
Unique stabilizing interactions identified in the two-stranded α -helical coiled-coil: Crystal structure of a cortexillin I/GCN4 hybrid coiled-coil peptide

DARIN L. LEE,^{1,2} SERGEI IVANINSKII,³ PETER BURKHARD,³ AND ROBERT S. HODGES^{1,2}

¹Department of Biochemistry, University of Alberta, Edmonton, Alberta T6G 2H7, Canada

²Department of Biochemistry and Molecular Genetics, University of Colorado Health Sciences Center, Denver, Colorado 80262, USA

³M.E. Müller Institute for Structural Biology, Biozentrum, University of Basel, CH-4056 Basel, Switzerland

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Abstract

We determined the 1.17 Å resolution X-ray crystal structure of a hybrid peptide based on sequences from coiled-coil regions of the proteins GCN4 and cortexillin I. The peptide forms a parallel homodimeric coiled-coil, with C α backbone geometry similar to GCN4 (rmsd value 0.71 Å). Three stabilizing interactions have been identified: a unique hydrogen bonding–electrostatic network not previously observed in coiled-coils, and two other hydrophobic interactions involving leucine residues at positions **e** and **g** from both **g-a'** and **d-e'** interchain interactions with the hydrophobic core. This is also the first report of the quantitative significance of these interactions. The GCN4/cortexillin hybrid surprisingly has two interchain Glu-Lys' ion pairs that form a hydrogen bonding network with the Asn residues in the core. This network, which was not observed for the reversed Lys-Glu' pair in GCN4, increases the combined stability contribution of each Glu-Lys' salt bridge across the central Asn15-Asn15' core to ~0.7 kcal/mole, compared to ~0.4 kcal mole⁻¹ from a Glu-Lys' salt bridge on its own. In addition to electrostatic and hydrogen bonding stabilization of the coiled-coil, individual leucine residues at positions **e** and **g** in the hybrid peptide also contribute to stability by 0.7 kcal/mole relative to alanine. These interactions are of critical importance to understanding the stability requirements for coiled-coil folding and in modulating the stability of de novo designed macromolecules containing this motif.

Keywords: Coiled-coil; protein folding; protein stability; protein oligomerization; electrostatics; side-chain packing; circular dichroism spectroscopy

The rodlike α -helical coiled-coil is one of the simplest yet most common structural motifs occurring in proteins. Consisting of two to five α -helices twisted into a left-handed supercoil, the occurrence of this structure is well docu-

mented, occurring in a wide variety of proteins including motor proteins, DNA binding proteins, extracellular proteins, and viral fusion proteins (Lupas 1996; Kohn et al. 1997; Burkhard et al. 2001). The presence of a continuous interface of hydrophobic amino acids along the length of the helices provides a major source of stability to the fold as the hydrophobes pack in a knobs-into-holes fashion shielded from the bulk solvent (Crick 1953). The pattern of repeating hydrophobic residues at positions **a** and **d** of the heptad repeat (denoted *abcdefg*) that are responsible for coiled-coil formation was first identified by Hodges et al. (1972) from the amino acid sequence of tropomyosin. This 3–4 or 4–3

Reprint requests to: Robert S. Hodges, Department of Biochemistry and Molecular Genetics, Box B121, University of Colorado Health Sciences Center, 4200 East 9th Avenue, Denver, CO 80262, USA; e-mail: robert.hodges@uchsc.edu; fax: (303) 315-1153.

Abbreviations: CD, circular dichroism; HPLC, high-performance liquid chromatography; TFE, 2, 2, 2-trifluoroethanol; PEG, polyethylene glycol; rmsd, root-mean-square deviation.

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hydrophobic repeat allows for the prediction of coiled-coils based on statistical occurrence of residues in these positions (Berger et al. 1995; Berger and Singh 1997; Wolf et al. 1997; Singh et al. 1999), and more recently, with STABLECOIL, an algorithm based on experimentally derived stability data (Wagschal et al. 1999a; Tripet et al. 2000; Tripet and Hodges 2001). However, deletion and/or mutational analysis of some protein sequences (such as GCN4, cortexillin I, macrophage scavenger receptor, and intermediate filament chains) has revealed that the presence of heptad repeats does not always guarantee the formation of coiled-coil structure (Steinmetz et al. 1998; Frank et al. 2000; Wu et al. 2000; Kammerer et al. 2001; Lee et al. 2001). The absence of a key region of only one or two heptads (dubbed the ‘trigger sequence’) prevented folding, whereas its presence induced folding of the coiled-coil structure. We previously took a 31-residue peptide that did not fold even with an excellent hydrophobic core (two valine, one leucine, and one methionine residue at position **a** and four leucine residues in position **d**, shown by Wagschal et al. [1999a] and Tripet et al. [2000] to provide excellent stability) and then introduced stabilizing amino acid substitutions to induce dimeric coiled-coil folding without matching a proposed ‘consensus trigger sequence,’ to prove that such a sequence is not necessary for coiled-coil folding (Lee et al. 2001). Although trigger sequences may be important for folding in naturally occurring coiled-coils of substantial length, we proposed that any sequence with enough stability above a critical threshold will fold, whether the stability is distributed evenly along the coiled-coil sequence or concentrated heavily in distinct regions. Thus, the ability to induce coiled-coil folding is not limited to a consensus trigger sequence, but should be present in any peptide sequence with sufficient individual helix and/or coiled-coil stability (Burkhard et al. 2000a, 2002). This is further supported by the large number and variety of designed synthetic and recombinant peptide sequences that do not contain the consensus trigger sequence, yet still fold into two-stranded coiled-coils (Dieckmann et al. 1998; Harbury et al. 1998; Kohn et al. 1998; Micklatcher and Chmielewski 1999; Pandya et al. 2000; McClain et al. 2001; Acharya et al. 2002; Arndt et al. 2002; Campbell et al. 2002; Litowski and Hodges 2002; Phelan et al. 2002; Havranek and Harbury 2003).

Here we determined the crystal structure for the most stable folded peptide from our previous study, which did not contain a ‘consensus trigger sequence’. This stable peptide contained potentially stabilizing design elements over the nonfolding sequence, including increased helix propensity, elimination of ionic repulsions, and introduction of ionic attractions. These stabilizing elements are identified in the 1.17 Å crystal structure. We also quantify the previously unreported stability contribution of three interactions using thermodynamic analysis of urea denaturation curves of pep-

tide analogs: a unique complex hydrogen bonding–electrostatic network involving i to $i' + 5$ electrostatic interactions and hydrogen bonding interactions with Asn in the hydrophobic core, and hydrophobic packing of leucine residues at positions **e** (d-e' interaction) and **g** (g-a' interaction).

Results

Peptide design

Our previous results showed that the native peptide (Hybrid 1) did not fold (Lee et al. 2001), even though it had an excellent hydrophobic core which makes a major contribution to overall stability because of the one leucine, one methionine, and two valine residues at position **a** (Wagschal et al. 1999a) and the four leucine residues at position **d** (Fig. 1; Tripet et al. 2000). This sequence was based primarily on the coiled-coil regions of the transcription factor, GCN4, and the actin bundling protein, cortexillin I. The N-terminal (residues 2 to 14) contains cortexillin I 270–282 (Burkhard et al. 2000b), whereas the C-terminal (residues 16–28) contains GCN4-p1 residues 3 to 15 (O’Shea et al. 1991). Both segments were inserted into the GCN4-p1 sequence 2–32, labeled as 1–31 in Figure 1. In the middle of this peptide sequence, residue 15 is a polar asparagine (**a** position) buried in the hydrophobic core. Polar residues in core **a** and **d** positions reduce coiled-coil stability (Wagschal et al. 1999a; Tripet et al. 2000; Zhu et al. 2000; Akey et al. 2001) versus more hydrophobic amino acids, but asparagine was left in the GCN4 template to specify dimerization over higher-order oligomerization states such as trimers or tetramers (Harbury et al. 1993). This central asparagine is analogous to the central Asn observed in the coiled-coil structure of GCN4-p1 (O’Shea et al. 1991).

#	Peptide	1	8	15	22	29	
		abcdefgabcdefgabcdefgabcdefgabc					
	GCN4-p1	Ac-MKQL	EDKVE	EELLS	KNYHLE	NEVARL	KKLVGE-amide
	Hybrid 1	Ac-MDQL	NSLLAS	LESENK	QLEDK	VEELL	LSKVGE-amide
1	Hybrid 2	Ac-MDQL	NALLAS	LEAENK	QLKAK	VEELL	AKVGE-amide
2	L7A	Ac-MDQL	NALLAS	LEAENK	QLKAK	VEELL	LAKVGE-amide
3	L26A	Ac-MDQL	NALLAS	LEAENK	QLKAK	VEELL	LAKVGE-amide
4	E14A	Ac-MDQL	NALLAS	LEAENK	QLKAK	VEELL	LAKVGE-amide
5	K19A	Ac-MDQL	NALLAS	LEAENK	QLKAK	VEELL	LAKVGE-amide
6	E14A, K19A	Ac-MDQL	NALLAS	LEAENK	QLKAK	VEELL	LAKVGE-amide
7	K19E	Ac-MDQL	NALLAS	LEAENK	QLKAK	VEELL	LAKVGE-amide

Figure 1. Sequences of peptides used in this study. Peptides are denoted by one-letter code and are labeled peptides 1 to 7. The sequence of GCN4-p1 (O’Shea et al. 1991) residues 2–32 is shown for comparison, where Met 2 of GCN4-p1 is labeled residue 1. Hybrid 1 is a GCN4-cortexillin I hybrid that did not fold into a coiled-coil, whereas Hybrid 2 formed a stable coiled-coil (Lee et al. 2001). The **a** and **d** positions of the heptad repeat **abcdefg** are boxed. Bolded residues indicate amino acid substitutions in Hybrid 2 relative to Hybrid 1. The circled and bolded residues in peptides 2 to 7 indicate amino acid substitutions relative to the Hybrid 2 peptide sequence. The residues 2–14 in the Hybrid 1 sequence are from cortexillin I residues 270–282, and residues 16–28 contain GCN4-p1 2–14, where both segments are inserted into the GCN4 template 1–31.

Because Hybrid 1 did not fold, we introduced amino acid substitutions into Hybrid 1 that would induce folding independently of a ‘consensus trigger sequence.’ The result was Hybrid 2, whose structure was determined in the present study. Hybrid 2 exhibited increased coiled-coil folding and stability over Hybrid 1 because of five amino acid substitutions: S6A, S13A, D20A, S27A, and E19K. Hybrid 2 contains four alanine residues in the **f** positions (A6, A13, A20, and A27) not found in GCN4 or cortexillin I, in order to increase α -helix propensity (Fig. 2). Also, a lysine replaced a glutamic acid at position 19 (**e** position), which eliminated a potential i to $i' - 5$ ionic repulsion with Glu 14' (**g'** position) in the original non-folding peptide, and could instead form a potential interhelical i to $i' - 5$ salt bridge across the hydrophobic interface (Fig. 2).

Overall fold

The x-ray crystal structure of Hybrid 2 (Table 1) is a parallel two-stranded coiled-coil (Fig. 3). The dimer of the coiled-coil is built up from an internal two-fold crystal symmetry element. Two core residues at the dimer interface (Leu 8 and Asn 15), each in **a** positions of the heptad repeat, show two rotamer conformations (Figs. 4, 5). Asn 15 adopts the two most preferred rotamer conformations (Lovell et al. 2000) with χ_1/χ_2 torsion angles of $-82^\circ/14^\circ$ and $169^\circ/-90^\circ$, respectively. In contrast to the two-fold symmetry observed in Hybrid 2, the crystal structure of another coiled-coil, GCN4-p1 (sequence in Fig. 1; O'Shea et al. 1991) does not show two-fold symmetry but still shows Asn 15 in two different conformations (in and out).

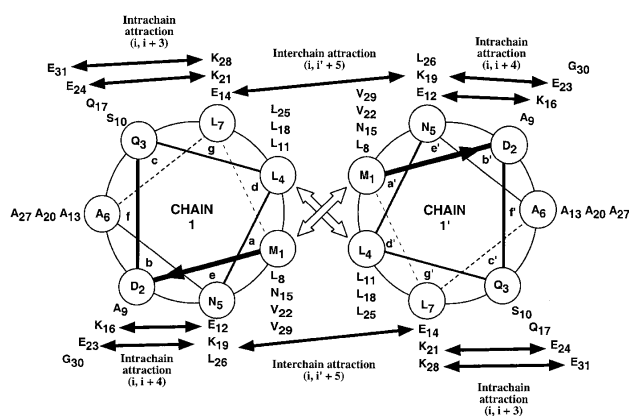


Figure 2. End-on view of the hybrid peptide analog, viewed from N- to C-terminus. Solid arrows denote predicted interchain (i to $i' + 5$) and intrachain (i to $i + 3$; i to $i + 4$) ion pairing between residues. Open arrows denote the hydrophobic interactions between residues **a** and **a'** or **d** and **d'**.

Table 1. Data collection and refinement statistics

Space group	C222 ₁
Unit cell a, b, c (Å)	18,098, 117,287, 2,209
α, β, γ (°)	90, 90, 90
Monomers/au	1
<i>Data collection statistics</i>	
Resolution (Å)	1.17
Observed reflections	263880
Unique reflections	8183
Completeness (%)	97.8
R_{sym}^a	0.037
<i>Refinement statistics</i>	
R factor (%) ^b	17.9
R_{free} (%) ^b	21.7
Mean B factor protein atoms (Å ²)	15.1
Mean B factor water atoms (Å ²)	31.3
rmsd bond distances (Å) ^c	0.009
rmsd bond angles (°) ^c	1.8

$$^a R_{\text{sym}} = \sum |I - \langle I \rangle| / \sum I.$$

$$^b R \text{ factor} = \sum |F_{\text{obs}}| - |F_{\text{calc}}| / \sum |F_{\text{calc}}|.$$

^c Root-mean-square deviation.

Interactions in the Asn 15 core region

The two asparagines in core **a** positions (Asn 15 and Asn 15'), together with the two Glu (14 and 14') and two Lys

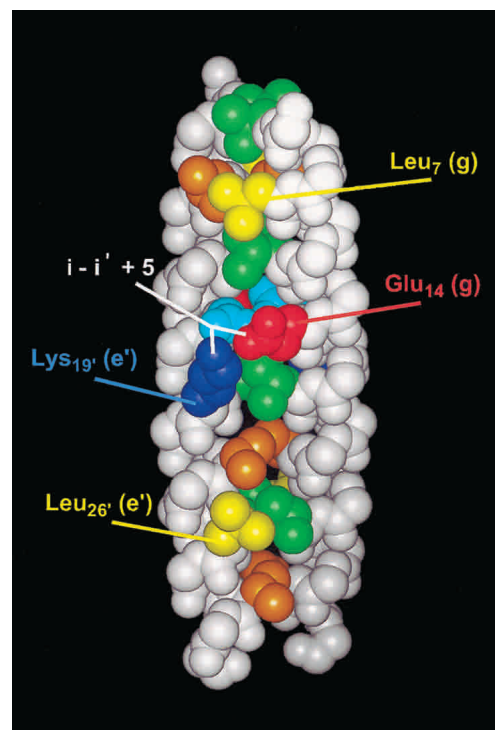


Figure 3. Space-filling representation of the GCN4/cortexillin Hybrid 2 structure. Residues in positions **a** and **d** along the hydrophobic core are displayed in orange and green, respectively, with the exception of Asn 15 (**a** position) in light blue. Leu 7 (**g**) and Leu 26 (**e**) side chains are displayed in yellow. Glu 14 (**g**) and Lys 19' (**e'**) side chains are displayed in red and dark blue, with the i to $i' + 5$ interchain interaction shown in white.

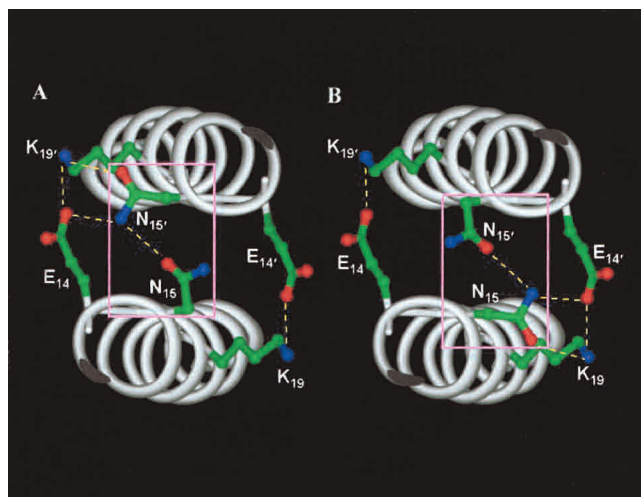


Figure 4. Cross-sectional view of the region surrounding the Asn 15 (a) core in Hybrid 2. Hydrogen bonds are denoted by dashed yellow lines. Oxygen atoms are red; nitrogen atoms are blue. The C_{α} helix backbone is shown in white (ribbon representation). The two interconverting conformers of Asn 15 and Asn 15' ('in' or 'out') are inside the pink boxes of panels A and B. (A) Asn 15 is 'in' (lower helix); Asn 15' is 'out' (upper helix). (B) Asn 15 is 'out' (lower helix); Asn 15' is 'in' (upper helix).

(19 and 19') residues, form an extended network of hydrogen bonds with a total of five (four interhelical and one intrahelical) hydrogen bonds not previously observed in the structures of coiled-coils (Fig. 4). The amide and carbonyl

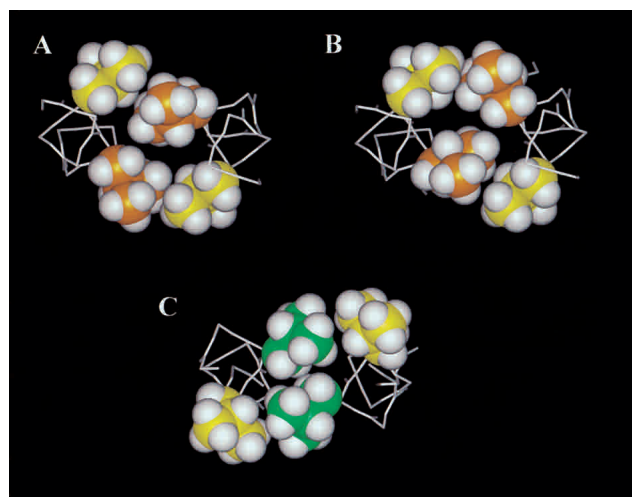


Figure 5. Cross-sectional comparison of **g-a** packing and **d-e** packing of leucines in the hydrophobic core of the hybrid peptide analog, viewed from N-terminus to C-terminus. The C_{α} helix backbone is shown as white sticks, and hydrogen atoms are depicted as white balls. Panels A and B: Leu 7 (g) and Leu 7' (g') side chains are shown in yellow; Leu 8 (a) and Leu 8' (a') side chains are shown in orange. The two conformations of Leu 8 (a) and Leu 8' (a'), labeled from *left to right*, are (a) 'out/in' and (b) 'in/out'. (C) Packing of Leu 25 (d), in green, and Leu 26 (e), in yellow, in the hydrophobic interface.

groups of the two Asn 15 side-chains are 3.1 Å apart from each other. The $N_{\delta 2}$ atom from asparagine with the 'out' conformation also forms a 2.8 Å hydrogen bond with the carboxyl group oxygen of Glu 14 (Fig. 4a) and Glu 14' (Fig. 4b). Glu 14 – Lys 19' and Glu 14' – Lys 19 form the two interhelical salt bridges with an N_{ζ} to O_{e1} distance of 3.2 Å on both sides of the coiled-coil. The lone intrahelical hydrogen bond occurs between the O_{δ} of Asn 15 in the 'out' conformation and the N_{ζ} of Lys 19.

In the GCN4-p1 coiled-coil, there is an interchain electrostatic interaction on only one side of the coiled-coil with an interatomic distance of 3.7 Å. However, there is no interaction with the nearest Asn residue in the hydrophobic core. On the other side of GCN4-p1, the side chains of Lys and Glu are 6.1 Å apart (Fig. 6b) and thus are too far apart to interact with each other, or with Asn 15 (indicated by the red X's in Fig. 6b). This can be explained by the fact that in the GCN4-p1 peptide, the charged residues are in opposite positions, and due to the stereochemical restraints, the favorable hydrogen bonding pattern as observed in the Hybrid 2 structure cannot form. This is a clear indication that the configuration with Glu in the **g** position and Lys in the **e'** position (as it is in Hybrid 2) is more stabilizing to the coiled-coil than the opposite arrangement, if Lys is in the **g** position and Glu is in the **e'** position (as it is in the GCN4-p1 coiled-coil).

Crystal contacts

Surprisingly, none of the proposed intrahelical salt bridges are formed in the crystal. Instead, all residues thought to form such interactions are involved in crystal lattice contacts. Lys 21 and Glu 24 (potential i to $i + 3$ ion pair) form

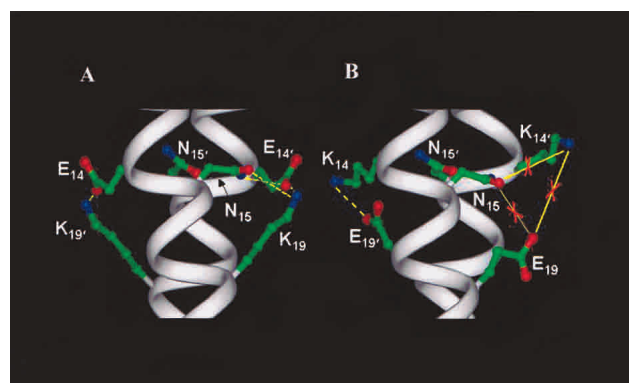


Figure 6. Comparison of the region around the central Asn (a) in Hybrid 2 (A) and GCN4 (B) coiled-coils. The numbering of residues in GCN4 has been aligned to the hybrid sequence in Fig. 1. Oxygen atoms are red; nitrogen atoms are blue. Hydrogen bonds are denoted by dashed yellow lines. The C_{α} helix backbone is shown in white (ribbon representation). Solid yellow lines with a red 'x' denote a lack of noncovalent bonding between atoms.

salt bridges with the residues Glu 24 and Lys 21 from a symmetry-related molecule; the distance between these residues is approximately 4 Å. Glu 12 is involved in an interaction with Asn 15 of a symmetry-related molecule, and does not form an i to $i + 4$ intrahelical salt bridge with Lys 16. Glu 23 forms a salt bridge with Lys 28 from a symmetry-related molecule, despite the possibility that these residues could each form intrahelical salt bridges with Lys 19 (i to $i + 4$) and Glu 31 (i to $i + 3$), respectively. Also, three hydrophobic alanine residues in **b** and **f** positions—Ala 6(**f**), Ala 9(**b**), and Ala 13(**f**)—pack against identical residues from a crystallographically related molecule. Thus, crystal packing forces may disrupt intrachain interactions that would be very important in solution, in favor of intermolecular interactions in the crystal.

Comparison with GCN4

The peptide backbone can be superimposed upon the GCN4-p1 coiled-coil dimer, giving an rmsd of 0.71 Å for C_{α} atoms from residues 3 to 30. The largest differences in the positions of the C_{α} backbone atoms occur at the N-terminus. In contrast, the C_{α} atoms of the C-terminal portion of the coiled-coil (residues 15–29) can be superimposed with an rmsd of only 0.36 Å. Hybrid 2 has a leucine residue in an N-terminal **a** position (Leu 8) compared to a smaller valine residue in GCN4-p1 that may explain the difference in the C_{α} backbone between GCN4-p1 and Hybrid 2. The C_{α} – C_{α} distance between Leu 8 and Leu 8' at position **a** is 6.3 Å, whereas in the GCN4-p1 structure the Val 8 and Val 8' C_{α} carbons are separated by 5.5 Å (see Fig. 1 for sequences). According to the program TWISTER (Strelkov and Burkhard 2002), the coiled-coil radius of Hybrid 2 increases towards the N-terminus and also the coiled-coil pitch of Hybrid 2 increases, indicating a local unwinding of the coiled-coil (Fig. 7): the C_{α} – C_{α} distance between the two

Leu 4 residues at position **d** gets as high as 8.4 Å, whereas in GCN4-p1 it is only 6.3 Å. Clearly, packing the larger Leu residue instead of Val at position **a** results in the increase in coiled-coil radius, as position 8 has the only residue that is different in the hydrophobic **a** and **d** positions of GCN4-p1 versus Hybrid 2.

Biophysical characterization of peptides

To quantitatively determine the extent that residues in the **e** and **g** positions stabilize the coiled-coil structure of Hybrid 2, we synthesized a set of analogs substituted at various **e** and **g** positions along the sequence (peptides 2 to 7 in Fig. 1). We characterized the secondary structure, oligomerization state, and thermodynamic stability of the peptides by circular dichroism (CD), sedimentation equilibrium, and chemical unfolding experiments (Figs. 8, 9; Table 2). All peptides formed α -helices in benign medium and are folded as α -helical coiled-coils based on the following characteristics: high negative molar ellipticity values in benign medium, with double minima at 208 and 222 nm and a maximum at 190 nm; and a high α -helix content such that the addition of 50% trifluoroethanol (TFE) does not significantly increase the α -helix content (determined at 222 nm; Table 2). Furthermore, the ratio of molar ellipticities at 222 nm and 208 nm suggests that all of the peptides studied formed coiled-coils: the $[\theta]_{222}/[\theta]_{208}$ ratio for coiled-coils in benign medium is usually greater than 1.0 (Lau et al. 1984), and this value decreases to 0.90 in the presence of the monomeric helix-inducing solvent, trifluoroethanol. Also, results from sedimentation equilibrium experiments indicated that most of peptides in the present study were globally fit to a single dimeric species in benign medium, with observed molecular weights that are approximately double the calculated weight of the monomers (Table 2).

Peptides E14A and E14A, K19A exhibited higher-order oligomerization in benign medium. E14A was best fit to a monomer-trimer equilibrium (MW_{obs} 8087), and E14A, K19A was best fit to a monomer-tetramer equilibrium (MW_{obs} 9896). These peptides showed essentially no change in molar ellipticities (or helicities) in the pretransition region (0 to 2 M urea) of the urea unfolding curves (Fig. 9). As shown previously by Wagschal et al. (1999b), the oligomerization state of coiled-coils can change from a higher-order state (in the absence of denaturant) to the two-stranded state at levels of denaturant that are below the denaturant concentration required for the start of the transition from the folded to unfolded state. Therefore, we performed the sedimentation equilibrium analyses of peptide E14A and peptide E14A, K19A in 2 M urea to show that the coiled-coil analogs are two-stranded prior to unfolding. Under these conditions, E14A existed as a monomer-to-dimer equilibrium, and E14A, K19A existed as a single dimeric species (Table 2). This is important because the unfolding

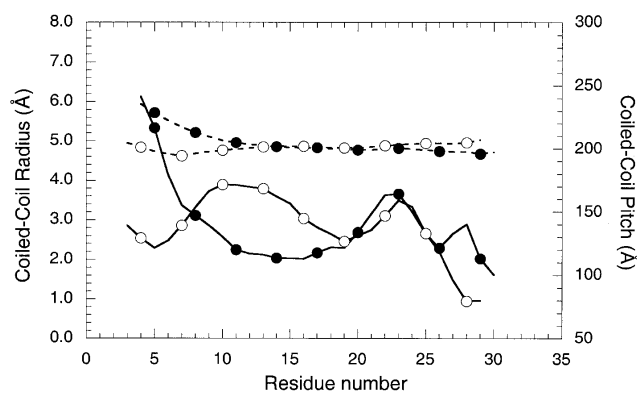


Figure 7. Coiled-coil pitch (solid lines) and coiled-coil radius (dashed lines) plotted against the residue number for the Hybrid 2 (●) and the GCN4-p1 peptide (○). The residue number is according to the Hybrid 2 peptide in both sequences, i.e., in GCN4-p1 they are shifted by one residue.

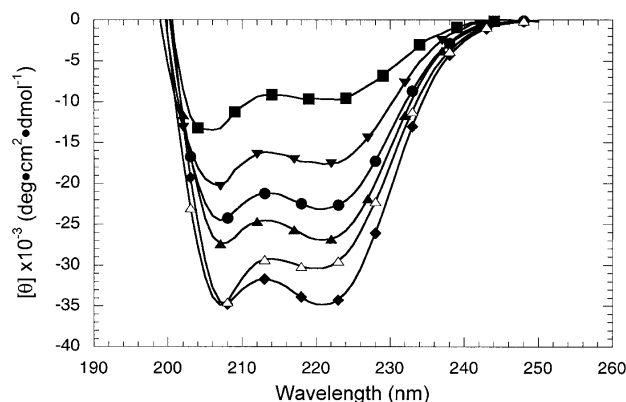


Figure 8. CD spectra of the peptide K19E obtained at 20°C in 50 mM phosphate, 100 mM KCl, pH 7.0 buffer (benign). Peptide concentrations were 2592 μM (\blacklozenge), 432 μM (\blacktriangle), 86 μM (\bullet), 22 μM (\blacktriangledown), 4 μM (\blacksquare) and in 1:1 (v:v) benign buffer and TFE at 432 μM (\triangle).

curve must represent the transition from the same oligomerization state to the same unfolded state for all analogs in order for the calculated $\Delta\Delta G_u$ values to be meaningful.

By monitoring the folded state of the coiled-coil as a function of denaturant concentration using CD spectroscopy, we obtained values for the urea concentration at the transition midpoint ($[\text{urea}]_{1/2}$), the slope at the transition midpoint (m), and the stability contributions for Hybrid 2 relative to less stable analogs ($\Delta\Delta G_u^{\text{obs}}$; Fig. 9, Table 3). Peptides were analyzed at $\sim 400 \mu\text{M}$ concentration to ensure that the total population of the entire set of peptides was essentially fully folded in the absence of denaturant; the effect of peptide concentration on the CD spectra of the least stable peptide in this study, K19E (peptide 7 in Fig. 1), is shown in Figure 8. Although K19E shows an increase in negative molar ellipticity from 432 μM to 2600 μM , indicating greater helicity at higher concentrations, it was calculated to be 91% helical at $\sim 400 \mu\text{M}$ by comparing $[\theta]_{222}$ values in benign conditions and in 50% TFE (Table 2). All the other peptides were more stable and more helical than K19E (Fig. 9) at 400 μM concentration in benign conditions (Table 2). Hybrid 2 was the most stable peptide in this study, so $\Delta\Delta G_u^{\text{obs}}$ was reported as a positive value, that is, a stabilizing contribution, by calculating the stability difference of Hybrid 2 over the less stable analog (Table 4).

The choice of substituting the residues by alanine to eliminate *g* to *e'* interchain effects such as ionic interactions or hydrophobic interactions also affects another parameter that influences coiled-coil stability: α -helix propensity. Alanine has the highest intrinsic helix propensity among the twenty naturally occurring amino acids (O'Neil and De-Grado 1990; Blaber et al. 1993; Monera et al. 1995; Myers et al. 1997; Pace and Scholtz 1998), and so the difference in helix propensities between Ala and the substituted residue was determined and corrected in the final column of Table 4. Alanine is an ideal residue to carry out substitutions in

our case because it does not disrupt the α -helical coiled-coil structure, as it might if one were inserting it into an alternate nonhelical structure (Minor Jr. and Kim 1994). After correction, the average value for the strength of a *g* to *e'* ion pair was 0.74 kcal mole $^{-1}$ (averaged over the $\Delta\Delta G_u^{\text{obs}}$ values of single substitutions E14A; K19A; and double substitutions E14A, K19A). Leucine was not corrected for helix propensity, although its value has previously been shown to be similar to Ala (Pace and Scholtz 1998).

Interactions of leucine residues at positions *e* and *g*

Peptides L7A and L26A contain destabilizing substitutions by replacing leucine residues at positions *g* and *e* respectively with the shorter, less hydrophobic alanine side chains. L7A reduced stability by 0.74 kcal mole $^{-1}$, whereas L26A reduced stability by 0.69 kcal mole $^{-1}$ per substitution. Overall, leucine residues at *e* and *g* each contributed ~ 0.7 kcal mole $^{-1}$ to stability relative to alanine, whereas the interchain Glu-Lys' salt bridge involved in the hydrogen bonding network contributed 0.74 kcal mole $^{-1}$ to stability after correction for helix propensity.

Discussion

Unlike the *a* and *d* positions in the hydrophobic core, the environment of side chains located at the *e* and *g* positions in a two-stranded coiled-coil is partially solvent-excluded but also partially solvent-exposed (Fig. 2). Therein lies the controversy over the effects of interchain ion pairs located at these positions, as the stabilizing benefit of an electrostatic attraction must outweigh the energy penalty from both desolvation and the presence of charged residues in a hydro-

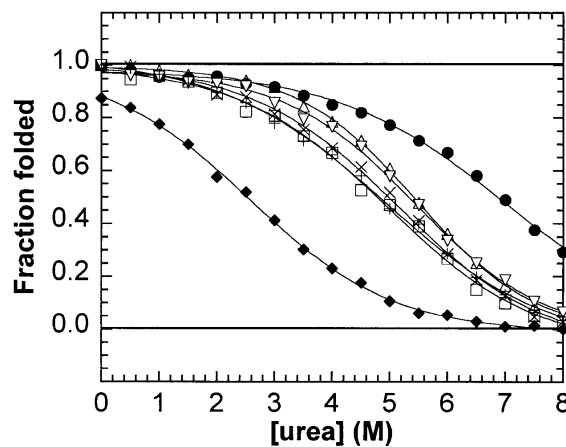


Figure 9. Chemical unfolding curves of the Hybrid 2 peptide and analogs. Fraction folded was plotted against denaturant concentration for Hybrid 2 (\bullet) and peptides E14A (∇); K19A ($+$); L7A (\square); L26A (\times); E14A, K19A (\triangle); and K19E (\blacklozenge).

Table 2. Biophysical characterization of hybrid peptide and analogs

Peptide	$[\theta]_{222}$ ($\times 10^3$ deg cm ² dmole ⁻¹) ^a		$[\theta]_{222}/[\theta]_{208}$ benign ^b	$[\theta]_{222}/[\theta]_{208}$ TFE ^c	Percent helix ^d benign	Helical residues ^e	m_{calc} (Da) ^f	m_{obs} (Da) ^g
	Benign	TFE						
Hybrid 2	-31.5	-30.6	1.00	0.86	103	33	3439	6698 (D) ^h
L7A	-29.9	-31.8	1.02	0.87	94	29	3397	6556 (D)
L26A	-29.8	-31.1	1.03	0.90	96	30	3397	6266 (D)
E14A	-29.4	-29.7	1.03	0.89	99	31	3381	6162 (M-D)*
K19A	-30.8	-30.3	1.03	0.89	102	32	3382	6383 (D)
E14A, K19A	-31.3	-33.0	1.02	0.89	95	29	3324	6524 (D)*
K19E	-27.7	-30.3	1.01	0.88	91	28	3440	6618 (D) ^h

^a Molar ellipticity at 222 nm at 20°C in benign buffer (50 mM potassium phosphate, 100 mM KCl, pH 7.0), 400 μM peptide concentration. For samples containing trifluoroethanol (TFE), the above buffer was diluted 1:1 (v:v) with TFE.

^b Ratio of molar ellipticities at 222 nm and 208 nm in benign buffer.

^c Ratio of molar ellipticities at 222 nm and 208 nm in a mixture of 1:1 (v:v) benign buffer and trifluoroethanol.

^d The percent α-helix was calculated by taking 100% α-helix as the $[\theta]_{222}$ value obtained in 50% TFE (the maximum inducible α-helical structure).

^e The number of α-helical residues was calculated by multiplying the percent α-helix (column d) by the number of residues in the polypeptide chain, 31.

^f Calculated molecular mass of monomeric peptides based on average isotopic masses.

^g Observed molecular mass based on global fits of sedimentation equilibrium data (see Materials and Methods). Data was best fit to a monomer to dimer equilibrium (M-D), or dimer (D). Values labeled with an asterisk (*) were determined in the presence of 2 M urea to eliminate higher order oligomerization states observed in benign buffer. The sedimentation equilibrium data in benign buffer was best fit to a monomer-trimer equilibrium for E14A (MW_{obs} 8087) and a monomer-tetramer equilibrium for E14A, K19A (MW_{obs} 9896). See results for further details.

^h Previously determined sedimentation equilibrium values (Lee et al. 2001).

phobic environment. Similarly, hydrophobic residues such as the long aliphatic chains of leucine and isoleucine, while gaining a favorable energy contribution to stability in the hydrophobic environment, may suffer a penalty because of their solvent accessibility at the **e** and **g** positions.

Electrostatic interactions at positions **e** and **g**

In comparison with other values for interchain Glu-Lys' attractions (ranging from 0.4 to 0.5 kcal mole⁻¹; Zhou et al. 1994; Krylov et al. 1998) and Glu-Glu' repulsions (0.45 kcal mole⁻¹; Kohn et al. 1995) in **g-e'** positions of coiled-

coils, our values for each electrostatic attraction in this study are greater by 0.2 to 0.3 kcal mole⁻¹ (Table 4). The increased stability of the interchain Glu-Lys' attraction can be explained by the two hydrogen bonds (Fig. 4a) in the Asn core region (one from E14 to N15', the other from K19' to N15') which contribute the added stability.

Hydrophobic residues at positions **e** and **g**

Hydrophobic residues at positions **a** and **d** in the hydrophobic core and their contributions to stability have previously been studied in detail. Here we examined Leu packing and

Table 3. Urea denaturation data

Peptide ^a	Heptad position ^b	[Peptide] (μM) ^c	[Urea] _{1/2} (M) ^d	m (kcal mole ⁻¹ M ⁻¹) ^e	$\Delta\Delta G_{\text{u}}^{\text{obs}}$ (kcal mole ⁻¹) ^f
Hybrid 2	—	341	6.9	0.69	—
L7A	g	350	4.8	0.68	1.44
L26A	e	465	5.0	0.68	1.30
E14A	g	450	5.4	0.73	1.06
K19A	e	316	4.8	0.63	1.39
E14A, K19A	g, e	375	5.4	0.86	1.08
K19E	e	460	2.5	0.70	3.06

^a The sequences are given in Fig. 1.

^b The position of the substituted residue within the heptad repeat, *abcdefg*.

^c Amino acid analysis of stock peptide solutions was used to determine the total peptide concentration in solutions analyzed by CD spectroscopy.

^d Denaturant concentration required to achieve a 50% reduction of the folded state or loss of α-helical content as measured by $[\theta]_{222}$.

^e Slope term in the equilibrium $\Delta G_{\text{u}}(\text{urea}) = \Delta G_{\text{u}}^{\text{H}_2\text{O}} - m[\text{urea}]$, where $\Delta G_{\text{u}}^{\text{H}_2\text{O}}$ is the Gibbs free energy change of unfolding in the absence of denaturant (Santoro and Bolen 1988). The $\Delta G_{\text{u}}^{\text{H}_2\text{O}}$ value for hybrid 2 was 9.4 kcal mole⁻¹.

^f $\Delta\Delta G_{\text{u}}^{\text{obs}}$ is the observed change in ΔG_{u} , the free energy change of unfolding, between hybrid 2 and the peptide analog in column 1, and is calculated $\Delta\Delta G_{\text{u}}^{\text{obs}} = ([\text{urea}]_{1/2, \text{hybrid 2}} - [\text{urea}]_{1/2, \text{analog}}) * (m_{\text{hybrid 2}} + m_{\text{analog}}) / 2$ (Serrano and Fersht 1989).

Table 4. Contributions of α -helix propensity, hydrophobic and electrostatic interactions to coiled-coil stability

Peptide comparisons		$\Delta\Delta G_u^{\text{obs}}$ coiled-coil (kcal mole ⁻¹)	Description of stabilizing contribution	$\Delta\Delta G_u^{\text{obs}}$ per substitution (kcal mole ⁻¹)	$\Delta\Delta G_u^{\text{obs}}$ corrected ^d (kcal mole ⁻¹)
More stable coiled-coil	Less stable coiled-coil				
Hybrid 2	L7A	1.44	Hydrophobic contribution of Leu (g) vs. Ala	0.74	0.74
Hybrid 2	L26A	1.38	Hydrophobic contribution of Leu (e) vs. Ala	0.69	0.69
Hybrid 2	E14A	1.06	Electrostatic and H-bond attractions uncorrected for α -helix propensity	0.53	0.75
Hybrid 2	K19A	1.39	Electrostatic and H-bond attractions uncorrected for α -helix propensity	0.70	0.70
Hybrid 2	E14A, K19A	1.08	Electrostatic and H-bond attractions uncorrected for α -helix propensity	0.54	0.76
E14A, K19A	K19A	0.45 ^a	Helix propensity of Ala vs. Glu (g)	0.22	—
E14A, K19A	E14A	0.00	Helix propensity of Ala vs. Lys (e)	0.00	—
K19A	K19E	1.53 ^b	Removal of electrostatic repulsions uncorrected for α -helix propensity	0.77	0.55
Hybrid 2	K19E	3.06 ^c	Removal of electrostatic repulsions and replacement with electrostatic and H-bond attractions	—	—

^a The α -helix propensity effect of E14 to A14 is calculated from the equation $\Delta\Delta G_u^{\text{obs}} = ([\text{urea}]_{1/2}(\text{E14A, K19A}) - [\text{urea}]_{1/2}(\text{K19A})) * (m_{\text{E14A, K19A}} + m_{\text{K19A}}) / 2$ (from Serrano and Fersht 1989).

^b The stabilizing effects of removing two electrostatic repulsions is calculated from $\Delta\Delta G_u^{\text{obs}} = ([\text{urea}]_{1/2}(\text{K19A}) - [\text{urea}]_{1/2}(\text{K19E})) * (m_{\text{K19A}} + m_{\text{K19E}}) / 2$ (from Serrano and Fersht 1989).

^c Observed values of $\Delta\Delta G_u^{\text{obs, coiled coil}} = 3.06$ kcal mole⁻¹ can be compared with the calculated corrected values from the loss of two electrostatic repulsions $2 * (0.55)$, plus the gain of the average value for two electrostatic attractions and H-bonding $2 * (0.73)$, = 2.56 kcal mole⁻¹.

^d $\Delta\Delta G_u^{\text{obs}}$ corrected is the $\Delta\Delta G_u^{\text{obs}}$ per substitution corrected for α -helix propensity effects. The α -helix propensity of Ala compared to Glu is 0.22 kcal mole⁻¹, which means that the stabilizing effect of an electrostatic and H-bond attraction is the $\Delta\Delta G_u^{\text{obs}}$ value of 0.54 kcal mole⁻¹ plus 0.22 kcal mole⁻¹ to give 0.76 kcal mole⁻¹.

stability at positions **e** and **g**, and found similar stability values in both positions. The side-chain packing of Leu (Fig. 5) in both cases is adjacent to another Leu, either **g-a'** (Leu 8) or **d-e'** (Leu 25). Our values of 0.69 and 0.74 kcal mole⁻¹ for leucine in positions **e** and **g** are less than half the stability contribution at the **a** and **d** positions (1.75 and 1.90 kcal mole⁻¹; Wagschal et al. 1999a; Tripet et al. 2000). The stability differences between Leu at **a** or **d** versus **e** or **g** positions match the differences in solvent-accessible surface area: after calculating accessible surface area in the Hybrid 2 structure using GETAREA 1.1 (Fraczkiewicz and Braun 1998), we observed that Leu 26 and Leu 7 at positions **e** and **g** were 48% exposed to solvent, whereas Leu 8 and Leu 25 at positions **a** and **d** were only 7% exposed at position **d**, and 9% ('in' conformation) or 20% ('out' conformation) exposed at position **a**. Thus, the more that leucine was shielded from solvent (in positions **a** or **d**), the more it contributed to stability via the hydrophobic effect.

How does side-chain packing affect the stability of hydrophobes in positions **e** and **g**? Figure 5 shows the **g-a** packing of Leu 7 and Leu 8, and the **d-e'** packing of Leu 25 and Leu 26. With two conformations of Leu 8 in each helix of Hybrid 2, two possible combinations for the coiled-coil are shown in Figure 5, A and B: 'out/in' and 'in/out'. These conformations would satisfy both the packing of the hydrophobic core as well as interactions with the neighboring Leu 7 at position **g** and could cause the coiled-coil radius to increase towards the N-terminal as observed. The neighboring Leu 7 (**g**) may stabilize Leu 8 (**a**) in the 'out' conformation, as none of the other hydrophobic residues in **a** or **d** positions (Val or Leu) show more than one rotamer confor-

mation in the X-ray structure. For example, Leu 25 (**d**) and Leu 26 (**e**) adopt a stable packing conformation, where the side chains at position **d** are densely packed in the hydrophobic core and also make close contact with the side chains at position **e** (Fig. 5, **e**).

If a hydrophobic residue such as leucine at positions **e** and **g** contributes the same amount of stability to a two-stranded coiled-coil as an ion pair, should we expect to see the same statistical occurrence of hydrophobic residues as charged amino acids in these positions? The database of two- and three-stranded coiled-coil proteins from GenBank showed that at positions **e** and **g**, charged residues were heavily favored over hydrophobic residues by over a factor of four: 51.7% of residues at positions **e** and **g** were charged, whereas only 11.8% were hydrophobic (Lupas et al. 1991). What can explain the preference for charged residues over hydrophobes at these positions? First, having hydrophobes at positions **e** and **g** reduces protein solubility in an aqueous environment. Second, hydrophobes at positions **e** and **g** have been shown to affect oligomerization state, allowing higher orders of self-association (trimers and tetramers; Harbury et al. 1993; Potekhin et al. 1994). So, if one decides to incorporate a relatively small number of hydrophobes in the **e** and **g** positions of de novo designed coiled-coils, these residues can increase overall stability but possibly at the cost of reducing solubility and decreasing specificity for the dimeric state. Nevertheless, despite the relatively low occurrence of leucine at positions **e** and **g** in two-stranded coiled-coils, it would not be possible to predict overall protein stability or the variations of stability along the sequence of coiled-coils without understanding

the quantitative contributions of the leucine residues at positions **e** and **g**.

Conclusions

The crystal structure of the GCN4/cortexillin I peptide (Hybrid 2) contains key details helping to explain its stability. In addition to the hydrophobic residues found in the core **a** and **d** positions are the alanine residues in the **f** positions, the leucine residues in the **e** and **g** positions of the hydrophobic interface, the interchain ionic Glu-Lys' salt bridges on both sides of the two central Asn 15 core residues, and the complex hydrogen bonding network in the surrounding region. Although structurally similar to GCN4, Hybrid 2 possesses more ionic and hydrogen bonds in the vicinity of the central asparagines (especially at the Asn in the 'out' conformation), and has a larger coiled-coil radius near the N-terminal due to the alternate packing conformers of leucine in an **a** position. Biophysical studies have shown the importance of Glu 14 and Lys 19 in establishing **g-e'** ionic interactions, as well as the contributions of Leu 26 and Leu 7 at positions **e** and **g** to coiled-coil stability. To our knowledge, this work is the first reported biophysical quantification of the stability contributions of **d-e'** or **g-a'** leucine packing in a two-stranded coiled-coil, as well as the first identification of a unique network of hydrogen bonds involved in the ion pairs from **g-e'** and the Asn residues in the hydrophobic core due to the reversed ion pairing compared to GCN4-p1. The results provide further information to assist in the de novo design of coiled-coils (Burkhard et al. 2000a, 2002) and understanding the folding and stability of coiled-coils, in ways that do not necessarily require the inclusion of a consensus trigger sequence.

Materials and methods

Peptide synthesis, purification, and analysis

Peptides were synthesized by conventional Boc solid-phase peptide synthesis methodology using methylbenzhydrylamine resin, as described (Litowski et al. 1999). N^{tert}-butyloxycarbonyl amino acid side-chain protecting groups were Asn(β -xanthyl), Asp(β -cyclohexyl), Glu(O-benzyl), Ser(O-benzyl), and Lys(N ϵ -2-Cl-benzyloxycarbonyl). After completion of synthesis, resin was neutralized with 10:90 (v/v) diisopropylethylamine: N, N-dimethylformamide, and N-terminal amino groups were acetylated with 25:75 (v/v) acetic anhydride: dichloromethane. The peptides were cleaved from the resin using anhydrous liquid HF (20 mL/g resin) containing 10% (v/v) anisole and 1% (v/v) 1,2-ethanedithiol for 1 h at -4°C, extracted with 0.05% trifluoroacetic acid (v/v) in 50% aqueous acetonitrile, and purified by reversed-phase HPLC (Wagschal et al. 1999a). Peptide purity was verified by mass spectrometry, analytical reversed-phase HPLC, and amino acid analysis (Litowski et al. 1999).

Circular dichroism spectroscopy and equilibrium unfolding measurements

CD measurements were obtained on a JASCO J-810 Spectropolarimeter at 20°C using Spectra Manager software, Version

1.10.00 running on a Pentium III under Microsoft Windows 2000. Data were collected at 0.1-nm intervals at 20°C from 190 to 250 nm for wavelength scans, with the average of five scans reported. The CD buffer was 50 mM potassium phosphate, 100 mM potassium chloride, pH 7. Ellipticity was reported as mean residue molar ellipticity ($[\theta]$) in deg \cdot cm² \cdot dmole⁻¹ using the following equation:

$$[\theta] = \theta(\text{MRW})/10lc$$

where θ is the observed ellipticity in degrees, MRW is the mean residue weight (molecular weight divided by number of residues), l is the optical path length of the cell in centimeters, and c is the peptide concentration in milligrams per milliliter.

Denaturation data were obtained by monitoring the ellipticity at 220 nm in 0.02 cm cells. Aliquots of ~10 mg/mL peptide stock solution were diluted with the appropriate amount of CD buffer and 10 M urea solution in CD buffer to give final urea concentrations ranging from 0 to 8 M and ~400 μ M peptide concentration. For each peptide, ellipticity values were converted to fraction folded values, using the equation

$$f_f = ([\theta]_o - [\theta]_u) / ([\theta]_n - [\theta]_u)$$

where f_f is the fraction of the population folded, $[\theta]_o$ is the observed molar ellipticity at a given urea concentration, $[\theta]_u$ is the peptide molar ellipticity in the fully unfolded state, and $[\theta]_n$ is the peptide molar ellipticity in the native (fully folded) state. For all peptides, the $[\theta]_n$ value was assumed to be the same as $[\theta]_o$ at 0 M, and the $[\theta]_u$ value was taken to be -2400 deg cm² dmole⁻¹ at 8 M, obtained from peptide K19E, which appeared to approach a minimal $[\theta]_u$ value at 8 M urea. Because peptide K19E was not fully folded at ~400 μ M peptide concentration in the absence of urea (Fig. 4), a value of -31,500 deg cm² dmole⁻¹ (taken from Hybrid 2 in benign, Table 2) was used for $[\theta]_n$ in the calculation of the K19E fraction folded, to give a f_f value less than 1.0 at 0 M urea (0.87, Fig. 9).

Thermodynamic analysis

For each urea denaturation curve, the midpoint of the unfolding curve ($[\text{urea}]_{1/2}$), the slope of the linear portion of the transition (m), and the stability contribution of Hybrid 2 relative to the analog ($\Delta\Delta G_u^{\text{obs}}$) were determined as described (Santoro and Bolen 1988; Serrano and Fersht 1989).

Analytical ultracentrifugation

Sedimentation equilibrium experiments were performed as described (Wagschal et al. 1999a), except that the concentrations used were 600, 150, and 50 μ M, and rotor speeds were 42,000, 46,000, and 50,000 rpm.

Crystallization

Crystals of the peptide were grown in 24-well Falcon plates by vapor diffusion using the hanging-drop method (McPherson 1982). The 1-mL well solution contained 0.1–0.2 M ammonium acetate, 30% PEG₄₀₀₀, and 0.1 M sodium acetate buffer at pH 4.6. The 4- μ L drop contained 2 μ L peptide at a concentration of 20 mg/mL and 2 μ L of well solution. Crystals grew within 2 d to dimensions up to 0.3 \times 0.3 \times 0.5 mm³. The crystals belong to the space group

C222₁ with unit cell dimensions $a = 18.098 \text{ \AA}$, $b = 117.287 \text{ \AA}$, and $c = 22.209 \text{ \AA}$ and contain one monomer in the asymmetric unit (Table 1).

Data collection and processing

The X-ray diffraction data set from the peptide crystals was collected at 100 K on the X11 beam-line (EMBL, DESY Hamburg) at a wavelength of $\lambda = 0.8033 \text{ \AA}$. The data set was processed and scaled using DENZO and SCALEPACK (Otwinowski and Minor 1997), respectively.

Model building and refinement

The structure was determined by molecular replacement methods using the program AMoRe (Navaza 1994). A polyalanine model of a 30-residue-long coiled-coil fragment based on the structure of the GCN4 coiled-coil (RSCB accession code 2zta) was used as a search model. The data set of the peptide was refined with the program SHELXL (Sheldrick and Schneider 1996) including anisotropic B-factor refinement to an R-factor of 17.9% and R-free calculated with 10% of the data set aside prior to refinement of 21.7%, at a resolution of 1.17 \AA . R.m.s. deviations from ideality in bond lengths and angles are 0.009 \AA and 1.8°, respectively (Table 1). The final model comprises one monomer, comprised of residues 3–31, and 32 water molecules in the asymmetric unit. The first two residues at the N-terminus appeared to be disordered in the crystal structure and displayed no interpretable electron density. The side chains of residues Leu 8, Ser 10, and Asn 15 were modeled in two alternative conformations. The overall quality of the model was good as judged by the low R values, the low deviations from stereochemical ideality, and the perfect appearance of the Ramachandran diagram with all residues in the most favored α -helical regions according to the program PROCHECK (Laskowsky et al. 1993). Graphical representations were performed using Insight II (Accelrys).

Calculation of solvent-accessible surface area

Models for L7A and L26A were built by modifying the PDB file for the Hybrid 2 peptide to substitute alanine residues for leucines. Accessible surface areas were calculated using GETAREA 1.1 (www.scsb.utmb.edu/getarea; Fraczkiwicz and Braun 1998).

Accession number

Coordinates have been deposited with the Research Collaboratory for Structural Bioinformatics under the accession code 1P9I.

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