Intracellular Concentrations of Coenzyme A and Its Derivatives from Clostridium acetobutylicum ATCC ⁸²⁴ and Their Roles in Enzyme Regulation

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Intracellular levels of coenzyme A (CoA) and its derivatives involved in the metabolic pathways for Clostridium acetobutylicum ATCC ⁸²⁴ were analyzed by using reverse-phase high-performance liquid chromatography (HPLC). During the shift from the acidogenic to the solventogenic or stationary growth phase, the concentration of butyryl-CoA increased rapidly and the concentrations of free CoA and acetyl-CoA decreased. These changes were accompanied by a rapid increase of the solvent pathway enzyme activity and a decrease of the acid pathway enzyme activity. Assays with several non-solvent-producing mutant strains were also carried out. Upon entry of the mutant strains to the stationary phase, the butyryl-CoA concentrations for these mutant strains were comparable to those for the wild type even though the mutants were deficient in solvent-producing enzymes. Levels of acetoacetyl-CoA, β-hydroxy-butyryl-CoA, and crotonyl-CoA compounds in both wild-type and mutant extracts were below HPLC detection thresholds $(<21 \mu M)$.

The anaerobic spore-forming bacterium Clostridium acetobutylicum is well known as a biological producer of acetone, ethanol, and butanol, all solvents of industrial interest (16). In batch culture, the typical acetone-butanol fermentation can be divided into two distinctive phases. In the acidogenic phase, the organism grows rapidly and produces acetate and butyrate, with the decrease in the pH of the medium resulting from the acid accumulation. In the solventogenic phase, the organism grows slowly and produces acetone, butanol, and ethanol, with the pH in the medium increasing slightly because of the acid reassimilation (6, 10, 24). The shift to solvent production is associated with the induction of solventogenic enzymes and a decrease in the activity of acidogenic enzymes (1, 10).

Although many investigators (7, 21, 22, 27, 32) have studied the biochemistry of acetone-butanol fermentation, the detailed mechanisms which control the shift to the solvent formation are not well understood. Significant progress on the analysis of the enzyme activities and internal pH involved in the clostridial pathway has been reported (4, 8-15, 17, 18, 20, 25-31). However, only recently have the in vivo levels of intermediate metabolites and cofactors, which may play an important role in enzyme regulation, been studied (9).

From the metabolic pathways related to the solventproducing clostridia, it is clear that different coenzyme A (CoA) derivatives (acetyl-CoA [Ac-CoA], acetoacetyl-CoA [AcAc-CoA], β-hydroxyl-butyryl-CoA, crotonyl-CoA, and butyryl-CoA [Bu-CoA]) are key intermediates in the metabolism of this organism. The metabolic fate of these compounds may directly affect the product pattern of the fer m entation (4, 5, 10). Reported here are the intracellular concentrations of free CoA and various CoA derivatives during the entire fermentation process, measured by using a reverse-phase chromatographic system. Assays on several

non-solvent-producing mutant cultures for comparison were also carried out. Furthermore, it was of interest to study the correlations of CoA compound concentrations with the specific enzyme activity levels before and during the onset of solventogenesis.

MATERIALS AND METHODS

Reagents. High-performance liquid chromatography (HPLC)-grade acetonitrile was purchased from EM Science (Gibbstown, N.J.), and sodium phosphate was from Mallinckrodt, Inc. (Paris, Ky.). Deionized water was further purified by passing it through a Milli-Q water purification system (Millipore Corp., Bedford, Mass.). CoA and its derivatives were obtained from Sigma Chemical Co. (St. Louis, Mo.). The dye-binding protein assay kit was obtained from Bio-Rad Laboratories (Richmond, Calif.)

HPLC apparatus. Separations of CoA compounds were performed with an LDC/Milton Roy GM ⁴⁰⁰⁰ HPLC system (Riviera Beach, Fla.). Two model III pumps were used to deliver eluants through a $5-\mu m$ octyldecyl silane column (4.6) by ²⁵⁰ mm) from Custom LC, Inc. (Houston, Tex.). A model 1203 UV-III detector was employed to monitor column effluent at ²⁵⁴ nm. The signal from the UV monitor was delivered to a CI-1OB integrator for quantification.

Chromatography. The two mobile-phase solvents used were buffer \bar{A} (0.2 M sodium phosphate [pH 5.0]) and buffer B (800 ml of 0.25 M sodium phosphate [pH 5.0] mixed with 200 ml of acetonitrile) (19). These buffers were passed through glass microfiber GF/F filters (Whatman International Ltd., Maidstone, England) prior to use. The chromatographic separations were performed at room temperature at a now rate of 1 m/min. The mobile-phase composition profile, after ^a 2.5-min constant buffer B level of 3% (i.e., 97% buffer A), was divided into several lineargradient segments, with sequential segment end points of 7.5 min (where 18% buffer B was reached), ¹⁰ min (28% buffer B), 15 min (30% buffer B), 25 min (40% buffer B), 26 min (42% buffer B), and 35 min (90% buffer B). At the 90% buffer B level, the percentage of buffer B was reduced, over ¹ min,

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to the original 3% level and then maintained to the end of the HPLC cycle (45 min). Data collection was stopped at ⁴⁰ min, and column reequilibration was complete after ⁴⁷ min. A constant injection volume of $10 \mu l$ was maintained for all HPLC runs.

CoA and the various acyl-CoA derivatives were made up as ⁵ mM stock solutions in ^a ¹⁰⁰ mM sodium phosphate buffer (pH 3.0) and stored at -80° C. Stock solutions prepared in this way were stable for at least ^a month.

Strains and culture conditions. The wild-type strain was C. acetobutylicum ATCC 824. Mutants 2BBD, M5, and 2BBR were isolated and described previously by Clark et al (3). Mutants 34, 4145, 1013, and 6a5 were Tn916 mutants prepared as described previously by Sass and Bennett (23). Maintenance of these organisms, including preparation of the medium, inoculation, heat shocking, and anaerobic growth, have been described previously (3, 23).

Method of culture preparation. Strains were grown anaerobically in 600 ml of reinforced soluble medium (RSM) (3) at 37°C. The pH was not controlled. Samples were taken at 2 to 4-h intervals to obtain data corresponding to the maximal solvent-producing stage. The appropriate harvest time was determined initially by measuring pH and A_{600} . Samples were centrifuged at 19,000 $\times g$ for 20 min at 4°C, the supernatants were analyzed for solvent production by the nitroblue tetrazolium method (3), and the pellets were resuspended in the cold soluble medium and centrifuged at 34,000 $\times g$ for another 20 min at 4°C. No significant loss of metabolite levels was observed with this second step, which allowed better chromatographic separations. The pellets were immediately powdered in a porcelain mortar and pestle cooled to -80° C with liquid nitrogen. The frozen powder was stored at -80° C until required.

Methods of enzyme assays. Cell extracts were prepared as described previously (3) except that the protein concentration was measured by the method of Bradford (2) with bovine serum albumin as the standard. Measurement of each enzyme activity was performed as described previously (3).

Methods of CoA compound extraction and recovery. Frozen powder (0.4 g) was weighed into a polystyrene centrifuge tube precooled with liquid nitrogen, and 0.8 ml of cold 6% perchloric acid (PCA) containing ² mM dithiothreitol was added. The mixture was sonicated at 90% power for 9 min for acidogenic cells and 15 to 18 min for solventogenic cells. The denatured protein was removed by centrifugation, and 0.8 ml of the PCA extract was transferred to ^a precooled Eppendorf centrifuge tube and adjusted to ^a pH of 3.0 by dropwise addition of 3 M K_2CO_3 with vortexing during the addition. The neutralized sample was centrifuged at 4°C, and the supematant was filtered through ^a 10,000 NMWL ultrafree-MC filter (Millipore Products Division, Bedford, Mass.) The resulting aliquot was stored at -20° C when not immediately analyzed by using the HPLC. Solutions stored in this manner were stable for a week or more.

Recovery of added standard CoA compounds was accomplished as follows. Either a 100 - μ l standard CoA mixture or the 100 mM $NaH₂PO₄$ (pH 3.0) solution was added to a centrifuge tube and cooled to liquid-nitrogen temperature. A 0.4-g amount of frozen powder was then added to the tube and extracted with 0.8 ml of ice-cold 6% PCA as described above. The resulting aliquot was analyzed in the HPLC. Each sample was run at least twice in the HPLC.

Four different samples from wild-type Clostridium cultures were extracted, and the recoveries were calculated. The percent recovery for free CoA was found to be ^a mean of 95% with a $\pm 3\%$ standard deviation; the other recoveries were $97\% \pm 18\%$ for Ac-CoA, $81\% \pm 6\%$ for AcAc-CoA, 93% \pm 2% for β -OH-butyryl-CoA, 82% \pm 3% for crotonyl-CoA, and $89\% \pm 13\%$ for Bu-CoA.

Determination of biomass concentration. Biomass concentration of the culture broth was measured by reading the optical density at 600 nm $(OD₆₀₀)$ and converting the readings to concentrations in grams (wet weight) per liter by using a mass extinction coefficient of 0.89 ± 0.03 liter gcm-'. The mass extinction coefficient was determined by measuring the actual biomass concentration and the corresponding OD_{600} under various conditions.

Intracellular and extracellular aqueous space measurement. To measure the intracellular volume at various times during the batch fermentation, a 5-ml cell suspension was incubated $\frac{1}{2}$ maerobically in a 15-ml disposable tube at $\frac{3}{5}$ C in the $\lim_{n \to \infty}$ with 1 μ C₁ of ["4C-carboxyl]dextran (American") Radiolabeled Chemicals Inc., St. Louis, Mo.) and 10μ Ci of ${}^{3}H_{2}O$ (New England Nuclear, Boston, Mass.) added to the assay. Dextran was used for the measurement of the extracellular water volume of the pellets and had a final concentration of 0.142 g/liter and a radioactivity of 1.4 mCi/g. Tritium water (i.e., ${}^{3}H_{2}O$) was used for measuring the total aqueous space in the pellet and had a final concentration of 110.7 mM (18.0 mCi/mol). The intracellular aqueous volume for the growing cells in their growth medium was determined by subtracting the extracellular space occupied by dextran from the total aqueous space (13, 14, 25). On the basis of 12 independent measurements each, the intracellular aqueous volumes of the acid-phase, solvent-phase, and late-solventphase cells are 0.21 ± 0.01 , 0.36 ± 0.02 , and 0.45 ± 0.02 ml/g (wet cell), respectively.

Calculations. Radioisotopes were used to measure the total (V_t) , intracellular (V_t) , and extracellular (V_o) aqueous volumes of the cell pellets. All units were milliliters per gram (wet weight).

PCA in the amount of 0.8 ml was added to 0.4 ^g of frozen cell powder (1:2 [wt/vol]). After extraction, the total aqueous volume for the CoA-compound solutions dissolved was 0.4 V_+ + 0.8, expressed in milliliters.

After centrifugation, neutralization, and filtration to separate out unwanted cell debris, the aliquot was analyzed by $HPLC$ while a 10 - μ l injection volume was maintained. The total amount of ^a measured CoA compound was determined from the expression $C_n(0.8 + 0.4 V_t)/0.4$ g, or $C_n(2 + V_t)$, expressed in nanomoles per gram (wet weight), where C_n is the concentration of a specific CoA compound, which was derived from the HPLC. The intracellular concentration is therefore $C_n(2 + V_t)/V_i$, expressed in micromolar.

RESULTS

Chromatography. A standard solution of all of the CoA compounds involved in the clostridial pathway was separated to a useful degree in 40 min by reverse-phase chromatography (Fig. 1A). CoA and its derivatives were distinctly separated while maintaining the baseline approximately level, thereby allowing for more accurate quantification of CoA compounds.

The typical chromatograms of PCA extract from ^a C. acetobutylicum wild-type solventogenic culture and a mutant culture are shown in Fig. 1B and C. Peaks were identified through their retention times and through spiking by adding corresponding standards. Peak areas were plotted against concentration for each CoA compound for quantification. CoA, Ac-CoA, and Bu-CoA were well separated and found in measurable quantities. AcAc-CoA, 3-OH-butyryl-

CoA, and crotonyl-CoA were separated in standard solutions but have not been found in measurable amounts in the

 $\sum_{i=1}^{n}$ Linearity, detection limits, and recovery. Linear relationships between the ratios of the peak areas of CoA com-
pounds and the amounts added in the range of from 25 to pounds and the amounts added in the range of from 25 to 4,000 pmol were observed by using full-scale absorbance of CoA 0.016 to 0.128. The respective lower limits of detection for CoA, A-CoA, AcAc-CoA, 3-OH-butyryl-CoA, crotonyl-CoA, and Bu-CoA were 15, 15, 25, 15, 12.5, and 15 pmol (with corresponding intracellular concentrations of 11, 11, 21, 11, 10, and 12 μ M for solvent-phase cells) at a minimum observed signal-to-noise ratio of 3.5.

Levels of intracellular CoA compounds from wild-type cultures. The intracellular CoA compound levels from a cultures. The intracellular CoA compound levels from ^a who-type pH-uncontrolled batch culture were measured, with the results shown in Table 1. The cultures were harvested more frequently during the onset of the solventogenic phase so that the rapidly changing metabolic levels

could be more clearly delineated.
It is known that the enzyme profiles of acid-phase and It is known that the enzyme profiles of acid-phase and
exist above alleges different. The example and delight solvent-phase cells are different. The enzymes needed for acid production, detected from their catalytic activity in cell enzyme activity, in contrast, is present only in cells har- μ enzyme activity, in contrast, is present only in cells harvested during the solvent phase (1, 10). To further characterize the correlation of enzyme induction with intracellular concentrations of key intermediates, the enzyme activity and the corresponding CoA compound concentrations around the onset of solventogenesis were measured. These results are also shown in Table 1.

Intracellular levels of CoA compounds from mutant cultures. To form a better understanding of the relationship between the CoA compound concentrations and solventogenesis, various non-solvent-producing mutants isolated prenesis, various non-solvent-producing mutants isolated previously in this laboratory $(3, 23)$ were analyzed. These mutants either did not produce any solvents (for mutants $\frac{dJ}{dt}$, 1013, MJ, and 4145 or produced smaller amounts of solvents than did the wild-type strains (for mutants $2BDD$) and 34). Except for mutant 34, all mutants are non-spore-forming strains. These mutants either lack enzymes or have only low levels of the enzymes involved in the solvent only low levels of the enzymes involved in the solvent production. The CoA compound concentrations for these mutants and the corresponding solvent pathway enzyme activities are listed in Table 2. Cultures were judiciously harvested during the early stationary phase. Cultures harvested past this point were difficult to sonicate and therefore vested past this point were difficult to some and therefore could lose their enzyme activity quickly; the CoA compound levels would also decrease rapidly.

DISCUSSION
The production of solvents by using clostridia is well The production of solvents by using clostridia is well known, but the initiation mechanisms for the switch to solvent production are still being widely studied. Current hypotheses proposed for the triggering mechanism that cause the shift from the acid to the solvent production phase involve the regulation of catalytic pathways in response to involve the regulation of catalytic pathways in response to the central concentrations of intermediates $(3, 0)$. Gottward

FIG. 1. (A) Elution profile of a mixture containing 1 nmol of CoA and noted acyl-CoA intermediate standards (detector sensitivity, and noted acyl-CoA intermediate standards (detector sensitivity),

1.128 full-scale absorbance); (B) elution profile of a 10-pl PCA extract from wild-type Clostridium solventogenic cells; (C) elution profile of a 10-µl PCA extract from *Clostridium* mutant culture M5.

TABLE 1. Intracellular CoA compound concentrations and the solvent pathway enzyme activities in ^a pH-uncontrolled batch culture of wild-type C. acetobutylicum

Sample	OD ₆₀₀	рH	Time (h)	Concn				Enzyme activity ^{a}							
				Butanol (mM)	CoA (μM)	Ac-CoA (μM)	Bu-CoA (μM)	PTB	BK^b	NADH- BDH	NADPH- BDH	NAD- BAD	NADP- BAD	CoA transferase	
Acid cells	2.40	4.27	- 11	0	185 ± 16	436 ± 22	574 ± 39								
Late acid cells	4.47	3.89	- 15	4.0	161 ± 14	456 ± 14	512 ± 18	20,100	900	5.7	22.5	14.6	12.6	17.9	
Early solvent cells	5.12		3.82 16.5	21	124 ± 7	316 ± 9	810 ± 24	16,500	840	16.6	50.0	30.4	47.0	316	
Solvent cells	5.14	3.82	- 18	>40	102 ± 3	327 ± 11	857 ± 23	12,400	680	4.8	47.7	22.3	22.3	613	
	5.86	3.79	-20		145 ± 6	396 ± 19	882 ± 30	13,200	600	7.8	85.7	51.6	34.4	1,200	
	6.13	3.80	- 22		122 ± 10	327 ± 5	1.002 ± 19								
	6.26	3.93 23			154 ± 12	383 ± 21	690 ± 25								
	7.82	3.92 27			140 ± 1	368 ± 11	645 ± 21								

^a Units for each enzyme are defined as nanomoles of product formed per milligram of protein.

 b BK, butyrate kinase.</sup>

phase, where the reactions which lead from Bu-CoA via butyryl-phosphate to butyrate are reversible, an elevated concentration of butyrate will result in higher levels of compound concentrations were also measured relative to the butyryl-phosphate and Bu-CoA accumulating in the cell. This will result in significant decreases in both the CoA and the phosphate pools, triggering activation or synthesis of the enzymes for solventogenesis. Results herein show that, during the switch to solvent production, the Bu-CoA concentration exhibited a marked increase from 512 to 810 μ M. centration exhibited a marked increase from 512 to 810 μ M. rapidly. Conversely, the activity of the acid-producing en-
Correspondingly, the CoA concentration decreased from 161 zymes, phosphotransbutyrylase (PTB) and b to 124 μ M, and the Ac-CoA concentration decreased from decreased.
456 to 316 μ M. In other stages of the fermentation, this As a control, the results from a wild-type culture were 456 to 316 μ M. In other stages of the fermentation, this concentration remains relatively constant (Table 1).

continuous culture of *C*. acetobutylicum DSM 1731, show solvent-producing strains (1013, 6a5, 4145, and M5) or are that, during the shift, contrary to the expected positive reduced-solvent-producing strains (2BBD and 34) that, during the shift, contrary to the expected positive reduced-solvent-producing strains (2BBD and 34) because pattern (first increase and then decrease) for Ac-CoA and they lack the solvent-producing enzymes mentioned pattern (first increase and then decrease) for Ac-CoA and they lack the solvent-producing enzymes mentioned above.
Bu-CoA, a negative (first decrease and then increase) pattern From these results (Table 2), it can be seen for CoA, Ac-CoA, and Bu-CoA was observed. Results generally exhibit reduced BDH levels and no CoA trans-
herein showed a similar negative pattern of CoA and Ac-
ferase or BAD activity. However, these mutants can proherein showed a similar negative pattern of CoA and Ac-
CoA compared with their results. However, during the shift, duce Bu-CoA concentrations comparable to, or even higher, CoA compared with their results. However, during the shift, duce Bu-CoA concentrations comparable to, or even higher, a consistent increase in Bu-CoA concentration was detected, than those for wild-type cultures, possibly until the highest level of 1,000 μ M was observed (Table 1). higher levels of butyrate cannot be reassimilated.
This discrepancy may result from the differences in the mode The main advantage of knowing the concentratio This discrepancy may result from the differences in the mode of the culture growth (continuous versus batch) and the of the culture growth (continuous versus batch) and the metabolic intermediates is that, after comparing the kinetic culture operation. The batch cultures are in less-steady-state and physiological properties of the enzymes involved, the conditions than continuous cultures; therefore, some tran-
influence of the metabolite in regulating

and Gottschalk (8) point out that, at the end of the acidogenic To further characterize the correlation of solvent pathway phase, where the reactions which lead from Bu-CoA via enzyme induction and intracellular concentrat intermediates, representative enzyme activities and CoA compound concentrations were also measured relative to the show that, along with the rapid increase in Bu-CoA concentration, the activity level of solvent pathway enzymes, namely CoA transferase, butyraldehyde dehydrogenase (BAD), and butanol dehydrogenase (BDH), also increase zymes, phosphotransbutyrylase (PTB) and butyrate kinase decreased.

orcentration remains relatively constant (Table 1). compared with those of various mutant cultures at the onset
Recent data from Grupe and Gottschalk (9), who used a contract of the stationary phase. These mutants are eith Recent data from Grupe and Gottschalk (9), who used a of the stationary phase. These mutants are either non-
continuous culture of C. acetobutylicum DSM 1731, show solvent-producing strains (1013, 6a5, 4145, and M5) or are Bu-CoA, a negative (first decrease and then increase) pattern From these results (Table 2), it can be seen that the mutants for CoA, Ac-CoA, and Bu-CoA was observed. Results generally exhibit reduced BDH levels and no CoA than those for wild-type cultures, possibly because the higher levels of butyrate cannot be reassimilated.

conditions than continuous cultures; therefore, some tran-
sient effect may not be observed.
can be evaluated. Those enzymes which react with the CoA can be evaluated. Those enzymes which react with the CoA

TABLE 2. Intracellular CoA compound concentrations and the solvent pathway enzyme activities from various Clostridium mutants at the onset of the stationary phase

Strain	OD ₆₀₀	рH		Enzyme activity ^a									
			Solvent (mM)	CoA (μM)	Ac-CoA (μM)	Bu-CoA (μM)	PTB	BK^b	NADH- BDH	NADPH- BDH	NAD- BAD	NADP- BAD	CoA transferase
Wild type	5.12	3.82	21	124 ± 7	316 ± 9	810 ± 24	16,500	840	16.6	50.0	30.4	47.0	316
1013	5.28	3.88	0	136 ± 2	319 ± 20	1.180 ± 61	10,900	1.480	1.1	17.2	0	0	0
6a5	5.34	3.95	0	76.4 ± 14.2	193 ± 18	1.101 ± 42	22,600	813	$\bf{0}$	20.8	0	0	
M5	6.00	3.92	0	284 ± 12	222 ± 8	1.169 ± 34	9.600	1,190	1.6	83.1	0	0	0
2BBD	6.26	3.76	NT ^c	162 ± 2	398 ± 1	679 ± 3	9,300	1,020	$\bf{0}$	26.0	NT	NT	NT
34	4.52	3.93	7.0	88.0 ± 6.9	302 ± 3	761 ± 51	12.900	1.380	4.4	35.7	1.5	1.5	79.2
4145	5.27	3.72	0	159 ± 4	225 ± 7	1.326 ± 16	14,900	1,170	21.2	27.0	28.4	34.0	39.0

Units for each enzyme are defined as nanomoles of product formed per milligram of protein.

 b BK, butyrate kinase.</sup>

^c NT, not tested.

FIG. 2. Metabolic pathway of C. acetobutylicum and the kinetic properties of the enzymes. The K_m values are given in parentheses. The intracellular CoA, Ac-CoA, and Bu-CoA concentrations are given in brackets. Notes: N_{m} values derived from C. acetobutylicum ATCC 824 (^{a1}) and B643 (^{a2}); ^b, K_m values derived from C. butylicum NRRL B593; °, NK, not known. Abbreviations: Ki, inhibition constant; PTA, phosphotransbutyrylase; AAD, acetaldehyde dehydrogenase; ADH, alcohol dehydrogenase; CoA Tr, CoA transferase.

derivatives are indicated in Fig. 2, along with the K_m s for the purified enzymes. Except for β -hydroxybutyryl-CoA dehypurified enzymes. Except for p-hydroxybutyryl-CoA dehy-
trogonose (BUBD), which was purified from Clostridium drogenase (BHBD), which was purified from Clostridium butylicum NRRL B593 (4), all other noted enzymes are derived from *C. acetobutylicum* (usually ATCC 824) (20, 27-30). $\frac{20}{2}$

The measured intracellular concentrations of CoA, Ac-CoA, and Bu-CoA (Table 1) were generally above 100, 316, and 512 μ M, respectively. These values are sufficient to allow maximal flux through the BAD and acetaldehyde dehydrogenase reactions for which the K_m values of 55 μ M for Bu-CoA and 45 μ M for Ac-CoA were determined (20).
These values should also be able to nearly saturate the PTB These values should also be able to nearly saturate the PTB reaction, for which a K_m value of 110 μ M for Bu-CoA in the forward direction was reported (29). In fact, a change in internal pH may be an important factor in the regulation of the PTB reaction (29). Within the physiological pH change from 5.5 to 7, the internal pH range for typical fermentations (12), the PTB activity decreases significantly in the forward direction as pH decreases and shows virtually no activity below ^a pH of ⁶ (29). Therefore, the level of Bu-CoA is not seen as an important regulating factor for PTB activity, as suggested previously (29).

In the condensation reaction with thiolase, the K_m value for Ac-CoA had been determined to be $270 \mu M$, and the thiolase activity was found to be inhibited by even micromolar levels of free CoA (28). Under these conditions, the intracellular CoA levels should be very inhibitory to this reaction, and the Ac-CoA would still be below a saturating level. The intracellular concentrations of CoA and Ac-CoA are likely important factors in regulating the net condensation of Ac-CoA to AcAc-CoA and in regulating the pattern of product formation.

Three of the CoA derivatives involved in the metabolic pathways, namely, AcAc-CoA, 3-OH-butyryl-CoA, and crotonyl-CoA, were not detected. It was known that these compounds are unstable, and thus several extractions with externally added CoA compounds were carried out to check the reproducibility and recoverability of the extraction and assay procedures. From four different samples, the recoveries of AcAc-CoA, β -OH-butyryl-CoA, and crotonyl-CoA were 81, 93, and 82%, respectively. On the basis of these recoveries, the corresponding intracellular concentrations of recoveries, the corresponding intracellular concentrations of ACAC-COA, p-OH-butyryl-COA, and crotonyl-COA should be less than 21 , 11, and 10 μ M. The enzymes which react with these CoA derivatives, except Bu-CoA dehydrogenase, have been purified from either *C. acetobutylicum* (i.e., CoA transferase and crotonase [27, 30]) or *C. butylicum* (i.e., BHBD [4]), and the K_m values are shown in Fig. 2. In the SHBD [4]), and the K_m values are shown in Fig. 2. In the crotonase reaction, the K_m value for the product crotonyl-
Can use 20 will and the entury we repeated to be CoA was 30 μ M, and the enzyme was reported to be sensitive to levels of crotonyl-CoA above those observed here. However, the K_m value for β -OH-butyryl-CoA was not determined (27). For the BHBD from C. butylicum, a K_m value for AcAc-CoA of 14 μ M was determined (4). This reference also reported that inhibition of BHBD activity by AcAc-CoA was observed at concentrations at or above its K_m when the NADH concentration was near its K_m , suggesting that AcAc-CoA plays an important role in the in vivo regulation of BHBD.

None of the three CoA intermediates could be detected when the mutant cultures were used (Table 2). Even in the absence of solvent pathway enzymes, these compounds do not accumulate, thus suggesting a tight regulation of their synthesis. Further investigation of the role of these three CoA derivatives in enzyme regulation awaits the isolation of mutants having specific enzymes removed.

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REFERENCES

- 1. Andersch, W., H. Bahl, and G. Gottschalk. 1983. Levels of enzymes involved in acetate, butyrate, acetone and butanol formation by *Clostridium acetobutylicum*. Eur. J. Appl. Micro-
biol. Biotechnol. 18:327-332.
- biol. Biotechnol. 18:327-332.
2. **Bradford, M. M.** 1976. A rapid and sensitive method for the quantitation of microgram quantities of protein utilizing the principle of protein-dye binding. Anal. Biochem. 72:248-254.
- 3. Clark, S. W., G. N. Bennett, and F. B. Rudolph. 1989. Isolation and characterization of mutants of Clostridium acetobutylicum ATCC ⁸²⁴ deficient in acetoacetyl-coenzyme A:acetate/butyrate:coenzyme A-transferase (EC 2.8.3.9) and in other solvent pathway enzymes. Appl. Environ. Microbiol. 55:970-976.
- 4. Colby, G. D., and J.-S. Chen. 1992. Purification and properties of 3-hydroxybutyryl-coenzyme A dehydrogenase from Clostridium beijerinckii ("C. butylicum") NRRL B593. Appl. Environ.
Microbiol. 58:3297-3302. MICrobiol. 56:3297-3302.
- 5. Datta, R., and J. G. Zeikus. 1985. Modulation of acetonebutanol-ethanol fermentation by carbon monooxide and organic acids. Appl. Environ. Microbiol. 49:522-529.
- 6. Davies, R. 1942. Studies on the acetone-butyryl alcohol fermentation. 2. Intermediates in the fermentation of glucose by Cl. acetobutylicum. 3. Potassium as an essential factor in the fermentation of maize meal by Cl. acetobutylicum (BY). Biochem. J. 36:582-599.
- 7. Davies, R., and M. Stephenson. 1941. Studies on the acetonebutyl alcohol fermentation. 1. Nutritional and other factors involved in the preparation of active suspensions of C. acetobutylicum (Weizmann). Biochem. J. 35:1320-1331.
- 8. Gottwald, M., and G. Gottschalk. 1985. The internal pH of Clostridium acetobutylicum and its effect on the shift from acid to solvent formation. Arch. Microbiol. 143:42-46.
- 9. Grupe, H., and G. Gottschalk. 1992. Physiological events in Clostridium acetobutylicum during the shift from acidogenesis to solventogenesis in continuous culture and presentation of a model for shift induction. Appl. Environ. Microbiol. **58:**3896-
3902.
- 3902. 10. Hartmanis, M. G., and S. Gatenbeck. 1984. Intermediary metabolism in Clostridium acetobutylicum: levels of enzymes involved in the formation of acetate and butyrate. Appl. Environ. Microbiol. 47:1277-1283.
- 11. Hartmanis, M. G., T. Klason, and S. Gatenbeck. 1984. Uptake and activation of acetate and butyrate in Clostridium acetobutylicum. Appl. Microbiol. Biotechnol. 20:66-71.
- 12. Huang, L., L. N. Gibbins, and C. W. Forsberg. 1985. Transmembrane pH gradient and membrane potential in Clostridium acetobutylicum during growth under acetogenic and solventogenic conditions. Appl. Environ. Microbiol. 50:1043-1047.
- 13. Husemann, M. H. W. 1989. Ph.D. thesis. Department of Chemical Engineering, Rice University, Houston, Tex.
- 14. Husemann, M. H. W., and E. T. Papoutsakis. 1988. Solventogenesis in Clostridium acetobutylicum fermentations related to

carboxylic acid and proton concentrations. Biotechnol. Bioeng.
32:843-852. $2.043-0.32.$

- 15. Husemann, M. H. W., and E. T. Papoutsakis. 1989. Enzymes limiting butanol and acetone formation in continuous and batch cultures of Clostridium acetobutylicum. Appl. Microbiol. Biocennol. $31.433 - 444.$
- $10.$ Jones, D. $1.$, and D. R. Woods. 1986. Acetone-butanol fermenation revisited. Microbiol. Rev. $50.484-524$.
- 17. Jungermann, K., E. Rupprecht, C. Ohrloff, R. Thauer, and K. Decker. 1971. Regulation of the reduced nicotinamide adenine dinucleotide-ferredoxin reductase system in Clostridium kluyveri. J. Biol. Chem. 246:960–963.
18. Jungermann, K., R. K. Thauer, G. Leimenstoll, and K. Decker.
- 1973. Function of reduced pyridine nucleotide-ferredoxin oxi-1973. Function of reduced pyridine nucleotide-ferredoxin oxidoreductases in saccharolytic clostridia. Biochim. Biophys. $\frac{303200 - 200}{256}$
- 19. King, M. T., and P. D. Reiss. 1985. Separation and measurement of short-chain coenzyme-A compounds in rat liver by reversephase high-performance liquid chromatography. Anal. Biochem. $146:173-179$.
20. **Palosaari, N. R., and P. Rogers.** 1988. Purification and proper-
- ties of the inducible coenzyme A-linked butyraldehyde dehyties of the inducible coenzyme A-linked butyraldehyde dehydrogenase from Clostridium acetobutylicum. J. Bacteriol. 170.
0051-0056 $271 - 2776.$
- 21. **Fapoutsakis, E. T.** 1904. Equations and calculations for fermen t_{at} of butyric acid bacteria. Biotechnol. Bioeng. 26:174-187.
- 22. Rosenfeld, B., and E. Simon. 1950. The mechanism of the butanol-acetone fermentation. 1. The role of pyruvate as an intermediate. 2. Phosphoenolpyruvate as a new intermediate. J. BIOL. Chem. 180:393-410.
- 23. Sass, C., and G. N. Bennett. 1993. Isolation of mutants of Clostridium acetobutylicum ATCC 824 deficient in protease activity. Curr. Microbiol. 26:151-153.
- activity. Curr. Microbiol. 26:151-153. 24. Spivey, M. J. 1978. The acetone-butanol-ethanol fermentation process. Process Biochem. 13:2-5.
25. Terracciano, J. S., and E. R. Kashket. 1986. Intracellular
- conditions required for initiation of solvent production of Closconditions required for initiation of solvent production of Clos t_{min} acetobutylicum. Appl. Environ. Microbiol. 52:86-91.
- 26. Thompson, D. K., and J. S. Chen. 1990. Purification and properties of an acetoatetyl coenzyme A-reacting phosphotransproperties of an acetoatetyl coenzyme A-reacting phosphotransbutyrylase from Clostridium beijerincki ("clostridium butylicum") NRRL B593. Appl. Environ. Microbiol. 56:607-613.
- 27. Waterson, R. M., P. J. Castellino, G. M. Haas, and R. L. Hill. 1972. Purification and characterization of crotonase from Clostridium acetobutylicum. J. Biol. Chem. 247:5266-5271.
- 28. Weisenborn, D. P., F. B. Rudolph, and E. T. Papoutsakis. 1988. Thiolase from Clostridium acetobutylicum ATCC 824 and its role in the synthesis of acids and solvents. Appl. Environ. Microbiol. 54:2717-2722. $\frac{1}{2}$
- $29.$ Weisenborn, D. P., F. R. Rudolph, and E. T. Fapoutsakis. 1989. Phosphotransbutyrylase from Clostridium acetobutylicum ATCC ⁸²⁴ and its role in acidogenesis. Appl. Environ. Microbiol. 55:317-322.
- 30. Weisenborn, D. P., F. R. Rudolph, and E. T. Papoutsakis. 1989. Coenzyme A transferase from Clostridium acetobutylicum ATCC 824 and its role in the uptake of acids. Appl. Environ. Microbiol. 55:323-329.
- Microbiol. 55:323-329. 31. Welch, R. W., S. W. Clark, G. N. Bennett, and F. B. Rudolph. 1992. Effects of rifampicin and chloramphenicol on product and enzyme levels of the acid- and solvent-producing pathways of Clostridium acetobutylicum (ATCC 824). Enzyme Microb. Technol. 14:277-283.
- 32. Wood, H. G., R. W. Brown, and C. H. Werkman. 1945. Mechanisms of the butyl alcohol fermentation with heavy carbon acetic and butyric acids and acetone. Arch. Biochem. 6:243-260.