

# Quantification of the characteristics of antagonists exhibiting both competitive antagonism and functional interaction

I.E. Hughes & D. Mackay<sup>1</sup>

Department of Pharmacology, Worsley Medical and Dental Building, University of Leeds, Leeds-2, LS2 9JT, Yorkshire.

1 Null equations have been derived which, when applied to  $\log_{10}$  concentration-tissue state curves for an agonist determined in the presence and absence of a competitive antagonist which also exhibits functional interaction, allow quantitation of the characteristics of the competitive and functional interactant effects.

2 Both the affinity constant of the antagonist for its receptors and numerical values characterizing the functional interaction can be obtained.

3 The null equations have been tested in a model system by using a mixture of papaverine hydrochloride (either 5 or 20  $\mu\text{M}$ ) and methyl atropine bromide (10 nM) to mimic a competitive antagonist which also shows functional interaction.

4 Agreement between values derived directly and indirectly from the model is good and validates the use of the null equations.

## Introduction

Antagonists tend to be classified according to their main pharmacological effects but it is not uncommon for antagonists to have more than one action. Such mixed actions may result in Schild plots which are non-linear or whose slopes differ from unity. Estimation of affinity constants by methods which assume competitive action is invalid in these circumstances and quantitation of antagonistic action presents a problem. This is especially acute in studies of structure-activity relationship where drugs may have a variety of actions and an estimate of the effectiveness of interaction of the drug with a particular receptor type may be required so that structures which optimise interaction with that receptor may be selected.

The present paper discusses the derivation and use of null equations which, when applied to  $\log_{10}$  concentration-tissue state curves, can be used to separate the competitive and 'non-competitive' (i.e. functional antagonistic or synergistic) activities of such mixed antagonists.

### Derivation of null equations

The simplest approach to the problem is to consider what would be expected to happen to the

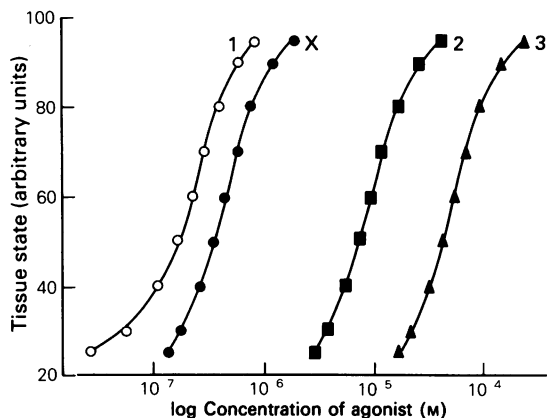
$\log_{10}$  concentration-tissue state curve for an agonist A if it is determined in the presence of a constant concentration of a pure functional interactant X (antagonist or synergist) and then in the presence of the same concentration of X together with a fixed concentration of a purely competitive antagonist  $I_1$ . The expected changes are shown in Figure 1.

The general null equation which summarises the relation between curve 1 and curve  $\times$  (Mackay, 1981) is

$$\frac{[A]_{\times}}{[A]_1} = \alpha_{\times} + \beta_{\times}[A]_{\times} + \frac{\gamma_{\times}}{[A]_1} \quad (1)$$

Where  $[A]_{\times}$  and  $[A]_1$  are equieffective molar concentrations of agonist A read from curve  $\times$  and curve 1 respectively. The quantities  $\alpha_{\times}$ ,  $\beta_{\times}$  and  $\gamma_{\times}$  are adjustable curve fitting constants. Since the presence of the pure functional interactant X merely changes the state of the tissue the addition of a fixed concentration,  $[I_1]$ , of the competitive antagonist  $I_1$ , in the presence of the previous concentration of X, would be expected to produce curve 2, to the right of curve  $\times$  and parallel to it. The appropriate null equation which relates curve 2 to curve  $\times$  is the well-known equation for competitive antagonism,

<sup>1</sup> Author for correspondence.



**Figure 1**  $\log_{10}$  agonist molar concentration-tissue state curves obtained by calculation from the null equations assuming the following values:  $\alpha_x 1.3$ ,  $\beta_x 0.05 \times 10^7$ ,  $\gamma_x 1.0 \times 10^{-7}$ ,  $K_{11} 2.0 \times 10^9 M^{-1}$ ,  $[I_1] = 1.0 \times 10^{-8} M$ ,  $K_{12} 1.0 \times 10^9 M^{-1}$ ,  $[I_2] = 1 \times 10^{-7} M$ . These have been chosen to mimic more closely the curves obtained from rat jejunum. Curve 1 is for the agonist alone, curve  $\times$  for agonist + functional interactant X, curve 2 for agonist + functional interactant X + competitive antagonist  $I_1$ , curve 3 for agonist + functional interactant + competitive antagonist  $I_1$  + competitive antagonist  $I_2$ .

$$\frac{[A]_2}{[A]_1} = 1 + K_{11}[I_1] \quad (2)$$

where  $K_{11}$  is the affinity constant of the antagonist-receptor complex.

Eliminating  $[A]_x$  between equations 1 and 2 gives the null equation which relates curve 2 to curve 1, namely

$$\frac{[A]_2}{[A]_1} = \alpha_{21} + \beta_{21}[A]_2 + \frac{\gamma_{21}}{[A]_1} \quad (3a)$$

where  $[A]_2$  and  $[A]_1$  are equieffective concentrations of agonist A read from curve 2 and curve 1 respectively,

$$\begin{aligned} \alpha_{21} &= \alpha_x \{1 + K_{11}[I_1]\} \\ \beta_{21} &= \beta_x \\ \text{and } \gamma_{21} &= \gamma_x \{1 + K_{11}[I_1]\} \end{aligned} \quad (3b)$$

The mixture of functional interactant and pure competitive antagonist imitates the properties of a mixed antagonist but in a real situation the 'components' of the mixture could not be separated and only curves 1 and 2 would be obtained experimentally.

Comparison of these curves would not allow calculation of  $K_{11}$ ,  $\alpha_x$ ,  $\beta_x$  and  $\gamma_x$  which are the quantities

of interest. However, this difficulty can be overcome if another antagonist  $I_2$  is available which is a pure competitive antagonist for the same receptors as those for which  $I_1$  and A compete. Curve 3 of figure 1 represents a  $\log_{10}[A]$ -tissue state curve for the agonist A estimated in the presence of the previously used concentrations of the functional interactant X and of the competitive antagonist  $I_1$  together with a fixed concentration of  $I_2$ . The null equation which relates curve 3 to curve 2 can readily be shown to be

$$\frac{[A]_3}{[A]_2} = 1 + \{K_{12}[I_2]/(1 + K_{11}[I_1])\} \quad (4)$$

where  $[A]_2$  and  $[A]_3$  are equieffective concentrations of agonist A read from curves 2 and 3 respectively and  $K_{11}$  and  $K_{12}$  are the affinity constants for the antagonist-receptor complexes of antagonists  $I_1$  and  $I_2$  used at concentrations  $[I_1]$  and  $[I_2]$  respectively.

Since  $K_{12}$  for the purely competitive antagonist  $I_2$  can be estimated by the standard methods for  $pA_2$  determinations, it follows that the value of  $K_{11}$  can be calculated from the parallel displacement of curve 3 relative to curve 2. Equation 4 should be valid whether or not  $I_1$  is a mixed antagonist. Use of equation 4 should therefore provide information about the value of  $K_{11}$  in all circumstances. If the functional interactant properties of a mixed antagonist are also of interest then the values of  $\alpha_{21}$ ,  $\beta_{21}$ , and  $\gamma_{21}$  can be obtained by applying equation 3a to the  $\log_{10}[A]$ -tissue state curves obtained in the absence and presence of the mixed antagonist. The values for  $\alpha_x$ ,  $\beta_x$  and  $\gamma_x$  can then be calculated from  $\alpha_{21}$ ,  $\beta_{21}$  and  $\gamma_{21}$  knowing the values of  $K_{11}$  and  $[I_1]$  (see equations 3b). Possible interpretation of the values of  $\alpha_x$ ,  $\beta_x$  and  $\gamma_x$  have been discussed elsewhere (Mackay, 1981).

## Methods

Experiments were carried out on jejunum taken from male rats (150 to 220 g) and on ileum taken from male guinea-pigs (200 to 350 g). The length of the tissue was recorded isotonicly and this was taken as a measure of the 'state' of the tissue in arbitrary units. In the case of the rat jejunum the tissue was bathed in Krebs solution (NaCl 118, KCl 4.7, MgSO<sub>4</sub> 0.6, KH<sub>2</sub>PO<sub>4</sub> 1.1, NaHCO<sub>3</sub> 25, CaCl<sub>2</sub> 2.5 and glucose 11 mM), gassed with 5% CO<sub>2</sub> in O<sub>2</sub>, at 35°C. For guinea-pig ileum the physiological solution was Tyrode solution (NaCl 137, KCl 2.7, MgSO<sub>4</sub> 1, NaH<sub>2</sub>PO<sub>4</sub> 0.3, NaHCO<sub>3</sub> 12, CaCl<sub>2</sub> 1.8 and glucose 5.5 mM), gassed with O<sub>2</sub>, at 32°C. For both tissues the resting tension was approximately 1 g. Only in the case of the rat jejunum were the  $\log_{10}$  concentration-tissue state curves determined using the cumulative technique and with both tissues the antagonists were added to the

physiological saline and allowed to remain in contact with the tissue for at least 20 min before measurements were made. For each experiment smoothed curves were drawn by eye through the experimentally-determined points and equi-effective concentrations of agonist were read from the smoothed curves. These equi-effective concentrations were then substituted into the appropriate null equations to obtain values of drug-receptor affinity constants and other parameters.

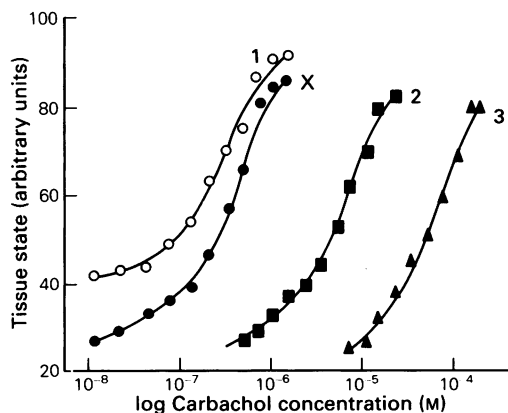
#### Drugs used

Atropine methyl bromide (Sigma), atropine sulphate (Sigma), carbachol chloride (BDH) and papaverine hydrochloride (Sigma) were used.

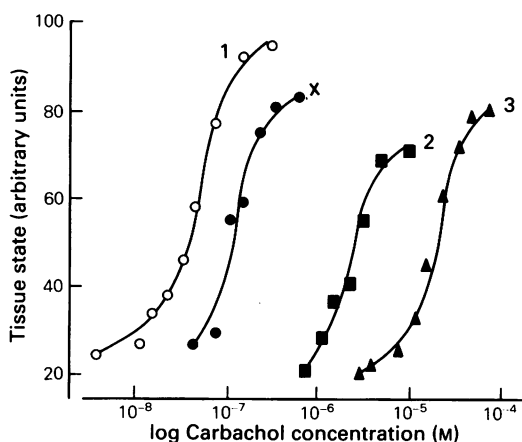
#### Results

Experiments were carried out on rat isolated jejunum using carbachol as the agonist and atropine as the antagonist, papaverine ( $5 \mu\text{M}$ ) being present in both cases. Comparison of the  $\log_{10}$ [carbachol]-tissue state curves gave a value of  $(0.99 \pm 0.32) \text{ nM}^{-1}$  ( $n = 4$ ) as the affinity constant of atropine for the muscarinic receptors of this tissue.

A series of experiments was then carried out on rat jejunum and on guinea-pig ileum. In each case the  $\log_{10}$ [carbachol]-tissue state curves were determined for the agonist alone (curve 1), in the presence of



**Figure 2**  $\log_{10}$ [carbachol]-tissue state curves obtained on rat isolated jejunum. The state of the tissue was recorded isotonicly and expressed in arbitrary units. Curve 1, carbachol alone (O); curve X, carbachol + papaverine ( $5 \mu\text{M}$ , ●); curve 2, carbachol + papaverine + methyl atropine ( $10 \text{ nM}$ , ■); curve 3, carbachol + papaverine + methyl atropine + atropine ( $100 \text{ nM}$ , ▲). From such data the values given in Table 1 were obtained.



**Figure 3**  $\log_{10}$ [carbachol]-tissue state curves obtained on guinea-pig isolated ileum. The state of the tissue was recorded isotonicly and expressed in arbitrary units. Curve 1, carbachol alone (O). Curve X, carbachol + papaverine ( $20 \mu\text{M}$ ; ●). Curve 2, carbachol + papaverine + methyl atropine ( $10 \text{ nM}$ ; ■). Curve 3, carbachol + papaverine + methyl atropine + atropine ( $100 \text{ nM}$ ; ▲). From such data the values given in Table 2 were obtained.

papaverine (curve X), in the presence of papaverine and methyl atropine (curve 2) and finally in the presence of papaverine, methyl atropine and atropine (curve 3). Typical examples of results obtained in such experiments are shown in Figures 2 and 3.

Comparison of curve X with curve 1 gave *direct* estimates of  $\alpha_x$ ,  $\beta_x$  and  $\gamma_x$  (see equation 1). Comparison of curve 2 and curve X provided a *direct* estimation of  $K_{11}$ , the affinity constant of methyl atropine for the muscarinic receptors (equation 2). By comparing curve 3 with curve 2 and assuming a value for  $K_{12}$  (i.e.  $1 \text{ nM}^{-1}$  for atropine) an *indirect* estimate was made of  $K_{11}$  using equation 4. Finally, by comparing curve 2 with curve 1 values were obtained for  $\alpha_{21}$ ,  $\beta_{21}$  and  $\gamma_{21}$  (see equation 3a). These were used to obtain *indirect* estimates of  $\alpha_x$ ,  $\beta_x$  and  $\gamma_x$  by using equations 3b together with the indirectly estimated value of  $K_{11}$ . The results of such calculations are shown in Tables 1 and 2 for experiments carried out on rat jejunum and guinea-pig ileum respectively.

#### Discussion

Examination of Figures 1, 2 and 3 shows that papaverine and combinations of papaverine with competitive antagonists produce shifts in the  $\log_{10}$ [concentration]-tissue state curve which are

**Table 1** Comparison of direct and indirect estimates of  $K_{11}$ ,  $\alpha_x$ ,  $\beta_x$  and  $\gamma_x$  obtained on rat isolated jejunum

Experiment number	Direct estimates				Indirect estimates			
	$\alpha_x$	Curve $\times$ cf. curve 1 $\beta_x \times 10^{-7}M$ $\gamma_x \times 10^7M^{-1}$		Curve 2 cf. curve $\times$ $K_{11}(nM^{-1})$	$\alpha_x$	Curve 2 cf. curve 1 $\beta_x \times 10^{-7}M$ $\gamma_x \times 10^7M^{-1}$		Curve 3 cf. curve 2 $K_{11}(nM^{-1})$
1	1.80	0.02	0.26	1.10	1.85	0.04	0.67	1.10
2	1.46	-0.04	1.30	1.20	1.77	0.03	2.10	0.95
3	0.52	0.08	1.30	1.60	0.60	0.11	0.97	3.10
4	1.11	0.26	0.17	1.00	2.20	0.04	0.45	0.46
5	2.34	-0.08	0.44	0.37	2.54	0.10	1.20	0.39

The curves used were for carbachol, Curve 1: alone; Curve  $\times$ : + papaverine ( $5 \times 10^{-6}M$ ); Curve 2: + papaverine ( $5 \times 10^{-6}M$ ) + methyl atropine,  $I_1$ , ( $10^{-8}M$ ); Curve 3: + papaverine ( $5 \times 10^{-6}M$ ) + methyl atropine ( $10^{-8}M$ ) + atropine ( $10^{-7}M$ ).

qualitatively as expected. In the model tested here the actions of papaverine and of methyl atropine can be analysed separately to provide *direct* estimates of  $\alpha_x$ ,  $\beta_x$ ,  $\gamma_x$  and  $K_{11}$ . However, normally, the mixed effect mimicked by papaverine and methyl atropine would be produced by a single compound with mixed actions and curve  $\times$  (see Figures 1 and 2) would not be obtainable experimentally. In such a case curves 1, 2 and 3 can be analysed using equations 4, 3a and 3b to obtain *indirect* estimates of  $K_{11}$ ,  $\alpha_x$ ,  $\beta_x$  and  $\gamma_x$ .

Comparison of the direct and indirect estimates of  $K_{11}$  (see Tables 1 and 2), shows quite good agreement, though the variability of the indirect estimates is greater. This increased variability is not surprising in view of the more complex and prolonged nature of the experiments required to obtain indirect estimates of  $K_{11}$ . Comparison of direct estimates of  $\alpha_x$ ,  $\beta_x$  and  $\gamma_x$  for each experiment with the indirect estimates also shows generally good agreement allowing for normal variation and changes in tissue sensitivity (see Tables 1 and 2).

The agreement between the direct and indirect estimates for  $K_{11}$  supports the use of equation 4 to obtain values of this quantity for antagonists with mixed competitive and functional properties. However, as already mentioned, the method should also be applicable if  $I_1$  is a pure competitive antagonist. In this special case the value of  $K_{11}$  obtained by applying equation 4 should be in agreement with that estimated by applying equation 2 to what should then be a parallel shift of the  $\log_{10}$  concentration-tissue state curve produced by  $I_1$  acting alone. In these special circumstances the technique described above would be closely related to that suggested by Paton & Rang (1965) for testing whether two antagonists compete for the same receptors, though the emphasis in this present paper is on the estimation of  $K_{11}$  knowing  $K_{12}$ .

In conclusion, the null equations presented here have been shown to be capable of giving satisfactory quantitative information about both the competitive and functional interactive properties of substances which have both types of action.

**Table 2** Comparison of direct and indirect estimates of  $K_{11}$ ,  $\alpha_x$ ,  $\beta_x$  and  $\gamma_x$  obtained on guinea-pig isolated ileum

Experiment number	Direct estimates				Indirect estimates			
	$\alpha_x$	Curve $\times$ cf. curve 1 $\beta_x \times 10^{-7}M$ $\gamma_x \times 10^7M^{-1}$		Curve 2 cf. curve $\times$ $K_{11}(nM^{-1})$	$\alpha_x$	Curve 2 cf. curve 1 $\beta_x \times 10^{-7}M$ $\gamma_x \times 10^7M^{-1}$		Curve 3 cf. curve 2 $K_{11}(nM^{-1})$
1	1.14	1.41	5.85	1.40	1.38	1.01	4.30	1.13
2	2.63	1.36	0.25	1.70	2.74	1.23	0.14	2.40
3	1.52	0.39	0.44	1.16	1.39	1.09	0.70	0.51

The curves used were for carbachol, Curve 1: alone; Curve  $\times$ : + papaverine ( $2 \times 10^{-5}M$ ); Curve 2: + papaverine ( $2 \times 10^{-5}M$ ) + methyl atropine,  $I_1$ , ( $10^{-8}M$ ); Curve 3: + papaverine ( $2 \times 10^{-5}M$ ) + methyl atropine ( $10^{-8}M$ ) + atropine ( $10^{-7}M$ ).

**References**

MACKAY, D. (1981). An analysis of functional antagonism and synergism. *Br. J. Pharmac.*, **73**, 127–134.  
PATON, W.D.M. & RANG, H.P. (1965). The uptake of atropine

and related drugs by intestinal smooth muscle of the guinea-pig in relation to acetylcholine receptors. *Proc. R. Soc. B.*, **163**, 1–44.

*(Received November 26, 1984.  
Accepted December 23, 1984.)*