

Structure of the membrane proximal oxidoreductase domain of human Steap3, the dominant ferrireductase of the erythroid transferrin cycle

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The daily production of 200 billion erythrocytes requires 20 mg of iron, accounting for nearly 80% of the iron demand in humans. Thus, erythroid precursor cells possess an efficient mechanism for iron uptake in which iron loaded transferrin (Tf) binds to the transferrin receptor (TfR) at the cell surface. The Tf:TfR complex then enters the endosome via receptor-mediated endocytosis. Upon endosomal acidification, iron is released from Tf, reduced to Fe²⁺ by Steap3, and transported across the endosomal membrane by divalent metal iron transporter 1. Steap3, the major ferrireductase in erythrocyte endosomes, is a member of a unique family of reductases. Steap3 is comprised of an N-terminal cytosolic oxidoreductase domain and a C-terminal heme-containing transmembrane domain. Cytosolic NADPH and a flavin are predicted cofactors, but the NADPH/flavin binding domain differs significantly from those in other eukaryotic reductases. Instead, Steap3 shows remarkable, although limited homology to FNO, an archaeal oxidoreductase. We have determined the crystal structure of the human Steap3 oxidoreductase domain in the absence and presence of NADPH. The structure reveals an FNO-like domain with an unexpected dimer interface and substrate binding sites that are well positioned to direct electron transfer from the cytosol to a heme moiety predicted to be fixed within the transmembrane domain. Here, we discuss possible gating mechanisms for electron transfer across the endosomal membrane.

iron transport | ferric | erythrocyte

Iron plays an integral role in many biochemical processes essential to life. However, unsequestered Fe²⁺ is deleterious, because it catalyzes production of the hydroxyl free radical via the Fenton reaction (1). For these reasons, iron homeostasis is critical in human health, and diseases of iron transport and metabolism are among the most prevalent causes of morbidity in humans (2). Iron deficiency is thought to affect more than one billion people worldwide, particularly pregnant women and young children (3, 4) and is largely due to dietary insufficiency or excess loss. In contrast, iron overload disorders, collectively known as hereditary hemochromatosis, are among the most frequent single gene disorders in humans (5, 6). The occurrence of a single disease associated allele, HFE^{C282Y}, is as high as 10% in individuals of Northern European descent (7). In homozygous individuals, progressive iron accumulation generates oxidative stress that results in significant cellular damage, with induction of inflammation and fibrosis that eventuates in hepatic cirrhosis, hepatocellular carcinoma, diabetes mellitus, cardiac insufficiency, and arthropathy (8). Consequently, the cellular machinery and mechanisms responsible for iron transport and homeostasis are worthy of significant investigation, because they may provide targets for pharmacological intervention to either promote or inhibit cellular iron uptake in a variety of human disorders.

The daily production of 200 billion erythrocytes requires ≈20 mg of iron, accounting for nearly 80% of the iron demand in humans (4). However, as the red blood cells senesce, they

undergo phagocytosis by macrophages, and much of the erythrocyte iron is recycled, drastically reducing the need for dietary uptake of additional iron (9). To meet their iron need, erythroid precursor cells are uniquely dependent on the transferrin cycle (4, 10). In this cycle, ferric (Fe³⁺) iron-loaded transferrin (Tf) binds to the transferrin receptor (TfR-1) on the cell surface. The Tf:TfR-1 complex then enters the endosome via receptor-mediated endocytosis. Within the endosome, iron is released from Tf and then is reduced from Fe³⁺ to Fe²⁺ by Steap3, permitting transport across the endosomal membrane by divalent metal iron transporter 1 (DMT1), which is selective for Fe²⁺. The apo-Tf:TfR-1 complex is then recycled to the cell surface, where, at neutral pH, the apo-Tf is released to participate in the cycle once again (4, 10).

Ohgami *et al.* (11, 12) recently demonstrated that Steap3, a member of a unique family of transmembrane reductases, is the major erythroid ferrireductase. Steap (six transmembrane epithelial antigen of the prostate) family members contain a C-terminal domain composed of six transmembrane helices and are thought to coordinate a single intramembrane heme via two conserved histidine residues (12–14). With the exception of Steap1, the transmembrane domain is accompanied by an N-terminal oxidoreductase domain predicted to lie against the cytosolic face of the membrane (13). Accordingly, Steap2 and Steap4 also exhibit ferrireductase activity *in vitro* and, along with Steap3, have been shown to stimulate uptake of non-transferrin-bound iron and copper (13). These activities strongly suggest a greater role for Steap proteins in iron and copper metabolism.

Not surprisingly then, Steap proteins appear to play important roles in human health. Steap1 is expressed in many human cancer cell lines (15) and, with Steap2, is found at particularly high levels in prostate cancer (16), making Steap1 an appealing target for cancer immunotherapy (15–19). Mice lacking Steap3 exhibit hypochromic microcytic anemia (11, 12). Although Steap4^{-/-} knockout mice develop spontaneous metabolic disease on a regular diet, manifesting insulin resistance, glucose intolerance, mild hyperglycemia, dyslipidemia, and fatty liver disease (20), reminiscent of “metabolic syndrome” in humans (20–22).

The Steap family oxidoreductase domain differs significantly from those of the bacterial Fre, yeast FRE, or plant FRO families of transmembrane metallo-reductases. In particular,

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Data deposition: The coordinates and structure factors have been deposited in the Protein Data Bank, www.pdb.org (PDB ID codes 2VNS and 2VQ3).

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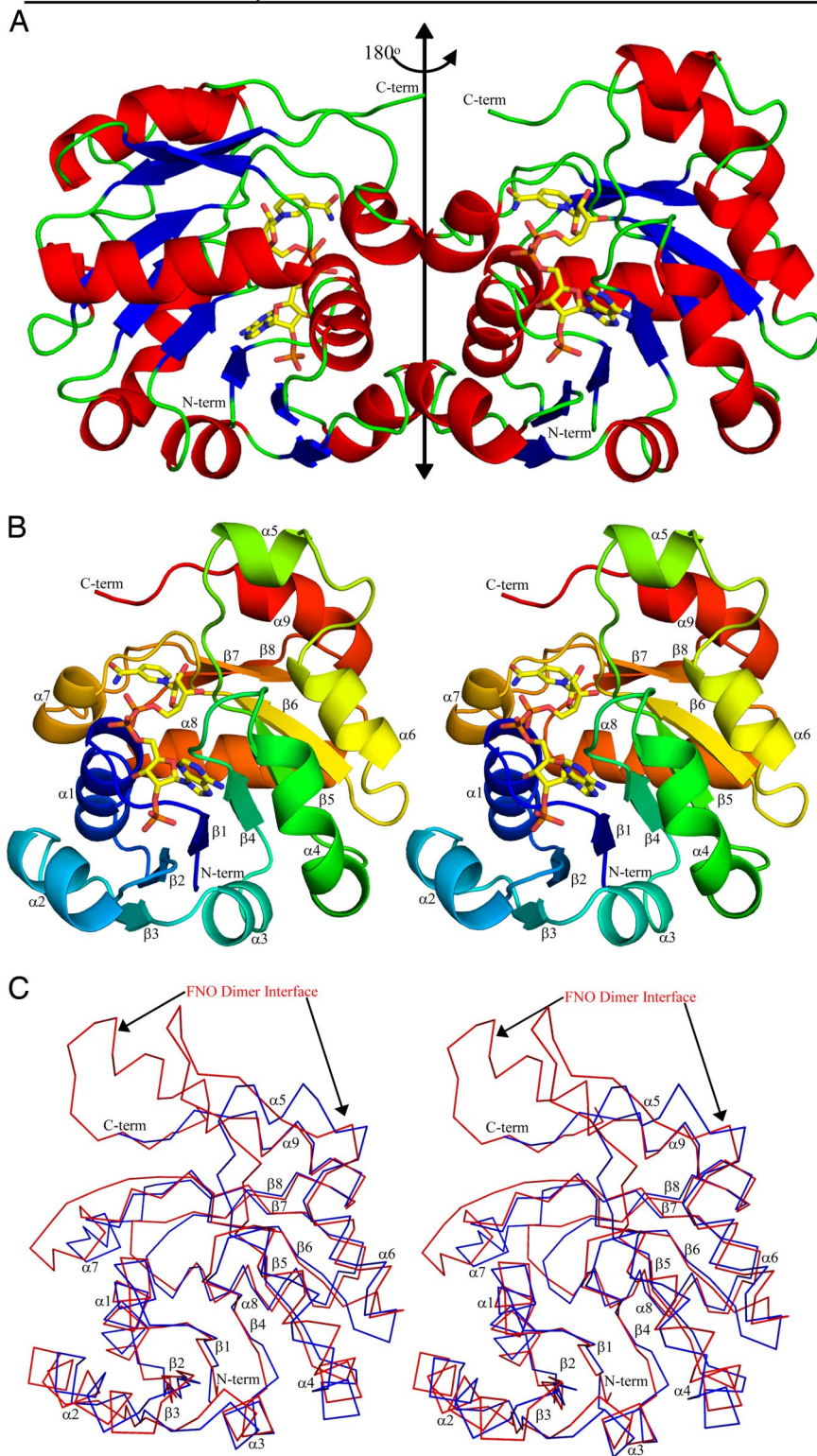


Fig. 1. Steap3 Structure. (A) The structure of the oxidoreductase dimer is depicted with α -helices in red, β -strands in blue, and connecting loops in green. The twofold axis runs vertically within the plane of the paper (double headed arrow). The truncated C termini, which must connect to the C terminal transmembrane domain, are in green at the top of the structure. NADPH (C, yellow; N, blue; O, red; and P, orange) runs up the front side of the right subunit (back side of the left subunit) with the adenine-ribose-2' phosphate moieties near the bottom and the nicotinamide ring near the top. (B) Stereo figure of the Steap3 subunit with labeled secondary structural elements. A color gradient runs from the N terminus (blue) to the C terminus (red). Note the proximity of the NADPH binding site to the dimer interface. (C) Stereo figure depicting the superposition of FNO on the Steap3 protomer. The Steap3 C_o trace is in blue, and FNO in red. The approximate location and extent of the FNO dimer interface is indicated by black arrows along the top of FNO. In contrast, the Steap3 interface is formed by $\alpha 7$, $\alpha 1$, and the C-terminal end of $\alpha 2$, which is significantly shorter in Steap3. Relocation of the dimer interface, combined with the shorter $\beta 5$ and $\alpha 9$ elements, allow the Steap3 NADPH binding site to approach the membrane.

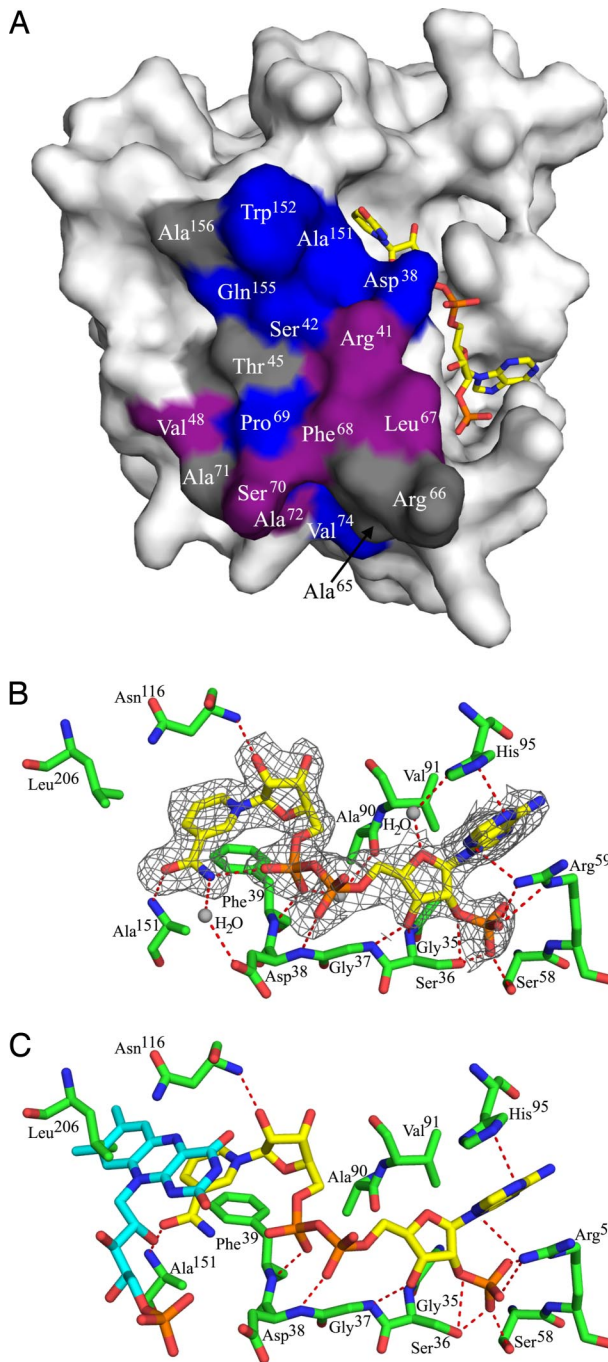


Fig. 2. Dimer interface and substrate binding sites. (A) Relative to Fig. 1, the Steap3 subunit has been rotated 90° about the twofold axis to expose the dimer interface. Strictly conserved residues at the dimer interface are shown in blue, conservative substitutions in violet, and variable residues in gray. (B) The orientation is intermediate between that of Figs. 1B and 3A, with NADPH colored as in Fig. 1. For details, see *NADPH Binding Site* under Results. Note that Asp³⁸ and Ala¹⁵¹ also serve as elements of the dimer interface. (C) FMN from biliverdin IX-beta reductase (PDB entry 1HE4) is docked to the Steap3-NADPH structure by superposition of the BVR-B nicotinamide ring onto that of the Steap3-NADPH structure. The isoalloxazine ring of the docked FMN clashes with the side chain of Leu²⁰⁶. FMN is colored similar to NADPH, but with carbons in cyan.

heme. Interestingly, we identify two hydrogen bonds to the amide group of the nicotinamide ring. The first is an intramolecular interaction within NADPH as the amide NH reaches

back to interact with the proximal phosphoryl group. The second is between the main chain NH of Ala¹⁵¹ and the amide oxygen of the nicotinamide ring (Fig. 2B). These hydrogen bonds appear to stabilize the amide oxygen in a rather unusual configuration (25), with the amide oxygen trans to C₄ of the nicotinamide ring. This trans configuration is also seen in FNO (23), and, in both Steap3 and FNO, the amide edge of the nicotinamide ring faces away from the protein and toward the solvent. The mode of binding is similar in other ways to that observed for other members of the dinucleotide binding family. The diphosphate moiety caps the N-terminal end of helix α 1 (Fig. 1B), interacting with the main chain NH of Ser³⁶, Gly³⁷, Asp³⁸, and Phe³⁹, which are present within the GXGXXG/A motif (Fig. S1). The 2' O of the nicotinamide ribose is caught up in a hydrogen bond to Asn¹¹⁶, whereas the 2' phosphate that differentiates NADPH from NADH is recognized by Ser⁵⁸ and Arg⁵⁹, and the adenine ring is sandwiched in between Arg⁵⁹ and His⁹⁵ (Fig. 2B).

Putative Flavin Binding Site. The *A. fulgidus* FNO structure was also determined as a non-Michaelis complex with NADP⁺ and F₄₂₀ (PDB entry 1JAY). The structure of this ternary complex allows docking of F₄₂₀ to Steap3 by structural superposition, and places the *Si*-face of the deazaflavin ring immediately “above” the *Si*-face (B-face) of the Steap3 nicotinamide ring, i.e., between the nicotinamide moiety and the predicted position of the lipid bilayer. However, this docking results in steric clash between the deazaflavin ring of F₄₂₀ and the side chain of Leu²⁰⁶. This led us to consider whether F₄₂₀ was the most suitable cofactor on which to model the interaction of Steap3 with FMN or FAD and led us to search the Protein Data Bank (PDB) for Steap3 homologs present as a ternary complex with a true flavin and a nicotinamide-derived cofactor. This search identified the ternary complex of Human Biliverdin IX Beta Reductase (BVR-B), FMN, and NADP⁺ (PDB entry 1HE4) (26). Although BVR-B, also known as flavin reductase, shows only modest structural similarity to Steap3 [rmsd = 2.954 Å for 113 C_α atoms with 17.7% identity, SSM Q score = 0.1738 (24)], the relative orientation of the isoalloxazine and nicotinamide rings are quite similar to those of the corresponding cofactors in FNO. Thus, BVR-B based modeling also predicts steric clash between Leu²⁰⁶ and the isoalloxazine ring of FAD or FMN (Fig. 2C). This may be an artifact of the Steap3 truncation, which disconnects the oxidoreductase and intramembrane domains. The disconnect may allow Leu²⁰⁶, the last well ordered C-terminal residue, to collapse in toward the active site. Alternatively, the position of Leu²⁰⁶ may reflect the situation in the intact protein, and conformational changes associated with electron gating might reorient Leu²⁰⁶, permitting the flavin to access the active site and the strictly conserved Leu²⁰⁶ side chain to lie over the top of the isoalloxazine ring. In this light, the Steap3 structures presented here might represent an inactive conformation. However, other than the steric clash with Leu²⁰⁶, the docked isoalloxazine ring would appear to be in an ideal position to mediate electron transfer between NADPH and the intramembrane heme, while, at the same time, serving to throttle-down from a two-electron process to a one-electron process (Figs. 2C and 3).

Interestingly, the ribitol and phosphate moieties of the docked F₄₂₀ and FMN cofactors project from the active site, closely approaching the neighboring Steap3 subunit (Fig. 3). Although the terminal γ -glutamyl-glutamate moiety of F₄₂₀ was not observed in the FNO structure, the position of the ribitol-phosphate moiety suggests it would clash with the neighboring Steap3 subunit. Thus, the adenine-ribose moiety of a similarly docked FAD molecule also shows steric clash with the neighboring Steap3 subunit. For this reason, FMN may be the preferred flavin for Steap3. However, an alternative FAD conformation that might allow the adenine-ribose moieties to bind along the subunit interface is a distinct possibility. Alter-

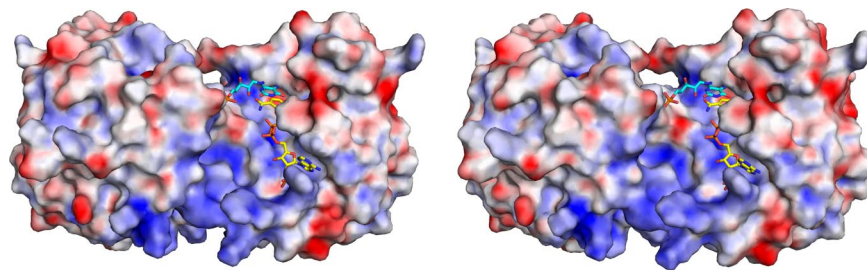


Fig. 3. The electrostatic potential mapped to the surface of the Steap3 dimer. Positive potentials are in blue, negative potentials in red (± 25 kT/e). Steap3-NADPH and the docked FMN from BVR-B are also shown. The orientation of the dimer and colors for NADPH are as in Fig. 1. FMN is similarly colored, but carbons are in cyan. Note the cleft running up the front side and across the top of the dimer interface. Because of the symmetry of the dimer, the cleft continues along the interface down the back side of the dimer.

natively, dissociation of the Steap3 dimer might result in exposure of an FAD binding site; or use of an altogether different cofactor might be indicated by the steric clash of the isoalloxazine ring with Leu²⁰⁶.

Proximity of Substrate Binding Sites to the Membrane. The relative locations of the membrane proximal face of Steap3 and the location of the NADPH binding site provides additional insight into the structural differences between Steap3 and FNO. Specifically, Steap3 utilizes Pro²⁰⁵ to truncate the C-terminal α -helix ($\alpha 9$). Our model places this helix on the membrane proximal face of Steap3, where it extends at an oblique angle toward the membrane. Similarly, the β -hairpin extension in FNO is replaced by the more compact $\alpha 5$ helix in Steap3 (Fig. 1C). These differences allow the nicotinamide ring of Steap3 to more closely approach the membrane surface, thereby facilitating electron transfer to the intramembrane heme, presumably via the isoalloxazine ring of the flavin.

Discussion

Data from size exclusion chromatography clearly indicate two different oligomeric forms of the Steap3 oxidoreductase domain, most likely an equilibrium mixture of monomer and dimer. Approximately equal concentrations of the two species are resolved on the sizing column when Steap3 is applied at concentrations of 100–200 μ M. This provides an approximate value for the dissociation constant if a negligible off rate over the time course of the chromatography experiment is assumed. However, full-length protein may exhibit an even tighter interaction. Not only may the C-terminal domain also participate in dimer formation, the orientation of the full-length protein within the membrane will result in the loss of rotational and translational degrees of freedom. Provided the C-terminal domain does not inhibit the interaction, the decreased entropy should strengthen it significantly.

Logically, the flow of electrons across the endosomal membrane should be a gated process; i.e., electron flow into the endosome is a direct response to ferric ions awaiting reduction. One possibility is that a physical interaction between Steap3 and endosomal Fe³⁺ might result in conformational changes within Steap3 that permit electron flow across the membrane. However, the structure of the Steap3 oxidoreductase domain suggests additional possibilities. For example, because of the proximity of the active site to the dimer interface, the oligomeric state of the enzyme might profoundly affect the structure of the catalytic site. Thus, the oligomeric state might represent a potential gating mechanism for electron transfer, serving to modulate the activity of the oxidoreductase domain. Further, Steap3 might function within a larger complex, with Steap3 activity coupled to the activity of other components of the iron transport process. The most obvious candidates are the divalent metal ion transporter

1 (DMT1) and the transferrin:transferrin receptor complex (Tf₂:TfR₂). We note with interest that the transferrin-transferrin receptor complex is also twofold symmetric, and a potential Steap3 homodimer, loaded with four electrons (two molecules of NADPH), nicely complements the 4 Fe⁺³ ions carried by the TfR₂:Tf₂ complex. In addition to specifying the correct stoichiometry between endosomal Fe³⁺ and cytosolic NADPH, the complex might also ensure that Steap3 is recruited into holotransferrin loaded endosomes. The mechanisms underlying reduction and uptake of non-transferrin bound iron and copper at the cell surface are less well understood, but it appears that Steap proteins will also play a role in this process, in which case regulation might occur with the assistance of other participating proteins.

Finally, as mentioned above, the nicotinamide ring is bound with a solvent exposed amide group, a rather uncommon orientation. In addition, the pocket harboring the nicotinamide ring opens upon a wider cleft that runs along the dimer interface (Fig. 3). The nature of the active site, combined with its proximity to the dimer interface, appears to be a unique feature of Steap3, one that may be shared with other members of the Steap family of metalloredoxases. These features might be exploited for the development of inhibitors that would specifically target the various members of the Steap metalloredoxase family.

Materials and Methods

Subcloning. The human *STEAP3* gene (National Center for Biotechnology Information accession no. BC042150) was obtained from the National Institutes of Health Mammalian Gene Collection. The N-terminal domain (residues 1–215) was amplified by overlapping PCR and inserted into Gateway entry vector pDONOR201. In addition to attB sites, the primers incorporated an N-terminal His₆-tag and a stop codon. The resulting entry clone was sequence verified and moved into destination vector pDEST14.

Expression and Purification of Steap3. Protein was expressed in BL21-CodonPlus (DE3)-RIL *Escherichia coli*, using standard protocols. The purification was done at 4°C. Cell pellets were thawed and resuspended in lysis buffer [10 mM Tris-HCl (pH 8.0), 150 mM NaCl, and 10 mM imidazole (pH 8.0)] at 5–6 ml/gm of wet pellet. PMSF (0.1 mM) was added to the cell suspension, and the cells were lysed by French press. The cleared lysate was applied to an HIS-select nickel affinity resin, washed with 10 column volumes of wash buffer (lysis buffer plus 10% glycerol), and eluted in elution buffer [10 mM Tris-HCl (pH 8.0), 150 mM NaCl, 200 mM imidazole (pH 8.0), 10% glycerol, and 0.5 mM β -mercaptoethanol]. Eluted protein was applied to a Superdex 75 gel filtration column equilibrated with 150 mM NaCl, 10% glycerol, 0.5 mM β -mercaptoethanol, and 10 mM Tris-HCl at (pH 8.0). The purified Steap3 was concentrated by centrifugal filtration to 10 mg/ml. Protein concentrations were determined by Bradford assay (27), using BSA as a standard.

Structure Determination. Steap3 was crystallized by hanging drop vapor diffusion, using equal volumes of protein and well solution [10 mM FeCl₃, 60–100 mM Na₃-citrate (pH 5.6), 4% Jeffamine M600, and 15% glycerol]. Crystals also grew in the presence of 1 mM NADPH. Crystals were flash-frozen in liquid nitrogen before data collection. All data were integrated, scaled, and reduced

in space group P2₁2₁2₁ (Table S1), using HKL-2000 (28). The unit cell constants suggested one or two (66% or 33% solvent) Steap3 protomers in the asymmetric unit. Although the self-rotation function failed to provide evidence for the dimer seen by size exclusion chromatography, a 50- σ peak in the native Patterson indicated a dimer in the asymmetric unit, with the twofold axis parallel to a crystallographic screw.

The structure of apo-Steap3 was solved by MIRAS. Heavy atom soaks were for 1 h in 10 mM K₂Pt(CN)₄ or 10 mM mersalyl acid. SOLVE (29) was used to determine heavy atom positions and to calculate initial phases. RESOLVE (30, 31) was used for density modification and initial model-building, followed by iterative model building and refinement with COOT (32) and REFMAC (33, 34), using NCS restraints and 22 TLS groups chosen by the TLS motion determination server (35, 36, 37).

The apo-Steap3 structure was used to phase the Steap3-NADPH structure.

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