

Potent selective nonpeptidic inhibitors of human lung tryptase

(asthma/protease inhibitors/bisbenzamidines)

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ABSTRACT Human lung tryptase, a homotetrameric serine protease unique to mast cell secretory granules, has been implicated in the pathogenesis of asthma. A hypothesis that tethered symmetrical inhibitors might bridge two adjacent active sites was explored via a rationally designed series of bisbenzamidines. These compounds demonstrated a remarkable distanced-defined structure–activity relationship against human tryptase with one series possessing subnanomolar potencies. Additional evidence supporting the concept of active-site bridging is also presented.

Human asthma is a complex immunoinflammatory disease (1–3). After allergen exposure, inflammatory cascades produce early- and late-phase responses as well as a state of hyperresponsiveness characterized by the release and synthesis of a myriad of mediators including the leukotrienes, prostaglandins, cytokines, histamine, growth factors, chemokines, heparin, and proteases. Of the various proteases, mast cell tryptase has recently become appreciated as a potential amplification component and driving force behind the changes to the airway seen in asthma. Inhibitors of tryptase have been reported (4–6), including one that has advanced to clinical trials (7).

Tryptase is a homotetrameric serine protease found in abundant preformed levels in human mast cells (11–35 pg/cell in human lung mast cells) that has been characterized as a protease, peptidase, and cytokine (8, 9). It has been implicated in a variety of pathological conditions including asthma (10) and other inflammatory (11–15) and autoimmune (16, 17) diseases. A variety of *in vitro* studies have demonstrated that this tryptic enzyme has many interesting effects such as promoting mast cell degranulation (43), inducing eosinophil and neutrophil migration (18), inactivating fibrinogen (19, 20), processing high and low molecular weight kininogen (21), degrading neurogenic bronchodilatory feedback mechanisms (22), amplifying the effects of histamine on lung tissue (23), and stimulating the growth of fibroblasts, bronchial smooth muscle cells, and airway epithelial cells while inducing IL-8 and intercellular adhesion molecule-1 expression (24). Tryptase can also process prostromelysin to mature stromelysin (matrix metalloproteinase type 3), which can further activate collagenase I (25, 26). All of these various activities of tryptase could significantly contribute to the early- and late-phase bronchoconstriction as well as to the development of airway hyperresponsiveness displayed in human asthma. In chronic asthma and other long-term respiratory diseases, these activities could also drive the profound changes to the airway such as desquamation of the epithelial lining, fibrosis, and thickening of the

underlying tissues (these changes are not treated by present therapeutics). It is the timing of tryptase release and its apparent autocrine effect in the allergen response that also make it such a compelling target.

Because tryptase is an enzyme consisting of four associated subunits, each capable of enzymatic proteolysis, the possibility exists for inhibition of more than one subunit with a single inhibitor molecule. This strategy of tethering two binding moieties together to produce an exponentially more potent inhibitor has been applied to relatively few medicinal chemistry problems; however, two recent examples are matrix metalloproteinase inhibitors, discovered via the “structure–activity relationship by NMR” technique (27–29), and acetylcholinesterase inhibitors (30). In each case, separate chemical moieties that displayed weak binding affinities were linked together to provide extremely potent enzyme inhibitors. The theoretical basis for the enhanced binding of these bifunctional molecules (A-B), originally proposed by Jencks (31), involves a summation of the observed intrinsic binding energy of moiety A, the observed intrinsic binding energy of moiety B, and a Gibbs energy of connection. This last term incorporates the change in the probability of binding that the connected molecule A-B displays over the individual fragments. In this study, weakly binding benzamidine moieties were bridged at various lengths and with various templates to provide subnanomolar inhibitors of human lung tryptase.

MATERIALS AND METHODS

Chemical Syntheses. The compounds described herein were prepared by standard synthetic organic chemistry procedures. Reagents, starting materials, and solvents were purchased from Aldrich or Maybridge (Cornwall, U.K.) Chemical Company and used without further purification. Intermediates and final products were purified by flash silica gel chromatography (32) or RP-HPLC by using a Waters Prep LC 2000 with Rainin Microsorb C18 columns (Rainin Instruments). Intermediates and final products were characterized by ¹H NMR (400 MHz, Bruker, Billerica, MA), ¹³C NMR (75 MHz, Bruker) and LRMS (Perkin–Elmer SCIEX electrospray). All compounds presented herein were determined to be ≥95% purity by ¹H NMR analysis.

Determination of K_i. Heparin-stabilized human lung tryptase, purchased from Cortex Biochem (San Leandro, CA), was further purified on a Superdex 200 gel-filtration column (Amersham Pharmacia) to remove the excess unbound heparin. The active-site concentration of the enzyme was determined by spectrophotometric titration with 4-nitrophenyl 4'-guanidinobenzoate (33). Tryptase activity was measured according to the procedures of Schwartz (34) with minor modifications, by using Tosyl-Gly-Pro-Arg-p-nitroanilide

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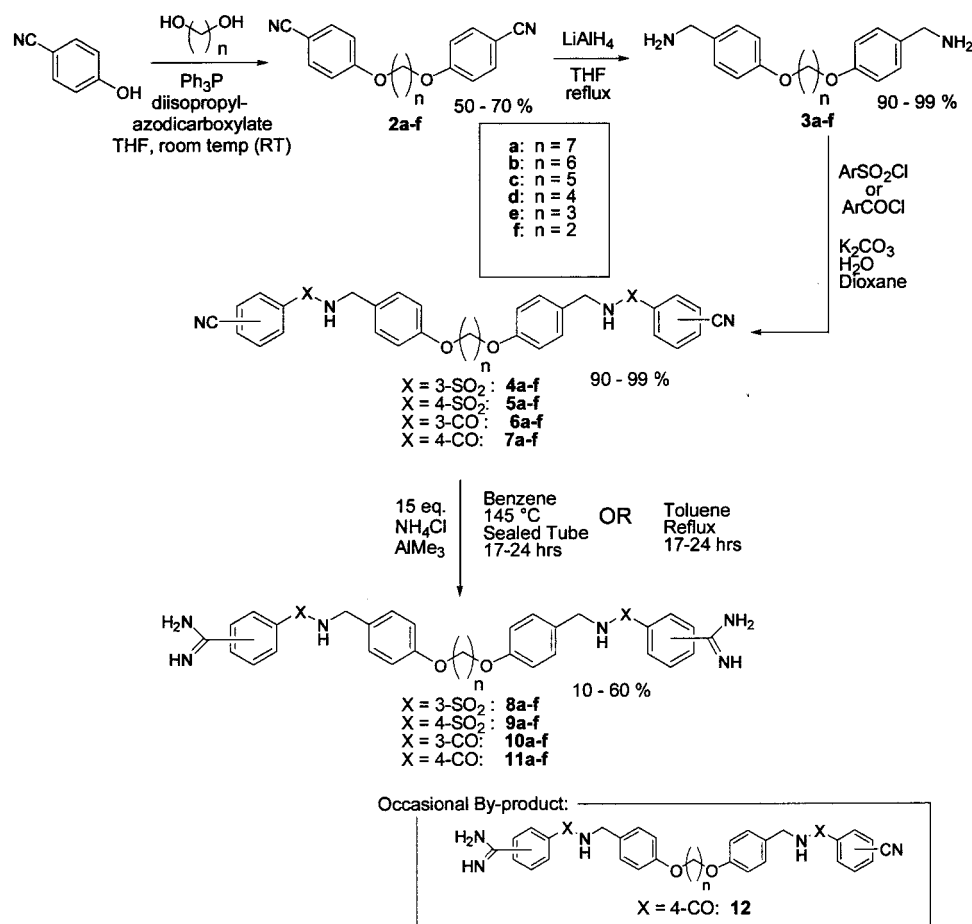
(GPR-*p*NA) as a chromogenic substrate. The reaction was carried out in 50 mM Tris-HCl, pH 8.0, containing 150 mM NaCl and 0.02% Triton X-100 at 37°C in Costar ultralow cluster 96-well microtiter plate. The amount of *p*NA produced by trypsin was determined by measuring the change in absorbance at 405 nm on a SpectraMAX 340 plate reader (Molecular Devices). The K_m for the substrate was determined by fitting the initial velocities of substrate hydrolysis to the Michaelis-Menten equation by using Sigma Plot (Jandel, San Rafael, CA) in an iterative fashion. Data were visualized by plotting according to the Lineweaver-Burk transformation. Inhibition assay was carried out in a total volume of 200 μ l wherein trypsin (30 μ l, final concentration 1 nM) was incubated with various concentrations of a sample (50 μ l) to be tested in the above assay buffer for 5 min. The reaction was started by the addition of substrate GPR-*p*NA (40 μ l, final concentration 320 μ M), and the residual activity was measured after 15 min of incubation. The inhibition constant, K_i , was determined by fitting the inhibition data (measured IC_{50} values) to a two-site competitive binding equation by using the data analysis program GRAPHPAD PRISM (GraphPad, San Diego). For very potent inhibitors, this program extrapolated the low K_i values. Determination of K_i values against trypsin and plasmin were obtained similarly.

RESULTS AND DISCUSSION

The ability of benzamidine to inhibit serine proteases of the trypsin family is well documented (35–37). This moiety, an isostere for the guanidine side chain of arginine, binds readily to the P_1 pocket via an amidine/carboxylate salt bridge with the aspartic acid at the base of this pocket. Indeed, benzami-

dine (**1**) was found to inhibit human lung trypsin ($K_i = 22 \mu$ M), bovine trypsin ($K_i = 33 \mu$ M), and human plasmin (22% inhibition at 100 μ M). However, because trypsin is the only member of this tryptic enzyme family that associates in a polymeric manner, we speculated that bisbenzamidine inhibitors, tethered together at varying distances, might provide much greater potency and selectivity by binding at adjacent active sites.

A concise synthetic route to a series of bisbenzamidines was designed and implemented (see Scheme 1). The particular structures targeted were chosen simply because the synthetic route to these compounds was highly precedented and likely to succeed. We felt that the linker region needed to possess only flexibility rather than any specific binding element and, as long as the benzamidine unit was presented at both ends, the compounds should effectively test the hypothesis of active site bridging. The aim of this study was not necessarily to identify a clinical candidate, but rather to produce pharmacological tools for probing the importance of trypsin in animal models of pulmonary disease. In the event, various bisethers **2a-f** were readily prepared via Mitsunobu chemistry by using 4-cyanophenol and the required diol (**38**). Reduction of the nitrile and sulfonylation (or acylation) under Schotten-Baumann conditions with the appropriate cyanoarylsulfonyl chloride (or cyanoaryl acid chloride) provided the bisbenzonitriles **4a-f**, **5a-f** or **6a-f**, **7a-f**, respectively (**39**). The use of the Garigapati-Weinreb amidination (**40**, **41**) in a sealed tube at high temperature or refluxing in toluene helps to overcome the insolubility of the monoamidine to return the required bisamidines **8–11** in moderate to low yields. Occasionally, the monoamidine intermediates **12** were isolated and tested. Although reverse-phase chromatography was needed for purifying sev-

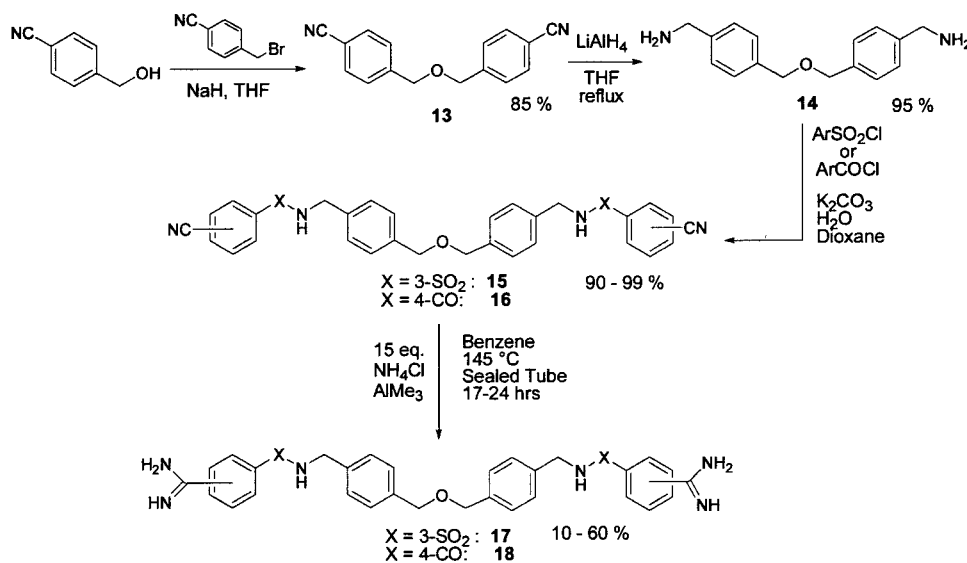


Scheme 1

eral of the smaller bisamidines, this amidination procedure can yield clean products after a simple quench with silica gel and methanol followed by filtration and concentration. The excess ammonium chloride can be washed out of the resulting solids with a small quantity of water.

A contracted tether was designed around the mono-ether **13** (see Scheme 2). Hydride reduction to diamine **14** followed by sulfonylation or acylation and amidination as before provided the smaller bisamidines **17** and **18**.

The most striking structure–activity relationship evident from these data is the effect of tether length on trypsin inhibition. Aside from the para-amidine sulfonamide series (4-SO₂), a distinct relationship between distance and inhibitory activity is manifested. A central chain of three to five carbon atoms produces very potent inhibition and, because trypsin is unique in its tetrameric structure, these inhibitors display excellent selectivity over other serine proteases (generally, trypsin and plasmin were the only other proteases



Scheme 2

This series of four potential active site bridging inhibitors was then evaluated against human lung trypsin (see Table 1—selectivities are provided as magnitude increases over trypsin activity). At first glance, one notices that the para-amidine amide series (4-CO) demonstrates excellent subnanomolar potencies against the enzyme, while the corresponding para-amidine sulfonamide (4-SO₂) series were generally much less active. One hypothesis for this observation is that the carbonyl group might favorably interact with the enzyme (presumably in and around the P₁ pocket) via its planarity and particular benzamidine vector, whereas the tetrahedral SO₂ cannot achieve the same geometry. Still another explanation could involve the electrophilic reactivity of the former and the nonreactivity of the latter. To test this hypothesis, the reversibility of these inhibitors was analyzed. When compounds **11c** and **8c** were separately incubated with the active enzyme and then dialyzed with 3,000× volumes of buffer, no activity could be recovered. On the other hand, denaturing the same enzyme-inhibitor complex with heat returned the intact inhibitors (data not shown). These results suggest that these compounds may have very slow off rates but that they are truly reversible, yet this does not rule out the possibility that the carbonyl moiety may be establishing a covalent interaction with the enzyme. An x-ray co-crystal would thus be required to probe the nature of carbonyl vs. sulfonyl activity against mast-cell trypsin.

Another observation concerning the nature of these potent inhibitors was the nonconventional curve shape of their inhibitory response. Although a typical enzyme inhibitor produces a standard curve with a single slope, these compounds display a biphasic curve with a double slope that fits well into a two-site binding equation (data not shown). Two *K*_i values can be derived from these plots, one of which may represent the binding to a single active site, and the other may reflect the high-affinity effect of binding two active sites within the same complex. The data reported here are the latter *K*_i values.

that were affected).

Intrigued by the potent activity of these C₂ symmetric bisamidine inhibitors, monoamidine intermediates (**12e**, **19**) were isolated, and several simple benzamidine derivatives were prepared (via chemistry depicted in Scheme 1) to further test the concept of active site bridging (see Table 2). None of these compounds were particularly active against trypsin relative to their bisamidine counterparts. For example, **12e**, which contains all the potential binding elements of **11e** except the second amidine, is 2,400 times less potent. This magnitude of potency enhancement is similar to that of the earlier examples cited (matrix metalloproteinase inhibitors and acetylcholinesterase inhibitors) and illustrates the important contribution of the Gibbs energy of connection.

On the basis of an analysis of the primary sequence of human lung trypsin and the x-ray crystal structures of other serine proteases, we had proposed that trypsin is not a C₄ symmetrical tetramer but rather a dimer of dimers possibly associated through loop 5 interactions. In that proposed dimeric assembly, adjacent catalytic domains would then be in relatively close proximity, and a single inhibitory molecule, like compound **11c**, could bridge these two active sites. In 1998, however, an x-ray crystal structure (3-Å resolution) of human trypsin was reported (42) that clarified the tetrameric geometry of the enzyme. In that study a square flat ringed arrangement was observed in which the four active sites are pointed toward the central core. Although this structure is slightly different than our proposal, the distances between adjacent and opposed active sites (≈20 Å and ≈40 Å) fit well with our results. For example, the distance between the two carbonyl groups in compound **11c**, in the fully extended conformation, is 22.2 Å, whereas the distance between amide nitrogens is 18.8 Å. Thus compound **11c** could occupy adjacent active sites in trypsin according to the dimensions reported.

Table 1. Biochemical profile of active site bridging inhibitors of human trypsinase

	Compound	Trypsin K_i , nM	Selectivity	
			Trypsin	Plasmin
X = 3-SO ₂	8a : $n = 7$	800	×15	×12
	8b : $n = 6$	105	×34	×25
	8c : $n = 5$	80	×31	×12
	8d : $n = 4$	30	×60	×2
	8e : $n = 3$	15	×180	×49
	8f : $n = 2$	27	×192	×74
	17	98	×31	×102
X = 4-SO ₂	9a : $n = 7$	1,500	—	—
	9b : $n = 6$	3,150	—	—
	9c : $n = 5$	250	×32	×40
	9d : $n = 4$	12,500	—	—
	9e : $n = 3$	800	×5	×8
	9f : $n = 2$	1,500	—	—
X = 3-CO	10a : $n = 7$	78	×15	×110
	10b : $n = 6$	10	×170	×1,000
	10c : $n = 5$	7	×130	×1,000
	10d : $n = 4$	9	×120	×1,400
	10e : $n = 3$	20	×120	×1,400
	10f : $n = 2$	350	×6	×131
X = 4-CO	11a : $n = 7$	1	×1,500	×4,000
	11b : $n = 6$	<0.01	×650,000	×800,000
	11c : $n = 5$	<0.01	×100,000	×450,000
	11d : $n = 4$	0.20	×7,500	×60,000
	11e : $n = 3$	0.50	×5,000	×23,000
	11f : $n = 2$	50	×28	×130
	18	375	×2	×27

In summary, potent selective nonpeptidic inhibitors of human lung trypsinase have been designed, prepared, and tested to take advantage of the tetrameric nature of the enzyme. A remarkable distance-defined structure–activity relationship resulted that, along with other data, supports the concept of

active site bridging. However, these trypsinase inhibitors are symmetrical and possess high molecular weights that make them fairly unattractive as drug-discovery leads. On the other hand, some of the inhibitors are quite potent and selective for mast cell trypsinase and are therefore potentially useful as

Table 2. Activity of monoamidines against human trypsinase

Structure	Compound	Trypsin K_i , nM
	12e	1,200
	19	1,050
	20	>1000,000
	21	100,000

pharmacological tools to delineate the role of tryptase in various animal models of inflammatory and pulmonary disorders. Studies of this aspect of our project will be reported in due course.

This paper is dedicated to Professor Edward M. Burgess on the occasion of his retirement from the Georgia Institute of Technology. We gratefully acknowledge Drs. Marv Caruthers, Jeff Yingling, Bhavana Shah, and Richard Nelson for helpful discussions.

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