

## CO<sub>2</sub> permeability and bicarbonate transport in microperfused interlobular ducts isolated from guinea-pig pancreas

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1. Permeabilities of the luminal and basolateral membranes of pancreatic duct cells to CO<sub>2</sub> and HCO<sub>3</sub><sup>-</sup> were examined in interlobular duct segments isolated from guinea-pig pancreas. Intracellular pH (pH<sub>i</sub>) was measured by microfluorometry in unstimulated, microperfused ducts loaded with the pH-sensitive fluoroprobe 2'7'-bis(2-carboxyethyl)-5(6)-carboxy-fluorescein (BCECF).
2. When HCO<sub>3</sub><sup>-</sup>/CO<sub>2</sub> was admitted to the bath, pH<sub>i</sub> decreased transiently as a result of CO<sub>2</sub> diffusion and then increased to a higher value as a result of HCO<sub>3</sub><sup>-</sup> uptake across the basolateral membrane by Na<sup>+</sup>-HCO<sub>3</sub><sup>-</sup> cotransport.
3. When HCO<sub>3</sub><sup>-</sup>/CO<sub>2</sub> was admitted to the lumen, pH<sub>i</sub> again decreased but no subsequent increase was observed, indicating that the luminal membrane was permeable to CO<sub>2</sub> but did not allow HCO<sub>3</sub><sup>-</sup> entry to the cells from the lumen. Only when the luminal HCO<sub>3</sub><sup>-</sup> concentration was raised above 125 mM was HCO<sub>3</sub><sup>-</sup> entry detected. The same was true of duct cells stimulated with forskolin.
4. Recovery of pH<sub>i</sub> from an acid load, induced by exposure to an NH<sub>4</sub><sup>+</sup> pulse, was dependent on basolateral but not luminal Na<sup>+</sup> and could be blocked by basolateral application of methylisobutylamiloride and H<sub>2</sub>DIDS. This indicates that the Na<sup>+</sup>-H<sup>+</sup> exchangers and Na<sup>+</sup>-HCO<sub>3</sub><sup>-</sup> cotransporters are located exclusively at the basolateral membrane.
5. In the presence of HCO<sub>3</sub><sup>-</sup>/CO<sub>2</sub>, substitution of basolateral Cl<sup>-</sup> with glucuronate caused larger increases in pH<sub>i</sub> than substitution of luminal Cl<sup>-</sup>. This suggests that the anion exchanger activity in the basolateral membrane is greater than that in the luminal membrane.
6. We conclude that the luminal and basolateral membranes are both freely permeable to CO<sub>2</sub>, but while the basolateral membrane has both uptake and efflux pathways for HCO<sub>3</sub><sup>-</sup>, the luminal membrane presents a significant barrier to the re-entry of secreted HCO<sub>3</sub><sup>-</sup>, largely through the inhibition of the luminal anion exchanger by high luminal HCO<sub>3</sub><sup>-</sup> concentrations.

The ductal system of the exocrine pancreas produces a bicarbonate-rich fluid secretion in response to secretin and some other stimuli. However, there are significant species-dependent variations in the pattern of pancreatic HCO<sub>3</sub><sup>-</sup> secretion *in vivo*. In the guinea-pig (Padfield *et al.* 1989) and in several other species, including dog, cat and human (Case & Argent, 1993), the HCO<sub>3</sub><sup>-</sup> concentration of the juice may reach 140–150 mM during maximal stimulation with secretin. In the rat, however, the highest HCO<sub>3</sub><sup>-</sup> concentration is about 70 mM (Sewell & Young, 1975). We have recently reported that interlobular duct segments isolated from guinea-pig pancreas secrete a HCO<sub>3</sub><sup>-</sup>-rich fluid (> 130 mM) during stimulation with 10 nM secretin (Ishiguro *et al.* 1998), which indicates that the *in vivo*

behaviour of pancreatic ducts is well preserved in this *in vitro* preparation.

In order to maintain such a high concentration of HCO<sub>3</sub><sup>-</sup> in pancreatic juice during sustained secretion, pancreatic duct cells must possess the following characteristics. (1) HCO<sub>3</sub><sup>-</sup> must be actively accumulated in the cell by transport across the basolateral membrane. We have previously shown that Na<sup>+</sup>-HCO<sub>3</sub><sup>-</sup> cotransport plays a dominant role in this process (Ishiguro *et al.* 1996*a*). (2) The luminal membrane must possess a mechanism which specifically transports HCO<sub>3</sub><sup>-</sup> into the lumen. In a previous paper (Ishiguro *et al.* 1996*b*), we demonstrated a secretin-stimulated HCO<sub>3</sub><sup>-</sup> efflux pathway in the luminal membrane that has yet to be identified. (3) The luminal membrane must not contain

transporters that would allow  $\text{HCO}_3^-$  secreted into the lumen to re-enter the cell.

To compare the passive permeability characteristics of the basolateral and luminal membranes of guinea-pig pancreatic duct cells to extracellular  $\text{CO}_2$  and  $\text{HCO}_3^-$ , we have examined the changes in intracellular pH that occur in unstimulated, microperfused, interlobular duct segments when either the basolateral or the luminal membrane is exposed to changes in  $\text{HCO}_3^-$  or  $\text{CO}_2$  concentration.

## METHODS

### Isolation and culture of interlobular ducts

Female Hartley guinea-pigs (350–450 g) were killed by cervical dislocation, in accordance with national guidelines. As described previously (Ishiguro *et al.* 1996a), the pancreas was removed and interlobular ducts (diameter 100–150  $\mu\text{m}$ , length 800–1200  $\mu\text{m}$ ) were isolated and cultured overnight, during which time the ends of the duct segments sealed spontaneously.

### Solutions

The standard Hepes-buffered solution contained (mM): 140 NaCl, 5 KCl, 1  $\text{CaCl}_2$ , 1  $\text{MgCl}_2$ , 10 D-glucose and 10 Hepes; and was equilibrated with 100%  $\text{O}_2$ . The  $\text{Na}^+$ -free Hepes-buffered solution contained *N*-methyl-D-glucamine (NMDG<sup>+</sup>) in place of  $\text{Na}^+$ . The standard  $\text{HCO}_3^-$ -buffered solution contained (mM): 115 NaCl, 5 KCl, 1  $\text{CaCl}_2$ , 1  $\text{MgCl}_2$ , 10 D-glucose and 25  $\text{NaHCO}_3$ ; and was equilibrated with 95%  $\text{O}_2$ –5%  $\text{CO}_2$ . The 25 mM  $\text{HCO}_3^-$ –0%  $\text{CO}_2$  solution was gassed with 100%  $\text{O}_2$ . High- $\text{HCO}_3^-$  solutions contained 125 or 145 mM  $\text{NaHCO}_3$  and the concentration of NaCl was reduced accordingly to maintain osmolarity. The  $\text{Cl}^-$  concentrations of these solutions were 24 and 4 mM, respectively, compared with 124 mM in the standard  $\text{HCO}_3^-$ -buffered solution. The  $\text{Na}^+$ -free  $\text{HCO}_3^-$ -buffered solution contained NMDG-Cl in place of NaCl, choline bicarbonate in place of  $\text{NaHCO}_3$  and 10  $\mu\text{M}$  atropine to avoid any muscarinic effects resulting from the high concentration of choline.  $\text{Cl}^-$ -free solutions were made by replacing  $\text{Cl}^-$  with glucuronate. In solutions containing  $\text{NH}_4^+$ , the concentration of  $\text{Na}^+$  was reduced to maintain osmolarity. All solutions, except for the high- $\text{HCO}_3^-$  solutions and the 25 mM  $\text{HCO}_3^-$ –0%  $\text{CO}_2$  solution, were adjusted to pH 7.4 at 37 °C. The pH values of the high- $\text{HCO}_3^-$  solutions equilibrated with 5%  $\text{CO}_2$  were in the range 8.2–8.3.

### Microperfusion of the isolated ducts

The lumen of the interlobular duct segment was microperfused as described previously (Ishiguro *et al.* 1999). Both ends of the duct were cut open using sharpened needles and one end was cannulated with concentric holding and perfusion pipettes. The duct lumen was perfused at a rate of approximately 10–20  $\mu\text{l min}^{-1}$  while the bath was maintained at 37 °C and continuously perfused at 3 ml  $\text{min}^{-1}$  in the same direction as the flow of luminal perfusate.

### Measurement of intracellular pH (pH<sub>i</sub>)

Intracellular pH in the duct cells was estimated by microfluorometry as described previously (Ishiguro *et al.* 1996a) using the pH-sensitive fluoroprobe BCECF. After cannulating the duct for luminal perfusion, the duct cells were loaded with BCECF for 10 min by adding the acetoxymethyl ester BCECF-AM (2  $\mu\text{M}$ ) to the bathing solution. Perfusion of both bath and lumen was stopped during the 10 min loading period. Experiments were begun after the duct had been perfused with the control solution for 15 min. Small regions of the duct epithelium (10–20 cells) were illuminated

alternately at excitation wavelengths of 430 and 480 nm. Values of pH<sub>i</sub> were calculated from the fluorescence ratio ( $F_{480}/F_{430}$ ) measured at 530 nm. The system was calibrated using the high- $\text{K}^+$ -nigericin technique (Thomas *et al.* 1979).

### Materials

BCECF-AM and dihydro-4,4'-diisothiocyanatostilbene-2,2'-disulphonic acid ( $\text{H}_2\text{DIDS}$ ) were obtained from Molecular Probes; bafilomycin A<sub>1</sub>, forskolin and atropine sulphate from Sigma; and *N*-methyl-*N*-isobutylamiloride (MIA) from Research Biochemicals International.

### Statistics

Data are presented as the means  $\pm$  S.E.M. unless otherwise indicated. Tests for statistically significant differences were made with Student's *t* test for paired or unpaired data.

## RESULTS

### Effects of $\text{CO}_2$ and $\text{HCO}_3^-$ on intracellular pH in unstimulated ducts

To investigate the  $\text{CO}_2$  permeability and  $\text{HCO}_3^-$  transport characteristics of the luminal and basolateral membranes of unstimulated pancreatic duct cells, changes in pH<sub>i</sub> were observed following the addition of  $\text{HCO}_3^-/\text{CO}_2$  first to the luminal perfusate and then to the bath (Fig. 1A). Initially both the duct lumen and the bath were perfused with a  $\text{HCO}_3^-$ -free, Hepes-buffered solution. When the luminal perfusate was switched to a  $\text{HCO}_3^-$ -buffered solution containing 25 mM  $\text{HCO}_3^-$ –5%  $\text{CO}_2$ , pH<sub>i</sub> decreased from  $7.25 \pm 0.06$  ( $n = 5$ ) to  $7.04 \pm 0.08$  over a period of 30 s. The cytosol remained acidified during the following 4 min period of luminal perfusion with the  $\text{HCO}_3^-$ -buffered solution, but when  $\text{HCO}_3^-/\text{CO}_2$  was removed pH<sub>i</sub> quickly returned to its control value.

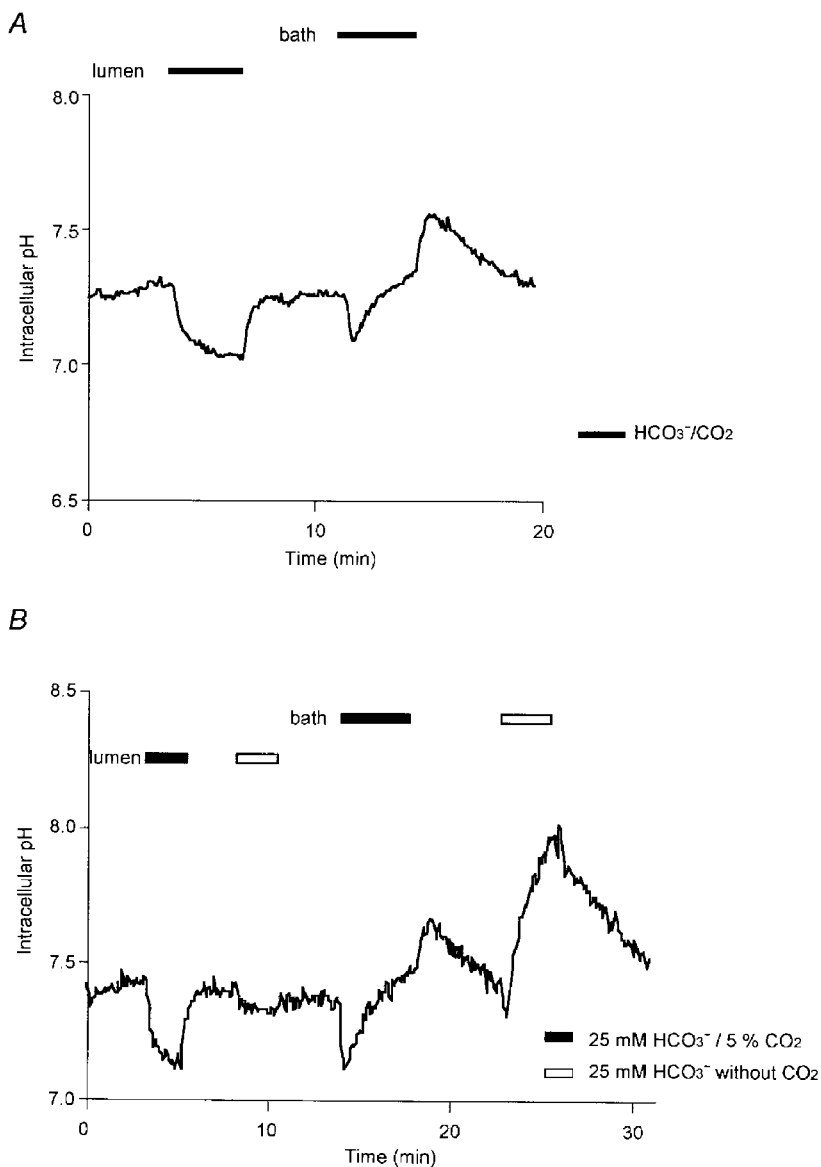
The sustained acidification observed during exposure of the luminal membrane to  $\text{HCO}_3^-/\text{CO}_2$  can best be explained by the continuous diffusion of  $\text{CO}_2$  from the lumen into the cell. This will occur as a result of the steep  $P_{\text{CO}_2}$  gradient that is maintained between the lumen and the bath – the  $P_{\text{CO}_2}$  in the cells attaining a steady-state value somewhere between that of the lumen (equilibrated with 5%  $\text{CO}_2$ ) and the bath (0%  $\text{CO}_2$ ). The diffusion of  $\text{CO}_2$  into the cells across the luminal membrane will continue to acidify the cytosol so long as there is an efflux pathway for  $\text{HCO}_3^-$ , and this most probably occurs via the anion exchangers that are present in the luminal and basolateral membranes (see below). Therefore the steady-state pH that is achieved during this period will reflect a balance between the rate of acidification due to  $\text{CO}_2$  entry and  $\text{HCO}_3^-$  efflux, and the rate of compensatory  $\text{H}^+$  extrusion via the  $\text{Na}^+$ – $\text{H}^+$  exchanger in the basolateral membrane, which is stimulated by the lower pH<sub>i</sub>.

In the second part of the experiment, the basolateral membrane was exposed to  $\text{HCO}_3^-/\text{CO}_2$  by switching the bath perfusate to the  $\text{HCO}_3^-$ -buffered solution. Intracellular pH decreased transiently but then gradually increased to a value which was significantly higher than before  $\text{HCO}_3^-/\text{CO}_2$  exposure ( $7.42 \pm 0.03$  in 3 min,  $P < 0.05$ ). When

$\text{HCO}_3^-/\text{CO}_2$  was removed from the perfusate, there was a further transient increase in  $\text{pH}_i$  followed by a gradual return towards the resting value.

In this case,  $\text{pH}_i$  recovered quickly from the acidifying effect of  $\text{CO}_2$  diffusion into the cells and increased to a higher value. This could be explained either by enhanced  $\text{H}^+$  extrusion across the basolateral membrane, which is unlikely given that it did not occur when  $\text{CO}_2$  entered across the luminal membrane, or by active  $\text{HCO}_3^-$  uptake into the cells, necessarily across the basolateral membrane. This would be consistent with our previous data (Ishiguro *et al.* 1996a, 1998), which established the presence of a  $\text{Na}^+-\text{HCO}_3^-$  cotransporter in the basolateral membrane.

To distinguish between the separate effects of  $\text{CO}_2$  and  $\text{HCO}_3^-$  on  $\text{pH}_i$ , a solution was prepared which contained 25 mM  $\text{HCO}_3^-$  but which was equilibrated with 100%  $\text{O}_2$  ( $\text{pH} \sim 8.2$ ) and thus was nominally free of  $\text{CO}_2$ . When this solution was applied to the lumen (first open bar, Fig. 1B),  $\text{pH}_i$  decreased by only  $0.06 \pm 0.01$  units ( $n = 4$ ), equivalent to only 26% of the acidification caused by the preceding application of both  $\text{HCO}_3^-$  and  $\text{CO}_2$  (first filled bar). This supports our hypothesis that the initial decrease in  $\text{pH}_i$  on exposure to luminal  $\text{HCO}_3^-/\text{CO}_2$  is due to  $\text{CO}_2$  entry by diffusion. The small residual decrease could have been due to the presence of a small amount of  $\text{CO}_2$  derived from the  $\text{HCO}_3^-$  in the solution.



**Figure 1.** Effects of basolateral or luminal  $\text{HCO}_3^-/\text{CO}_2$  on intracellular pH

Initially the bath and lumen of guinea-pig pancreatic duct segments were separately perfused with the standard  $\text{HCO}_3^-/\text{CO}_2$ -free Hepes-buffered solution. Thereafter, in *A* (representative of 5 experiments), first the luminal perfusate and then the bath solution was switched to a  $\text{HCO}_3^-$ -buffered solution containing 25 mM  $\text{HCO}_3^-$ -5%  $\text{CO}_2$  (filled bar). In *B* (representative of 4 experiments), the luminal and bath solutions were switched first to 25 mM  $\text{HCO}_3^-$ -5%  $\text{CO}_2$  (filled bar) and then to 25 mM  $\text{HCO}_3^-$ -0%  $\text{CO}_2$  (open bar).

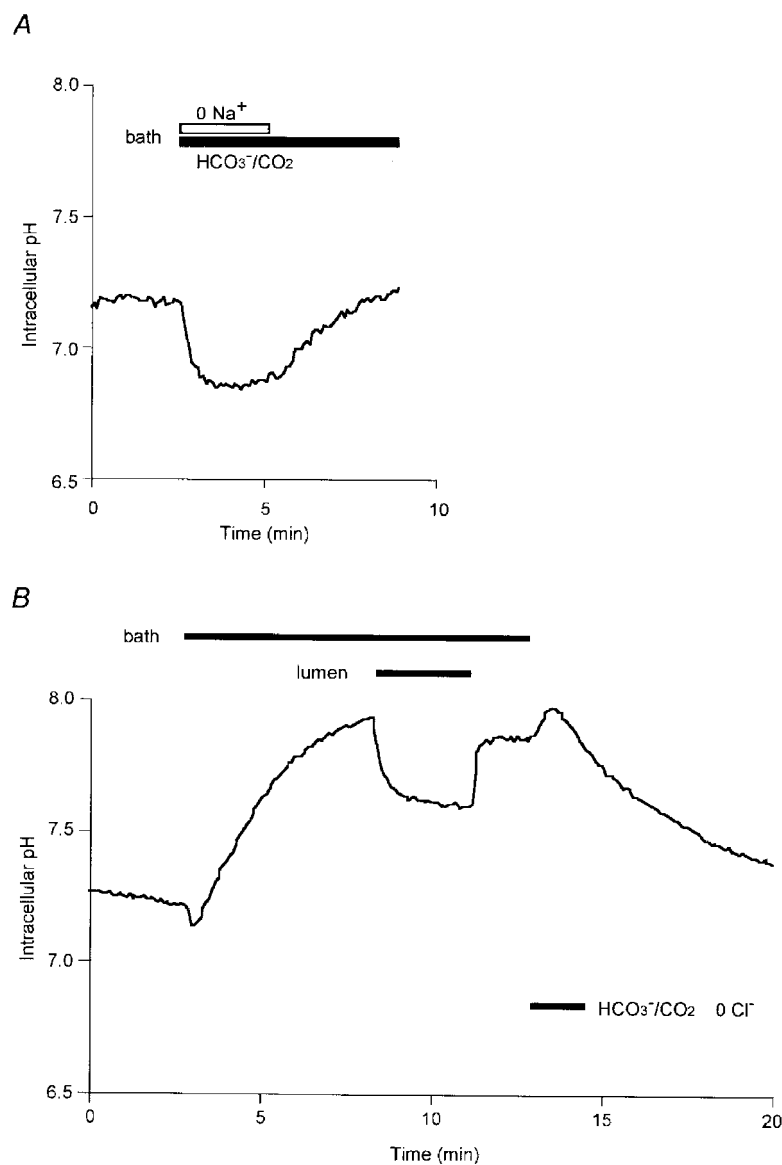
Application of the 25 mM  $\text{HCO}_3^-$ -0%  $\text{CO}_2$  solution to the bath (second open bar) also caused a much reduced transient acidification, but this was followed by a dramatic increase in  $\text{pH}_i$  to  $7.94 \pm 0.05$  in 3 min. This supports our hypothesis that the secondary increase in  $\text{pH}_i$ , on exposure of the basolateral membrane to  $\text{HCO}_3^-/\text{CO}_2$ , is due to  $\text{HCO}_3^-$  entry, exaggerated in these conditions by the absence of the acidifying effect of  $\text{CO}_2$  diffusion into the cell.

#### Dependence of $\text{HCO}_3^-$ influx across the basolateral membrane on $\text{Na}^+$ and $\text{Cl}^-$

The transport mechanisms responsible for the influx of  $\text{HCO}_3^-$  across the basolateral membrane were examined further in ion substitution experiments. When the bath perfusate was switched from the standard HEPES-buffered

solution to a  $\text{Na}^+$ -free  $\text{HCO}_3^-$ -buffered solution (Fig. 2A),  $\text{pH}_i$  decreased rapidly but then remained acidified throughout the 3 min period of  $\text{Na}^+$ -free perfusion. Restoration of basolateral  $\text{Na}^+$  caused  $\text{pH}_i$  to increase, indicating that the rise in  $\text{pH}_i$  previously observed during exposure to basolateral  $\text{HCO}_3^-/\text{CO}_2$  (Fig. 1A) was dependent on basolateral  $\text{Na}^+$ .

When the bath superfusate was switched to a  $\text{Cl}^-$ -free  $\text{HCO}_3^-$ -buffered solution (Fig. 2B),  $\text{pH}_i$  transiently decreased and then increased to  $7.93 \pm 0.05$  in 5 min ( $n = 4$ ). This was a much higher value than that observed in the standard ( $\text{Cl}^-$ -containing)  $\text{HCO}_3^-$ -buffered solution (Fig. 1A) and suggests that  $\text{HCO}_3^-$  influx via the basolateral membrane does not require the presence of basolateral  $\text{Cl}^-$ . The fact



**Figure 2.** Dependence of  $\text{HCO}_3^-$  influx across the basolateral membrane on  $\text{Na}^+$  or  $\text{Cl}^-$

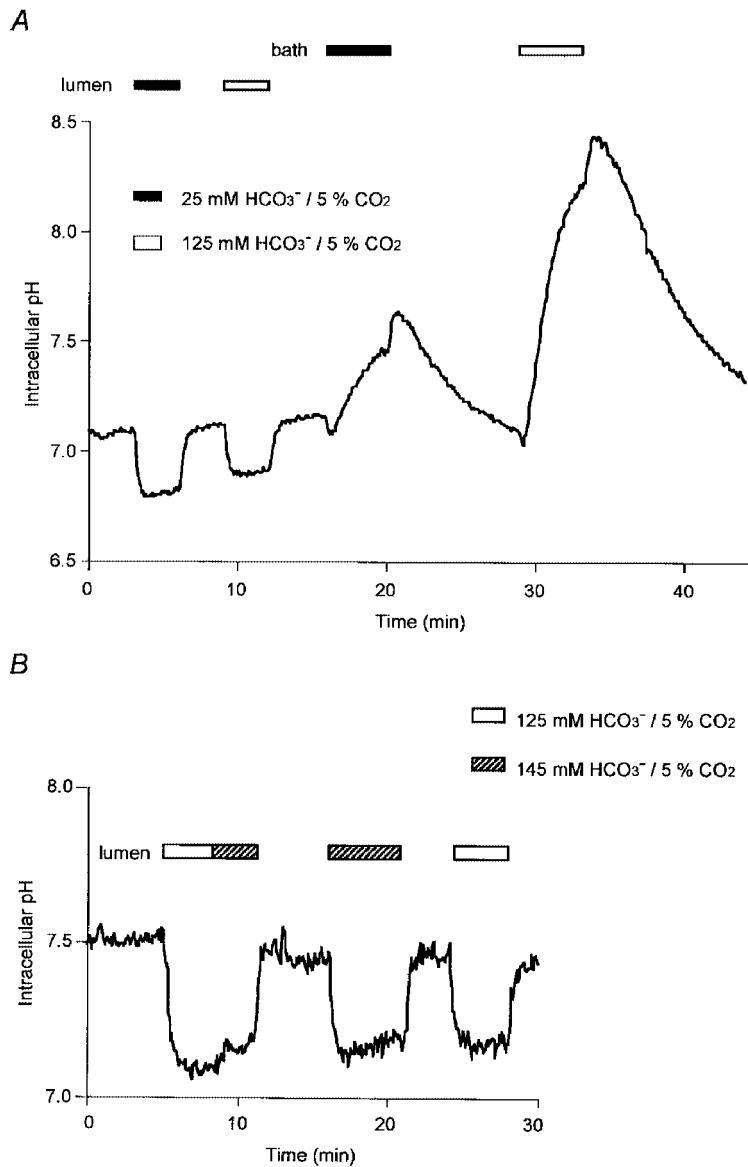
Initially the bath and lumen of the duct segments were separately perfused with the standard HEPES-buffered solution. Thereafter, in *A* (representative of 4 experiments), the bath solution was switched to a  $\text{Na}^+$ -free  $\text{HCO}_3^-$ -buffered solution, while in *B* (also representative of 4 experiments), the bath solution and then the luminal perfusate were switched to a  $\text{Cl}^-$ -free  $\text{HCO}_3^-$ -buffered solution.

that the rise in  $\text{pH}_i$  was enhanced by the withdrawal of basolateral  $\text{Cl}^-$  suggests that a basolateral  $\text{Cl}^-$ - $\text{HCO}_3^-$  exchanger (see below) may have contributed to the uptake of  $\text{HCO}_3^-$ . Subsequent application of the  $\text{Cl}^-$ -free  $\text{HCO}_3^-$ -buffered solution to the lumen (Fig. 2*B*) caused a decrease in  $\text{pH}_i$  to a new steady-state value. This would be expected since the rise in cytosolic  $P_{\text{CO}_2}$  resulting from bilateral exposure to  $\text{HCO}_3^-/\text{CO}_2$  would inevitably shift  $\text{pH}_i$  to a lower value.

**Effects of higher concentrations of  $\text{HCO}_3^-$  on intracellular pH**

To assess further the ability of the basolateral and luminal membranes to permit the entry of extracellular  $\text{HCO}_3^-$ , a

larger transmembrane gradient of  $\text{HCO}_3^-$  was applied using a solution containing 125 mM  $\text{HCO}_3^-$ , 24 mM  $\text{Cl}^-$ , 5%  $\text{CO}_2$ . In the experiment shown in Fig. 3*A*, the standard 25 mM  $\text{HCO}_3^-$  solution and the 125 mM  $\text{HCO}_3^-$  solution were each applied in turn, first to the luminal membrane and then to the basolateral membrane. Luminal application of the 125 mM  $\text{HCO}_3^-$  solution (first open bar) caused a rapid decrease in  $\text{pH}_i$  of  $0.20 \pm 0.04$  ( $n = 5$ ) units, which was not significantly different from the decrease in  $\text{pH}_i$  observed with the 25 mM  $\text{HCO}_3^-$  solution (first filled bar;  $0.21 \pm 0.03$ ). No significant recovery of  $\text{pH}_i$  was observed during luminal perfusion with the 125 mM  $\text{HCO}_3^-$  solution, indicating that little if any  $\text{HCO}_3^-$  entered the cell across the luminal



**Figure 3. Effects of high extracellular  $\text{HCO}_3^-$  concentrations on intracellular pH**

Initially the bath and lumen of the duct segments were separately perfused with the standard HEPES-buffered solution. Thereafter, in *A* (representative of 5 experiments), the luminal perfusate and then the bath solution was switched first to 25 mM  $\text{HCO}_3^-$ -5%  $\text{CO}_2$  (filled bar) and then to 125 mM  $\text{HCO}_3^-$ -5%  $\text{CO}_2$  (open bar). In *B* (representative of 5 experiments), the lumen was exposed either to 125 mM  $\text{HCO}_3^-$ -5%  $\text{CO}_2$  (open bar) or to 145 mM  $\text{HCO}_3^-$ -5%  $\text{CO}_2$  (hatched bar).

membrane even with a large inward concentration gradient for  $\text{HCO}_3^-$ . In contrast, when the 125 mM  $\text{HCO}_3^-$  solution was applied to the basolateral membrane (Fig. 3A, second open bar),  $\text{pH}_i$  increased rapidly to  $8.09 \pm 0.11$ . Although this change would have been partly due to  $\text{HCO}_3^-$  influx via the basolateral  $\text{Cl}^-$ - $\text{HCO}_3^-$  exchanger (driven by the reduction in basolateral  $\text{Cl}^-$  concentration as in Fig. 2B), it indicates a marked difference in the capacity of the luminal and basolateral membranes to transport  $\text{HCO}_3^-$  into the cells. In particular, it suggests that the activity of the luminal anion exchanger may actually be suppressed under these conditions.

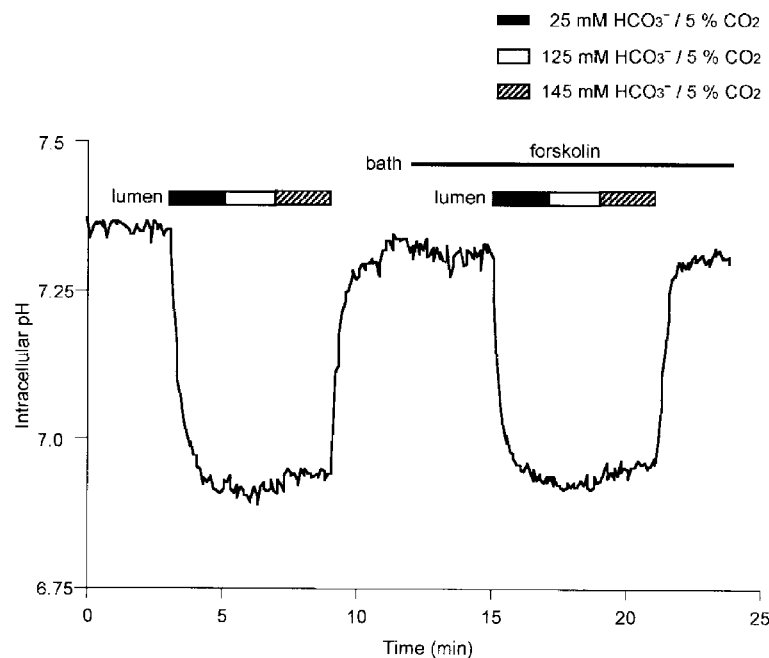
At an even higher luminal  $\text{HCO}_3^-$  concentration (145 mM) there began to be some evidence of  $\text{HCO}_3^-$  entry across the luminal membrane (Fig. 3B). Initially the 125 mM  $\text{HCO}_3^-$  solution was applied to the lumen for 3 min (first open bar), during which period no obvious  $\text{pH}_i$  increase was observed (as previously in Fig. 3A). The luminal concentration of  $\text{HCO}_3^-$  was then raised to 145 mM (first hatched bar) and this caused a slight rise in  $\text{pH}_i$  over a 3 min period. Direct application of the 145 mM  $\text{HCO}_3^-$  solution to the lumen (second hatched bar) had a similar effect:  $\text{pH}_i$  decreased quickly and then increased slowly by  $0.10 \pm 0.01$  units over 5 min ( $n = 5$ ). Direct application of the 125 mM  $\text{HCO}_3^-$  solution, on the other hand, caused a sustained reduction in  $\text{pH}_i$  (second open bar). Thus  $\text{HCO}_3^-$  entry across the luminal membrane appears only to occur at very high luminal  $\text{HCO}_3^-$  concentrations.

Since the activities of the luminal membrane transporters inevitably change during stimulated secretion, a similar

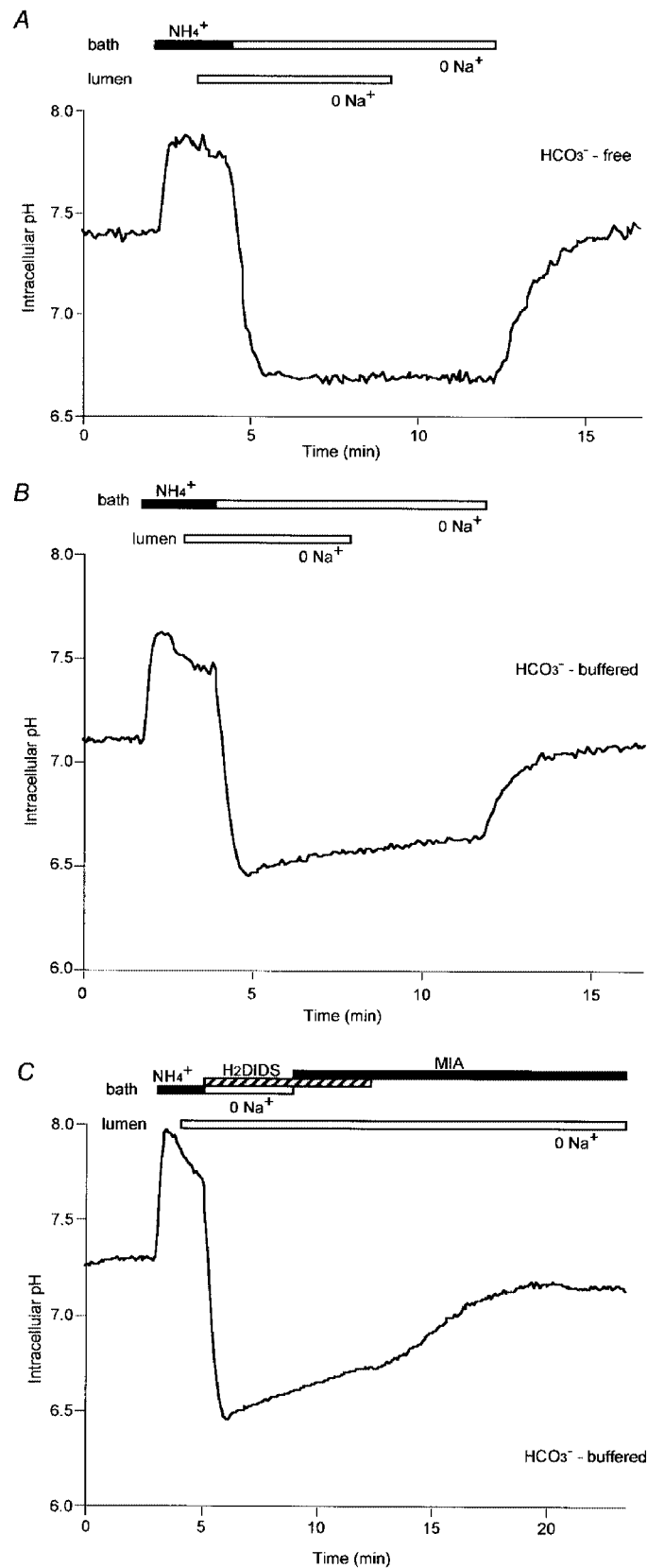
experiment was performed before and during application of 1  $\mu\text{M}$  forskolin, which is known to mimic the effect of secretin in this preparation (Ishiguro *et al.* 1996b) by elevating intracellular cyclic AMP (Fig. 4). As before, the lumen was exposed to progressively higher  $\text{HCO}_3^-$  concentrations at constant  $P_{\text{CO}_2}$  while the bath remained  $\text{HCO}_3^-$  free. Application of forskolin itself had little effect on  $\text{pH}_i$  as reported previously (Ishiguro *et al.* 1996b). The pattern of changes in  $\text{pH}_i$  when the lumen was exposed to the different  $\text{HCO}_3^-$  concentrations in the presence of forskolin was essentially the same as in the unstimulated state. Thus, whatever changes in transport occur in the luminal membrane during stimulation, there is still little evidence of passive  $\text{HCO}_3^-$  re-entry into the cells despite the steep concentration gradient.

#### Recovery of intracellular pH following an acid load

In order to localise the pH regulatory mechanisms responsible for  $\text{HCO}_3^-$  uptake and  $\text{H}^+$  extrusion in unstimulated ducts, we examined whether the recovery of  $\text{pH}_i$  from an acid load was dependent on basolateral or luminal  $\text{Na}^+$ . Acid loading was achieved by adding 20 mM  $\text{NH}_4^+$  to the bath for 2 min, followed immediately by  $\text{Na}^+$  withdrawal from both the bath and lumen. We first examined the recovery of  $\text{pH}_i$  in ducts in the absence of  $\text{HCO}_3^-/\text{CO}_2$  (Fig. 5A). While the bath and lumen were perfused with the  $\text{Na}^+$ -free HEPES-buffered solution, there was no recovery of  $\text{pH}_i$ . Restoration of luminal  $\text{Na}^+$  caused no change in  $\text{pH}_i$ . However, restoration of basolateral  $\text{Na}^+$  caused  $\text{pH}_i$  to return rapidly to the resting value. This indicates that a  $\text{Na}^+$ -dependent,  $\text{HCO}_3^-$ -independent acid



**Figure 4.** Effect of forskolin stimulation on  $\text{pH}_i$  changes with high luminal  $\text{HCO}_3^-$  concentrations. Initially the bath and lumen were separately perfused with the standard HEPES-buffered solution. Thereafter, the lumen was exposed to increasing concentrations of  $\text{HCO}_3^-$  equilibrated with 5%  $\text{CO}_2$ . This was then repeated during stimulation with 1  $\mu\text{M}$  forskolin. Data are representative of 4 experiments.



**Figure 5. Recovery of intracellular pH from an acid load**

Duct cells were acid-loaded by adding 20 mM  $\text{NH}_4^+$  to the bath for 2 min followed by  $\text{Na}^+$  removal (from the bath and lumen) in the absence (A) and presence (B and C) of  $\text{HCO}_3^-/\text{CO}_2$ . In C,  $\text{H}_2\text{DIDS}$  (0.5 mM) and MIA (10  $\mu\text{M}$ ) were added to the bath as indicated to block basolateral  $\text{Na}^+-\text{HCO}_3^-$  cotransport and  $\text{Na}^+-\text{H}^+$  exchange, respectively. Data are representative of at least 4 experiments.

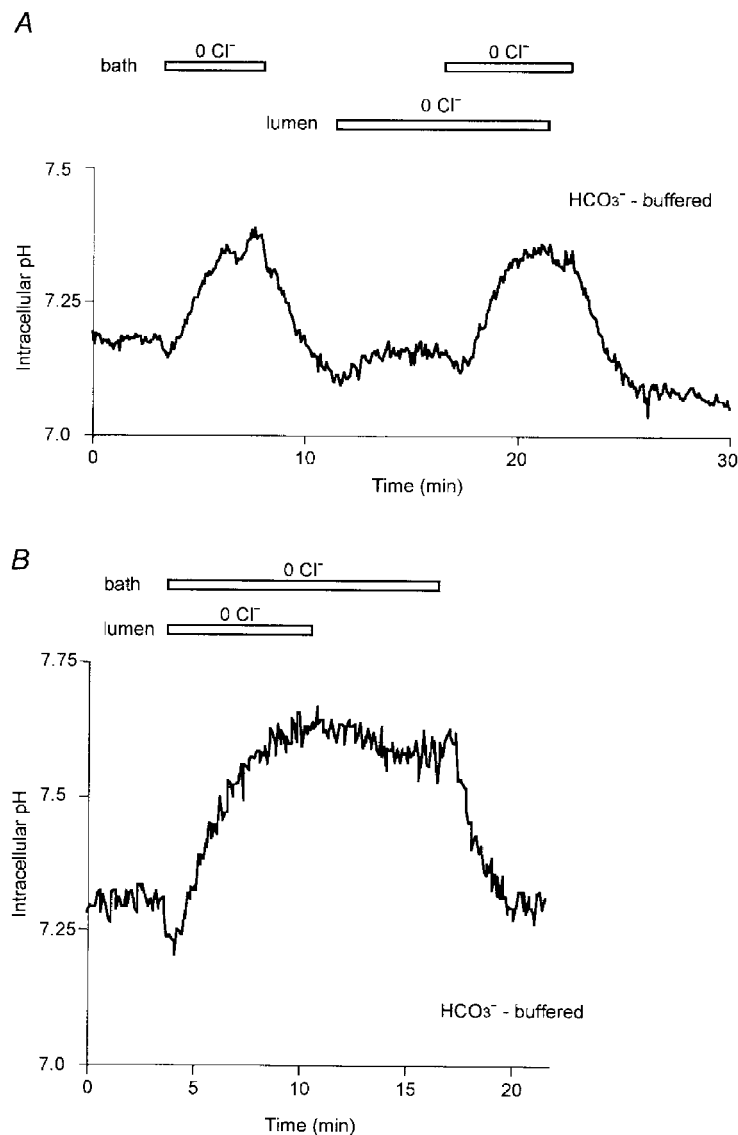
extruder, most probably an  $\text{Na}^+-\text{H}^+$  exchanger, is located exclusively in the basolateral membrane.

The same experiment was repeated in the bilateral presence of  $\text{HCO}_3^-/\text{CO}_2$  and rather similar observations were made (Fig. 5B). However, in these conditions, a slow but significant recovery of  $\text{pH}_i$  was observed in the absence of  $\text{Na}^+$  from both the lumen and the bath ( $0.13 \pm 0.02$  units in 4 min,  $n = 5$ ). The mechanism responsible for this phenomenon is unclear since it was not inhibited by bafilomycin  $\text{A}_1$  (data not shown) and therefore cannot be attributed to the vesicular-type  $\text{H}^+$ -ATPase previously reported in guinea-pig duct cells (De Ondarza & Hootman, 1997).

While restoration of  $\text{Na}^+$  to the lumen did not increase the rate of recovery above this slow basal level, restoration of

basolateral  $\text{Na}^+$  caused  $\text{pH}_i$  to return rapidly to its resting value. This suggests that not only the  $\text{Na}^+-\text{H}^+$  exchanger but also the  $\text{Na}^+-\text{HCO}_3^-$  cotransporter, which we have described previously (Ishiguro *et al.* 1996a), are located solely in the basolateral membrane.

To test this hypothesis, the effects of methylisobutylamiloride (MIA), an inhibitor of  $\text{Na}^+-\text{H}^+$  exchange, and the disulphonic stilbene  $\text{H}_2\text{DIDS}$ , an inhibitor of anion transport, were examined using a similar protocol (Fig. 5C). In this experiment, the increase in the rate of recovery of  $\text{pH}_i$  when basolateral  $\text{Na}^+$  was restored after acidification (Fig. 5B) was blocked by the presence of  $10 \mu\text{M}$  MIA and  $0.5 \text{ mM}$   $\text{H}_2\text{DIDS}$  in the bath. When  $\text{H}_2\text{DIDS}$ , which acts as a reversible inhibitor of the basolateral  $\text{Na}^+-\text{HCO}_3^-$  cotransporter, was withdrawn, there was a gradual recovery



**Figure 6.** Effects of  $\text{Cl}^-$  removal on intracellular pH in the presence of  $\text{HCO}_3^-$

Initially the bath and lumen of the duct segments were separately perfused with the standard  $\text{HCO}_3^-$ -buffered solution. Thereafter, the luminal perfusate and/or the bath solution was switched to a  $\text{Cl}^-$ -free,  $\text{HCO}_3^-$ -buffered solution ( $\text{Cl}^-$  replaced with glucuronate) (open bar). Experiments in A and B are each representative of 5 experiments.



of  $\text{pH}_i$  in the continued presence of MIA. This is consistent with there being a significant role for  $\text{HCO}_3^-$  uptake via the cotransporter in the regulation of  $\text{pH}_i$ .

### Effects of $\text{Cl}^-$ removal on intracellular pH in the presence of $\text{HCO}_3^-$

Studies on rat pancreatic ducts suggest that  $\text{HCO}_3^-$  secretion across the luminal membrane may involve the parallel operation of a  $\text{Cl}^-$ - $\text{HCO}_3^-$  exchanger and a cAMP-activated  $\text{Cl}^-$  channel, the cystic fibrosis transmembrane conductance regulator (CFTR) (Gray *et al.* 1988, 1993; Novak & Greger, 1988). In our work, using guinea-pig pancreatic ducts, we have demonstrated that  $\text{HCO}_3^-$  transport across the luminal membrane is mediated by  $\text{Cl}^-$ - $\text{HCO}_3^-$  exchange under resting conditions but seems to involve an alternative mechanism during stimulation with secretin (Ishiguro *et al.* 1996b, 1998). To determine the localisation of  $\text{Cl}^-$ - $\text{HCO}_3^-$  exchangers in the guinea-pig ducts, we have investigated the effects on  $\text{pH}_i$  of removing  $\text{Cl}^-$  from the bath or lumen in the bilateral presence of  $\text{HCO}_3^-/\text{CO}_2$ .

Removal of  $\text{Cl}^-$  from the bath solution by replacement with glucuronate (Fig. 6A) caused a rapid increase in  $\text{pH}_i$  of  $0.24 \pm 0.03$  units ( $n = 5$ ) over a 4 min period. Removal of  $\text{Cl}^-$  from the luminal solution caused a slower and much smaller increase in  $\text{pH}_i$  ( $0.07 \pm 0.01$  units). In Fig. 6B, the bath and lumen were initially both switched to the  $\text{Cl}^-$ -free solution and then  $\text{Cl}^-$  was restored first to the luminal solution and then to the bath. Restoring  $\text{Cl}^-$  to the lumen caused  $\text{pH}_i$  to decrease very slowly (by  $0.06 \pm 0.01$  in 6 min,  $n = 4$ ) whereas the subsequent restoration of  $\text{Cl}^-$  to the bath caused  $\text{pH}_i$  to decrease much more rapidly (by  $0.28 \pm 0.04$  in 5 min) to the basal value. Taken together the data in Fig. 6 appear to suggest that the activity of the  $\text{Cl}^-$ - $\text{HCO}_3^-$  exchangers in the basolateral membrane of unstimulated pancreatic duct cells is considerably greater than that in the luminal membrane.

## DISCUSSION

### $\text{HCO}_3^-/\text{CO}_2$ permeability of the basolateral and luminal membranes

Transport of  $\text{HCO}_3^-$  and  $\text{Cl}^-$  across pancreatic duct epithelium has previously been examined by perfusing the main pancreatic duct of the cat *in vivo* (Case *et al.* 1969; Case & Scratcherd, 1970; Reber *et al.* 1986). These studies revealed that the main duct allowed very little exchange of  $\text{Cl}^-$  for  $\text{HCO}_3^-$ , even at the slowest perfusion rates, and therefore must be relatively impermeable to  $\text{HCO}_3^-$ . Our present paper is the first to study the permeability of small, intra- and interlobular ducts where there is active  $\text{HCO}_3^-$  secretion (Case & Argent, 1993). By measuring  $\text{pH}_i$  in microperfused, interlobular duct segments isolated from guinea-pig pancreas, we have demonstrated that (1) both basolateral and luminal membranes are permeable to extracellular  $\text{CO}_2$ , (2) basolaterally applied  $\text{HCO}_3^-$  easily gains access to the cell, whereas (3) intraluminally applied  $\text{HCO}_3^-$  does not. Regardless of the mechanisms involved,

this polarity is a basic property of the pancreatic duct epithelium which is necessary to achieve vectorial transport of  $\text{HCO}_3^-$  from blood to duct lumen.

Only a few previous reports have compared the permeability of the basolateral and luminal membranes of epithelial tissues to  $\text{HCO}_3^-/\text{CO}_2$ . In perfused gastric glands (Waisbren *et al.* 1994), the luminal membranes of both parietal and chief cells were impermeable to  $\text{HCO}_3^-$  and to  $\text{CO}_2$  while the basolateral membranes were permeable to both. Not surprisingly the permeability characteristics of pancreatic duct cells are somewhat different.

### Membrane localization of $\text{H}^+$ - $\text{HCO}_3^-$ transporters

Our present data demonstrate the presence of an  $\text{Na}^+$ - $\text{H}^+$  exchanger, an  $\text{Na}^+$ - $\text{HCO}_3^-$  cotransporter and an unidentified  $\text{Na}^+$ -independent acid-extrusion mechanism in the basolateral membrane. Anion exchangers are found in both basolateral and luminal membranes. These results are largely similar to those of a previous study of the rat pancreatic duct (Zhao *et al.* 1994). However, the observation that the  $\text{Cl}^-$ - $\text{HCO}_3^-$  exchanger activity in the basolateral membrane of unstimulated guinea-pig duct cells was greater than that in the luminal membrane was unexpected. We have previously demonstrated that the luminal exchanger plays a role in the spontaneous secretion of  $\text{HCO}_3^-$  and fluid in guinea-pig ducts (Ishiguro *et al.* 1998). This is achieved by the passive exchange of intracellular  $\text{HCO}_3^-$  with luminal  $\text{Cl}^-$  via the anion exchanger in the luminal membrane, while  $\text{Cl}^-$  ions recycle via a luminal  $\text{Cl}^-$  conductance. A  $\text{Cl}^-$ - $\text{HCO}_3^-$  exchanger in the basolateral membrane would provide an alternative efflux pathway for intracellular  $\text{HCO}_3^-$  which would tend to dissipate the accumulation of  $\text{HCO}_3^-$  across the basolateral membrane whilst favouring the accumulation of  $\text{Cl}^-$  instead. We would therefore predict that the fluid secreted by unstimulated ducts might be richer in  $\text{Cl}^-$  than  $\text{HCO}_3^-$ .

### $\text{HCO}_3^-$ transport across the basolateral membrane

Two mechanisms have been postulated to account for  $\text{HCO}_3^-$  accumulation across the basolateral membrane: forward (blood to cell) transport of  $\text{HCO}_3^-$ , or backward (cell to blood) transport of  $\text{H}^+$  derived from carbonic acid (Case & Argent, 1993). The latter concept has been supported by several studies in perfused pancreas preparations, which have shown that a variety of buffer anions can substitute for  $\text{HCO}_3^-$ .

However, in this study, when solutions containing 25 or 125 mM  $\text{HCO}_3^-$  were applied to the basolateral membrane, extracellular  $\text{HCO}_3^-$  entered the cells rapidly. Since this was equally true in the nominal absence of  $\text{CO}_2$ , these results support the hypothesis that  $\text{HCO}_3^-$  can be transported actively into the cell. The results are also consistent with a previous study in the isolated perfused cat pancreas in which Ammar *et al.* (1987) altered the pH of the arterial perfusate either at a constant  $P_{\text{CO}_2}$  or at a constant  $\text{HCO}_3^-$  concentration. They found that  $\text{HCO}_3^-$  secretion was independent of arterial pH but critically dependent upon

the arterial  $\text{HCO}_3^-$  concentration. In our study, the influx of  $\text{HCO}_3^-$  across the basolateral membrane was largely dependent on basolateral  $\text{Na}^+$  and was inhibited by  $\text{H}_2\text{DIDS}$ , and thus is most probably mediated by  $\text{Na}^+-\text{HCO}_3^-$  cotransport as we have suggested previously (Ishiguro *et al.* 1996a, 1998).

### $\text{HCO}_3^-$ transport across the luminal membrane

In the present study we have demonstrated that luminally applied  $\text{HCO}_3^-$  does not appear to enter the ductal cells until the luminal concentration of  $\text{HCO}_3^-$  is elevated above 125 mM. Given the presence of an anion exchanger in the luminal membrane this is surprising. When the lumen is perfused with solutions containing 125 mM  $\text{HCO}_3^-$  and 24 mM  $\text{Cl}^-$ , the concentration gradients for  $\text{HCO}_3^-$  and  $\text{Cl}^-$  across the luminal membrane should strongly favour the exchange of intracellular  $\text{Cl}^-$  with luminal  $\text{HCO}_3^-$ , and a consequent rise in  $\text{pH}_i$  as  $\text{HCO}_3^-$  enters the cells. The fact that this does not happen suggests that the luminal membrane  $\text{Cl}^-$ - $\text{HCO}_3^-$  exchanger may be inhibited under these conditions, perhaps as a result of the high luminal pH or the high concentration of luminal  $\text{HCO}_3^-$ . Interestingly, the need for this type of feedback inhibition of the exchanger has already been indicated by computer simulation studies of the rat pancreatic duct (Sohma *et al.* 1996, 1997). Most importantly, this finding again casts doubt on the ability of the exchanger to contribute to net  $\text{HCO}_3^-$  secretion during secretin stimulation where, in the guinea-pig, the luminal  $\text{HCO}_3^-$  concentration is already high. Also, as we have previously demonstrated (Ishiguro *et al.* 1998), secretin-stimulated  $\text{HCO}_3^-$  secretion in the guinea-pig appears to be mediated mainly by a transport mechanism which has, at most, a minimal requirement for  $\text{Cl}^-$ .

An anion channel with a significant  $\text{HCO}_3^-$  conductance could account for  $\text{Cl}^-$ -independent  $\text{HCO}_3^-$  secretion across the luminal membrane. The present finding, that  $\text{HCO}_3^-$  begins to enter the cells at luminal concentrations above 125 mM, suggests that there may be an anion conductance in the luminal membrane that is permeable to  $\text{HCO}_3^-$ . Although electrophysiological data are not yet available for the guinea-pig pancreatic duct, a luminal membrane potential of approximately  $-60$  mV would be close to the reversal potential for  $\text{HCO}_3^-$  at luminal concentrations of this magnitude (assuming that the intracellular  $\text{HCO}_3^-$  concentration was approximately 10 mM since  $\text{pH}_i$  was approximately 7.0). Thus, an increase in luminal  $\text{HCO}_3^-$  concentration from 125 to 145 mM might begin to allow  $\text{HCO}_3^-$  ions to enter the cells from the lumen and hence cause the small alkalinization observed in Fig. 3B. Provided that the membrane potential does not depolarise significantly during stimulation, there could therefore be a small luminally directed electrochemical gradient for  $\text{HCO}_3^-$  that would drive  $\text{HCO}_3^-$  secretion via an anion channel.

In summary, we have demonstrated that, while the basolateral membrane of interlobular duct cells from guinea-pig pancreas has powerful mechanisms for  $\text{HCO}_3^-$  uptake, the

luminal membrane presents a significant barrier to the re-entry of secreted  $\text{HCO}_3^-$  ions from the lumen. Given the likely  $\text{Cl}^-$  and  $\text{HCO}_3^-$  gradients across the luminal membrane during stimulated secretion, this can only be explained by a marked inhibition of the luminal anion exchanger. If this is the case, what is the pathway for  $\text{HCO}_3^-$  secretion across the luminal membrane? There is a possibility that  $\text{HCO}_3^-$  is secreted via a luminal anion channel. If the membrane potential is large enough, there could be a small electrochemical gradient favouring  $\text{HCO}_3^-$  efflux to the lumen even when the luminal  $\text{HCO}_3^-$  concentration exceeds 125 mM. However, if future electrophysiological measurements in the guinea-pig ducts show evidence of significant depolarisation during stimulation, as occurs in the rat (Novak & Pahl, 1993), alternative mechanisms will have to be considered.

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