino acids	ratio	$\alpha$ -Helices	$\beta$ -Sheets	
16(3.67)	26(5.96)	13 (2.98)	24(5.50)	436	23.4	48.4	8.0	12.8	5.0	10.5	38.5	15.4	
22(4.92)	29 (6.49)	15(3.36)	23(5.15)	447	19.7	40.9	4.0	5.6	14.8	3.1	38.9	16.1	
20(4.27)	28 (5.98)	10(2.15)	37 (7.91)	468	26.7	38.5	4.3	7.1	9.0	0.4	40.6	15.8	
17(3.79)	21 (4.68)	16(3.56)	30(6.68)	449	22.0	41.4	4.0	6.7	11.1	1.2	39.0	16.0	
20(4.08)	21(4.29)	12(2.45)	33(6.73)	490	23.1	41.2	5.3	7.4	10.4	0.8	39.2	15.5	
24(4.69)	21(4.10)	12(2.34)	32(6.25)	512	23.6	39.1	6.3	5.7	12.3	0.7	36.1	15.2	
34(6.79)	20(3.99)	10(2.00)	37 (7.39)	501	20.2	36.1	5.2	4.6	12.8	0.7	38.1	17.0	
20(4.09)	30(6.13)	17 (3.48)	34 (6.95)	489	23.9	39.0	5.3	5.7	10.8	1.4	34.7	17.4	
11(2.29)	41 (8.52)	17(3.53)	28 (5.82)	481	23.9	44.7	6.2	6.0	10.2	0.83	36.2	14.6	

sidases are very similar (Table 3), but two significant amino acid composition changes in  $\text{Gly}_{\text{Ta}}$  (increased content of Ala and decreased content of  $\beta$ -branched residues) are both helpful for the stabilization of  $\alpha$ -helices. The  $\alpha$ -helices can be stabilized by the introduction of residues with a high level of helix-forming propensity, such as Ala. The level of Ala content in the  $\alpha$ -helices of Gly<sub>Tn</sub> (17.9%) is much higher than those in mesophilic glycosidases and hyperthermophilic enzymes from archaea. The  $\beta$ -branched residues Ile, Val, and Thr were found to destabilize helices (38, 40). Their effects were ascribed to conformational entropy loss upon transfer from the extended conformation to the helix (10). The percentage of Val, Ile, and Thr in the helices of Gly<sub>Tn</sub> is 9.5%, which is much lower than those of other glycosidases. Facchiano et al. had also found that the helices of thermophilic proteins contain a lower percentage of  $\beta$ -branched residues than their mesophilic equivalents (16). These two features that may contribute to the thermostability of  $\mathrm{Gly}_{\mathrm{Tn}}$  were not observed in the hyperthermophilic glycosidases from archaea.

**Pro residues.** Matthews et al. have proposed that the protein structures can be stabilized by decrease in their entropy of unfolding (32). Pro differs from all other amino acids because the side chain curls back to the preceding peptide-bond nitrogen and forms the five-member pyrrolidine ring. It can adopt only a few configurations and can restrict the configurations allowed for the preceding residue (44); thus, it has the lowest conformational entropy. Replacement of other residues by Pro at suitable positions can enhance protein thermostability (19). It has been noted that Pro has an increased occurrence in thermophilic proteins, especially in loops (5, 30, 52). The Pro content in Gly<sub>Tn</sub> (8.0%) is the highest of all nine glycosidases, and the positional distribution in  $\text{Gly}_{\text{tn}}$  of the total 35 Pro residues was examined. In order of frequency, these Pro residues lie in random coil, helices, sheet, and turn. There are 23 Pro residues occurring in random coils, and the Pro-richest regions are the four loops: 248 to 265, 291 to 297, 301 to 315, and 330 to 333 (Fig. 2). There are six (252, 255, 259, 260, 261, and 263), three (291, 295, and 297), three (302, 303, and 306), and two (331 and 333) Pro residues in these loops. These constrained loops often strengthen the stabilizing interactions between the two adjacent core elements, thus increasing the protein rigidity (48). Of the seven Pro residues occurring in helices, five of them are at the N1 position. They are Pro93, Pro165, Pro228, Pro316, and Pro356. At the N1 position, a kink in the peptide backbone introduced by Pro can thermodynamically stabilize an  $\alpha$ -helix (57).

**Surfaces and cavities.** The importance of the hydrophobic effect on the stability of proteins has been accepted (13). The calculation of the ASA was used to simplify the complex analytical solution for the hydrophobic interaction (15, 56). It has been suggested that the reduction of hydrophobic ASA contributes to the stability of proteins (2, 20, 27). The charged and hydrophobic surface areas of the nine glycosidases were calculated and are listed in Table 5. From the results, we can see that reduction of hydrophobic surface areas does not contribute to the stability of  $\text{Gly}_{\text{Ta}}$ .

A decrease in the number and volume of the internal cavities and pockets has also been indicated as providing thermal stability (45). The nine glycosidases were analyzed for the presence of the cavities to try to ascertain whether they contribute to the thermal stability of these enzymes (Table 5). No significant trend for the decrease of the number and volume of cavities was observed in the thermostable  $\text{Gly}_{\text{Tr}}$ , 1GOW, and 1QVB.

TABLE 4. Amino acid composition in the  $\alpha$ -helices

Amino acid	No. of amino acids											
								Gly <sub>Tn</sub> 1BGA 1PBG 1QOX 1CBG 1E1E 1MYR 1GOW 1QVB				
Ala	30	12	23	17	18	12	7	16	20			
Cys	1	$\overline{2}$	$\overline{2}$	$\overline{c}$	$\mathbf{1}$	$\overline{2}$	3	$\theta$	$\overline{0}$			
Asp	8	13	17	16	12	12	16	8	9			
Glu	17	11	20	13	14	17	10	11	13			
Phe	6	9	9	$\tau$	9	10	11	9	$\tau$			
Gly	6	$\overline{4}$	$\overline{7}$	5	7	$\overline{4}$	9	6	$\overline{\mathbf{c}}$			
His	9	10	12	6	8	7	$\overline{7}$	5	6			
Ile	7	13	12	11	10	11	17	9	$\overline{7}$			
Lys	$\overline{2}$	$\overline{2}$	15	6	13	16	17	10	14			
Leu	25	16	15	26	20	20	14	10	12			
Met	$\mathbf{1}$	$\overline{4}$	$\overline{4}$	6	5	5	5	$\overline{4}$	5			
Asn	$\overline{0}$	4	5	3	6	10	5	10	10			
Pro	$\overline{7}$	3	3	3	5	6	5	5	6			
Gln	3	13	6	$\overline{7}$	8	6	11	5	$\overline{4}$			
Arg	20	15	6	11	10	7	8	16	11			
Ser	3	$\overline{7}$	6	5	7	9	10	8	8			
Thr	$\overline{c}$	6	3	$\overline{4}$	8	7	8	5	$\overline{c}$			
Val	$\overline{7}$	12	10	9	8	6	9	11	17			
Trp	$\overline{4}$	6	3	6	5	5	5	7	8			
Tyr	10	12	12	12	18	13	14	15	13			
Total	168	174	190	175	192	185	191	170	174			
$%$ Ala	17.9	6.9	12.1	9.7	9.4	6.5	3.7	9.4	11.5			
% ß-branched residues <sup>a</sup>	9.5	17.8	13.2	13.7	13.5	13.0	17.8	14.7	14.9			

<sup>a</sup> β-branched residues: Ile, Val, and Thr.

TABLE 5. Surface areas and cavities

Characteristic	Enzyme											
	$\mathrm{Gly}_{\mathrm{Tn}}$	1BGA	1PBG	100X	1CBG	1E1E	1MYR	$1$ GOW	10VB			
Total ASA of the monomer	16.120	16.251	17.522	15.968	19.938	18.715	18,087	18.361	18.741			
Hydrophobic ASA	5,139	3,856	4.241	4,337	5,110	5,180	4.003	4,353	5,625			
Charged ASA	7,723	6,090	8.434	6.733	8,734	8,098	6.456	7.942	8,830			
Hydrophobic ASA/total ASA $(\%)$	32	24	24	27	26	28	22	24	30			
No. of cavities												
Vol. of cavities	38.4	46.0	86.7	19.9	56.8	92.2	23.5	54.0	71.7			

**Hydrogen bonds.** Hydrogen bonding was compared among the nine glycosidases (Table 6). The hydrogen bonds were divided into three classes: main chain-main chain (MM Hbonds), main chain-side chain (MS H-bonds), and side chainside chain (SS H-bonds). There is no significant difference in the number of hydrogen bonds among mesophilic, thermophilic, and hyperthermophilic enzymes.

**Ion pairs.** Earlier work by Perutz suggested that electrostatic interactions represent a significant stabilizing factor in folded protein (39). After comprehensive studies of proteins with mesophilic and thermophilic structures in the PDB, Szilagyi and Zavodszky proposed that the only parameter with a consistent and significant contribution to thermal stability is the number of charged pairs and ion pair networks (45). An extensive analysis of ion pairs in the nine glycosidases was carried out, and the results are listed in Table 6. The number of ion pairs per residue was 0.10 for thermophilic  $\text{Gly}_{\text{Tn}}$ , 1GOW, and mesophilic 1PBG, a number which was the highest among the nine glycosidases. For thermophilic 1QVB, the number was 0.07, not departing significantly from the value of other mesophilic enzymes. After analyzing the constitutions of ion pairs in the nine enzymes, it was found that the ion pairs formed by Arg had the largest percentage for the three thermophilic enzymes. The percentages of ion pairs with Arg were 80, 70, and 72% in Gly<sub>Tn</sub>, 1GOW, and 1QVB, respectively. In Gly<sub>Tn</sub>, the high percentage of ion pairs with Arg is related to its extremely high Arg/Lys ratio. The Arg-forming ion pairs are over 70% of all of the ion pairs in 1GOW and 1QVB, although the Arg/Lys ratios are 1.4 and 0.8. The strong preference for utilization of Arg in ion pairs is determined by its properties. Arg is better at maintaining ion pairs at elevated temperature, because it has a higher  $pK_a$  than Lys. The  $pK_a$  value typically drops with increasing temperature (35, 50). It might also be due to its -guanido and longer side chain, which offer a wider range of possible ion pairs than those of Lys  $(35)$ . In Gly<sub>Tn</sub>,  $68\%$  of the ion pairs occur as part of multiple ion pair networks involving three or more charged residues. Many of the ion pair networks help to cross-link noncontiguous parts of the structure at the protein surface. One network (involving six charged residues, Asp235, Arg240, Asp244, Arg249, Arg325, and Glu329) acts as electrostatic links between two helices,  $\alpha$ 5 and  $\alpha$ 6 (Fig. 3). The other six-residue-forming ion pair network, which links the two connections  $\beta$ 7- $\alpha$ 7 and  $\beta$ 8- $\alpha$ 8, includes Asp345, Asp355, Arg358, Asp389, Arg400, and Lys416.

**Reduction in thermolabile residues.** Of the 20 amino acids, Asn, Gln, Cys, and Met can be classified as thermolabile due to their tendency to undergo deamidation or oxidation at high temperature and therefore may be naturally discriminated against in thermostable proteins (41). In Gly<sub>Tn</sub>, the level of Asn, Gln, Met, and Cys is 5.0%, which is much lower than those of other glycosidases. The difference mainly comes from the decrease of Asn, Gln, and Met content, and the difference of Cys is not significant because the content of Cys in all the nine glycosidases is very low. From the structural view, there are seven occurrences of the Gln residues in 1BGA changing to Glu or Asp that form an ion pair in  $\text{Gly}_{\text{Ta}}$ . Three Asn residues in 1BGA are changed to Glu or Arg that form an ion pair. Six Gln and Asn residues in 1BGA are changed to Pro that locate in loops of  $\mathrm{Gly}_\mathrm{Ta}$ .

**Conclusion.** The crystal structure of  $\beta$ -glycosidase  $\text{Gly}_{\text{Tr}}$ from thermophilic *T. nonproteolyticus* HG102 has been determined at a resolution of 2.4 Å. Analysis of its primary structure and the positions of two Glu residues in the active center indicate that it belongs to the glycosyl hydrolase of retaining

Characteristic	Enzyme										
	$\mathrm{Gly}_{\mathrm{Tn}}$	1BGA	1PBG	100X	1CBG	1E1E	1MYR	1GOW	10VB		
All protein-protein H-bonds	376	450	445	444	459	432	494	495	433		
MM H-bonds	254	285	284	283	283	263	293	285	275		
MS H-bonds	61	100	85	92	101	99	121	119	91		
SS H-bonds	61	65	76	69	75	70	80	91	67		
No. of hydrogen bonds per residue	0.88	1.01	0.98	0.99	0.94	0.88	0.99	1.01	0.90		
No. of ion pairs	44	32	45	34	27	32	23	47	32		
No. of ion pairs per residue	0.10	0.07	0.10	0.08	0.06	0.07	0.05	0.10	0.07		
No. of ion pairs formed by Arg/Lys/His	35/3/6	19/5/8	17/20/8	21/7/6	12/11/4	14/13/5	13/7/3	33/3/11	23/3/6		
No. of ion pairs formed by Glu/Asp	23/21	15/17	23/22	11/23	10/17	14/18	9/14	19/28	14/18		
$%$ of ion pairs including Arg	80	59	38	62	44	44	57	70	72		
$%$ of ion pairs occurring in ion network with three or more charged residues	68	47	60	56	41	28	17	62	47		

TABLE 6. Ion pairs and hydrogen bonds

family 1. Of all the structural features analyzed, those that may contribute to the thermostability of  $\mathrm{Gly}_{\mathrm{Tn}}$  are increased stabilization of the  $\alpha$ -helices, high content of Pro residues and the resulting restricted flexibility of loops, increased number of ion pairs and high percentage of Arg occurrence in them, and a reduction in thermolabile residues.

## **ACKNOWLEDGMENT**

The project was supported by the National Key Research Development Project of China (Project No. G1999075601).

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