# Isolation of a Cytochrome-Deficient Mutant Strain of Sporomusa sphaeroides Not Capable of Oxidizing Methyl Groups

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The homoacetogenic anaerobic bacterium Sporomusa sphaeroides was mutagenized with UV light. Taking advantage of the ampicillin enrichment technique and a newly developed test for the detection of heme in bacterial colonies, the cytochrome-deficient mutant strain S. sphaeroides BK824 was isolated. In contrast to the wild type, this mutant strain failed to grow on betaine, betaine plus methanol,  $H_2$  plus  $CO_2$ , and methanol plus  $CO_2$ . Growth on betaine plus formate, betaine plus H<sub>2</sub>, betaine plus pyruvate, methanol plus H<sub>2</sub> and CO<sub>2</sub>, and acetoin was not impaired. All enzymes of the Wood pathway as well as hydrogenase and carbon monoxide dehydrogenase were detectable at comparable activities in both the wild type and the cytochrome-deficient mutant. Labeling experiments with  $[14C]$ methanol demonstrated the inability of S. sphaeroides BK824 to oxidize methyl groups. The role of cytochromes in electron transport steps associated with the Wood pathway enzymes and their possible role in energy conservation during autotrophic growth in acetogens are discussed.

Sporomusa sphaeroides is an obligately anaerobic, gramnegative, endospore-forming bacterium which converts a number of substrates, such as betaine, methanol plus  $CO<sub>2</sub>$ ,  $H_2$  plus  $CO_2$ , pyruvate, lactate, and acetoin mainly to acetate (24). Betaine is either dismutated according to equation 1 or, in the presence of an external electron donor such as formate or hydrogen, reductively split according to equation 2:

4 betaine  $\rightarrow$  3 acetate + 3 trimethylamine

 $+ N_x N$ -dimethylglycine  $+ CO_2$  (1)

betaine +  $2e^- + 2H^+$   $\rightarrow$  acetate + trimethylamine (2)

A reductive cleavage of betaine according to equation <sup>2</sup> was previously reported for Clostridium sporogenes and Eubacterium acidaminophilum (13, 26). The betaine reductase is assumed to function analogously to the glycine reductase system elucidated for E. acidaminophilum (5).

After the first report on the presence of cytochromes in S. sphaeroides and Sporomusa ovata (24), these electron carriers were also detected in other species of this genus (2, 4). A thorough characterization of the cytochromes of S. sphaeroides and S. ovata indicated the presence of two band two c-type cytochromes (19). Although these electron carriers are not unusual for homoacetogenic bacteria (10), their physiological function is not really understood. In Clostridium thermoautotrophicum, membrane-bound b-type cytochromes were found to interact with carbon monoxide dehydrogenase (14) or hydrogenase (15). Furthermore, b-type cytochromes were thought to be involved in the fumarate reductase reaction as performed by the homoacetogenic bacterium Clostridium formicoaceticum (6). In contrast to S. sphaeroides, S. ovata displays a clear dependence of cytochrome expression on the growth substrate: an 11fold increase in the cytochrome content was observed after growth on betaine compared with that after growth on betaine plus formate (19), indicating a role of cytochromes during betaine disproportionation.

In order to find a clue to the role of these electron carriers, a cytochrome-deficient mutant strain of S. sphaeroides was isolated. Physiological investigations of this mutant demonstrated that the mutation affects the ability of the mutant to oxidize methyl groups.

### MATERIALS AND METHODS

Bacterial strain. The experiments were performed with S. sphaeroides DSM 2875, which was obtained from the Deutsche Sammlung fur Mikroorganismen (Braunschweig, Germany). For maintenance, the organism was grown at 37°C in the standard medium described below containing <sup>50</sup> mM betaine as the energy source.

Chemicals. Enzymes and cofactors, except tetrahydrofolate  $(H_4F)$  and NADH, were purchased from Boehringer (Mannheim, Germany). H<sub>4</sub>F, dithioerythritol (DTE), heme, and 8-aminolevulinic acid were from Sigma (Deisenhofen, Germany), and NADH was obtained from Biomol (Hamburg, Germany). 3,3',5,5'-Tetramethylbenzidine was purchased from Serva (Heidelberg, Germany). All other chemicals were of high purity and purchased from commercial sources. Gases were from Messer Griesheim (Kassel, Germany). All radiochemicals except <sup>14</sup>CH<sub>3</sub>-H<sub>4</sub>F were purchased from New England Nuclear, Dreieich, Germany.  $^{14}CH_{3}$ -H<sub>4</sub>F was from Amersham (Braunschweig, Germany).

Growth media and cultivation. S. sphaeroides was grown at 37°C under  $N_2$  in the medium described previously (19) except that the phosphate concentration was increased to 3.48 g of  $K_2HPO_4$  and 2.27 g of  $KH_2PO_4$  per liter. Betaine, betaine plus formate, and acetoin served as the energy sources and were added to <sup>a</sup> <sup>50</sup> mM final concentration. The acetoin stock solution was filter sterilized. Solid medium was prepared in an anaerobic chamber (Megaplex, Grenchen,

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Switzerland) after addition of 1.5% agar (Oxoid, Basingstoke, England) to the medium and autoclaving.

When  $CO<sub>2</sub>$  is explicitly listed as a substrate component, S. sphaeroides was grown in bicarbonate-buffered medium (24) under  $N_2$ -CO<sub>2</sub> (80:20) or, when  $H_2$  served as the electron donor, under  $H_2$ -CO<sub>2</sub> (80:20); the latter gas mixture was pressurized to 1.5 atm (151.9 kPa). Methanol and pyruvate were added from concentrated stock solutions to a final concentration of 50 mM. The pyruvate solution was filter sterilized. Autotrophic cultures were shaken at 30'C. For cultivation of the mutant strain on methanol plus  $H_2$  and  $CO<sub>2</sub>$ , the bicarbonate concentration was increased to 3 g/liter. Where indicated, heme and 8-aminolevulinic acid were added from stock solutions to the growth medium to give final concentrations of 4  $\mu$ M and 1 mM, respectively. The stock solutions were prepared as follows. Hemin was dissolved in <sup>10</sup> mM KOH containing 50% methanol and filter sterilized; b-aminolevulinic acid was neutralized, made anaerobic, and filter sterilized before addition to the medium.

The cultivation was performed under strictly anaerobic conditions. For liquid medium, rubber-stoppered bottles or tubes were used and additions were made with sterile syringes. Growth was monitored by measuring the optical density at 600 nm (OD $_{600}$ ) in a Bausch & Lomb photometer with medium alone as a blank. Agar plates were inoculated in an anaerobic chamber, subsequently transferred to an anaerobic jar (Oxoid, Wesel, Germany) with  $N_2-H_2$  (95:5) as the gas phase, and incubated at  $37^{\circ}$ C.

Isolation of cytochrome-deficient mutant strain S. sphaeroides BK824 after UV mutagenesis. All media used in these experiments were additionally supplied with <sup>8</sup> mg of vitamin  $\overrightarrow{B_{12}}$  (Serva, Heidelberg, Germany) per liter. Ten milliliters of a betaine-grown culture of S. sphaeroides (OD<sub>600</sub> of 0.3) was centrifuged at 4,000  $\times$  g, and the cells were suspended in 15 ml of sterile <sup>50</sup> mM potassium phosphate buffer (pH 7.0) containing 3% NaCl and 0.5 g of L-cysteine hydrochloride per liter. The cell suspension was transferred to an anaerobic chamber, and 10 ml was poured into a sterile glass petri dish. An aliquot of 0.1 ml was diluted and plated onto plates containing betaine plus formate in order to determine the number of viable cells. The cell suspension in the petri dish was irradiated with short-wavelength UV light (254 nm) for <sup>20</sup> <sup>s</sup> with <sup>a</sup> prewarmed UV lamp. Irradiation and all following steps were performed in dim light. A portion (0.1 ml) of the irradiated cell suspension was serially diluted and spread onto betaine-formate plates in order to estimate the rate of survival, which was 0.6%. A survival rate of 0.1 to 1% has been recommended for UV mutagenesis (3).

Five tubes, each containing 5 ml of betaine-formate medium, were inoculated with 0.5 ml of the irradiated cell suspension. After 40 h of incubation, the optical density was about 0.9. This incubation under conditions not requiring cytochromes served to dilute out the residual cytochromes in the mutants in order to prevent them from starting to grow during the following ampicillin enrichment step. The cultures were centrifuged at  $4,000 \times g$ , and the cells were washed once in 10 ml of the above-mentioned sterile buffer. Washed cells were suspended in 5 ml of buffer, 3 ml of which served as an inoculum for 10 ml of betaine medium for the mutant enrichment. The tubes were incubated until the optical density had doubled. Subsequently, 0.2 ml of sterile anaerobic ampicillin (15 mg/ml) was added to each tube. After 12 h of shaking, the cultures had cleared up, whereas an ampicillin-free control reached an optical density of 1.4. The cultures were centrifuged, and the pellets were washed twice

in the aforementioned buffer to remove the ampicillin. The pellet was suspended in 3 ml of betaine-formate medium, and 0.2-ml aliquots were spread onto betaine-formate agar plates. After 7 days of incubation, colonies were picked. Each colony picked was transferred in parallel to a sterile nylon membrane (Biodyne A;  $0.2 \mu m$ ; Pall, Dreieich, Germany), which was placed on a betaine-formate agar plate and on <sup>a</sup> betaine-formate agar plate. When new colonies had appeared, the filters were tested for heme as described below. Four of 2,000 colonies picked stained heme negative. Since all of them originated from the same enrichment culture, they are assumed to be the same clone. Colonies staining negative for heme were picked and transferred to liquid betaine-formate medium. After growth had occurred, the stability of the mutation and purity of the clone were checked again by plating and staining for heme as described above.

Heme staining of colonies. Colonies grown on nylon transfer membranes as described above were lysed by incubating the filters in 1% sodium dodecyl sulfate (SDS)-0.2% NaOH for <sup>5</sup> min at room temperature with gentle shaking. The filters were subsequently rinsed with water for a few seconds and incubated in the dark for <sup>1</sup> to 2 h at room temperature in <sup>a</sup> mixture of seven parts 0.25 M sodium acetate (pH 5.0) and three parts of <sup>a</sup> freshly prepared solution of 6.3 mM 3,3',5,5' tetramethylbenzidine in methanol (31). The reaction was started by adding 0.1% (vol/vol) 30%  $\text{H}_2\text{O}_2$ . Heme-containing colonies stained turquoise blue within a few seconds, while heme-deficient colonies remained white. The reaction was stopped by rinsing the filters with water.

Redox difference spectroscopy and enzyme assays. Washed membrane fractions were prepared under aerobic conditions as described before (19). The redox difference spectra of washed membranes were recorded at room temperature by reducing a sample with a few grains of  $Na<sub>2</sub>S<sub>2</sub>O<sub>4</sub>$  and measuring it against an air-oxidized sample with a Perkin-Elmer/ Hitachi spectrophotometer (model 556) in the double-beam mode (18). The spectral bandwidth was set to <sup>1</sup> nm, and the scanning speed was 30 nm/min.

Cell extracts for the determination of enzyme activities were prepared under strictly anaerobic conditions and assayed as described before (19). Hydrogenase activity was assayed under  $H_2$  in 1.6-ml stoppered glass cuvettes filled with 0.8 ml of <sup>50</sup> mM anaerobic potassium phosphate buffer (pH 7.0) containing 5 mM DTE; 1  $\mu$ l of 1 M benzyl viologen, 10  $\mu$ l of 15 mM NAD, or 10  $\mu$ l of 15 mM NADP was added as the electron acceptor. The reaction was started with <sup>1</sup> to  $2 \mu$ I of crude extract, and the reaction was monitored at 578 and 365 nm, respectively. The extinction coefficients used for calculating the activity were  $\varepsilon_{578} = 8.6 \text{ mM}^{-1} \text{ cm}^{-1}$  for benzyl viologen and  $\varepsilon_{365} = 3.4 \text{ m}\text{M}^{-1} \text{ cm}^{-1}$  for NAD(P)H. When  $H_2$  was replaced by  $N_2$  as a control, no reaction occurred. CO dehydrogenase was assayed in the same way as hydrogenase except that  $H_2$  was replaced by CO and only benzyl viologen served as the electron acceptor.

Acetogenesis from  $CH_3$ - $H_4F$  by crude extracts was assayed by the method of Drake et al. (7). Crude extracts were prepared under strictly anaerobic conditions in <sup>50</sup> mM potassium phosphate buffer, pH 6.5, containing <sup>5</sup> mM  $MgCl<sub>2</sub>$ , 10 mM DTE, and 1 mg of resazurin per liter. The test was performed in 4.7-ml rubber-stoppered tubes under  $H_2$ -CO (75:25) containing 250  $\mu$ l of crude extract, 2 mM ATP, and 0.2 mM coenzyme A (CoA). The tubes were incubated at 30°C for <sup>1</sup> min, and the reaction was started by the addition of <sup>14</sup>CH<sub>3</sub>-H<sub>4</sub>F to a final concentration of 2 mM, with a specific radioactivity of 136.9 kBq/ $\mu$ mol. Samples (25  $\mu$ l) were withdrawn at 1-min intervals and injected into 250  $\mu$ l of the aforementioned buffer without DTE but with 10 mM sodium acetate as a carrier. After addition of  $125 \mu l$  of 2.2 M HClO<sub>4</sub> and then of 300  $\mu$ l of 1 M KOH to bring the pH to about 3, the samples were incubated at  $0^{\circ}$ C for 15 min and centrifuged. The supernatants were loaded onto 300  $\mu$ l of cation exchanger (Dowex <sup>50</sup> WX8, 200-400 mesh, H' form, equilibrated with water). Acetate was eluted by washing the columns three times with 0.5 ml of water.  $^{14}CH_{2}$ -H<sub>4</sub>F bound tightly to the cation exchanger at pH <sup>3</sup> and was eluted by washing the columns three times with 0.5 ml of <sup>1</sup> M KOH. The acetate fraction was alkalinized by addition of 0.25 ml of <sup>1</sup> M KOH. Radioactivity was determined by scintillation counting as described below.

Labeling experiments. Growth experiments with  $[{}^{14}C]$ methanol were performed in 16-ml rubber-stoppered Hungate tubes containing 5 ml of medium under  $H_2$ -CO<sub>2</sub> (80:20) pressurized to 2 atm (203 kPa) or under  $N_2$ -CO<sub>2</sub> (80:20) (101 kPa). Since phosphate disturbs the quantification of  ${}^{14}CO<sub>2</sub>$ (see below), the potassium phosphate concentration was decreased to 4 mM. [<sup>14</sup>C]methanol (specific radioactivity,  $160.6$  Bq/ $\mu$ mol) was added to a final concentration of 46 mM. Immediately after inoculation, a sample (1 ml) was removed and mixed with 0.2 ml of stop mix (0.2 M NaOH, 0.1 M sodium acetate, 0.4 M  $Na_2CO_3$ ). Cells were removed by centrifugation, and the supernatant was stored at  $-20^{\circ}$ C. The cultures were shaken at 30'C in <sup>a</sup> water bath. When growth was complete, 0.8 ml of stop mix was added by syringe, resulting in <sup>a</sup> pH of about 10. The tubes were shaken vigorously in order to absorb the  $CO<sub>2</sub>$  from the gas phase as carbonate. Cells were pelleted by centrifugation, and the supernatants from the beginning and the end of the growth experiments were tested for [<sup>14</sup>C]methanol, [<sup>14</sup>C]acetate, and  ${}^{14}CO_2$ .

For the determination of the total radioactivity added (100%), 0.3 ml of each sample was mixed with 5 ml of scintillation fluid (Opti-Fluor; Canberra Packard, Frankfurt/ Main, Germany), and the radioactivity was determined in a scintillation counter (Tri-Carb; Canberra Packard). Another 0.3-ml aliquot of each sample was mixed with 0.6 ml of 0.15 M Ba(OH)<sub>2</sub> solution to precipitate <sup>14</sup>CO<sub>2</sub> as Ba<sup>14</sup>CO<sub>3</sub>. After centrifugation, the supernatants were carefully removed, and the pellets were resuspended in  $1 \text{ ml of } 0.1 \text{ M BaCl}_2$ . Centrifugation and removal of the supernatants were repeated, and the  $Ba^{14}CO_3$  pellets were resuspended in 0.2 ml of 0.1 M BaCl<sub>2</sub>. Radioactivity was determined by scintillation counting. To verify that the material precipitated with  $Ba^{2+}$  was  $BaCO<sub>3</sub>$ , it was resuspended in water and placed in <sup>a</sup> rubber-stoppered tube. Subsequently, <sup>2</sup> M HCl was added by syringe until the precipitate was dissolved (about 0.5 ml). The released  $CO<sub>2</sub>$  was carried by a stream of air via rubber tubing to a vial containing 4 ml of Carbo-Sorb (Packard, Frankfurt, Germany). By this method, 98% of the radioactivity was recovered in the Carbo-Sorb fraction by scintillation counting, demonstrating that the material precipitated with Ba<sup>2+</sup> was actually <sup>14</sup>CO<sub>3</sub><sup>2-</sup>.

The supernatants from the first centrifugation step were loaded onto Dowex anion-exchange columns equilibrated with <sup>2</sup> M NaOH and then washed with water until the pH of the eluent was  $\lt 9$  (Dowex 1 $\times 8$ , 200-400 mesh; bed volume, 0.3 ml; Serva). The columns were washed three times with 0.5 ml of distilled water. The radioactive compound which eluted in these fractions was assumed to be [<sup>14</sup>C]methanol because the corresponding standard eluted in the same fraction. The radioactivity was quantified by scintillation counting. [14C]acetate bound tightly to Dowex type <sup>1</sup> and

was eluted by washing three times with 0.3 ml of 0.6 M KI. The pooled KI fractions were mixed with <sup>1</sup> ml of distilled water, and the radioactivity was determined by scintillation counting. The efficiency of this separation method was determined by adding known amounts of  $NaH^{14}CO_3$ ,  $[$ <sup>14</sup>C]methanol, and  $[$ <sup>14</sup>C]acetate (alone and mixed) to the culture medium and processing as described above. The recovery of radioactivity was in the range of 94 to 98%.

Identification of Dowex-binding radioactivity as acetate. Dowex-1-binding radioactivity was identified as  $[{}^{14}C]$ acetate by its enzymatic conversion to  $[$ <sup>14</sup>C]acetylchloramphenicol via  $14^{\circ}$ C acetylphosphate and  $14^{\circ}$ C acetyl-CoA by the combined action of acetate kinase, phosphotransacetylase, and chloramphenicol acetyltransferase. An ATP-generating system was established by addition of phosphoenolpyruvate and pyruvate kinase. [14C]acetylchloramphenicol was specifically separated from  $[14\text{C}]$ acetate and  $[14\text{C}]$ acetyl-CoA by extraction with ethyl acetate (29). A 1-ml portion of the culture supernatant and a  $[14]$ C]acetate standard solution were separated over Dowex as described above. The acetate in the KI-eluted fractions was converted to acetylchloramphenicol as described before (29). The reaction mixture contained, in <sup>a</sup> total volume of 0.5 ml, <sup>100</sup> mM Tris-HCl (pH 7.8), 6 mM  $MgCl<sub>2</sub>$ , 3 mM ATP, 20 mM phosphoenolpyruvate, 2.6 mM CoA, 4.5 mM chloramphenicol, <sup>10</sup> U of acetate kinase, <sup>20</sup> U of pyruvate kinase, <sup>10</sup> U of phosphotransacetylase, <sup>30</sup> U of chloramphenicol acetyltransferase, and 2.4 to <sup>3</sup> mM [14C]acetate (about 12,000 dpm). A negative control received  $H_2O$  instead of chloramphenicol acetyltransferase. After 11 h of incubation at 25°C, acetylchloramphenicol was extracted with 0.7 ml of ethyl acetate by vortexing for <sup>1</sup> min followed by centrifugation for phase separation. The extraction was repeated four times, resulting in recovery of 70 to 80% of the total radioactivity in the pooled ethyl acetate extracts. In the absence of chloramphenicol acetyltransferase, only 0.8% of the radioactivity was detectable in the ethyl acetate fraction. The pooled ethyl acetate fractions were dried in an air stream, taken up in  $40 \mu$ l of ethyl acetate, and spotted onto a silica gel thin-layer sheet (Merck, Darmstadt, Germany). The chromatogram was developed in ascending fashion with water-saturated ethyl acetate, dried, and subjected to autoradiography. As a result, all spots appearing could be identified by their  $R_f$  values (28) as 1-acetoxy chloramphenicol, 3-acetoxy chloramphenicol, or 1,3-diacetoxy chloramphenicol, demonstrating that the fractions eluted from Dowex contained  $[$ <sup>14</sup>C]acetate.

SDS-PAGE. Extracts for SDS gel electrophoresis were prepared as described before (19) except that aerobic conditions were employed and 0.04 M Tris-HCl buffer (pH 7.1) was used instead of potassium phosphate buffer. SDSpolyacrylamide gel electrophoresis (PAGE) was performed as described before (20) with 12% (wt/vol) polyacrylamide gels (7 by <sup>15</sup> by 0.1 cm) which were run at <sup>25</sup> mA for about 1 h. Standard proteins of low molecular weight were from Pharmacia (Freiburg, Germany). Proteins were stained with Coomassie brilliant blue (32). Proteins to be stained for heme after separation by SDS-PAGE were prepared and stained as described before (31) except that the SDS concentration in the sample preparation buffer was increased to 8%.

Determination of acetate and protein. Acetate content was determined enzymatically by coupling acetyl-CoA formation by acetyl-CoA synthetase via myokinase and pyruvate kinase with the NADH-dependent reduction of pyruvate to lactate (6). Protein was determined by the method of Lowry et al. (22). Bovine serum albumin served as a standard.



FIG. 1.  $Na<sub>2</sub>S<sub>2</sub>O<sub>4</sub>$  reduced minus  $O<sub>2</sub>$  oxidized difference spectra of the washed membrane fractions from two strains of S. sphaeroides after growth on betaine plus formate. Wild-type (a) and BK824 (b), each contained 10.8 mg of protein per ml in <sup>50</sup> mM potassium phosphate buffer (pH 7.0). The peaks at 555 nm ( $\alpha$ -band), 532 and 526 nm ( $\beta$ -band), and 429 nm ( $\gamma$ -band) in spectrum a demonstrate the presence of cytochromes.

# RESULTS AND DISCUSSION

Isolation of a cytochrome-deficient mutant of S. sphaeroides. In order to obtain a clue about the role of cytochromes in the metabolism of S. sphaeroides, the organism was mutagenized with UV light, and cytochrome-deficient mutants were isolated. It was previously found that the cytochrome content of S. ovata is 11 times higher during growth on betaine than during growth on betaine plus formate, suggesting a role of these electron carriers in the disproportionation of betaine (19). Mutant enrichment was based on the assumption that cytochromes are needed for the conversion of betaine alone but not for the conversion of betaine in the presence of formate. Therefore, S. sphaeroides cells mutagenized with UV light were first grown on betaine plus formate and subsequently on betaine in the presence of ampicillin to select for mutants not capable of growth on betaine alone. Mutants were detected by testing for the presence of heme after plating the mutagenized cells on agar plates with betaine plus formate as the substrate. For this purpose, a new test was developed which allowed the detection of heme-free mutant colonies on nylon filters by applying a heme stain. (Details are given in Materials and Methods.) This procedure led to isolation of the S. sphaeroides mutant BK824, which was used for all further experiments.

Characteristics of the mutant strain. Membranes isolated from the mutant strain were tested for the presence of cytochromes by redox difference spectroscopy. The spectrum in Fig. <sup>1</sup> shows that the mutant membranes were devoid of cytochromes. This is in agreement with the negative heme stain behavior of the mutant. SDS gels of crude extracts and membrane proteins from the wild type and the mutant showed the same protein pattern, but only the





<sup>a</sup> Both strains were grown first on betaine plus formate and then transferred to fresh medium containing the substrate(s) listed. Growth was monitored photometrically at 600 nm, and the final optical density is given. Uninoculated medium served as a blank. Without substrate, the final  $OD<sub>600</sub>$  was 0.

Formate in the absence of betaine is a very poor growth substrate (24).  $c$  ND, not determined.

membrane proteins of the wild type displayed bands when stained for the presence of heme (not shown).

From these results, it was assumed that the mutant strain was not capable of synthesizing heme. In contrast to this idea, addition of heme or 8-aminolevulinic acid to the growth medium did not result in reversion of the observed phenotypical defects of the mutant. However, this finding does not exclude a defect in heme synthesis, since it has to be kept in mind that *Escherichia coli* or Salmonella typhimurium mutants defective in heme synthesis are unable to take up heme from the medium unless secondary mutations are introduced that allow heme to enter the cells (17, 23).

It was now of interest to find out which catabolic functions were affected by this mutation. Therefore, as a first approach, the ability of the mutant strain to grow on a number of substrates was compared with that of the wild type. Table <sup>1</sup> shows that growth on betaine was impaired in the mutant but not in the wild type, whereas betaine utilization in the presence of formate was equally good in the wild-type and the mutant strain. No or only little growth of the mutant was observed on  $H_2$  plus  $CO_2$ , betaine plus methanol, or methanol plus  $CO<sub>2</sub>$ . In contrast, growth of the mutant on acetoin, betaine plus  $\overline{H}_2$  and  $CO_2$ , betaine plus pyruvate, or methanol plus  $H_2$  and  $CO_2$  could not be distinguished from that of the wild type. These results suggest that the mutation does not affect betaine reduction itself but methyl group oxidation. This idea was substantiated by the finding that  $H<sub>2</sub>$ , formate, and pyruvate could serve as electron donors for the reductive betaine cleavage to trimethylamine and acetate (Fig. 2), whereas methanol or betaine could not.

From the recent demonstration of the Wood pathway enzymes in S. *sphaeroides* (19), it is reasonable to assume that oxidation of methanol and betaine involves the formation of methyl- $H_4F$  and its subsequent oxidation to  $CO<sub>2</sub>$  (Fig. 2) (21, 27). That the mutation affects a reaction associated with this pathway is supported by the inability of the mutant to grow on  $H_2$  plus  $CO_2$ . The absolute dependence on methanol for growth on  $H_2$  plus  $CO_2$  points to the inability of the mutant to form the methyl moiety of acetate from  $CO<sub>2</sub>$ ; therefore, when an external methyl group donor such as methanol was supplied, acetate was formed. Acetogenesis from betaine or  $H_2$  plus  $CO_2$  involves reactions of the Wood pathway, although in opposite directions (Fig. 2). From the ability of the mutant to grow on formate plus betaine or  $H<sub>2</sub>$ 



FIG. 2. Scheme depicting the central role of the Wood pathway for the dismutation of betaine (left) and acetogenesis from  $H_2$  plus  $CO<sub>2</sub>$  (right) as performed by S. sphaeroides. Boldface letters indicate the  $C_1$  moieties of the Wood pathway intermediates. 1, Formate dehydrogenase; 2, formyl-H4F synthetase (ATP dependent); 3, methenyl-H<sub>4</sub>F cyclohydrolase; 4, methylene-H<sub>4</sub>F dehydrogenase; 5, methylene-H4F reductase; 6, CO dehydrogenase; 7, phosphotransacetylase; 8, acetate kinase; 9, betaine reductase; 10, hydrogenase; 11, methyltransferase and corrinoid-FeS protein. CoA-SH, coenzyme A;  $CH_3$ -Co~SCoA, acetyl coenzyme A.

plus betaine, it must be concluded that the oxidation of neither formate nor  $H_2$  is affected by the mutation.

To further substantiate these conclusions, wild-type and mutant cells were grown on  $[^{44}C]$ methanol under N<sub>2</sub>-CO<sub>2</sub> (80:20) or under  $\text{H}_2\text{-}\text{CO}_2$  (80:20), and the fate of the label was determined. It is evident from experiments 1 to 4 in Table 2 that 73 to 99% of the label was converted to  $14^{\circ}CO$ , and  $1^{14}$ C acetate, with a  $1^{14}$ C acetate/ $1^{4}$ CO<sub>2</sub> ratio of 1.3 to 1.4. It can be assumed that the majority of the methanol was converted directly into the methyl moiety of acetate by means of a specific methyl group transfer (30). Since methanol has to be oxidized to  $CO<sub>2</sub>$  in order to gain reducing power for the generation of the carboxyl group of acetate, a small portion of the label was probably also incorporated into the carboxyl group of acetate. With  $H<sub>2</sub>$  as the electron donor (Table 2, experiments <sup>5</sup> to 8), much less methanol was oxidized to  $CO<sub>2</sub>$ . Only 55 to 71% of the methanol was converted, 46 to 52% being incorporated into acetate and 4 to 9% being oxidized to  $CO<sub>2</sub>$ .

The situation encountered in growing cultures of mutant strain BK824 was completely different. A negligible proportion (<1%) of the label was recovered in the  ${}^{14}CO_2$  fraction under both  $N_2$ -CO<sub>2</sub> (Table 2, experiments 9 to 12) and  $H_2$ -CO<sub>2</sub> (Table 2, experiments 13 to 15). This shows that the mutant cells were not capable of oxidizing methanol to  $CO<sub>2</sub>$  and is in agreement with the very little growth on methanol plus  $CO<sub>2</sub>$  (Table 1) and with the observation that the majority of the added methanol was not converted, the rest being converted almost exclusively to acetate (Table 2, experiments 9 to 12).

That some  $\int_0^{14}$ C acetate could be formed may be explained as follows. Since methanol was obviously not oxidized, it has to be concluded that it is converted into the methyl group of acetate and that the reducing equivalents for the reduction of  $CO<sub>2</sub>$  to the carboxyl group of acetate were derived either from some storage material or from the yeast extract and casamino acids present in the growth medium. Under  $H_2$ plus  $CO<sub>2</sub>$ , reducing equivalents for the reduction of  $CO<sub>2</sub>$  to the carboxyl group of acetate can be generated from  $H_2$  (see Fig. 2) (Table 2, experiments 13 and 14). This agrees with the finding that S. sphaeroides BK824 is not able to grow on either methanol or  $H_2$  plus  $CO_2$  but does grow on the combination of these substrates (Table 1). Interestingly, in experiments performed with resting cells, free CO could substitute for  $H_2$  plus  $CO_2$  when methanol was present (data not shown). In conclusion, these experiments show that methanol can serve as a methyl group donor in acetate synthesis but not as an electron donor in the mutant.

The ability of the cytochrome-deficient mutant to form acetate from  $H_2$  plus  $CO_2$  and methanol implies that cytochromes are not required for the reduction of  $CO<sub>2</sub>$  to the carboxyl carbon of acetate, as catalyzed by CO dehydrogenase. This is in contrast to experiments performed with vesicles of Clostridium thermoautotrophicum, which suggest the involvement of b-type cytochromes in the reduction of  $CO<sub>2</sub>$  to  $CO$  and vice versa (14).

Enzyme activities in the mutant. From these results, only the following reactions could theoretically be affected by the mutation: formyl- $H_4F$  synthetase, methenyl- $H_4F$  cyclohydrolase, methylene- $H_4F$  dehydrogenase, methylene- $H_4F$  reductase, methyltransferase, or an electron transfer reaction connected to these reactions.

Therefore, the activity of the enzymes of the acetyl-CoA pathway and of enzymes associated with this pathway were tested in crude extracts of both the wild type and the mutant strain. It is evident from Table 3 that there was no marked difference in the activity of the Wood pathway enzymes between the wild-type and the mutant strains of S. sphaeroides. The differences encountered are normal variations in the measurements; they cannot explain the inability of the mutant to grow on certain substrates. Acetogenesis from  $CH<sub>3</sub>-H<sub>4</sub>F$  and CO by crude extracts occurred only at low specific activities (Table 3). However, since low in vitro rates of acetate formation are a general phenomenon observed in crude extracts (8), the observed activities are considered significant.

From these data, formyl- $H_4F$  synthetase, methenyl- $H_4F$ cyclohydrolase, methylene- $H_4F$  dehydrogenase (Fig. 2), and the synthesis of acetate from  $\text{CH}_3\text{-}\text{H}_4\text{F}$  and CO are considered to be cytochrome-independent activities, since these activities with physiological substrates were detectable in comparable amounts in both the wild type and the cytochrome-deficient mutant.

Furthermore, our results allow conclusions about the hydrogen metabolism of S. sphaeroides. Acetogenesis from  $H<sub>2</sub>$  plus CO<sub>2</sub> necessitates an electron transfer from H<sub>2</sub> to  $NAD(P)$  as well as to methylene- $H_4F$  to generate the methyl moiety of acetate. In addition,  $CO<sub>2</sub>$  has to be reduced to the carboxyl group of acetate (Fig. 2). However, information on the enzymes involved in these electron transfer reactions is very limited. In case of  $H_2$ -dependent NAD(P) reduction,





<sup>a</sup> Immediately after inoculation, radioactivity was found exclusively in the methanol fraction (data not shown). At the end of the experiments, the radioactivity in an aliquot of alkalized cell culture supernatant was arbitrarily set to 100%, and the radioactivity recovered in the various fractions is given as the percentage of that value.

<sup>*o*</sup> Calculated from <sup>14</sup>CH<sub>3</sub>OH added (100%) minus <sup>14</sup>CH<sub>3</sub>OH unconverted.<br><sup>*c*</sup> Sum of [<sup>14</sup>C]acetate, <sup>14</sup>CO<sub>2</sub>, and <sup>14</sup>CH<sub>3</sub>OH unconverted.

the information available suggests the involvement of a ferredoxin-reactive hydrogenase, ferredoxin, and ferredoxin:NAD(P) oxidoreductase in acetogenic bacteria (21). Since extracts of both the wild-type and mutant strains of S. sphaeroides catalyze the  $H_2$ -dependent reduction of NAD(P) (Table 3), it has to be concluded that this electron transfer is not affected by the mutation and therefore does not involve

cytochromes. This is in contrast to C. thermoautotrophicum, in which b-type cytochromes have been shown to act as primary electron acceptors of a periplasmic hydrogenase (15). The reduction of  $CO<sub>2</sub>$  by  $H<sub>2</sub>$  to CO is likewise not affected by the mutation, as clearly shown by the growth experiments with [14C]methanol (Table 2).

Nothing is known about the physiological electron donor

TABLE 3. Comparison of various enzyme activities in cell extracts of S. sphaeroides wild type and BK824 after growth on betaine plus formate

Enzyme activity	Electron acceptor or substrate	Sp act ( $\mu$ mol min <sup>-1</sup> [mg of protein] <sup>-1</sup> )	
		Wild type	<b>BK 824</b>
CO dehydrogenase	Benzyl viologen	1.0	2.0
Hydrogenase	Benzyl viologen <b>NAD</b> <b>NADP</b>	15.6 0.09 0.21	15.0 0.12 0.96
Formate dehydrogenase	Benzyl viologen <b>NAD</b> <b>NADP</b>	0.39 0.02 0.11	0.61 0.02 0.10
Formyl- $H_4F$ synthetase		0.44	0.15
Methenyl- $H_4F$ cyclohydrolase		0.10	0.13
Methylene-H <sub>4</sub> F dehydrogenase	<b>NAD</b> <b>NADP</b>	$\bf{0}$ 1.47	0 1.22
Methylene- $H_4F$ reductase	Benzyl viologen <b>NAD</b> <b>NADP</b>	0.76 0 $\bf{0}$	0.70 0 $\mathbf{0}$
Acetogenesis	$CH3-H4F + CO$	0.0016	0.001

in the methylene- $H_4F$  reductase reaction during acetogenesis from  $H_2$  plus  $CO_2$  in S. sphaeroides and in other acetogenic bacteria, the only exception being Peptostreptococcus productus, in which NADH was shown to be the physiological electron donor (33). The methylene- $H_4F$  reductase has been proposed to be involved in energy generation in acetogens (8). The membrane location of this enzyme in C. thermoautotrophicum and in P. productus is in favor of this suggestion (14, 33). Since the extracts of S. sphaeroides were prepared by French pressing, it is not surprising that the majority of the methylene- $H_4F$  reductase was recovered from the soluble fraction (data not shown); this is also observed in C. thermoautotrophicum and in P. productus. It is most noteworthy that the benzyl viologendependent methylene-H4F reductase activity was present in both the wild type and the cytochrome-deficient mutant of S. sphaeroides. Therefore, the enzyme itself cannot be affected by the cytochrome deficiency. However, it is conceivable that cytochromes serve as primary physiological electron donors in this reaction. This hypothesis is in very good agreement with the observed features of the cytochromedeficient mutant strain. Future experiments will have to show whether this is really the case.

It is not presently understood how acetogenic bacteria gain net ATP during acetogenesis from  $H_2$  plus  $CO<sub>2</sub>$  (8). However, there seems to be a general agreement that net ATP synthesis needs the generation of <sup>a</sup> transmembrane gradient of protons or sodium ions (8, 15, 25). The reaction of the acetogenic pathway coupled to the extrusion of Na' has not yet been identified, but experiments performed with Acetobacterium woodii and P. productus indicate that either methylene-H4F reduction or the formation of the methyl group of acetyl-CoA from methyl- $H_4F$  has to be involved (9, 11). The methyl group transfer from methyltetrahydromethanopterin to coenzyme M was recently shown in Methanosarcina strain Göl to be involved in the generation of a transmembrane sodium ion potential (1). It is conceivable that an analogous reaction exists in acetogenic bacteria that depend on  $Na<sup>+</sup>$  for growth and acetogenesis from  $H<sub>2</sub>$  plus  $CO<sub>2</sub>$ .

 $\tilde{Na}^+$  dependence has been demonstrated for A. woodii, P. productus, and Acetogenium kivui (9, 11, 34), and it seems possible that these organisms generate a primary  $Na<sup>+</sup>$  gradient which is taken advantage of for ATP synthesis via an Na<sup>+</sup>-dependent ATP synthase. In agreement with this notion, Heise et al. reported the presence of an  $Na<sup>+</sup>$ -translocating ATPase in  $\overline{A}$ . woodii (12). Autotrophic growth of a second group of acetogenic bacteria (e.g., C. thermoaceticum and Sporomusa species) seems to be independent of  $Na<sup>+</sup>$  (34; unpublished results) and to require cytochromes for autotrophic growth (16; this study). Cytochromes were suggested to participate in electron transport from various electron donors to methylene- $H_4F$  in C. thermoautotrophicum (16). Our study clearly shows that cytochromes are absolutely required in S. sphaeroides for autotrophic growth and methyl group oxidation. The cytochrome-deficient mutant described here will give us new insights into the role of these electron carriers in this group of acetogenic bacteria.

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