

Counting membrane-embedded KCNE β -subunits in functioning K^+ channel complexes

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Ion channels are multisubunit proteins responsible for the generation and propagation of action potentials in nerve, skeletal muscle, and heart as well as maintaining salt and water homeostasis in epithelium. The subunit composition and stoichiometry of these membrane protein complexes underlies their physiological function, as different cells pair ion-conducting α -subunits with specific regulatory β -subunits to produce complexes with diverse ion-conducting and gating properties. However, determining the number of α - and β -subunits in functioning ion channel complexes is challenging and often fraught with contradictory results. Here we describe the synthesis of a chemically releasable, irreversible K^+ channel inhibitor and its iterative application to tally the number of β -subunits in a KCNQ1/KCNE1 K^+ channel complex. Using this inhibitor in electrical recordings, we definitively show that there are two KCNE subunits in a functioning tetrameric K^+ channel, breaking the apparent fourfold arrangement of the ion-conducting subunits. This digital determination rules out any measurable contribution from supra, sub, and multiple stoichiometries, providing a uniform structural picture to interpret KCNE β -subunit modulation of voltage-gated K^+ channels and the inherited mutations that cause dysfunction. Moreover, the architectural asymmetry of the K^+ channel complex affords a unique opportunity to therapeutically target ion channels that coassemble with KCNE β -subunits.

charybdotoxin | KCNQ1 | stoichiometry

Voltage-gated KCNQ1 K^+ channels are tetrameric integral membrane proteins that open and close in response to changes in membrane potential. To regulate cellular potassium flow in both excitable and nonexcitable tissues, Q1 channels coassemble with regulatory KCNE peptides, forming membrane-embedded K^+ channel complexes with various voltage-sensing and gating properties (1). The importance of forming a properly assembled Q1-KCNE complex is underscored by the mutations that give rise to long QT syndrome and congenital hearing loss (2, 3). Although the fourfold arrangement of the Q1 α -subunits along the ion conduction pathway is established and unquestioned, the number of KCNE β -subunits in the K^+ channel complex has been a long-standing and heated debate (4–7). Goldstein and coworkers (6) have strongly argued that Q1 channels average two KCNE peptides per complex; however, they could not definitively rule out K^+ channel complexes containing a solitary KCNE peptide. Another complicating factor is the notion that Q1 channels form complexes with KCNE β -subunits with multiple stoichiometries (5, 8). Thus, approaches that measure the average number of KCNE peptides in the Q1 channel complex at the cell surface are unable to readily discern between fixed and various stoichiometries.

To determine the stoichiometry or stoichiometries of Q1-KCNE K^+ channel complexes, we devised an iterative cell surface modification scheme where the labeling reagent specifically binds to the outer vestibule of the K^+ conduction pore while covalently modifying a cysteine-bearing KCNE peptide in the channel complex (Fig. 1). Once the reaction is complete, excess reagent is removed and the irreversibly bound reagent is

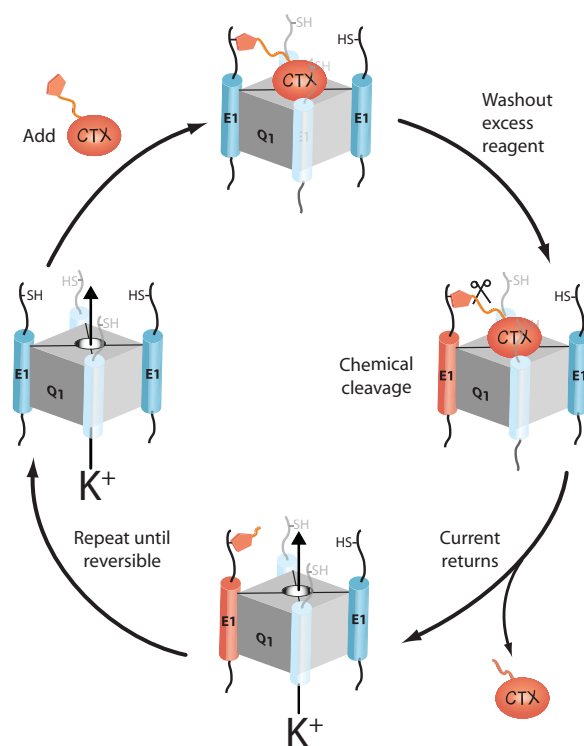


Fig. 1. A cartoon depiction of one modification cycle of the iterative counting strategy used to determine the number of KCNE1 subunits in a KCNQ1 K^+ channel complex by using a derivatized charybdotoxin (CTX). Each complete round of chemical treatment modifies one E1 subunit, allowing for the direct counting of KCNE subunits in the K^+ channel complex.

chemically liberated, restoring K^+ channel complex function, and thereby modifying one KCNE peptide. Cysteine modification renders the KCNE peptide chemically inactive, and thus the number of KCNE peptides in the K^+ channel complex is simply the number of complete modification cycles (reagent in, wash-out, cleaved) required before the washout step in the cycle becomes reversible. If Q1 and KCNE subunits form complexes with multiple stoichiometries, then these subpopulations would manifest themselves as reversibility during the labeling steps of the modification scheme. The measured reversibility would directly correspond to the percentage of subpopulations of K^+

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Table 1. Recovery parameters after CTX-Clv application

Construct	Treatment	τ_{recovery} , s	% recovery	<i>n</i>
Q1/E1	CTX-Clv	98 ± 1	95 ± 2	5
Q1/E1 _{T14C}	CTX-Clv, TCEP	97 ± 1	96 ± 2	4
Q1/E1 _{T14C}	CTX-Clv + excess CTX	155 ± 4	96 ± 1	3
Q1/E1 _{T14C}	CTX-Clv, CTX-Mal, TCEP	94 ± 3	94 ± 1	3
Q1/E1 _{T14C}	Round 1, TCEP	96 ± 2	96 ± 2	3
Q1/E1 _{T14C}	Round 2, TCEP	98 ± 3	98 ± 4	3
Q1/E1 _{T14C}	Round 3	97 ± 1	94 ± 3	3

Data are from individual experiments obtained from 3–5 oocytes. Treatment describes the sequential application of chemical reagents to the indicated construct. Round 1, Round 2, and Round 3 indicates the first, second, and third rounds of subunit counting (Fig. 4a), respectively. Time constants of recovery (τ_{recovery}) were fit to a single exponential as described in *Materials and Methods*. Percent recovery is a comparison between the amount of current remaining after CTX-Clv release with the amount of current before CTX-Clv application. Values are mean ± SEM.

red trace). Irreversible inhibition requires the cysteine in E1 because CTX-Clv is an entirely reversible inhibitor with wild-type Q1/E1 complexes (Fig. 3b). These results demonstrate that each step in the modification cycle occurs quantitatively within minutes.

Compared with model reactions between positively charged maleimides and thiols, the modification of E1_{T14C} by CTX-Clv is ≈1,000-fold faster (19). This faster modification rate was expected on the basis of previous affinity-driven modifying agents, which increase the reagent's effective molarity, thereby accelerating the chemical reaction (20, 21). In addition, this accelerated reaction rate indicates that toxin binding is required for modification and suggests that the nonspecific bimolecular reaction between CTX-Clv and E1_{T14C} does not measurably occur. To directly demonstrate that the nonspecific reaction does not occur, we performed a competition experiment in which we treated Q1/E1_{T14C} complexes with 10 nM CTX-Clv and 500 nM unlabeled CTX for an extended duration (Fig. 3c). As predicted for a reagent that requires binding before modification, CTX-Clv did not modify the Q1/E1_{T14C} complex in the presence of excess competitor (unlabeled CTX), as evidenced by complete washout (Fig. 3c). This result demonstrates that the bimolecular reaction between CTX-Clv and E1_{T14C} does not contribute to the irreversible inhibition of the K⁺ channel complex. In addition, the vast excess of underivatized CTX inhibited the same amount of total current as 10 nM CTX-Clv (compare traces in Fig. 3a and c *Left*), further demonstrating that the residual current is from endogenous CTX-insensitive channels.

To determine whether a covalently tethered CTX-Clv prevents further toxin-accelerated modifications of the other KCNE peptides in the channel complex, we performed a second competition experiment (Fig. 3d), this time using CTX-Mal (Fig. 2a), a noncleavable version of CTX-Clv. CTX-Mal irreversibly inhibits Q1/E1_{T14C} with a modification rate similar to that of CTX-Clv (16), yet the irreversible inhibition by CTX-Mal is completely resistant to TCEP treatment (SI Fig. 5). Q1/E1_{T14C} complexes were first irreversibly inhibited by CTX-Clv, as observed after washout (Fig. 3d). The inhibited complexes were then treated with CTX-Mal for the same duration and the reagent was washed out. A final treatment with TCEP resulted in complete restoration of the current that was measured before the toxin treatment sequence began. Because the current fully returned after TCEP treatment and CTX-Mal inhibition is not reversible by TCEP, this result confirms that K⁺ channel complexes with a tethered CTX-Clv are protected from subsequent toxin-accelerated modifications. Table 1 shows that CTX-Clv dissociates from the Q1/E1 channel complex with similar kinetics and percent recovery. The one exception is the competition

experiment (CTX-Clv + excess CTX), where the measured recovery kinetics reflects unmodified CTX dissociation, which takes ≈50% longer than CTX-Clv.

With this now battle-tested reagent, we used CTX-Clv to count the number of KCNE peptides in the Q1 K⁺ channel complex. Q1/E1_{T14C} complexes were repetitively subjected to the modification cycle: CTX-Clv treatment, washout, and TCEP reductive cleavage (Fig. 4a). The first two rounds required TCEP treatment to release CTX-Clv from the K⁺ channel complex. The second round of CTX-Clv treatment proceeded as smoothly as the first, demonstrating that the molecular remnants left on the KCNE peptide after TCEP cleavage do not interfere with subsequent rounds of binding and modification. In the third round, CTX-Clv once again inhibited the K⁺ channel complex; however, inhibition was completely reversible with a simple washout, showing that it takes two modification cycles to react with all of the KCNE peptides in a K⁺ channel complex. In addition, we never observed irreversible modification by CTX-Clv after the second round, nor did we observe partial reversibility during the first or second rounds, which shows that K⁺ channel complexes with multiple stoichiometries do not significantly contribute to the cell surface population. Moreover, the near quantitative reactions observed in each round indicate that Q1/E1 complexes at the cell surface are essentially devoid of endogenous *Xenopus* KCNE peptides (22), since their incorporation would result in incomplete modification cycles and reversibility with CTX-Clv. The kinetics of CTX-Clv unbinding and the percent recovery did not vary from round to round (Table 1).

Although we show that a CTX-Clv bound and tethered to a K⁺ channel complex prevents subsequent KCNE modifications by CTX-Mal (Fig. 2d), we have not directly shown that one, and only one, KCNE peptide reacts per CTX-Clv treatment. To do this, we took advantage of the difference in the probability mass function when *n* = 2, 4, or any even number of KCNE subunits because the results in Fig. 4a eliminate stoichiometries with an odd number of KCNE peptides (*Appendix*). We first pretreated Q1/E1_{T14C} complexes with a membrane-impermeant maleimide (Mal-ES) (19) to randomly and independently react with a fraction of the E1_{T14C} peptides. We then determined the percentage of Q1/E1_{T14C} complexes that were fully modified by this pretreatment by measuring the reversibility of CTX-Clv after washout (Fig. 4b). Based on the 19% reversible inhibition of the CTX-sensitive current that we observed, the binomial distribution predicts that a second treatment with CTX-Clv should result in 69% reversibility if only one KCNE is modified per CTX-Clv treatment; 88%, if two KCNE peptides react per treatment. Cleavage of the previously tethered CTX-Clv, followed by another bolus of CTX-Clv, and washout resulted in 65 ± 4% (SEM) reversibility (*n* = 3). These results demonstrate that only one KCNE peptide is modified per exposure to CTX-Clv. In addition, this binomial determination of the stoichiometry of the complex corroborates the counting experiment in Fig. 4a, confirms the previous stoichiometric determinations by Goldstein and coworkers (4, 6), and indicates that the lack of reactivity in round 3 was not due to steric hindrance from previous CTX-Clv modifications. Moreover, these results show that CTX-Clv can accurately measure (within 5%) different subpopulations of ion channel complexes at the cell surface. Thus, the counting results in Fig. 4a demand that the vast majority (>95%) of the Q1 K⁺ channel complexes have only two KCNE peptides, resulting in complexes that are not fourfold symmetric.

Discussion

The 4:2 (α : β) stoichiometry of the Q1/E1 complex has several broad implications for KCNE modulation of voltage-gated K⁺ channels. If modulation is due to a direct KCNE–K⁺ channel protein–protein interaction, then KCNE peptides either interact

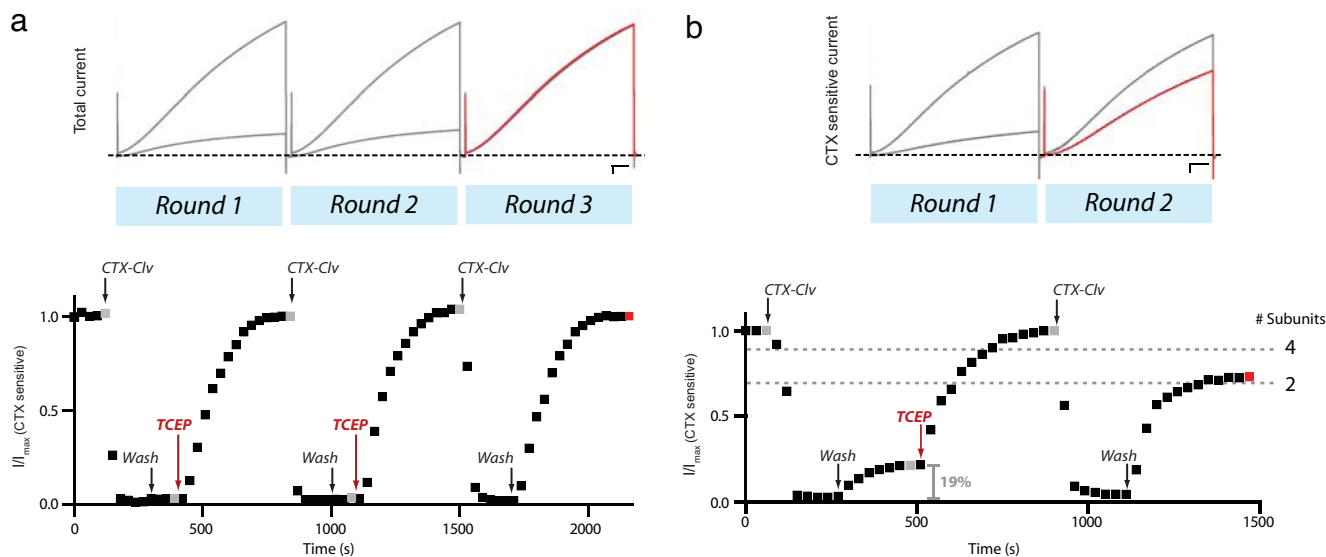


Fig. 4. KCNQ1 K^+ channel complexes contain two KCNE1 peptides. (a) Iterative counting of E1_{T14C} subunits in a Q1 K^+ channel complex. (Upper) Raw data showing the total current for each round. Scale bar is $0.5 \mu A$ and $0.5 s$. (Lower) Q1/E1_{T14C} complexes were treated with two rounds (10 nM CTX-Clv, washout, 1 mM TCEP) of treatment before CTX-Clv binding became reversible. Oocytes were depolarized every 30 s and the data points were obtained at the end of a 5-s pulse. Gray and red data points correspond to the raw traces above. (b) Only one KCNE peptide reacts per CTX-Clv treatment. (Upper) CTX-sensitive current for each round. Scale bar is $0.5 \mu A$ and $0.5 s$. (Lower) Chemical modification of 19% of the cell surface Q1/E1_{T14C} complexes with $10 \mu M$ maleimide *N*-ethylsulfonate, sodium salt (Mal-ES), predicts that the reversibility of CTX-Clv in the second round will be 69% if one KCNE peptide reacts per round; 88% if two KCNE are modified per round (dotted lines, which correspond to two or four KCNE subunits in the K^+ channel complex). Washout of CTX-Clv after the second round of treatment results in $65 \pm 4\%$ (SEM) reversibility ($n = 3$). I_{max} is the CTX-sensitive current that was normalized before chemical treatment. Gray and red data points correspond to the CTX-sensitive currents above.

with only two of the α -subunits in a K^+ channel complex or they bifurcate adjacent α -subunits, making two unique sets of protein contacts with all four ion-conducting subunits. A recent mutagenic perturbation study of the KCNE1 transmembrane domain suggests that a stretch of residues 56–63 alternately interacts with two protein faces of the KCNQ1 channel, providing a potential bifurcation site (23). Alternatively, KCNE modulation of K^+ channels could be allosteric. Likewise, mutant KCNE peptides that produce diseased K^+ channel complexes will impart their deleterious effects on ion conduction and channel gating by the molecular mechanisms outlined above. Interpreting the effects of KCNE peptides on K^+ channel function is made even more complex with the discovery that a single Q1 channel can assemble with two different KCNE peptides to form a functional heteromeric complex (16). However, because each K^+ channel complex contains only two KCNE peptides, the number of potential heteromeric complexes that can form is greatly reduced.

How then does a K^+ channel, which is destined to be tetrameric when expressed alone, incorporate only two modulatory KCNE peptides? Some insight into the Q1/E1 assembly mechanism comes from the study of K^+ channel α -subunit biogenesis, which has shown that individual subunits first form dimers, then subsequently dimerize to form a tetrameric ion channel (24). In addition, kinetic modeling indicates that tetrameric proteins exclusively use a dimer-of-dimers pathway because other assembly pathways would lead to substantial amounts of partially assembled, dead-end intermediates (25). Therefore, KCNE peptides may not be breaking the symmetry of the K^+ channel complex, but are instead infiltrating the dimeric assembly processes of the tetrameric ion conduction pore. For Kv-type channels, the N-terminal T1 domain tetramerizes while the nascent polypeptides are still attached to ribosomes (26). Thus, for this class of K^+ channels, KCNE peptides would have to coassemble with channel subunits very early in biogenesis to use the nascent dimeric structures. For KCNQ1 channels, coassem-

bly could occur later in biosynthesis because the subunit specificity domain for tetramerization resides in the C terminus (27). In either case, to exploit the dimer-of-dimers pathway, coassembly would have to occur in the endoplasmic reticulum, which is consistent with previous assembly studies with wild-type (28) and mutant KCNE1 peptides (29).

CTX-Clv and the iterative cell surface modification approach have enabled us to definitively determine the stoichiometry of the Q1/E1 K^+ channel complex. Moreover, this counting approach has the capacity to identify and measure subpopulations of stoichiometric mixtures, which may be operational for other K^+ channel complexes. The structural conservation of the outer vestibule of K^+ channels allows CTX sensitivity to be transplanted into any K^+ channel of interest, a strategy that we have used here, and should be applicable for most K^+ channel complexes. In addition, the modular design of the reagent allows for the incorporation of different targeting ligands and chemical modifying agents. Given the vast assortment of high-affinity scorpion and spider toxins available, this iterative chemical approach can be adapted to identify and enumerate the protein subunits in a variety of multimeric ion channel complexes.

Materials and Methods

Chemical Synthesis. Details describing the synthesis and characterization of all chemical compounds can be found in *SI Materials and Methods*.

Mutagenesis and *in Vitro* Transcription. For optimal CTX binding to the Q1/E1 complex, CTX-sensitive Q1 and hemagglutinin-tagged E1 constructs were used (6) and subcloned into a vector containing the 5' and 3' UTRs from the *Xenopus* β -globin gene for optimal expression in *Xenopus* oocytes. Site-directed mutagenesis and mRNA *in vitro* transcription were performed as described in ref. 16.

Electrophysiology. Standard techniques and solutions were used for the preparation of and recording from *Xenopus* oocytes by two-electrode voltage clamp (30–34). Oocytes were injected with a Q1:E1 mRNA ratio of 15:5 ng to ensure that E1 was in excess. Oocytes were incubated for 3–5 days with 3 mM

L-glutathione and pretreated with 1 mM TCEP for 15 min before recording, and the current was measured by holding at -80 mV and pulsing to $+20$ mV for 5 s every 30 s.

Cell Surface Modifications. Labeling reagents CTX-Clv and CTX-Mal (10 nM) were in ND96 (30) solution with $50 \mu\text{g}\cdot\text{ml}^{-1}$ BSA and were added to the bath by a gravity-fed perfusion system with a chamber clearing time of ≈ 10 s. CTX-Clv cleavage was accomplished by 1 mM TCEP for ≈ 120 s. The recovery time constant (τ_{recovery}) was measured by fitting the return of current after CTX-Clv washout to a single exponential. To randomly label cell surface cysteines with Mal-ES (19), TCEP-pretreated oocytes were incubated with $10 \mu\text{M}$ Mal-ES in ND96 for 500 s.

Synthesis of CTX-Clv. Recombinant CTX R19C was purified and the methanethiosulfonate ethyltrimethylammonium (MTSET)-protected disulfide of CTX R19C was prepared as described in ref. 15. CTX-Clv was synthesized as follows: 16 nmol of CTX-MTSET in 2 ml of low-salt buffer (10 mM NaCl, 10 mM KPi , pH 7.4) was reduced with 1 mM dithiothreitol (DTT) for 45 min. Reduced CTX was purified by using a C_{18} HPLC column (4.6 mm \times 250 mm) and eluting with solvent A [0.1% trifluoroacetic acid (TFA)] and solvent B (acetonitrile), forming a gradient of 10–40% B over 30 min. The fractions containing reduced CTX were adjusted to pH 7.0 by using 1 M KPi , pH 7.1, and a solution of $16 \mu\text{mol}$ of bismaleimide **1b** in $90 \mu\text{l}$ of acetonitrile was slowly added and allowed to react for 30 min at room temperature. The reaction mixture was placed on ice for 10 min to precipitate excess unreacted bismaleimide, which was removed by filtration (GHP Acrodisc $0.45 \mu\text{m}$, Pall Gelman Laboratory). CTX-Clv was HPLC purified as described above, using a 20–50% gradient B over 30 min. The concentration of purified CTX-Clv was determined by UV spectrometry (A_{280} of $1.0 = 100 \mu\text{M}$ CTX) (14) and labeling efficiency of CTX-R19C was determined to be $30 \pm 8\%$, $n = 7$). The purified product was confirmed by electrospray ionization mass spectrometry (SI Fig. 6) and was aliquoted, lyophilized, and stored at -20°C . Immediately before use, individual aliquots were resuspended in ND96 containing 30% acetonitrile to ensure CTX-Clv went into solution. This solution was then diluted such that the final acetonitrile concentration was 0.3% and the final CTX-Clv concentration was 10 nM.

Appendix

The distribution of modified KCNE-K⁺ channel complexes generated by a partial treatment of Mal-ES can be defined by the probability mass function if each modification occurs independently:

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$$F(k;n,p) = \binom{n}{k} p^k (1-p)^{n-k},$$

where n is the number of KCNE peptides, k is the number of modified KCNE peptides, and F is the fraction of K⁺ channel complexes with all KCNE peptides modified when $k = n$. If one KCNE is modified per CTX-Clv treatment cycle, then Fig. 4a demands that there are two KCNE peptides in the complex. Therefore, $p = 0.44$ when $k = 2$, $n = 2$ and $F = 0.19$, which was experimentally determined after CTX-Clv washout (Fig. 4b). The complementary cumulative distribution function predicts the reversibility in the second CTX-Clv washout:

$$R(x;n,p) = 1 - \sum_{k=0}^x \binom{n}{k} p^k (1-p)^{n-k},$$

where x is equal to the maximum number of Mal-ES modifications a K⁺ channel complex can have such that there is at least one modifiable cysteine remaining in the second cycle. For $n = 2$, $x = 0$ because all K⁺ channel complexes will have one CTX-Clv modification, which occurred during the measuring of F . Thus, the expected reversibility for one CTX-Clv modification per cycle with two KCNE peptides in a K⁺ channel complex ($x = 0$; $n = 2$, and $p = 0.44$) is $R = 0.69$. Conversely, if two KCNE modifications occur per exposure to CTX-Clv, then each complex must have four KCNE peptides (Fig. 4a) and $p = 0.66$ when $k = 4$, $n = 4$, and $F = 0.19$. In this scenario, $x = 1$ for the cumulative distribution function because only complexes with less than two Mal-ES modifications will have a modifiable cysteine in the second round of treatment. Thus, the expected reversibility for two CTX-Clv modifications per cycle with a four KCNE peptide-K⁺ channel complex ($x = 1$; $n = 4$, and $p = 0.66$) is $R = 0.88$. Stoichiometries with an odd number of KCNE peptides were not considered because they would require partial CTX-Clv irreversibility, which was not observed (Fig. 4a). Integer modifications greater than two KCNE peptides are possible; however, these would result in $R > 0.95$ when $F = 0.19$.

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