

## Ammonia Assimilation Pathways in Nitrogen-Fixing *Clostridium kluverii* and *Clostridium butyricum*†

KEIKO KANAMORI,<sup>1,2\*</sup> RICHARD L. WEISS,<sup>1</sup> AND JOHN D. ROBERTS<sup>2</sup>

Department of Chemistry and Biochemistry, University of California, Los Angeles, California 90024-1569,<sup>1</sup> and Gates and Crellin Laboratories of Chemistry, California Institute of Technology, Pasadena, California 91125<sup>2</sup>

Received 12 September 1988/Accepted 3 January 1989

Pathways of ammonia assimilation into glutamic acid were investigated in ammonia-grown and N<sub>2</sub>-fixing *Clostridium kluverii* and *Clostridium butyricum* by measuring the specific activities of glutamate dehydrogenase, glutamine synthetase, and glutamate synthase. *C. kluverii* had NADPH-glutamate dehydrogenase with a  $K_m$  of 12.0 mM for NH<sub>4</sub><sup>+</sup>. The glutamate dehydrogenase pathway played an important role in ammonia assimilation in ammonia-grown cells but was found to play a minor role relative to that of the glutamine synthetase/NADPH-glutamate synthase pathway in nitrogen-fixing cells when the intracellular NH<sub>4</sub><sup>+</sup> concentration and the low affinity of the enzyme for NH<sub>4</sub><sup>+</sup> were taken into account. In *C. butyricum* grown on glucose-salt medium with ammonia or N<sub>2</sub> as the nitrogen source, glutamate dehydrogenase activity was undetectable, and the glutamine synthetase/NADH-glutamate synthase pathway was the predominant pathway of ammonia assimilation. Under these growth conditions, *C. butyricum* also lacked the activity of glucose-6-phosphate dehydrogenase, which catalyzes the regeneration of NADPH from NADP<sup>+</sup>. However, high activities of glucose-6-phosphate dehydrogenase as well as of NADPH-glutamate dehydrogenase with a  $K_m$  of 2.8 mM for NH<sub>4</sub><sup>+</sup> were present in *C. butyricum* after growth on complex nitrogen and carbon sources. The ammonia-assimilating pathway of N<sub>2</sub>-fixing *C. butyricum*, which differs from that of the previously studied *Bacillus polymyxa* and *Bacillus macerans*, is discussed in relation to possible effects of the availability of ATP and of NADPH on ammonia-assimilating pathways.

The N<sub>2</sub>-fixing prokaryotes studied to date fall into two groups with respect to the pathways of ammonia assimilation; (i) *Bacillus polymyxa* and *Bacillus macerans*, which have glutamate dehydrogenases (GDHs) with  $K_m$ s for ammonia of 2.2 to 2.9 mM and utilize the GDH pathway during N<sub>2</sub> fixation (13, 14), and (ii) *Bacillus azotofixans* (15) and other prokaryotes such as *Clostridium pasteurianum* (3, 20, 21) and *Klebsiella pneumoniae* (16, 21), which utilize the glutamine synthetase (GS)-glutamate synthase (GOGAT) pathway because they either have barely detectable levels of GDH even in ammonia-rich medium (3, 20) or else GDH with a high  $K_m$  for NH<sub>4</sub><sup>+</sup> (16). Our results with *Bacillus* spp. suggest that, for N<sub>2</sub>-fixing prokaryotes capable of synthesizing GDH with a moderate affinity for NH<sub>4</sub><sup>+</sup>, the GDH pathway may be more advantageous than the ATP-requiring GS-GOGAT pathway for assimilating ammonia during the energy-demanding process of nitrogen fixation (14). To determine whether the GDH pathway plays an important role in ammonia assimilation during N<sub>2</sub> fixation in genera other than *Bacillus*, we studied the pathways of ammonia assimilation in two nitrogen-fixing species of obligate anaerobes, *Clostridium kluverii* and *Clostridium butyricum* (25). *C. kluverii*, which utilizes ethanol, acetate, and bicarbonate as carbon sources, has GDH activity (27) when grown with NH<sub>4</sub><sup>+</sup> as the nitrogen source, although its  $K_m$  for NH<sub>4</sub><sup>+</sup> has not been measured and its coenzyme specificity has been controversial (10, 27). *C. butyricum* grown on complex medium (11) possesses GDH, and measurements of the NADH- and NADPH-dependent GDH activities in crude extracts as a function of NH<sub>4</sub><sup>+</sup> concentration suggest that the enzyme(s) have moderate affinity for NH<sub>4</sub><sup>+</sup> (2).

Studies of NAD(P)H-dependent enzymes such as GDH and GOGAT in the *Clostridium* species have been hampered by the presence in cell extracts of NADH oxidase and NAD(P)H-ferredoxin reductases, which cause very rapid oxidation of NAD(P)H. As a result, measurements of the specific activities and  $K_m$ s for substrates of GDH and GOGAT through the standard spectrophotometric method of observing the rate of oxidation of NAD(P)H are very difficult. Assays under anaerobic conditions (8) and/or through detection of [<sup>14</sup>C]glutamate formed from <sup>14</sup>C-labeled substrates (27) have been reported. However, <sup>15</sup>N nuclear magnetic resonance (NMR) spectroscopy provides an especially convenient method of measuring the specific activities of these enzymes through observation of the time-dependent formation of the product [<sup>15</sup>N]glutamic acid from <sup>15</sup>NH<sub>4</sub><sup>+</sup> (GDH) or from [<sup>15</sup>N]glutamine (GOGAT) without separation from the substrates in assay solutions containing excess NAD(P)H or an NADH-regenerating system. We report here an investigation of the pathways of ammonia assimilation in ammonia-grown and N<sub>2</sub>-fixing *C. kluverii* and *C. butyricum* through measurements of the specific activities of GDH, GS, and GOGAT.

### MATERIALS AND METHODS

**Strains, media, and growth.** *C. kluverii* (ATCC 8527) and *C. butyricum* (ATCC 8260) were obtained from the American Type Culture Collection. For *C. kluverii*, the medium (for large-scale culture) described by Stadtman and Burton (26) was used with 22 mM NH<sub>4</sub>Cl as the nitrogen source. Growth was monitored in a Klett-Summerson colorimeter with a no. 540 filter. Cells were grown from an inoculum of 2 to 3 Klett units to the midexponential phase (120 Klett units) in a closed vessel at 30°C. For N<sub>2</sub> fixation, the same medium without NH<sub>4</sub>Cl was supplemented with FeSO<sub>4</sub> · 7H<sub>2</sub>O (85 mg/liter) and Na<sub>2</sub>MoO<sub>4</sub> · 2H<sub>2</sub>O (30 mg/liter). A

\* Corresponding author.

† This article is contribution no. 7845 from the Gates and Crellin Laboratories of Chemistry.

10% inoculum of culture grown in limited (5 mM)  $\text{NH}_4\text{Cl}$  was added to the N-free medium and grown from 10 Klett units to 60 Klett units at 37°C with a continuous flow (200 ml/min) of  $\text{N}_2$  containing 3%  $\text{CO}_2$ . A continuous supply of  $\text{CO}_2$  was found to be essential for growth under this condition. When an uninoculated medium was sparged with  $\text{N}_2$  alone, the concentration of  $\text{HCO}_3^-$ , an essential carbon source for *C. kluyverii* supplied at 20 mM in the medium, was found to decrease to one-half within 3 h (as measured by decrease in the radioactivity of  $\text{H}^{14}\text{CO}_3^-$  added at 0.1  $\mu\text{Ci/ml}$  of medium) as the result of  $\text{CO}_2$  loss by the reaction  $\text{HCO}_3^- + \text{H}^+ \rightleftharpoons \text{H}_2\text{O} + \text{CO}_2 \uparrow$ .

*C. butyricum* was maintained in reinforced *Clostridium* medium (Difco Laboratories) containing (per liter) tryptose (10 g), beef extract (10 g), yeast extract (3 g), dextrose (5 g), NaCl (5 g), starch (1 g), cysteine hydrochloride (0.5 g), sodium acetate (3 g), and Bacto-Agar (Difco) (0.5 g). For growth on  $\text{NH}_4\text{Cl}$  (22 mM) as the nitrogen source, the medium described by Carnahan and Castle (6) was used with the following modifications;  $\text{CaCl}_2 \cdot \text{H}_2\text{O}$  (133 mg/liter) was substituted for  $\text{CaCO}_3$ , glucose (20 g/liter) was substituted for sucrose, and  $\text{K}_2\text{HPO}_4\text{-KH}_2\text{PO}_4$  buffer was used at 0.1 M. For  $\text{N}_2$  fixation, the nitrogen-free medium was supplemented with  $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$  (85 mg/liter) and  $\text{Na}_2\text{MoO}_4 \cdot 2\text{H}_2\text{O}$  (20 mg/liter). Cultures were grown anaerobically with a continuous flow of  $\text{N}_2$  at 37°C from a 1.5% inoculum of ammonia-grown cells to 195 Klett units for large-scale growth on  $\text{NH}_4^+$  and from a 6% inoculum of  $\text{N}_2$ -fixing cells to 107 Klett units for large-scale growth on  $\text{N}_2$ . The purity of the culture was checked periodically by microscopic observation of smears stained with safranin.

**Enzyme assays.** The cells were harvested by centrifugation at  $14,680 \times g$  and washed in 0.1 M Tris hydrochloride buffer (pH 7.4) containing 1 mM cysteine (*C. kluyverii*) or in 50 mM  $\text{KH}_2\text{PO}_4$  buffer (pH 7.0) containing 0.004% sodium thioglycollate (*C. butyricum*). The cell pellet was either used immediately or stored at  $-20^\circ\text{C}$  until use. For preparation of cell extracts for GDH and GOGAT assays, the cell pellet was suspended in the same buffer (supplemented with 20 mM mercaptoethanol for *C. kluyverii*) and disrupted by sonication at 4°C as described previously (13). For GS and glucose-6-phosphate dehydrogenase (G6PD) assays, the buffers used for preparation of cell extracts were 0.2 M  $\text{KH}_2\text{PO}_4$  (pH 7.5) and 5 mM Tris hydrochloride (pH 7.6) containing 10 mM  $\text{MgCl}_2$ , respectively. The cell debris was removed by centrifugation at  $36,600 \times g$ , and the supernatant containing 5 to 20 mg of protein was used for enzyme assays. Protein was measured by the method of Lowry et al. (19) with bovine serum albumin as the standard. All enzyme assays were performed at 20°C and completed within 1 or 2 h after preparation of the cell extracts.

GS activity was measured by a modification of the radiochemical method of Prusiner and Milner (24) as described previously (13). The assay solution, modified on the basis of the known properties of the GS of *C. pasteurianum* (18), contained the following in a final volume of 1.1 ml: 50 mM 3-(*N*-morpholino)propanesulfonic acid buffer (pH 7.2), 12 mM  $\text{NH}_4\text{Cl}$ , 25 mM glutamic acid with L-[U- $^{14}\text{C}$ ]glutamic acid (1.72  $\mu\text{Ci/ml}$ ), 12 mM ATP, 24 mM  $\text{MgCl}_2$ , and 0.1 mM aminooxyacetate.

GOGAT activity was assayed by  $^{15}\text{N}$  NMR as described previously (14) with the following modifications. The assay solution contained 5 mM [ $\gamma$ - $^{15}\text{N}$ ]glutamine, 5 mM  $\alpha$ -ketoglutarate, and 7.5 mM NAD(P)H for *C. butyricum* and 15 mM NAD(P)H for *C. kluyverii*. The reaction was performed under  $\text{N}_2$  to minimize oxidation of NADH by endogenous

NADH oxidase in the cell extracts. For assay of NADPH-GDH activity, cell extracts were added to an assay solution containing 25 mM  $^{15}\text{NH}_4\text{Cl}$ , 5 mM  $\alpha$ -ketoglutarate, and 15 mM NADPH in 0.1 M Tris buffer (pH 7.0) and incubated under  $\text{N}_2$ . At various time intervals, a 2-ml sample of the reaction mixture was withdrawn anaerobically, and the reaction was terminated by acidification to pH 2.0. After removal of denatured protein by centrifugation and neutralization to pH 6 to 7, the extent of formation of [ $^{15}\text{N}$ ]glutamic acid was measured by  $^{15}\text{N}$  NMR. The amount of [ $^{15}\text{N}$ ]glutamic acid formed in the reaction mixture was calculated from its peak intensity in the NMR spectra by comparison with the peak intensity of a known amount of [ $^{15}\text{N}$ ]glutamic acid. For assay of NADH-GDH activity, the assay solution contained 7.5 mM NADH or an NADH-regenerating system consisting of 20 mM ethanol, 0.27 mM  $\text{NAD}^+$ , and alcohol dehydrogenase (2.8 U/ml) in Tris buffer at pH 7.8. The system generated NADH to a maximum concentration of approximately 0.15 mM as observed by UV absorbance in the assay solution before the addition of cell extracts. GDH and GOGAT activities are reported as milliunits (nanomoles of [ $^{15}\text{N}$ ]glutamic acid formed per minute) per milligram of protein.

The  $^{15}\text{N}$  NMR spectra were obtained at 50.68 MHz as described previously (13).  $^{15}\text{N}$  chemical shifts are reported in parts per million upfield from 1 M  $\text{H}^{15}\text{NO}_3$ .

When the spectrophotometric method was used to assay for GDH (for *C. butyricum* grown on complex medium), the assay solution described above was modified by decreasing the NAD(P)H concentration to 0.25 mM. The activity for reductive amination is reported as milliunits (nanomoles of NAD(P)H oxidized per min) per milligram of protein. For measurement of the  $K_m$  for glutamate of NADPH-GDH in the oxidative deamination reaction, the assay solution contained 0.3 mM  $\text{NADP}^+$  and L-glutamate at concentrations ranging over 5 to 50 mM in 0.1 M Tris buffer (pH 7.8). The rates of oxidation of NAD(P)H in the absence of substrates for GDH and GOGAT—the “background” oxidation—by cell extracts of *C. kluyverii* and *C. butyricum* were measured in 0.1 M Tris hydrochloride buffer (pH 7.0) containing 0.25 mM NAD(P)H. G6PD activity was measured spectrophotometrically by the standard procedure (29), except that the concentration of  $\text{NADP}^+$  in the assay solution was 0.3 mM. Activity is reported as milliunits (nanomoles of  $\text{NADP}^+$  reduced per minute) per milligram of protein.

Intracellular  $\text{NH}_4^+$  concentrations in *C. kluyverii* were determined by assaying for  $\text{NH}_4^+$  after perchlorate extraction and, for ammonia-grown cells, correcting for the amount of  $\text{NH}_4^+$  in the medium trapped in the cell pellet, as described previously (13).

## RESULTS

***C. kluyverii.*** Cell extracts of ammonia-grown *C. kluyverii* oxidized NADPH and NADH at the rates of  $1,246 \pm 323$  and  $3,000 \pm 138$  mU/mg of protein, respectively. This high background oxidation precluded spectrophotometric measurements of the GDH and GOGAT activities. The GDH activity was therefore measured through observation of the time-dependent formation of [ $^{15}\text{N}$ ]glutamic acid as described in Materials and Methods. Figure 1 shows representative  $^{15}\text{N}$  NMR spectra of the reaction mixtures for ammonia-grown cells (Fig. 1A through D) and  $\text{N}_2$ -fixing cells (Fig. 1E through G). Significant formation of [ $^{15}\text{N}$ ]glutamic acid (335.1 ppm) was observed in the presence of NADPH (Fig. 1A, B, E, and F) but not in the presence of 15 mM NADH

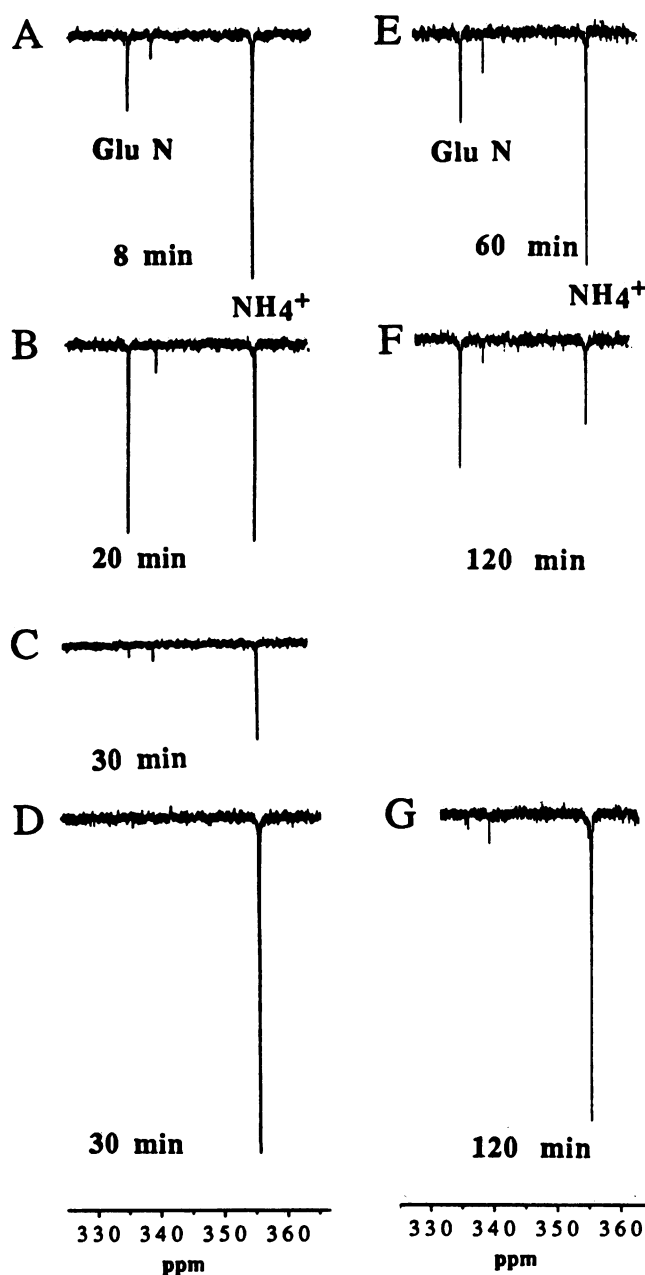


FIG. 1.  $^{15}\text{N}$  NMR spectra of GDH assay solutions at various time intervals after addition of cell extracts of ammonia-grown (A through D) and of  $\text{N}_2$ -fixing (E through G) *C. kluyverii*. Assay preparations contained 15 mM NADPH (A, B, E, F), 15 mM NADH (C), or no coenzyme (D, G). The small peak at 338 ppm represents the nitrogen in the Tris buffer.

(Fig. 1C) or in the absence of coenzyme (Fig. 1D and G). The GDH activity was also assayed in the presence of an NADH-regenerating system that regenerates NADH to a maximum concentration of 0.15 mM, because the activities of NADH-GDHs from many organisms have been shown to be inhibited at high NADH concentrations (9, 31). No  $^{15}\text{N}$ glutamic acid formation was observed (data not shown). The results show that the GDH of *C. kluyverii* is NADPH specific. Figure 2 shows the time-dependent formation of  $^{15}\text{N}$ glutamic acid calculated from the observed peak intensities in the NMR spectra to determine the specific

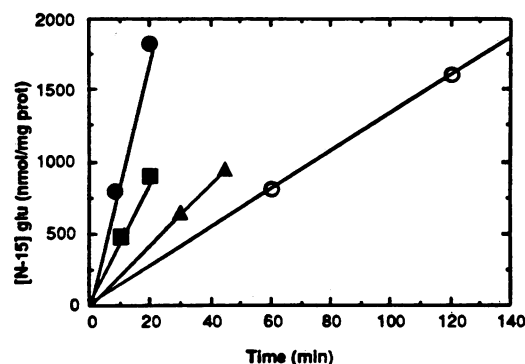


FIG. 2. Time-dependent formation of  $^{15}\text{N}$ glutamic acid after the addition of cell extracts of ammonia-grown *C. kluyverii* to GDH assay solutions with initial  $^{15}\text{NH}_4^+$  concentrations of 25 mM (●), 5 mM (■), and 2 mM (▲) and after addition of cell extracts of  $\text{N}_2$ -fixing cells to an assay solution with a  $^{15}\text{NH}_4^+$  concentration of 25 mM (○).

activity of NADPH-GDH. The activity for  $\text{NH}_4^+$ -grown *C. kluyverii* was 91 mU (nmol of  $^{15}\text{N}$ glutamic acid formed per min) per mg of protein when assayed with a  $^{15}\text{NH}_4^+$  concentration of 25 mM. To estimate the affinity of the enzyme for  $\text{NH}_4^+$ , the activity was measured at various concentrations of  $\text{NH}_4^+$ . At  $^{15}\text{NH}_4^+$  concentrations of 5 and 2 mM, the activity decreased to 47 and 20.5 mU/mg of protein, respectively (Fig. 2). An assay with a duplicate culture confirmed that the activity at the  $\text{NH}_4^+$  concentration of 5 mM is very close to one-half of that at 25 mM. Lineweaver-Burk plots of  $1/[\text{NH}_4^+]$  versus  $1/v_0$  gave a  $K_m^{app}$  of 12.0 mM for  $\text{NH}_4^+$ . For  $\text{N}_2$ -fixing *C. kluyverii*, the specific activity of NADPH-GDH, calculated from the data in Fig. 2, was 13.4 mU/mg of protein at a  $^{15}\text{NH}_4^+$  concentration of 25 mM.

In GOGAT assays performed with NAD(P)H as the coenzyme,  $^{15}\text{N}$ glutamic acid was formed only in the presence of NADPH. NADPH-GOGAT activities were 32 and 5.4 mU/mg of protein for ammonia-grown and  $\text{N}_2$ -fixing *C. kluyverii*, respectively.

Table 1 lists the specific activities of GDH, GS, and GOGAT and the average intracellular  $\text{NH}_4^+$  concentrations in ammonia-grown and  $\text{N}_2$ -fixing *C. kluyverii*. To estimate the actual contribution of the GDH pathway to ammonia assimilation in the cell, the affinity of GDH for  $\text{NH}_4^+$  and the intracellular  $\text{NH}_4^+$  concentration must be taken into account. In ammonia-grown cells, where the average intracellular  $\text{NH}_4^+$  concentration was 13.8 mM, the activity of GDH with a  $K_m$  of 12.0 mM for  $\text{NH}_4^+$  can be calculated to be 72 mU/mg of protein from the known activity at 25 mM  $\text{NH}_4^+$  and the Michaelis-Menten equation. Comparison with the observed GS and GOGAT activities shows that GDH is an important pathway of ammonia assimilation in ammonia-grown cells. In  $\text{N}_2$ -fixing cells, the in vitro GDH activity measured at an  $\text{NH}_4^+$  concentration of 25 mM (13.4 mU/mg of protein) was 2.5-fold higher than the GOGAT activity (5.4 mU/mg of protein). However, the average intracellular  $\text{NH}_4^+$  concentration was 0.99 mM. At such an  $\text{NH}_4^+$  concentration, the GDH activity can be calculated to be 11% of that at  $\text{NH}_4^+$  concentration of 25 mM. Therefore, the actual rate of ammonia assimilation by the GDH pathway in  $\text{N}_2$ -fixing cells is expected to be  $13.4 \times 0.11 = 1.5$  mU/mg of protein. Assuming that the GS and GOGAT of *C. kluyverii*, like those of other bacteria, have very low  $K_m$ s for their substrates and therefore can operate at the rates observed in the in vitro assay, 5 to 6 mU/mg of protein (Table 1), the

TABLE 1. Ammonia-assimilating enzymes in *C. kluyverii* and *C. butyricum* grown on different nitrogen sources

Species	Nitrogen source	Doubling time (h)	Intracellular NH <sub>4</sub> <sup>+</sup> concn (mM)	Sp act (mU/mg of protein)					
				GDH		GS	GOGAT		G6PD
				NADPH	NADH		NADPH	NADH	
<i>C. kluyverii</i>	NH <sub>4</sub> <sup>+</sup>	19 ± 1 <sup>a</sup>	13.8	91	ND <sup>b</sup>	23 ± 12	32	ND	
	N <sub>2</sub>	60 <sup>c</sup>	0.99	13.4	ND	5.9	5.4	ND	
<i>C. butyricum</i>	NH <sub>4</sub> <sup>+</sup>	2.4 ± 0.1 <sup>c</sup>		ND	ND	20 ± 3	ND	110	ND
	N <sub>2</sub>	15 ± 2 <sup>c</sup>		ND	ND	27	ND	45	ND
	Complex			207 ± 20	ND				152

<sup>a</sup> Growth at 30°C.<sup>b</sup> ND, Not detectable.<sup>c</sup> Growth at 37°C.

contribution of the GDH pathway to ammonia assimilation appears to be minor compared with that of the GS-GOGAT pathway in N<sub>2</sub>-fixing *C. kluyverii*.

***C. butyricum*.** (i) Cells grown in glucose-salt medium with ammonia or N<sub>2</sub>. The cell extracts of *C. butyricum* growing exponentially in glucose-salt medium with NH<sub>4</sub><sup>+</sup> or N<sub>2</sub> as the nitrogen source were found to oxidize NADPH and NADH at the rates of 30 to 50 and 425 to 1,750 mU/mg of protein, respectively. The GDH and GOGAT activities were therefore assayed through observation of [<sup>15</sup>N]glutamic acid formation. GDH activity was undetectable with NADPH or NADH as the coenzyme in either ammonia-grown or N<sub>2</sub>-fixing cultures. The results of the GOGAT assay for ammonia-grown *C. butyricum* are shown in Fig. 3. [<sup>15</sup>N]Glutamic acid was formed in the presence of NADH (Fig. 3A through D) but not in the presence of NADPH (Fig. 3E) or in the absence of coenzyme (Fig. 3F). The results show that the GOGAT of *C. butyricum* is NADH specific. Figure 4 shows the time-dependent formation of [<sup>15</sup>N]glutamic acid. From the initial rates of formation, NADH-GOGAT activities of 110 and 45 mU/mg of protein were obtained for ammonia-grown and N<sub>2</sub>-fixing *C. butyricum*, respectively. The specific activities of ammonia-assimilating enzymes summarized in Table 1 clearly show that the GS-GOGAT pathway is the predominant pathway of ammonia assimilation in both ammonia-grown and N<sub>2</sub>-fixing *C. butyricum* in glucose-salt medium.

(ii) Cells grown in complex medium. When a 2% inoculum of *C. butyricum* growing exponentially in reinforced *Clostridium* medium was added to the nitrogen-free glucose-salts medium and grown under N<sub>2</sub> flow, the culture showed an effective growth of 80 Klett units in 17 h but showed no further growth in the following 9 h. Because the reinforced *Clostridium* medium is very rich in complex carbon and nitrogen sources (see Materials and Methods), the observed growth was probably due to complex nitrogen sources carried over from the inoculum and the cessation of growth due to the lag required for induction of nitrogenase on exhaustion of the combined nitrogen sources. Surprisingly, cell extracts prepared from cells harvested just before the cessation of growth had NADPH-GDH activity of 207 ± 20 mU/mg of protein. No NADH-dependent GDH activity was observed. The activity was measured by the spectrophotometric method because the background oxidations of NADPH and NADH in the absence of substrates were low, at 33 and 65 mU/mg of protein, respectively. The NADPH-GDH activity remained stable in the cell extract at -20°C for more than 5 months. The detection of GDH activity in *C. butyricum* grown on complex medium is in agreement with the report of Bachofen and Herr (2), who detected GDH

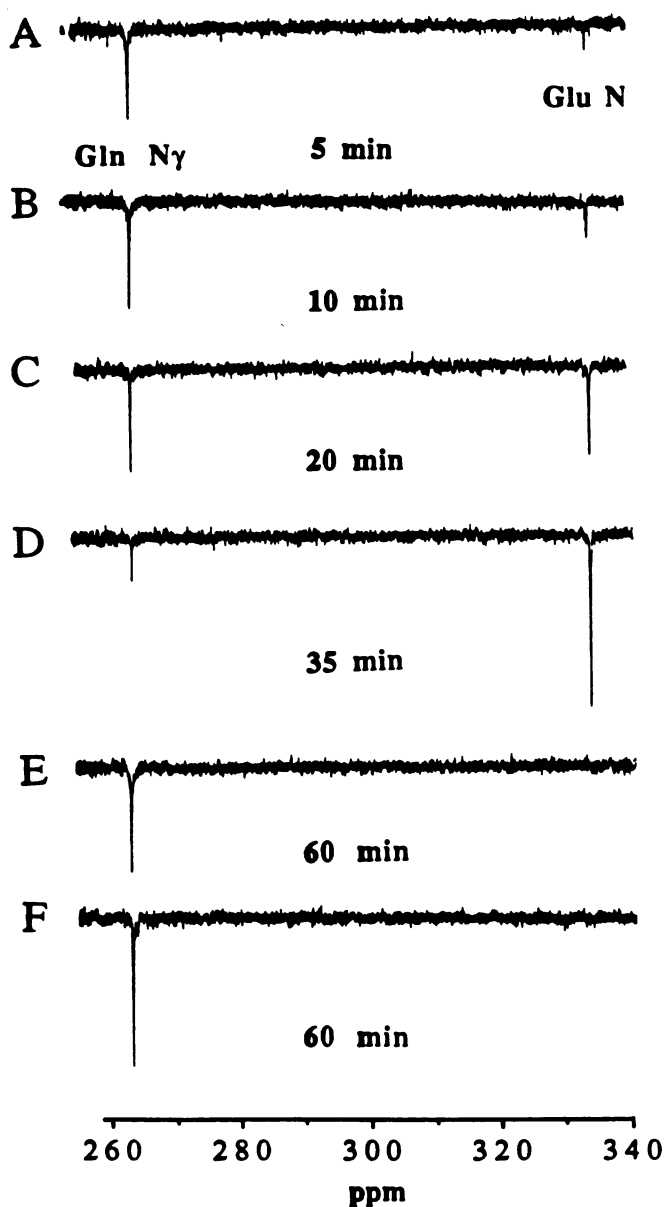


FIG. 3. <sup>15</sup>N NMR spectra of GOGAT assay solutions at various time intervals after addition of cell extracts of ammonia-grown *C. butyricum*. Assay preparations contained 7.5 mM NADH (A through D), 7.5 mM NADPH (E), or no coenzyme (F).

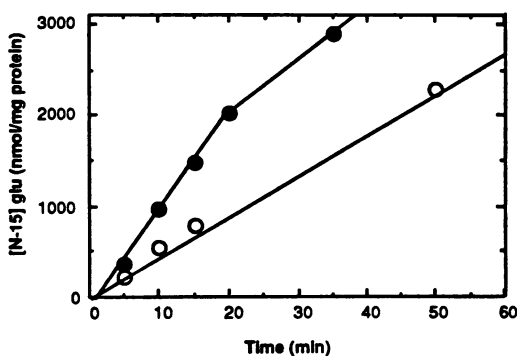


FIG. 4. Time-dependent formation of [ $^{15}\text{N}$ ]glutamic acid in NADH-GOGAT assay solutions on addition of cell extracts of ammonia-grown (●) and  $\text{N}_2$ -fixing (○) *C. butyricum*.

activity in *C. butyricum* grown on complex nitrogen sources for 36 h (the growth phase and the specific activity of the enzyme were not reported). However, under their growth conditions, the GDH activity was higher with NADH than with NADPH.

The affinities of NADPH-GDH for  $\text{NH}_4^+$  in the reductive amination reaction and for glutamate in the oxidative deamination reaction were measured to assess the physiological function of the enzyme. The Lineweaver-Burk plots of  $1/v_0$  versus  $1/[\text{NH}_4^+]$  were linear for  $\text{NH}_4^+$  concentrations of 1 to 25 mM, with a  $K_m^{app}$  of 2.8 mM, but at higher (25 to 100 mM)  $\text{NH}_4^+$  concentrations the slope of the least-squares line through the points was greater, with a  $K_m^{app}$  of 25.3 mM. Such biphasic kinetics with respect to  $\text{NH}_4^+$  have been observed in GDHs from a number of organisms (33). With glutamate as the substrate, the plots were linear for glutamate concentrations of 5 to 50 mM with a  $K_m$  of 8.9 mM. The results suggest that the enzyme has an adequate affinity for  $\text{NH}_4^+$  in the physiological  $\text{NH}_4^+$  concentration range of 1 to 25 mM.

Attempts were made to better understand the growth conditions under which NADPH-GDH is derepressed. No NADPH-GDH activity was detected in *C. butyricum* growing exponentially in reinforced *Clostridium* medium, which is expected to be rich in complex nitrogen sources including glutamate. No GDH activity, NADPH or NADH dependent, could be detected in *C. butyricum* growing exponentially on the synthetic glucose-salt medium with 50 mM L-glutamate as the sole nitrogen source—a growth condition under which GDH with a catabolic function, if present, is expected to be maximally induced. These results suggest that the metabolic function of NADPH-GDH of *C. butyricum*, like that of most bacterial NADPH-specific GDHs, is to assimilate ammonia. Our results, in combination with that of Bachofen and Herr (2), suggest that derepression of NADPH-GDH in *C. butyricum* occurs in the presence of complex nutrients but not in the synthetic glucose-salt medium tested here. Further studies are needed to understand the factors that cause such derepression.

G6PD, an enzyme that catalyzes the first reaction in the pentose phosphate pathway and is responsible for formation of NADPH from  $\text{NADP}^+$  in most organisms, has been reported to be undetectable in glucose-grown *C. butyricum* (12) (see Discussion). To investigate whether the absence of G6PD, which may result in an unusually low rate of regeneration of NADPH, can be one of the factors responsible for the absence of NADPH-dependent GDH activity in glucose-grown cells, the activity of G6PD was measured under

various growth conditions. Cells fixing  $\text{N}_2$  in glucose-salt medium had no G6PD activity (Table 1), in agreement with the reported result for glucose-grown cells (12). By contrast, the cells grown in the presence of complex carbon and nitrogen sources in addition to glucose under the conditions described above had a G6PD activity of 152 mU/mg of protein as well as an NADPH-GDH activity of  $207 \pm 20$  mU/mg of protein. Thus, both enzymes are derepressed under this growth condition.

## DISCUSSION

In *C. kluverii* which has an NADPH-GDH with  $K_m^{app}$  for  $\text{NH}_4^+$  of 12.0 mM, the GDH pathway was found to play a minor role relative to the GS-GOGAT pathway in ammonia assimilation during  $\text{N}_2$  fixation. In this respect, *C. kluverii* resembles *K. pneumoniae*, *Rhodospirillum rubrum*, and *Chromatium* sp. strain D, which have GDHs with high  $K_m$ s for  $\text{NH}_4^+$  (6 to 16 mM) and utilize the GS-GOGAT pathway for ammonia assimilation during  $\text{N}_2$  fixation (4, 5, 16, 17, 21).

In *C. butyricum* grown on synthetic glucose-salt medium with  $\text{NH}_4^+$  or  $\text{N}_2$  as the nitrogen source, the GDH activity was undetectable, and GS-GOGAT was the predominant pathway of ammonia assimilation. In this respect, *C. butyricum* resembles *C. pasteurianum* and *Azotobacter vinelandii*, which have undetectable levels of GDH even in ammonia-rich medium and assimilate ammonia by the GS-GOGAT pathway during  $\text{N}_2$  fixation (3, 8, 17, 20, 21). However, *C. butyricum* is capable of synthesizing NADPH-GDH with a  $K_m^{app}$  for  $\text{NH}_4^+$  of 2.8 mM, although the activity has so far been detected only after growth in complex media. Why is this enzyme repressed in *C. butyricum* during growth on glucose-salt media with  $\text{NH}_4^+$  or  $\text{N}_2$  as the nitrogen source? In *B. polymyxa* and *B. macerans*, NADPH-GDHs with comparable affinities for  $\text{NH}_4^+$  are the major pathways of ammonia assimilation in both ammonia-grown and  $\text{N}_2$ -fixing cells (13, 14).

Two unusual characteristics of glucose fermentation in *C. butyricum* may have some bearing on this question. One is the apparent inability of glucose-grown *C. butyricum* to regenerate NADPH from  $\text{NADP}^+$  directly through fermentation of glucose. The other is the presence of acetate and butyrate kinases in *C. butyricum*, which may allow efficient production of ATP from fermentation of glucose; as a result, the pool of ATP available for energy-requiring pathways such as the GS-GOGAT pathway may be somewhat higher in *C. butyricum* than in the *Bacillus* species during the energy-demanding process of nitrogen fixation.

The rate of regeneration of NADPH as a reductant for biosynthetic pathways may be unusually low in glucose-grown *C. butyricum*. Two key enzymes that are responsible for the regeneration of NADPH from  $\text{NADP}^+$  in many other organisms, G6PD and the  $\text{NADP}^+$ -specific malic enzyme, have been reported as undetectable in glucose-grown *C. butyricum* and *C. pasteurianum* (12). The only known pathway by which NADPH is regenerated from  $\text{NADP}^+$  in these two *Clostridium* species is through oxidation of NADH by  $\text{NAD}^+$ -ferredoxin oxidoreductase to form reduced ferredoxin, which in turn reduces  $\text{NADP}^+$  to NADPH in a reaction catalyzed by  $\text{NADP}^+$ -ferredoxin oxidoreductase (12). NADH, on the other hand, is produced in abundance during the fermentation of glucose and has to be rapidly oxidized to allow glycolysis to proceed (7). Measurements of the intracellular concentrations of the two coenzymes in sucrose-grown *C. pasteurianum* showed the  $(\text{NADPH} + \text{NADP}^+)/(\text{NADH} + \text{NAD}^+)$  ratio to be very low, 0.024 (23).

In contrast, the NADP<sup>+</sup>/NAD<sup>+</sup> ratio is 0.28 in glucose-grown *Bacillus licheniformis*, which has G6PD activity (22). These considerations suggest that a low rate of regeneration of NADPH resulting from the lack of G6PD activity may be one of the factors responsible for the absence of NADPH-GDH activity in ammonia-grown and N<sub>2</sub>-fixing *C. butyricum* in glucose-salt medium. Furthermore, with excess NADH, which must be reoxidized to allow glycolysis to proceed, the alternative ammonia-assimilating pathway with an NADH-specific enzyme, the GS/NADH-GOGAT pathway, is expected to be more favorable in terms of efficient recycling of reductants. Significantly, in one of the growth media used in this study that contained complex carbon sources in addition to glucose, *C. butyricum* was found to have G6PD activity (Table 1) and thus is capable of regenerating NADPH from NADP<sup>+</sup> directly through oxidation of glucose. The same culture had high NADPH-GDH activity, suggesting that such a cellular environment may be more favorable for derepression and utilization of the NADPH-GDH pathway. It is hoped that future studies will lead to observation of the activities of these enzymes in *C. butyricum* growing in chemically defined media and thus permit investigation of the factors that regulate the synthesis of these enzymes.

In terms of ATP production, the fermentation pathways of *C. butyricum* may be more efficient than those of the N<sub>2</sub>-fixing *Bacillus* species. *C. butyricum* ferments glucose to acetate and butyrate, yielding 3 to 4 mol of ATP per mol of glucose; the average is 3.3 mol of ATP per mol of glucose fermented in an ammonia-grown chemostat culture (7). *B. polymyxa* ferments glucose to ethanol and 2,3-butanediol in acidic media (32) and to ethanol and acetate in glucose-limited culture (1); the former pathway is expected to yield 2 mol of ATP per mol of glucose, and the latter is expected to yield 2 to 3 mol of ATP per mol of glucose depending on whether acetate kinase is present in *B. polymyxa* and on the ratio of ethanol/acetate formation. For obligate anaerobes such as *C. pasteurianum* and *C. butyricum*, development of high-energy yielding fermentation pathways was probably more crucial for survival than for facultative anaerobes such as the *Bacillus* species. If the efficiency of ATP production from fermentation of glucose is indeed significantly lower in *B. polymyxa* and *B. macerans* compared with that in the N<sub>2</sub>-fixing *Clostridium* species, the drain on the ATP pool imposed by N<sub>2</sub> fixation can be more severe for the *Bacillus* species than has been observed in *C. pasteurianum* (30). Consequently, assimilation of ammonia through the GDH pathway without further expenditure of energy may well be more crucial to those N<sub>2</sub>-fixing species having low energy-yielding fermentation pathways. We are currently investigating the efficiency of ATP production from fermentation of glucose as well as the levels of adenylate pools and ADP/ATP ratios in N<sub>2</sub>-fixing *Bacillus* species to obtain a better understanding of the possible correlation between cellular energy state and ammonia-assimilating pathways in nitrogen-fixing bacteria.

#### ACKNOWLEDGMENT

This work was supported by National Science Foundation grant DMB85-01617.

#### LITERATURE CITED

1. Arbige, M., and W. R. Chesbro. 1982. Very slow growth of *Bacillus polymyxa*; stringent response and maintenance energy. *Arch. Microbiol.* **132**:338-344.
2. Bachofen, R., and E. Herr. 1966. Über die synthese von alanin und aspartat aus acetat und CO<sub>2</sub> bei *Clostridium butyricum*.

3. Pathol. Microbiol. **29**:747-755.
3. Brown, C. M. 1976. Nitrogen metabolism in bacteria and fungi, p. 170-183. In A. C. R. Dean, D. C. Ellwood, C. G. T. Evans, and J. Melling (ed.), *Continuous culture 6: applications and new fields*. Ellis Horwood, Chichester, England.
4. Brown, C. M., and R. A. Herbert. 1977. Ammonia assimilation in purple and green sulphur bacteria. *FEMS Lett.* **1**:39-42.
5. Brown, C. M., and R. A. Herbert. 1977. Ammonia assimilation in members of the *Rhodospirillaceae*. *FEMS Lett.* **1**:43-46.
6. Carnahan, J. E., and J. E. Castle. 1958. Some requirements of biological nitrogen fixation. *J. Bacteriol.* **75**:121-124.
7. Crabbendam, P. M., O. M. Neijssel, and D. W. Tempest. 1985. Metabolic and energetic aspects of the growth of *Clostridium butyricum* on glucose in chemostat culture. *Arch. Microbiol.* **142**:375-382.
8. Dainty, R. H. 1972. Glutamate biosynthesis in *Clostridium pasteurianum* and its significance in nitrogen metabolism. *Biochem. J.* **126**:1055-1056.
9. Fisher, H. F. 1973. Glutamate dehydrogenase-ligand complexes and their relationship to the mechanism of the reaction. *Adv. Enzymol.* **39**:369-417.
10. Gottschalk, G., and H. A. Barker. 1966. Synthesis of glutamate and citrate by *Clostridium kluyveri*. A new type of citrate synthase. *Biochemistry* **5**:1125-1133.
11. Heer, E., and R. Bachofen. 1966. Pyruvatstoffwechsel von *Clostridium butyricum*. *Arch. Mikrobiol.* **54**:1-13.
12. Jungermann, K., R. K. Thauer, G. Leimenstoll, and K. Decker. 1973. Function of reduced pyridine nucleotide-ferredoxin oxidoreductases in saccharolytic *Clostridia*. *Biochim. Biophys. Acta* **305**:268-280.
13. Kanamori, K., R. L. Weiss, and J. D. Roberts. 1987. Ammonia assimilation in *Bacillus polymyxa*: N-15 NMR and enzymatic studies. *J. Biol. Chem.* **262**:11038-11045.
14. Kanamori, K., R. L. Weiss, and J. D. Roberts. 1987. Role of glutamate dehydrogenase in ammonia assimilation in nitrogen-fixing *Bacillus macerans*. *J. Bacteriol.* **169**:4692-4695.
15. Kanamori, K., R. L. Weiss, and J. D. Roberts. 1988. Glutamate biosynthesis in *Bacillus azotofixans*. *J. Biol. Chem.* **263**:2817-2823.
16. Kleiner, D. 1976. Ammonium uptake and metabolism by nitrogen-fixing bacteria. II. *Klebsiella pneumoniae*. *Arch. Microbiol.* **111**:85-91.
17. Kleiner, D., S. Phillips, and E. Fitzke. 1981. Pathways and regulatory aspects of N<sub>2</sub> and NH<sub>4</sub><sup>+</sup> assimilation in N<sub>2</sub>-fixing bacteria, p. 131-140. In H. Bothe and A. Trebst (ed.), *Biology of inorganic nitrogen and sulfur*. Springer-Verlag, New York.
18. Krishnan, I. S., R. K. Singhal, and R. D. Dua. 1986. Purification and characterization of glutamine synthetase from *Clostridium pasteurianum*. *Biochemistry* **25**:1589-1599.
19. Lowry, O. H., N. J. Rosebrough, A. L. Farr, and R. J. Randall. 1951. Protein measurement with the Folin phenol reagent. *J. Biol. Chem.* **193**:265-275.
20. Mortenson, L. E. 1978. Regulation of nitrogen fixation. *Curr. Top. Cell. Regul.* **13**:179-232.
21. Nagatani, H., M. Shimizu, and R. C. Valentine. 1971. The mechanism of ammonia assimilation in nitrogen fixing bacteria. *Arch. Mikrobiol.* **79**:164-171.
22. Opheim, D., and R. W. Bernlohr. 1973. Purification and regulation of glucose-6-phosphate dehydrogenase from *Bacillus licheniformis*. *J. Bacteriol.* **116**:1150-1159.
23. Petitdemange, H., D. Lambert, and R. Gay. 1973. Détermination des nucléotides pyridiniques oxydés et réduits chez *Clostridium pasteurianum*. *C. R. Soc. Biol.* **167**:111-115.
24. Prusiner, S., and L. Milner. 1970. Rapid radioactive assay for glutamine synthetase, glutaminase, asparagine synthetase and asparaginase. *Anal. Biochem.* **37**:429-438.
25. Rosenblum, E. D., and P. W. Wilson. 1949. Fixation of isotopic nitrogen by *Clostridium*. *J. Bacteriol.* **57**:413-414.
26. Stadtman, E. R., and R. M. Burton. 1955. Aldehyde dehydrogenase from *Clostridium kluyveri*. *Methods Enzymol.* **1**:518-523.
27. Stern, J. R., and G. Bambers. 1966. Glutamate biosynthesis in

- anaerobic bacteria. I. The citrate pathways of glutamate synthesis in *Clostridium kluyveri*. *Biochemistry* **5**:1113-1118.
28. **Thauer, R. K., K. Jungermann, and K. Decker.** 1977. Energy conservation in chemotrophic anaerobic bacteria. *Bacteriol. Rev.* **41**:100-180.
29. **Ujita, S., and K. Kimura.** 1982. Glucose-6-phosphate dehydrogenase, vegetative and spore *Bacillus subtilis*. *Methods Enzymol.* **89**:258-261.
30. **Upchurch, R. G., and L. E. Mortenson.** 1980. In vivo energetics and control of nitrogen fixation: changes in the adenylate energy charge and adenosine 5'-diphosphate/adenosine 5'-triphosphate ratios of cells during growth on dinitrogen versus growth on ammonia. *J. Bacteriol.* **143**:274-284.
31. **Winnacker, E. L., and H. A. Barker.** 1970. Purification and properties of a NAD-dependent glutamate dehydrogenase from *Clostridium* SB4. *Biochim. Biophys. Acta* **212**:225-242.
32. **Wood, W. A.** 1961. Fermentation of carbohydrates and related compounds, p. 59-150. *In* I. E. Gunsalus and R. Y. Stanier (ed.), *The bacteria. II. Metabolism*. Academic Press, Inc., New York.
33. **Wootton, J. C.** 1983. Re-assessment of ammonium-ion affinities of NADP-specific glutamate dehydrogenase. *Biochem. J.* **209**:527-531.