

Yuji Ashikawa,^a Zui Fujimoto,^b
Haruko Noguchi,^a Hiroshi
Habe,^a Toshio Omori,^{a,‡}
Hisakazu Yamane^a and Hideaki
Nojiri^{a*}

^aBiotechnology Research Center, The University of Tokyo, 1-1-1 Yayoi, Bunkyo-ku, Tokyo 113-8657, Japan, and ^bDepartment of Biochemistry, National Institute of Agrobiological Sciences, 2-1-2 Kannondai, Tsukuba, Ibaraki 305-8602, Japan

‡ Present address: Shibaura Institute of Technology, Shibaura 3-9-14, Minato-ku, Tokyo 108-8548, Japan.

Correspondence e-mail:
anojiri@mail.ecc.u-tokyo.ac.jp

Received 25 February 2005
Accepted 6 May 2005
Online 1 June 2005

Crystallization and preliminary X-ray diffraction analysis of the electron-transfer complex between the terminal oxygenase component and ferredoxin in the Rieske non-haem iron oxygenase system carbazole 1,9a-dioxygenase

Carbazole 1,9a-dioxygenase, which consists of an oxygenase component (CARDO-O) and the electron-transport components ferredoxin (CARDO-F) and ferredoxin reductase (CARDO-R), catalyzes dihydroxylation at the C1 and C9a positions of carbazole. The electron-transport complex between CARDO-O and CARDO-F crystallizes at 293 K using hanging-drop vapour diffusion with the precipitant PEG MME 2000 (type I crystals) or PEG 3350 (type II). Blossom-shaped crystals form from a pile of triangular plate-shaped crystals. The type I crystal diffracts to a maximum resolution of 1.90 Å and belongs to space group $P2_1$, with unit-cell parameters $a = 97.1$, $b = 89.8$, $c = 104.9$ Å, $\alpha = \gamma = 90$, $\beta = 103.8^\circ$. Diffraction data for the type I crystal gave an overall R_{merge} of 8.0% and a completeness of 100%. Its V_M value is $2.63 \text{ \AA}^3 \text{ Da}^{-1}$, indicating a solvent content of 53.2%.

1. Introduction

Rieske non-haem iron oxygenase systems (ROSSs) are the initial catalysts in the degradation pathways of various environmentally significant aromatic compounds, including dioxins, polychlorinated biphenyls and crude-oil components such as polycyclic aromatic hydrocarbons and carbazole (Wittich, 1998; Bressler & Fedorak, 2000; Nojiri & Omori, 2002; Habe & Omori, 2003; Furukawa *et al.*, 2004). In the initial dioxygenation of carbazole and dioxins, one carbon that is bonded to the heteroatom and its adjacent carbon in the aromatic ring are both hydroxylated. This reaction, called angular dioxygenation, is catalyzed by a limited number of ROSSs, which are called angular dioxygenases (Nojiri & Omori, 2002).

Since the *car* operon containing the genes encoding the angular dioxygenase system carbazole 1,9a-dioxygenase (CARDO; Fig. 1) was first isolated from a carbazole-degrader, *Pseudomonas resinovorans* strain CA10 (Sato *et al.*, 1997), we have studied the enzymatic function of this novel enzyme system extensively (Nojiri & Omori, 2002; Inoue *et al.*, 2004). We have purified and characterized each component of CARDO of strain CA10 (CARDO_{CA10}): the terminal

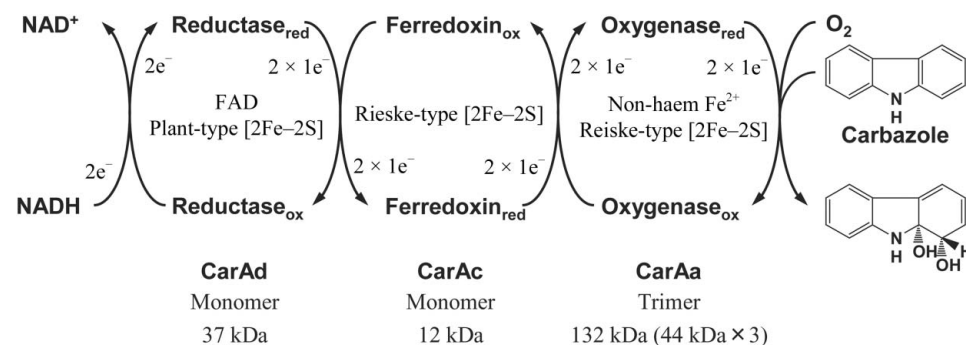
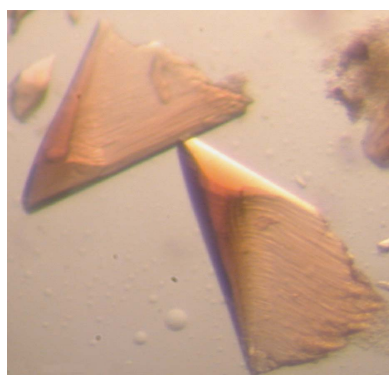


Figure 1 Components and functions of the CARDO system, showing the electron-transfer reactions and conversion of carbazole to an unstable product, 1,9a-dihydroxy-1-hydrocarbazole, which has not been detected directly.

oxygenase component (CarAa_{CA10} trimer; CARDO-O_{CA10}) and the electron-transport components ferredoxin (CarAc_{CA10} monomer; CARDO-F_{CA10}) and ferredoxin reductase (CarAd_{CA10} monomer; CARDO-R_{CA10}) (Nam *et al.*, 2002). A single subunit of CARDO-O_{CA10} contains a Rieske [2Fe–2S] cluster and a non-haem iron. CARDO-F_{CA10} and CARDO-R_{CA10} each contain a Rieske [2Fe–2S] cluster and a plant-type [2Fe–2S] cluster and an FAD, respectively. Although these characteristics of the electron-transport components suggest that the CARDO_{CA10} system can be classified as a class III ROS according to Batie's classification (Batie *et al.*, 1991), CARDO is a novel ROS owing to several important distinguishing characteristics. For example, the amino-acid sequence of CarAa_{CA10} shares rather low homology (<19% identity) with the corresponding proteins of other ROSs and shows greatest homology (35% identity) with the monooxygenase component (OxoO) from *Pseudomonas putida* strain 86 (Rosche *et al.*, 1995), except for other CarAa proteins from different bacterial sources. In addition, CARDO-O_{CA10} consists of only one catalytic subunit with an α_3 configuration, which is a typical feature of class IA ROSs in Batie's system. Therefore, the electron-transport mechanism in CARDO_{CA10} has several unusual features. In addition, studies of its substrate specificity have shown that CARDO_{CA10} has a broad substrate range and that it catalyzes not only angular dioxygenation but also other oxygenation reactions, such as lateral dioxygenation (*e.g.* 1,2-dihydroxylation of naphthalene) and monooxygenation (9-hydroxylation of fluorene and sulfoxidation of dibenzothiophene) (Nojiri *et al.*, 1999; Takagi *et al.*, 2002). CARDO_{CA10} also attacks some chlorinated dioxin congeners (Habe *et al.*, 2001). Considering these unique properties of the CARDO system, structure-based elucidation of CARDO function is of great interest.

From a survey of the distribution and diversity of this novel enzyme, we have obtained several bacterial strains that have a *car* operon similar to that of strain CA10. One of these strains, *Janthinobacterium* sp. strain J3, has a *carAa* gene that has 47 nucleotide mismatches with the gene of strain CA10, which results in three amino-acid substitutions in its 384-amino-acid length. No functional differences were observed between CARDO-O of strain J3 (CARDO-O_{J3}) and CARDO-O_{CA10} in the analysis of substrate specificity using resting cells of CARDO-expressing *Escherichia coli* (data not shown). Therefore, we tried to determine the crystal structure of the CARDO components using proteins of strains CA10 and J3. Recently, we succeeded in determining the crystal structures of CARDO-F_{CA10} (Nam *et al.*, 2005) and CARDO-O_{J3} (Nojiri *et al.*, 2005). In this study, as a first attempt to reveal the electron-transfer

mechanism of CARDO, we report the crystallization and preliminary X-ray diffraction studies of the complex of CARDO-O_{J3} with CARDO-F_{CA10}.

2. Materials and methods

2.1. Protein expression, purification and characterization

In this study, we used CARDO-O_{J3} and CARDO-F_{CA10} as CARDO components because expression and purification procedures for their crystallization have been established and because the crystal structures of both components have been determined (Nam *et al.*, 2005; Nojiri *et al.*, 2005). Both proteins were expressed in *E. coli* strains as six-histidine-tagged proteins at their C-terminal regions. A thorough description of the expression and purification of CARDO-O_{J3} will be provided elsewhere (Nojiri *et al.*, 2005). CARDO-F_{CA10} was expressed and purified as described previously (Nam *et al.*, 2002), except that 20 mM Tris–HCl buffer pH 7.5 containing 0.5 M NaCl was used to resuspend the *E. coli* cells.

The two proteins were purified separately using metal-chelation chromatography and were then mixed and subjected to gel-filtration chromatography using Superdex-200 (26 × 600 mm; Amersham Bioscience, NJ, USA). As the two proteins were eluted separately with gel-filtration chromatography, the respective fractions were collected. Each protein was concentrated separately and buffer-exchanged with 50 mM Tris–HCl pH 7.5 using Vivaspin 20 membranes (10 000 MWCO; Sartorius KK, Gottingen, Germany) or Amicon Centriprep YM-10 membranes (Millipore, MA, USA). Protein concentrations were estimated with a protein assay kit (Bio-Rad, CA, USA; Bradford, 1976) using BSA as a standard.

The electron transferability between the purified CARDO-O_{J3} and CARDO-F_{CA10} was confirmed by detecting oxygenase activity for carbazole of reconstituted CARDO (Nam *et al.*, 2002).

Western blot analysis was performed using the ECL Plus Western Blotting Detection System (Amersham Bioscience) according to the manufacturer's instructions. Denatured samples of dissolved crystals were separated using 15% SDS–PAGE and electrophoretically transferred onto Trans-Blot Transfer Medium Nitrocellulose membranes (Bio-Rad). The CarAc_{CA10} protein was detected immunologically using a mouse polyclonal antibody as the primary antibody and a horseradish-peroxidase-labelled antibody as the secondary antibody. The CarAc_{CA10} protein signals were visualized using the Luminescent Image Analyzer LAS-1000 Plus (Fuji Photo Film, Tokyo, Japan).

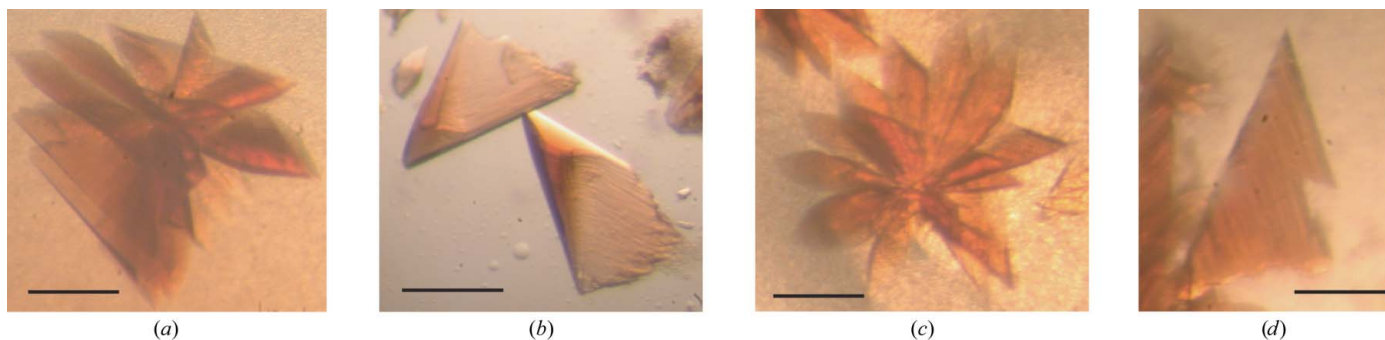


Figure 2

Photographs of the CARDO-O_{J3} and CARDO-F_{CA10} complex crystals. The lines indicate 0.2 mm. (a) Type I crystals grown in 1–2 mM sodium dithionite, 14%(v/v) PEG MME 2000 and 0.05 M Bis-Tris pH 6.5. (b) Triangular plate-shaped crystals; pieces of the blossom-like crystals shown in (a). (c) Type I crystals grown in 0.1 M ammonium acetate, 12.5%(v/v) PEG 3350 and 0.05 M Bis-Tris pH 5.5. (d) Triangular plate-shaped crystals; pieces of the blossom-like crystals shown in (c).

2.2. Crystallization

For crystallization experiments, purified CARDO-O_{J3} and CARDO-F_{CA10} proteins were mixed in an approximate molar ratio of 1:6 or 1:3. The concentration of the protein mixture in 50 mM Tris-HCl pH 7.5 was adjusted to 15–30 mg ml⁻¹. The proteins were crystallized using the hanging-drop vapour-diffusion method at 293 K. Drops containing 3 µl protein solution and 3 µl mother liquor were equilibrated against 700 µl reservoir solution. The initial crystallization was screened using Crystal Screen I/II, Grid Screens (PEG 6000 and MPD), Index Screens (Hampton Research, CA, USA), Wizard I/II (Jena Bioscience, Jena, Germany) and JBScreen (Emerald Biostructures, WA, USA). The initial crystallization trials showed that protein crystals appeared under a variety of conditions. After optimizing the crystallization parameters, protein concentration and molar ratio in the protein mixture, blossom-shaped crystals built up from piles of triangular plates were obtained with two reservoir solutions using VDX Plates (Hampton Research) and 20 mg ml⁻¹ protein solution (CARDO-O_{J3}:CARDO-F_{CA10} = 1:6) at 293 K (drops contained 3 µl protein solution and 3 µl precipitating solution). Type I crystals (Figs. 2a and 2b) were formed with 1–2 mM sodium dithionite and 14% (v/v) PEG MME 2000 in 0.05 M Bis-Tris pH 6.5 and type II crystals (Figs. 2c and 2d) were formed with 0.1 M ammonium acetate and 12.5% (v/v) PEG 3350 in 0.05 M Bis-Tris pH 5.5. Growth of type I and type II crystals was observed after two weeks and 2–3 d, respectively. The triangle plate-shaped crystal pieces grew to maximum dimensions of 0.3 × 0.2 × 0.05 mm after four weeks and one week for types I and II, respectively.

2.3. Data collection

Owing to the sensitivity of the crystals to X-ray radiation, cryo-cooling was necessary during data collection to avoid crystal decay at room temperature. Both crystals were directly transferred into cryoprotectant solution (reservoir solution containing 20% glycerol), mounted in a nylon loop and flash-frozen in a nitrogen stream at 100 K. Diffraction experiments were conducted at beamlines AR-NW12 and BL-5A, Photon Factory, Tsukuba, Japan. Diffraction data were gathered with a wavelength of 1.0 Å using a Quantum 210 CCD X-ray detector (ADSC, CA, USA) at AR-NW12 and a Quantum 315 CCD X-ray detector (ADSC, CA, USA) at BL-5A. For the type I crystal, two data sets were collected initially using a single crystal in 0.5° oscillation steps over a range of 180° with 15 s exposure per frame and in 0.5° oscillation steps over a range of 270° with 2 s exposure per frame to cover the low-resolution diffraction; the two

Table 1

Crystal parameters and data-collection statistics.

The data were collected on AR-NW12 and BL-5A at the Photon Factory, Tsukuba, Japan. Values in parentheses are for the highest resolution shell.

Crystal type	Type I	Type II
Beamline	AR-NW12	BL-5A
Wavelength (Å)	1.0	1.0
Space group	<i>P</i> 2 ₁	<i>P</i> 2 ₁
Unit-cell parameters (Å, °)	<i>a</i> = 97.1, <i>b</i> = 89.8, <i>c</i> = 104.9, α = γ = 90.0, β = 103.8	<i>a</i> = 98.2, <i>b</i> = 90.0, <i>c</i> = 105.1, α = γ = 90.0, β = 104.0
Resolution range (Å)	50.0–1.90 (1.97–1.90)	50.0–2.05 (2.12–2.05)
Total No. of reflections	1131785	402412
No. of unique reflections	138930 (13797)	109845 (10911)
Completeness (%)	100 (100)	99.6 (99.7)
Average <i>I</i> /σ(<i>I</i>)	47.8 (3.3)	31.7 (3.7)
<i>R</i> _{merge} (%)	8.0 (41.9)	7.2 (37.5)
Multiplicity	8.1 (3.7)	3.7 (3.7)

data sets were scaled and merged into a single data set using the *HKL2000* program suite (Otwinowski & Minor, 1997). The type II crystal data set was collected in 0.5° oscillation steps over a range of 180° with 30 s exposure per frame. All diffraction images were indexed, integrated and scaled using the *HKL2000* program suite.

3. Results and discussion

Before crystallization, the purified CARDO-O_{J3} was confirmed to retain its angular dioxygenation activity for carbazole when the electron-transfer proteins CARDO-F_{CA10} and CARDO-R_{CA10} were supplied (data not shown). This result clearly indicates that CARDO-O_{J3} can receive electrons from CARDO-F_{CA10}.

To ensure that the two types of crystal consisted of both CARDO-O_{J3} and CARDO-F_{CA10}, SDS-PAGE followed by Western blot analysis was performed using the dissolved crystals. The SDS-PAGE analysis clearly indicated that both types of crystals contained CarAa_{J3} (CARDO-O_{J3} monomer; Figs. 3a and 3b, lanes 2 and 3). This result was further confirmed by Western blot analysis, which showed clear detection of CarAa_{J3} with ~44 kDa molecular weight (data not shown). On the other hand, because the degradation peptide that originated from CarAa_{J3} had a molecular size similar to that of CarAc_{CA10} (Fig. 3a, lanes 1, 2 and 3), we could not conclude that both types of crystals contained CarAc_{CA10} (CARDO-F_{CA10}). Western blots using anti-CarAc_{CA10} antibody indicated that both crystals contained CarAc_{CA10} protein, although the bands at about 30 kDa reacted slightly with anti-CarAc_{CA10} antibody (Fig. 3b).

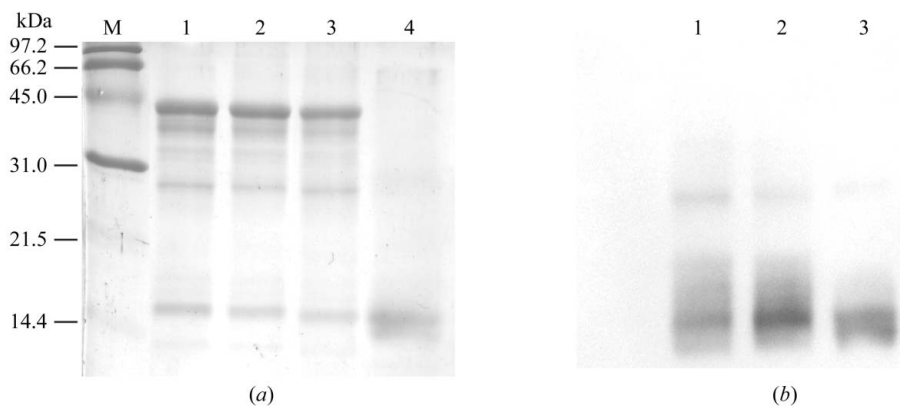


Figure 3

SDS-PAGE and Western blot analysis of the dissolved crystals. (a) SDS-PAGE of the dissolved crystals (5 µg each). Lane M, low-molecular-weight markers; lane 1, dissolved crystals of CARDO-O_{J3}; lane 2, dissolved type I crystals of the complex; lane 3, dissolved type II crystals of the complex; lane 4, dissolved crystals of CARDO-F_{CA10}. (b) Western blots using anti-CarAc_{CA10} antibody. Lane numbers are the same as in (a).

The merged data sets for the type I and II crystals were collected to 1.90 and 2.05 Å resolution, respectively. The data-collection and processing statistics are summarized in Table 1. The space groups of both crystals were determined to be $P2_1$, with similar unit-cell parameters. During purification using gel-filtration chromatography, the molecular weights of CARDO- O_{J3} and CARDO- F_{CA10} were calculated to be 132 and 13 kDa (data not shown), suggesting that CARDO- O_{J3} and CARDO- F_{CA10} have α_3 homotrimeric and monomeric configurations, respectively. These quaternary structures suggested by biochemical experiments are in accordance with the crystal structures of CARDO- O_{J3} and CARDO- F_{CA10} as single proteins (Nam *et al.*, 2005; Nojiri *et al.*, 2005). The trimeric structure of CARDO- O_{J3} has three completely separate active sites (catalytic non-haem irons), suggesting that three regions capable of interacting with CARDO- F_{CA10} exist around three Rieske [2Fe–2S] clusters in CARDO- O_{J3} . Therefore, assuming one molecule of CARDO- O_{J3} and three molecules of CARDO- F_{CA10} per asymmetric unit, the Matthews coefficient V_M (Matthews, 1968) is $2.63 \text{ \AA}^3 \text{ Da}^{-1}$ (corresponding to a solvent content of 53.2%). Consequently, the asymmetric unit in the crystal may contain one complex that is a trimer of a heterodimer consisting of CARDO- O_{J3} and CARDO- F_{CA10} molecules.

We are now attempting to determine the structure of this complex by the molecular-replacement method using the determined structures of CARDO- O_{J3} and CARDO- F_{CA10} and the refinement is now in progress. Until now, various types of electron transport between proteins have been reported, but there are only a few reports of their three-dimensional structures. Three structures of complexes of ferredoxins with their reductases have been determined: the complexes of *Anabaena* and maize ferredoxin with their respective ferredoxin:NADP⁺ oxidoreductase in oxygenic photosynthesis (Morales *et al.*, 2000; Kurisu *et al.*, 2001) and the complex of mitochondria adrenodoxin with adrenodoxin reductase in steroid biosynthesis (Müller *et al.*, 2001). To the best of our knowledge, the complex structure of the terminal oxygenase with its electron-donor protein has not been determined yet, except for the complex structure of the haem- and FMN-binding domains of the bacterial cytochrome P450BM-3, a prototype for the complex between eukaryotic microsomal P450 monooxygenase and P450 reductase (Sevrioukova *et al.*, 1999). Therefore, the complex structure of CARDO- O_{J3} with CARDO- F_{CA10} will represent the first structure of a biologically relevant complex between an oxygenase and its electron donor, ferredoxin.

Several reports have indicated that while the reductase components of ROSSs can be replaced by unrelated reductases, the ferredoxin components cannot be replaced (Subramanian *et al.*, 1981, 1985; Haiger & Gibson, 1990; Fukuda *et al.*, 1994). In the CARDO- $CA10$ system, CARDO- R_{CA10} can be replaced by an unrelated reductase, such as the ferredoxin reductase from spinach, but CARDO- F_{CA10} is indispensable for electron transfer to CARDO- O_{CA10} , suggesting that there is a specific interaction between CARDO- O_{CA10} and CARDO- F_{CA10} (Nam *et al.*, 2002). As no functional difference has been observed between CARDO- O_{J3} and CARDO- O_{CA10} in several experiments, the forthcoming structure of the CARDO- O_{J3} and CARDO- F_{CA10} complex will provide insight

into the structural basis of the specific interaction between the CARDO components. In addition, as the first structural example of an ROS electron-transport complex structure, it will also provide useful information for understanding the general architecture of the component interactions and electron transport in this important class of multicomponent oxygenase systems.

Synchrotron radiation was used for this work with the approval of the Photon Factory Advisory Committee and KEK (High Energy Accelerator Research Organization), Tsukuba (proposal No. 2003G124). This work was supported in part by PROBRAIN (Promotion of Basic Research Activities for Innovative Bioscience).

References

- Batie, C. J., Ballou, D. P. & Correll, C. C. (1991). *Chemistry and Biochemistry of Flavoenzymes*, Vol. 3, edited by F. Muller, pp. 543–556. Boca Raton, FL, USA: CRC Press.
- Bradford, M. M. (1976). *Anal. Biochem.* **72**, 248–254.
- Bressler, D. C. & Fedorak, P. M. (2000). *Can. J. Microbiol.* **46**, 397–409.
- Fukuda, M., Yasukouchi, Y., Kikuchi, Y., Nagata, Y., Kimbara, K., Horiuchi, H., Takagi, M. & Yano, K. (1994). *Biochem. Biophys. Res. Commun.* **202**, 850–856.
- Furukawa, K., Suenaga, H. & Goto, M. (2004). *J. Bacteriol.* **186**, 5189–5196.
- Habe, H., Chung, J.-S., Lee, J.-H., Kasuga, K., Yoshida, T., Nojiri, H. & Omori, T. (2001). *Appl. Environ. Microbiol.* **67**, 3610–3617.
- Habe, H. & Omori, T. (2003). *Biosci. Biotechnol. Biochem.* **67**, 225–243.
- Haiger, B. E. & Gibson, D. T. (1990). *J. Bacteriol.* **172**, 465–468.
- Inoue, K., Widada, J., Nakai, S., Endoh, T., Urata, M., Ashikawa, Y., Shintani, M., Saiki, Y., Yoshida, T., Habe, H., Omori, T. & Nojiri, H. (2004). *Biosci. Biotechnol. Biochem.* **68**, 1467–1480.
- Kurisu, G., Kusunoki, M., Katoh, E., Yamazaki, T., Teshima, K., Onda, Y., Kimata-Aruga, Y. & Hase, T. (2001). *Nature Struct. Biol.* **8**, 117–121.
- Matthews, B. W. (1968). *J. Mol. Biol.* **33**, 491–497.
- Morales, R., Charon, M.-H., Kachalova, G., Serre, L., Medina, M., Gómez-Moreno, C. & Frey, M. (2000). *EMBO Rep.* **1**, 271–276.
- Müller, J. J., Lapko, A., Bourenkov, G., Ruckpaul, K. & Heinemann, U. (2001). *J. Biol. Chem.* **276**, 2786–2789.
- Nam, J.-W., Noguchi, H., Fujimoto, Z., Mizuno, H., Ashikawa, Y., Abo, M., Fushinobu, S., Kobashi, K., Wakagi, T., Iwata, K., Yoshida, T., Habe, H., Yamane, H., Omori, T. & Nojiri, H. (2005). *Proteins*, **58**, 779–789.
- Nam, J.-W., Nojiri, H., Noguchi, H., Uchimura, H., Yoshida, T., Habe, H., Yamane, H. & Omori, T. (2002). *Appl. Environ. Microbiol.* **68**, 5882–5890.
- Nojiri, H., Ashikawa, Y., Noguchi, H., Nam, J.-W., Urata, M., Fujimoto, Z., Uchimura, H., Terada, T., Nakamura, S., Shimizu, K., Yoshida, T., Habe, H. & Omori, T. (2005). Submitted.
- Nojiri, H., Nam, J.-W., Kosaka, M., Morii, K., Takemura, T., Furihata, K., Yamane, H. & Omori, T. (1999). *J. Bacteriol.* **181**, 3105–3113.
- Nojiri, H. & Omori, T. (2002). *Biosci. Biotechnol. Biochem.* **66**, 2001–2016.
- Otwinowski, Z. & Minor, W. (1997). *Methods Enzymol.* **276**, 307–326.
- Rosche, B., Tshisuaka, B., Fetzner, S. & Lingens, F. (1995). *J. Biol. Chem.* **270**, 17836–17842.
- Sato, S., Nam, J.-W., Kasuga, K., Nojiri, H., Yamane, H. & Omori, T. (1997). *J. Bacteriol.* **179**, 4850–4858.
- Sevrioukova, I. F., Li, H., Zhang, H., Petron, J. A. & Poulos, T. L. (1999). *Proc. Natl Acad. Sci. USA*, **96**, 1863–1868.
- Subramanian, V., Liu, T.-N., Yeh, W.-K., Narro, M. & Gibson, D. T. (1981). *J. Biol. Chem.* **256**, 2723–2730.
- Subramanian, V., Liu, T.-N., Yeh, W.-K., Serdar, C. M., Wakett, L. P. & Gibson, D. T. (1985). *J. Biol. Chem.* **260**, 2355–2363.
- Takagi, T., Nojiri, H., Yoshida, T., Habe, H. & Omori, T. (2002). *Biotechnol. Lett.* **24**, 2099–2106.
- Wittich, R.-M. (1998). *Appl. Microbiol. Biotechnol.* **49**, 489–499.