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## Tandem MS can Distinguish Hyaluronic Acid from N-Acetylheparosan

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### Abstract

Isobaric oligosaccharides enzymatically prepared from hyaluronic acid (HA) and *N*-acetylheparosan (NAH), were distinguished using tandem mass spectrometry. The only difference between the two series of oligosaccharides was the linkage pattern (in HA 1→3 and in NAH 1→4) between glucuronic acid and *N*-acetylglucosamine residues. Tandem mass spectrometry afforded spectra, in which glycosidic cleavage fragment ions were observed for both HA and NAH oligosaccharides. Cross-ring cleavage ions <sup>0,2</sup>A<sub>n</sub> and <sup>0,2</sup>A<sub>n</sub>-h (n is even number) were observed only in GlcNAc residues of NAH oligosaccharides. One exception was an <sup>0,2</sup>A<sub>2</sub> ion fragment observed for the disaccharide from HA. These cross-ring cleavage fragment ions are useful to definitively distinguish HA and NAH oligosaccharides.

### Keywords

hyaluronic acid; *N*-acetylheparosan; oligosaccharide; tandem mass spectrometry; fragmentation

### Introduction

Glycosaminoglycans (GAGs) are linear, acidic polysaccharides found on cell surfaces and in the surrounding extracellular matrix. GAGs participate in and regulate many cellular events and physiological and pathophysiological processes, such as cell proliferation and differentiation, cell-cell and cell-matrix interactions, viral infection through their interaction with different proteins [1-3]. GAGs are divided into four main categories: hyaluronic acid, or hyaluronan (HA), chondroitin/dermatan sulfate, *N*-acetylheparosan (NAH)/heparan sulfate/heparin, and keratan sulfate, based on their monosaccharide composition and the configuration and position of the glycosidic bonds between these monosaccharides. The specificity of the interactions between GAGs and proteins results from structural diversity of GAGs defined by their size, saccharide composition and sequence, charge density [4,5]. Thus, understanding the structure of a GAG is essential in understanding its activity and biological functions. HA and NAH are biosynthesized by both prokaryotic and eukaryotic cells. HA is co-polymer linked polymer of β-1,4 D-glucuronic acid (GlcA) and β-1,3 D-*N*-acetylglucosamine (GlcNAc). [6]

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NAH is a co-polymer of  $\beta$ -1,4 GlcA and  $\alpha$ -1,4 GlcNAc and the biosynthetic precursor of the mammalian GAGs, heparin and heparan sulfate [7]. HA can be prepared by bacterial fermentation of *Streptococcus zooepidemicus* [8], or enzymatically using biosynthetic enzymes prepared from *Pasturella multocida* [9]. Bacteria use HA as camouflage to enhance their ability to infect higher animals [10]. Similarly, *Escherichia coli* strain K5 produces capsular polysaccharide comprised of NAH [11]. NAH is also important as it represents a considerable portion of the sequence of mammalian heparan sulfate [12,13]. HA and NAH are the two simplest GAGs since neither have sulfo groups and, thus, consist of only a single sequence (Figure 1). HA and NAH have very similar structures, differing only in the position of their glycosidic linkage between GlcA and GlcNAc. This difference in structure results from very different biosynthesis pathways for HA and NAH [14-17] and gives each polysaccharide unique biological functions and distinctive structure-activity relationships [18].

The oligosaccharides prepared from GAGs, by controlled enzymatic depolymerization, are often studied to define the minimum structural requirements for biological activity [19,20]. ESI-MS is particularly useful to monitor these GAG-derived oligosaccharides due to its soft ionization [21,22]. LC-ESI-MS, relying on reversed-phase ion-pairing (RPIP)-high performance liquid chromatography (HPLC), has employed volatile ion-pairing reagents [23, 24] to successfully analyze HA and NAH [25]. Unfortunately, this method results in nearly identical retention times for the HA and NAH oligosaccharides and since they have identical masses, it is not possible to differentiate these oligosaccharides. Most laboratories routinely rely on nuclear magnetic resonance (NMR) spectroscopy, requiring relatively large amounts of samples, to distinguish HA and NAH oligosaccharides [26,27]. These two series of oligosaccharides can also be differentiated based on their susceptibilities to specific polysaccharide degrading enzymes, where the degradation can be detected by ultraviolet spectroscopy (UV), HPLC, thin layer chromatography (TLC), polyacrylamide gel electrophoresis or change in viscosity. A new method, which could confirm the identity of polysaccharide made by small samples of new isolates containing both HA and NAH, would be useful.

Tandem mass spectrometry offers a sensitive and more readily available alternative to NMR for the analysis of GAG oligosaccharides. ESI tandem MS has been previously used to analyze HA oligosaccharides and glycosidic fragment ions were previously observed with single or multiple charges [28,29]. Unfortunately, these fragment ions would be expected to form from both HA and NAH oligosaccharides.

This article describes a method for distinguishing HA and NAH oligosaccharide mixtures prepared by partial enzymatic digestion. RPIP-LC-ESI-MS, and tandem MS are applied to analyze these oligosaccharides and distinguish HA and NAH oligosaccharides based on the relative abundance of their characteristic fragment ions.

## Materials and Methods

### Preparation of oligosaccharides mixture

*N*-acetylheparosan (NAH) was isolated through fermentation of *E. coli* K5 and purified as previously described [11]. The NAH 500  $\mu$ g/50  $\mu$ l was incubated in 50 mM sodium phosphate buffer pH 7.0 with heparin lyase III (10 m-units, Sigma Chemicals, St Louis, MO, U.S.A.) at 37°C for 10 h with aliquots removed at various time points for analysis. The aliquots were heated in a boiling water bath for 10 min. to halt the reaction. The denatured protein was removed by centrifugation at 12000  $\times$  g for 10 min. Hyaluronic acid (HA, from rooster comb) was purchased from Sigma. Chondroitin lyase ABC (5 m-units, Seikagaku Biochemical, Tokyo, Japan) was used to digest HA (500  $\mu$ g/50  $\mu$ l) in 50 mM sodium phosphate buffer pH 7.0 at 37°C for 10 h and aliquots were removed periodically and collected as described above.

## TLC conditions

The progress of the enzymatic depolymerization was monitored by measuring the HA and NAH oligosaccharides present in aliquots taken throughout the reaction by TLC on a precoated silica gel-60 aluminum plates (Merck, Darmstadt, Germany) (1 × 5 cm) or (10 cm × 5 cm) eluted with a solvent system consisting of n-butanol/formic acid/water, 4:8:1 (v/v). The eluted plate was stained by dipping in a reagent containing 1 ml of 37.5% HCl, 2 ml of aniline, 10 ml of 85% H<sub>3</sub>PO<sub>3</sub>, 100 ml of ethyl acetate and 2 g diphenylamine) for 3 sec and heating at 150 °C for 10 sec. [30].

## LC-MS/MS

LC MS analyses were performed on Agilent 1100 LC/MSD instrument (Agilent Technologies, Inc. Wilmington, DE, U.S.A.) equipped with an ion trap, binary pump and a UV detector. The column was a 5 μm Agilent Zorbax SB-C18 (0.5 × 250 mm) from Agilent Technologies. Eluent A was water/acetonitrile (85:15), v/v and eluent B was water/acetonitrile (35:65) v/v. Both eluents contained 12 mM tributylamine (TBA) and 38 mM NH<sub>4</sub>OAc and their pH was adjusted to 6.5 with HOAc. The product mixtures of the HA and NAH enzyme digestions were each diluted 10-fold with water and 1 μl of the resulting analytes were individually injected by auto-sampler. A gradient of 0% B for 15 min, and 0-100% B over 85 min. was used at a flow rate of 10 μl/min. Mass spectra were obtained using an Agilent 1100 series Classic G2445D LC/MSD trap (Agilent Technologies, Inc. Wilmington, DE, U.S.A.). The electrospray interface was set in negative ionization mode with the skimmer potential -40.0 V, capillary exit -120 .5 V and a source temperature of 325 °C to obtain maximum abundance of the ions in a full scan spectra (150–1500 Da, 10 full scans/s). Nitrogen was used as a drying (5 liters/min) and nebulizing gas (20 p.s.i.). Auto MS/MS was turned on in these experiments using an estimated cycle time of 0.07 min. Total ion chromatograms (TIC) and mass spectra were processed using Data Analysis 2.0 (Bruker software).

## Results and Discussion

Chondroitin lyase ABC was used to digest HA and heparin lyase III was used to digest NAH and both digestions were carried out to ~30% completion. The extent of digestion, reflected in the value “m” (at 100 % digestion m = 0) was monitored by TLC (Figure 2A). The oligosaccharides resulting from HA had the general structure ΔUA (1[→3] GlcNAc (1→4) GlcA (1→3]<sub>m</sub>) GlcNAc (ΔUA, 4-deoxy-α-L-threo-hex-4-enopyranosyluronic acid, m = 0,1,2...). The oligosaccharides resulting from NAH had the general structure ΔUA (1[→4] GlcNAc (1→4) GlcA (1→4]<sub>m</sub>) GlcNAc.

Next, LC-ESI-MS analysis was performed on HA and NAH oligosaccharides obtained at 30% reaction completion (Figure 2B and C). Peaks corresponding to a degree of polymerization (dp) from 2 to 20 were observed in the mass spectra of both HA and NAH oligosaccharides (Table 1). The ESI-MS spectra of HA and NAH oligosaccharides were identical across a range of dp values. For example, the largest oligosaccharides from HA and NAH (dp 20) afforded a calculated mass of 3791.1 based on the fully protonated acidic form. The mass spectra of both showed identical multiply charged molecular-ion peaks [M-5H]<sup>5-</sup> and [M-4H]<sup>4-</sup> at *m/z* 757.3 and 946.9, respectively (Figure 3), confirming a length of 20 saccharide units.

Tandem mass spectrometry can result in both glycosidic linkage and cross-ring cleavages. All product ions are labeled according to the Domon and Costello nomenclature. [31] The cross-ring cleavages hold the key in distinguishing HA and NAH oligosaccharides. Tandem mass spectra of the smallest products, the two disaccharides (dp2) from HA and NAH, were examined to see if they could be distinguished (Figure 2A and B). Tandem MS, on the singly-charged precursor at *m/z* 378.1, afforded a cross-ring cleavage ion fragment <sup>0,2</sup>A<sub>2</sub> at *m/z* 276.9

in both the HA and NAH disaccharides but also a unique ion fragment  $^{0,2}A_2\text{-h}$  at  $m/z$  258.9 in the spectrum of NAH disaccharide, corresponding to a loss of  $H_2O$  from  $^{0,2}A_2$  (Figure 4).

The tandem mass spectra of two tetrasaccharides (dp4) were next examined (Figure 5 A and B) to determine whether this unique fragment could be used to distinguish oligosaccharides derived HA and NAH. Glycosidic cleavage fragment ions were observed, from the singly charged the precursors molecular ions at  $m/z$  757.1, for both HA and NAH tetrasaccharides. Again, the cross-ring cleavage ion fragments  $^{0,2}A_2\text{-h}$   $m/z$  258.9 and  $^{0,2}A_4\text{-h}$   $m/z$  638.1 were observed only in the NAH. When the tandem MS was performed on the doubly charged molecular ions of the two hexasaccharides (dp6) at  $m/z$  576.8 (Figure 6A and B), the fragments observed were similar to those observed in the tandem mass spectra of the dp 4  $[M-H]^-$  precursor. In addition to the glycosidic cleavage ion fragments, observed in tandem mass spectra of both HA and NAH hexasaccharides, A-type ion fragments  $^{0,2}A_2$ ,  $^{0,2}A_4$ ,  $^{0,2}A_6$ ,  $^{0,2}A_2\text{-h}$ ,  $^{0,2}A_4\text{-h}$  and  $^{0,2}A_6\text{-h}$  were also observed in the tandem mass spectrum of the NAH hexasaccharide. The  $^{0,2}A_6$  and  $^{0,2}A_6\text{-h}$  fragments were mainly observed as doubly charged ions at  $m/z$  517.1 and  $m/z$  508.1, respectively. Cross-ring cleavage ion fragments were not observed in spectrum of HA hexasaccharides. In the spectra of HA tetrasaccharide and hexasaccharide, the relative intensities of fragment ions were different. In the spectrum of HA hexasaccharide, the  $C_3$  fragment ion at  $m/z$  554.1 is of much lower intensity than the  $C_3$  fragment ion in the spectrum of HA tetrasaccharide (Figures 5A and 6A). The  $C_4/Z_4$  fragment ion at  $m/z$  757.2 dominates the spectrum of NAH hexasaccharide, while all the peaks in the spectrum of NAH tetrasaccharide are of similar relative intensities (Figure 5B and 6B). These differences between tetrasaccharide and hexasaccharides would appear to suggest that the charge state influences the product ion pattern. All ion fragments in the tandem mass spectra of dp2 to dp12, derived from HA and NAH, are presented in Table 2 and Table 3, respectively. Tandem MS was performed on singly or multiply charged molecular ions (labeled with asterisks in Table 1).  $^{0,2}A$ -type ion fragments were observed in the GlcNAc residue of NAH oligosaccharides, but were not observed in the GlcNAc residues of HA oligosaccharides. The larger oligosaccharides of HA and NAH from dp14 to dp20 shown in the TIC (Figure 2) have a similar fragmentation to that of smaller oligosaccharides, dp2 to dp 12, in which the difference between HA and NAH also could be observed (data not shown). Interestingly, only dp 2 of HA gave an  $^{0,2}A$  fragment (but not an  $^{0,2}A\text{-h}$  fragment), making  $^{0,2}A\text{-h}$  fragment a unique marker for NAH.

Ion fragments, of the  $^{0,2}A$ -type observed in the NAH GlcNAc residues, containing 1→4 linkage suggest the fragmentation mechanism is a retro-aldol rearrangement of oligosaccharide ions in the negative mode [32-35]. This mechanism requires an open-ring reducing terminal aldehyde, and such structures are formed by C-type fragmentation [34-38]. The  $^{0,2}A_n$  ( $n$ =even number) ions formed from cross-ring cleavage to the internal glucosamine residues are therefore likely to arise from the  $C_n$  ( $n$ =even number) ions by retro-aldol rearrangement. These A-type fragment ions are observed in residues that are 4- or 6- linked [21,34,36-42]. Furthermore, in NAH oligosaccharides, the free hydroxy group at position 3 can accept a proton from position 4 after cleavage of the GlcNAc ring by  $^{0,2}A$  type fragmentation (Figure 7) affording water and the negatively charged  $^{0,2}A_n\text{-h}$  fragment. These dehydrated A-type fragment ions also are observed with the residue requiring an unsubstituted hydroxyl in the 3-position in other oligosaccharides [34,36,39,43]. In the HA oligosaccharides, the bonds at C-3 of GlcNAc are substituted perhaps making these more stable than NAH oligosaccharides. The exception to this rule is dp2 from HA, where the  $^{0,2}A_2$  ion fragment is observed. This may result from the small size of the dp 2 molecule, preventing it from dissipating vibrational energy, resulting in its fragmentation. Despite its small size, the HA disaccharide still fails to afford a  $^{0,2}A_n\text{-h}$  fragment ion because of its substitution at position 3. Here, the presence of  $^{0,2}A$ -type fragment ions can be used as a diagnostic tool to distinguish HA disaccharide from longer HA oligosaccharides, in cases where the charge state is unknown.

In conclusion, the current study clearly demonstrates LC-MS/MS can be used to differentiate HA and NAH oligosaccharides. The sensitivity of this method is better than 100 ng for each oligosaccharide separated using RPIP-HPLC. A-type fragment ions from GlcNAc residues were shown as diagnostic peaks in MS/MS spectra of NAH oligosaccharides, but not in the spectra of HA oligosaccharides. One exception was an  $^{0,2}A_2$  ion fragment observed for the disaccharide from HA. This approach should be useful for screening the capsules of bacterial isolates to determine whether they are composed of HA or NAH polysaccharides. This method might also be useful in distinguishing between small quantities of HA and NAH derived oligosaccharides isolated from animal tissues. This method should also be useful as a complement approach to confirm the structure of unsulfated chondroitin oligosaccharides that have the same linkage pattern as HA, containing 3-linked HexNAc residues.

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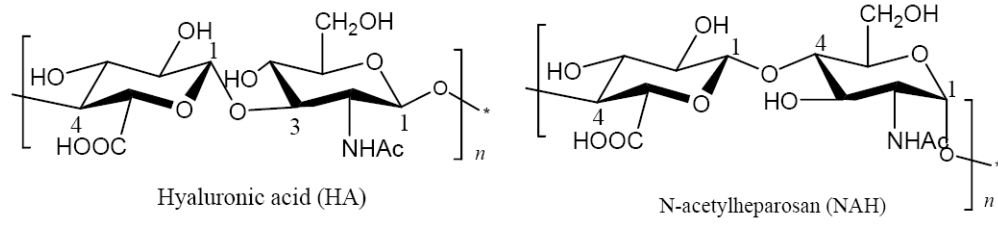
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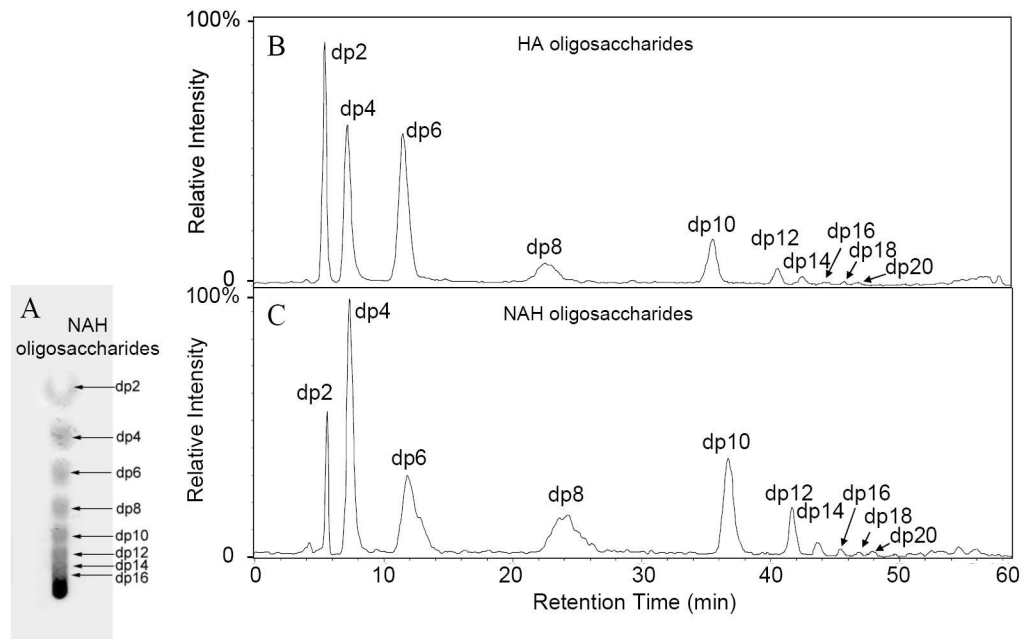
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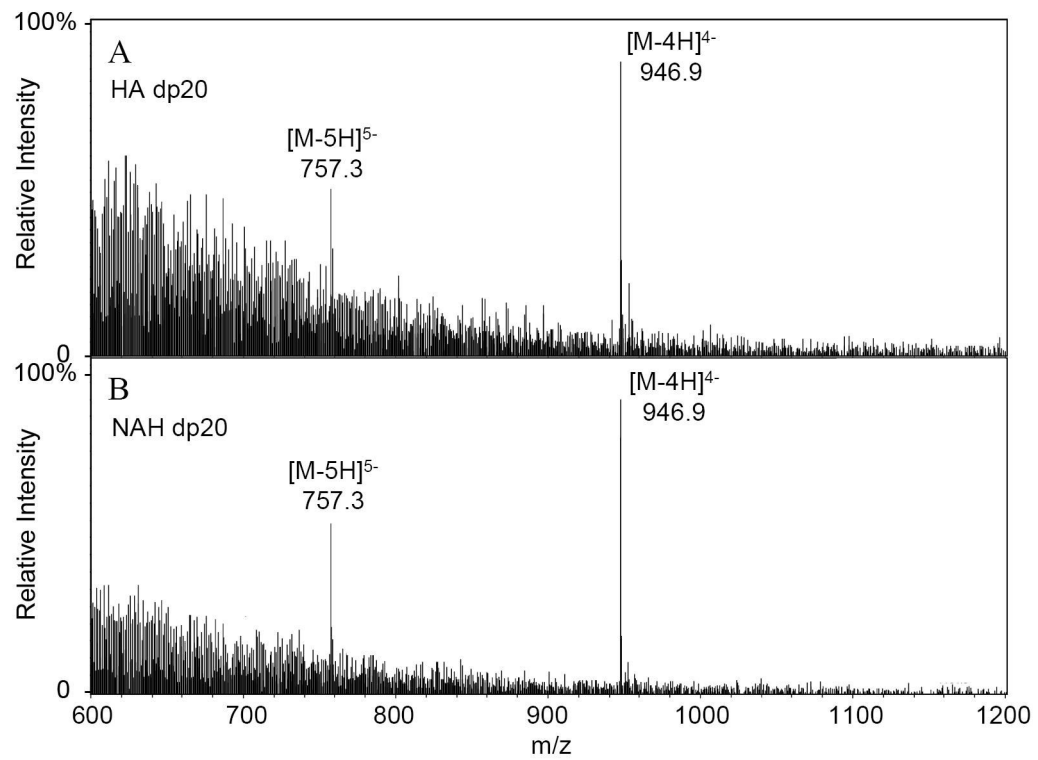


**Figure 1.**  
Structure of HA and NAH.

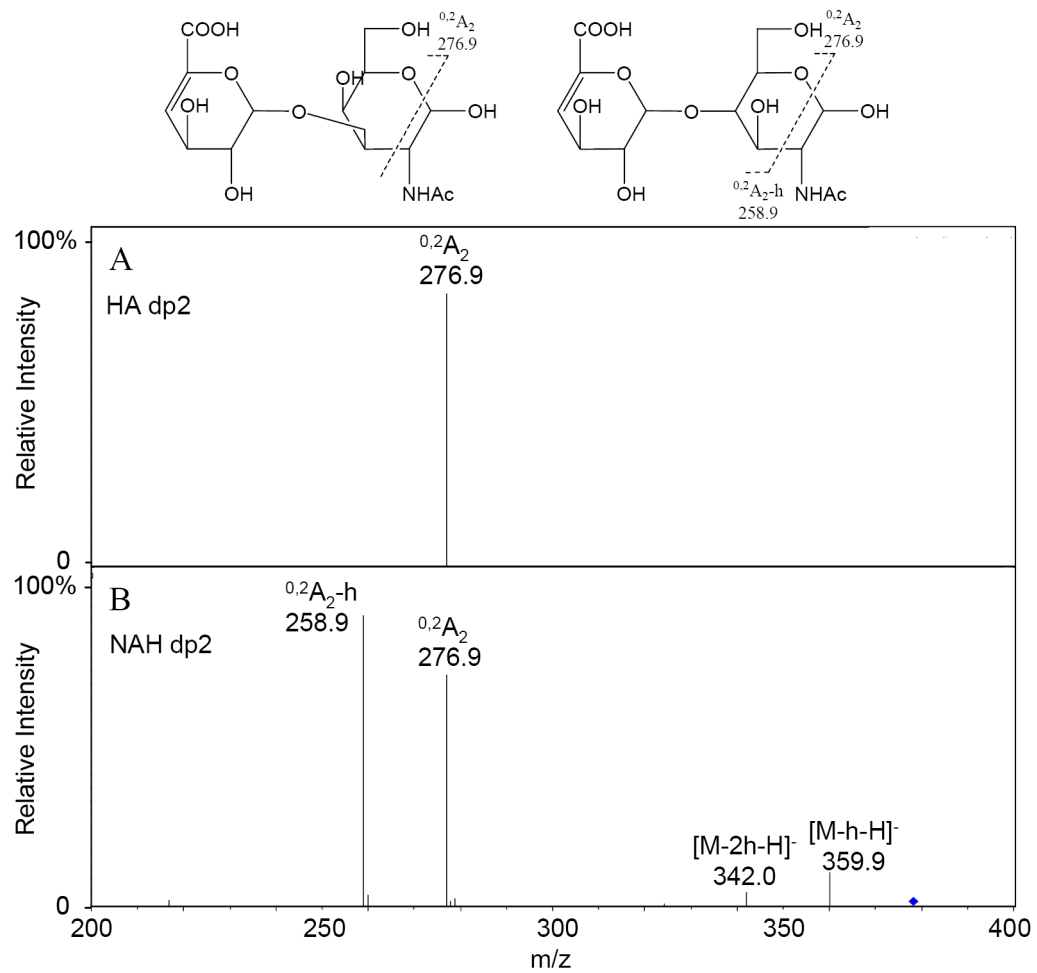




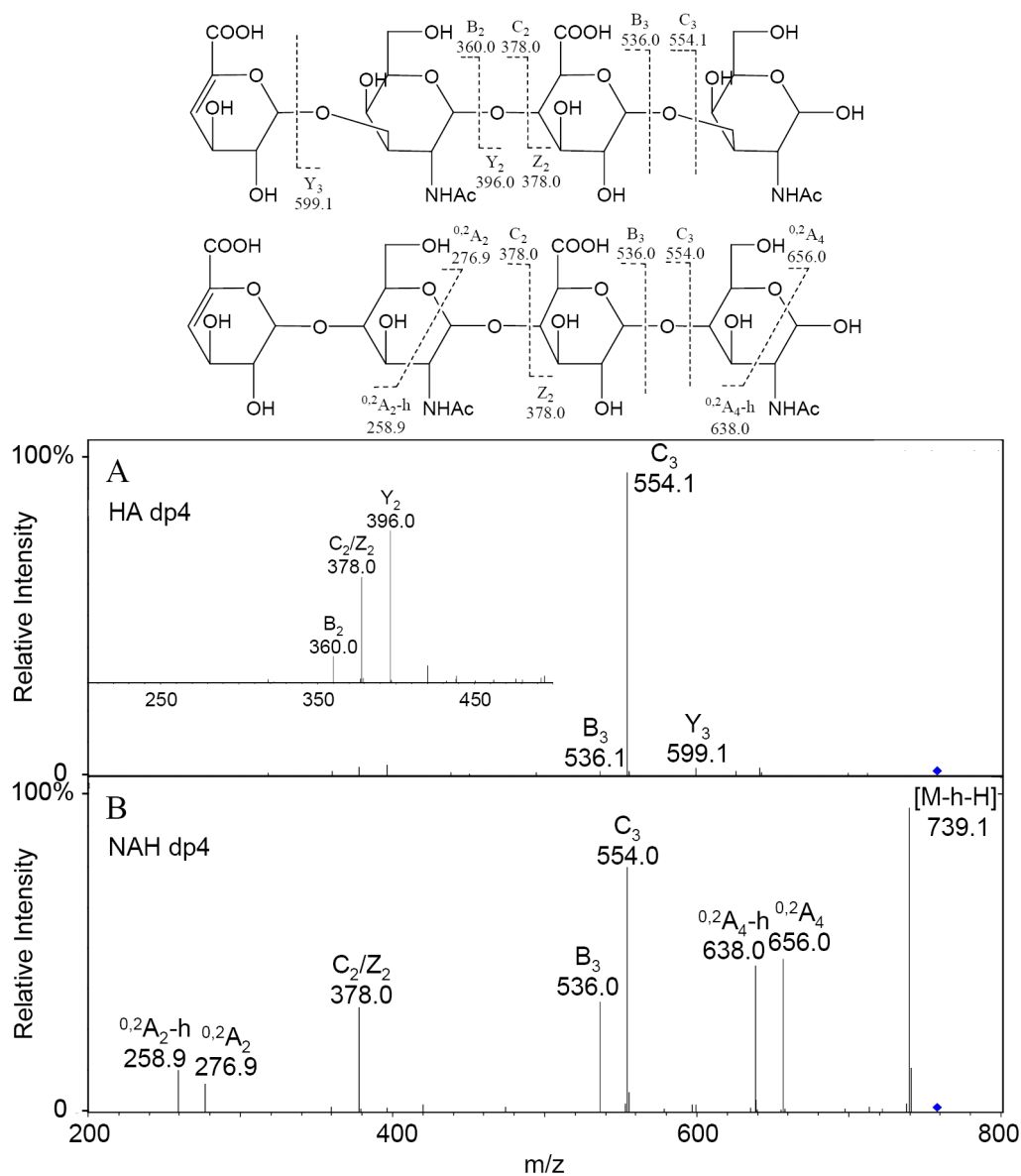
**Figure 2.** Total ion chromatogram (TIC) of HA oligosaccharides (A). TLC analysis of NAH oligosaccharides (B). TIC of NAH oligosaccharides (C).



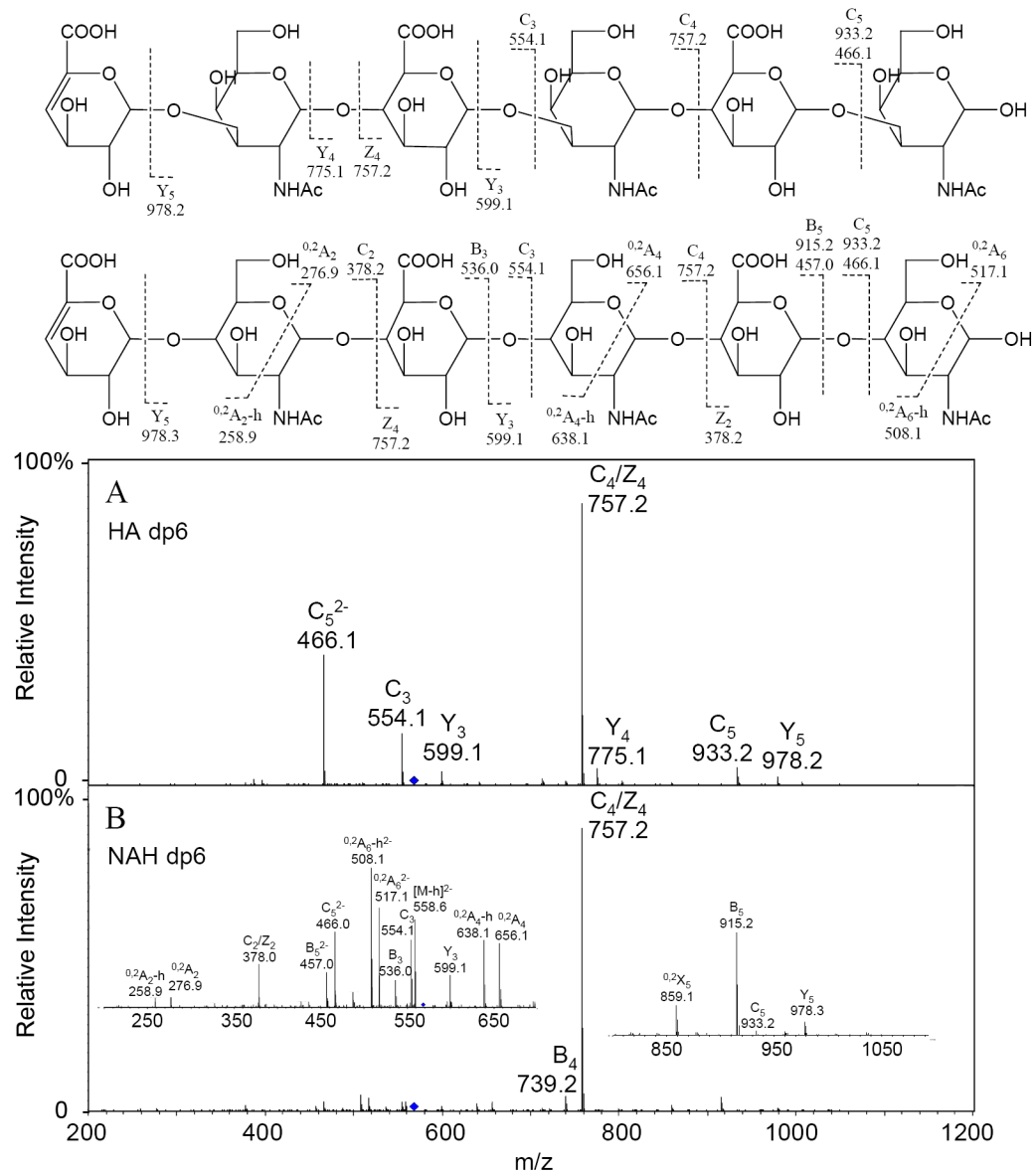
**Figure 3.**  
ESI mass spectra of dp20 of HA (A) and NAH (B).



**Figure 4.** MS/MS spectra of disaccharides (dp2) from HA (A) and NAH (B).



**Figure 5.** MS/MS spectra of tetrasaccharides (dp4) from HA (A) and NAH (B).



**Figure 6.** MS/MS spectra of hexasaccharides (dp6) from HA (A) and NAH (B).

Table 1

Major molecular ions observed in the ESI mass spectra of HA and NAH oligosaccharides.

	Calculated mass	[2M-H] <sup>-</sup>	[M-H] <sup>-</sup>	[M-2H] <sup>2-</sup>	[M-3H] <sup>3-</sup>	[M-4H] <sup>4-</sup>	[M-5H] <sup>5-</sup>
dp2	379.1		<b>378.1</b> *				
dp4	758.2	757.2	<b>757.1</b> *				
dp6	1137.3		1136.3	<b>567.8</b> *			
dp8	1516.4			<b>757.2</b> *	630.9		
dp10	1895.5			<b>946.8</b> *	<b>757.3</b> *		
dp12	2274.7			1136.8	<b>883.8</b> *		
dp14	2653.8			1326.8	1010.3		
dp16	3032.9				1136.9	<b>757.4</b> *	681.5
dp18	3412.0					<b>852.2</b> *	
dp20	3791.1					<b>947.2</b> *	757.3

\* Fragments marked were used as precursor ions in tandem MS.

Table 2

Fragment ions observed in the product-ion spectra of HA oligosaccharides

	dp2	dp4	dp6	dp8	dp10	dp12
<sup>0,2</sup> A <sub>n</sub> *	276.9					
<sup>0,2</sup> A <sub>n</sub> -h						
B <sub>n-1</sub>	536.1	915.2			1497.5	
B <sub>n-2</sub>	359.9	739.2			1294.3	
B <sub>n-3</sub>				915.2	1118.3	1497.3
B <sub>n-4</sub>				739.1	915.2	1294.3
B <sub>n-5</sub>				536.8	739.1	1118.3
B <sub>n-6</sub>						915.2
B <sub>n-7</sub>					845.5	1035.5
C <sub>n-1</sub>	554.1	933.2/466.1		1312.3/655.7	845.5	1035.5
C <sub>n-2</sub> /Z <sub>n-2</sub>	378.0	757..2		1136.3	757.2	947.2
C <sub>n-3</sub>		554.1		933.2	1312.3	845.6
C <sub>n-4</sub> /Z <sub>n-4</sub>		378.0		757.1	1136.3	757.1
C <sub>n-5</sub>				554.1	933.2	1312.3
C <sub>n-6</sub> /Z <sub>n-6</sub>					757.2	1136.3
C <sub>n-7</sub>						933.2
C <sub>n-8</sub> /Z <sub>n-8</sub>						757.2
Y <sub>n-1</sub>	599.1	978.2				
Y <sub>n-2</sub>	396.0	775.1		1154.3	1375.4	
Y <sub>n-3</sub>		599.1		978.3	1154.3	
Y <sub>n-4</sub>		396.0		775.1	978.3	1375.4
Y <sub>n-5</sub>				599.1	775.2	1154.3
Y <sub>n-6</sub>						978.3
Y <sub>n-7</sub>						775.2
Y <sub>n-8</sub>						

\* dp = 2 n

Table 3

Fragment ions observed in the product-ion spectra of NAH oligosaccharides

dp n	dp2	dp4	dp6	dp8	dp10	dp12
[M-h] <sup>+</sup> *	359.9	739.1	558.1	748.2	937.8	
0.2A <sub>n</sub> <sup>-</sup>	276.9	656.1	517.1	706.6	896.3	723.7
0.2A <sub>n</sub> <sup>-h</sup>	258.9	638.1	508.1	697.7	887.3	717.7
0.2A <sub>n-2</sub> <sup>-h</sup>		276.8	656.1	1035.1/517.1	1414.4/706.6	896.7
0.2A <sub>n-4</sub> <sup>-h</sup>		258.8	638.1	1017.1/508.1	1396.4/697.7	887.7
0.2A <sub>n-4</sub> <sup>-h</sup>			276.8	656.1	1035.1/517.1	1414.4/706.6
0.2A <sub>n-4</sub> <sup>-h</sup>			258.8	638.1	1017.1/508.1	1396.4/697.7
0.2A <sub>n-6</sub> <sup>-h</sup>				276.8	656.1	1035.1/517.1
0.2A <sub>n-6</sub> <sup>-h</sup>				258.8	638.1	1017.1/508.1
0.2A <sub>n-8</sub> <sup>-h</sup>					276.8	656.1
0.2A <sub>n-8</sub> <sup>-h</sup>					258.8	638.1
0.2A <sub>n-10</sub> <sup>-h</sup>						276.8
0.2A <sub>n-10</sub> <sup>-h</sup>						258.8
B <sub>n-1</sub>		536.0	915.2/457.1			
B <sub>n-2</sub>			739.2	1118.3	1497.4/748.8	
B <sub>n-3</sub>			536.2	915.2	1294.3/646.2	
B <sub>n-4</sub>				739.1	1118.3	1497.4
B <sub>n-5</sub>				536.0	915.2	1294.3
B <sub>n-6</sub>					739.1	1118.2
B <sub>n-7</sub>					536.1	915.2
B <sub>n-8</sub>						739.1
B <sub>n-9</sub>						536.1
C <sub>n-1</sub>		554.1	933.2/466.1	1312.3/655.7	845.8	
C <sub>n-2</sub> /Z <sub>n-2</sub>		378.0	757.2	1136.3	757.2	947.2
C <sub>n-3</sub>			554.1	933.2	1312.3/656.2	845.8
C <sub>n-4</sub> /Z <sub>n-4</sub>			378.0	757.1	1136.3	757.1
C <sub>n-5</sub>				554.1	933.2	1312.3/656.1
C <sub>n-6</sub> /Z <sub>n-6</sub>				378.0	757.2	1136.3
C <sub>n-7</sub>						933.2
C <sub>n-8</sub> /Z <sub>n-8</sub>						757.2
Y <sub>n-1</sub>		599.1	978.2			
Y <sub>n-2</sub>			775.1	1154.3		
Y <sub>n-3</sub>			599.1	978.3	1375.4	
Y <sub>n-4</sub>			396.0	775.1	1154.3	
Y <sub>n-5</sub>				599.1	978.3	
Y <sub>n-6</sub>				396.0	775.2	
Y <sub>n-7</sub>						1375.4
Y <sub>n-8</sub>						1154.3
						978.3
						775.2

\* dp = 2n