

Wilson's disease: assessment of D-penicillamine treatment

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SUMMARY Serum copper and zinc concentrations and 24 hour urinary copper and zinc excretion were determined serially from the beginning of treatment with D-penicillamine in four children with Wilson's disease. The data show a progressive decrease in both serum copper and zinc concentrations in all. Urinary copper excretion gradually levelled off to approximately 50% of initial values, but zinc excretion increased. Urinary zinc:copper ratios therefore increased with the duration of treatment. Copper elimination was considered adequate as soon as challenge with a test dose of D-penicillamine did not result in an increase in copper excretion. Urinary zinc excretion was increased further by the test dose. Zinc depletion was suspected clinically in one patient on D-penicillamine maintenance treatment. Lowering the dose alleviated the symptoms, urinary zinc loss decreased from 64 to 34 $\mu\text{mol}/24$ hours, and copper excretion remained largely unchanged.

Data obtained indicate that D-penicillamine alters the metabolism of both copper and zinc. The extent of this is not only dose dependent but is also related to the efficacy of copper elimination. Both copper and zinc concentrations must be monitored to assess the benefits of treatment and the risks of inducing manifest or subclinical zinc deficiency.

The clinical manifestations of Wilson's disease are due to the accumulation of toxic amounts of copper in the liver, kidney, brain, cornea, and other tissues. Treatment with D-penicillamine (dimethylcysteine), combined with a low copper diet, first described by Walshe¹ and further confirmed by others², has substantially improved the outcome of the disease. The ultimate prognosis, however, still depends on the age, symptoms, and degree of tissue damage at the time of diagnosis.^{3,4}

D-penicillamine is an effective chelator of copper, is well absorbed from the gastrointestinal tract and promotes the urinary excretion of copper.¹ Thus it allows the removal of the excessive amounts stored and prevents further accumulation of copper. According to current recommendations^{3,5} treatment should be continued for life and the dosage adapted to the patient's tolerance. Practical guidelines are few, however: a small dose may not be effective enough while a larger amount can be toxic.⁵ Other essential metals, especially zinc, may, by an increased dosage, be removed simultaneously. To assess the efficacy of treatment and avoid possible

deficiency, we studied both zinc and copper values in serum and urine from the start of treatment with D-penicillamine in four children with the hepatic form of Wilson's disease.

Patients and methods

Patients. The four children were aged 5, 3, 7, and 10 years respectively when treatment was started. Criteria for diagnosis³ were (1) liver copper values greater than 325 $\mu\text{g/g}$ dry weight, (2) urinary copper excretion greater than 1.26 $\mu\text{mol}/24$ hours, (3) abnormal radiocopper kinetics in patients 1 and 2 (no substantial incorporation of ⁶⁴Cu in ceruloplasmin), (4) low serum ceruloplasmin values (less than 20 mg/dl), and (5) familial occurrence in patients 2, 3, and 4. None had Kayser-Fleischer rings or neurologic symptoms. Patient 1 presented with a chronic, active hepatitis-like syndrome, patient 3 with liver fibrosis, and patients 2 and 4 were asymptomatic siblings of known patients with Wilson's disease. None was jaundiced.

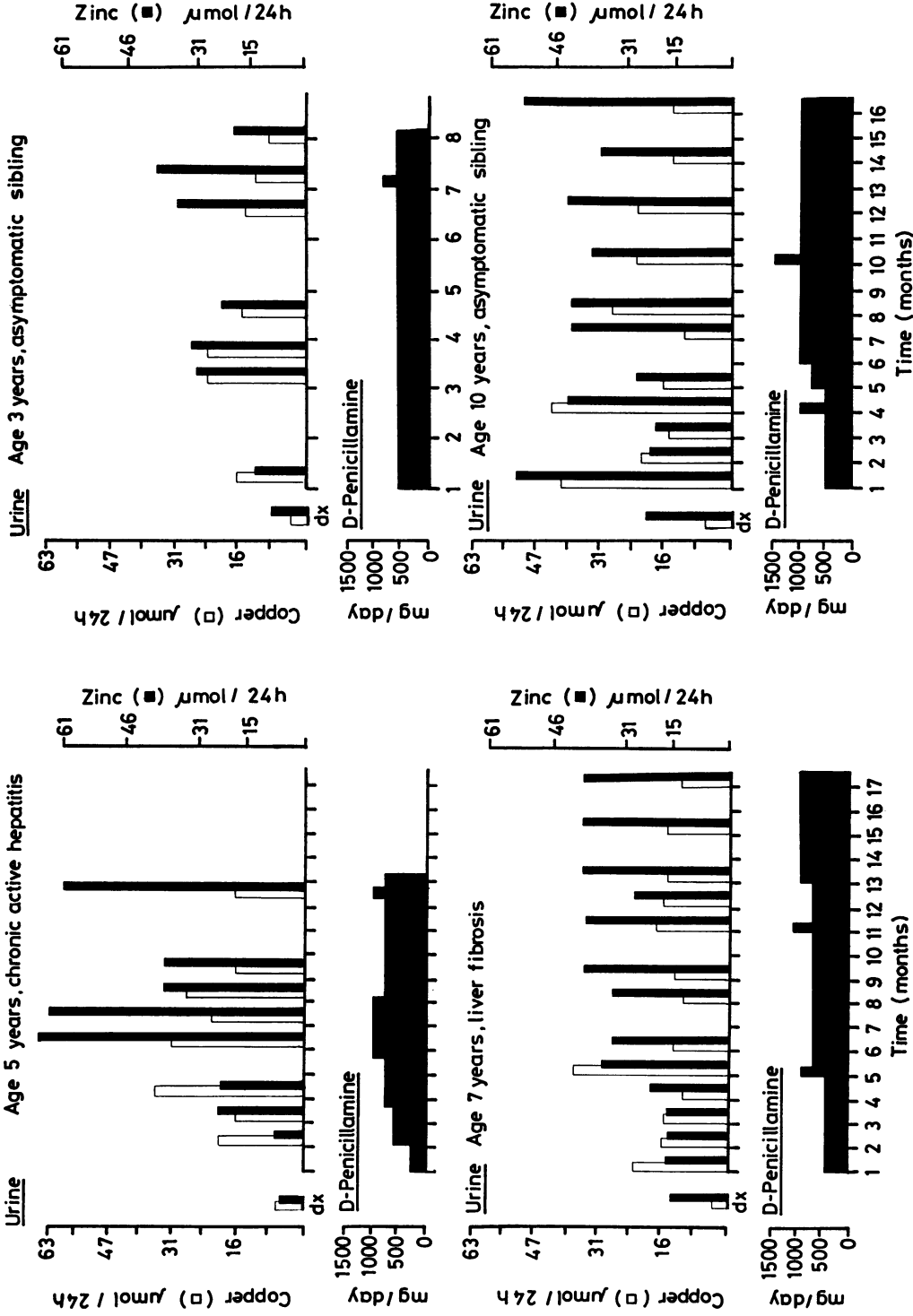


Figure Twenty four hour urinary zinc and copper excretion in relation to D-penicillamine dosage from the start of treatment and during steady state in four patients with Wilson's disease.

Conversion—SI to traditional units: copper 1 μmol/24 hours=63.6 μg/24 hours; zinc 1 μmol/24 hours=65.4 μg/24 hours. dx=start of treatment.

Treatment. Soon after confirmation of the diagnosis of Wilson's disease the patients were treated with D-penicillamine and a low copper diet was prescribed. The drug was given in three divided doses with meals. The initial dose was 500 mg daily, except for patient 1 who received 300 mg. This dose was maintained during the first six months in patients 2, 3, and 4, after which a challenge dose was given equal to the maintenance dose +250 mg. Dependent upon the results obtained, the maintenance dose was increased or continued unchanged.

Methods. Liver function tests were carried out by standard methods. Serum and 24 hour urine samples for zinc and copper assays were collected in metal free containers at monthly intervals during the first year and at bimonthly intervals thereafter. All metal analyses were performed in duplicate by means of atomic absorption spectrophotometry.

Results

Clinical course. All children were asymptomatic with normal liver function tests at the last follow up. They did not develop any neurological signs of the disease or Kayser-Fleischer rings. Patient 1, however, who received 1 g D-penicillamine daily, developed skin lesions seven months after the start of treatment. These lesions consisted of alterations in hair texture with hair loss, parakeratosis of the hands and feet, and deep grooves on the soles of the feet. Urinary zinc excretion at that time was 64 $\mu\text{mol}/24$ hours. Reducing the dosage of D-penicillamine to 750 mg alleviated the symptoms, the 24 hour urinary zinc excretion decreased to 34 $\mu\text{mol}/\text{l}$, and the copper excretion remained largely unchanged at 19 $\mu\text{mol}/24$ hours.

Copper and zinc data. Serum copper concentrations decreased steadily during treatment from 8.0 to 4.4, 8.7 to 3.1, 11.4 to 1.9, and 10.1 to less than 1.5 $\mu\text{mol}/\text{l}$ in patients 1, 2, 3, and 4 respectively. Serum zinc concentrations, which were initially high (mean 27 $\mu\text{mol}/\text{l}$, range 38 to 22 $\mu\text{mol}/\text{l}$) followed the same trend and reached the low normal range (mean 12.7 $\mu\text{mol}/\text{l}$, range 12 to 13.3 $\mu\text{mol}/\text{l}$) during maintenance treatment. The lowest value (9.8 $\mu\text{mol}/\text{l}$) was seen in patient 3, 15 months after beginning treatment.

Urinary copper and zinc excretions are summarised in the Figure. With the exception of patient 4, copper excretion exceeded zinc excretion, even on a relatively low dose of D-penicillamine (500 mg) at the start of treatment. As treatment continued the urinary copper excretion decreased progressively to approximately 50% of initial values, but the urinary zinc excretion increased further. Consequently the

Table Twenty four hour urinary zinc copper ratios at the start of D-penicillamine (D-P) treatment and during treatment.

Patient No	Start of treatment		Steady state	
	(No)	After D-P challenge	(No)	After D-P challenge
1	0.63 (3)	-	2.18 (1)	2.88
2	1.12 (4)	-	2.15 (2)	3.14
3	1.01 (4)	0.84	2.33 (6)	3.64
4	1.09 (3)	0.88	2.09 (16)	3.76
mean (+SD)	0.98 (0.4)	(14)	2.17 (0.6)	(25)

urinary zinc copper ratios increased with the duration of treatment as shown in the Table.

At the end of the first phase of treatment (usually after more than six months) a challenge dose of D-penicillamine resulted in a rise in zinc excretion. No further increase in copper excretion was then observed, suggesting that copper elimination was adequate. During the initial phase, however, this challenge dose did cause a rise in copper excretion as shown in patients 3 and 4, four and five months respectively after the start of treatment.

Discussion

The data obtained show that D-penicillamine alters the excretion of both copper and zinc in patients with Wilson's disease. The extent of this is not only dose dependent but is also related to the efficacy of copper elimination. A comparison of the dissociation constants for metal-penicillamine complexes show a decreasing stability in the order $\text{Hg} > \text{Cu} > \text{Ag} > \text{Pb} > \text{Ni} > \text{Cd} > \text{Zn} > \text{Co} > \text{Fe} > \text{Mn}$.⁸ Consequently more copper is excreted than zinc in the copper loaded patient at the beginning of treatment. As the mass of copper decreases, zinc having a lower affinity constant but being in relatively greater abundance will be excreted in greater amounts. Therefore, increased zincuria during D-penicillamine challenge is important both in assessing the benefits of treatment and the risks of inducing zinc as well as other trace metal deficiencies.

The effectiveness of treatment is usually assessed by serial determinations of transaminases, serum copper, and 24 hour urinary copper excretion.^{3 5 6} Our observations confirm previous reports (in children as well as in adults) that serum copper falls with time, as the non-ceruloplasmin bound fraction of serum copper returns to normal.^{1 7} Twenty four hour urinary excretion rises initially, but levels off to approximately 50% of the initial values (9.5 to 15 $\mu\text{mol}/24$ hours) during maintenance treatment.

Serial serum zinc and 24 hour urinary zinc excretion, however, are not routinely measured. The Figure shows that these determinations give important complementary information for assessing the adequacy of treatment. If copper elimination has been effective, the increase in zinc excretion is not accompanied by any appreciable increase in copper excretion when the patient is challenged by a larger dose of D-penicillamine. If the challenge dose steps up copper excretion, however, a higher maintenance dose is indicated. This approach may help to improve the effectiveness and thus prevent partial failure of the treatment.

Because of its non-specificity as a chelator, D-penicillamine may also affect other biologically important metals. Although in clinical use for more than 25 years, few data exist about its long term effects on the homeostasis of essential trace metals other than copper.⁹ Our findings support a previous report claiming the zinc deficiency may occur during treatment if urinary excretion exceeds the zinc absorption rate,¹⁰ although this was not substantiated with balance studies. Some of the known side effects of the drug strongly suggest a zinc deficiency: skin lesions on pressure points, desquamations, psoriasis like eruptions, delayed wound healing, alopecia and sometimes loss of taste acuity, glossitis, and stomatitis.¹¹ Some of the toxic effects may be attributable to hypersensitivity, others to the fact that D-penicillamine is an important anti-metabolite of pyridoxine or to its direct action upon collagen.¹² A few are almost certainly due to zinc deficiency and possibly also to deficiencies of other essential trace metals.

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