

# Activation of chloride current by P<sub>2</sub>-purinoceptors in rat ventricular myocytes

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1 Rat ventricular myocytes were dissociated and their responses to extracellularly applied ATP were recorded using patch pipettes under the whole cell configuration.

2 ATP initially induced an inward current followed by an outward current at –50 mV. With a Cs-rich pipette solution the late outward current was blocked, leaving a sustained inward current ( $I_{ATPs}$ ) suggesting that a K<sup>+</sup> conductance underlies the late response.

3 When the extracellular Cl<sup>–</sup> concentration was changed, the reversal potential of  $I_{ATPs}$  corresponded well to the shift of the Cl<sup>–</sup> equilibrium potential.  $I_{ATPs}$  was reversibly blocked by the chloride channel blocker, 4,4'-diisothiocyanatostilbene-2,2'-disulphonic acid (DIDS).

4 The concentration-response curve of  $I_{ATPs}$  had a Hill coefficient of 0.98 and an EC<sub>50</sub> value of  $5.2 \times 10^{-6}$  M.

5 ATP was more potent than ADP, while AMP and adenosine were ineffective, suggesting that P<sub>2</sub>-purinoceptor activation induced  $I_{ATPs}$ .

6 The activation of  $I_{ATPs}$  was depressed by depleting the extracellular Mg<sup>2+</sup> and increased by adding Mg<sup>2+</sup>.

7 Our results strongly suggest that P<sub>2</sub>-purinoceptor activation by ATP induces both a Cl<sup>–</sup>-conductance ( $I_{ATPs}$ ) and a K<sup>+</sup>-conductance in rat ventricular myocytes.

**Keywords:** Ventricular myocytes; ATP; chloride current; P<sub>2</sub>-purinoceptor

## Introduction

There are many reports that adenosine 5'-triphosphate (ATP) acts as a neurotransmitter through activation of P<sub>2</sub>-purinoceptors in a variety of tissues (Hoyle & Burnstock, 1991). In the heart, after the establishment of the presence of P<sub>2x</sub>-purinoceptors (frog: Hoyle & Burnstock, 1986; rat: Fleetwood & Gordon, 1987), later reports focused on the ionic mechanisms of purinoceptors in the heart. In guinea-pig atrium, it has been shown that ATP responses are mediated by an activation of Cl conductance (Matsuura & Ehara, 1992). However, the ionic mechanism in the rat ventricle is confusing. Christie *et al.* (1992) reported that the activation of P<sub>2</sub>-purinoceptor phosphorylates extracellular membrane-bound proteins, leading to the activation of a channel which carries Na<sup>+</sup> and Ca<sup>2+</sup>. On the other hand, Puceat *et al.* (1991b) reported that ATP binding to a putative P<sub>3</sub>-purinoceptor activates an electrically neutral HCO<sub>3</sub><sup>–</sup>-Cl<sup>–</sup> exchanger causing intracellular acidification. The intracellular acidification was thought to induce an increase in [Ca<sup>2+</sup>]<sub>i</sub> (Scamps & Vassort, 1990; Puceat *et al.*, 1991a) and finally the activation of a Ca-dependent cation conductance (Colquhoun *et al.*, 1981). Thus, the relationship between the ionic mechanisms of ATP-induced responses and the types of purinoceptors was unclear in the rat ventricle. In the present experiments, therefore, we have used the patch clamp recording method to analyse the ionic mechanisms of ATP-induced responses and the possible types of purinoceptors in rat ventricular myocytes. We find that activation of a P<sub>2</sub>-purinoceptor induced two different conductances, a Cl<sup>–</sup>-dependent current and a K<sup>+</sup>-dependent current, which have not been reported previously. In the discussion the possible relationship between Cl<sup>–</sup> movement through an activation of P<sub>2</sub>-purinoceptor and an ATP-induced intracellular acidification has been suggested.

## Methods

### Preparation

Isolated myocytes were obtained from male Sprague-Dawley rats (300–400 g) by collagenase digestion (0.1 mg ml<sup>–1</sup>, Yakult, Tokyo, Japan) (Powell & Twist, 1976). The isolated cells were collected by sedimentation. The pellet was diluted with control solution to give a final concentration of 10<sup>5</sup> cells ml<sup>–1</sup>; 200 μl of cell suspension was transferred to concanavalin A (Sigma Chemical Co., St Louis, MO, USA)-coated dishes containing 5 ml of control solution (extracellular Ca<sup>2+</sup> concentration ([Ca<sup>2+</sup>]<sub>o</sub>) = 0.1 mM). Within 1 h cells were anchored on the dish. Cells were kept at room temperature until use and [Ca<sup>2+</sup>]<sub>o</sub> was gradually increased up to 1 mM just before use. Cells were viable for 6–8 h in the present experiments.

### Solutions

External and internal solutions used in the present experiments are shown in Table 1. To isolate Cl<sup>–</sup> conductance we used a N-methyl-D-glucamine (NMG)-rich solution intra- and extracellularly. In the NMG-rich solution replacement of the membrane-permeable cations (Na<sup>+</sup>, K<sup>+</sup> and Ca<sup>2+</sup>) by the membrane-impermeable cation, NMG, blocks cation conductances, background conductances (Hagiwara *et al.*, 1992), the Na-K pump and the Na-Ca exchanger (Kimura *et al.*, 1986; Mechmann & Pott, 1986). Second, the possibility of the passage of Mg<sup>2+</sup> or contaminating Ca<sup>2+</sup> through voltage-gated Ca channels was excluded with verapamil. Third, the movement of residual intracellular K<sup>+</sup> through voltage-gated K<sup>+</sup> channels was also inhibited by perfusing the cells intracellularly with tetraethyl ammonium (TEA). Fourth, the Na-K pump was directly inhibited by ouabain.

A rapid superfusion system was used for solution exchange (Suzuki *et al.*, 1990). In this system solution exchange around the cells was complete within 100 ms. All experiments were

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**Table 1** Composition of solutions

External solutions (in mM)										
	NaCl	NaH <sub>2</sub> PO <sub>4</sub>	KCl	CaCl <sub>2</sub>	MgCl <sub>2</sub>	HEPES	Glucose	NMG		
Control	144	0.33	5.4	1	2	5	10			
K-free	150	0.33		1	2	5	10			
NMG-rich					2	5	10	150		

Internal solutions (in mM)											
	CsOH	CsCl	EGTA	HEPES	AA	ATP	NaCr	MgCl <sub>2</sub>	CaCl <sub>2</sub>	NMG	TEACl
Cs-rich	110	20	10	5	90	5	5	1	0.5		
NMG-rich			10	5		5		1		110	20

Solutions contained 0.1 mg ml<sup>-1</sup> bovine serum albumin and 0.001% phenol red. pH was adjusted at 7.4 with NaOH (control and K-free) or HCl (NMG-rich). NMG-rich solutions contained 5 μM verapamil and 10 μM ouabain. NMG: N-methyl-D-glucamine.

pH was adjusted at 7.2 with CsOH (Cs-rich) or HCl (NMG-rich). Final Cl<sup>-</sup> concentration of NMG-rich solution was 101 mM. AA: aspartic acid. ATP salts used are as follows: Cs-rich, Na<sub>2</sub>-ATP; NMG-rich, Tris ATP.

carried out at room temperature. Unless otherwise mentioned, experiments were repeated on at least 3 cells.

#### Current recording

Currents were recorded by use of a patch clamp recording system in the whole cell configuration (Hamill *et al.*, 1981). The resistance of fire polished patch pipettes, filled with internal solution was 5–10 MΩ. The outer wall of the pipette, except for the tip, was coated with apiezon wax dissolved in chloroform (Apiezon Products Ltd., London, U.K.) to reduce the capacitance. The reference electrode was an Ag-AgCl wire connected to the dish by a 140 mM NaCl agar bridge. Currents were recorded by a patch clamp amplifier (CEZ-2200, Nihon Kohden, Tokyo, Japan). The ramp pulse was generated by a function generator (FG-121B, NF Circuit Design Block Co. Ltd., Yokohama, Japan). Current and voltage signals were recorded on a video tape recorder (HV-S21, Mitsubishi, Tokyo, Japan) through a PCM processor (PCM-DP16, Shoshin EM Corp., Okazaki, Japan). Current and voltage signals were also monitored by a chartrecorder (WR3701, Graphtec, Yokohama, Japan). The recorded current and voltage signals were filtered at 1 kHz through a low pass filter (E3201-A, NF Circuit Design Block Co. Ltd., Yokohama, Japan) and analysed by an off-line computer (PC-9801RX, NEC, Tokyo, Japan). A hard copy of individual current-voltage (*I-V*) relationship was plotted by an X-Y plotter (MILOT III, Graphtec, Yokohama, Japan) through the off-line computer.

#### Correction of holding potential

In the present experiments the holding potential ( $V_H$ ) was corrected for the effect of liquid junction potentials (Hagiwara & Ohmori, 1982).

#### Correction of chloride concentration

The equilibrium potential of Cl<sup>-</sup> was calculated using the activity coefficient of Cl<sup>-</sup>. The activity coefficient of Cl<sup>-</sup> in NMG-Cl solution was approximated by that in KCl solution. The activity coefficient of [Cl<sup>-</sup>]<sub>o</sub> was estimated as 0.74 at 153 mM, 0.82 at 51 mM and 0.88 at 15 mM. The activity coefficient of the intracellular Cl<sup>-</sup> concentration ([Cl<sup>-</sup>]<sub>i</sub>) was estimated as 0.77 when cells were fully perfused by NMG-rich intracellular solution ([Cl<sup>-</sup>]<sub>i</sub> = 101 mM). The corrected [Cl<sup>-</sup>]<sub>o</sub> and [Cl<sup>-</sup>]<sub>i</sub> was expressed as the product of the activity coefficient and Cl<sup>-</sup> concentration.

## Results

### ATP-induced response

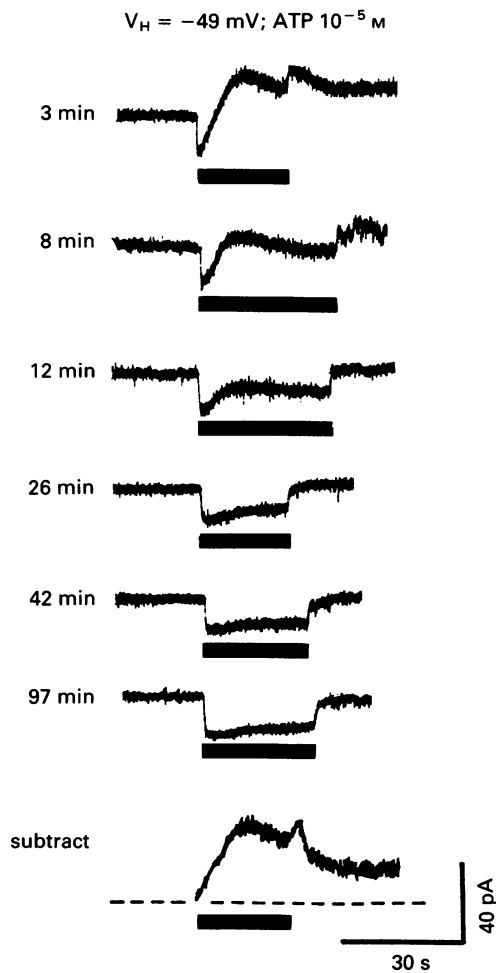
As shown in Figure 1, the application of 10<sup>-5</sup> M ATP induced a biphasic current response at a  $V_H$  of -49 mV. The initial inward current was observed immediately after the start of ATP superfusion and was followed by a gradual outward current. Removal of ATP induced a further transient outward current. In the experiment shown in Figure 1, a Cs-rich pipette solution was used to suppress outward K<sup>+</sup> conductances. The gradual loss of the ATP-induced outward current (compare 3 min and 42 min) probably reflects equilibration of this Cs-containing pipette solution with the intracellular medium. Moreover, as the outward current was not observed when cells were held at -89 mV (not shown), this current may be carried by K<sup>+</sup>. Separation of the inward and outward responses was performed by subtraction of the response recorded at 97 min from that at 3 min (Figure 1, subtract). The outward response reached a peak at around 10 s and decayed gradually. The average time after patch rupture until the disappearance of the outward current was 21.5 ± 7.5 min ( $n = 23$ ). Out of 57 cells tested at 4 min after the establishment of intracellular perfusion, 52 had both an inward and an outward current and 5 cells had only an inward current. In the present experiment we have analysed only the sustained inward current ( $I_{ATP_s}$ ) in detail.

$I_{ATP_s}$  was repeatedly evoked when 10<sup>-4</sup> M ATP was applied every 4 min. Therefore, in the present experiments the interval of ATP application was set at 4–5 min.

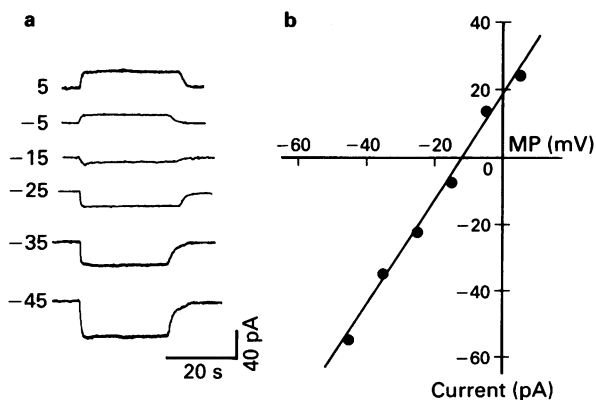
### Ionic selectivity of $I_{ATP_s}$

In previous reports, two different cation conductances activated by ATP have been reported in rat ventricular myocytes; a cation conductance (Christie *et al.*, 1992), and a Ca-dependent cation conductance (Puceat *et al.*, 1991a). In our experiments, cells were intra- and extracellularly perfused with an NMG-rich solution designed to block cation conductances (see Methods). However, even under these ionic conditions, the application of ATP induced  $I_{ATP_s}$ . In the NMG-rich solution the average time until the isolation of  $I_{ATP_s}$  was 7.7 ± 2.9 min ( $n = 42$ ). Therefore, we conclude that  $I_{ATP_s}$  response is not mediated by an increase in cation conductance. In addition,  $I_{ATP_s}$  could be observed when an NMG-rich solution containing 10 mM EGTA was used for the extracellular perfusate, suggesting that an influx of Ca<sup>2+</sup> is not involved in the generation of  $I_{ATP_s}$ .

The  $\text{Cl}^-$ -sensitive nature of  $I_{\text{ATPs}}$  was indicated by measurement of the reversal potential. In the experiment shown in Figure 2a, ATP was applied at various holding potentials. The response was sustained during the application of ATP at every membrane potential and reversed between  $-5$  and



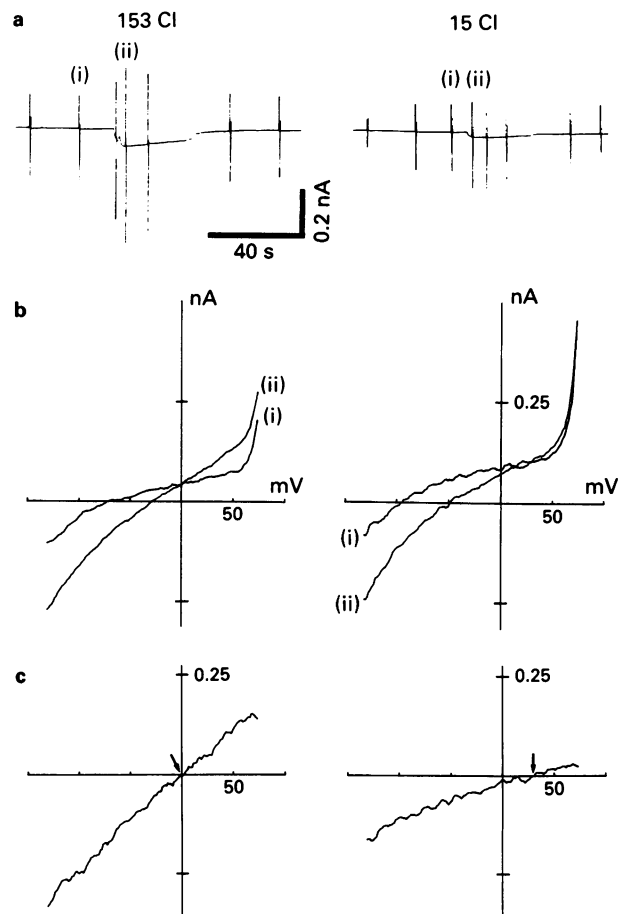
**Figure 1** ATP-induced current recorded at different times. Times shown at the left, indicate the time after the establishment of the intracellular perfusion. Bars shown below each trace correspond to the duration of ATP application.  $V_H = -49$  mV. External solution: K-free, pipette solution: Cs-rich.



**Figure 2** Current-voltage ( $I$ - $V$ ) relationship of  $I_{\text{ATPs}}$ . (a)  $I_{\text{ATPs}}$  recorded at different holding potentials. (b)  $I$ - $V$  relationship of  $I_{\text{ATPs}}$ . (a) and (b) were obtained from the same cell. Applied ATP concentration:  $10^{-5}$  M, external solution: NMG-rich, pipette solution: NMG-rich.

$-15$  mV. The peak amplitude voltage relation plotted in Figure 2b, was linear and indicated a reversal potential ( $E_{\text{ATPs}}$ ) of  $-12$  mV. In 6 experiments  $E_{\text{ATPs}}$  was  $-13.3 \pm 1.7$  mV. This value is quite close to the equilibrium potential for  $\text{Cl}^-$  ( $E_{\text{Cl}}$ ) of  $-9.5$  mV.

To facilitate the measurement of  $E_{\text{ATPs}}$ , the  $I$ - $V$  relation was determined by the use of a ramp pulse (Figure 3a). The background conductance obtained before the application of ATP showed an outward going rectification. When  $[\text{Cl}^-]_o$  and  $[\text{Cl}^-]_i$  were 153 and 101 mM, respectively, the  $I$ - $V$  curve in the presence of ATP showed large slope conductance and crossed with the control  $I$ - $V$  curve at 0 mV (Figure 3b, left). When  $[\text{Cl}^-]_o$  was reduced to 15 mM, the  $I$ - $V$  curve also showed rectification and two  $I$ - $V$  curves determined before and after ATP application crossed at  $+31$  mV (Figure 3b, right). The difference obtained by subtracting the control  $I$ - $V$  curve from those in the presence of ATP was almost linear in every case (Figure 3c). The reversal potentials thus determined were corrected for the liquid junction potential and plotted against the activity of  $[\text{Cl}^-]_o$  (Figure 4). These values corresponded well with the shift of  $\text{Cl}^-$  equilibrium potential calculated from the Nernst equation. When  $[\text{Cl}^-]_o$  was changed from 153 mM to 15 mM, for a theoretical shift of  $\text{Cl}^-$  equilibrium potential of 54.1 mV, the observed shift of  $E_{\text{ATPs}}$  was 38.2 mV.



**Figure 3**  $I$ - $V$  relationship of  $I_{\text{ATPs}}$ . (a) Response to ramp pulse application. The sharp vertical lines on the current trace indicate the current evoked by the ramp pulse. Speed of ramp pulse was  $420 \text{ mV s}^{-1}$  (range  $-140$  mV to  $70$  mV). In the presence of ATP, ramp pulse was applied at around the peak of  $I_{\text{ATPs}}$ . (b) Estimation of  $E_{\text{ATPs}}$  by the ramp pulse. Individual  $I$ - $V$  relationships correspond to (i) and (ii) shown in (a). The crossing points correspond to  $E_{\text{ATPs}}$ . (c) Subtraction of ramp pulse in the presence and absence of ATP. Arrows indicate  $E_{\text{ATPs}}$ . Applied ATP concentration:  $10^{-5}$  M, external solution: NMG-rich, pipette solution: NMG-rich.

*Effects of DIDS on  $I_{ATPs}$* 

The  $Cl^-$ -sensitive nature of  $I_{ATPs}$  was further confirmed by the use of  $Cl^-$  channel blocker, DIDS (de Lisle & Hopfer, 1986). In the presence of 0.2 mM DIDS,  $I_{ATPs}$  induced by  $10^{-5}$  M ATP was suppressed by 70% (Figure 5). The inhibitory effects of DIDS on  $I_{ATPs}$  was reversible. The average of inhibition was  $61 \pm 8\%$  ( $n = 3$ ).

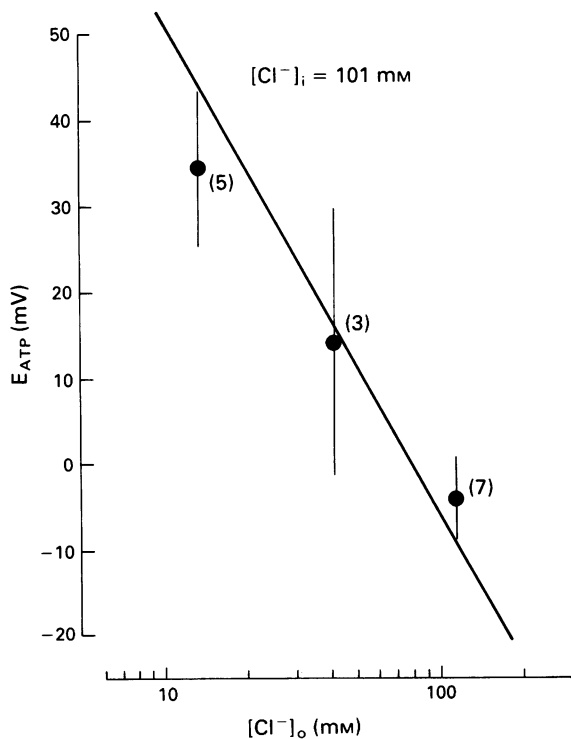
*Response to various purinoceptor agonists*

Purinoceptors can be pharmacologically classified into  $P_1$  and  $P_2$  types;  $P_1$  was characterized by a high affinity for AMP and adenosine while  $P_2$  is characterized by the high affinity for ATP and ADP (Burnstock, 1978). At a  $V_H$  of  $-45$  mV,  $10^{-5}$  M ATP induced  $I_{ATPs}$  (Figure 6). Similarly  $10^{-4}$  M ADP also evoked  $I_{ATPs}$ , though the amplitude of  $I_{ATPs}$  was only 15% of that evoked by  $10^{-5}$  M ATP. In contrast, AMP and adenosine did not evoke any response at  $10^{-4}$  M (Figure 6). These results indicate that  $I_{ATPs}$  is mediated by a  $P_2$ -purinoceptor in the rat ventricular myocytes.

However, the  $P_{2X}$ -purinoceptor agonist,  $\alpha$ ,  $\beta$ -methyladenosine 5'-phosphate (AMP-CPP) and  $\beta$ ,  $\gamma$ -methyladenosine 5'-phosphate (AMP-PCP) (Burnstock, 1990; Olsson & Pearson, 1990) did not evoke any response at  $10^{-4}$  M (not shown). The subtype classification of  $P_2$ -purinoceptor therefore remains to be determined although our data might also indicate the presence of  $P_{2Y}$ -purinoceptors in rat ventricular myocytes (Björnsson *et al.*, 1989).

*Concentration-response relationship of  $I_{ATPs}$* 

At a  $V_H$  of  $-45$  mV, application of  $3 \times 10^{-7}$  M ATP failed to evoke a detectable response (Figure 7a).  $I_{ATPs}$  was first observed at  $6 \times 10^{-7}$  M ATP and  $I_{ATPs}$  increased its amplitude with increasing ATP concentration. Figure 7b shows the

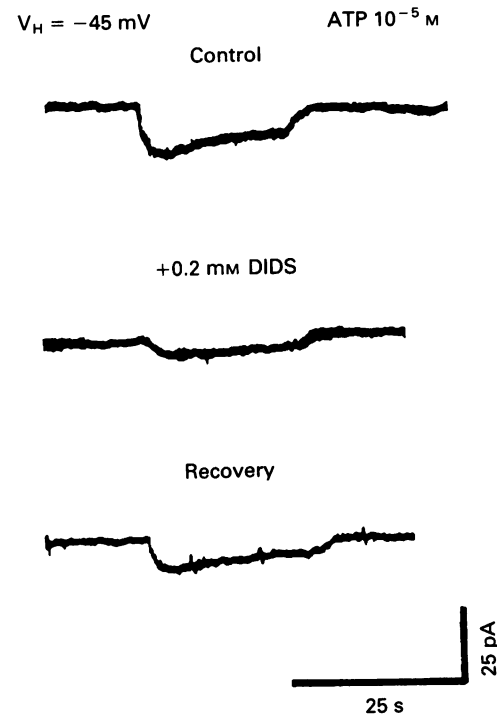


**Figure 4** Relationship between  $[Cl^-]_o$  and  $E_{ATPs}$ .  $[Cl^-]_o$  was corrected by the activity coefficient of  $Cl^-$ .  $E_{ATPs}$  were corrected by the tip potential and the liquid junction potential. Symbols show the average with s.d. Solid line represents the equilibrium potential of  $Cl^-$  calculated from the Nernst equation. Number in parentheses indicates  $n$  values. Applied ATP concentration:  $10^{-5}$  M, external solution: NMG-rich, pipette solution: NMG-rich.

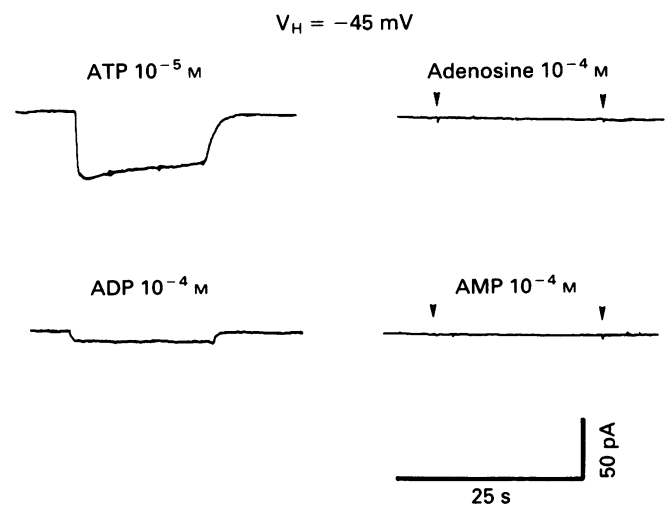
concentration-response relationship of  $I_{ATPs}$ . In this figure the amplitude of individual responses was normalized to the amplitude of  $I_{ATPs}$  evoked by  $10^{-5}$  M ATP.  $I_{ATPs}$  had a threshold at around  $3 \times 10^{-7}$  M and increased its amplitude in a sigmoidal fashion with increasing ATP concentration. The half maximal response was obtained at  $5.2 \times 10^{-6}$  M. At  $10^{-4}$  M  $I_{ATPs}$  reached maximum. The concentration-response relationship of  $I_{ATPs}$  was fitted by the following equation:

$$I = I_{max}(C^n / (C^n + EC_{50}^n))$$

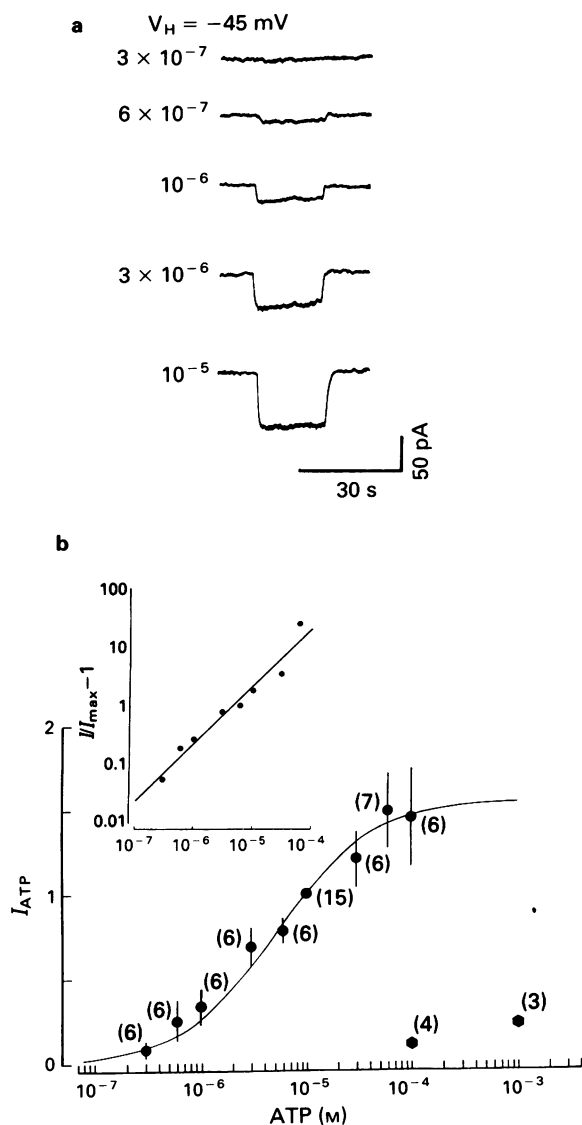
where  $I$  is the observed  $I_{ATPs}$ ,  $I_{max}$  is the maximum current,  $C$  is the applied ATP concentration,  $EC_{50}$  is the ATP concen-



**Figure 5** Effects of DIDS on  $I_{ATPs}$ . DIDS was perfused for 30 s prior to the ATP application.  $V_H = -45$  mV. Applied ATP concentration:  $10^{-5}$  M, external solution: NMG-rich, pipette solution: NMG-rich.



**Figure 6** Responses to purinoceptor agonists. Arrows shown in the trace of AMP and adenosine indicate the start and the finish of the application. All responses are recorded from the same cell.  $V_H = -45$  mV. External solution: NMG-rich, pipette solution: NMG-rich.

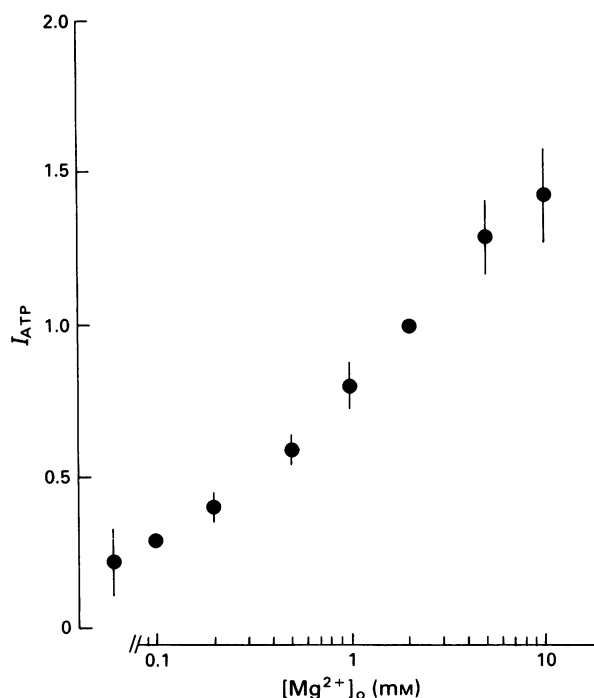


**Figure 7** (a)  $I_{ATP}$ s evoked by different ATP concentrations (M). (b) Concentration-response relationship of  $I_{ATP}$ s.  $I_{ATP}$ s were normalized by  $I_{ATP}$ s evoked by  $10^{-5}$  M ATP and plotted against ATP concentration. Symbols shown in the figure indicate the average with s.d. Numbers in parentheses indicate  $n$  values. Inset shows the Hill plot of  $I_{ATP}$ s. Solid line was drawn by the least squares method.  $V_H = -45$  mV. External solution: NMG-rich, pipette solution: NMG-rich. (●): ATP; (◆): ADP.

tration that evokes half maximal response, and  $n$  is the Hill coefficient. From the Hill plot of peak  $I_{ATP}$ s (inset), the Hill coefficient was determined as 0.98, indicating that the binding of one ATP molecule to  $P_2$ -purinoceptors activates  $I_{ATP}$ s in rat ventricular myocytes. Individual  $EC_{50}$  was also determined by the same equation in 7 cells having more than 5 points. An average of  $EC_{50}$  was  $7.15 \times 10^{-6} \pm 6.20 \times 10^{-6}$  M (mean  $\pm$  s.d.). In the Cs-rich solution the concentration-response relationship of  $I_{ATP}$ s was fitted by the similar parameters ( $EC_{50}$ ,  $2.2 \times 10^{-6}$  M; Hill coefficient, 0.95).

#### Effects of extracellular $Mg^{2+}$ ( $[Mg^{2+}]_o$ ) on $I_{ATP}$ s

It has been reported that ATP responses are sensitive to  $[Mg^{2+}]_o$  (Scamps *et al.*, 1990). We tested the effect of  $[Mg^{2+}]_o$  on  $I_{ATP}$ s evoked by  $10^{-5}$  M ATP (Cs-rich pipette solution; Figure 8). The individual currents evoked at different  $[Mg^{2+}]_o$  were normalized to the  $I_{ATP}$ s evoked in 2 mM  $Mg^{2+}$ . Then, normalized currents were averaged and plotted against



**Figure 8** Effects of  $[Mg^{2+}]_o$  on  $I_{ATP}$ s.  $I_{ATP}$ s were normalized by  $I_{ATP}$ s evoked in 2 mM  $[Mg^{2+}]_o$ . Concentration of applied ATP was  $10^{-5}$  M.  $MgCl_2$  was used for  $[Mg^{2+}]_o$  change. Symbols show the average with s.d.  $V_H = -49$  mV. External solution: K-free, pipette solution: Cs-rich.

$[Mg^{2+}]_o$ . When cells were perfused with  $Mg^{2+}$ -free solution, the observed amplitude of  $I_{ATP}$ s was  $22 \pm 11\%$  of that obtained in 2 mM  $Mg^{2+}$  ( $n = 5$ ).  $I_{ATP}$ s increased its amplitude with increasing  $[Mg^{2+}]_o$ . In 10 mM  $Mg^{2+}$  the amplitude of  $I_{ATP}$ s was 1.4 times larger than in 2 mM  $Mg^{2+}$  ( $n = 7$ ). Thus in rat ventricular myocytes  $I_{ATP}$ s is affected by  $[Mg^{2+}]_o$ .

#### Discussion

We have identified an ATP-activated  $Cl^-$ -current in rat ventricular myocytes by the intra- and extracellular perfusion of Na, K, Ca-free NMG solution. Under these conditions, the only permeant ions are anions, including  $Cl^-$ . In fact, the shift of  $E_{ATP}$ s fitted well to the theoretical value of  $E_{Cl}$  calculated from the Nernst equation, strongly suggesting that  $I_{ATP}$ s is carried by  $Cl^-$ . A similar ATP-induced  $Cl^-$  current was recently identified in guinea-pig atrial cells (Matsuura & Ehara, 1992).

In the present experiments we also observed ATP-induced  $K^+$  currents. Thus, in ventricular myocytes,  $P_2$ -purinoceptors activation appears to induce both  $Cl^-$ - and  $K^+$ -conductances independently. In previous reports it has been shown that ATP activates two different types of conductances in guinea-pig (Hirano *et al.*, 1991) and bullfrog atria (Friel & Bean, 1988), one component of which was considered to be carried by  $K^+$ . In bullfrog atrium, ATP-induced  $K^+$  conductances are thought to be identical to muscarinic ATP-induced  $K^+$  conductances, which do not exist in ventricular cells (Friel & Bean, 1988). Thus, it would be interesting to see if ATP-induced  $K^+$  conductances are sharing same  $K^+$  channels with other  $K^+$  conductances in rat ventricular myocytes.

In symmetrical  $Cl^-$   $I_{ATP}$ s reversed its polarity at around 0 mV and had a linear  $I-V$  relationship. Previous studies of ATP-currents reported linear  $I-V$  relationships and reversal potentials around 0 mV in symmetrical  $Cl^-$  solution in the rat ventricular myocytes (Scamps & Vassort, 1990; Christie *et al.*, 1992). From the characteristics of  $I-V$  curve these authors

considered the existence of cation conductances. In the present experiments, however, the observed  $I_{ATP}$  was mediated by  $Cl^-$ . Therefore, we conclude that the ATP-induced conductance having a linear  $I-V$  relationship is mediated by  $Cl^-$  and not by cations.

In the rat ventricular myocytes, previous reports show an increase in  $[Ca^{2+}]_i$  by an application of ATP (Danziger *et al.*, 1988; Björnsson *et al.*, 1989; de Young & Scarpa, 1987, 1989). In ventricular myocytes, the equilibrium potential for  $Cl^-$  is thought to be more positive than the resting potential. Activation of  $I_{ATP}$  would therefore induce depolarization within a few seconds. In fact, ATP-induced depolarization has been reported in rat ventricular myocytes, where superimposition of action potentials occurs within 1–2 s after the start of depolarization (Christie *et al.*, 1992). Therefore,  $[Ca^{2+}]_i$  increase as a result of the activation of voltage-gated  $Ca^{2+}$  channels was considered. Actually this hypothesis corresponds well to the previous observation of simultaneous recording of membrane potentials and  $[Ca^{2+}]_i$  in the rat (Björnsson *et al.*, 1989). As to the initiation of an increase in  $[Ca^{2+}]_i$ , the possibilities of the involvement of the intracellular acidification by a putative  $P_3$ -purinoceptor (Puceat *et al.*, 1991b) and extracellular membrane phosphorylation (Christie *et al.*, 1992) have been reported. However, these two possibilities are not likely to be the triggering mechanism since it seems difficult to explain the time course of  $[Ca^{2+}]_i$  increase by these mechanisms. Using fluorescent dyes they reported that the increase in  $[Ca^{2+}]_i$  reached a peak at around 15–20 s after the ATP application and then elevated  $[Ca^{2+}]_i$  started to return to the control level. On the other hand, peak res-

ponses for both intracellular acidification (Puceat *et al.*, 1991b) and extracellular phosphorylation (Christie *et al.*, 1992) were observed at 60 s after the ATP application. Thus,  $[Ca^{2+}]_i$  increase reached a peak before the other two responses reached that peak. Therefore, we concluded that the initiation of  $[Ca^{2+}]_i$  increase was triggered by an activation of ATP-induced depolarization which is followed by voltage gated  $Ca^{2+}$  channel activation.

In the rat, it has been reported that ATP induces intracellular acidification by the activation of a putative  $P_3$ -purinoceptor which is coupled to a  $HCO_3^-Cl^-$  exchanger (Puceat *et al.*, 1991b). In the present experiments we found  $P_2$ -purinoceptor-activated  $Cl^-$  conductance. If the activation of  $I_{ATP}$  occurs at the resting potential,  $Cl^-$  ions would move from the intracellular space to the extracellular space. This depletion of the intracellular  $Cl^-$  could activate the  $HCO_3^-Cl^-$  exchanger leading to intracellular acidification. Thus, it is also possible to explain the previously reported intracellular acidification by the activation of a  $P_2$ -purinoceptor. Actually, the extracellular ATP concentrations which gave the threshold,  $EC_{50}$  and the maximal response of  $I_{ATP}$  in the present experiments were similar to those reported for the ATP-induced intracellular acidification (Puceat *et al.*, 1991b). Thus, it will be interesting to see if the present result could explain some of the previous observation that have been ascribed to the existence of a putative  $P_3$ -purinoceptors.

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