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Structural Basis for Catalysis by Onconase

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Abstract

Onconase (ONC) is a homolog of bovine pancreatic ribonuclease (RNase A) from the frog *Rana pipiens*. ONC displays antitumoral activity and is in advanced clinical trials for the treatment of cancer. Here, we report the first atomic structures of ONC-nucleic acid complexes: a T89N/E91A ONC·5'-AMP complex at 1.65 Å resolution and a wild-type ONC-d(AUGA) complex at 1.90 Å resolution. The latter structure and site-directed mutagenesis was used to reveal the atomic basis for substrate recognition and turnover by ONC. The residues in ONC that are proximal to the scissile phosphodiester bond (His10, Lys31, and His97) and uracil nucleobase (Thr35, Asp67, and Phe98) are conserved from RNase A and serve to generate a similar bell-shaped pH- k_{cat}/K_M profile for RNA cleavage. Glu91 of ONC forms two hydrogen bonds with the guanine nucleobase in d(AUGA), and Thr89 is in close proximity to that nucleobase. Installing a neutral or cationic residue at position 91 or an asparagine residue at position 89 virtually eliminated the 10²-fold guanine:adenine preference of ONC. A variant that combined such substitutions, T89N/E91A ONC, actually preferred adenine over guanine. In contrast, installing an arginine residue at position 91 increased the guanine preference and afforded an ONC variant with the highest known k_{cat}/K_M value. These data indicate that ONC discriminates between guanine and adenine by using Coulombic interactions and a network of hydrogen bonds. The structure of the ONC·d(AUGA) complex was also used to probe other aspects of catalysis. For example, the T5R substitution, designed to create a favorable Coulombic interaction between ONC and a phosphoryl group in RNA, increased ribonucleolytic activity by twofold. No variant, however, was more toxic to human cancer cells than wild-type ONC. Together, these findings provide a cynosure for understanding catalysis of RNA cleavage in a system of high medicinal relevance.

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Abbreviations used: 5'-AMP, adenosine 5'-monophosphate; 5'-GMP, guanosine 5'-monophosphate; 6-FAM, 6-carboxyfluorescein; 6-TAMRA, 6-tetraethylaminerhodamine; BIS-TRIS, bis(2-hydroxyethyl)amino-tris(hydroxymethyl)methane; CCD, charge-coupled device; DTT, dithiothreitol; <E, pyroglutamic acid; EDTA, ethylenediaminetetraacetic acid; EF-Tu, elongation factor Tu; FBS, fetal bovine serum; HEPES, 4-(2-hydroxyethyl)piperazine-1-ethanesulfonic acid; HIV, human immunodeficiency virus; *I*, inhibitor; IPTG, isopropyl-1-thio-β-D-galactopyranoside; MALDI-TOF, matrix-assisted laser desorption ionization-time-of-flight; MEPEG, polyethylene glycol monomethyl ether; MES, 2-(*N*-morpholino)-ethanesulfonic acid; ONC, Onconase® (a registered trademark of Alfacell, Inc.), also known as “ranpirnase”; PBS, phosphate-buffered saline; PCA, pyroglutamic acid; PDB, protein data bank; PEG, polyethylene glycol; RC-RNase, ribonuclease from *Rana catesbeiana*; RI, ribonuclease inhibitor protein; RMSD, root-mean-square deviation; RNase A, unglycosylated bovine pancreatic ribonuclease; RNase T1, microbial ribonuclease from *Aspergillus oryzae*; Tris, 2-amino-2-(hydroxymethyl)-1,3-propanediol; UpA, 6-carboxyfluorescein-dArUdAdA-6-carboxytetramethylrhodamine; UpG, 6-carboxyfluorescein-dArUdGdA-6-carboxytetramethylrhodamine.

Keywords

cytotoxin; ribonuclease; RNA; substrate specificity; X-ray crystallography

Introduction

The viability of organisms relies on the ability of proteins to recognize nucleic acids. In contrast, the ability of an enzyme to both recognize a nucleic acid and catalyze its cleavage can have deleterious consequences. For example, ribonucleases can be cytotoxic because cleaving RNA renders indecipherable its encoded information.^{1,2}

Onconase (ONC; Figure 1(a)) is a ribonuclease found in the eggs and early embryos of the frog *Rana pipiens*. ONC is a homolog of bovine pancreatic ribonuclease (RNase A³), and the two proteins share 30% amino-acid sequence identity and a similar three-dimensional structure.⁴ ONC is in confirmatory Phase IIb clinical trials for the treatment of malignant mesothelioma,^{5,6} and has been granted both orphan-drug and fast-track status by the US Food and Drug Administration. ONC also inhibits HIV-1 replication.⁷

ONC is a unique ribonuclease. The protein exhibits remarkable conformational stability ($T_m = 90^\circ\text{C}$ ⁸). Four disulfide bonds and the absence of a *cis* prolyl peptide bond contribute to this attribute.⁸⁻¹¹ ONC evades the cellular ribonuclease inhibitor protein (RI12), to which other ribonucleases bind with femtomolar affinity.¹³⁻¹⁹ The exceptional conformational stability and the RI-evading ability contribute to its cytotoxic activity.^{17,8,20}

ONC is a poor catalyst. The ribonucleolytic activity of ONC is three to five orders of magnitude lower than that of RNase A, due in large part to low affinity for its substrate.²¹ Nonetheless, the catalytic activity of ONC is necessary for cytotoxicity.²²

Homologs of RNase A bind a pyrimidine residue on the 5' side of the scissile phosphodiester bond in a small, conserved nucleobase-binding site.²³⁻²⁶ ONC displays a distinct preference for a guanosine nucleoside on the 3' side of the scissile phosphodiester bond.²¹ This guanine preference is also found in other frog ribonucleases,²⁷ but not in mammalian homologs.²⁸ The basis for this guanine preference in ONC is unknown. tRNA appears to be the major cellular substrate for ONC.²⁹ A recent study revealed an unconventional cleavage sequence for ONC—the guanosine–guanosine phosphodiester bond in the variable loop or the D-arm in tRNA.³⁰

Here, we report the first crystal structures of ONC–nucleic acid complexes. We use this structural information to address key issues in ONC catalysis. First, we determine the molecular basis for the nucleobase specificity of ONC through a systematic site-directed mutagenesis study. Next, we ask whether the low catalytic activity of ONC can be enhanced by a rational design approach. Finally, we seek to confirm the cellular target sequence of ONC *in vitro* using two novel fluorogenic substrates. We anticipate that the development of ONC as a cancer chemotherapeutic will benefit from the incipient understanding of its catalysis.

Results

Structural overview

The crystalline structures of the T89N/E91A ONC·5'-AMP and ONC·d(AUGA) complexes were solved to resolutions of 1.65 and 1.90 Å, respectively. Data collection, refinement, and model statistics are summarized in Table 1. The electron density was continuous for main-chain and side-chain atoms. Asymmetric units of the structures contain a single monomer with

a chain-fold virtually identical to that of free ONC (PDB entry 1ONC⁴). Both structures exhibit the typical bilobal shape of the RNase A superfamily, with two anti-parallel β -sheets flanked by two α -helices. In addition, our structures contain the rare N-terminal pyroglutamate residue and C-terminal disulfide bond, which are conserved among the amphibian members of the superfamily.

Structure of the T89N/E91A ONC-5'-AMP complex

In the crystalline T89N/E91A ONC-5'-AMP complex, four 5'-AMP molecules are bound to each enzyme molecule in a nonproductive mode (Figure 1(b)). The nucleobases of the 5'-AMP molecules form a stack that makes contact with multiple enzyme molecules in the crystal lattice. One of the four 5'-AMP molecules has its 5'-phosphate group bound in the active site of the enzyme. Another 5'-AMP molecule is located at the interface between two enzyme molecules. The remaining two 5'-AMP molecules show no significant interaction with the enzyme and have relatively high B -factors.

Structure of the wild-type ONC-d(AUGA) complex

In the crystalline ONC-d(AUGA) complex, one nucleic acid molecule is bound to one enzyme molecule (Figure 1(c)). There was no electron density above the noise level for the 3' adenosine. The electron density for the 5' adenosine was less than that of a typical water molecule, was therefore not included in the model. The disorder in these two flanking nucleosides suggests that they do not interact significantly with the enzyme.

The conformation of the Glu91 and His97 side chains in the ONC-d(AUGA) complex differed dramatically from those in free ONC (PDB entry 1ONC⁴). The χ_2 dihedral angle of His97 was investigated in detail and the residue was assigned to a conformation that had the most uniform B -values around the imidazole ring. This conformation, in which N ^{δ 1} forms a 2.89-Å hydrogen bond with O ^{γ 1} of Thr89 (Figure 1(d,e)), is the one supported best by the diffraction data. The alternative conformation, which puts N ^{δ 1} close to O^{5'} of guanosine, had refined B -factors around the imidazole ring that varied by nearly twofold, with the B -factors for the nitrogen atoms being too large and those of the carbon atoms being too small relative to surrounding residues.

Effect of pH on catalysis

RNase A cleaves RNA molecules by concerted general acid–base catalysis using the active-site residues His12 and His119.^{31–34} In this mechanism, both a protonated and an unprotonated histidine residue are required for efficient catalysis. As a result, the pH- k_{cat}/K_M profile of RNase A is a classic bell-shaped curve with a maximal catalytic activity near pH 6.^{35,36} These histidine residues are conserved in ONC and other amphibian homologs, suggesting that the catalytic mechanism of RNase A could be operative within the amphibian subfamily (Figure 2). A recent study by Liao and coworkers showed, however, that the $\text{p}K_a$ values of the catalytic histidine residues in RC-RNase 3, an amphibian homolog from the frog *Rana catesbeiana*, were significantly lower (4.26 and 5.96) than those manifested in RNase A.³⁷ Because of these low $\text{p}K_a$ values, the catalytic activity of RC-RNase 3 is maximal near pH 5.³⁸

To ensure that assays of the enzymatic activity of ONC were performed herein at an optimal pH, values of k_{cat}/K_M for the cleavage of 6-FAM-dArUdGdA-6-TAMRA by ONC were determined in buffers of various pH. The pH- k_{cat}/K_M profile for ONC has a symmetric bell shape with $\text{p}K_a$ values of 5.84 ± 0.05 and 6.77 ± 0.04 (Figure 3), which are similar to those of RNase A.^{39–41} Accordingly, assays were performed in a buffer of pH 6.0, which enables nearly maximal activity and allows for direct comparison with data on RNase A.

Residues in the B_1 subsite

RNase A and its homologs bind to RNA substrates in a cleft, leading to interactions that extend beyond the scissile bond. In RNase A, two subsites for a nucleobase (B_1 and B_2)^{42,25,43,26} and four subsites for a phosphoryl group (P_{-1} , P_0 , P_1 , and P_2)⁴⁴⁻⁴⁶ have been characterized in detail.³ The enzyme catalyzes the cleavage of the P–O^{5'} bond of a substrate bound in the B_1 – P_1 – B_2 subsites.

In the structure of the ONC·d(AUGA) complex, several residues of ONC are in close contact with the nucleobase moieties of d(AUGA). Lys33, Thr35, Asp67, and Phe98 are proximal to the uracil nucleobase (Figure 1(d)). O^{γ1} and backbone amide nitrogen atom of Thr35 form two hydrogen bonds with N₄ and O₂, respectively, of that nucleobase. The distance between N^ζ of Lys 33 and O⁴ of the uracil base is 3.73 Å. O^{δ1} of Asp67 forms a hydrogen bond with O^{γ1} of Thr35. Each of these residues except Lys33 is conserved in the RNase A superfamily, and hence constitute the B_1 subsite in ONC.

Residues in the B_2 subsite

In the structure of crystalline ONC·d(AUGA) complex, Glu91 forms two hydrogen bonds with the guanine nucleobase of d(AUGA) (Figure 1(d)). In addition, Thr89 is located in close proximity to this nucleobase. We hypothesized that these two residues constitute the B_2 subsite of ONC, where B_2 refers to the nucleobase on the 3' side of the scissile bond. To reveal the molecular basis for the guanine preference, we used site-directed mutagenesis to replace Thr89 and Glu91 with residues that differ in their net charge and ability to form hydrogen bonds (Table 2), and determined whether these variants showed an altered nucleobase preference.

Variants with an increased preference for guanine—Insertion of a positive charge at position 89 increased the guanine preference of ONC for cleavage of UpG relative to UpA by 10-fold (Figure 4; Table 2). The T89R and T89K variants displayed a 10³-fold preference for a guanine nucleobase over an adenine one. Furthermore, the value of k_{cat}/K_M for the cleavage of a guanine-containing substrate (UpG) by T89R ONC was increased by 3-fold from that of the wild-type enzyme.

Variants with a decreased preference for guanine—Elimination of the negative charge at position 91 not only diminished the preference of ONC for guanine *versus* adenine, but also increased the value of k_{cat}/K_M for the cleavage of an adenine-containing substrate (UpA, Table 2). The E91A, E91Q, and E91N variants of ONC cleaved UpG 2- to 9-fold more quickly than UpA, indicative of substantially lower guanine preferences than that of the wild-type enzyme. The largest change in preference was observed with E91K ONC, which actually had a 1.6-fold preference for UpA over UpG. Each of these variants above had 3- to 10-fold greater UpA-cleaving activity than did the wild-type enzyme.

Replacing Thr89 with an asparagine residue increased the value of k_{cat}/K_M for UpA cleavage by 3-fold (Table 2). In addition, the T89N substitution decreased the guanine preference of ONC by 50-fold. An aspartate or glutamine residue at this position likewise decreased the guanine preference, though without increasing the k_{cat}/K_M value. Combining the T89N and E91A substitutions augmented their individual effect. T89N/E91A ONC had the highest catalytic activity toward UpA among all of the variants, with a greater than 12-fold increase from the wild-type enzyme (Figure 4; Table 2). Moreover, T89N/E91A ONC had a complete shift in nucleobase preference, favoring adenine over guanine by 2.6-fold.

Inhibition of catalysis

Catalysis of 6-FAM–dArUdGdA–6-TAMRA cleavage by wild-type ONC and its variants was inhibited by the nucleotides 5'-GMP and 5'-AMP (Table 3). Wild-type ONC was inhibited by

5'-GMP and 5'-AMP with K_i values of $(6.7 \pm 0.5) \times 10^2 \mu\text{M}$ and $(3.7 \pm 0.3) \times 10^3 \mu\text{M}$, respectively. The K_i values of the T89K and T89R variants, which manifested an increased preference for the guanine nucleobase, did not differ markedly from that of wild-type ONC. On the contrary, variants with a decreased guanine preference showed relatively large changes in K_i values. Inhibition by 5'-AMP was 3-fold and 6-fold more pronounced for T89N ONC and T89N/E91A ONC, respectively. Inhibition by 5'-GMP was twofold less pronounced for E91A ONC.

Rational design of ONC variants with enhanced catalytic activity

Previous structural investigations have shown that the structure of ONC is less flexible than that of other ribonucleases.^{26,47} This rigidity could hinder facile interaction with the substrate. The structure of the ONC-d(AUGA) complex supports this hypothesis. The RMSD of the backbone C^α of ONC in the free and complexed form is just 0.4 Å,⁴⁸ suggesting little conformational change upon substrate binding.

Global approach—We sought to increase the catalytic activity of ONC by enhancing its flexibility. Sica and coworkers have implicated the short loops of ONC as causing rigidity.⁴⁹ Amphibian ribonucleases are grouped into two subfamilies.⁵⁰ The members of the subfamily with two short loops connecting the $\alpha 2$ -helix and $\beta 2$ -strand, and $\beta 4$ - and $\beta 5$ -strands (ONC, RC-RNase 2, RC-RNase 4, and RC-RNase 6) show low catalytic activity.^{38,51,52} The second subfamily, represented by RC-RNase and RC-RNase L1, contain loops with additional amino acid residues (IVGG and ITP, Figure 1(a)), and exhibit relatively high catalytic activity. The difference in catalytic activity between the two subfamilies reaches up to five orders of magnitude. We reasoned that installing the longer loops from RC-RNase onto ONC could enhance its catalytic activity. Accordingly, two ONC variants (IVGG/ITP and L27I/F28Y/IVGG/ITP ONC) were designed and tested for their ribonucleolytic activity. The L27I/F28Y substitutions were added to mimic the loop structure of RC-RNase more closely. Both variants did not, however, exhibit increased catalytic activity.

Local approach—A conspicuous feature of the ONC-d(AUGA) complex was the absence of significant electron density for the two flanking adenosine nucleosides in the nucleic acid (Figure 1(b)). This lack of electron density suggests a negligible interaction of these nucleosides with the enzyme. We asked whether the incorporation of residues that would promote “local” interaction with substrate could revive the catalytic activity of ONC. In RNase A, residues denoted as “*P* subsites” participate in substrate binding. These *P*-subsite residues are cationic lysine or arginine residues that can promote binding with the anionic phosphoryl groups in the nucleic acid backbone through favorable Coulombic interaction.⁴⁴ Removal of any residues in the *P* subsites leads to a loss in catalytic activity and substrate binding.^{53,46} In the ONC-d(AUGA) complex structure, no residues other than the active-site lysine residues (Lys9 and Lys31; *P*₁ subsite, Figure 1(d)) interact with phosphoryl groups in the nucleic acid. We hypothesized that installing a novel *P* subsite in ONC would increase its catalytic activity by enhancing substrate binding. A detailed structural comparison between the RNase A-d(ATAAG) complex (PDB entry 1RCN⁴⁴) and the ONC complex pointed towards Thr5 in ONC as being a promising location for a novel *P* subsite. Consequently, we created the T5K and T5R variants and measured their catalytic activity. In this regard, the T5R substitution performed better than did the T5K substitution, conferring a twofold increase in catalytic activity.

Val37 in RC-RNase is involved in nucleobase recognition (*B*₁ subsite).⁵² The corresponding residue in ONC is Lys33. K33V ONC was designed to test whether this substitution could increase nucleobase recognition in ONC. The catalytic activity of K33V ONC did not, however, differ from that of wild-type ONC.

Catalytic activity toward GpG

Excess wild-type ONC (10 μM) did not catalyze the cleavage of 6-FAM-dUrGdGdA-6-TAMRA or 6-FAM-dArGdGdA-6-TAMRA with a rate measurable by our assay. We estimate that the values of $k_{\text{cat}}/K_{\text{M}}$ to be $<1.0 \text{ M}^{-1}\text{s}^{-1}$. As a positive control we used RNase T1, which cleaves RNA at the 3'-side of guanosine residues.⁵⁴ One unit of RNase T1 (which produces an increase of 0.0004 A_{260} units in 1 min in 1 mL at room temperature using GpA as substrate) catalyzed the cleavage of our two GpG substrates to completion within a minute (data not shown).

Cytotoxicity of ONC variants

All ONC variants with altered nucleobase preference were less toxic than the wild-type enzyme for K-562 cells (Table 2). Likewise, ONC variants designed to possess increased catalytic activity displayed less cytotoxic activity than did wild-type ONC.

Discussion

Molecular basis for the B₂-subsite specificity of ONC

The contribution hydrogen bonds in an enzyme–substrate interface to catalysis has been established.⁵⁵⁻⁵⁹ In the structure of an ONC-nucleic acid complex, O ^{ϵ 1} and O ^{ϵ 2} of Glu91 form two hydrogen bonds with N₁ and N₂ of the guanine nucleobase (Figure 1(d)). In addition, Thr89 is located proximal to the guanine nucleobase. The distances between O ^{γ 1} and C ^{γ 2} of Thr89, and O⁶ of guanine are 3.57 and 4.25 Å, respectively. Similar interactions were observed in the RC-RNase-d(ACGA) complex.⁵² Specifically, Lys95 and Glu97 from RC-RNase establish an extensive network of hydrogen bonds with the nucleobase. Replacement of these residues with alanine resulted in a significant change in nucleobase preference.⁵² To understand the molecular basis underlying the nucleobase specificity of ONC, we took a more systematic approach—we replaced Thr89 and Glu91 with residues that vary in their ability to form hydrogen bonds or in their net charge (or both).

Catalysis of UpG cleavage by wild-type ONC is 110-fold more efficient than is the cleavage of UpA (Table 2). Each of the substitutions made at position 91 (alanine, glutamine, asparagine, and lysine) necessarily led to the loss of the two Glu91–guanine hydrogen bonds observed in the crystalline complex (Figure 1(d)). And, each of these variants exhibited a lower preference for guanine. The lysine substitution had the largest effect, actually producing a slight preference for adenine. It is worth noting that this specificity change originates from both a decrease in the catalytic activity toward the UpG substrate and an increase toward the UpA substrate.

To understand the substrate specificity on ONC in greater detail, we performed electrostatic potential calculations. The results are in agreement with previous calculations,⁶⁰ showing that the electron density on N¹ and N² of guanine is low (Figure 5). Thus, the interaction of this region with the anionic side chain of Glu91 would be promoted by favorable Coulombic interaction. The electrostatic potential of adenine differs significantly from that of guanine. The electron density on the corresponding region of adenine is high, and the anionic side chain of Glu91 would produce unfavorable interactions. We conclude that the 110-fold preference of ONC for a guanine nucleobase can be explained, at least in part, by electrostatic interactions between Glu91 and a purine nucleobase.

The structure of the ONC-d(AUGA) complex predicts that O₆ of the guanine nucleobase would be located close to a cationic side chain at position 89. In the RC-RNase-d(ACGA) complex, a lysine residue at the corresponding position makes a hydrogen bond with O₆ of guanine.⁵² Our data demonstrate that a residue bearing a positive charge at this position (lysine or arginine) strengthens the guanine preference of ONC, while a negative charge (aspartate) attenuates that

preference. Because O₆ on guanine is electron-rich, and N₆ on adenine is electron-poor (Figure 2), this result suggests again that the substrate specificity of ONC originates from its electrostatic interaction with the nucleobases. It is not clear, however, why T89N ONC has increased catalytic activity for the UpA substrate, because this substitution does not involve a change in net charge. We have also shown that the increase in catalytic activity toward the UpA substrate by the single substitutions can be enhanced further by their combination. T89N/E91A ONC, a variant that would possess two compatible interfaces for adenine, displays an 11-fold enhanced catalytic activity toward the UpA substrate. Furthermore, this double variant now prefers adenine to guanine by 2.6-fold.

We used the inhibitory effect of 5'-GMP or 5'-AMP on the cleavage of 6-FAM-dArUdGdA-6-TAMRA to assess the effect of substitutions on ground-state binding (Table 2). In this analysis, we assume that the mononucleotides bind at the B₂ and P₁ subsites, and that this binding competitively inhibits the cleavage of the substrate. The inhibitory effect of 5'-AMP is somewhat enhanced in T89K ONC and T89R ONC, indicating that binding of adenine at the B₂ subsite is slightly stronger in these variants, despite the adverse Coulombic effect of adding a positive charge. The binding of guanine became only slightly stronger after the substitutions. Thus, the major ramification of the extra positive charge at position 89 is the destabilization of transition-state binding of adenine as shown in the considerably decreased k_{cat}/K_M values (Table 2). The T89N substitution lowers the ground-state energy of the enzyme-adenine complex as indicated by the 3-fold lower value of K_i . The E91A substitution destabilizes the ground-state binding of guanine.

Guanine and adenine are part of nucleotides with a broad range of distinct biological functions. Discrimination between these two nucleobases, similar in shape and size, is critical for many enzymes (*e.g.*, kinases, ATPases, and GTPases). These proteins usually exhibit a strong preference for their cognate nucleotide over the other. Statistical analyses of protein-nucleotide structures in the RCSB Protein Data Bank has suggested that Coulombic interactions and a network of hydrogen bonds are the two major determinants of selectivity.^{61,60} A hydrogen bond is primarily an electrostatic interaction,⁶² and a correlation between hydrogen bonds and the electrostatic potential distribution in the protein-ligand interface was indeed found in the statistical analyses. In accord with these findings, our study indicates that Glu91 of ONC uses both Coulombic interactions and hydrogen bonds to confer nucleobase specificity.

Rational design of ONC with enhanced catalytic activity

Implanting extra amino acid residues (IVGG and ITP) designed to affect the “global” breathing motion on ONC⁴⁹ did not increase its catalytic activity (Table 2). This result implies that the factors causing the low catalytic activity of ONC reside outside of the loops. The greater catalytic activity and flexibility attained by the M23L substitution^{9,47,49} supports this notion. Met23 is located in a hydrophobic cavity created by the side chains of Ile22, Phe28, Lys31, Phe36, Cys68, and Tyr77. A bulky leucine residue at this position is thought to alter the position of residues that are critical for catalysis, especially that of Lys31. In this regard, it would be interesting to identify those amino acid residues responsible for rigidity in the β -sheet region by thorough sequence and structural comparison of amphibian ribonucleases. The “local” modifications (T5K, T5R, and K33V substitutions) did not result in a sizeable increase in catalytic activity. It is possible that the envisioned local interactions were not realized due to the low flexibility of ONC.

It is noteworthy that His97 in the ONC-d(AUGA) complex assumes a conformation distinct from that of corresponding histidine residues in other members of the RNase A superfamily. His97 is thought to function as an acid during catalysis by ONC (Figure 2), providing a proton to the displaced O^{5'}. In the RNase A-d(ATAAG) and RC-RNase-d(ACGA) complexes, the corresponding histidine residue adopts a suitable conformation for the transfer of a proton from

$N^{\delta 1}$ to $O^{5'}$.^{44,52} The distance between $N^{\delta 1}$ and $O^{5'}$ is 2.99 and 3.01 Å in the RNase A-d(ATAAG) and RC-RNase-d(ACGA) complex, respectively. In the ONC-d(AUGA) complex, however, the imidazole ring of His97 is rotated by nearly 180° about the χ_2 dihedral angle, such that the $N^{\delta 1}$ - $O^{5'}$ distance is 5.35 Å (Figure 1(d)). In addition, $N^{\delta 1}$ forms a hydrogen bond with $O^{\gamma 1}$ of Thr89, which is not conserved in either RNase A or RC-RNase. Thus in ONC, $N^{\epsilon 2}$ of His97 rather than $N^{\delta 1}$ could be the proton source.

Under physiological conditions, His97 could adopt a more conventional conformation. Twenty-five years ago, the corresponding histidine residue in RNase A was refined in two alternative conformations, indicative of its intrinsic flexibility.^{63,64} This histidine residue in RC-RNase exhibits conformational flexibility upon ligand binding.⁵² Likewise, the χ_1 dihedral angle of His97 differs by nearly 180° in unliganded ONC (PDB entry 1ONC) and the ONC-d(AUGA) complex. Regardless of this apparent flexibility, both His10 and His97 appear to participate in the catalysis of RNA cleavage (Figure 2), as the pH- k_{cat}/K_M profile of ONC retains the bell shape that is renowned in the RNase A superfamily (Figure 3)³⁵⁻³⁷ and is known to arise from the titration of the two active-site histidine residues.⁶⁵

***In vivo* cleavage sites for ONC**

tRNA has shown to be a predominant cellular target for ONC.²⁹ Still, it remains a matter of debate whether cellular tRNA is the sole substrate for ONC in the cell.⁶⁶ Members of the RNase A superfamily usually cleave single-stranded region of RNA that is not base-paired. How tRNA, which is extensively base-paired, becomes a target for ONC is unknown. Moreover, calculations predict that most of the tRNA in mammalian mitochondria and *Escherichia coli* exists as a ternary complex with elongation factor Tu (EF-Tu) and GTP,⁶⁷ which would limit accessibility to ONC even further. A recent report shed some light on this issue. Through an extensive sequence analysis on the cleavage site in tRNA, Suhasini & Sirdeshimukh demonstrated that ONC preferentially targets the variable loop or the D-arm in tRNA.³⁰ These regions of tRNA lack base pairing and are thus candidates for cleavage by ONC. In our ONC-d(AUGA) structure, ONC makes minimal contacts with substrate (only one *P* and two *B* subsites), implying that a compact single-stranded region in RNA is sufficient for an ONC substrate. In addition, the structure of the EF-Tu-GMP-tRNA^{Phe} complex (PDB entry 1TTT⁶⁸) confirms that the variable loop and the D-arm regions are not blocked by EF-Tu. Remarkably, the authors found that the predominant cleavage site is a guanosine-guanosine phosphodiester bond. In the structure of the ONC-d(AUGA) complex, the narrow *B*₁ subsite composed of Lys33, Thr35, and Phe98 makes intimate contact with the small uracil nucleobase (Figure 1(d)), and would likely occlude the large guanine nucleobase. To determine whether ONC catalyzes the cleavage of a guanosine-guanosine phosphodiester bond *in vitro*, we developed two novel substrates (6-FAM-dUrGdGdA-6-TAMRA and 6-FAM-dArGdGdA-6-TAMRA). The inability of ONC to cleave either substrate leads us to conclude that the *B*₁ subsite of ONC is too constricted to accommodate a large guanine nucleobase in an unstructured RNA molecule, at least in its *anti* orientation. We conclude that as yet unidentified elements of tRNA structure are involved in the cleavage of the guanosine-guanosine phosphodiester bond in tRNA.

Conclusions

We have determined the crystalline structure of two ONC-nucleic acid complexes at resolutions of 1.90 and 1.65 Å. Guided by these structures, we have revealed the atomic basis for substrate recognition and turnover by ONC. We have discovered that ONC utilizes Coulombic interactions (especially from Glu91) and a hydrogen bonding network to mediate substrate specificity, and have demonstrated that rational amino-acid substitutions can alter this specificity. Finally, we have probed structural attributes responsible for the low catalytic

activity of ONC and the unusual cleavage of a guanosine–guanosine phosphodiester bond in tRNA.

Materials and Methods

Materials

Human RI (as RNasin[®]) was from Promega (Madison, WI). RNase T1 was from Amibon (Austin, TX). 6-Carboxyfluorescein–dArUdAdA–6-carboxytetramethylrhodamine (6-FAM–dArUdAdA–6-TAMRA), 6-FAM–dArUdGdA–6-TAMRA, 6-FAM–dUrGdGdA–6-TAMRA, and 6-FAM–dArGdGdA–6-TAMRA were from Integrated DNA Technologies (Coralville, IA). 2-(*N*-Morpholino)ethanesulfonic acid (MES), adenosine 5′-monophosphate (5′-AMP), and guanosine 5′-monophosphate (5′-GMP) were from Sigma Chemical (St. Louis, MO). MES was purified further by anion-exchange chromatography prior to its use so as to eliminate contaminating oligo(vinylsulfonic acid), which is a potent inhibitor of ribonucleases. ⁶⁹ [methyl-³H]Thymidine was from PerkinElmer Life Sciences (Boston, MA). Phosphate-buffered saline (PBS) contained (in 1 liter) 0.20 g of KCl, 0.20 g of KH₂PO₄, 8.0 g of NaCl, and 2.16 g of Na₂HPO₄·7H₂O. All other chemicals and reagents were of commercial grade or better, and were used without further purification.

K-562 cells, which derive from a continuous human chronic myelogenous leukemia line, were from the American Type Culture Collection (Manassas, VA). Cell culture medium and supplements were from Invitrogen (Carlsbad, CA).

Biophysical measurements

Mass was measured by matrix-assisted laser desorption ionization–time-of-flight (MALDI–TOF) mass spectrometry using a Voyager-DE-PRO Biospectrometry Workstation Applied Biosystems, Foster City, CA) and a 3,5-dimethoxy-4-hydroxycinnamic acid (sinapinic acid) matrix (Sigma Chemical). Fluorescence measurements were performed with a QuantaMaster 1 Photon Counting Fluorometer equipped with sample stirring (Photon Technology International, South Brunswick, NJ). Radioactivity was quantitated with a Microbeta TriLux liquid scintillation and luminescence counter (PerkinElmer, Wellesley, MA).

Production of ONC and its variants

Wild-type ONC and its variants were produced in *E. coli* strain BL21(DE3) as described previously.²¹ The observed molecular mass of each purified protein was within 0.1% of its expected molecular mass (Table 2). The yield of each purified protein was 20 mg per liter of culture.

Crystal preparation

Crystals of T89N/E91A ONC were grown at 20 °C by the hanging-drop method from a solution containing protein (21.4 mg/ml) in water mixed with an equivalent volume of reservoir solution, which was 90 mM bis(2-hydroxyethyl)amino-tris(hydroxymethyl)methane (BIS-TRIS) buffer, pH 6.5, containing polyethylene glycol monomethyl ether (MEPEG) 2K (30.6% w/v) and 5′-AMP (50 mM). Crystals were soaked in a reservoir solution supplemented with ethylene glycol (5% v/v), and then flash-frozen in a stream of cryogenic N₂(g).

Crystals of the wild-type ONC-d(AUGA) complex were grown at 20 °C by the hanging-drop method from a solution containing protein (10 mg/ml) and nucleic acid (3.0 mM) in water mixed with an equivalent volume of reservoir solution, which was 0.10 M 4-(2-hydroxyethyl) piperazine-1-ethanesulfonic acid (HEPES) buffer, pH 7.5, containing polyethylene glycol (PEG) 3350 (25% v/v). Crystals were soaked in reservoir solutions supplemented with

increasing amounts of ethylene glycol (up to 20% v/v), and then flash-frozen in a stream of cryogenic N₂(g).

X-ray data collection

X-ray diffraction data for the T89N/E91A ONC·5'-AMP and the wild-type ONC·d(AUGA) complex were collected with a Bruker AXS Proteum R CCD detector and Microstar rotating-anode generator using copper K_α radiation. All data were processed and scaled with the programs SAINT and SADABS from the Proteum software suite (Bruker, Madison, WI).

Structure refinement

Structures of the T89N/E91A ONC·5'-AMP and the wild-type ONC·d(AUGA) complex were solved by molecular replacement with apo-ONC (PDB entry 1ONC⁴) as a model using the program MOLREP.⁷⁰ The structures were completed using alternate cycles of manual building in COOT⁷¹ and refinement in REFMAC5.⁷² The stereochemical quality of the final models was assessed with MolProbity.⁷³ The final coordinates were deposited in the RCSB Protein Data Bank⁷⁴ with accession numbers 2GMK and 2I5S for the T89N/E91A ONC·5'-AMP and wild-type ONC·d(AUGA) complexes, respectively.

Assays of catalytic activity

Ribonucleolytic activity of wild-type ONC and its variants was measured with a hypersensitive assay based on the cessation of fluorescence quenching as described in previous studies.²¹ To assess the B₂-subsite specificity of ONC, two fluorogenic substrates that contain distinct nucleobase sequences were used. Preference for the interaction with the guanine nucleobase was measured by using 6-FAM-dArUdGdA-6-TAMRA, and the interaction with the adenine nucleobase was measured by using 6-FAM-dArUdAdA-6-TAMRA. We also assessed the catalytic activity of wild-type ONC toward two novel ONC substrates, each containing a single cleavable guanosine-guanosine phosphodiester bond: 6-FAM-dUrGdGdA-6-TAMRA and 6-FAM-dArGdGdA-6-TAMRA.

pH- k_{cat}/K_M profile

The effect of pH on the value of k_{cat}/K_M for the cleavage of 6-FAM-dArUdGdA-6-TAMRA by wild-type ONC was determined in 2.00 mL of 1.0 mM buffer containing NaCl (1.0 M), 6-FAM-dArUdGdA-6-TAMRA (50–500 nM), and wild-type ONC (50 nM–1.0 μM). Buffers were sodium formate-HCl (pH 4.07), sodium acetate-HCl (pH 4.43–5.54), BIS-TRIS-NaOH (pH 6.0–6.64), MOPS-NaOH (pH 7.09–7.43), and TRIS-HCl (pH 7.87). The pH values of buffers were determined with a Φ40 pH meter from Beckman instruments (Fullerton, CA). Buffer concentration was kept low to ensure an identical salt environment, which is known to affect catalysis by ribonucleases.^{53,75} No change in the pH values of reaction mixtures was detectable during the course of the reaction. Higher concentration of the substrate was used at acidic pH (350–500 nM) than at basic pH (50 nM) because the protonation of the fluorescein moiety of the substrate decreases its fluorescence intensity. To obtain values of K_1 and K_2 , which are the apparent macroscopic acid dissociation constants of two functional groups important in catalysis, data were fitted by non-linear least squares regression analysis with the program PRISM 4 (GraphPad Software, San Diego, CA) to equation (1):

$$k_{\text{cat}}/K_M = \frac{(k_{\text{cat}}/K_M)_{\text{max}}}{\frac{[H^+]}{K_1} + 1 + \frac{K_2}{[H^+]}} \quad (1)$$

where $(k_{\text{cat}}/K_M)_{\text{max}}$ is the pH-independent specificity constant.

Inhibition of catalytic activity

Inhibition of ribonucleolytic activity was measured using 6-FAM-dArUdGdA-TAMRA as a substrate. Inhibition by 5'-AMP or 5'-GMP was assessed at 25 °C in 2.0 mL of 50 mM imidazole-HCl buffer, pH 6.0, containing NaCl (0–0.25 M), 6-FAM-dArUdAdA-6-TAMRA (60 nM), and ONC (1–5 nM), as described previously.^{76,77} We assumed that the nucleobases of 5'-AMP and 5'-GMP bind exclusively at the B_2 subsite, because the narrow B_1 subsite cannot accommodate a purine. This assumption is supported by our finding that ONC does not cleave our novel substrates containing a guanosine-guanosine phosphodiester bond (*vide supra*).

In the crystalline T89N/E91A ONC·5'-AMP complex, four 5'-AMP molecules are bound in a mode that is not productive in that the active site is occupied by phosphoryl groups from two 5'-AMP molecules and no adenine nucleobase is proximal to the B_2 subsite (*i.e.*, Asn89 and Ala91). The formation of such a nonproductive complex is unlikely during our inhibition assay. 5'-AMP molecules in the crystalline T89N/E91A ONC·5'-AMP complex are shared by multiple enzyme molecules in the crystal lattice, which is comprised of an extensive array of enzymes and nucleotides. Such protein-nucleotide interactions are unlikely to form in dilute aqueous solution.

Fluorescence (F) was measured with 493 and 515 nm as the excitation and emission wavelengths, respectively. The value of $\Delta F/\Delta t$ was measured for 3 min after the addition of ONC. Next, an aliquot of inhibitor (I) dissolved in the assay buffer was added, and $\Delta F/\Delta t$ was measured in the presence of the inhibitor for 3 min. The concentration of inhibitor in the assay was doubled repeatedly at 3-min intervals. Excess wild-type RNase A was then added to the mixture to ensure that <10% of the substrate had been cleaved prior to completion of the inhibition assay. Apparent changes in ribonucleolytic activity due to dilution were corrected by comparing values to those from an assay in which aliquots of buffer were added. To obtain values of K_i , data were fitted by non-linear least squares regression analysis with the program DELTAGRAPH 5.5 (Red Rock Software, Salt Lake City, UT) to equation (2)^{76,77}:

$$\Delta F/\Delta t = (\Delta F/\Delta t)_0 \left(\frac{K_i}{[I] + K_i} \right) \quad (2)$$

where $(\Delta F/\Delta t)_0$ is the ribonucleolytic activity prior to the addition of the inhibitor.

Assays of cytotoxic activity

The effect of wild-type ONC, its variants, and RNase A on cell proliferation was determined as described in previous studies.²¹ Cytotoxicity data were analyzed with the programs SIGMAPLOT (SPSS Science, Chicago, IL) and DELTAGRAPH 5.5. Each data point represents the mean (\pm SE) of at least three experiments performed in triplicate. To obtain values of IC_{50} , data were fitted by non-linear least squares regression analysis with the program DELTAGRAPH 5.5 to equation (3)⁷⁸:

$$S = \frac{IC_{50}}{IC_{50} + [\text{ribonuclease}]} \times 100 \quad (3)$$

where S is the percent of total DNA synthesis during the 4-h pulse as compared to that of a PBS control.

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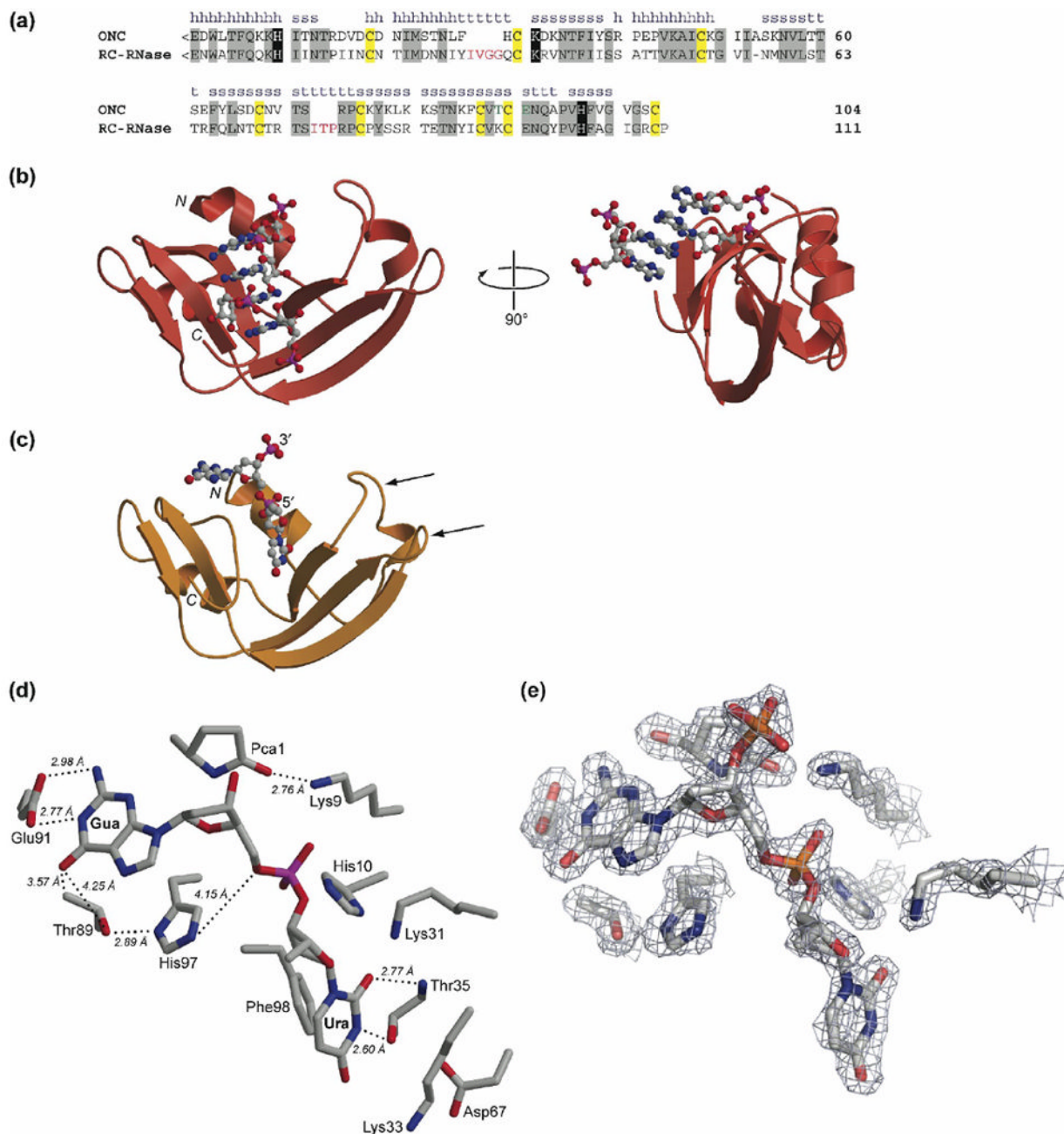


Figure 1.

Primary and tertiary structure of ONC. (a) Amino-acid sequence alignment of ONC and RC-RNase. The secondary structure of ONC is labeled as h (α -helix), s (β -strand), or t (turn). Residues conserved between the two ribonucleases are in gray boxes. <E denotes a pyroglutamate residue. Key active-site residues are in black boxes. Cysteine residues that form intramolecular disulfide bonds are in yellow boxes. Residues that participate in the nucleobase recognition at the B_2 subsite in ONC are in green. Extra amino acid residues in the loops of RC-RNase are in red. (b) Ribbon diagram of the three-dimensional structure of the T89N/E91A ONC·5'-AMP complex (PDB entry 2GMK). (c) Ribbon diagram of the three-dimensional structure of the ONC·d(AUGA) complex (PDB entry 2I5S). The two flanking adenosine

nucleotides are not included in the diagram because of their low electron density. The two arrowheads indicate the two loops that were subjected to elongation herein. (d) Stick diagram of the active site and B_1 and B_2 subsites in the ONC-d(AUGA) complex. Thr35 forms two hydrogen bonds with the uracil nucleobase. Glu91 forms two hydrogen bonds with the guanine nucleobase. Thr89 is located close to the carbonyl oxygen of guanine. (e) Electron density map of the active site (σ -weighted $2F_o - F_c$) contoured at 1.5σ over background. Images in panels (b)–(e) were created with the program MOLSCRIPT and rendered with the program RASTER3D.⁷⁹

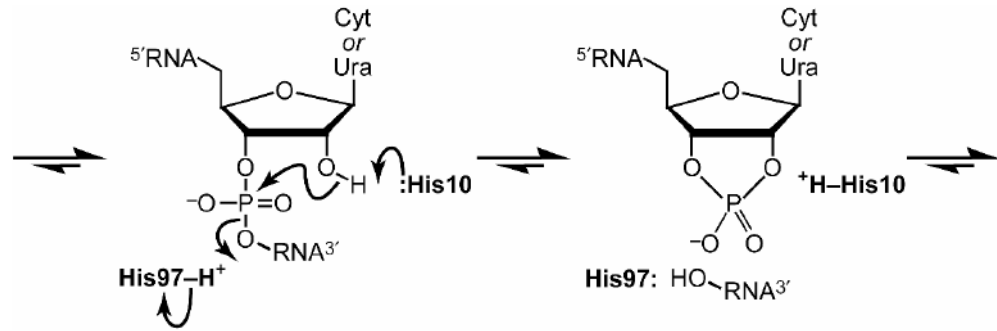


Figure 2.
Putative mechanism of catalysis of RNA cleavage by ONC. In this transphosphorylation reaction, His10 acts as a base, and His97 acts as an acid.

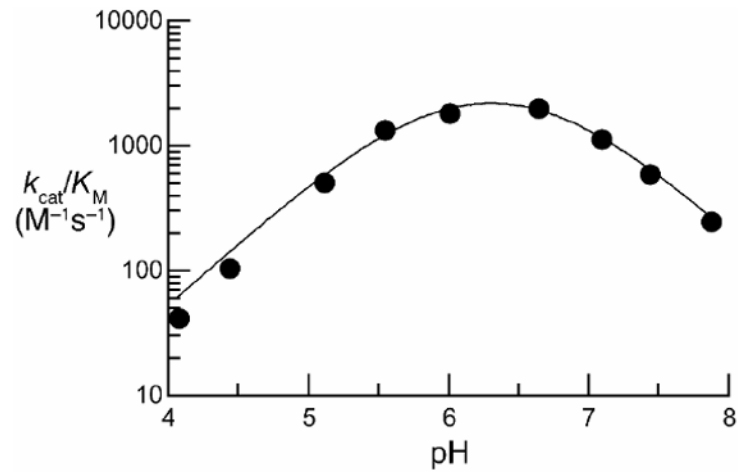


Figure 3. pH- k_{cat}/K_M profile for the cleavage of 6-FAM-dArUdGdA-6-TAMRA by ONC. Assays were performed at 23 °C in 1.0 mM buffer containing NaCl (1.0 M). Determination of k_{cat}/K_M values was performed in triplicate. Data were fitted to equation (1) by non-linear least squares regression analysis to give pK_a values of 5.84 ± 0.05 and 6.77 ± 0.04 .

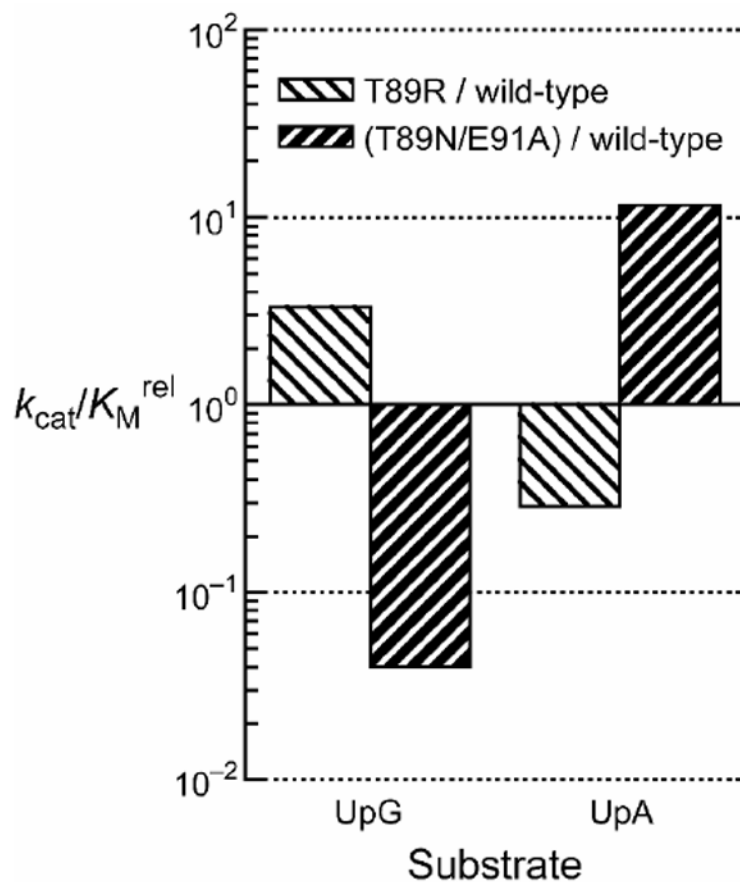


Figure 4.

Effect of Thr89 and Glu91 on the substrate specificity of ONC. Bars indicate the effect of replacing Thr89 or Glu91 on the value of k_{cat}/K_M for the cleavage of 6-FAM-dArUdGdA-6-TAMRA (UpG) and 6-FAM-dArUdAdA-6-TAMRA (UpA). The guanine:adenine preference of T89R ONC is 3.3×10^3 -fold greater than that of T89N/E91A ONC (Table 2).

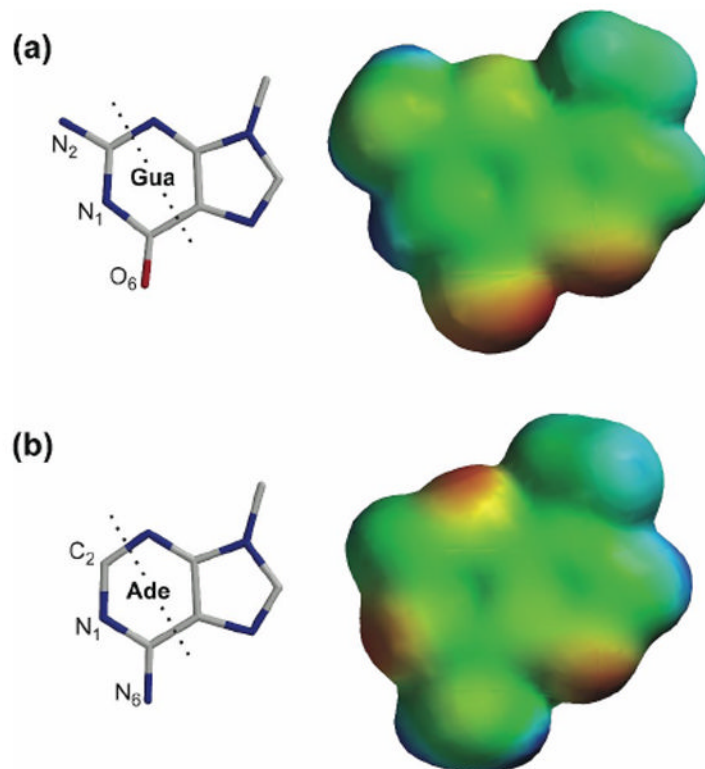


Figure 5. Structure and electron density diagram of (a) 9-methylguanine and (b) 9-methyladenine. Atoms to the left of the dashed lines interact with Thr89 and Glu91 of the B_2 subsite of ONC (Figure 1(d)). Electron density of the nucleobases was calculated with the program SPARTAN (Wavefunction, Irvine, CA). Electron-rich regions are colored red; electron-poor regions are blue; neutral regions are green.

Table 1Summary of data collection and refinement statistics^a

	T89N/E91A ONC-5'-AMP	Wild-type ONC-d(AUGA)
Space group	P2 ₁ 2 ₁ 2 ₁	P2 ₁ 2 ₁ 2
Unit cell parameters (Å)	a=29.0, b=52.1, c=66.1	a=129.2, b=26.1, c=32.5
Data collection statistics		
Resolution range (Å)	66.14–1.65 (1.70–1.65)	64.62–1.90 (1.95–1.90)
Number of reflections (measured/unique)	336231/12645	91217/9270
Completeness (%)	99.9 (99.6)	99.7 (95.9)
R_{merge}^b	0.054 (0.663)	0.125 (0.517)
Redundancy	26.6 (5.1)	9.8 (3.3)
Mean σ/I	40.66 (3.32)	12.30 (2.52)
Refinement statistics		
Resolution range (Å)	40.93–1.65	32.49–1.90
Number of reflections (total/test)	12597/612	9229/441
$R_{\text{cryst}}^c/R_{\text{free}}^d$	0.165/0.217	0.178/0.240
RMSD bonds (Å)	0.010	0.013
RMSD angles (deg)	1.718	1.415
Average B factor (Å ²)	19.35	16.60
Number of water molecules	196	155
Ramachandran favored (%)	98.1	100.0
Ramachandran allowed (%)	1.9	0.0

^aValues in parentheses are for the highest-resolution shell

^b $R_{\text{merge}} = \sum_h \sum_i |I_i(h) - \langle I(h) \rangle| / \sum_h \sum_i I_i(h)$, where $I_i(h)$ is the intensity of an individual measurement of the reflection and $\langle I(h) \rangle$ is the mean intensity of the reflection.

^c $R_{\text{cryst}} = \sum_h \|F_{\text{obs}} - F_{\text{calc}}\| / \sum_h \|F_{\text{calc}}\|$, where F_{obs} and F_{calc} are the observed and calculated structure factor amplitudes, respectively.

^d R_{free} was calculated as R_{cryst} using 5.0% of the randomly selected unique reflections that were omitted from structure refinement.

Table 2

Values of k_{cat}/K_M , m/z , and IC_{50} for ONC and its variants

ONC	k_{cat}/K_M		UpG/UpA	m/z^c		IC_{50} (μM) ^d
	UpG ($M^{-1}s^{-1}$) ^a	UpA ($M^{-1}s^{-1}$) ^b		expected	observed	
Wild-type	(4.2 ± 0.5) × 10⁴	(3.8 ± 0.2) × 10²	1.1 × 10²	11820	11820	0.2
T89A	(1.5 ± 0.1) × 10 ⁴	(4.7 ± 0.4) × 10 ²	32	11790	11793	0.4
T89K	(4.1 ± 0.3) × 10 ⁴	(7.0 ± 0.4) × 10 ¹	5.9 × 10 ²	11820	11824	1.0
T89R	(1.4 ± 0.1) × 10 ⁵	(1.1 ± 0.1) × 10 ²	1.3 × 10 ³	11875	11868	0.8
T89D	(1.2 ± 0.1) × 10 ³	(1.1 ± 0.1) × 10 ²	11	11834	11840	2.0
T89Q	(1.4 ± 0.1) × 10 ³	(3.5 ± 0.2) × 10 ¹	40	11847	11852	ND
T89N	(2.6 ± 0.2) × 10 ³	(1.2 ± 0.1) × 10 ³	2.2	11833	11826	0.7
E91A	(3.9 ± 0.3) × 10 ³	(1.6 ± 0.1) × 10 ³	2.4	11762	11750	1.5
E91Q	(9.4 ± 0.3) × 10 ³	(1.1 ± 0.1) × 10 ³	8.6	11819	11827	0.6
E91N	(4.2 ± 0.3) × 10 ³	(1.6 ± 0.3) × 10 ³	2.6	11811	11805	ND
E91K	(2.4 ± 0.1) × 10 ³	(3.8 ± 0.3) × 10 ³	0.63	11819	11828	ND
T89N/E91A	(1.7 ± 0.3) × 10 ³	(4.4 ± 0.2) × 10 ³	0.39	11775	11780	2.5
T89K/E91A	(7.5 ± 0.3) × 10 ³	(5.6 ± 0.2) × 10 ²	13	11788	11793	1.7
T5K	(5.8 ± 0.1) × 10 ⁴	ND ^e	ND	11846	11852	0.4
T5R	(8.3 ± 0.5) × 10 ⁴	ND	ND	11874	11881	1.2
K33V	(4.4 ± 0.3) × 10 ⁴	(3.7 ± 0.3) × 10 ²	1.2 × 10 ²	11791	11795	ND
IvGGITP	(7.1 ± 0.7) × 10 ⁴	(1.6 ± 0.2) × 10 ²	4.4 × 10 ²	11818	11823	ND
L27/F28Y IvGGITP	(1.9 ± 0.2) × 10 ⁴	(5.3 ± 0.4) × 10 ²	36	12447	12454	4.0
	(3.4 ± 0.2) × 10 ⁴	(6.0 ± 0.2) × 10 ²	57	12463	12470	4.0

^{a,b} Values of k_{cat}/K_M (\pm SE) are for cleavage of 6-FAM-dArUdGdA-6-TAMRA or 6-FAM-dArUdAdA-6-TAMRA in 0.10 M MES-NaOH buffer (pH 6.0) containing NaCl (0.10 M) at 23 \pm 2 °C.

^c Values of m/z were determined by MALDI-TOF mass spectrometry.

^d Values of IC_{50} (\pm SE) are for incorporation of [*methyl*]-³H]thymidine into the DNA of K-562 cells exposed to ONC or a variant, and were calculated with equation (3).

^e ND, not determined.

Table 3Values of K_i for ONC and its variants

ONC	K_i (μM) ^a	
	5'-GMP	5'-AMP
Wild-type	$(6.7 \pm 0.5) \times 10^2$	$(3.7 \pm 0.3) \times 10^3$
T89K	$(5.3 \pm 0.2) \times 10^2$	$(2.0 \pm 0.2) \times 10^3$
T89R	$(4.3 \pm 0.6) \times 10^2$	$(2.4 \pm 0.4) \times 10^3$
T89N	$(1.0 \pm 0.1) \times 10^3$	$(1.2 \pm 0.1) \times 10^3$
E91A	$(1.3 \pm 0.3) \times 10^3$	$(3.2 \pm 0.4) \times 10^3$
T89N/E91A	$(9.9 \pm 0.7) \times 10^2$	$(5.9 \pm 0.2) \times 10^2$

^aValues of K_i (\pm SE) are for inhibition of catalysis of 6-FAM-dArUdGdA-6-TAMRA cleavage in 0.10 M MES-NaOH buffer (pH 6.0) containing NaCl (0.10 M) at 23 ± 2 °C.