
The HoxB1 hexapeptide is a prefolded domain: Implications for the Pbx1/Hox interaction

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Abstract

Hox proteins are transcriptional regulators that bind consensus DNA sequences. The DNA-binding specificity of many of these Hox proteins is modulated by the heterodimerization with partners, such as the Pbx proteins. This cooperative heterodimerization is accomplished through a conserved hexapeptide motif found N-terminal to the Hox DNA-binding homeodomain. Several human leukemias have been associated with a chromosomal translocation involving either the Hox gene (i.e., NUP98/HOXA9) or the gene encoding Pbx1 (E2A/PBX1). The transforming ability of these fusion oncoproteins relies at least partially on the ability to interact with one another through this hexapeptide motif. Herein we describe NMR structural calculations of the hexapeptide of HoxB1 (N α -acetyl-Thr-Phe-Asp-Trp-Met-Lys-amide) that has been shown to mediate binding between HoxB1 and Pbx1 and a hexapeptide consensus sequence (N α -acetyl-Leu-Phe-Pro-Trp-Met-Arg-amide). The consensus peptide exists in two conformations caused by *cis-trans* isomerization of the Phe-Pro peptide bond. The structures of the HoxB1 peptide and the *trans* form of the consensus peptide reveal a turn very similar to that found as part of the HoxB1/Pbx1/DNA complex in the X-ray crystal structure. This observation implies that this region is at least partially 'preformed' and thus ready to interact with Pbx1 and stabilize binding of Pbx1 and HoxB1 to DNA. The structural results presented here provide a starting point for synthesizing potential nonpeptide or cyclical peptide antagonists that mimic the interaction of these transcriptional cofactors resulting in a potential chemotherapeutic for certain types of leukemias.

Keywords: Hox; Pbx; DNA; transcription; inhibitor; NMR; peptide

The Homeobox genes are a family of developmental regulatory genes that encode nuclear homeoproteins that act as transcription factors (Gehring et al. 1994). These proteins contain a highly conserved common 60-63-residue DNA-binding homeodomain (HD) that is capable of binding DNA

as a monomer (Laughon 1991). Although the homeobox genes were initially described as crucial to the correct anteroposterior patterning of the embryo in vertebrates, *Drosophila*, and *Caenorhabditis elegans* (for review, see McGinnis and Krumlauf 1992), the homeobox has subsequently been identified in a wide range of ~100 mammalian proteins (Stein et al. 1996).

The class-I homeobox genes are defined by homology with the *Drosophila Antennapedia* (Antp) homeodomain (Akam 1987). The 40 mammalian class-I genes (Hox genes) have attracted the most attention because of their clustered organization on four chromosomes (Scott et al. 1989). The Hox proteins are very specific regulators of transcription, and yet, as monomers *in vitro*, they exhibit similar DNA-

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Abbreviations: NMR, nuclear magnetic resonance; DSS, 2,2-dimethyl-2-silapentane-5-sulfonate; TOCSY, total correlation spectroscopy; DQF-COSY, double-quantum filtered correlation spectroscopy; ROESY, rotating frame Overhauser effect spectroscopy.

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binding specificities (Laughon 1991). However, although the monomeric Hox proteins show little discrimination among different DNA sequences, their specificity and DNA-binding affinity are greatly enhanced through cooperative DNA binding with Hox cofactors (for review, see Mann and Chan 1996). The identification of extradenticle (Exd) in *Drosophila* as well as the vertebrate ortholog Pbx1 showed how a hox/cofactor heterodimer could achieve precise transcriptional regulation by enhanced DNA-sequence specificity (Chan et al. 1994; van Dijk and Murre 1994).

Pbx1 (as well as Pbx2 and Pbx3) and Exd are members of the TALE (three-amino-acid loop extension) family of atypical homeodomain proteins, whose members are characterized by a three-residue insertion in the first helix of the homeodomain involved in their interaction with Hox proteins (Bertolino et al. 1995). Examination of Pbx1 has shown that, in addition to the homeodomain, a short 16-residue C-terminal tail (conserved in Exd) is essential for maximal cooperative interactions with Hox partners as well as for maximal monomeric binding of Pbx1 to DNA (Lu and Kamps 1996). Cooperative binding of Hox proteins of paralog groups 1–10 is also dependent on a conserved hexapeptide sequence (consensus Y/F-P-W-M-K/R) located N-terminal to the Hox homeodomain (Knoepfler and Kamps 1995). This hexapeptide is separated from the N terminus of the homeodomain by a linker that varies in length and sequence among the different Hox proteins. Disruption of this hexapeptide by deletion or by mutation of one or more of its residues abrogates this in vitro cooperative binding between Pbx1 and Hox (Knoepfler and Kamps 1995).

The importance of regulated Hox gene expression was first recognized in the study of homeotic mutations in *Drosophila* and mice. In humans, the importance of Hox genes and their cofactors is well exemplified in the study of hematopoiesis in which one sees an ordered (3'–5' in the A, B, and C gene clusters) expression of Hox during maturation of the hematopoietic stem cell to the fully differentiated blood cells (Sauvageau et al. 1994). Furthermore, the examination of human leukemias has revealed both a general pattern of Hox gene dysregulation as well as specific recurring translocations involving Hox genes and their cofactors. For example, the t(10;14) translocation leads to the overexpression of Hox11 accompanied by T-cell acute lymphocytic leukemia (ALL; Hatano et al. 1991), whereas the t(7;11) translocation joins the Nucleoporin98 gene with HoxA9, resulting in acute myeloid leukemia (Borrow et al. 1996; Nakamura et al. 1996). In addition, the Hox cofactor Pbx1 was originally identified as the C-terminal portion of the fusion oncogene created in the t(1;19) translocation found in 25% of pediatric pre-B cell ALL (Kamps et al. 1990; Nourse et al. 1990). The t(1;19) results in the fusion of the transcriptional activation domains of E2a with the majority of Pbx1 including its DNA-binding homeodomain. Presumably, the transforming ability of the resultant

E2aPbx1 fusion oncoprotein is dependent on its ability to interact with Hox proteins.

Recently, the structure of the human HoxB1–Pbx1 heterodimer bound to DNA was elucidated by X-ray crystallography (Piper et al. 1999). The structure was comprised of the minimal fragments necessary for cooperative DNA binding: The HoxB1 homeodomain + N-terminal hexapeptide and the Pbx1 homeodomain + C-terminal helix. The structure showed that HoxB1 and Pbx1 bind to overlapping sites on opposite faces of the DNA helix. All contacts with Pbx1 are mediated through the conserved Hox hexapeptide that contacts Pbx1 in a pocket located between the three-residue insertion and the third helix of the Pbx1 homeodomain. In contrast, the C-terminal residues of Pbx1 are not involved in binding either DNA or HoxB1. The linker region between the HoxB1 homeodomain and the hexapeptide (not conserved within the Hox family of proteins) is unstructured, suggesting that it is not involved in establishing the contact between the Hox hexapeptide and the Pbx1 homeodomain. The structure of the homologous *Drosophila* Ubx–Exd heterodimer on a similar DNA sequence has also been solved by X-ray crystallography (Passner et al. 1999) and shows very similar features. In particular, contact between the two proteins is mediated entirely by the Ubx hexapeptide region that adopts an essentially identical structure to that of HoxB1 and binds to a hydrophobic pocket in the Exd homeodomain (Wilson and Desplan 1999). Previous in vitro studies have shown that disruption of the cooperative DNA binding by Pbx1 and Hox proteins can be accomplished by simple mutations in the hexapeptide domain as well as by the inclusion, in the binding reaction of a high concentration of a synthetic ~15-residue peptide containing the hexapeptide domain (Knoepfler and Kamps 1995). The structure of the Pbx1 homeodomain, both free in solution and complexed to DNA, has also been determined by NMR spectroscopy. A model for the interaction was proposed that suggested the formation of the additional helix of Pbx1 involved in the interaction with HoxB1 protein was triggered by the interaction of Pbx1 with the DNA oligonucleotide (Jabet et al. 1999).

Herein we investigate the question of whether the HoxB1 hexapeptide Thr-Phe-Asp-Trp-Met-Lys or the consensus sequence hexapeptide Leu-Phe-Pro-Trp-Met-Arg are self-assembling domains, capable of independently folding into the same structure as the wild-type protein in the absence of Pbx1, DNA, or any other portion of HoxB1. NMR studies over the last decade have focussed on the observation of nascent structures in peptides in solution as well as their relevance to function (Dyson et al. 1988). Very recent NMR studies (Forman-Kay 1999; Cavanagh and Akke 2000) have highlighted the importance of backbone and side-chain dynamics in the formation of protein–protein complexes and have focussed on the understanding of the entropic contributions in the interface of the protein–protein interac-

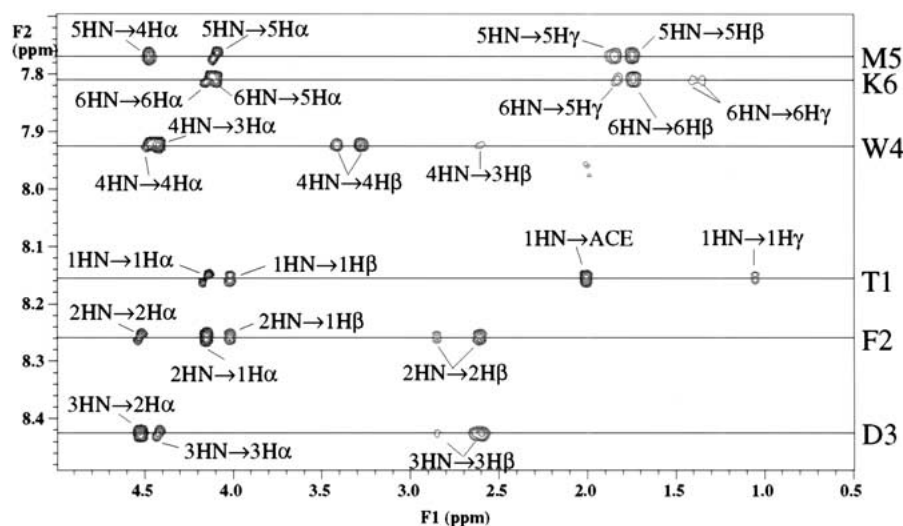


Fig. 1. NH-aliphatic region of the ^1H 2D NMR ROESY spectrum of Ac(TFDWMK) NH_2 at 600 MHz taken at 5°C . with a 150 ms mixing time.

tion to the stability of the complex. We show that both of these small peptides are capable of folding into stable turn structures that are equivalent to the structure of this region in the HoxB1 protein bound to Pbx1. These results are important for understanding the mechanism and energetics of the Hox/Pbx1 interaction. In addition, these studies are critical for the potential design and synthesis of small, specific, stable nonpeptide or cyclical peptide inhibitors that might be important in the treatment of specific types of leukemias.

Results

In the HoxB1–Pbx1–DNA complex (Piper et al. 1999) and similarly in the Ubx–Exd–DNA structure (Passner et al. 1999), contacts between the HoxB1 homeodomain (or Ubx) and Pbx1 (or Exd) are mediated entirely by the hexapeptide

(TFDWMK). This peptide, corresponding to residues –24 to –19 of the HoxB1 homeodomain, was synthesized for solution structural studies with the addition of acetyl and amide groups to remove effects of artificial charges at the N- and C termini, which would not be present in the intact protein. The hexapeptide region has a core pentapeptide motif that is highly conserved among Hox proteins (F/Y-P-W-M-R/K; Knoepfler and Kamps 1995). Thus, the hexapeptide Ac(LFPWMR) NH_2 representing the consensus sequence was also studied. The consensus peptide exists in two conformations caused by *cis*–*trans* isomerization of the Phe–Pro peptide bond.

^1H 1D and ^1H – ^1H 2D NMR spectra were obtained at 300 MHz, 500 MHz, and 600 MHz for both peptides. Spectra were collected at 5°C to maximize nascent or existing structure in the peptide. The amide region of the 2D ROESY NMR spectrum for Ac(TFDWMK) NH_2 is shown in Fig-

Table 1. Chemical shifts^a and coupling constants for Ac(TFDWMK) NH_2

Residue	HN	H α	H β	Others	$^3J_{\text{HNH}\alpha}$
Acetyl				CH ₃ 2.02	
Thr-1	8.15	4.15	4.02	H γ 1 8.03, H γ 2* 1.06	7.31
Phe-2	8.26	4.53	2.86	H δ * 7.11, H ϵ * 7.30, H ζ 7.30	7.49
			2.61		
Asp-3	8.43	4.43	2.64		5.35
			2.60		
Trp-4	7.93	4.48	3.42	H δ 1 7.40, H ϵ 1 10.27, H ϵ 3 7.56	4.82
			3.28	H ζ 2 7.41, H ζ 3 7.17, H η 2 7.21	
Met-5	7.77	4.11	1.75	H γ * 1.87, 1.83, H ϵ * 2.00	6.80
Lys-6	7.81	4.14	1.75	H γ * 1.41, 1.35, H δ * 1.64, H ϵ * 2.93	6.93
				H ζ * 7.60	
Amide	7.45				

^a Residue chemical shifts were recorded at 5°C and corrected to an internal shift standard of DSS.

Table 2. Chemical shifts^a and coupling constants for *trans* Ac(LFPWMR)NH₂

Residue	HN	H α	H β	Others	³ J _{HNHα}
Acetyl				CH ₃ 1.91	
Leu-1	8.14	4.07	1.18	H γ 1.38	6.65
			1.31	H δ * 0.81, 0.75	
Phe-2	7.81	4.63	2.30	H δ * 7.19	8.22
			2.15	H ϵ * 7.38	
Pro-3	—	4.25	2.30	H γ * 2.09	—
			2.02	H δ * 3.73, 3.59	
Trp-4	7.36	4.65	3.43	H δ 1 7.29, H ϵ 1 10.37, H ϵ 3 7.61	
			3.32	H ζ 2 7.18, H ζ 3 7.22, H η 2 7.30	
Met-5	7.68	4.34	2.06	H γ * 2.17, 1.98	7.61
			1.79	H ϵ * 2.08	
Arg-6	7.95	4.20	1.86	H γ * 1.59, H δ * 3.17, H ϵ 7.22	7.19
			1.75	HH11 6.90, HH21 6.47	
Amide	7.59				

^a Residue chemical shifts were recorded at 5°C and corrected to an internal shift standard of DSS.

ure 1. Complete NMR resonance assignments of both peptides were obtained using a combination of DQF-COSY, TOCSY and ROESY experiments and standard NMR sequence assignment methods (Wüthrich 1986). ³J_{HNH α} coupling constants were measured from high resolution 1-D NMR spectra obtained at 600 MHz for Ac(TFDWMK)NH₂ and 500 MHz for Ac(LFPWMR)NH₂. The Ac(LFPWMR)NH₂ peptide exhibits *cis-trans* isomerism about the proline residue leading to ~10% of the *cis* form of the peptide. The presence of the bulky tryptophan residue following the proline should result in a relatively higher proportion than normal of the *cis* conformation of the peptide (Nardi et al. 2000). The chemical shift assignments and coupling constants for these peptides are listed in Table 1, Table 2, and Table 3.

For small peptides, such as Ac(TFDWMK)NH₂ and Ac(LFPWMR)NH₂, distance restraints were obtained from

cross-peak intensities in 2D ROESY NMR spectra obtained at 600 MHz and 300 MHz, both with mixing times of 150 ms. For the Ac(TFDWMK)NH₂ peptide, strong d _{α N} connectivities were found between residues 1 and 2, residues 2 and 3, residues 3 and 4, residues 4 and 5, as well as residues 5 and 6 (Fig. 1). Relatively weak d_{NN} connectivities were found between residues 2 and 3, as well as residues 3 and 4 with a stronger d_{NN} ROE between residues 4 and 5. For the Ac(LFPWMR)NH₂ peptide in the *trans* form, strong d _{α N} connectivities were found between residues 1 and 2, residues 3 and 4, residues 4 and 5, as well as residues 5 and 6. Relatively weak d_{NN} connectivities were found between residues 4 and 5, and residues 5 and 6. For the Ac(LFPWMR)NH₂ peptide in the *cis* form, strong d _{α N} connectivities were found between residues 1 and 2, residues 3 and 4, residues 4 and 5, as well as residues 5 and 6. Relatively weak d_{NN} connectivities were found between residues 1 and

Table 3. Chemical shifts^a and coupling constants for *cis* Ac(LFPWMR)NH₂

Residue	HN	H α	H β	Others	³ J _{HNHα}
Acetyl				CH ₃ 2.00	
Leu-1	8.23	4.37	1.54	H γ 1.63	7.59
				H δ * 0.92, 0.88	
Phe-2	8.30	4.33	2.96	H δ * 7.14	6.48
			2.91	H ϵ * 7.31	
Pro-3	—	3.57	1.72	H γ * 1.58, 1.29	—
			1.47	H δ * 3.63, 3.55	
Trp-4	8.41	4.53	3.27	H δ 1 7.24, H ϵ 1 10.21, H ϵ 3 7.59	6.44
				H ζ 2 7.63, H ζ 3 7.18, H η 2 7.26	
Met-5	8.13	4.34	1.95	H γ * 2.41, 2.34	—
			1.84	H ϵ * 2.03	
Arg-6	8.19	4.10	1.78	H γ * 1.57, H δ * 3.24	6.85
			1.67		
Amide	7.65				

^a Residue chemical shifts were recorded at 5°C and corrected to an internal shift standard of DSS.

2 and between residues 4 and 5. A complete summary of the NMR ROE information is shown in Figure 2. Experimental distance restraints were calibrated using the program NMRView (Johnson and Blevins 1994) with some in-house modifications. Torsion angle ϕ restraints were obtained by converting measured $^3J_{\text{HNH}\alpha}$ NMR coupling constants to ϕ angles and adding a range of $\pm 10^\circ$. Only the Trp and Met residues in Ac(TFDWMK)NH₂ had coupling constants <6.0 Hz, which is indicative of ϕ values that are significantly different from values typical of random coil or extended structure. In the *trans* form of the Ac(LFPWMR)NH₂ peptide, only F2 had a coupling constant >8 Hz, indicative of a

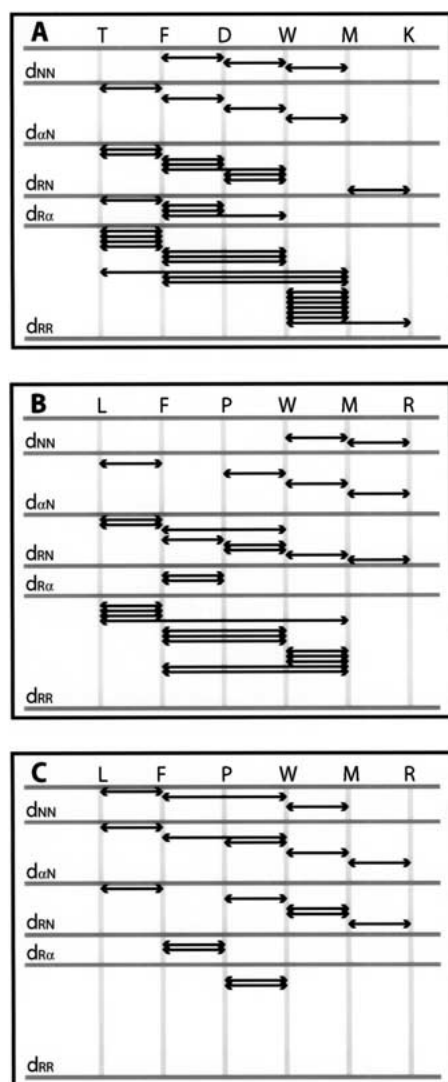


Fig. 2. Summary of ROE interactions defining the preferred conformations of (A) Ac(TFDWMK)NH₂, (B) *trans*-Ac(LFPWMR)NH₂, and (C) *cis*-Ac(LFPWMR)NH₂. Shown are the d_{NN} , $d_{\alpha\text{N}}$, d_{RN} , $d_{\text{R}\alpha}$, and d_{RR} connectivities. d_{NN} refers to the sequential NH–NH ROE, $d_{\alpha\text{N}}$ refers to the sequential H α –NH ROE, d_{RN} refers to any side chain–NH ROE, $d_{\text{R}\alpha}$ refers to any side chain–H α ROE, and d_{RR} refers to any side chain–side chain ROE.

ϕ value different from random coil. The coupling constant for Trp4 in the *trans* form of the Ac(LFPWMR)NH₂ peptide could not be measured because of overlap with the side-chain resonances of Phe2. In total, for the Ac(TFDWMK)NH₂ peptide there were 36 intraresidue, 31 sequential ($|i - j| = 1$) and nine short-range ($1 < |i - j| \leq 5$) distance restraints as well as two dihedral ϕ restraints used in the structural calculations. For the *trans* form of the Ac(LFPWMR)NH₂ peptide, there were 52 intraresidue, 22 sequential, and eight short-range distance restraints in addition to one dihedral ϕ restraint used for the structural calculations. For the *cis* form of Ac(LFPWMR)NH₂, there were 30 intraresidue, 19 sequential, and two short-range restraints.

Ensembles of 60 structures were computed from the experimental restraints derived from the NMR data as described in the Materials and Methods section. For Ac(TFDWMK)NH₂, 23 were accepted with no ROE violations >0.2 Å, no dihedral violations $>5^\circ$, and low E_{TOT} (<15 kcal/mole). The structural statistics are shown in Table 4. For the *trans* form of Ac(LFPWMR)NH₂, 17 structures were accepted with no ROE violations >0.2 Å, no dihedral violations $>5^\circ$, and low E_{TOT} (<17 kcal/mole). For the *cis* form of Ac(LFPWMR)NH₂, 18 structures were accepted with no ROE violations >0.2 Å; and low E_{TOT} (<13 kcal/mole). The ensembles of NMR structures superimposed on the representative structures are shown for all three structures in Figure 3A–C. The backbone RMSDs within the families of structures for residues 1–5 are given in Table 4.

The preferred conformations of Ac(TFDWMK)NH₂ and the *trans* form of Ac(LFPWMR)NH₂ resemble a type-I turn (Fig. 3A,B). Type-I turns, or $\alpha\alpha$ in the nomenclature of Wilmot and Thornton (1990), are characterized by (ϕ, ψ) angles of $(-60^\circ, -30^\circ)$ in the $i + 1$ position followed by $(-90^\circ, 0^\circ)$ in the $i + 2$ position of the turn. These angles can deviate as much as 30° in the actual structure (Wilmot and Thornton 1990). Table 5 and Table 6 illustrate the (ϕ, ψ) angles of the 6-mer peptides in solution. In the Ac(TFDWMK)NH₂ peptide, residue 3 has (ϕ, ψ) angles of $(-81^\circ, -32^\circ)$ and residue 4 has (ϕ, ψ) angles of $(-71^\circ, -20^\circ)$. A backbone H-bond is possible in the structure involving the carbonyl oxygen of residue 2 and the amide hydrogen of residue 5 (distance is ~ 2.9 Å in the average NMR structure and ~ 2.4 Å in the crystal structure). In addition, another hydrogen is possible involving the backbone carbonyl oxygen of residue 3 with the backbone amide hydrogen of residue 6 (~ 2.5 Å in the average NMR structure and ~ 1.9 Å in the crystal structure). For the *trans* form of Ac(LFPWMR)NH₂, residue 3 has (ϕ, ψ) angles of $(-80^\circ, -15^\circ)$ and residue 4 has (ϕ, ψ) angles of $(-101^\circ, 2^\circ)$. A backbone H-bond is possible in the structure involving the carbonyl oxygen of residue 3 and the amide hydrogen of residue 6. For the *cis* form of Ac(LFPWMR)NH₂, the (ϕ, ψ) angles are completely different from the *trans* form of

Table 4. Structural statistics and atomic r.m.s. differences for 23 calculated Ac(TFDWMK)NH₂ structures, 17 calculated *trans* Ac(LFPWMR)NH₂ structures, and 18 calculated *cis* Ac(LFPWMR)NH₂ structures

	Ac(TFDWMK)NH ₂	Trans-Ac(LFPWMR)NH ₂	Cis-Ac(LFPWMR)NH ₂
E _{TOT} (kcal/mol) ^a	11.36 ± 2.09	14.04 ± 1.32	12.18 ± 0.44
E _{NOE} (kcal/mol) ^a	1.44 ± 0.38	0.41 ± 0.13	0.27 ± 0.12
E _{VDW} (kcal/mol) ^a	0.42 ± 0.74	0.41 ± 0.56	0.05 ± 0.16
Deviations from Ideal Geometry ^b			
Bonds (Å)	0.0036 ± 0.0002	0.0039 ± 0.0002	0.0035 ± 0.0002
Angles (°)	0.4758 ± 0.0407	0.5360 ± 0.0216	0.5098 ± 0.0077
Improper (°)	0.1929 ± 0.0323	0.2929 ± 0.0105	0.2878 ± 0.0112
Atomic r.m.s. differences (Å) ^c			
Backbone atoms	0.36 ± 0.06	0.32 ± 0.08	0.55 ± 0.05
Heavy atoms	0.63 ± 0.06	0.54 ± 0.07	1.17 ± 0.16

^a The values for E_{NOE} are calculated from a square well potential with a force constant of 50 kcal · mole⁻¹ · Å². E_{VDW} is calculated with a force constant of 4 kcal · mole⁻¹ · Å⁻⁴ and the final van der Waal's radii were set to 0.75 times the value used in the CHARMM force field.

^b The values for bonds, angles, and impropers show the deviation from ideal values based on perfect stereochemistry.

^c R.m.s. differences of the final simulated annealing structures superimposed on the average structures (residues 1–5).

Ac(LFPWMR)NH₂, the Ac(TFDWMK)NH₂ peptide and the X-ray crystal structure of the TFDWMK complex (Table 6). The *cis* form of the peptide appears to form a structure that may be described at a type-VIII turn, with characteristic (ϕ , ψ) angles of (–60°, –30°) in the *i* + 1 position, and (–120°, 120°) in the *i* + 2 position (Wilmot and Thornton 1990).

Figure 4 depicts a comparison of a member of the major conformation of each ensemble of NMR structures to the X-ray structure (coordinates 1B72 from the Protein Data Bank) of the same region. It is immediately clear that the Ac(TFDWMK)NH₂ peptide and the *trans* form of the Ac(LFPWMR)NH₂ peptide has adopted an overall structure that is very similar to that of the same region in the intact ternary complex (Table 5). The RMSD between the Ac(TFDWMK)NH₂ peptide and the X-ray structure is 0.92 over residues 1–5. Between the *trans* form of the Ac(LFPWMR)NH₂ peptide and the X-ray structure, the RMSD is 0.94. The *cis* form of the Ac(LFPWMR)NH₂ peptide has an RMSD of 1.66 with the X-ray structure.

To further characterize the structuring of the 6-mer peptides, amide proton NMR chemical shift temperature coefficients ($\Delta\delta/\Delta T$) were measured. Exposed NHs typically have gradients in the range of –6.0 ppb/°C to –8.5 ppb/°C, whereas hydrogen-bonded NHs have $\Delta\delta/\Delta T$ of –2.0 ppb/°C ± 1.4 ppb/°C (Andersen et al. 1997). To measure $\Delta\delta/\Delta T$ values for the peptides, 1D spectra were acquired at 5°C, 10°C, 15°C, 20°C, and 25°C. Chemical shift deviations were derived from the lowest temperature set included (5°C). Random coil chemical shifts (Wishart et al. 1995) were corrected to 5°C according to (Mertuka et al. 1995). Table 7 shows the $\Delta\delta/\Delta T$ for each residue in addition to the NH and H α chemical shift deviations for the 6-mer peptides at 5°C and 20°C. For the Ac(TFPWMK)NH₂ peptide, it is possible that residues 5 and 6 are able to form hydrogen

bonds. Indeed, the greatest chemical shift deviations occur for residues M5 and K6 (Table 4), and their $\Delta\delta/\Delta T$ values are within the –2.0 ppb/°C ± 1.4 ppb/°C range for hydrogen-bonded NHs. The hydrogen bond potential of these residues could account for the good stability of the peptide. Interestingly, the *trans* form of the Ac(LFPWMR)NH₂ peptide has a small $\Delta\delta/\Delta T$ value for residue M5, indicating that it could be hydrogen bonded (most likely to F2). The largest chemical shift deviations from random coil values occur for residues W4 and M5. The *cis* form of this peptide does not have significant chemical shift deviations from random coil and has no $\Delta\delta/\Delta T$ values that indicate hydrogen bonds are present to stabilize the structure.

Discussion

The preferred conformation of the hexapeptide Ac(TFDWMK)NH₂, which corresponds to residues –24 to –19 of the HoxB1 homeodomain, was studied to determine if it is a self-assembling domain retaining the structure of the bound form of this part of the intact HoxB1 protein. In addition, Ac(LFPWMR)NH₂, which corresponds to the consensus sequence of the hexapeptide, was studied to determine if it has a similar structure to the hexapeptide Ac(TFDWMK)NH₂ and the intact HoxB1 protein.

NMR spectroscopic analysis of the Ac(TFDWMK)NH₂ peptide reveals the major conformation of the peptide to be a stable folded structure that is strikingly similar to this portion of HoxB1 in the X-ray structure where HoxB1 was complexed with Pbx1 and DNA (Fig. 4). In addition, the *trans* form of the consensus sequence Ac(LFPWMR)NH₂ is also structured and the preferred conformation is similar to the X-ray structure. Similar ϕ , ψ , and χ_1 angles are found between the X-ray structure bound to Pbx1 and DNA and

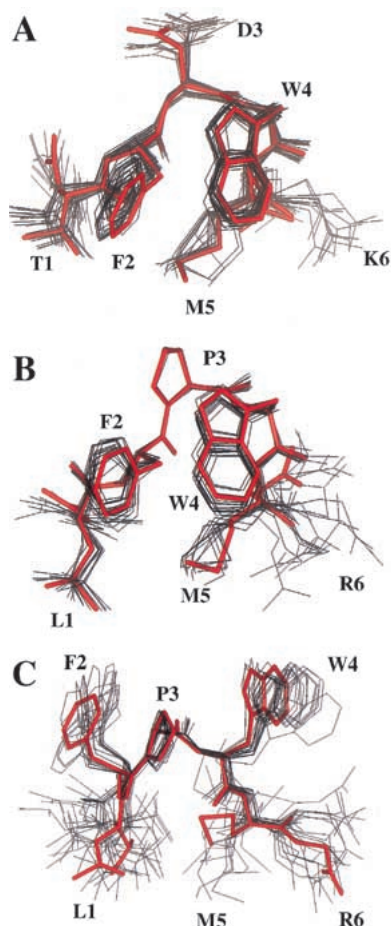


Fig. 3. (A) Superimposition of 23 calculated NMR structures for Ac(TFDWMK)NH₂ (black) with a representative structure (red). (B) Superimposition of 17 calculated NMR structures for the *trans* form of Ac(LFPWMR)NH₂ (black) with a representative structure (red). (C) Superimposition of 18 calculated NMR structures for the *cis* form of Ac(LFPWMR)NH₂ (black) with a representative structure (red). The representative structure is a structure in the ensemble that has the lowest RMSD to its own average structure. Figure was prepared using Insight II (MSI).

the two hexapeptides (Table 5 and Table 6). In addition, similar hydrogen-bonding patterns are found between the three structures. Although the preferred NMR and X-ray structures are not identical (RMSDs between the X-ray structure and the NMR structures are close to 1 Å), this is to be expected because the rest of the Hox1 protein is missing in the NMR-determined structure and the X-ray structure was of the complexed form rather than the free form of this region of the protein.

The *cis* form of the Ac(LFPWMR)NH₂ peptide has a different preferred structure from the *trans* form, the X-ray structure or the hexapeptide Ac(TFDWMK)NH₂ in solution. The presence of an aromatic group C-terminal to the proline appears to enhance the population of the *cis* form of the peptide (Nardi et al. 2000). It is likely, therefore, that the

presence of the proline in the consensus sequence would attenuate the binding ability of this peptide to Pbx1.

The well-defined preferred peptide conformational structures determined here, as evident from the low RMSDs, are mainly caused by the specific hexapeptide sequence. The hydrophobic residues (phenylalanine, tryptophan, and methionine) have a tendency to interact with one another (Fig. 2). In addition, because of the interacting hydrophobic residues, the backbone can orient itself into a turn structure such that hydrogen bonds form to stabilize the backbone conformation. For a type-I turn, it is expected that one would see the ROE $d_{\alpha\text{N}}(i, i + 2)$ between Asp3-Met5 and Trp4-Lys6. These ROEs are expected to be weak (because, in the X-ray structure, these atoms are 3.5 Å–4 Å apart). One of the ROEs (Asp3 H α to Met5 NH) overlaps with another ROE (Trp4 H α to Met5 NH) and thus was not included in the data set. The other ROE is present (Trp4 H α to Lys6 NH), but is weak, and was not chosen to be part of the original data set. Interestingly, the major structures of the 6-mers are dependent mostly on side-chain interactions, rather than strictly backbone ROEs as shown in Figure 2.

The major structures observed here for Ac(TFDWMK)-NH₂ and the *trans* form of Ac(LFPWMR)NH₂ appear to be relatively stable in solution since the chemical shift deviations at 20°C are similar to those at 5°C (Table 7), indicating a high proportion of structure is still present at 20°C. Analysis of the free energy of folding of the ensemble of Ac(TFDWMK)NH₂ peptide structures utilizing the STC program (Lavigne et al. 2000), which bases its calculations on the differences in accessible surface area of polar and nonpolar residues between unfolded and folded states, indicates a ΔG of -0.1 ± 0.3 kcal/mole at 5°C, indicating that $\geq 50\%$ of the peptide is folded at this temperature. Further STC analysis of the ΔG of binding of this peptide to the Pbx1–DNA complex (using the X-ray structure complex as the protein complex structure and the NMR derived structure for the peptide free structure) reveals a ΔG of ~ -5 to -6 kcal/mole, corresponding to a K_D of ~ 100 μM . This binding constant is consistent with the notion that the Pbx1 and HoxB1 proteins independently bind to the DNA after which

Table 5. Comparison of (ϕ, ψ) and χ^1 angles for the Ac(TFDWMK)NH₂ calculated NMR structures¹ with the X-ray structure of the hexapeptide in complex with Pbx1 and DNA

Residue	NMR structures			X-ray structure		
	ϕ	ψ	χ^1	ϕ	ψ	χ^1
T1	—	143 ± 26	47 ± 5	—	99	65
F2	44 ± 54	131 ± 17	-56 ± 4	-76	176	-51
D3	-81 ± 0	-32 ± 7	-146 ± 40	-55	-46	-69
W4	-71 ± 7	-20 ± 14	44 ± 8	-60	-30	64
M5	-68 ± 15	28 ± 74	-65 ± 79	-72	-10	-51
K6	-31 ± 86	—	-50 ± 37	-112	—	—

Table 6. Comparison of (ϕ , ψ) and χ^1 angles for the Ac(LFPWMR)NH₂ peptide

Residue	Trans			Cis		
	ϕ	ψ	χ^1	ϕ	ψ	χ^1
L1	—	-63 ± 11	-75 ± 33	—	77 ± 36	-127 ± 36
F2	-129 ± 13	158 ± 3	-68 ± 9	139 ± 70	143 ± 7	-95 ± 32
P3	-80 ± 1	-15 ± 9	24 ± 0	-80 ± 1	-42 ± 4	24 ± 0
W4	-101 ± 14	2 ± 13.0	43 ± 2	-120 ± 13	165 ± 11	-97 ± 4
M5	-87 ± 20	58 ± 90	-64 ± 42	-138 ± 12	92 ± 43	160 ± 89
R6	-75 ± 59	—	-75 ± 79	-103 ± 50	—	-90 ± 72

the complex is stabilized against dissociation by the binding of the hexapeptide of HoxB1 to Pbx1 (Wilson and Desplan 1999).

A truncated form of the hexapeptide studied here (a pentapeptide consisting of residues 2–6) has been previously shown to inhibit cooperative binding between Hox proteins and Pbx1 while enhancing Pbx1 DNA binding (Knoepfler and Kamps 1995). These results are important in that certain leukemias result from chromosomal translocations involving Hox or Pbx1 genes and that the genesis of these leukemias most likely requires the hexapeptide motif of the Hox protein to stabilize Pbx binding to DNA. The ability to block the interaction between these two proteins should pro-

vide a means of inhibiting transcriptional activation. Therefore, the design and synthesis of analogs that mimic the Hox hexapeptide structure may be potentially chemotherapeutic for these types of cancers by virtue of inhibiting the Hox/Pbx1 interaction.

Materials and methods

The peptides N α -acetyl-Thr-Phe-Asp-Trp-Met-Lys-amide, and N α -acetyl-Leu-Phe-Pro-Trp-Met-Arg-amide were synthesized using solid-phase synthesis methodology by the Alberta Peptide Institute. The correct mass was verified by electrospray mass spectrometry, and the overall purity confirmed by reverse-phase

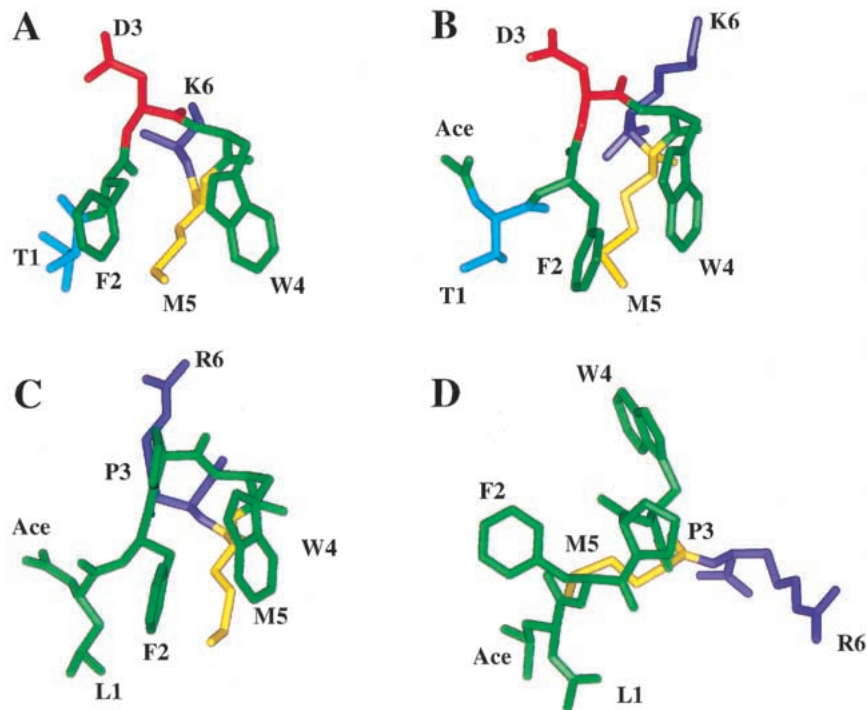


Fig. 4. (A) Crystal structure of the 6-mer peptide from complex with Pbx1 and DNA. (B) A representative structure of the Ac(TFDWMK)NH₂ peptide. (C) A representative structure of the *trans* form of the Ac(LFPWMR)NH₂ peptide. (D) A representative structure of the *cis* form of the Ac(LFPWMR)NH₂ peptide. The representative structure is a structure in the ensemble that has the lowest RMSD to its own average structure. Figure was prepared using Insight II (MSI).

Table 7. NH and H α ^1H NMR chemical shift deviations and NH temperature coefficients

Residue	Ac(TFDWMK)NH ₂			H α CSD
	NH CSD (5°C)	NH CSD (20°C)	$\Delta\delta/\Delta T$	
T1	-0.12	-0.14	-9.0	-0.20
F2	-0.18	-0.19	-9.0	-0.09
D3	-0.03	0.00	-5.6	-0.21
W4	-0.46	-0.46	-7.8	-0.18
M5	-0.66	-0.58	-2.6	-0.37
K6	-0.63	-0.55	-3.1	-0.18
Trans Ac(LFPWMR)NH ₂				
L1	-0.18	-0.16	-8.2	-0.27
F2	-0.46	-0.47	-7.1	-0.27
P3	—	—	—	-0.19
W4	-1.03	-1.05	-7.6	-0.01
M5	-0.74	-0.68	-2.9	-0.14
R6	-0.41	-0.37	-4.4	-0.14
Cis Ac(LFPWMR)NH ₂				
L1	-0.07	-0.06	-6.8	0.03
F2	0.04	0.03	-8.4	-0.57
P3	—	—	—	-0.87
W4	0.02	0	-9.4	-0.13
M5	-0.3	-0.24	-4.5	-0.14
R6	-0.16	-0.16	-6.2	-0.24

NH chemical shift deviations were calculated at the appropriate temperature with random coil chemical shift correction as described in Andersen et al. (1997); Mertuka et al. (1995); and Wishart et al. (1995). H α chemical shift deviations were calculated from peptide data at 5°C and random coil data at 25°C from Wishart et al. (1995).

HPLC. NMR experiments were performed on a 2-mM sample of either Ac(TFDWMK)NH₂ or Ac(LFPWMR)NH₂ dissolved in 100 mM KCl in 90% H₂O/10% D₂O at pH 6.5, containing 0.5 mM DSS as an internal NMR chemical shift reference standard.

All NMR data were recorded at 5°C with either a Varian Unity 600 NMR spectrometer, a Varian INOVA 500 NMR spectrometer, or a Varian Unity 300 NMR spectrometer operating at ^1H resonance frequencies of 600 MHz, 500 MHz, and 300 MHz, respectively. 2D ^1H - ^1H DQF-COSY, TOCSY, and ROESY spectra were collected at 300 MHz, and DQF-COSY and ROESY spectra at 600 MHz were collected for the Ac(TFDWMK)NH₂ and Ac(LFPWMR)NH₂ peptides. In addition, a TOCSY spectrum was acquired at 500 MHz for the Ac(LFPWMR)NH₂ peptide. The ROESY spectra were collected with a mixing time of 150 ms, and the TOCSY spectrum with a mixing time of 80 ms. All NMR data were processed using Vnmr and analyzed using the program nmr-View (Johnson and Blevins 1994).

Three-dimensional structures were computed from experimental restraints using a simulated annealing protocol (Nilges et al. 1988) implementing the SHAKE algorithm with X-PLOR version 3.8 (Brünger 1993). The initial structure was an extended chain, and the target function contained only potential terms for covalent geometry, torsion angle restraints, experimental distance restraints, and a van der Waals repulsion term for nonbonded contacts. Force constants for the NOE-derived distance restraints were set to 50 kcal/mole Å^{-2} , and dihedral angle restraints were initialized at 5 kcal/mole $^{-1}$ rad $^{-2}$ during the high-temperature dynamics and increased to 200 kcal/mole rad $^{-2}$ during the annealing stage. An-

nealing proceeded stepwise from 800 K to 298 K in decrements of 50 K. The final energy-minimization step employed the 6-12 Lennard-Jones potential.

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