Stabilization of a β -hairpin in monomeric Alzheimer's amyloid- β peptide inhibits amyloid formation

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Edited by Adriaan Bax, National Institutes of Health, Bethesda, MD, and approved February 1, 2008 (received for review December 13, 2007)

According to the amyloid hypothesis, the pathogenesis of Alzheimer's disease is triggered by the oligomerization and aggregation of the amyloid- β (A β) peptide into protein plagues. Formation of the potentially toxic oligometric and fibrillar A β assemblies is accompanied by a conformational change toward a high content of β -structure. Here, we report the solution structure of A β (1–40) in complex with the phage-display selected affibody protein $Z_{A\beta3}$, a binding protein of nanomolar affinity. Bound A β (1–40) features a β -hairpin comprising residues 17–36, providing the first highresolution structure of A β in β conformation. The positions of the secondary structure elements strongly resemble those observed for fibrillar A β . Z_{A β 3} stabilizes the β -sheet by extending it intermolecularly and by burying both of the mostly nonpolar faces of the A β hairpin within a large hydrophobic tunnel-like cavity. Consequently, $Z_{A\beta3}$ acts as a stoichiometric inhibitor of A β fibrillation. The selected A β conformation allows us to suggest a structural mechanism for amyloid formation based on soluble oligomeric hairpin intermediates.

Aβ-peptide | engineered binding protein | molecular recognition | protein structure | nuclear magnetic resonance

The amyloid- β (A β) peptide is a 39–43 residue cleavage product of the amyloid precursor protein. It is the main component of senile plaques, which are neuropathological hallmarks of Alzheimer's disease (AD), and its critical role for AD etiology has been supported by genetic studies (1, 2). Substantial evidence attributes the role of the principal neurotoxic species to soluble oligomeric forms of A β (3–7). The precise cellular and molecular mechanisms underlying their toxicity are currently a matter of intense research (3, 7).

Antiamyloid approaches, which target the production, aggregation, and clearance of $A\beta$, constitute the focus of current efforts to develop AD therapeutics (1, 2, 8). A promising example is anti-amyloid immunotherapy, which has demonstrated effectiveness in animal models and entailed reduced plaque burden in clinical trials (8-10). Three different modes of action of amyloid-dissolving antibodies have been suggested: (i) dissolution as a consequence of direct binding to oligomeric/ fibrillar A β in the brain, (ii) phagocytosis by microglial cells, and (*iii*) increased efflux of $A\beta$ from the brain as a result of binding of monomeric A β in the plasma (peripheral sink mechanism) (8, 11). Whereas antibodies to the A β N terminus, which is located at the fibril surface, are effective in direct binding to aggregated A β (12, 13), the binding and stabilization of monomeric A β might be a specific feature of antibodies to its hydrophobic central part (14). Recognition of the hydrophobic central part is also a strategy pursued in the search for peptide or peptidomimetic inhibitors of A β fibrillation (15–17). Recently, the structures of the antigen binding fragments of two antibodies to the N-terminal $A\beta(1-8)$ peptide were reported (18). However, the structural basis of interactions with the central and C-terminal parts of $A\beta$ remains elusive.

The peripheral sink mechanism implies that not only antibodies, but any peripherally administered $A\beta$ binding molecule of sufficiently high affinity could potentially dissolve plaques by shifting the dynamic equilibrium between central nervous system $A\beta$ and plasma $A\beta$ toward the latter (14, 19). On this account, affibody ligands to monomeric $A\beta(1-40)$ have recently been selected (20). Affibody ligands represent one class of engineered affinity proteins with applications in biotechnology, biochemical assays, disease diagnosis, and therapy (21–23). They are based on the Z domain derived from staphylococcal protein A and selected by phage display from a combinatorial protein library in which 13 of the 58 amino acid residues are randomized. The randomized positions, distributed over helix 1 and 2 of the three-helix bundle scaffold, have been chosen because of their location to the binding interface in the complex of the Z domain with its target, the Fc fragment from IgG (22).

Here, we investigate the interaction of $A\beta(1-40)$ with the affibody protein $Z_{A\beta3}$ and report the solution structure of the complex. $Z_{A\beta3}$ binds to the central/C-terminal part of $A\beta(1-40)$, which adopts a β -hairpin conformation reminiscent of the $A\beta$ fibril structure.

Results and Discussion

Disulfide-Linked $Z_{A\beta3}$ Dimer Binds $A\beta(1-40)$ with Nanomolar Affinity. The 16 Affibody variants, for which binding to both $A\beta(1-40)$ and $A\beta(1-42)$ was tested and confirmed (20) (Fig. 1A), all have a cysteine residue at position 28, suggesting that disulfide-linked dimers were selected. Dimeric $Z_{A\beta3}$ binds monomeric $A\beta(1-40)$ with 1:1 stoichiometry and an affinity of $K_d = 17$ nM as determined by isothermal titration calorimetry (ITC) (Fig. 24). In contrast, a monomeric mutant, obtained by replacing Cys-28 with serine (Z_{AB3}C28S), binds A β (1-40) with much lower affinity and in agreement with a cooperative association of two $Z_{A\beta}C28S$ molecules with one $A\beta(1-40)$ monomer (Fig. 2B). Cooperative binding of two affibody units to distinct sites on $A\beta(1-40)$ is supported by ¹⁵N heteronuclear single quantum correlation (HSQC) NMR spectroscopy, as shown for the glycine region: The two glycines Gly-13 an Gly-14 give rise to one peak each for free $Z_{A\beta3}C28S$ (Fig. 2C). Upon titration with unlabeled A β (1-40), four new peaks appear with identical intensities, corresponding to Gly-13 and Gly-14 of ZAB3C28S bound to two distinct A β (1–40) sites (Fig. 2D). The chemical shifts of these peaks are practically identical to those observed for saturated $Z_{A\beta3}$ (Fig. 2*E*), suggesting a common binding mode for the monomeric and dimeric constructs. Binding is coupled to folding

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Author contributions: W.H. and T.H. designed research; W.H., C.G., A.J., S.S., and T.H. performed research; C.G., A.J., and S.S. contributed new reagents/analytic tools; W.H. and T.H. analyzed data; and W.H. and T.H. wrote the paper.

Conflict of interest statement: A.J. is an employee of Affibody AB.

This article is a PNAS Direct Submission.

Data deposition: Atomic coordinates and experimental constraints have been deposited in the Protein Data Bank, www.pdb.org [accession no. 20TK ($Z_{A\beta3}$:A β (1–40) complex)].

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This article contains supporting information online at www.pnas.org/cgi/content/full/ 0711731105/DCSupplemental.

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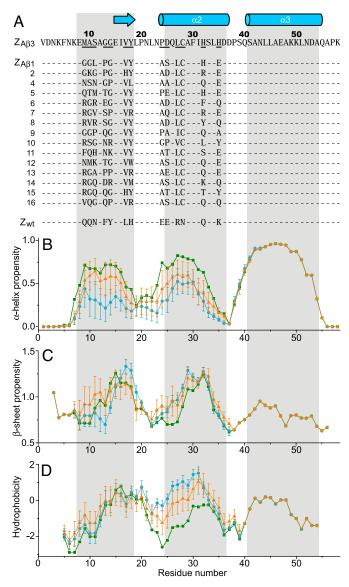


Fig. 1. Sequences, secondary structure propensities, and hydrophobicity of A β -binding affibody proteins. (A) Amino acid sequences of A β -binding affibody proteins aligned to Z_{A $\beta\beta$} (20). Helical or β -sheet secondary structure in the Z_{A $\beta\beta$}:A β (1–40) complex is indicated by cylinders or arrows, respectively. Areas highlighted in gray correspond to helical structure in the Z scaffold (bottommost sequence). Randomized amino acids are underlined for Z_{A $\beta\beta$}. (*B*–*D*) Averaged α -helix propensity (*B*), β -sheet propensity (*C*), and hydrophobicity (*D*) of the 16 Z_{A $\beta\beta} affibody protein sequences (blue circles), and 30 previously published affibody protein sequences selected to bind 10 different targets (orange triangles) (see SI Text). The corresponding values of the Z domain are given as green squares. Error bars indicate estimated standard deviations.</sub>$

of both $A\beta(1-40)$ and the affibody ligand, as indicated by the greatly improved resonance dispersion in HSQC spectra upon complex formation (Fig. 2 *F* and *G*). Particularly, several amide proton resonances are shifted downfield to values typical for β -sheet conformation upon binding [supporting information (SI) Fig. S1]. Concomitantly, the thermostability of $Z_{A\beta3}$ increases from a melting temperature of 47°C for free $Z_{A\beta3}$ to 64°C for the complex (Fig. 2*H*).

 $Z_{A\beta3}$ Inhibits $A\beta(1-40)$ Fibrillation at Stoichiometric Concentrations. Thioflavin T fluorescence was used to monitor $A\beta(1-40)$ fibrillation in the absence and presence of $Z_{A\beta3}$ (Fig. 21). $Z_{A\beta3}$ acts as a potent fibrillation inhibitor. Stoichiometric concentrations

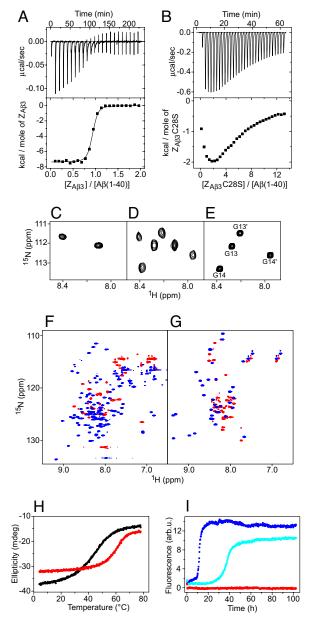


Fig. 2. Biophysical characterization of the $Z_{A\beta3}$:A β (1–40) interaction. (A and *B*) Titration of $Z_{A\beta3}$ dimer (A) or $Z_{A\beta3}$ C28S (B) into A β (1–40) monitored by ITC at 20 °C. For the experiment in *A*, *N* = 0.89 ± 0.01, *K*_d = 17 ± 2 nM, Δ H = -7.4 ± 0.1 kcal mol⁻¹. (*C*-*E*) Glycine region of the ¹H-¹⁵N HSQC spectrum of ¹⁵N-labeled $Z_{A\beta3}$ C28S (*C* and *D*) or $Z_{A\beta3}$ dimer (*E*) in the absence (*C*) or presence (*D* and *E*) of an excess of unlabeled A β (1–40). Peaks in *E* are assigned to the subunits Z and Z' as shown in Fig. 3. (*F* and *G*) Sections of the ¹H-¹⁵N HSQC spectrum of ¹⁵N-labeled $Z_{A\beta3}$ dimer (*F*) or A β (1–40) (*G*) in the absence (red) or presence (blue) of a 15% excess of the respective unlabeled binding partner. (*H*) Thermal melting of free $Z_{A\beta3}$ dimer (black), and an equimolar mixture of the $Z_{A\beta3}$ dimer and A β (1–40) in the absence (blue) and presence of 0.5 (cyan) or 1.1 (red) molar equivalents of $Z_{A\beta3}$ dimer monitored by thioflavin T fluorescence.

of $Z_{A\beta3}$ dimer are required for complete inhibition, revealing that the binding of monomeric $A\beta(1-40)$ is responsible for the inhibitory function.

Structure of the Z_{A\beta3}:Aβ(1-40) Complex. NMR was used to determine the structure of the complex between A β (1-40) and the disulfide-linked dimer of Z_{Aβ3} (Fig. 3). The complex consists of

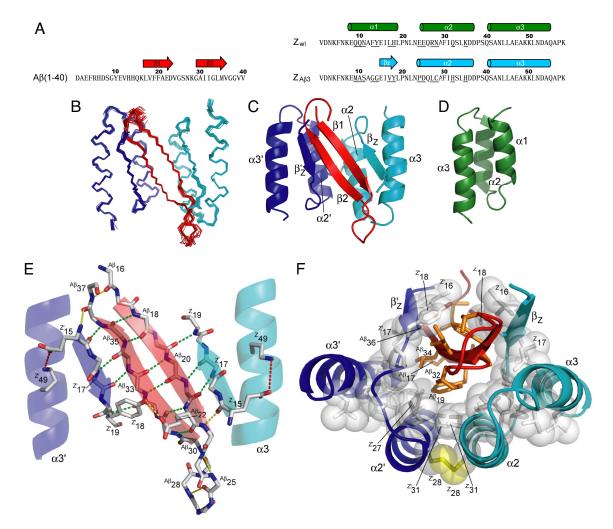


Fig. 3. Structure of the $Z_{A\beta3}$: $A\beta(1-40)$ complex. (*A*) Amino acid sequences of $A\beta(1-40)$, $Z_{A\beta3}$, and the wild-type Z domain (Z_{wt}). Residues randomized in the phagemid library are underlined. Helical or β -sheet secondary structures in the $Z_{A\beta3}$: $A\beta(1-40)$ complex and in Z_{wt} are indicated by cylinders or arrows, respectively. (*B*) Superimposed simulated annealing structures of the $Z_{A\beta3}$: $A\beta(1-40)$ complex [$A\beta(1-40)$ in red, $Z_{A\beta3}$ subunits in blue or cyan]. (*C* and *D*) Ribbon drawings illustrating the topologies of the $Z_{A\beta3}$: $A\beta(1-40)$ complex (*C*) and the original Z domain scaffold (*D*). (*E*) Polar contacts in the $Z_{A\beta3}$: $A\beta(1-40)$ complex. Experimentally validated hydrogen bonds (green), hydrogen bonds observed in >50% of the simulated annealing structures (yellow), and salt bridges (red) are displayed. Residues of the two $Z_{A\beta3}$ subunits are labeled Z or Z', respectively. (*F*) The hydrophobic core of the complex. Nonpolar side chains with water exposure <33% are displayed as orange sticks [for $A\beta(1-40)$] or white sticks and spheres (for $Z_{A\beta3}$). The disulfide bond is shown in yellow.

a four-stranded antiparallel β -sheet and four α -helices. The selected conformation of $A\beta(1-40)$ is a β -hairpin in which the $^{A\beta}17^{-A\beta}23$ and $^{A\beta}30^{-A\beta}36$ fragments make intramolecular backbone hydrogen bonds to form the two central strands of the β -sheet (Fig. 3*E*). Both faces of the A β hairpin are predominantly nonpolar, and both are buried within a large hydrophobic tunnel-like cavity in the $Z_{A\beta3}$ dimer (Fig. 3F). Hence, 1,400 Å² of surface area, of which 71% is nonpolar, is inaccessible to water in the complex. In the cavity, the A β hairpin is flanked on each side by β -strands formed by residues 15–18 of the two Z_{AB3} subunits, respectively. This fragment is part of helix 1 in the originating Z domain (24) and affibody complexes reported in refs. 25 and 26. However, it is unfolded in $Z_{A\beta3}$ and presumably in all $Z_{A\beta}$ binders, because they contain helix-destabilizing glycine and proline replacements at positions 9-11 and 13-14 (Fig. 1 A and B). Selected residues at positions 17 and 18 in the (former) helix do, however, show β -sheet propensity in agreement with the observed structure (Fig. 1C). A further consequence of helix 1 unfolding is that it opens a large hydrophobic cleft in which the core of the $Z_{A\beta3}$ dimer becomes exposed. The "interior" face of the A β hairpin containing the ^{A β}Leu-17, ^{A β}Phe-19, ^{A β}Ile-32, ^{A β}Leu-34, and ^{A β}Val-36 side chains docks into the cleft to form a large intermolecular hydrophobic core.

The $Z_{A\beta3}$ side of the core is held tightly by the selected Cys-28-Cys-28 disulfide linking helices 2 of the two subunits, and it includes the conservatively selected Leu-27, entailing a comparatively strong hydrophobicity in this sequence region (Fig. 1D), and the two Ile-31 side chains. Interestingly, although Ile-31 is not varied in phage display selection, it is still found at the hydrophobic interface of all Z domain and affibody complexes studied so far (25). The "exterior" face of the A β hairpin is embraced from both sides by the Ile-16 and (selected) Tyr-18 side chains. In this position, Tyr-18 of one $Z_{A\beta3}$ subunit (Z'18 in Fig. 3*E*) also forms a hydrogen bond to the $^{A\beta}$ Glu-22 carboxyl. The N-terminal $Z_{A\beta3}$ β -strands are further anchored against helix 3 in both subunits by nonpolar interactions involving (selected) Val-17 and a salt bridge between Glu-15 and Lys-49. This salt bridge is in fact also present in other Z domain and affibody structures when Glu-15 is in a helical conformation. The selection of alanine or proline at position 24 can be rationalized

Hover et al.

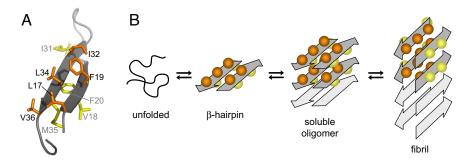


Fig. 4. Hypothetical aggregation mechanism involving the $A\beta\beta$ -hairpin. (A) β -hairpin with the interior hydrophobic core side chains (orange) and the exterior side chains (yellow). (B) Schematic of a hypothetical aggregation mechanism that involves the β -hairpin as a transiently populated conformation sampled by the disordered monomer and/or as a constituent of oligomeric $A\beta$. Soluble oligomers form by hydrophobic stacking of β -hairpins. A concerted conformational transition establishes a fibril seed with in-register, parallel β -sheets. The side chains incorporated in the hydrophobic core within one molecular layer of $A\beta(1-40)$ fibrils correspond to the interior side chains in the $Z_{A\beta3}$: $A\beta(1-40)$ complex. The mechanism is suggested based on the accessibility of the $A\beta\beta$ -hairpin reported here and ATR-FTIR data reported in reference 35. However, it remains to be proven experimentally.

by the tight packing of the disulfide-linked helices, which requires a small nonpolar residue at this position. Residues 25, 32, and 35 are solvent-exposed and not involved in binding, resulting in a less-conserved selection of polar side chains. The terminal fragments ${}^{A\beta}1-{}^{A\beta}15$ of $A\beta(1-40)$ and 1–13 and 57–58 of both $Z_{A\beta3}$ subunits are not well defined by NMR data; NMR chemical shifts close to random coil values and lack of observable NOEs indicate that they are disordered. The C-terminal ${}^{A\beta}$ Val-39 and ${}^{A\beta}$ Val-40 of $A\beta(1-40)$ are ordered as judged from several medium- and long-range NOEs, but their conformation is nevertheless not uniquely defined following structure determination by simulated annealing. NOEs supporting the formation of a salt bridge between the side chains of ${}^{A\beta}$ Asp-23 and ${}^{A\beta}$ Lys-28, which is populated in $A\beta$ fibrils formed under certain conditions (27, 28), could not be detected.

Short peptides, which are homologous to $A\beta$ but contain proline residues as β -sheet blockers, have been developed as fibrillogenesis inhibitors (16). The aim of this and related peptide- and protein-based approaches that target A β aggregation is to bind the hydrophobic part of A β by exploiting the same intermolecular interactions formed in A β self-assembly, e.g., β -sheet backbone hydrogen bonds, and to consequently disrupt the potential for further β -sheet extension (15–17, 29). Although no structure of a β -sheet breaker peptide in complex with A β has been reported, certain aspects of the concept appear to be reflected in the $Z_{A\beta3}$: A β (1-40) interaction: The $Z_{A\beta3}$ β -strands cap the edges of the A β (1-40) β -sheet; the strands are short and terminated on their C-terminal side by a proline and on the N-terminal side by a sequence region selected to have little propensity for either α - or β -structure, resulting in an inability to serve as a template for further β -sheet extension.

Relation of the A β β -Hairpin to the Conformation Within Amyloid

Fibrils. The structure of $A\beta(1-40)$ bound to $Z_{A\beta3}$ shares important characteristics with fibrillar $A\beta$. Within amyloid fibrils, $A\beta$ peptides form β -strand-turn- β -strand motifs, with the precise extent of β -strands and turn varying somewhat depending on $A\beta$ variant and preparation conditions (28, 30–32). Overall, the positions of the secondary structure elements are in good agreement with those determined in this study for bound $A\beta(1-40)$. The hydrophobic cluster identified in $A\beta(1-40)$ fibrils (28), including ${}^{A\beta}Leu$ -17, ${}^{A\beta}Phe$ -19, ${}^{A\beta}Ile$ -32, ${}^{A\beta}Leu$ -34, and ${}^{A\beta}Val$ -36, is also present in the $A\beta(1-40)$: $Z_{A\beta3}$ complex (Figs. 3F and 4A). $A\beta$ amyloid fibrils are stabilized by backbone hydrogen bonding within β -sheets and packing of hydrophobic side-chains. Similarly, $Z_{A\beta3}$ forms an intermolecular β -sheet with $A\beta(1-40)$ and provides a hydrophobic interface for both faces of the $A\beta(1-40)\beta$ -sheet. Thus, $Z_{A\beta3}$ captures $A\beta(1-40)$ in an amyloid-like, but monomeric, conformation and consequently inhibits fibrillation.

Although the two $A\beta(1-40)\beta$ -strands form in-register, intermolecular, parallel β -sheets in amyloid fibrils, they hydrogen bond to each other in the $A\beta(1-40)$:Z_{A\beta3} complex. The two conformations are related by a 90° rotation of both β -strands around their axes (Fig. 4B). Applying a 90° rotation with either direction of rotation to both β -strands in the Z_{A\beta3} bound conformation yields four possible molecular conformations for fibrillar A β (1–40) (28, 32). Although the conformation allowing for formation of the hydrophobic cluster ABLeu-17, ABPhe-19, ^{Aβ}Ile-32, ^{Aβ}Leu-34, and ^{Aβ}Val-36 within one molecular layer appears to be preferred, other conformations might also be populated depending on the conditions of fibril formation, reflecting the structural plasticity of amyloid fibrils (28, 32). In this context, it can be speculated that the β -hairpin presented here constitutes an intermediate conformation on the pathway to amyloid fibrils, e.g., in the form of a transiently populated conformation sampled by the disordered monomer (33) or as a constituent of oligomeric A β (Fig. 4B). In small oligomers, which presumably account for most of the toxicity of amyloidogenic proteins (3, 4, 34), intramolecular β -sheets might be preferred to intermolecular ones because the fibril core structure with its characteristic long-range order is not fully established. Thus, oligomers might form by hydrophobic stacking of β -hairpins and remain soluble as a consequence of the hydrogen bonding capacity of exposed peptide backbones. Fibril seeds could subsequently be generated by a concerted conformational transition toward intermolecular in-register β -sheets (Fig. 4B). The presence of the β -hairpin conformation in oligomers would be in agreement with the observation of a clearly resolvable peak at 1,693 cm⁻¹, indicating an antiparallel β -sheet structure, in the attenuated total reflectance Fourier-transform infrared (ATR-FTIR) spectrum of A β oligomers but not fibrils (35).

Amyloid pores, oligomeric assemblies that have been suggested to insert into membranes and confer neurotoxicity by disrupting cell homeostasis (36), could also be composed of $A\beta(1-40)$ hairpins (data not shown). It is unclear, however, whether their dimensions would suffice to span neuronal membranes and whether their mainly nonpolar inner surface would be compatible with a membrane-permeabilizing activity.

Conclusion

This study establishes the β -hairpin as an accessible conformational state of A β peptides. Its relation to the conformation of A β within amyloid fibrils suggests a role of the β -hairpin in oligomerization and fibrillation. The successful selection of an affibody binding protein that adapts to the conformational preferences of its target by tolerating limited but essential modifications of the scaffold structure highlights the potential of the combinatorial engineering approach. The availability of a binder to monomeric $A\beta$ and its detailed structural and biophysical description will potentiate further investigation of $A\beta$ oligomerization, aggregation, and disaggregation and will help to elucidate to what extent binding and stabilization of monomeric $A\beta$ can interfere with early pathogenic events in AD.

Methods

Proteins. ZAB3 and ZAB3C28S were expressed in Escherichia coli BL21 DE3 cells from plasmid pAY442 containing a T7 promoter (20). In addition to the $Z_{A\beta3}$ affibody sequence displayed in Fig. 1A, the constructs contained an Nterminal (His) $_{\rm 6}$ tag (sequence MGSSHHHHHHLQ) and two C-terminal residues (VD). Cultures were grown at 37°C in LB medium or minimal medium enriched with 1 g/liter ¹⁵NH₄SO₄ and/or 2 g/liter ¹³C-glucose supplemented with 30 μ g/ml kanamycin. Protein expression was induced at OD 0.6–0.8 with IPTG (final concentration 1 mM), followed by further incubation for 4 h. Cells were harvested by centrifugation at 4,000 imes g and frozen at -20 °C. For purification, cells were resuspended in 50 mM Na-phosphate (pH 7.0), 0.2 mM NaCl, and 1 mM PMSF and lysed by sonication. Insoluble material was removed by centrifugation at 16,000 \times g. The (His)_6-tagged protein was isolated by TALON metal affinity chromatography (BD Biosciences) according to the manufacturer's instructions. Further purification was achieved by size exclusion chromatography employing an ÄKTA Explorer system equipped with a HiLoad 16/60 Superdex 75 prep grade column (GE Healthcare). The purified affibody proteins were dialysed against 20 mM Na-phosphate (pH 7.2).

A β (1–40) was obtained either unlabeled (NaOH-purified) or uniformly ¹⁵N or ¹³C/¹⁵N-labeled [trifluoroacetic acid (TFA)-purified] from a commercial source (rPeptide). A β (1–40) was dissolved in 30 mM NH₄OH (NaOH-purified samples) or 100 mM NH₄OH (TFA-purified samples) at a concentration of \approx 1.5 mM, centrifuged at 20,800 × g to remove any larger aggregates, and diluted into the final experiment buffer. Protein concentrations were determined by UV absorbance at 280 nm, using extinction coefficients validated by amino acid analysis (Amino Acid Analysis Center, Uppsala University, Uppsala, Sweden). The purities of all proteins were >98% as estimated by SDS/PAGE.

Isothermal Titration Calorimetry (ITC). ITC was carried out on a VP-ITC calorimeter (MicroCal) at 20°C in 20 mM Na-phosphate (pH 7.2). $A\beta(1-40)$ at concentrations of 7–16 μ M was used as titrant in the cell and $Z_{A\beta3}$ at a 9-fold higher concentration (calculated for the dimer) or $Z_{A\beta3}C28S$ at a 60-fold higher concentration as titrant in the syringe. All solutions were degassed before experiments. Baseline correction and integration of the calorimeter response was carried out with the Origin software (MicroCal) provided with the calorimeter. To correct for heats of dilution and viscous mixing, the average heat of postsaturation injections was subtracted from each injection. The obtained binding isotherms were fitted to a model in which the variable parameters are the stoichiometry of identical sites (*N*), an apparent dissociation constant (*K*_d), and an apparent association enthalpy (ΔH).

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Circular Dichroism (CD) Spectroscopy. CD was performed on a JASCO J-810 spectropolarimeter. Melting curves were recorded at 220 nm, using a 0.1 cm path-length cell containing proteins at concentrations of 17.5 μ M [A β (1–40), Z_{A β 3} dimer] or 35 μ M (Z_{A β 3}C28S) in 20 mM Na-phosphate, pH 7.2.

Aggregation Assay. Thioflavin T fluorescence was recorded in 96-well plates (Nunc), using a FLUOstar Optima reader (BMG Labtech) equipped with 440-nm excitation and 480-nm emission filters. The samples contained 100 μ l of 115 μ M A β (1–40) in 50 mM Na-phosphate (pH 7.1), 0.1 M NaCl, 10 μ M thioflavin T, and 0.1% Na-azide supplemented with the indicated amount of Z_{A β 3}. Plates were sealed with polyolefin tape (Nunc) and incubated at 37°C. Data points were recorded every 15 min with 5 min of orbital shaking (width 5 mm) before the measurement.

NMR and Structure Determination. NMR data were collected at 25°C, using Varian Inova 800 and 900-MHz spectrometers, the latter of which was equipped with a cryogenic probe. NMR samples for structure determination contained \approx 400 μ M ¹⁵N-labeled or ¹³C,¹⁵N-labeled A β (1-40) or disulfidelinked $Z_{A\beta3}$ dimer, and 15% molar excess of unlabeled $Z_{A\beta3}$ dimer or $A\beta(1-40)$, respectively, in 20 mM Na-phosphate buffer at pH 7.2. Resonance assignments were obtained from standard triple-resonance experiments. Interproton distance constraints were derived from 3D ¹⁵N-NOESY and 3D ¹³C-NOESY spectra recorded with mixing times of 120 ms and calibrated by using known distances in regular secondary structure elements. Intermolecular nuclear Overhauser effects (NOEs) were also identified in 3D F1(13C, 15N)-filtered F2(13C or 15N)edited NOESY experiments (37). Backbone dihedral angle constraints were derived from chemical shifts, using TALOS (38). Backbone hydrogen bond donors were identified in amide hydrogen exchange experiments (see SI Text and Fig. S2) and acceptor carbonyl oxygens were identified based on initial structure calculations (see SI Text). The final constraint dataset (Table S1) contained 3,438 NOE distances, of which 387 are unambiguous intermolecular distances, 160 are backbone dihedral angle constraints, and 39 are hydrogen bonds. Structures were calculated with Xplor-NIH 2.15.0 (39), using ab initio simulated annealing with r⁻⁶-averaging. A pseudopotential for the radius of gyration (40) was used on residues 16–40 of A β (1–40) and residues 14–56 of the two ZAB3 subunits to improve packing, and a conformational database potential (41) was used to improve dihedral angle distributions. The full Lennard-Jones potential, an electrostatic potential scaled to 25% of the default value, and parameters for proper covalent disulfide bond geometry were applied during the final slow-cooling refinement. An ensemble of 24 (of 100) structures was selected. Backbone and all-heavy atom rms deviations between structures in the ensemble are 0.32 Å and 0.71 Å, respectively, and 93% of the residues are found in the most favored regions of the Ramachandran diagram (see Table S2 for additional statistics). Hydrogen bonds shown in Fig. 3*E* have donor hydrogen-acceptor distances <2.4 Å and angles between the donor bond vector and the vector connecting the two heavy atoms of < 35° in at least 50% of the structures in the ensemble. Molecular graphics figures were created by using PyMOL (DeLano Scientific).

ACKNOWLEDGMENTS. This work was supported by the Swedish Research Council, the German Academic Exchange Service, and the Mucosal Immunobiology and Vaccine Swedish Foundation for Strategic Research Center.

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