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Structure of the Cytochrome *b***6***f* **Complex: Quinone Analogue Inhibitors as Ligands of Heme** *c***ⁿ**

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Abstract

A native structure of the cytochrome $b₆f$ complex with improved resolution was obtained from crystals of the complex grown in the presence of divalent cadmium. Two Cd^{2+} binding sites with different occupancy were determined: (i) a higher affinity site, Cd1, which bridges His143 of cytochrome *f* and the acidic residue, Glu75, of cyt b_6 ; in addition, Cd1 is coordinated by 1-2 H₂O or 1-2 Cl- ; (ii) a second site, Cd2, of lower affinity for which three identified ligands are Asp58 (subunit IV), Glu3 (PetG subunit) and Glu4 (PetM subunit).

Binding sites of quinone analogue inhibitors were sought in order to map the pathway of transfer of the lipophilic quinone across the $b₆f$ complex and to further define the function of the novel heme *c*n. Two sites were found for the chromone ring of the tridecyl-stigmatellin (TDS) quinone analogue inhibitor, one near the *p*-side [2Fe-2S] cluster. A second TDS site was unexpectedly found on the *n-side* of the complex facing the quinone exchange cavity as an axial ligand of heme c_n . A similar binding site proximal to heme *c*n was found for the *n*-side quinone analogue inhibitor, NQNO. Binding of these inhibitors required their addition to the complex before lipid that is used to facilitate crystallization.

The similar binding of NQNO and TDS as axial ligands to heme c_n implies that this heme utilizes plastoquinone as a natural ligand, thus defining an electron transfer complex consisting of hemes *b*n, *c*n, and PQ, and the pathway of *n*-side reduction of the PQ pool. The NQNO binding site provides an explanation for several effects associated with its inhibitory action: the large negative shift in midpoint redox potential of heme c_n the increased amplitude of light-induced reduction of heme b_n , and an altered EPR spectrum attributed to interaction between hemes c_n and b_n . A decreased extent of heme *c*n reduction by reduced ferredoxin in the presence of NQNO allows observation of the heme *c*n Soret band in a chemical difference spectrum.

Keywords

crystal structure; cytochrome b_6f ; cyclic electron transfer; heme c_n ; plastoquinone; quinone analogue inhibitors

Introduction

The hetero-oligomeric dimeric cytochrome $b₆f$ complex mediates electron transfer between the photosystem II and photosystem I reaction center complexes in oxygenic photosynthesis

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by oxidizing the relatively low potential $(E_{m7} \approx +0.1V)$ lipophilic plastoquinol and reducing soluble plastocyanin or cytochrome $c₆$, as discussed in several reviews with different perspectives 1-7. These electron transfer events are linked to proton translocation across the low dielectric membrane that generates a proton electrochemical potential gradient of approximately 250 mV, positive on the p- or lumen side of the membrane. A detailed understanding of charge transfer and movement of lipophilic quinone across the membrane in the $b₆f$ complex requires information derived from structure analysis. Three-dimensional structures of the b_f complex obtained from x-ray diffraction have been obtained at a resolution of 3.0 - 3.1 Å from the thermophilic filamentous cyanobacterium, *Mastigocladus laminosus* (pdb accession, 1VF5) 8, and the green alga, *Chlamydomonas reinhardtii* (pdb, 1Q90) 9 in the presence of the quinone analogue inhibitor, tri-decyl-stigmatellin. Structures of the native complex, and the complex with the inhibitor DBMIB (pdb, 2D2C) bound tightly at a site very distal to the [2Fe-2S] cluster, have also been obtained from *M. laminosus*8; 10. These structures show that the *b*₆*f* complex contains eight polypeptide subunits with 13 trans-membrane helices in each monomer of a functional dimer^{5-9; 11; 12}. Four of the eight subunits, petA, B, C, D, that contain or confine the redox prosthetic groups are "large" (16 – 31 kDa: cytochrome *f,* cytochrome b_6 , the Rieske iron-sulfur protein, and subunit IV; the four small (3.3 - 4.1 kDa) hydrophobic subunits, petG, L, M, and N form a "fence" at the outside periphery of each monomer, with each small subunit containing one trans-membrane helix. The *b₆f* complex isolated from plant (spinach) thylakoid membranes contains one additional subunit, FNR 13 ; ¹⁴ that is bound more weakly to the spinach complex, and is not present in the cyanobacterial *b*6*f* complex.

Given the similarity found between the core of the cytochrome $bc₁$ complex from the mitochondrial respiratory chain and photosynthetic bacteria and that of the b_6f complex 15 , structure information derived from the bc_1 complex¹⁶⁻²⁶ is relevant to an understanding of structure-function of the $b₆f$ complex.

Unique prosthetic groups; heme c_n . The b_6f complex from both the cyanobacterial and green algal sources shows the presence of three unusual or unique prosthetic groups, each present at a unit stoichiometry, chlorophyll a^{27-29} and β-carotene 29 , shown previously by biochemical analysis, whose presence would not have been expected *a priori* in the $b₆f$ complex that functions in the 'dark' reactions of oxygenic photosynthesis. In the structure studies, a unique covalently bound heme, heme c_n , was found on the electrochemically negative side of the complex. Heme *c*n has no amino acid side chain serving as an axial ligand, but only an axial H_2O that bridges the 4 Å distance between the Fe atom of heme c_n and a propionate oxygen of heme $b_n^{\{8\}}$. In retrospect, it was realized that heme c_n was previously defined spectrophotometrically and designated as component 'G.³⁰; 31. The position of heme c_n in close proximity to the previously well-defined heme b_n that has been proposed to function in a 'Q cycle' $32\dot{ }40$ and alternatively, or as well, in a ferredoxin-linked cyclic pathway $7\dot{ }9$, begs the question of the location of the plastoquinone that should provide the electron acceptor in these pathways.

Together with atomic structures of the PSI 41 and PSII $^{42-45}$ reaction center complexes, also obtained from a thermophilic cyanobacterium, a structure framework has been completed of the three integral membrane protein complexes that sustain linear electron transport in oxygenic photosynthesis. These structures have provided deeper insight into the pathways and mechanisms that govern redox energy transduction and, as well, trans-membrane movement of hydrophobic plastoquinone(ol) in the photosynthetic membrane.

The present study focuses on the binding site(s) of the quinone analogue inhibitors, TDS and NQNO, in the *M. laminosus b*₆*f* complex, which lead to the inference that plastoquinone is an axial ligand of heme *c*n, and the question of conformational changes transmitted across the

complex by NQNO. The question of whether trans-membrane conformational changes are transmitted across the $b₆f$ complex is of interest not only for a detailed understanding of the structure-function of cytochrome *bc* complexes, but also for the mechanism of trans-membrane transmission of signals arising from receptor activation by various stimuli ranging from light to hormones, which must involve conformational changes of trans-membrane α-helices 46 .

Results

1. The native b_6f **complex with cadmium**

Crystallization of the native b_6f complex in the presence of divalent Cd ions, Cd²⁺, resulted in an improvement of the native structure from 3.4 Å ($R_{cryst} = 0.256$; $R_{free} = 0.336$)⁹ to 3.00 Å $(R_{crvst} = 0.222$ and $R_{free} = 0.268$; Table I). These parameters were obtained from refinement using space group $P6₁22$ instead of $P6₁$, based on the assumption that the asymmetric unit contains the monomeric unit of the complex, and that the resulting structure of the $b₆f$ complex is described by a symmetric dimer.

In the native structure, obtained from crystallization in the presence of the divalent cation, Cd^{2+} , two Cd^{2+} binding sites are located in each monomer (Figs. 1A, B), a relatively high affinity (Cd1) and low (Cd2) site. Cd1, close to the p-side of the complex and the inter-monomer interface (Fig. 1A), is coordinated by His143 of cytochrome f and Glu75 of cytochrome b_6 (Fig. 1B^{),} and is also coordinated by 1-2 H₂O or 1-2 Cl⁻ (not shown). The second lower affinity $Cd²⁺$ binding site, located at the *p*-side aqueous interface of the complex near the small subunits of the complex and the external lipid medium, is coordinated by a carboxylate residue of subunit IV (Asp58) and carboxylate residues Glu3 and Glu4 of the small subunits. From effects of Cu^{2+} and Zn^{2+} on the EPR spectra of the Rieske [2Fe-2S] cluster in the spinach b_6f complex⁴⁷; 48, His143 was inferred to participate in one of three binding sites for divalent Cu^{2+} or Zn^{2+} ions.

The amplitudes of the anomalous scattering by the Cd sites and those of the other metal centers, in order of peak height relative to the background and scattering from other metal sites, are summarized in the legend of Fig. 1B.

The distances from the Cd1 and Cd2 sites are: Cd1 to Cd2, 23.2 Å; from Cd1 to heme b_p , 24.0 Å, and to the [2Fe-2S] cluster on the same and opposite side monomer, 38.9 Å and 40.1 Å (average distances to the 4 atoms of the cluster). From Cd2, the distances to the [2Fe-2S] cluster on the same and opposite side monomer are 57.1 and 28.0 Å. Thus, inhibitory effects of Cd^{2+} are not related to any direct effect on PQ binding to the [2Fe-2S] cluster. Furthermore, occlusion of a *p*-side H+ transfer pathway seems unlikely because the shortest distances from a Cd^{2+} binding site to the PEWY sequence that is implicated in *p*-side H⁺ transfer are 13, 12, and 16 Å to Glu78, Trp79, and Tyr80. Conformation changes that are allosteric in nature caused by Cu^{2+} binding to the $b₆f$ complex, which could be responsible for inhibition of electron transport, have been reported 47 ; 48 . The existence of the small changes in conformation, a change in orientation (ca. 5°) of the cytochrome *f* heme relative to the plane of the membrane, and a 7° rotation of the [2Fe-2S] cluster and the soluble domain of the iron-sulfur protein around the S-S axis of the cluster, calculated from the effects of Cu^{2+} and Zn^{2+} on the EPR spectra 48, cannot be determined at the present resolution of the structure of the $b₆f$ - Cd²⁺ complex.

2. *p***-side (near the Rieske cluster), and** *n-side* **(near heme** *c***n) difference maps in the absence of inhibitors**

Fo – Fc maps of the native structure (with Cd^{2+}), in the absence of quinone analogue inhibitors, show no significant density immediately adjacent to the *p*-side 2Fe-2S cluster (Fig. 2A) and the *n*-side hemes b_n or c_n (Fig. 2B). On the *p*-side, additional electron density seen in the Fo –

Fc map under the cd2 loop connecting the C and D TM helices (Fig. 2A) presumably arises from lipid head group(s) or bound detergent (see **Discussion**). The small patches of background electron density seen in the Fo-Fc difference map on the *n*-side of heme b_n (Fig. 2B) and on the *n*-side of β-carotene near the E helix, have not been identified and are ascribed to bound lipid, detergent, or PEG.

3. *p***- side binding site of TDS**

The site of inhibition by stigmatellin on the p -side of the $bc₁$ complex, and the site of inhibition by tri-decyl stigmatellin in the b_6f complex were defined in structure studies of the bc_1 ^{17;} 20; 23-25 and $b_6 f^9$ complex. At his site the chromone ring of stigmatellin or TDS is seen to be close enough to the imidazole of a histidine ligand of the [2Fe-2S] cluster to form an H-bond. This bond would provide the pathway for H^+ transfer from the bound quinol, PQH_2 , to the *p*side aqueous phase.

In the original study on the structure of the $b₆f$ complex in *M. laminosus*, it was concluded that the TDS was oriented "ring-out" based on electron density seen outside the entrance portal that connects the quinone exchange cavity and the *p*-side inhibitor binding niche 8. Now that a better native structure is available, it can be seen that this density under the cd2 loop and outside the entrance portal from the inter-monomer cavity to the p-side quinone binding niche is present in both the native (Fig. 2A) and TDS (Fig. 3A) structures. The R-factors of the TDS structure are $R_{crvst} = 0.201$ and $R_{free} = 0.258$ (Table I). The previous inference $\frac{8}{3}$ about the 'ring-out' TDS structure was a consequence of the fact that addition of the lipid necessary for crystallization 49 prior to TDS prevented binding of TDS. The presence of electron density at this position in the absence of TDS was not detected because of a native structure of sufficient resolution was not available. When TDS was added before augmentation with lipid, electron density attributable to the double ring of the TDS in a "ring-in" conformation is seen to be sufficiently close to form an H-bond between O4 of the stigmatellin chromone ring and the imidazole of the His129 ligand to the [2Fe-2S] cluster (Fig. 3A). The additional electron density outside the entrance portal (Figs. 2A, 3A) $\frac{8}{3}$ is now attributed to unidentified lipid head-group (s) and/or bound detergent.

4. *n***- side binding site of TDS**

TDS is considered to be a well-defined *p*-side inhibitor in cytochrome bc_1 and b_6 *f* complexes. The appearance of electron density arising from TDS in the Fo – Fc difference map on the *n*side of the complex near heme c_n (Fig. 3B) was not expected. No such site of inhibition by stigmatellin in the bc_1 complex has been inferred from spectroscopic/electron transfer studies, nor seen in the *bc*1 crystal structures. Electron density tentatively ascribed to plastoquinone in a figure of the environment around heme c_n in *C. reinhardtii* 9 may arise from TDS, which was present during the crystallization of the complex. An *n*-side binding site for stigmatellin in bc_1 and b_6f complexes was suggested by the perturbation by stigmatellin cytochrome *b* difference spectra for hemes b_p and b_n and in both complexes⁵⁰.

5. An NQNO heme *c***n ligand binding site**

The effect of the quinone analogue NQNO on the light-induced reduction of heme b_n in the $b₆f$ complex resembles the effect of antimycin A, a potent inhibitor of the cytochrome $bc₁$ complex 25 , on heme b_n in the bc_1 complex. However, antimycin does not affect the lightinduced reduction of heme b_n in the b_6f complex⁵¹, either because antimycin binds more weakly to heme c_n or that heme c_n occludes the projected antimycin binding site in the b_6f complex (not shown). The binding site of NQNO was sought through co-crystallization with the complex with R-factors, $R_{cryst} = 0.224$ and $R_{free} = 0.273$ (Table I). The $F_0 - F_c$ difference map, which shows the [2Fe-2S] cluster and the *p*-side portal in the "roof" of the cavity, did not show any additional density on the *p*-side (Fig. 4A) in addition to that seen in the absence of

added inhibitor (Fig. 2A). The *p*-side plastoquinone binding niche that includes the [2Fe-2S] cluster and the *p*-side portal in the "roof" of the cavity is shown. On the *n*-side, the difference map revealed a density consistent with NQNO serving as an axial ligand of heme *c*n, distal to the water ligand and heme b_n (Fig. 4B). No electron density was detected at this position in the native $F_0 - F_c$ difference map (Fig. 2B). The binding of NQNO as an apparent ligand to heme *c*n, together with the ability of TDS to bind similarly, reinforces the inference that the physiological ligand of heme c_n is plastoquinone.

6. Redox consequences of interaction of NQNO with heme *c***n; spectrophotometric detection of** *c***n reduction by reduced ferredoxin**

Detection of heme c_n in optical difference spectra of the b_6f complex is difficult because its α-band is flat with a small extinction coefficient and the Soret band is masked by the absorbance bands of three hemes, f , b_n , and b_p , and also chlorophyll *a*. The presence of FNR in the b_6f complex prepared from spinach 14 ; 52 and the ability of NADPH-reduced ferredoxin to reduce cytochrome *b* in the complex suggested that the complex might participate in the pathway of cyclic electron transport linked to reduction of ferredoxin by photosystem $I¹⁴$. However, at the time of the FNR studies, the existence of heme c_n was not known, and the possibility that it might also be reduced by ferredoxin was not realized. In the presence of pre-reduced cytochrome f , the amplitude of the Soret band of heme b_n caused by addition of NADPH as reductant, is seen one minute after addition to be somewhat less than half that caused by dithionite (Fig. 5A). This reduction of heme b_n requires both FNR, which is present in the spinach but not the *M. laminosus* complex, and ferredoxin (Fig. 5B). Ferredoxin-dependent reduction of cytochrome b_6 ⁵³, inferred to be heme b_n ⁵⁴, has previously been observed.

In the presence of NQNO, the Soret spectrum is shifted by 1 nm to longer wavelengths. A double difference spectrum, $b₆f$ complex in the absence minus the presence of NQNO, shows a difference band for reduced heme with a peak at 428 nm (Fig. 5C), presumably belonging to heme c_n ^{55; 56}. A similar result was obtained with butyl-isocyanide (data not shown), previously shown to interact with heme c_n ⁵⁶. It is inferred that the plus/minus NQNO difference spectrum of the Soret band, having a peak at 428 nm diagnostic for heme c_n , is a consequence of the interaction of heme *c*n with NQNO, which results in the large (ca. - 225 mV) decrease in the heme c_n midpoint potential⁵⁵. It should be noted that the \pm NQNO difference spectrum in the α-band region is featureless.

The changes caused by NQNO can be understood by assuming an electron transport pathway from reduced ferredoxin, which is the physiological acceptor of reducing equivalents provided by photosystem I (Fig. 5D), via FNR to hemes b_n and c_p ordered in terms of their midpoint redox potentials, $E_{m7} \cong -50$ mV 54 ; 57 and $+ 75$ mV 55 , respectively. The increase in the amplitude of flash-induced reduction of heme b_n in the presence of NQNO 54 ; 58 ; 59 is inferred to be a consequence of the NQNO-induced decrease in E_{m7} to \sim -150 mV of heme c_n , and a resulting unfavorable equilibrium and large energy barrier for electron transfer from *b*n to *c*ⁿ (Fig. 5D). The complex formed by hemes *b*n, *c*n, and the inferred bound PQ ligand of heme *c*n, that would be a donor to the PQ pool (larger font), is shown (Fig. 5D).

Discussion

1. *p***-side cadmium binding sites**

As previously found for the effect of Cu^{2+} on the electron transfer rate in the b_6f complex⁴⁷, addition of Cu²⁺ or Cd²⁺ at a concentration of 10 μ M also caused substantial inhibition, although in the present experiments the extent of this inhibition was only about 50 % of the uninhibited rate (data not shown). In the bovine mitochondrial cytochrome bc_1 complex, Zn^{2+} was found to bind more tightly than any other divalent cation, and was proposed to bind

to a protonatable group, perhaps a histidine, in the proton channel associated with the p-side quinol oxidation ⁶⁰. In the *Rb. sphaeroides* photosynthetic bacterial bc_1 complex, Zn^2 ⁺ inhibited the electrochromic band shift, oxidation of heme b_n re-reduction after an oxidizing light flash of cytochrome c_1 , and proton release into the chromatophore lumen ⁶¹. In the bacterial photosynthetic reaction center, a Cd^{2+} binding site bridging an Asp and two His residues, a coordination very similar to the major Cd1 binding site of Cd^{2+} in the *b₆f* complex (Fig. 1), is associated with inhibition of the reduction and protonation of the Q_B quinone 62 . In the case of the two Cd^{2+} sites on the *p*-side of the *b₆f* complex, it is not known at present whether the comparatively small $~50$ % inhibition of electron transport caused by added Cd^{2+} could be a consequence of the presence of background divalent ions in the medium. Regarding a possible effect of divalent ions on charge transfer, the smallest distance between a Cd²⁺ binding site and residues known to be involved in H⁺ or electron transfer is 12-13 Å between Cd^{2+} and glutamate-78 or tryptophan-79 of the "PEWY" segment implicated in the transfer pathway of the second H^+ to the *p*-side aqueous phase 63 .

2. *p***-side "ring-in" TDS binding site**

The p -side $F_o - F_c$ difference map for the quinone analogue inhibitor TDS shows electron density in the quinone exchange cavity outside the 11 $\AA \times 12$ Å portal entry to the iron-sulfur cluster binding niche, similar to the density that has previously been reported and inferred to be associated with a "ring-out" conformation of the TDS $\frac{8}{3}$. The comparison with a native structure having much better resolution (Fig. 1[;] Table I) indicated that this density is intrinsic to the structure, presumably associated with lipid head-group(s) and/or detergent, in the roof of the exchange cavity. The addition of lipid, which is necessary to stabilize the b_6f complex for crystallization 49, prior to addition of TDS, can apparently block the binding of the quinone analogue inhibitors. In the present study, TDS was added before the addition of lipid. This resulted in a prominent electron density that can be associated with the TDS chromone ring close to the His129 imidazole ligand of the [2Fe-2S] cluster (Fig. 3A). A study of the properties of the quinone passage portal indicated that residue 81 is critical for the efficiency of inhibition of TDS or stigmatellin 64. The efficiency of inhibition of the reduction of cytochrome *f* after oxidation by a light flash in the *Synechococcus* sp. PCC 7002 was increased by a factor of 10-100 when Leu81 at the entrance to the portal was changed by site-directed mutagenesis to Phe. An explanation alternative to the original inference that the Leu81Phe mutation allowed the more efficient "ring-in" conformation to replace the 'ring-out' orientation ⁶⁴ is that the entrance portal either has a smaller size or smaller degree of flexibility in the cyanobacterial $b₆f$ complex, in which inhibition by TDS and stigmatellin is substantially less effective.

3. An *n***-side TDS binding site**

It was unexpected to find that TDS also has a prominent binding site on the *n*-side of the complex, where it is in a position to be an axial ligand of heme c_n (Fig. 3B). The possibility of an *n*-side binding site near heme b_n was indicated in a previous study on both b_6f and bc_1 complexes 50. The existence of both *n*- and *p*-side binding sites for TDS raises the question as whether it should be considered that many of the quinone analogue inhibitors, presently described as *n*- or *p*-side in their action, may in fact be able to bind on both sides of the complex, although with different affinities. With the commonly used flash spectroscopic probes of inhibitor action via the rate and amplitude of cytochrome $f/c₁$ turnover, it would be difficult to discern an *n*-side mode of action of an inhibitor with significant *p*-side activity.

4. The *n***-side NQNO binding site**

NQNO axial ligation of heme c_n , and the resulting large (ca. 225 mV) negative shift in midpoint potential of heme c_n ⁵⁵, as summarized in the pathway shown in Fig. 5D, provide an explanation for the decreased rate of heme b_n oxidation and resulting increased amplitude of

light-induced reduction of heme b_n ^{54; 59; 65}, and the decreased extent of heme c_n reduction by reduced ferredoxin which allows detection of the heme *c*n Soret band in a chemical difference spectrum (Fig. 5C). The low field EPR spectrum of the $b₆f$ complex, attributed to spin interaction between the closely spaced hemes c_n and b_n , is altered by the presence of NQNO ⁶⁶. The interaction between hemes b_n and c_n suggests that the two hemes may be able to act as a two electron donor to PQ 66 (Fig. 5D). A two electron reduction of PQ, which would avoid formation of plasto-semiquinone that could reduce O_2 to superoxide and thus form reactive oxygen species, could be a crucial protection reaction in an oxygen-rich membrane.

5. PQ as the physiological ligand of heme *c***n; pathways of linear and cyclic electron transfer**

The binding of the two quinone analogue inhibitors, TDS and NQNO, as axial ligands of heme c_n implies that plastoquinone positioned at the interface between heme c_n and the large intermonomer quinone exchange cavity is the physiological ligand of heme c_n (Fig. 5D). Thus, the *n*-side pathway to PQ, which serves as the trans-membrane proton and electron carrier would be the same, whether electrons are provided to heme b_n by a Q cycle connected to the linear or non-cyclic electron transport chain, or from ferredoxin-FNR as part of a cyclic pathway connected to photosystem I (Fig. 5D).

O2 as a potential ligand of heme cn—By analogy with hemoglobin, myoglobin, cytochrome oxidase, etc., the apparently free axial ligand position on the cavity side of heme c_n would seem to be a potential ligand binding site for molecular O_2 . However, such a molecular O_2 cannot be seen and at best could be only weakly bound, because a water or OH⁻ molecule on the other side of heme *c*n can be seen to connect its Fe atom with a propionate O atom of heme b_n .

6. Role of the *b***6***f* **complex in the photosystem I cyclic pathway**

A function of the $b₆f$ complex in the PSI cyclic pathway has been discussed extensively. The absence of an effect of antimycin A on the flash-induced reduction of heme b_n led to the conclusion that the $b₆f$ complex is not involved in the cyclic pathway, but that there is an alternate ferredoxin:quinone oxidoreductase (FQR) 51 . However, it has not been possible to isolate this hypothetical oxido-reductase. The overlapping of heme c_n with the projection of the antimycin site to the $b₆f$ complex from its binding site in the $bc₁$ complex might suggest that antimycin A would not be a good inhibitor of a cyclic pathway that passes through hemes *b*n- *c*n. However, inhibition of a ferredoxin-mediated PSI-linked cyclic pathway by antimycin is well documented 51 ; $67-69$. On the basis of the data for TDS (Fig. 3B) and NQNO (Fig. 4B) binding as a ligand to heme c_n , the absence of an antimycin-induced increase in amplitude of flash-induced cytochrome $b₆$ reduction 51 could be explained by weaker antimycin binding with marginal inhibitory effects.

Following the description of FNR in the b_6f complex ¹⁴, and the re-discovery of the unique heme c_n in the structures of the b_6f complex $\frac{8}{9}$, there have been further studies on the cyclic pathway in higher plants and the role of the $b₆f$ complex in re-entry of electrons and protons into the membrane 12 ; 70; 71. Recent data describe a high efficiency of photosystem I-linked cyclic electron transport that implies a separate domain, i.e., that of the non-appressed thylakoid membranes for the cyclic pathway $70; 71$. It had been proposed that the $b₆f$ complex responsible for cyclic electron transport is structurally separated in the membrane, and is located in the chloroplast non-appressed membrane domain 70 , where it is distinguished by the presence of bound FNR, which would not be present in the appressed membrane fraction of the $b₆f$ complex 70 . More recently, the mechanism of regulation of cyclic vs. linear flow has been inferred to be regulated by the redox state of FNR in the stromal compartment 12 , a model that bears some resemblance to an earlier proposal 67 . A role of heme c_n in the pathway of PSI-linked cyclic electron transport is an obvious general hypothesis.

7. Side specificity of binding of Q analogue inhibitors; *n***- to** *p***-side conformational changes**

The finding that there is a well-defined binding site of TDS on the *n*-side of the $b₆f$ complex was unexpected. The TDS binding site on the *p*-side close to a His ligand of the [2Fe-2S] cluster, is documented in both bc_1^2 (25); 25 and b_6 *f*⁹ complexes. Although NQNO is regarded as an *n*-side inhibitor, an NQNO binding site on both sides of the bc_1 complex has been defined in crystal structures of the bovine bc_1 complex²². The existence of a TDS site on both sides of the complex documented above (Figs. $3A \cdot 4A$) raises the question of the side specificity of binding of these inhibitors, which is relevant to the question of conformational changes transmitted across the complex by inhibitor binding. In the *Rb. capsulatus* bacterial *bc*¹ complex, it has been concluded that binding of the *n*-side inhibitors, HQNO and antimycin A, influences the interactions and environment of the ubiquinol donor to the [2Fe-2S] cluster on the *p*-side, implying the existence of inhibitor-induced *n* to *p*-side conformational changes that span the complex⁷². *n* to *p*-side conformational changes involving antimycin A binding have also been proposed in the yeast bc_1 complex 73 . The importance of such trans-complex and trans-membrane conformational changes extends beyond the cytochrome *bc* complexes, as it relates to the mechanism of trans-membrane signal transduction ⁴⁶.

In the case of the $b₆f$ complex, it has been found that NQNO inhibits the rate of cytochrome *f* reduction at the smallest concentrations that cause an increased amplitude of light-induced heme b_n reduction (59; S. Heimann, unpublished data). This implies: (i) NQNO exerts a structure change on the *p*-side from its binding site on the *n-side*; or (ii) there is a second NQNO binding site on the *p*-side that is not detected at the present resolution (3.55 Å) of the $b₆f$ structure with NQNO. At the present level of resolution (Table I), the electron density data show no *p*-side density attributable to NQNO, nor any *p*-side structure change.

Materials and Methods

Purification of the cytochrome b6f complex

The cytochrome $b₆f$ complex from spinach thylakoid membranes and the thermophilic cyanobacterium, *Mastigocladus laminosus*, was purified as described previously 74.

Crystallization of the cytochrome b₆^f complex

Crystals of the cytochrome $b₆f$ complex were obtained by hanging drop vapor diffusion at 4° C or 20 \degree C. For crystallization, the *b*₆*f* complex was concentrated to 20 mg/ml in a solution containing 50 mM Tris-HCl (pH 7.5), 50 mM NaCl, 1 mM EDTA, 0.05% β-D-undecylmaltoside, and 0.01% (w/v) dioleolyl-phosphatidylcholine $\frac{49; 74}{ }$. 2 µl of protein solution was mixed with an equal volume of reservoir solution containing 100 mM Tris-HCl, (pH 8.5), 200 mM MgCl₂[,] 40 mM CdCl₂ and 15-18% PEG 550. Crystals were grown for 5-7 days at 4^oC or for 2-3 days at 20°C. Co-crystallization with the quinone analogue inhibitors, NQNO and TDS, were carried out, respectively, in the presence of 125 μM NQNO diluted 80-fold from a 10 mM stock solution in ethanol, and 100 μM TDS diluted 200-fold from a 20 mM stock solution in ethanol, with a ratio in each case of inhibitor: protein, mol: mol \cong 5. Additions of inhibitors were necessarily made before the augmentation with DOPC lipid, as addition of lipid before inhibitor seemed to prevent or decrease binding of inhibitors.

X-ray data collection and structure refinement

Native, NQNO and TDS data sets were collected under cryo conditions (100° K) using beamline SBC 19-ID at the Advanced Photon Source (APS), Argonne, IL. Data reduction and scaling were processed using HKL2000 75 . The structure of the native crystal was solved by molecular replacement using MOLREP from the 'CCP4' suite 76 and the $b₆f$ monomer (pdb) accession number, 1VF5) as the starting model. The structure refinements for native, NQNO

and TDS crystals were carried out with REFMAC5 from the CCP4 suite. Structure corrections and model building were carried out using the program 'O' 77 . The data collection and refinement statistics are summarized in Table I. The programs 'O,' 'Molscript' 78, POVScript 79 and raster3D 80 were used for presentation of molecular graphics.

Reduction of cytochrome b6/cⁿ

Chemical and enzymatic reduction of cytochrome $b₆$ were measured on a Cary 4000 UV-VIS spectrophotometer using a half-bandwidth for the measuring light of 1 nm, a scan rate of 150 nm/min, sampling interval, 0.5 nm, and averaging time, 0.2 sec. The measurements were carried out in a 3 ml anaerobic cuvette containing 2.5 ml TNE buffer (30 mM Tris-HCl, pH 7.5, 50 mM NaCl, 1 mM EDTA, 0.05% UDM), and 1.5 μ M spinach *b₆f* complex containing intrinsically bound FNR 14 or *M. laminosus* complex not containing FNR, together with 10 mM glucose, glucose oxidase (type X-S from *A. nidulans*, 160 units), 3 μM ferredoxin, 50 μM decyl-plastoquinol, and a quinol analogue inhibitor in some experiments. The mixture was magnetically stirred and argon passed over the sample for 1 h, after which a baseline spectrum was measured. Reduction of cytochrome $b₆$ was initiated by addition of NADPH to a final concentration of 0.3 mM. The difference spectrum of the cytochrome complex reduced by NADPH, relative to that in which cytochrome *f* alone was reduced by decyl-plastoquinol, was measured as a function of time after addition of NADPH.

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References

- 1. Kallas, T. The cytochrome $b₆f$ complex. In: Bryant, DA., editor. The Molecular Biology of Cyanobacteria. Kluwer Academic Publishers; Dordrecht: 1994. p. 259-317.
- 2. O'Keefe DP. Structure and function of the chloroplast cytochrome *bf* complex. Photosynth Res 1988;17:189–216.
- 3. Allen JF. Cytochrome b6f: structure for signaling and vectorial metabolism. Trends in Plant Science 2004;9:130–137. [PubMed: 15003236]
- 4. Kurisu G, Zhang H, Smith JL, Cramer WA. Structure and function of the cytochrome $b₆f$ complex of oxygenic photosynthesis. Proteins, Nucleic Acids, and Enzymes 2004;49:1265–1273.
- 5. Smith JL, Zhang H, Yan J, Kurisu G, Cramer WA. Cytochrome bc complexes: a common core of structure and function surrounded by diversity in the outlying provinces. Curr Opin Struct Biol 2004;14:432–9. [PubMed: 15313237]
- 6. Cramer WA, Zhang H, Yan J, Kurisu G, Smith JL. Evolution of photosynthesis: time-independent structure of the cytochrome *b*₆*f* complex. Biochemistry 2004;43:5921–9. [PubMed: 15147175]
- 7. Cramer WA, Zhang H, Yan J, Kurisu G, Smith JL. Trans-membrane traffic in the cytochrome $b₆f$ complex. Annu Rev Biochem 2006;75:769–790. [PubMed: 16756511]
- 8. Kurisu G, Zhang H, Smith JL, Cramer WA. Structure of the cytochrome *b*6*f* complex of oxygenic photosynthesis: tuning the cavity. Science 2003;302:1009–1014. [PubMed: 14526088]
- 9. Stroebel D, Choquet Y, Popot J-L, Picot D. An atypical heam in the cytochrome *b*6*f* complex. Nature 2003;426:413–418. [PubMed: 14647374]
- 10. Yan J, Kurisu G, Cramer WA. Structure of the cytochrome *b*6*f* complex: Binding site and intraprotein transfer of the quinone analogue inhibitor 2,5-dibromo-3-methyl-6-isopropyl-*p*-benzoquinone. Proc Nat Acad Sci USA 2006;103:67–74.

- 11. Finazzi G, Forti G. Metabolic flexibility of the green alga Chlamydomonas reinhardtii as revealed by the link between state transitions and cyclic electron flow. Photosynth Res 2004;82:327–338. [PubMed: 16143844]
- 12. Joliot P, Joliot A. Cyclic electron flow in C3 plants. Biochim Biophys Acta 2006;1757:362–368. [PubMed: 16762315]
- 13. Whitelegge JP, Zhang H, Taylor R, Cramer WA. Full subunit coverage liquid chromatography electrospray-ionization mass spectrometry (LCMS+) of an oligomeric membrane protein complex: the cytochrome b_6f complex from spinach and the cyanobacterium, *M. laminosus*. Molecular Cellular Proteomics 2002;1:816–827. [PubMed: 12438564]
- 14. Zhang H, Whitelegge JP, Cramer WA. Ferredoxin:NADP+ oxidoreductase is a subunit of the chloroplast cytochrome *b6f* complex. J Biol Chem 2001;276:38159–38165. [PubMed: 11483610]
- 15. Widger WR, Cramer WA, Herrmann RG, Trebst A. Sequence homology and structural similarity between the *b* cytochrome of mitochondrial complex III and the chloroplast $b₆f$ complex: position of the cytochrome *b* hemes in the membrane. Proc Nat Acad Sci USA 1984;81:674–678. [PubMed: 6322162]
- 16. Xia D, Yu C-A, Kim H, Xia J-Z, Kachurin AM, Yu L, Deisenhofer J. Crystal structure of the cytochrome *bc*1 complex from bovine heart mitochondria. Science 1997;277:60–66. [PubMed: 9204897]
- 17. Zhang Z, Huang L, Shulmeister VM, Chi YI, Kim KK, Hung LW, Crofts AR, Berry EA, Kim SH. Electron transfer by domain movement in cytochrome *bc*1. Nature 1998;392:677–84. [PubMed: 9565029]
- 18. Iwata S, Lee JW, Okada K, Lee JK, Iwata M, Rasmussen B, Link TA, Ramaswamy S, Jap BK. Complete structure of the 11-subunit mitochondrial cytochrome *bc*1 complex. Science 1998;281:64– 71. [PubMed: 9651245]
- 19. Kim H, Xia D, Yu CA, Xia JZ, Kachurin AM, Zhang L, Yu L, Deisenhofer J. Inhibitor binding changes domain mobility in the iron-sulfur protein of the mitochondrial bc_1 complex from bovine heart. Proc Nat Acad Sci USA 1998;a 95:8026–33. [PubMed: 9653134]
- 20. Hunte C, Koepke J, Lange C, Rosmanith T, Michel H. Structure at 2.3 Å resolution of the cytochrome *bc*1 complex from the yeast *Saccharomyces cerevisiae* with an antibody Fv fragment. Structure Fold Des 2000;8:669–84. [PubMed: 10873857]
- 21. Palsdottir H, Lojero CG, Trumpower BL, Hunte C. Structure of the yeast cytochrome bc_1 complex with a hydroxyquinone anion Qo site inhibitor bound. J Biol Chem 2003;278:31303–11. [PubMed: 12782631]
- 22. Gao X, Wen X, Esser L, Quinn B, Yu L, Yu C, Xia D. Structural basis for the quinone reduction in the *bc*1 complex: A comparative analysis of crystal structure of mitochondrial cytochrome *bc*1 with bound substrate and inhibitors at the Qi site. Biochemistry 2003;42:9067–9080. [PubMed: 12885240]
- 23. Esser L, Quinn B, Li YF, Zhang M, Elberry M, Yu L, Yu CA, Xia D. Crystallographic studies of quinol oxidation site inhibitors: a modified classification of inhibitors for the cytochrome *bc*1 complex. J Mol Biol 2004;341:281–302. [PubMed: 15312779]
- 24. Berry EA, Huang L-S, Saechao L-K, Pon NG, Valkova-Valchanova M, Daldal F. X-ray structure of Rhodobacter capsulatus cytochrome *bc*1: comparison with its mitochondrial and chloroplast counterparts. Photosynth Res 2004;81:251–275. [PubMed: 16034531]
- 25. Huang LS, Cobessi D, Tung EY, Berry EA. Binding of the respiratory chain inhibitor antimycin to the mitochondrial bc_1 complex: a new crystal structure reveals an altered intramolecular hydrogenbonding pattern. J Mol Biol 2005;351:573–97. [PubMed: 16024040]
- 26. Esser L, Gong X, Yang S, Yu L, Yu CA, Xia D. Surface-modulated motion switch: capture and release of iron-sulfur protein in the cytochrome bc1 complex. Proc Natl Acad Sci U S A 2006;103:13045– 50. [PubMed: 16924113]
- 27. Huang D, Everly RM, Cheng RH, Heymann JB, Schägger H, Sled V, Ohnishi T, Baker TS, Cramer WA. Characterization of the chloroplast cytochrome $b₆f$ complex as a structural and functional dimer. Biochemistry 1994;33:4401–4409. [PubMed: 8155658]
- 28. Pierre Y, Breyton C, Lemoine Y, Robert B, Vernotte C, Popot J-L. On the presence and role of a molecule of chlorophyll a in the cytochrome *b6f* complex. J Biol Chem 1997;272:21901–21908. [PubMed: 9268323]

- 29. Zhang H, Huang D, Cramer WA. Stoichiometrically bound beta-carotene in the cytochrome $b₆f$ complex of oxygenic photosynthesis protects against oxygen damage. J Biol Chem 1999;274:1581– 7. [PubMed: 9880536]
- 30. Joliot P, Joliot A. The low-potential electron-transfer chain in the cytochrome *b/f* complex. Biochim Biophys Acta 1988;933:319–333.
- 31. Lavergne J. Membrane potential-dependent reduction of cytochrome *b*-6 in an algal mutant lacking photosystem I centers. Biochim Biophys Acta 1983;725:25–33.
- 32. Crofts, AR. The mechanism of the ubiquinol:cytochrome *c* oxidoreductases of mitochondria and of *Rhodopseudomonas sphaeroides*. In: Martonosi, AN., editor. The Enzymes of Biological Membranes. 4. Plenum Press; New York: 1985. p. 347-382.
- 33. Crofts AR. Proton-coupled electron transfer at the Q_0 -site of the bc_1 complex controls the rate of ubihydroquinone oxidation. Biochim Biophys Acta 2004;1655:77–92. [PubMed: 15100020]
- 34. Graan T, Ort DR. Initial events in the regulation of electron transfer in chloroplasts. J Biol Chem 1983;258:2831–2836. [PubMed: 6826542]
- 35. Hope AB. The chloroplast cytochrome *bf* complex: A critical focus on function. Biochimica et Biophysica Acta 1993;1143:1–22. [PubMed: 8388722]
- 36. Mitchell P. The protonmotive Q cycle: A general formulation. FEBS Letters 1975;59:137–139. [PubMed: 1227927]
- 37. Joliot P, Joliot A. Mechanism of proton-pumping in the cytochrome *b/f* complex. Photosynthesis Research 1985;9:113–124.
- 38. Rich PR. Mechanisms of quinol oxidation in photosynthesis. Photosynthesis Research 1985;6:335– 348.
- 39. Joliot P, Joliot A. Electron transfer between photosystem II and the cytochrome *b/f* : mechanistic and structural implications. Biochim Biophys Acta 1992;1102:53–61.
- 40. Sacksteder CA, Kanazawa A, Jacoby ME, Kramer DM. The proton to electron stoichiometry of steady-state photosynthesis in living plants: A proton-pumping Q cycle is continuously engaged. Proc Natl Acad Sci U S A 2000;97:14283–8. [PubMed: 11121034]
- 41. Jordan P, Fromme P, Witt H, Klukas O, Saenger W, Krauss N. Three-dimensional structure of cyanobacterial photosystem I at 2.5 A resolution. Nature 2001;411:909–917. [PubMed: 11418848]
- 42. Zouni A, Witt H, Kern J, Fromme P, Krauss N, Saenger W, Orth P. Crystal structure of photosystem II from Synechococcus elongatus at 3.8 A resolution. Nature 2001;409:739–743. [PubMed: 11217865]
- 43. Ferreira KN, Iverson TM, Maghlaoui K, Barber J, Iwata S. Architecture of the photosynthetic oxygenevolving center. Science 2004;303:1831–1838. [PubMed: 14764885]
- 44. Kern J, Loll B, Zouni A, Saenger W, Irrgang KD, Biesiadka J. Cyanobacterial Photosystem II at 3.2 A resolution - the plastoquinone binding pockets. Photosynth Res 2005;84:153–9. [PubMed: 16049768]
- 45. Yano J, Kern J, Irrgang K-D, Latimer MJ, Bergmann U, Glatzel P, Pushkar Y, Biesiadka J, Loll B, Sauer K, Messinger J, Zouni A, Yachandra VK. X-ray damage to the Mn₄Ca complex in single crystals of photosystem II: A case study for metalloprotein crystallography. Proc Nat Acad Sci USA 2005;102:12047–12052. [PubMed: 16103362]
- 46. Bourne HR. How receptors talk to trimeric G proteins. Curr Opin Cell Biol 1997;9:134–142. [PubMed: 9069253]
- 47. Rao S, Tyryshkin AM, Roberts AG, Bowman MK, Kramer DM. Inhibitory Copper Binding Site on the Spinach Cytochrome $b₆f$ Complex: Implication for Qo Site Catalysis. Biochemistry 2000;39:3285–3296. [PubMed: 10727220]
- 48. Roberts AG, Bowman MK, Kramer DM. Certain metal irons are inhibitors of cytochrome *b*₆*f* complex 'Rieske' iron-sulfur protein domain movements. Biochemistry 2002;41:4070–4079. [PubMed: 11900550]
- 49. Zhang H, Kurisu G, Smith JL, Cramer WA. A defined protein-detergent-lipid complex for crystallization of integral membrane proteins: The cytochrome $b₆f$ complex of oxygenic photosynthesis. Proc Natl Acad Sci USA 2003;100:5160–5163. [PubMed: 12702760]
- 50. Hauska G, Herald E, Huber C, Nitschke W, Sofrova D. Stigmatellin affects both hemes of cytochrome b in cytochrome b_6f /bc₁ complexes. Z Naturforsch 1989;44c:462–467.

- 51. Moss DA, Bendall DS. Cyclic electron transport in chloroplasts. the Q-cycle and the site of action of antimycin. Biochim Biophys Acta 1984;767:389–395.
- 52. Clark RD, Hawkesford MJ, Coughlan SJ, Hind G. Association of ferredoxin-NADP⁺ oxidoreductase with the chloroplast cytochrome *b-f* complex. FEBS Lett 1984;174:137–142.
- 53. Lam E, Malkin R. Ferredoxin-mediated reduction of cytochrome *b*-563 in a chloroplast cytochrome *b*-563/*f* complex. FEBS Lett 1982;141:98–101.
- 54. Furbacher PN, Girvin ME, Cramer WA. On the question of interheme electron transfer in the chloroplast cytochrome b_6 *in situ*. Biochemistry 1989;28:8990-8998. [PubMed: 2605237]
- 55. Alric J, Pierre Y, Picot D, Lavergne J, Rappaport F. Spectral and redox characterization of the heme ci of the cytochrome *b*6*f* complex. Proc Natl Acad Sci U S A 2005;102:15860–5. [PubMed: 16247018]
- 56. Zhang H, Primak A, Bowman MK, Kramer DM, Cramer WA. Characterization of the high-spin heme x in the cytochrome *b6f* complex of oxygenic photosynthesis. Biochemistry 2004;43:16329–36. [PubMed: 15610027]
- 57. Girvin ME, Cramer WA. A redox study of the electron transport pathway responsible for generation of the slow electrochromic phase in chloroplasts. Biochim Biophys Acta 1984;767:29–38. [PubMed: 6487614]
- 58. Selak MA, Whitmarsh J. Kinetics of the electrogenic step and cytochrome $b₆$ and f redox changes in chloroplasts. FEBS Letters 1982;150:286–292.
- 59. Jones RW, Whitmarsh J. Inhibition of electron transfer and electrogenic reaction in the cytochrome *b/f* complex by 2-n-nonyl-4-hydroxyquinoline N-oxide (NQNO) and 2,5-dibromo-3methyl-6 isopropyl-p-benzoquinone (DBMIB). Biochim Biophys Acta 1988;933:258–268.
- 60. Link TA, von Jagow G. Zinc ions inhibit the Qp center of bovine heart mitochondria *bc*1 complex by blocking a protonatable group. J Biol Chem 1995;270:25001–25006. [PubMed: 7559629]
- 61. Klishen, SS.; Mulkijanian, AY. Photosynthesis: Fundamental Aspects to Global Perspectives. 2. Montreal: 2005. p. 460-462.
- 62. Axelrod HL, Abreusch EC, Paddock ML, Okamura MY. Binding of Cd^{2+} and Zn^{2+} inhibit the rate of reduction and protonation of QB. Proc Nat Acad Sci USA 2000;97:1542–1547. [PubMed: 10677497]
- 63. Finazzi G. Redox-coupled proton pumping activity in cytochrome $b₆f$, as evidenced by the pH dependence of electron transfer in whole cells of Chlamydomonas reinhardtii. Biochemistry 2002;41:7475–7482. [PubMed: 12044181]
- 64. Yan J, Cramer WA. Molecular control of a bimodal distribution of quinone-analogue inhibitor binding sites in the cytochrome $b₆f$ complex. J Mol Biol 2004;344:481–93. [PubMed: 15522300]
- 65. Rich PR, Madgwick SA, Moss DA. The interactions of duroquinol, DBMIB, and NQNO with the chloroplast cytochrome *bf* complex. Biochim Biophys Acta 1991;1058:312–328.
- 66. Zatsman AI, Zhang H, Gunderson WA, Cramer WA, Hendrich MP. Heme-heme interactions in the cytochrome $b₆f$ complex: EPR spectroscopy and correlation with structure. J Amer Chem Soc 2006;128:14246–14247. [PubMed: 17076484]
- 67. Hosler JP, Yocum CF. Evidence for two photophosphorylation reactions concurrent with ferredoxincatalyzed non-cyclic electron transport. Biochim Biophys Acta 1985;808:21–31.
- 68. Bendall DS, Manasse RS. Cyclic phosphorylation and electron transport. Biochim Biophys Acta 1995;1229:23–38.
- 69. Tagawa K, Tsujimoto HY, Arnon DL. Role of chloroplast ferredoxin in the energy conversion process of photosynthesis. Biochemistry 1963;49:567–572.
- 70. Joliot P, Joliot A. Cyclic electron transfer in plant leaf. Proc Nat Acad Sci USA 2002;99:10209– 10214. [PubMed: 12119384]
- 71. Joliot P, Joliot A. Quantification of cyclic and linear electron flows in plants. Proc Nat Acad Sci USA 2005;102:4913–4918. [PubMed: 15781857]
- 72. Cooley JW, Ohnishi T, Daldal F. Binding Dynamics at the Quinone Reduction (Q(i)) Site Influence the Equilibrium Interactions of the Iron Sulfur Protein and Hydroquinone Oxidation (Q(o)) Site of the Cytochrome *bc*1 Complex. Biochemistry 2005;44:10520–32. [PubMed: 16060661]
- 73. Covian R, Trumpower BL. Regulatory interactions between ubiquinol oxidation and ubiquinone reduction sites in the dimeric cytochrome *bc*1 complex. J Biol Chem 2006;281:30925–30932. [PubMed: 16908520]
- 74. Zhang, H.; Cramer, WA. Purification and crystallization of the cytochrome $b₆f$ complex in oxygenic photosynthesis. In: Carpentier, R., editor. Photosynthesis Research Protocols. 274. Humana Press; Totowa, New Jersey: 2004. p. 67-78.
- 75. Otwinowski Z, Minor W. Processing of X-ray diffraction data collected in oscillation mode. Methods in Enzymology 1997;276:307–326.
- 76. CCP4. The collaborative computional project, number 4: Programs for protein crystallography. Acta Cryst 1994;D50:760–763.
- 77. Jones TA, Zou JY, Cowan SW, Kjeldgaard M. Improved methods for the building of protein models in electron density maps and the location of errors in these models. Acta Cryst 1991;A47:110–119.
- 78. Kraulis PJ. MOLSCRIPT: A program to produce both detailed and schematic plots of protein structures. J Applied Crystallography 1991;24:946–950.
- 79. Fenn TD, Ringe D, Petsko GA. POVScript+: A program for model and data visualization using persistence of vision ray-tracing. J Appl Cryst 2003;36:944–947.
- 80. Merritt EA, Bacon DJ. Raster3D: Photorealistic molecular graphics. Methods in Enzymology 1997;277:505–524. [PubMed: 18488322]

Abbreviations

TDS

tridecyl-stigmatellin

TM

trans-membrane

Fig. 1B

Fig. 1. Two cadmium (Cd²⁺) binding sites on the *p***-side of the** *M. laminosus b***₆***f* **complex**

(A) Position of higher occupancy (Cd1) is close to the inter-monomer interface, and that of lower occupancy (Cd2) site is near the small subunits and the exterior of the complex. View is parallel to the plane of the membrane. Distances: (i) from Cd1 site, and (ii) from Cd2, to the [2Fe-2S] cluster on the same and opposite side monomer, (i) 38.9 Å and 40.1 Å, and (ii) 57.1 and 28.0 Å. Color code: cytochrome $b₆$ (cyan), subunit IV (purple), cytochrome f (red), ISP (yellow), PetG, PetL, PetM, and PetN (green). **(B,** stereo**)** Environment of Cd1 and Cd2 sites shown in more detail. Lower occupancy of the Cd2 site is shown by the smaller cage of electron density. A lipid molecule (possibly galactolipid) described in the coordinates of the *C. reinhardtii* b_6f complex (pdb; 1Q90), but not previously discussed, is closer to Cd2. Distances: higher (Cd1) to lower occupancy (Cd2) Cd²⁺ site, 23.2 Å; higher occupancy Cd1 site to heme

 b_p : 24.0 Å; Cd2 to residues Glu78, Trp79 and Y80 of the 'PEWY' loop on the same monomer, 13, 12, and 16 Å. The seven peaks of largest amplitude in the anomalous difference map are: Cd1 site (18.5 σ), heme b_p (18.1 σ), heme c_n (17.6 σ), heme b_n (17.0 σ), heme f (13.3 σ), [2Fe-2S] site (11.1 σ), Cd2 site (6.3 σ). The anomalous scattering factors (f'') for Fe and Cd at 0.98 A are 1.50 and 2.13, respectively. The B factors of the Cd1 and Cd2 sites are 80 \AA^2 and 178 Å² .

Fig. 2A

Fig. 2B

Fig. 2.

Fo-Fc difference map on $[A]$ p - and $[B]$ n -side of the native $b₆f$ complex in the absence of added inhibitor ("native" structure). TM helices and surface helices within loops are labeled in upper and lower case, respectively. [**A**] Fo-Fc difference map, contoured at 4σ, in the native (with Cd^{2+}) b_6f complex. Origin of electron density under the cd2 loop, presumably lipid and/ or detergent, is not known. [**B**] Fo-Fc difference map of *n*-side background electron density in the native (with Cd²⁺) b_6f complex. Fo-Fc map contoured at 4 σ .

Fig. 3A

Fig. 3B

Fig. 3. Fo-Fc difference map of (A) *p***- and (B)** *n***-side binding site of TDS in the** *b***6***f* **complex** (**A**) *p*-side. The extra density within H-bond distance of the His129 ligand of the [2Fe-2S] cluster and between the 'ef' and 'cd1' loops is attributed to the chromone ring of TDS. As in Fig. 2A, the origin of the electron density under the cd2 loop, presumably lipid and/or detergent, is not known. (**B**) *n*-side. The position of TDS is shown relative to heme c_n , which is exposed to the quinone-exchange cavity. TDS is on the side of heme c_n distal to heme b_n . TM helices and surface helices within loops labeled as in Fig. 2. Fo-Fc maps contoured at 4σ.

Fig. 4A

Fig. 4B

Fig. 4. Fo-Fc difference map of (a) *p***-side and (b)** *n***-side binding site of NQNO in the** $b₆f$ **complex** (**A**) As in Fig. 2A, the origin of the electron density under the cd2 loop, presumably lipid and/ or detergent, is not known. The *p*-side plastoquinone binding niche that includes the [2Fe-2S] cluster and the *p*-side portal in the "roof" of the cavity is shown. (**B**) The Fo – Fc difference map in the region of heme c_n on the *n*-side of the b_6f complex shows the *n*-side binding site of NQNO. TM helices and surface helices within loops are labeled as in Fig. 2. Fo-Fc maps are contoured at 4σ.

Fig. 5. Reduction of hemes b_n/c_n **by reduced ferredoxin/FNR**

The reaction solution contained 1.5 μM spinach *b*₆*f* complex with bound FNR and 3 μM spinach ferredoxin. Semi-anaerobic conditions were obtained through the addition of 10 mM glucose and 160 units of glucose oxidase. Decyl-plastoquinol (50 μ M) was present to eliminate any contribution of cytochrome *f* to the difference spectra. Cytochrome b_6/c_n reduction was initiated by addition of 0.3 mM NADPH. **(A)** Comparison of reduction of spinach $b₆f$ complex by NADPH/ferredoxin ¹⁴ and dithionite. The NADPH/ferredoxin minus decyl-plastoquinol difference spectrum (blue trace) was measured 1 min after addition of NADPH. The spectrum of fully reduced cytochrome b_6 (pink) was obtained by addition of dithionite. **(B)** Dependence of reduction of hemes b_n/c_n reduction on the presence of FNR and ferredoxin. Reduction of spinach *b₆f* complex following addition of NADPH in the presence (blue) or absence (red) of ferredoxin (3 μM), and of *M. laminosus b*6*f* complex, which does not have bound FNR, in the presence (green) or absence (purple) of ferredoxin. All spectra were measured 1 min after addition of NADPH. **(C)** Double difference spectrum, normalized at 563 nm, showing the Soret band spectrum of heme *c*n (green), measured as the difference between the spectrum obtained in the presence and absence of NQNO. Difference spectra obtained in the absence (blue) and presence (pink) of NQNO (final concentration, 25 μM). **(D)** Proposed pathway of reduction of hemes b_n and c_n by reduced ferredoxin (Fd) in a Q cycle or cyclic pathway in which ferredoxin is reduced physiologically by the photosystem I reaction center and by NADPH in the present experiment. The requirement of FNR and Fd for the pathway of heme b_n reduction by NADPH is defined by the data shown in Fig. 5B. The ability of NQNO to block the pathway at the site of heme c_n is ascribed to NQNO binding as a ligand of this heme (Fig. 4B) and the NQNOinduced decrease in E_{m7} of heme c_n ⁵⁵. The inference that hemes b_n and c_n , together with PQ,

are part of a complex in which PQ serves as a ligand of heme *c*n is based on the proximity of the two hemes $8;9$, their interaction measured by EPR 66 , and binding of the quinone analogue inhibitors, TDS (Fig. 3B) and NQNO (Fig. 4B).

Intensity data and refinement statistics

Values in parentheses apply to the highest resolution shell.