

# Environmental Parameters Regulating Gaseous Nitrogen Losses from Two Forested Ecosystems via Nitrification and Denitrification†

ERIC A. DAVIDSON<sup>1‡\*</sup> AND WAYNE T. SWANK<sup>2</sup>

Department of Forestry, North Carolina State University, Raleigh, North Carolina 27695-8002,<sup>1</sup> and U.S. Department of Agriculture Forest Service Southeastern Experiment Station, Coweeta Hydrologic Laboratory, Otto, North Carolina 28763<sup>2</sup>

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**Gaseous N losses from disturbed and reference forested watersheds at the Coweeta Hydrologic Laboratory in western North Carolina were studied by in situ N<sub>2</sub>O diffusion measurements and laboratory incubations throughout a 10-month period. Soil temperature, percent base saturation, and water-filled pore space accounted for 43% of the variation in in situ N<sub>2</sub>O diffusion measurements. Laboratory incubations distinguished the gaseous N products of nitrification and denitrification. Nitrifying activity, ambient NO<sub>3</sub><sup>-</sup>, and nitrification N<sub>2</sub>O were positively correlated with percent base saturation. However, differences between watersheds in soil N substrate caused by presence of leguminous black locust in the disturbed watershed were confounded with differences in soil acidity. Denitrification was most strongly affected by soil moisture, which in turn was determined by precipitation events and slope position. Gaseous N losses from well-drained midslope and toeslope landscape positions appeared to be minor relative to other N transformations. Favorable conditions for denitrification occurred at a poorly drained site near the stream of the disturbed watershed. Laboratory incubations revealed high rates of NO<sub>3</sub><sup>-</sup> reduction in these soils. We speculate that the riparian zone is a major site of depletion of NO<sub>3</sub><sup>-</sup> from the soil solution via denitrification.**

Studies of gaseous nitrogen losses from agricultural systems are abundant, but few studies have addressed losses from forested ecosystems (5, 12, 17, 21). Difficulty in carrying equipment into forested sites has hampered these efforts, but the paucity of studies in forested relative to agricultural ecosystems also results from differences in the economic value of determining the fate of N. However, as intensive silviculture emulates agriculture by becoming increasingly reliant on N fertilizers, an understanding of the processes regulating nitrification and denitrification in forested ecosystems will become valuable.

Environmental parameters which affect nitrification and denitrification have been identified in laboratory studies (4, 6, 16). The relative importance of these parameters has been investigated in field studies of agricultural and grassland systems (14, 18, 19). The objective of this study was to examine the relative importance of selected environmental parameters on gaseous N losses in forested ecosystems. Specifically, the effects of soil temperature, acidity, redox potential, moisture, inorganic N, total N, available C, and total C were examined.

## MATERIALS AND METHODS

**Study sites and sampling regime.** Study sites at the U.S. Department of Agriculture Forest Service Coweeta Hydrologic Laboratory are described in detail in the accompanying paper (2). Two watersheds (WSs) were studied. (i) WS18 (reference WS) supported an uneven-aged, aggrading, mixed-hardwood forest that had not undergone major disturbance since the chestnut blight in the late 1920s. (ii) WS6 (disturbed WS) supported an 18-year-old mixed hardwood

forest dominated by declining black locust (*Robinia pseudo-acacia* L., Fabaceae); disturbance history on WS6 includes clear cutting, liming, fertilization, conversion to grassland, herbicide treatments, and a recent infestation by the locust stem borer (*Megacyllene robiniae* Forster). Streambank and toeslope positions were identified on both WSs, and a midslope area dominated by black locust was also studied in the disturbed WS. Within each of these study areas, three soil sampling sites and adjacent frames for measuring N<sub>2</sub>O diffusion in situ were established. Because of spatial heterogeneity, the three sampling sites within each area were not considered true replicates in statistical analyses.

Soil sampling and in situ N<sub>2</sub>O measurements were carried out on a routine bimonthly basis from September 1984 through July 1985. More intensive sampling before and after a major precipitation event was conducted at the midslope locust site during July 1985.

**Gaseous N determinations.** In situ N<sub>2</sub>O diffusion was measured with closed covers, and estimates of nitrification N<sub>2</sub>O, denitrification N<sub>2</sub>O, and denitrification N<sub>2</sub> were determined from laboratory incubations under three levels of C<sub>2</sub>H<sub>2</sub> partial pressure and at field soil temperatures, as described in the companion paper (2). At each date and sampling site, 12 cores (2.5-cm diameter, 6 cm deep) were composited in plastic bags, returned to the laboratory on ice, and stored at 4°C. All analyses on field-moist soils were conducted within 1 week of sampling. Root fragments and pebbles were removed by hand to the extent practical.

**Edaphic parameters.** Soil temperature was measured with soil thermometers inserted to 6 cm below the mineral soil surface. Redox potential (E<sub>h</sub>) was determined in situ by constructing platinum electrodes (10), inserting the electrodes 3 cm below the mineral soil surface, placing a calomel reference electrode in contact with the soil, reading E<sub>h</sub> from a portable pH meter, and adding 244 mV to correct for the redox potential of the reference electrode. Measurements

\* Corresponding author.

† Paper no. 10424 of the journal series of the North Carolina Agricultural Research Service, Raleigh, NC 27695-7601.

‡ Present address: Department of Plant and Soil Biology, University of California, Berkeley, CA 94720.

TABLE 1. Multiple regression models<sup>a</sup>

Dependent Variable	Independent variable <sup>b</sup>	Cumulative R <sup>2</sup>
In situ N <sub>2</sub> O <sup>c</sup>	-0.326 + 0.047Temp + $\epsilon$	0.25
	-0.588 + 0.047Temp + 0.004BS + $\epsilon$	0.32
	-1.034 + 0.055Temp + 0.005BS + 0.006WFPS + $\epsilon$	0.43
Nitrification N <sub>2</sub> O <sup>c</sup>	-1.045 + 0.011BS + $\epsilon$	0.51
	-1.402 + 0.010BS + 0.028Temp + $\epsilon$	0.59
Denitrification N <sub>2</sub> O <sup>c</sup>	-1.455 + 0.022WFPS + $\epsilon$	0.66
	0.249 + 0.021WFPS - 0.003E <sub>h</sub> + $\epsilon$	0.71
Denitrification N <sub>2</sub> <sup>c</sup>	-1.325 + 0.020WFPS + $\epsilon$	0.44
	0.849 + 0.019WFPS - 0.004E <sub>h</sub> + $\epsilon$	0.50
NP	-29.22 + 0.92BS + $\epsilon$	0.44
	-65.93 + 0.98BS + 0.10available C + $\epsilon$	0.73
Ambient NO <sub>3</sub> <sup>-</sup>	0.653 + 0.091NP + $\epsilon$	0.58
	-1.388 + 0.090NP + 0.164Temp + $\epsilon$	0.62

<sup>a</sup> All models are statistically significant at  $\alpha = 0.01$ .

<sup>b</sup> All independent variables included are significant at  $\alpha = 0.01$ .

<sup>c</sup> Gaseous N estimate data are log transformed.

from seven electrodes were made per site. Electrodes were removed and cleaned between sampling dates.

All other measurements were made in the laboratory on the composite soil samples described above. Soil moisture was determined gravimetrically by drying for 24 h at 105°C. Ambient NH<sub>4</sub><sup>+</sup> and NO<sub>3</sub><sup>-</sup> were determined by duplicate extraction of 10-g field-moist soil samples in 50 ml of 2 M KCl and colorimetric analysis of filtered extracts with a Technicon Autoanalyzer. Available C was estimated by measuring CO<sub>2</sub> liberated during a 7-day aerobic incubation of triplicate 10-g field-moist soil samples at 20°C (E. A. Davidson, L. F. Galloway, and M. K. Strand, *Commun. Soil Sci. Plant Anal.*, in press). Nitrogen mineralization potentials and nitrification potentials were estimated from changes in NH<sub>4</sub><sup>+</sup> and NO<sub>3</sub><sup>-</sup> concentrations after 30-day aerobic incubations of duplicate 10-g field-moist soil samples at 20°C. Water-filled pore space (WFPS) was determined from gravimetric moisture content, bulk density, and an assumed 2.65 particle density (23). Standing water caused WFPS to exceed 100% on two occasions.

The remaining soil was air dried and passed through a 2-mm sieve. The air-dried, sieved samples collected throughout a year from a given site were then composited. pH was measured with a glass electrode, and exchangeable acidity, cation exchange capacity (CEC), and percent base saturation (BS) were determined with Mehlich III by the North Carolina Department of Agriculture, Agronomic Division, Soil Testing Laboratory. Total N was determined by using a micro-Kjeldahl digestion procedure (Coweeta files, unpublished manuscript). Total C was determined by combustion in a Leco total C analyzer, which uses infrared spectroscopy to measure liberated CO<sub>2</sub>.

**Statistical analyses.** Multiple regression models were evaluated by PROC STEPWISE of SAS (20). The three laboratory estimates of gaseous N production and in situ N<sub>2</sub>O measurements were considered as dependent variables. The edaphic parameters listed above were considered as independent variables. The factors regulating nitrifying activity were also examined by considering ambient NO<sub>3</sub><sup>-</sup> and nitrification potential as dependent variables with the remaining edaphic parameters as independent variables. Four criteria were employed for selecting the best models: (i)

maximizing the R<sup>2</sup> value, (ii) requiring significance at  $\alpha = 0.05$  for each independent variable, (iii) minimizing covariation among independent variables, and (iv) affording meaningful interpretations.

Site variability was investigated by analysis of variance with PROC GLM of SAS (20). Only data from the toeslope, mixed hardwood and midslope, locust areas of the disturbed WS were considered, using the following model:  $Y = \beta_0 + \text{date} + \text{area} + \beta_1 X(\text{area}) + \epsilon$ , where  $Y$  is one of the estimates of gaseous N, the date is the class variable for sampling date; the area is the class variable for sampling area (i.e., slope position),  $X(\text{area})$  is a continuous variable for an environmental parameter ( $X$ ) nested within the sampling area, and  $\epsilon$  is the error term.

## RESULTS

**In situ N<sub>2</sub>O diffusion.** Soil temperature, BS, and WFPS accounted for 43% of the variation in in situ N<sub>2</sub>O diffusion rates (Table 1). Exchangeable acidity and pH were strongly correlated with BS ( $r = -0.91$  and  $r = 0.98$ , respectively; significant at  $\alpha = 0.01$ ) and could be substituted for BS in all of the models in Table 1 with only a minor effect on the R<sup>2</sup> values. Reference WS sites, where BS was low, generally exhibited low N<sub>2</sub>O diffusion rates, and low temperatures generally reduced rates within sites (Fig. 1). Temperature reflects seasonal variation, whereas BS (or acidity) reflects site variation. Rates of N<sub>2</sub>O diffusion also increased as soil moisture increased across both sites and dates.

**Nitrification N<sub>2</sub>O.** Temperature and BS also accounted for 59% of the variation in the laboratory estimate of nitrification N<sub>2</sub>O (Table 1). In contrast to the model for in situ N<sub>2</sub>O, BS accounted for most of the variation; temperature was also significant, but WFPS did not significantly contribute to reduction in sums of squares. Nitrification N<sub>2</sub>O also showed significant positive correlations with ambient NO<sub>3</sub><sup>-</sup>, nitrification potential (NP), and mineralization potential (Table 2).

**Denitrification N<sub>2</sub>O and denitrification N<sub>2</sub>.** WFPS and E<sub>h</sub> accounted for 71% of the variation in denitrification N<sub>2</sub>O and 50% of the variation in denitrification N<sub>2</sub> (Table 1). The maximum observed rates of gaseous N production in the laboratory differed among processes, with denitrification

TABLE 2. Correlation coefficients (*r*) among gaseous N estimates and edaphic parameters

Parameter	<i>r</i>										
	WFPS	E <sub>h</sub>	BS	Temp	NO <sub>3</sub> <sup>-</sup>	NH <sub>4</sub> <sup>+</sup>	NP	Mineralization potential	Available C	Total C	Total N
In situ N <sub>2</sub> O	0.19	-0.04	0.33 <sup>a</sup>	0.54 <sup>a</sup>	0.51 <sup>a</sup>	-0.09	0.39 <sup>a</sup>	0.37 <sup>a</sup>	0.01	-0.14	-0.09
Nitrification N <sub>2</sub> O	0.18	-0.35 <sup>a</sup>	0.73 <sup>a</sup>	0.36 <sup>a</sup>	0.75 <sup>a</sup>	-0.32 <sup>a</sup>	0.54 <sup>a</sup>	0.50 <sup>a</sup>	-0.06	0.02	0.09
Denitrification N <sub>2</sub> O	0.81 <sup>a</sup>	-0.35 <sup>a</sup>	0.20	0.02	0.17	0.07	0.33 <sup>a</sup>	0.30 <sup>a</sup>	0.23 <sup>b</sup>	0.06	0.08
Denitrification N <sub>2</sub>	0.67 <sup>a</sup>	-0.36 <sup>a</sup>	-0.03	-0.03	-0.16	0.12	0.05	0.02	0.03	-0.11	-0.12
WFPS		-0.17	0.11	-0.18	0.18	0.19	0.37 <sup>a</sup>	0.36 <sup>a</sup>	0.35 <sup>a</sup>	0.14	0.15
E <sub>h</sub>			-0.45 <sup>a</sup>	-0.00	-0.19	0.10	-0.29 <sup>a</sup>	-0.25 <sup>b</sup>	-0.04	-0.10	-0.17
BS				0.06	0.71 <sup>a</sup>	-0.16	0.67 <sup>a</sup>	0.63 <sup>a</sup>	-0.06	-0.11	0.08
Temp					0.24 <sup>b</sup>	-0.09	0.10	0.11	0.06	-0.07	-0.07
NO <sub>3</sub> <sup>-</sup>						-0.19	0.77 <sup>a</sup>	0.74 <sup>a</sup>	0.26 <sup>b</sup>	0.25 <sup>b</sup>	0.33 <sup>a</sup>
NH <sub>4</sub> <sup>+</sup>							0.15	0.19	0.42 <sup>a</sup>	0.10	0.09
NP								0.98 <sup>a</sup>	0.24 <sup>b</sup>	0.33 <sup>a</sup>	0.44 <sup>a</sup>
Mineralization potential									0.59 <sup>a</sup>	0.35 <sup>a</sup>	0.48 <sup>a</sup>
Available C										0.73 <sup>a</sup>	0.72 <sup>a</sup>
Total C											0.92 <sup>a</sup>

<sup>a</sup> Statistically significant at α = 0.01.  
<sup>b</sup> Statistically significant at α = 0.05.

N<sub>2</sub>O and denitrification N<sub>2</sub> being 1 and 2 orders of magnitude higher than nitrification N<sub>2</sub>O, respectively. Denitrification N<sub>2</sub>O was positively correlated with ambient NO<sub>3</sub><sup>-</sup>, with the exception of one site that exhibited the highest laboratory denitrifying activity and low levels of ambient NO<sub>3</sub><sup>-</sup> (Fig. 2).

**Nitrifying activity.** Most of the observed variation in NPs could be accounted for by variation in BS and available C (Table 1). Disturbed WS soils, which were high in BS, exhibited high NPs. However, large residuals in the NP versus BS regression for disturbed WS samples (Fig. 3A, sites A through I) indicate that another variable related to seasonal differences also affected NP. Variation in available C reflected both slope position (stream site soils having higher levels) and seasonal changes in soil moisture (available C increased with increasing soil moisture at all sites). The reference WS soils (sites J through O) exhibited low NP regardless of available C, whereas the disturbed WS soils (sites A through I) exhibited a significant correlation between NP and available C (Fig. 3b). Temperature and NP accounted for 62% of the variation in ambient NO<sub>3</sub><sup>-</sup> (Table 1).

**Site variability.** Understory vegetation and ground cover varied within each sampling area. Patches of fescue, appar-

ently residual from grassland conversion, were particularly notable on the disturbed WS. In both the midslope, locus and the toeslope, mixed hardwood sampling areas, one of the three sampling sites was located on patches of fescue. These fescue sites (sites C and E in all figures) consistently exhibited the highest in situ diffusion measurements and the highest laboratory N<sub>2</sub>O estimates for both areas. When variation due to season and overstory vegetation were partitioned out, variation in BS within these two sampling areas was significantly correlated (α = 0.05) with both in situ N<sub>2</sub>O and denitrification N<sub>2</sub>O variation. Soils supporting patches of fescue had the highest BS of all sites studied.

**Effects of precipitation.** The effects of a major precipitation event were studied after a dry period in July at the midslope, locust area, where high rates of N<sub>2</sub>O diffusion had been observed during other seasons. In situ N<sub>2</sub>O diffusion was determined before the rain, the morning after 11 mm of precipitation, 6 days later after 51 mm more precipitation

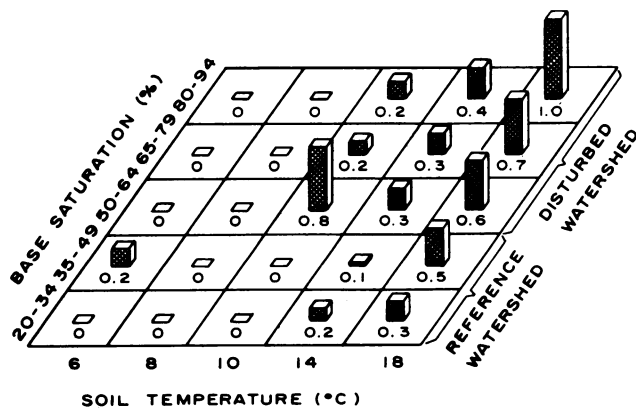


FIG. 1. Effects of temperature and base saturation on rates of in situ N<sub>2</sub>O diffusion. Bars represent means of transformed [log (X + 1)] diffusion rates for temperature-BS combinations.

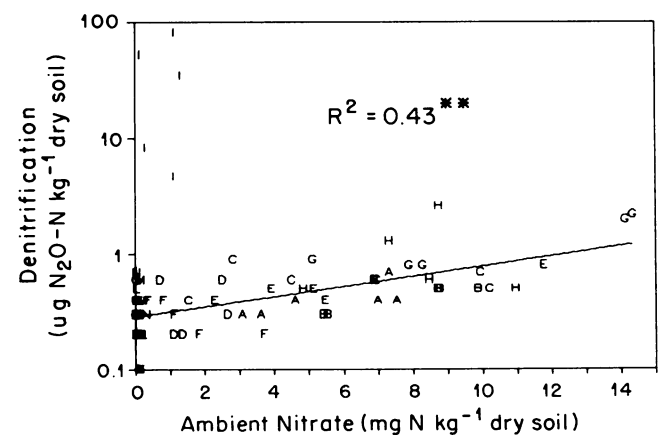


FIG. 2. Regression of denitrification N<sub>2</sub>O estimate versus ambient NO<sub>3</sub><sup>-</sup> concentration. Measurements were taken from September 1984 through July 1985 from the midslope, locust area of the disturbed WS (sampling sites A, B, and C), toeslope, hardwood area of the disturbed WS (D, E, and F), stream area of the disturbed WS (G, H, and I), toeslope area of the reference WS (J, K, and L), and the stream area of the reference WS (M, N, and O). R<sup>2</sup> was significant at α = 0.01.

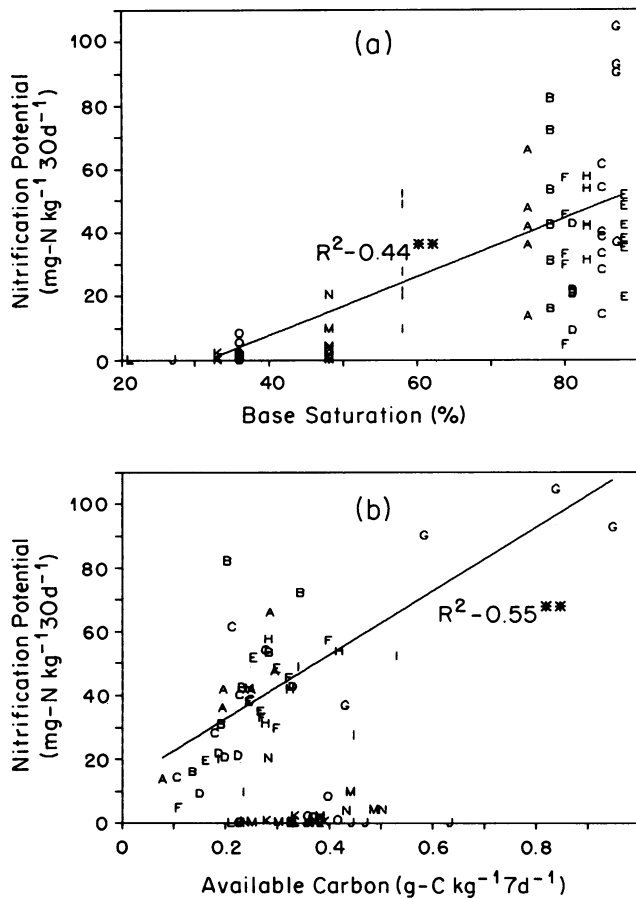


FIG. 3. Regressions of NP versus BS (a) and available carbon (b). The regression line for NP versus available C was calculated for disturbed WS samples only (sites A through I). Symbols are the same as in Fig. 2.  $R^2$  was significant at  $\alpha = 0.01$  for both panels.

during that period, and 3 days after the rain stopped. Rates of  $N_2O$  diffusion increased dramatically immediately after the precipitation. The high rate remained through the following 6-day rainy period and declined to prerin levels within 3 days after the rain ceased (Fig. 4). Laboratory estimates of both nitrification  $N_2O$  and denitrification  $N_2O$  were also higher for soils sampled immediately after the precipitation.

## DISCUSSION

Observational data must be interpreted in light of the range of sites and variables considered. For example, Melillo et al. (12) found a correlation between ambient  $NO_3^-$  and denitrification potentials across a series of well-drained sites that represented a successional chronosequence. We found a similar relationship across well-drained sites that exhibited low denitrification rates, but the highest denitrification rates occurred at a poorly drained site with low ambient  $NO_3^-$  (Fig. 2). Although  $NO_3^-$  is important as a denitrification substrate, soil aeration appears to be extremely important in controlling the rate at which  $NO_3^-$  is reduced. Therefore, we suggest that slope position and its effect on soil drainage within a WS must be considered in addition to differences among WSs. The regulating influence of soil moisture as affected by precipitation (19) and irrigation (18) has been well demonstrated for denitrification in agricultural systems. We

have demonstrated a similar dominating effect of soil moisture as affected by slope position in a forested ecosystem.

Regardless of WS management history, the highest rates of N loss due to denitrification are likely to be found in riparian zones. Not only are soils wetter near the stream than upslope (9), but also cumulation enriches upper horizons of riparian zone soil with organic matter and thus increases the potential for microbial respiration and oxygen depletion. Although  $E_h$  and WFPS are usually related,  $E_h$  can reflect differences in  $O_2$  consumption as well as  $O_2$  diffusion. Therefore, both WFPS and  $E_h$  were included in the models accounting for most of the variation observed in denitrification estimates (Table 1).

The overwhelming effect of WFPS may have masked the effect of NP on denitrifying activity which has been previously observed (17). Although well-drained sites exhibited a wide range of NPs, we observed very low laboratory denitrification rates in these soils. Therefore, the opportunity to correlate these two parameters across this data subset was poor. The correlation between NP and denitrification  $N_2O$  was significant, but the  $r$  value was low (Table 2).

Soil moisture also affected many other edaphic parameters that relate to microbial activity. Significant positive correlations between WFPS and available C, NP, and mineralization potential (Table 2) suggest that soil moisture not only affects  $O_2$  diffusion, but also directly stimulates microbial activity. The September 1984 sample date followed a long dry period, resulting in the very low values observed for these parameters. In light of this covariation of edaphic parameters, it is difficult to identify regulating mechanisms for microbial processes. For example, the inclusion of available C in the model for NP (Table 1) may result from sensitivity of the available C bioassay to moisture differences rather than any particular role that available C plays in the nitrification process.

The importance of black locust in the N cycle of successional eastern hardwood forests has been demonstrated by symbiotic N fixation estimates of  $30 \text{ kg of N ha}^{-1} \text{ year}^{-1}$  (1). Presumably, rapid turnover of N-rich black locust tissue caused by the stem borer infestation would also influence microbial N transformations. Evidence for this effect can be seen in an 80-fold increase in mean annual hydrologic export of  $NO_3^-$  from the disturbed WS relative to the reference WS. Mineralization of black locust tissue probably provides sufficient  $NH_4^+$  to permit high rates of nitrification in the

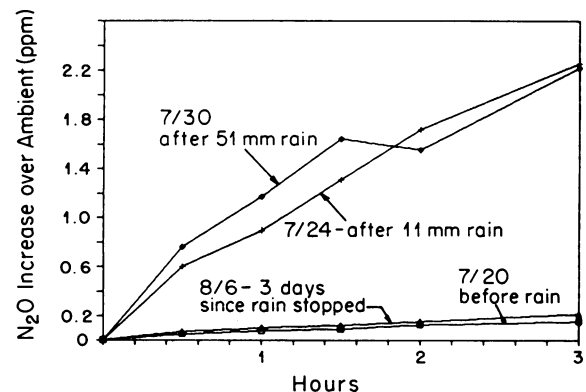


FIG. 4. Intensive sampling of in situ  $N_2O$  diffusion before and after a precipitation event in July 1985 at the midslope, locust area of the disturbed WS. Concentrations of  $N_2O$  within the soil cover are plotted against the time the cover was in place.

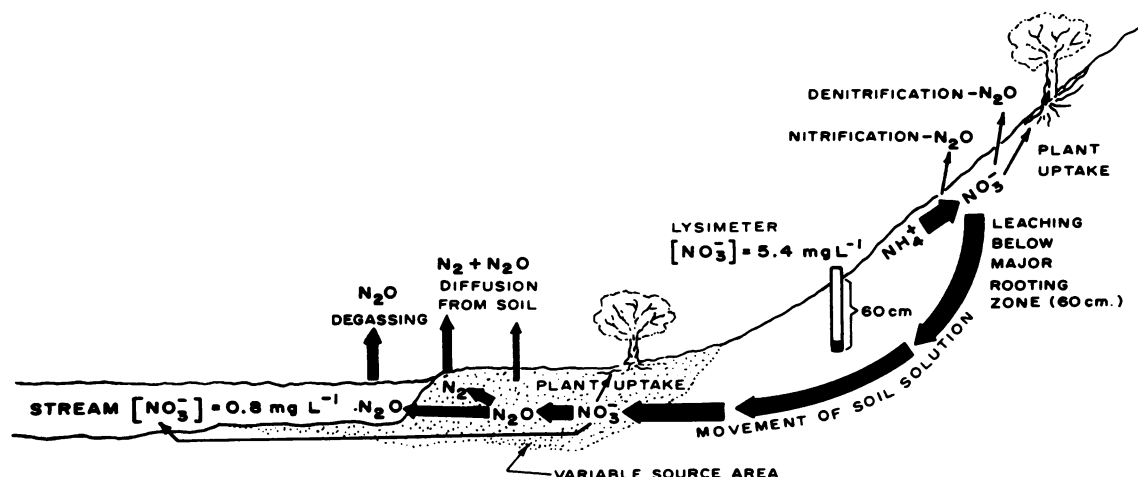


FIG. 5. Proposed relative importance of gaseous N loss pathways in the disturbed WS. Thickness of arrows corresponds to relative flux magnitude. Data for lysimeter  $\text{NO}_3^-$  solutions are from Montagnini (Ph.D. thesis).

disturbed WS, whereas intense competition among plants and microorganisms for inorganic N probably precludes nitrification in the mature, aggrading reference WS (13, 22). However, differences in BS across sites could also account for most of the observed variability in nitrifying activity (Table 1). Indeed, acidity appears to be important both between and within WSs, since variation in BS across sampling sites within the well-drained areas of the disturbed WS was correlated with in situ  $\text{N}_2\text{O}$  measurements and denitrification  $\text{N}_2\text{O}$  estimates. Unfortunately, differences in successional development and vegetation type between the two WSs are confounded with differences in soil acidity. Therefore, the effects of acidity and black locust mortality upon nitrification and denitrification cannot be separated. Both factors probably affect relative rates of N transformations.

Nitrifying activity and  $\text{N}_2\text{O}$  production from nitrification are obviously related, but the effect of acidity on  $\text{NO}_3^-$  versus  $\text{N}_2\text{O}$  production by nitrifying bacteria appears to differ. Whereas autotrophic nitrification in general is favored by circumneutral conditions (16),  $\text{N}_2\text{O}$  production by nitrifying bacteria may be enhanced by acidic conditions (11). At first, it appears that the results reported here are contradictory to those of Martikainen (11), since our estimates of nitrification  $\text{N}_2\text{O}$  are positively correlated with BS. However, we again emphasize that our results are observational across soils exhibiting a broad range of nitrifying activity. Nitrification  $\text{N}_2\text{O}$  increased as NPs increased across sites. Differences in nitrifying activity across sites appeared to be more important than differences in factors that affect  $\text{NO}_3^-/\text{N}_2\text{O}$  ratios produced by nitrification. In contrast, Martikainen experimentally manipulated a single soil sample, so that only those factors affecting the  $\text{NO}_3^-/\text{N}_2\text{O}$  ratio varied.

Similarly, the present study revealed no clear relationship between nitrification  $\text{N}_2\text{O}$  and WFPS, whereas the proportion of  $\text{NH}_4^+$  oxidized by nitrifiers which becomes reduced to  $\text{N}_2\text{O}$  should increase with increasing soil moisture (6) and decreasing  $\text{O}_2$  partial pressure (8). Again, although this relationship is probably true for a given soil sample, differences in overall nitrifying activity across sites in the present study appeared to affect  $\text{N}_2\text{O}$  production more than did differences in factors affecting ratios of  $\text{NO}_3^-$  production to  $\text{N}_2\text{O}$  production.

In situ  $\text{N}_2\text{O}$  diffusion was strongly affected by soil temperature, but laboratory estimates of denitrification were not correlated with incubation temperature. The temperature effect, however, is confounded with moisture differences. Winter months at Coweeta provide both the lowest temperatures and the wettest soil conditions. Diffusion of gases is restricted by water-saturated pore spaces, and  $\text{N}_2\text{O}$  solubility increases with decreasing temperature, thus contributing to the low diffusion rates observed in winter. Denitrification, however, is enhanced by wet conditions but retarded by low temperatures (3). In situ measurements in the present study cannot distinguish between the processes of  $\text{N}_2\text{O}$  diffusion and  $\text{N}_2\text{O}$  production. In contrast, laboratory incubations of small soil samples at simulated field temperatures provide estimates of biological activity under laboratory conditions with only minimal errors associated with restriction of diffusion. Laboratory rates of biological denitrification in very wet soil samples collected in December and incubated at  $8^\circ\text{C}$  were similar to rates observed in drier soil samples collected in July and incubated at  $18^\circ\text{C}$ . Apparently, the effects of temperature and moisture roughly cancelled each other. Although we cannot demonstrate that similar high rates of denitrification were occurring in situ during winter months with the  $\text{N}_2\text{O}$  failing to diffuse to the soil surface, this speculation is consistent with observations of high rates of  $\text{N}_2\text{O}$  diffusion during spring thaws in Wisconsin (5, 7).

Diffusion of  $\text{N}_2\text{O}$  from upslope positions increased after a precipitation event, but the effect on these well-drained soils was short lived (Fig. 4). If extrapolated across an entire hectare and throughout an entire year, the highest observed rate of  $\text{N}_2\text{O}$  diffusion from a well-drained site after the precipitation event studied would be equivalent to  $15 \text{ kg of N ha}^{-1} \text{ year}^{-1}$ . Since losses are 1 to 2 orders of magnitude lower during drier periods during the summer and colder periods during the winter, annual losses from these well-drained sites via  $\text{N}_2\text{O}$  diffusion are probably at least 1 order of magnitude lower than the highest projected rate, thus rendering  $\text{N}_2\text{O}$  loss by diffusion insignificant relative to other N transformations in this disturbed ecosystem. The highest observed rate of  $\text{N}_2\text{O}$  diffusion from any of the reference WS sites, when similarly extrapolated, was equivalent to  $0.4 \text{ kg of N ha}^{-1} \text{ year}^{-1}$ . Actual gaseous N losses from this middle-aged, aggrading forest are also probably at least 1 order of magnitude below this maximum estimate. Laboratory incu-

bations indicated that  $N_2$  production was not important in the reference WS or in well-drained soils of the disturbed WS.

In contrast, gaseous N losses in riparian zones of the disturbed WS are more difficult to quantify and are potentially of greater magnitude. At the wettest sites,  $N_2$  may be the main product of denitrification as it was in our laboratory incubations. Furthermore, diffusion of  $N_2O$  may be impeded in wet soils so that  $N_2O$  does not escape to the atmosphere until soil water reaches the stream. Quantifying N losses in the riparian zone was not possible, but riparian zone soils from the disturbed WS exhibited high rates of denitrification in the laboratory. Since these same soil samples also exhibited very low ambient  $NO_3^-$  levels, denitrification in the riparian zone had apparently depleted soil  $NO_3^-$  relative to upslope concentrations.

Although hydrologic export of  $NO_3^-$  is elevated in the disturbed WS relative to the reference WS, stream water  $NO_3^-$  concentrations are still 1 order of magnitude lower than soil solution concentrations measured by porous cup lysimeters at 60 cm depth at midslope positions in the disturbed WS (F. Montagnini, Ph.D. thesis, University of Georgia, Athens, 1985). Either most of this  $NO_3^-$  must be taken up by plants as soil water passes through surface horizons near the stream—an unlikely process due to low oxygen availability for roots when the variable source area is saturated—or it must be reduced before entering the stream (Fig. 5). Low estimates of gaseous N production in toeslope and midslope soils and high estimates in stream-side soils support the hypothesis that the riparian zone is a major site of depletion of soil solution  $NO_3^-$  via denitrification. Similar conclusions have been reached for forested riparian zones that receive agricultural drainage (10, 15). The present study suggests that much of the  $NO_3^-$  produced within a disturbed forested ecosystem is reduced in the riparian zone.

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